# Novel Regio- and Stereoselective Synthesis of Functionalized 3-Spiropyrrolidines and 3-Spiropyrrolizidines Using the Baylis–Hillman Adducts Derived from Nitroolefins

Manickam Bakthadoss,\* Nagappan Sivakumar

Department of Organic Chemistry, University of Madras, Guindy Campus, Chennai 600025, Tamilnadu, India Fax +91(44)22352494; E-mail: bhakthadoss@yahoo.com *Received 2 October 2008* 

**Abstract:** A facile regio- and stereoselective synthesis of functionalized 3-spiropyrrolidines and 3-spiropyrrolizidines using the Baylis–Hillman adducts derived from nitroolefins via intermolecular [3+2] cycloaddition reaction is reported.

**Key words:** Baylis–Hillman reaction, intermolecular [3+2] cycloaddition, azomethine ylides, 1,3-dipolar cycloaddition, spiro compounds

The Baylis–Hillman reaction is an important carbon– carbon bond-forming reaction which is atom economical and simple providing densely functionalized molecules.<sup>1–5</sup> Due to the multifunctionality of Baylis–Hillman adducts they represent valuable starting materials for a variety of novel classes of organic compounds.

The spiropyrrolidine framework is an integral part of many natural products such as horsfiline (1),<sup>6</sup> elacomine (2),<sup>7</sup> alstonisine (3),<sup>8</sup> and spirotryprostatin A (4)<sup>9</sup> (Figure 1). In addition spiropyrrolidines have been shown to posses anticancer, antimicrobial, antibiotic, and antineoplastic properties.<sup>10–12</sup>





The synthesis of nitrogen- and oxygen-containing heterocycles continues to be an important and challenging area in the field of organic chemistry.<sup>13–16</sup> The 1,3-dipolar cyclo-

SYNLETT 2009, No. 6, pp 1014–1018 Advanced online publication: 16.03.2009 DOI: 10.1055/s-0028-1088206; Art ID: D34808ST © Georg Thieme Verlag Stuttgart · New York addition reaction is one of the most important and useful methods for the preparation of five-membered heterocycles. The reaction of azomethine ylides with alkenes provides the pyrrolidine moiety which is present in numerous natural products and biologically active molecules.<sup>17,18a</sup> Utilizing azomethine ylide based [3+2] cycloaddition reaction, a variety of spiropyrrolidines have been reported in the literature.<sup>18a-18e</sup>

In the Baylis–Hillman reaction electron-deficient olefins are utilized although nitroolefins have only been used as the activated olefinic component in a couple of reports.<sup>19</sup> Recently a novel type of Baylis–Hillman adduct **5** derived from  $\beta$ -substituted nitroolefins has been reported.<sup>19b</sup> but so far no further transformations have been studied.

We envisaged that this novel class of  $\beta$ -substituted Baylis– Hillman adduct **5** derived from  $\beta$ -aryl nitroolefins would be a useful starting material for the construction of spiropyrrolidine compounds via azomethine ylide based [3+2] cycloaddition. It is well documented in the literature that Baylis–Hillman adducts have been utilized for the synthesis of various heterocycles<sup>1,3,20</sup> but the synthesis of heterocyclic spiro compounds using Baylis–Hillman adducts derived from nitroolefins was not known to date.

In continuation of our interest in the field of Baylis– Hillman chemistry,<sup>21–24</sup> we herein report a simple and convenient route for the regio- and stereoselective synthesis of 3-spiropyrrolidine and 3-spiropyrrolizidine frameworks using the Baylis–Hillman adducts derived from nitroolefins with sarcosine/proline-based dipoles generated via in situ imine formation, decarboxylation, and intermolecular [3+2] cycloaddition as shown below in the retrosynthetic strategy (Scheme 1).

We first selected (*E*)-2-nitro-3-phenylprop-2-en-1-ol (**5a**), a Baylis–Hillman adduct derived from nitrostyrene, as a starting material for [3+2] cycloaddition reaction with dipoles generated from sarcosine with isatin. Best results were obtained when **5a** was treated with isatin and sarcosine without any catalyst in acetonitrile as solvent for 5 hours at reflux temperature, which successfully provided the desired 3-spiropyrrolidine compound **6a** in very good yield (82%) after usual workup followed by column chromatography. The compound **6a** was characterized by IR, <sup>1</sup>H NMR and <sup>13</sup>C NMR spectroscopy, mass spectrometry, and elemental analysis (Scheme 2).



 
 Table 1
 Synthesis of 3-Spiropyrrolidine Compounds from the
 Baylis-Hillman Adducts<sup>a</sup>

Allyl alcohol	R	Product <sup>b</sup>	Yield (%) <sup>c</sup>
5a	Н	<b>6a</b> <sup>d</sup>	82
5b	2-Me	6b	75
5c	4-Me	6c	76
5d	4-Et	6d	77
5e	4- <i>i</i> -Pr	6e	60
5f	4-OMe	6f	61
5g	3,4-di-OMe	6g	62
5h	3,4-OCH <sub>2</sub> O	6h	60
5i	4-F	6i	64
5j	2-CI	6j	80
5k	3-CI	6k	64
51	4-CI	61	70

<sup>a</sup> All reactions were carried out using Baylis-Hillman alcohol 5a-l (2 mmol) with sarcosine (2 mmol) and isatin (2 mmol) in MeCN (8 mL) under reflux conditions for 5 h.

<sup>b</sup> All products gave satisfactory IR, <sup>1</sup>H NMR (300 MHz), <sup>13</sup>C NMR (75 MHz), mass spectrometry, and elemental analyses.

° Yields of the pure products 6a-l obtained after column chromatography (SiO<sub>2</sub>, 20% EtOAc in hexanes).

<sup>d</sup> Structure was further confirmed by single-crystal X-ray analysis.

The X-ray crystal-structure analysis of compound 6a showed that the relative stereochemistry (Figure 2) of the phenyl group and CH<sub>2</sub>OH group in the vicinal positions is cis. Similarly the NO<sub>2</sub> group and benzylic proton are also in a cis orientation.

To probe further the generality of the reaction, we subjected adducts 5a-d,j,l with the dipole generated from isatin and proline. Best results were obtained when 5a-d,j,l were treated with isatin and proline without catalyst in acetonitrile as solvent for two hours at reflux temperature,



Figure 2 ORTEP diagram of 6a

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## Scheme 2

Encouraged by this result, we prepared a variety of (E)-2nitro-3-arylprop-2-en-1-ols 5b-l as starting materials for the preparation of 3-spiropyrrolidine compounds. As before treatment of these with isatin and sarcosine in acetonitrile for five hours at reflux temperature successfully led to the desired 3-spiropyrrolidine compounds 6b-l in 60-80% yields (Scheme 2). The results are summarized in Table 1.

Worthy of note was the high regio- and stereoselectivity obtained in the each case, which was clearly evidenced by <sup>1</sup>H NMR data and X-ray crystal analysis.<sup>25</sup> The <sup>1</sup>H NMR spectrum of compound **6a** showed a singlet for NCH<sub>3</sub> proton at  $\delta = 2.18$  ppm and triplet for benzylic proton at  $\delta =$ 4.97 ppm. The NCH<sub>2</sub> protons of pyrrolidine ring appeared as two triplets at  $\delta$  = 3.53 and 4.04 ppm. The O-attached CH<sub>2</sub> protons appeared as multiplet in the region of  $\delta$  = 3.75–3.94. The NH proton of oxindole ring appeared at  $\delta = 7.74$  ppm. The aromatic protons appeared in the region of  $\delta = 6.82 - 7.57$  ppm.

Had the other regioisomer been formed, the benzylic proton should have been appeared as singlet which we did not observe in any of the cases 6a-I. Comparison of the <sup>1</sup>H NMR spectra of the crude and recrystallized products showed them to be identical which confirming the stereoselectivity of the reaction. Furthermore the structure 6a was confirmed by X-ray crystallographic analysis.

Table 2 Synthesis of 3-Spiropyrrolizidine Compounds from the Baylis–Hillman Adducts<sup>a</sup>

Allyl alcohol	R	Product <sup>b</sup>	Yield (%) <sup>c</sup>	Product <sup>b</sup>	Yield (%) <sup>c</sup>	Yield of <b>7</b> + <b>8</b> (%)
5a	Н	7a	52	8a	33	85
5b	2-Me	7b	65	8b	25	90
5c	4-Me	7c	59	8c	35	94
5d	4-Et	$\mathbf{7d}^{d}$	52	8d	37	89
5j	2-CI	7j	60	8j	20	80
51	4-CI	71	62	81	31	93

<sup>a</sup> All reactions were carried out using Baylis–Hillman alcohol **5a–d.j.l** (2 mmol) with proline (2 mmol) and isatin (2 mmol) in MeCN (8 mL) under reflux conditions for 2 h.

<sup>b</sup> All products gave satisfactory IR, <sup>1</sup>H NMR (300 MHz), <sup>13</sup>C NMR (75 MHz), mass spectrometry, and elemental analyses.

<sup>c</sup> Yields of the pure products **7a–d,j,l** and **8a–d,j,l** obtained after column chromatography (SiO<sub>2</sub>, **7a–d,j,l**: 30% EtOAc in hexanes; **8a–d,j,l**: 40% EtOAc in hexanes).

<sup>d</sup> Structure was further confirmed by single-crystal X-ray analysis.

successfully provided the desired 3-spiropyrrolizidine compounds **7a–d,j,l** in good yields (52–65%) along with other regioisomers **8a–d,j,l** in 20–37% yields (Scheme 3). The compounds **7a–d,j,l** and **8a–d,j,l** were characterized by IR, <sup>1</sup>H NMR and <sup>13</sup>C NMR spectroscopy, mass spectrometry, and elemental analyses (Scheme 3). The results are summarized in Table 2.

The NMR data and X-ray crystal-structure analysis<sup>25</sup> showed that the proline- and isatin-based [3+2] cycloaddition reaction with the BH adducts yielded regioisomeric compounds that were successfully separated by column chromatography.



#### Scheme 3

The <sup>1</sup>H NMR spectrum of compound **7a** showed a doublet for H<sub>a</sub> proton at  $\delta = 4.29$  ppm and the H<sub>b</sub> proton appeared as multiplet in the region of  $\delta = 4.47-4.56$  ppm. The pyrrolidine ring protons appeared as multiplets in the region of  $\delta = 1.50-2.76$  ppm. The hydroxymethylene protons appeared as a multiplet in the region of  $\delta = 4.39-4.44$  ppm and a doublet of doublet at  $\delta = 4.98$  ppm.

The <sup>1</sup>H NMR spectrum of compound **8a** showed a singlet for the H<sub>a</sub> proton (benzylic proton) at  $\delta = 5.29$  ppm and the



Figure 3 ORTEP diagram of 7d

 $H_b$  proton appeared as a triplet at  $\delta = 4.77$  ppm. The pyrrolidine ring protons appeared as multiplets in the region of  $\delta = 1.37-3.30$  ppm. Hydroxymethylene protons appeared as two doublet of doublets at  $\delta = 4.15$  and 4.49 ppm.

The X-ray crystal structure of the compound **7d** showed that the relative stereochemistry (Figure 3) of the aryl and  $CH_2OH$  groups to be *cis*. Similarly the  $NO_2$  group and benzylic proton are also in a *cis* orientation. The benzylic proton ( $H_a$ ) and pyrrolidizine proton ( $H_b$ ) at the ring junction are in a *trans* orientation as shown in Figure 4.

In conclusion, we have successfully developed a simple and novel protocol for the facile synthesis of functionalized 3-spiropyrrolidine with high regio- and stereoselectivity. We have also demonstrated that this method is useful for making novel 3-spiropyrrolizidine frameworks using Baylis–Hillman adducts derived from nitroolefins.



Figure 4

### **Typical Experimental Procedure of 6a**

A mixture of (*E*)-2-nitro-3-phenylprop-2-en-1-ol (**5a**, 2 mmol, 0.36 g), isatin (2 mmol, 0.29 g), and sarcosine (2 mmol, 0.18 g) in MeCN (8 mL) was refluxed for 5 h. After the completion of the reaction as indicated by TLC, the reaction mixture was concentrated, and the resulting crude mass was diluted with H<sub>2</sub>O (10 mL) and extracted with EtOAc ( $3 \times 10$  mL). The combined organic layers were washed with brine ( $2 \times 10$  mL) and dried over anhyd Na<sub>2</sub>SO<sub>4</sub>. The organic layer was concentrated and the residue purified by column chromatography on SiO<sub>2</sub> (Acme 100–200 mesh), using EtOAc–hexanes (2:8) to afford **6a** as a colorless solid in 82% (0.58g) yield.

## Typical Experimental Procedure of 7a and 8a

A mixture of (*E*)-2-nitro-3-phenylprop-2-en-1-ol (**5a**, 2 mmol, 0.36 g), isatin (2 mmol, 0.29 g), and L-proline (2 mmol, 0.23 g) in MeCN (8 mL) was refluxed for 2 h. After the completion of the reaction as indicated by TLC, the reaction mixture was concentrated, and the resulting crude mass was diluted with H<sub>2</sub>O (20 mL) and extracted with EtOAc ( $3 \times 10$  mL). The combined organic layer obtained was washed with brine ( $3 \times 10$  mL) and dried over anhyd Na<sub>2</sub>SO<sub>4</sub>. The organic layers were concentrated and purified by column chromatography on SiO<sub>2</sub> (Acme 100–200 mesh), using EtOAc–hexanes (3:7) to provide **7a** as a colorless solid in 52% (0.11 g) yield and **8a** as a colorless solid (Acme 100–200 mesh), using EtOAc–hexanes (4:6) in 33% (0.07 g) yield.

#### Spectroscopic Data for Selected Compounds Compound 6a

Mp 160–162 °C. IR (KBr): 3484, 3227, 1717, 1544, 1337 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 2.18 (s, 3 H), 2.89 (dd, 1 H, *J* = 2.7, 9.0 Hz), 3.53 (t, 1 H, *J* = 9.0 Hz), 3.78 (dd, 1 H, *J* = 3.0, 10.5 Hz), 3.90–4.07 (m, 2 H), 4.97 (t, 1 H, *J* = 9.0 Hz), 6.82–7.57 (m, 9 H), 7.74 (s, 1 H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  = 34.77, 49.40, 56.78, 64.17, 77.23, 105.05, 110.15, 123.38, 124.46, 125.20, 128.18, 128.78, 129.99, 130.48, 134.75, 141.55, 176.06. MS: *m/z* = 354 [M<sup>+</sup> + 1]. Anal. Calcd for C<sub>19</sub>H<sub>19</sub>N<sub>3</sub>O<sub>4</sub>: C, 64.58; H, 5.42; N, 11.89. Found: C, 64.56; H, 5.39; N, 11.94.

## **Compound 7a**

Mp 148–150 °C. IR (KBr): 3463, 3197, 1709, 1540, 1336 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 1.50-1.62$  (m, 1 H), 1.87–2.10 (m, 3 H), 2.48–2.56 (m, 1 H), 2.69–2.76 (m, 1 H), 3.87 (t, 1 H, *J* = 12.0 Hz), 4.27 (d, 1 H, *J* = 10.2 Hz), 4.39–4.56 (m, 2 H), 4.95 (dd, 1 H, *J* = 3.6, 8.1 Hz), 6.87–7.81 (m, 10 H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 27.88$ , 32.74, 47.89, 56.59, 64.51, 69.88, 78.03, 105.54, 111.22, 122.96, 124.98, 125.52, 127.99, 128.56, 130.70, 130.88, 134.19, 140.86, 179.78. MS: *m/z* = 380 [M<sup>+</sup> + 1]. Anal. Calcd for C<sub>21</sub>H<sub>21</sub>N<sub>3</sub>O<sub>4</sub>: C, 66.48; H, 5.58; N, 11.08. Found: C, 66.41; H, 5.46; N, 11.14.

## **Compound 8a**

Mp 170–172 °C. IR (KBr): 3467, 3185, 1709, 1535, 1345 cm<sup>-1. 1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.38–1.50 (m, 1 H), 1.80–2.05 (m, 2 H), 2.71–2.31 (m, 2 H), 2.67 (td, 1 H, *J* = 2.4, 5.4 Hz), 3.26 (q, 1 H, *J* = 8.4 Hz), 4.15 (dd, 1 H, *J* = 4.5, 8.7 Hz), 4.49 (t, 1 H, *J* = 10.5

Hz), 4.78 (t, 1 H, J = 5.4 Hz), 5.29 (s, 1 H), 6.72–7.52 (m, 10 H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 25.04$ , 27.69, 48.50, 59.23, 65.28, 69.60, 73.67, 104.73, 110.22, 122.73, 125.31, 126.25, 128.33, 128.96, 129.43, 130.01, 131.79, 141.40, 178.87. MS: m/z = 380[M<sup>+</sup> + 1]. Anal. Calcd for C<sub>21</sub>H<sub>21</sub>N<sub>3</sub>O<sub>4</sub>: C, 66.48; H, 5.58; N, 11.08. Found: C, 66.42; H, 5.49; N, 11.16.

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