Catalytic Activity Studies of Aminocarbonyl Group Containing Hoveyda– Grubbs-Type Complexes for the Syntheses of Herbarumin I and Stagonolide A

Debendra K. Mohapatra,** R. Somaiah,* M. Mallikarjuna Rao,* Fréderic Caijo,* Marc Mauduit,** J. S. Yadav**

^a Organic Chemistry Division-I, Indian Institute of Chemical Technology (CSIR), Uppal Road, Hyderabad 500 007, India Fax +91(40)27160512; E-mail: mohapatra@iict.res.in

^b Ωcat SYSTEM®, Ecole Nationale Supérieure de Chimie de Rennes, Av. Du Général Leclerc, 35700 Rennes, France *Received 2 March 2010*

Abstract: The catalytic activity of four aminocarbonyl group containing 'boomerang'-type ring-closing metathesis catalysts have been studied for ten-membered lactone and compared well with the Grubbs I and II as well as the Hoveyda–Grubbs catalysts. The activity was found to be superior to the above three ring-closing metathesis catalysts and suggesting novel stereoselective total syntheses of herbarumin I and stagonolide A.

Key words: decanolides, stagonolides, herbarumin I, Wittig reaction, ring-closing metathesis

The generic name decanolides encompasses a relatively small group of naturally occurring ten-membered lactones of polyketides origin isolated in most cases from various species of fungi.¹ Stagonospora cirsii is a pathogen of Cirsium arvense and demonstrated development of a mycoherbicide. A new decanolide, named stagonolide A, was isolated and characterized as (8R,9R)-8-hydroxy-7oxo-9-propyl-5-nonen-9-olide by spectral and analytical methods.² Stagonolide A was shown to be a nonhostspecific but selective phytotoxin. Stagonolide A is structurally related to several known fungal ten-membered lactones, for instance, putaminoxins³ isolated from *Phoma* putaminum, a pathogen of Erigeron annuus (Asteraceae), pinolidoxins⁴ isolated from Ascochyta pinodes, a pathogen of peas (Fabaceae), lethaloxins⁵ and herbarumins isolated from Phoma herbarum, a pathogen of Zea mays (Poaceae).⁶ Among them, herbarumin I⁷ is the most related to stagonolide A (Figure 1). Surprisingly, among investigated natural phytotoxic ten-membered lactones, there are no structures with a ketonic carbonyl group in the lactone ring at C7, similar to stagonolide.⁸ These naturally occurring ten-membered lactones, isolated from fungal metabolites, commonly known as decanolides, have attracted considerable attention from synthetic organic chemists as well as bioorganic chemists because of their interesting structures and various biological activity like plant-growth inhibition, antifeedant, antifungal, and antibacterial activities.

The construction of lactones through the formation of a C-C bond and particularly by ring-closing metathesis reaction (RCM)⁹ is a promising tool for decanolides. In this communication, we found that the catalytic activity of

SYNLETT 2010, No. 8, pp 1223–1226 Advanced online publication: 09.04.2010 DOI: 10.1055/s-0029-1219807; Art ID: G05410ST © Georg Thieme Verlag Stuttgart · New York four well defined, air-stable 'boomerang'-type metathesis catalysts¹⁰ compared well with the Grubbs first- and second-generation and Hoveyda–Grubbs catalysts for decanolide suggesting a novel route towards the syntheses of herbarumin I and stagonolide A. Recently, the first stereoselective total synthesis of stagonolide A has been reported.¹¹



Figure 1 Structures of stagonolide A, herbarumin I, pinolidoxin, and lethaloxin



Scheme 1 Retrosynthetic analysis of herbarumin I and stagonolide A

According to our retrosynthetic analysis (Scheme 1), the construction of the ten-membered lactones 1 and 2 would arise from the formation of C5–C6 olefin linkage from bisalkene 5 which in turn would be prepared via the esterification of alcohol 6 and acid 7. The synthesis of alcohol 6 could be obtained from D-glucose in a concise and high-yielding manner.

The synthesis of the alcohol fragment 6 began with a known intermediate 8 which in turn was obtained from Dglucose.¹² Deprotection of the 1,2-O-isopropylidene group of 8 with 20% AcOH in H₂O and a catalytic amount of H_2SO_4 afforded the lactol **9** in 88% yield, which after was subjected to one-carbon workup Wittig homologation¹³ with in situ generated methylenetriphenylphosphorane to afford the olefinic intermediate 10^{14} in 84% yield. Selective protection of the allylic hydroxyl group as its TBDMS ether and esterification with 5-hexenoic acid under EDCI and DMAP conditions at room temperature gave the intermediate 11^{15} in 85% yield over two steps (Scheme 2). RCM reaction on substrate 11 did not work with the above metathesis catalysts under different reaction conditions. Deprotection of the TBS ether with HF-pyridine complex afforded the desired RCM precursor 5 in 92% yield.¹⁶

The RCM reaction was then investigated using Grubbs first- and second-generation catalysts. The reaction of 5 with Grubbs first-generation catalyst (20 mol%) in

CH₂Cl₂ with high dilution and under reflux conditions gave the desired lactone 12^{17} in 45% yield as the only product (E-isomer) after 48 hours with 42% recovery of the starting material, whereas the second-generation catalyst (10 mol%) afforded the required product in 72% yield in 12 hours. When the Hoveyda-Grubbs catalyst (10 mol%) was used for the same reaction, lactone 12 obtained in 75% yield as the only product. Next, the scope of the metathesis transformation catalyzed by the amino carbonyl group containing Hoveyda-Grubbs-type ('boomerang'-type catalysts) complexes was investigated. The 'boomerang"-type catalysts (Scheme 2) were found to be extremely competent because only 1 mol% catalyst and a reaction time of 1 hours was required to complete the RCM reaction of 5 at room temperature to afford the desired compound 12 in 84–95% yield (Table 1) as the only product.¹⁸

After having the desired compound **12** in good yield and selectivity, the next task was to deprotect the benzyl group under excess TiCl_4^{19} (10 equiv) and CH_2Cl_2 at room temperature to afford the herbarumin I (**2**)²⁰ in 89% yield whose spectral and analytical data were in good agreement with the reported values (Scheme 3).^{6a,7b} Finally, selective oxidation of the allylic hydroxyl group with MnO₂ in CH₂Cl₂ gave the stagonolide A (**1**)^{21,11} in 84% yield. The spectral and analytical data were in an excellent accord with the natural product which reconfirmed the absolute configuration to be (8*R*,9*R*).^{2,11}



Scheme 2 Catalytic activity studies of 'boomerang' catalysts

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Table 1 Results of	RCM Reaction
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Entry	Catalyst	Conditions	Time (h)	Yield (%) ^a
1	G-I	CH ₂ Cl ₂ , reflux	48	45
2	G-II	CH ₂ Cl ₂ , reflux	12	72
3	HG	CH ₂ Cl ₂ , reflux	12	75
4	M71SIMes	CH ₂ Cl ₂ , r.t.	1	84
5	M73SIMes	CH ₂ Cl ₂ , r.t.	1	90
6	M71SIPr	CH ₂ Cl ₂ , r.t.	1	92
7	M73SIPr	CH ₂ Cl ₂ , r.t.	1	95

^a Isolated yield and only *E*-isomer.



Scheme 3 Syntheses of herbarumin I and stagonolide A

In conclusion, we have shown that well-defined, airstable, four aminocarbonyl group containing Hoveyda– Grubbs-type complexes are more effective for ten-membered lactones compared to Grubbs and Hoveyda–Grubbs catalysts and en route achieved a convergent synthesis of pharmacologically active herbarumin I and stagonolide A starting from commercially available D-glucose. All these new air-stable catalysts are under investigation for RCM reaction for different substrates leading to lactones of different ring sizes and substituents at the double bond.

Supporting Information for this article is available online at http://www.thieme-connect.com/ejournals/toc/synlett.

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- (14) Analytical and Spectral Data of 10 $[\alpha]_D^{25}$ +7.6 (*c* 1.1, CHCl₃). IR (neat): 3455, 2924, 2852, 2097, 1461 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 7.43– 7.19 (m, 5 H), 6.05 (m, 1 H), 5.34 (dd, *J* = 17.3, 1.5 Hz, 1 H), 5.18 (dd, *J* = 10.6, 1.5 Hz, 1 H), 4.60 (ABq, *J* = 13.6, 11.3 Hz, 2 H), 4.31 (t, *J* = 5.3 Hz, 1 H), 3.74 (m, 1 H), 3.25 (t, *J* = 6.04 Hz, 1 H), 2.90 (br s, 1 H), 2.59 (br s, 1 H), 1.69–1.24 (m, 4 H), 0.92 (t, *J* = 6.8 Hz, 3 H). ¹³C NMR (75 MHz, CDCl₃): δ = 137.7, 137.9, 137.7, 128.4, 127.8, 128.1, 116.1, 84.2, 73.9, 73.7, 72.6, 35.3, 18.8, 14.0. HRMS: *m/z* calcd for C₁₅H₂₂O₃Na: 273.1466; found: 273.1472.
- (15) Analytical and Spectral Data of 11 $[\alpha]_D^{25}$ +27.0 (*c* 0.7, CHCl₃). IR (neat): $v_{max} = 3398, 2963, 2925, 2855, 1723, 1711, 1639, 1461, 1129 cm⁻¹. ¹H NMR$ $(300 MHz, CDCl₃): <math>\delta = 7.38-7.23$ (m, 5 H), 6.02–5.68 (m, 2 H), 5.29–5.12 (m, 2 H), 5.07–4.95 (m, 2 H), 4.66 (ABq, *J* = 17.2, 11.5 Hz, 2 H), 4.21–4.06 (m, 2 H), 3.50 (t, *J* = 4.7 Hz, 1 H), 2.33–2.24 (m, 2 H), 2.14–2.06 (dd, *J* = 7.4, 6.9 Hz, 2 H), 1.78–1.63 (m, 4 H), 1.40–1.22 (m, 2 H), 0.96–0.84 (m, 12 H), 0.07 (s, 3 H), 0.04 (s, 3 H). ¹³C NMR (75 MHz, CDCl₃): $\delta = 172.7, 138.5, 138.1, 137.6, 128.2, 127.8, 127.5, 116.5, 115.3, 83.9, 74.6, 74.0, 73.7, 34.9, 33.7, 33.1, 31.5, 25.8, 24.1, 18.7, 14.0, –4.2, –5.0. HRMS:$ *m/z*calcd for C₂₇H₄₄O₄NaSi: 483.2906; found: 483.2914.
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(300 MHz, CDCl<sub>3</sub>): \delta = 7.40–7.30 (m, 5 H), 5.68–5.57 (m, 1 H), 5.50 (dd, J = 15.7, 1.1 Hz, 1 H), 5.21 (dt, J = 9.5, 1.1 Hz, 1 H), 4.73–4.53 (m, 3 H), 3.37 (dd, J = 9.6, 1.9 Hz, 1 H), 2.34–2.27 (m, 2 H), 2.03–1.78 (m, 4 H), 1.49 (m, 1 H), 1.36–1.22 (m, 4 H), 0.89 (t, J = 7.4 Hz, 3 H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): \delta = 175.5, 137.3, 133.1, 130.3, 128.5, 128.0, 124.6, 80.9, 72.3, 68.8, 68.8, 34.5, 33.6, 33.5, 24.5, 17.8, 13.9. HRMS: m/z calcd for C<sub>19</sub>H<sub>26</sub>O<sub>4</sub>Na: 341.1728; found: 341.1722.
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- (18) General Procedure for the Metathesis Reaction To a stirred solution of diene (1.0 mmol) in freshly prepared dry CH_2Cl_2 (200 mL) at r.t. was added the metathesis catalyst (0.01 mmol) under argon. The progress of the reaction was monitored by TLC. After completion of the reaction, solvent was removed under reduced pressure. The crude product was purified by flash column chromatography (EtOAc–hexane = 1:6) to afford the *E*-isomer as the sole product.
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(20) Analytical and Spectral Data of 2

 $[a]_D^{25}$ +10.6 (*c* 1.0, EtOH); lit.:^{7b} $[a]_D^{20}$ +10.8 (*c* 0.51, EtOH). ¹H NMR (300 MHz, CDCl₃): δ = 5.61 (d, *J* = 15.9 Hz, 1 H), 5.49 (m, 1 H), 4.95 (td, *J* = 9.4, 2.2 Hz, 1 H), 4.42 (br s, 1 H), 3.51 (d, *J* = 9.8 Hz, 1 H), 2.76 (br s, 1 H), 2.45–2.25 (m, 2 H), 2.06–1.81 (m, 4 H), 1.56 (m, 1 H), 1.43–1.18 (m, 3 H), 0.91 (t, *J* = 7.3 Hz, 3 H). ¹³C NMR (75 MHz, CDCl₃): δ = 176.3, 130.7, 124.6, 73.6, 73.3, 70.1, 34.4, 33.6, 33.3, 24.6, 17.9, 13.8. IR (neat): v_{max} = 3455, 3038, 2955, 2920, 2867, 1711, 1639, 1219 cm⁻¹. HRMS: *m/z* calcd for C₁₂H₂₀O₄Na: 251.1259; found: 251.1264.

(21) Analytical and Spectral Data of 1

Mp 71–72 °C; $[\alpha]_D^{25}$ –59.2 (*c* 0.5, EtOH). ¹H NMR (300 MHz, CDCl₃): $\delta = 6.42$ (d, J = 16.0 Hz, 1 H), 6.32 (m, 1 H), 4.65 (dt, J = 9.5, 2.4 Hz, 1 H), 4.05 (dd, J = 9.5, 6.2 Hz, 1 H), 3.57 (d, J = 6.2 Hz, 1 H), 2.57–2.41 (m, 2 H), 2.20–1.90 (m, 6 H), 1.43–1.18 (m, 3 H), 0.93 (t, J = 7.2 Hz, 3 H). ¹³C NMR (75 MHz, CDCl₃): $\delta = 199.5$, 174.2, 143.1, 131.9, 76.5, 74.4, 34.2, 34.0, 33.5, 24.9, 18.0, 13.7. IR (neat): $v_{max} = 3468$, 2925, 1740, 1700, 1638, 1219 cm⁻¹. HRMS: *m/z* calcd for C₁₂H₁₉O₄: 227.1278; found: 227.1274.

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