A new synthesis of spiro-fused 1,4-benzothiazines

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A new preparative procedure was developed for the synthesis of spiro-fused 1,4-benzo-thiazines by thermal heterocyclization of S-benzyl and S-phenacyl derivatives of o-indothiophenol.

Key words: *o*-indothiophenol, benzothiazine.

ortho-Substituted quinone N-arylimines^{1,2} are promising chelating ligands³ and starting compounds for the preparation of previously unknown spiro-fused benzo-azines.⁴ However, mercapto-substituted quinone N-arylimines have been previously inaccessible because of the formation of S—S dimers under the standard conditions of condensation of o-aminothiophenol with quinones acting as oxidants.

Results and Discussion

We succeeded in preparing *o*-indothiophenol (1) from *o*-aminobenzenethiol with the use of diethyl malonate as the SH-protective group, which has previously been unknown in the traditional chemistry of thiols.⁵ The resulting sulfide was treated without isolation with

2,6-di-*tert*-butyl-1,4-benzoquinone (2) in MeOH to obtain the target compound 1 in 58% yield (Scheme 1).

The present study was aimed at employing intramolecular heterocyclization of S-methylene derivatives of o-indothiophenol for the synthesis of previously inaccessible spiro-fused benzothiazines. The starting quinone N-(o-benzylthio)phenylimines (3) were prepared (see Scheme 1) by either condensation of o-aminophenyl benzyl sulfides (4) with benzoquinone 2 (method A) or alkylation of o-indothiophenol (1) with benzyl halides (method B). The method A is advantageous because of its simplicity, higher yields of the target products, and the possibility of performing the one-pot synthesis.

However, the corresponding phenacyl sulfides cannot be prepared according to the method *A* because the reaction of *o*-aminobenzenethiol with *p*-bromophenacyl

Scheme 1

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bromide (5) is known⁶ to afford directly 3-p-bromophenyl-2H-1,4-benzothiazine (Scheme 2).

Scheme 2

$$Ar = Ph (a), 4-NO_2C_6H_4 (b)$$

Like analogous o-indophenol derivatives, 4 sulfides 3 are converted into spirobenzothiazines 6 upon heating (see Scheme 2). However, intramolecular cyclization of the thio analogs proceeds much more rapidly. Benzyl sulfide 3a undergoes cyclization on refluxing in toluene, whereas cyclization of 4-nitrobenzyl sulfide 3b proceeds even on refluxing in benzene. The formation of spirobenzothiazines 6 from sulfides 3 is evidenced by the appearance of two one-proton signals of the CH and NH groups instead of the two-proton signal of the methylene group. The reaction of o-indothiophenol (1) with phenacyl bromide 5 gave rise directly to spirobenzothiazine 7 (see Scheme 2). Its structure was confirmed by the fact that the signals for two tert-butyl groups and two protons of the cyclohexadiene ring are isochronous and also by the presence of one-proton singlet signals for the protons of the CH and NH groups.

The spirane structure of benzothiazine **6b** was unambiguously established by X-ray diffraction analysis (Fig. 1, Tables 1 and 2).

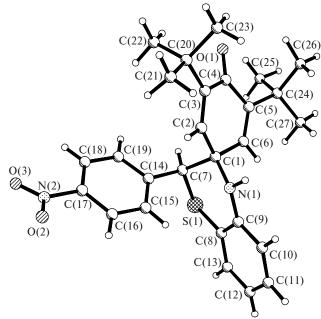


Fig. 1. Structure of molecule 6b.

Table 1. Principal bond lengths (d) in molecule **6b**

	7.18		1/9
Bond	d/Å	Bond	d/Å
S(1)-C(8)	1.762(2)	C(3)-C(4)	1.496(3)
S(1)-C(7)	1.821(2)	C(3)-C(20)	1.532(3)
N(1)-C(9)	1.392(3)	C(4)-C(5)	1.485(3)
N(1)-C(1)	1.458(3)	C(5)-C(6)	1.334(3)
N(2) - O(3)	1.215(3)	C(5)-C(24)	1.548(3)
N(2) - O(2)	1.231(3)	C(7)-C(14)	1.502(3)
N(2)-C(17)	1.458(3)	C(8)-C(13)	1.396(3)
O(1)-C(4)	1.229(2)	C(8)-C(9)	1.407(3)
C(1)-C(2)	1.493(3)	C(9)-C(10)	1.398(3)
C(1)-C(6)	1.509(3)	C(10)-C(11)	1.372(4)
C(1) - C(7)	1.565(3)	C(11) - C(12)	1.389(4)
C(2) - C(3)	1.338(3)	C(12)-C(13)	1.382(4)

Table 2. Principal bond angles (ω) in molecule 6b

Angle	ω/deg	Angle	ω/deg
C(8)-S(1)-C(7)	101.0(1)	O(1)-C(4)-C(3)	120.4(2)
C(9)-N(1)-C(1)	123.8(2)	C(5)-C(4)-C(3)	118.4(2)
O(3)-N(2)-O(2)	122.8(2)	C(6)-C(5)-C(4)	118.4(2)
O(3)-N(2)-C(17)	119.1(2)	C(6)-C(5)-C(24)	123.2(2)
O(2)-N(2)-C(17)	118.0(2)	C(4)-C(5)-C(24)	118.3(2)
N(1)-C(1)-C(2)	107.0(2)	C(5)-C(6)-C(1)	123.9(2)
N(1)-C(1)-C(6)	111.9(2)	C(14)-C(7)-C(1)	114.3(2)
C(2)-C(1)-C(6)	111.8(2)	C(14)-C(7)-S(1)	113.1(2)
N(1)-C(1)-C(7)	109.0(2)	C(1)-C(7)-S(1)	111.9(2)
C(2)-C(1)-C(7)	107.3(2)	C(13)-C(8)-S(1)	117.6(2)
C(6)-C(1)-C(7)	109.7(2)	C(9)-C(8)-S(1)	123.1(2)
C(3)-C(2)-C(1)	125.2(2)	N(1)-C(9)-C(10)	119.1(2)
C(2)-C(3)-C(4)	117.1(2)	N(1)-C(9)-C(8)	122.3(2)
C(2)-C(3)-C(20)	123.0(2)	C(19)-C(14)-C(7)	118.5(2)
C(4)-C(3)-C(20)	119.9(2)	C(15)-C(14)-C(7)	123.1(2)
O(1)-C(4)-C(5)	121.2(2)		

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the average $C(sp^2)-C(sp^3)$ bond length (1.507 Å).⁸ The O(1) atom of the carbonyl group forms two intermolecular C—H...O hydrogen bonds with the H atoms of the aryl rings: H(12)...O(1') (0.5 - y, x, 0.5 - z), 2.66(3) Å; C(12)...O(1'), 3.522(4) Å; the C(12)-H(12)...O(1') angle is $151(2)^\circ$, and H(16)...O(1'') (x, y, z - 1), 2.41(2) Å; C(16)...O(1''), 3.304(3) Å; the C(16)-H(16)...O(1'') angle is $153(2)^\circ$.

the C(3)—C(20) and C(5)—C(24) bonds are elongated

to 1.532(3) and 1.548(3) Å, respectively, compared to

The six-membered heterocycle adopts a half-chair conformation with the C(1) and C(7) atoms deviating from the plane through the remaining atoms of the ring by -0.337(4) and 0.418(3) Å, respectively. The aryl substituent at the C(7) atom is in the axial position (the C(8)–S(1)–C(7)–C(14) torsion angle is $-86.7(2)^{\circ}$) and is substantially rotated with respect to the C(7)–S(1) bond (the S(1)–C(7)–C(14)–C(15) torsion angle is $65.2(3)^{\circ}$). Such position of the substituent gives rise to the shortened intramolecular contacts: H(7)...H(19), 2.26(3) Å; N(1)...C(15), 3.097(3) Å (the sum of the van der Waals radii⁷ of the C and N atoms is 3.21 Å); C(8)...H(15), 2.80(2) Å; and C(9)...H(15), 2.71(2) Å.

The nitro group in the molecule **6b** is rotated with respect to the plane of the benzene ring (the O(2)-N(2)-C(17)-C(16) torsion angle is $22.9(4)^{\circ}$). The H(1) atom forms a bifurcated intermolecular hydrogen bond with the O atoms of the nitro group: H(1)...O(2''') (y - 0.5, 1 - x, 0.5 + z), 2.22(3) Å; N(1)...O(2'''), 3.077(3) Å; the N(1)-H(1)...O(2''') angle is $173(2)^{\circ}$; and H(1)...O(3'''), 2.64(2) Å; N(1)...O(3'''), 3.224(3) Å; the N(1)-H(1)...O(3''') angle is $127(2)^{\circ}$. In addition, the O(3) atom is involved in the formation of the intermolecular C-H...O hydrogen bond: H(18)...O(3'''') (-x+1, -y+1, -z), 2.55(3) Å; C(18)...O(3''''), 3.187(4) Å; the C(18)-H(18)...O(3'''') angle is $127(2)^{\circ}$.

Experimental

The 1H NMR spectra were recorded on a Bruker DPX-250 spectrometer (250 MHz) at 25 $^{\circ}C$ (CDCl $_3$) with Me $_4Si$ as the internal standard.

o-Indothiophenol (1). A solution of o-aminobenzenethiol (0.02 mol) in MeOH (15 mL) was added to a solution of KOH (0.02 mol) in MeOH (15 mL) and the reaction mixture was stirred for 0.5 h. Then a solution of diethyl bromomalonate (0.02 mol) in MeOH (15 mL) was added. The reaction mixture was brought to boiling and kept at ~20 °C for 1 h. Then benzoquinone 2 (0.02 mol) was added. The mixture was refluxed for 15 h and cooled. The precipitate that formed was filtered off, purified by column chromatography (Al₂O₃, benzene—light petroleum, 1:2), and recrystallized from MeOH. The yield was 58%, m.p. 58 °C. Found (%): C, 73.35; H, 7.69; N, 4.28. C₂₀H₂₅NOS. Calculated (%): C, 73.48, H, 7.55, N, 4.39. ¹H NMR, δ: 1.20 and 1.33 (both s, 9 H each, Bu^t); 6.65 (d, 1 H, H(2), J = 7.4 Hz); 6.77 and 7.09 (both d, 1 H each, H(2'), H(6'), J = 2.4 Hz); 7.10-7.22 (m, 2 H, H(3), H(4)); 7.34 (s, 1 H, SH); 7.67 (d, 1 H, H(5), J = 7.4 Hz).

3,5-Di-*tert*-butyl-*p*-benzoquinone *N*-(*o*-benzylthio)phenylimine (3a). *A*. A solution of *o*-aminobenzenethiol (0.02 mol) in MeOH (15 mL) was added to a solution of KOH (0.02 mol) in MeOH (15 mL). The reaction mixture was stirred for 0.5 h and a solution of benzyl chloride (0.02 mol) in MeOH (15 mL) was added. The reaction mixture was brought to boiling and kept at ~20 °C for 1 h. Then benzoquinone 2 (0.02 mol) was added and the mixture was refluxed for 5 h. The solvent was evaporated, the residue was purified by column chromatography (Al₂O₃, benzene—light petroleum, 1 : 2), and compound 3a was obtained in 80% yield, m.p. 98—100 °C (from MeOH).

B. o-Indothiophenol 1 (0.02 mol) was added to a solution of KOH (0.02 mol) in MeOH (35 mL) and the reaction mixture was stirred with gentle heating until the precipitate completely dissolved. Then benzyl chloride (0.02 mol) was added and the reaction mixture was refluxed for 5 h. The product was isolated as described above, the yield was 35%.

Found (%): C, 77.75; H, 7.39; N, 3.45. $C_{27}H_{31}NOS$. Calculated (%): C, 77.65; H, 7.48; N, 3.35. ¹H NMR, δ : 1.18 and 1.31 (both s, 9 H each, Bu^t); 4.10 (s, 2 H, CH₂); 6.63 (d, 1 H, H(2), J = 7.4 Hz); 6.70 and 7.08 (both d, 1 H each, H(2'), H(6'), J = 2.5 Hz); 7.10—7.35 (m, 8 H, H(3), H(4), H(5), Ph).

3,5-Di-*tert*-butyl-*p*-benzoquinone N-[o-(4-nitrobenzyl-thio)]phenylimine (3b). A. o-Aminobenzenethiol (0.02 mol) was alkylated with p-nitrobenzyl bromide in MeOH in the presence of KOH analogously to the synthesis of 3a. The reaction with benzoquinone 2 lasted for 4 h. After column chromatography (Al_2O_3 , benzene—light petroleum, 1:2), compound 3b was obtained in 76% yield, m.p. 153 °C.

B. Alkylation of *o*-indothiophenol **1** (0.02 mol) with *p*-nitrobenzyl bromide in MeOH in the presence of KOH was carried out as described for **3a**. The yield was 30%. Found (%): C, 70.31; H, 6.43; N, 6.18. $C_{27}H_{30}N_2O_3S$. Calculated (%): C, 70.10; H, 6.54; N, 6.06. ¹H NMR, δ: 1.17 and 1.32 (both s, 9 H each, Bu^t); 4.13 (s, 2 H, CH₂); 6.65 and 7.06 (both d, 1 H each, H(2'), H(6'), J = 2.4 Hz); 6.66 (d, 1 H, H(2), J = 6.8 Hz); 7.08–7.26 (m, 3 H, H(3), H(4), H(5)); 7.44 and 8.11 (both d, 2 H each, H(2"), H(6"), H(3"), H(5"), J = 8.7 Hz).

3′,5′-Di-tert-butyl-4′-oxo-2-phenyl-2,3-dihydrospiro(4*H*-1,4-benzothiazine-3,1′-cyclohexa-2′,5′-diene) (6a). A solution of sulfide 3a (0.005 mol) in toluene (15 mL) was refluxed for 5 h and concentrated. Then the residue was purified by column chromatography (Al_2O_3 , benzene). The product was dried and triturated with light petroleum until the crystalline precipitate formed. The yield was 78%, m.p. 175 °C. Found (%): C, 77.81; H, 7.44; N, 3.39. $C_{27}H_{31}NOS$. Calculated (%): C, 77.65; H, 7.48; N, 3.35. 1H NMR, δ : 1.02 and 1.14 (both s, 9 H each, Bu¹); 4.17 (s, 1 H, NH); 4.39 (s, 1 H, H(2)); 6.39 and 6.89 (both d, 1 H each, H(2′), H(6′), J = 3.0 Hz); 6.62 (d,

1 H, H(8), J = 8.0 Hz); 6.72 (dd, 1 H, H(7), J = 7.3 and 8.0 Hz); 6.99 (dd, 1 H, H(6), J = 7.3 and 7.9 Hz); 7.10—7.22 (m, 6 H, H(5), Ph).

3′,5′-Di-*tert*-butyl-4′-oxo-2-(*p*-nitrophenyl)-2,3-dihydrospiro(4*H*-1,4-benzothiazine-3,1′-cyclohexa-2′,5′-diene) (6b). A solution of sulfide 3b (0.005 mol) in benzene (15 mL) was refluxed for 0.5 h. The solvent was evaporated and the residue was purified by column chromatography (Al₂O₃, benzene). The yield was 88%, m.p. 202 °C (from benzene). Found (%): C, 70.26; H, 6.45; N, 6.14. C₂₇H₃₀N₂O₃S. Calculated (%): C, 70.10; H, 6.54; N, 6.06. ¹H NMR, δ: 1.04 and 1.13 (both s, 9 H each, Bu¹); 4.21 (s, 1 H, NH); 4.38 (s, 1 H, H(2)); 6.48 and 6.67 (both d, 1 H each, H(2′), H(6′), J = 3.0 Hz); 6.65 (d, 1 H, H(8), J = 7.5 Hz); 6.79 (dd, 1 H, H(7), J = 7.5 and 8.7 Hz); 7.03 (dd, 1 H, H(6), J = 8.7 and 7.8 Hz); 7.12 (d, 1 H, H(5), J = 7.8 Hz); 7.32 and 8.08 (both d, 2 H each, H(2″), H(6″), H(3″), H(5″), J = 8.8 Hz).

2-p-Bromobenzoyl-3',5'-di-tert-butyl-4'-oxo-2,3-dihydrospiro(4H-1,4-benzothiazine-3,1'-cyclohexa-2',5'-diene) (7). o-Indothiophenol 1 (0.02 mol) was added to a solution of KOH (0.02 mol) in MeOH (35 mL) and the reaction mixture was stirred with gentle heating until thiophenol 1 completely dissolved. Then a solution of p-bromophenacyl bromide 5 (0.02 mol) in MeOH (20 mL) was added and the mixture was refluxed for 3 h. The solvent was evaporated and the residue was purified by column chromatography (Al₂O₃, benzene—light petroleum, 1:1). The yield was 28%, m.p. 200 °C. Found (%): C, 75.38; H, 7.13; N, 3.28. C₂₈H₃₁NO₂S. Calculated (%): C, 75.47; H, 7.01; N, 3.14. ¹H NMR, δ: 1.40 (s, 18 H, Bu^t); 5.37 (s, 1 H, NH); 6.00 (d, 1 H, H(8), J = 7.8 Hz); 6.70—6.88 (m, 3 H, H(5), H(6), H(7)); 6.89 (s, 1 H, H(2)); 7.01 (s, 2 H, H(2'), H(6')); 7.42 and 7.49 (both d, 2 H each, H(3"), H(5"), H(2''), H(6''), J = 8.6 Hz).

X-ray diffraction study of compound 6b. Crystals of $C_{27}H_{30}O_3N_2S$ are tetragonal, at 138(2) K, a=22.015(8) Å, c=10.697(6) Å, V=5184(4) ų, crystal dimensions are $0.5\times0.3\times0.2$ mm, space group $P4_2/n$, Z=8, $d_{calc}=1.185$ g cm⁻³, F(000)=1968, $\mu=0.154$ mm⁻¹.

The intensities of 5953 reflections (of which 5657 reflections were independent, $R_{\rm int}=0.068$) were measured on an automated four-circle Siemens P3/PC diffractometer (graphite monochromator, Mo-K α radiation, $\omega/2\theta$ scanning technique, $2\theta_{\rm max}=54^{\circ}$).

The structure was solved by the direct method using the SHELXTL PLUS program package. The positions of all H atoms were revealed from the difference electron density synthesis. The anisotropic (isotropic for the H atoms) refinement

by the full-matrix least-squares method (418 parameters) based on F^2 using 5600 reflections converged to $R_1 = 0.044$ (for 2515 reflections with $F > 4\sigma(F)$), w $R_2 = 0.116$, S = 0.82. The atomic coordinates for the structure of **6b** and the complete tables of the bond lengths and bond angles were deposited with the Cambridge Structural Database. The principal bond lengths and bond angles are listed in Tables 1 and 2, respectively.

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References

- L. P. Olekhnovich, S. N. Lyubchenko, I. N. Shcherbakov, S. V. Kurbatov, and V. A. Kogan, *Ros. Khim. Zh.*, 1996, 40, 139 [Mendeleev Chem. J., 1996, 40 (Engl. Transl.)].
- L. P. Olekhnovich, S. N. Lyubchenko, V. I. Simakov, A. I. Shif, S. V. Kurbatov, A. V. Lesin, G. S. Borodkin, E. P. Ivakhnenko, and Yu. A. Zhdanov, *Dokl. Akad. Nauk*, 1999, 369, 632 [*Dokl. Chem.*, 1999 (Engl. Transl.)].
- S. N. Lyubchenko, V. V. Litvinov, T. A. Ryskina, E. P. Ivakhnenko, V. A. Kogan, and L. P. Olekhnovich, *Zh. Obshch. Khim.*, 1990, 60, 1618 [*J. Gen. Chem. USSR*, 1990, 60 (Engl. Transl.)].
- L. P. Olekhnovich, V. I. Simakov, N. G. Furmanova, E. P. Ivakhnenko, O. Yu. Rekhlova, and T. A. Ryskina, Yu. A. Zhdanov, *Zh. Obshch. Khim.*, 1992, 62, 885 [*Russ. J. Gen. Chem.*, 1992, 62 (Engl. Transl.)].
- Protective Groups in Organic Chemistry, Ed. J. F. W. McOmie, Plenum Press, London—New York, 1973.
- 6. M. Wilhelm and P. Schmidt, *J. Heterocycl. Chem.*, 1969, **6**, 635.
- 7. Yu. V. Zefirov and P. M. Zorkii, *Usp. Khim.*, 1989, **58**, 713 [*Russ. Chem. Rev.*, 1989, **58** (Engl. Transl.)].
- 8. F. H. Allen, O. Kennard, D. G. Watson, L. Brammer, A. J. Orpen, and R. Taylor, *J. Chem. Soc.*, *Perkin Trans.* 2, 1987. 1.
- G. M. Sheldrick, SHELXTL PLUS. PC Version., A System of Computer Programs for the Determination of Crystal Structure from X-Ray Diffraction Data, Rev. 5.02, Seimens Analytical X-ray Instruments Inc., Germany, 1994.

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