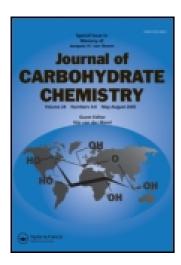
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A. S. M. Sofian ^a & C. Kuan Lee ^b

^a Department of Chemistry, National University of Singapore, Kent Ridge, 0511, Singapore

^b Department of Chemistry, National University of Singapore, Kent Ridge, 0511, Singapore

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SYNTHESIS AND TASTE PROPERTIES OF 4,1', 6'-TRIHALODEOXY SUCROSE ANALOGUES^{1,2}

A. S. M. Sofian and C. Kuan Lee*

Department of Chemistry, National University of Singapore, Kent Ridge, Singapore 0511

ABSTRACT

Treatment of 3,4-di-O-acetyl-1,6-di-O-trityl- β -D-fructofuranosyl 2,3,6-tri-O-acetyl-4-O-triflyl- α -D-galactopyranoside (3) with a halide source gave, via $S_{\rm N}2$ displacement, the corresponding C-4 halogenated compounds which were subsequently transformed into the monohalogenated 4-deoxy-4-halogenosucrose derivatives, 10–12. Trihalogenated sucrose derivatives, 4-bromo-1',6'-dichloro 17, 1',6'-dibromo 18, and 1',6'-diiodosucrose 19, were synthesized from 5. Exploiting the distinct reactivity at C-1' and C-6' trifluoromethanesulfonates enabled the introduction of different halogens at these positions. This pathway led to the preparation of the 4-bromo-1',6'-dihalodeoxysucrose derivatives, 24–27. Preliminary taste properties of these mono- and tri-halogenated sucrose analogues were also investigated.

INTRODUCTION

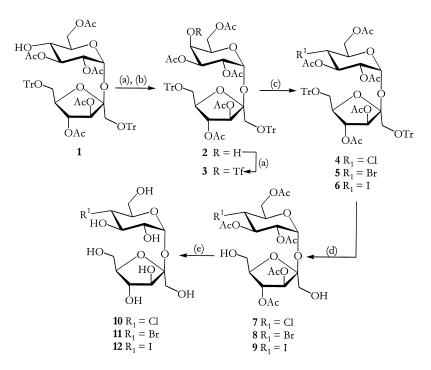
The intense sweetness of halodeoxy analogues of sucrose, many of which are several thousand times sweeter than sucrose, had puzzled researchers since the discovery of the first of these in 1976. Currently, the Shallenberger-Kier AH, B, γ model^{4,5} is the most widely accepted model used to explain the initial mechanism of sweet-taste response. However, the location of the AH,B, γ glucophore for this group of sucrose compounds is still uncertain and intensely debated. ^{6–12}

^{*}Corresponding author.

The high sweetness of these sucrose derivatives is clearly inexplicable in terms of either the glucopyranosyl or fructofuranosyl residues alone since the chlorodeoxy hexopyranoses or fructofuranoses are either very bitter or only moderately sweet. The most likely location of the AH, B entity is located at hydroxyl group of C-3' and the oxygen of the C-2 hydroxyl. 12 The intense sweetness of the halodeoxy sucrose derivatives clearly shows the importance of a halogen substituent(s) in enhancing sweetness, and it has been widely assumed that one or more of these could be acting as the third γ -site in these halodeoxy analogues. In addition, the stereochemical disposition of the halogen substituent(s) is/are critical. 13 1,6-Dichloro-1,6-dideoxy- β -D-fructofuranosyl 4-chloro-4-deoxy- α -Dgalactopyranoside (sucralose)¹⁴ and related 4-chloro-4-deoxy-galactosucrose derivatives have been reported but, apart from 4,6,1',6'-tetrachloro-4,6,1',6'tetradeoxysucrose, ¹⁴ we are not aware of any reports of the synthesis and taste of the 4-chlorodeoxy sucrose derivatives. These derivatives are critical for determining the importance of a halogen substituent at C-4. We now report the synthesis as well as the taste properties of these compounds.

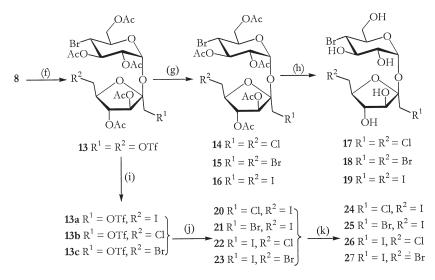
RESULTS AND DISCUSSION

A convenient starting intermediate for the synthesis of the mono- and tri-halogenated derivatives of sucrose is 3,4-di-*O*-acetyl-1,6-di-*O*-trityl-β-D-



Scheme 1. (a) Tf₂O, pyridine/CH₂Cl₂; (b) NaNA₂, DMF, rt, (c) **4:** LiCl, acetone, rt, **5:** LiBr, acetone, rt; **6:** KI, acetone, rt; (d) AcOH-CH₂Cl₂ (1:1), concd HCl, 0 °C; (e) NaOMe, MeOH, rt.





Scheme 2. (f) Tf₂O, pyridine/CH₂Cl₂; (g) **14:** LiCl, acetone, rt, **15:** LiBr, acetone, rt; **16:** KI, acetone, rt; (h) NaOMe, MeOH, rt; (i) **13a:** KI, acetone, 0 °C; **13b:** LiCl, acetone, 0 °C; **13c:** LiBr, acetone, rt; (j) **20:** LiCl, acetone, rt, **21:** LiBr, acetone, rt; **22** and **23:** KI, acetone, rt; (k) NaOMe, MeOH, rt.

fructofuranosyl 2,3,6-tri-O-acetyl- α -D-glucopyranoside (1)¹⁵ (Scheme 1). This was conveniently converted into the *galacto* isomer 2 *via* the C-4 triflate followed by displacement with nitrite ion and *in situ* hydrolysis of the nitrite intermediate. ¹⁶ The *galacto* configuration was evident from the small $J_{3,4}$ coupling (2.7 Hz) in its ¹H NMR spectrum.

The C-4 halodeoxy sucrose derivatives were synthesized by displacement of the triflyl substituent of **3** using lithium chloride (\rightarrow **4**), lithium bromide (\rightarrow **5**) or potassium iodide (\rightarrow **6**). The ¹H NMR spectra of compounds **4–6** all showed H-3 and H-4 as triplets with $J_{2,3} = J_{3,4} = \sim 10$ Hz, characteristic of the *gluco*-configuration. Further evidence was obtained from their ¹³C NMR spectra. The resonances for C-4 in these compounds showed upfield shifts of between 13 to 43 ppm compared to that of **1**.

Detritylation of **4–6** in dichloromethane-acetic acid (1:1) in the presence of concd HCl at 0° C (\rightarrow **7–9**, respectively), followed by conventional deacetylation gave the 4-halodeoxy derivatives **10–12**, respectively, in good yields (Scheme 1).

Compounds **7 –9** served as key intermediates in the synthesis of a variety of trihalogenated sucrose analogues. Thus, triflation of **8** gave the corresponding 1',6'-disulfonate, which was easily converted to the 4-bromo-1',6'-dichloro (**14**), 4,1',6'-tribromo (**15**) and 4-bromo-1',6'-diiodo (**16**) derivatives, respectively, on treatment with lithium chloride, lithium bromide and potassium iodide. Zemplén deacetylation of these compounds then gave the free trihalogenated derivatives **17 –19** (Scheme 2).

The triflate groups at C-1′ and C-6′ have been reported ¹⁷ to show considerable difference in reactivity. This enables the introduction of different halogen substituents at these two positions. The reaction of **13** with potassium iodide at 0°C gave the 4-bromo-6′-iodo-4,6′-dideoxy-1′-*O*-triflate (**13a**), which was then converted to the corresponding chloro derivative **20** by reaction with lithium chloride in acetone at room temperature. Compounds **21–23** were synthesized in a similar manner. These were then deacetylated to give **24–27** (Scheme 2).

Preliminary taste properties of the above C-4 monohalodeoxy derivatives showed all to possess little or no sweetness. The presence of only one halogen substituent in the sucrose is clearly not sufficient to produce an optimum hydrophilic-lipophilic balance for interaction with the receptor. The trihalodeoxy derivatives, on the other hand, were very much sweeter but all were lower than that of sucralose. Thus, if the halo substituent at C-4 functions as a γ -site, hen it appears that the stereochemistry of the halogen substituent at this carbon is extremely important for optimum binding to the receptor. The results of these taste tests will be reported in detail elsewhere.

EXPERIMENTAL

General Methods. Melting points were measured with a Thermo Galen Hot Stage Microscope. Optical rotations were taken with a Perkin Elmer 241 polarimeter at 26°C. NMR spectra were recorded at 298 K in CDCl₃ (unless otherwise specified) on a Bruker DPX 300 spectrometer (300.13 MHz for ¹H and 75.47 MHz for ¹³C). Mass spectra were recorded on either on a Finnigan MAT95XL-T instrument using electronic ionisation (EI) (70 eV) mode with an accelerating voltage of 5 KV and a resolution of 1000 (10% valley definition), or on a Finnigan TSQ 7000 (ion trap) spectrometer using electron spray ionisation (ESI) with a spray voltage of 4.5 KV. The elemental composition of ions was determined with a resolution of 7,000 (10% valley definition). Microanalyses were carried out using a Perkin Elmer 2400 Elemental Analyser. Flash chromatography was performed on Silica Gel 60 (0.63–0.200 nm, Merck) at 5–10 p.s.i. Thin-layer chromatography was run on glass plates precoated with silica gel 60F₂₅₄ (Merck, Darmstadt, Germany); detection was effected by observation under short wavelength UV light (254 nm), then spraying with 10% sulphuric acid in ethanol and charring them on a hot plate.

3,4-Di-O-acetyl-1,6-di-O-trityl- β -D-fructofuranosyl 2,3,6-tri-O-acetyl- α -D-galactopyranoside (2). To a solution of 1 (15.3 g, 14.8 mmol) in dry CH₂Cl₂-pyridine (15:1, 64 mL) at -78° C, was added trifluoromethanesulfonic anhydride (4.9 mL, 29.9 mmol). The mixture was stirred for 15 min at -78° C, then for about 2 h at 0° C. The mixture was diluted with dichloromethane and the organic solution was washed successively with aq KHSO₄ (10%), satd NaHCO₃ and water, dried (Na₂SO₄) and concentrated to give a syrupy product. This was dissolved in DMF (60 mL) and sodium nitrite (5.6 g) was added. The solution was then stirred at room temperature overnight, concentrated, and the residue diluted with

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dichloromethane. The filtered solution was again concentrated. Flash chromatography (ethyl acetate/hexane, 1:2) gave **2**, (10.6 g, 69%), mp 103–105°C; [α]_D +70.5° (c 1.21, CHCl₃); ¹H NMR (CDCl₃) δ 1.83–2.06 (s, 15H, 5 × CH₃), 3.20 (s, 2H, H-1'a, b), 3.33–3.52 (m, 2H, H-6'a, b), 4.02 (d, 1H, $J_{3,4}$ 2.7 Hz, H-4), 4.10–4.36 (m, 4H, H-5, 5', 6'a, b), 5.05 (dd, 1H, $J_{2,3}$ 10.8 $J_{3,4}$ 2.7 Hz, H-3), 5.14 (dd, 1H, $J_{1,2}$ 3.7 $J_{2,3}$ 10.8 Hz, H-2), 5.37 (t, 1H, $J_{3',4'}$ = $J_{4',5'}$ 7.3 Hz, H-4'), 5.56 (d, 1H, $J_{1,2}$ 3.7 Hz, H-1), 5.84 (d, 1H, $J_{3',4'}$ 7.3 Hz, H-3') and 7.21–7.51 (m, 30H, Ar-H). ¹³C NMR δ 170.9, 170.0, 169.9, 169.8, 169.6 (COCH₃), 143.6, 143.4 (CPh₃), 129.0–127.0 (Ar-C), 104.1 (C-2'), 89.5 (C-1), 78.8 (C-5'), 75.7, 74.7 (C-3', 4'), 69.9, 67.8, 67.2, 66.9 (C-2, 3, 4, 5), 63.8, 62.1 (C-1',6,6') and 20.8, 20.7, 20.5, 20.4 (COCH₃). EI-MS m/z (%): 1036 [M⁺] (<0.5), 731 [M - C₁₂H₁₇O₉]⁺ (<0.5), 289 [M - C₄₈H₄₃O₈]⁺ (6), 244 (32), 243 (100), 242 (26), 228 (9), 215 (6), 43 (100).

Anal. Calcd for C₆₀H₆₀O₁₆: C, 69.49; H, 5.83. Found: C, 69.87; H, 5.46.

3,4-Di-O-acetyl-1,6-di-O-trityl- β -D-fructofuranosyl 2,3,6-tri-O-acetyl-4-chloro-4-deoxy- α -D-glucopyranoside (4). A solution of 2 (1.66 g, 1.60 mmol) in CH₂Cl₂- pyridine (15:1, 32 mL) at -78° C was triflated as described above to give crude 3.

Without purification and characterization, the crude triflate 3 (1.70 g, 1.45 mmol) in dry acetone (20 mL), was stirred with lithium chloride (0.2 g) at room temperature under argon for 4 h. When TLC (ether/hexane, 3:1) showed that all starting material had reacted, the solution was concentrated and dichloromethane added and filtered. The organic layer was washed with satd NaHCO₃, brine, dried (Na₂SO₄) and concentrated. Flash chromatography (ether/hexane, 1:1) gave 4 as colourless crystals (1.25 g, 81%), mp 97–99°C; $[\alpha]_D$ +44.0° (c 1.17, CHCl₃); ¹H NMR δ 1.83–2.06 (s, 15H, 5 × CH₃), 3.20 (s, 2H, H-1'a, b), 3.31–3.45 (2 × dd, 2H, $J_{5',6'a}$ 5.6 $J_{5',6'b}$ 4.8 $J_{6'a,6'b}$ 10.0 Hz, H-6'a, b), 3.81 (t, 1H, $J_{3,4} = J_{4,5}$ 10.0 Hz, H-4), 4.15 (m, 1H, H-5'), 4.23 (m, 3H, H-5, 6a,b), 4.66 (dd, 1H, $J_{1,2}$ 4.0 $J_{2,3}$ 10.0 Hz, H-2), 5.33 (t, 1H, $J_{2,3} = J_{3,4}$ 10.0 Hz, H-3), 5.42 (d, 1H, $J_{1,2}$ 4.0 Hz, H-1), 5.47 (d, 1H, $J_{3',4'} = J_{4',5'}$ 6.0 Hz, H-4'), 5.80 (d, 1H, $J_{3',4'}$ 6.0 Hz, H-3') and 7.20–7.47 (m, 30H, Ar-H). ¹³C NMR δ 170.3, 169.8, 169.7, 169.5, 169.4 (COCH₃), 143.6, 143.3 (CPh₃), 128.7–127.1 (Ar-C), 104.7 (C-2'), 89.5 (C-1), 79.7 (C-5'), 76.0, 75.2 (C-3', 4'), 71.4, 70.5, 70.2 (C-2, 3, 5), 63.7, 63.6, 62.4 (C-1', 6, 6'), 55.1 (C-4) and 21.0, 20.8, 20.6, 20.5, 20.3 (COCH₃). EI-MS m/z (%): 1054 [M⁺] (<0.5), 731 [M $-C_{12}H_{16}ClO_{8}$]⁺ (1.5), 307, 309 [M - $C_{48}H_{43}O_{8}$]⁺ (3:1, 47), 244 (56), 243 (52), 242 (70), 241 (72), 215 (43), 43 (100).

Anal. Calcd for $C_{60}H_{59}ClO_{15}$: C, 68.22; H, 5.63; Cl, 3.31. Found: C, 67.98; H, 5.92; Cl, 3.67.

3,4-Di-*O*-acetyl-1,6-di-*O*-trityl- β -D-fructofuranosyl 2,3,6-tri-*O*-acetyl-4-bromo-4-deoxy- α -D-glucopyranoside (5). A solution of crude 3 (11.3 g, 9.67 mmol) in dry acetone (60 mL) was treated with lithium bromide (2.5 g) as for 4 to give after flash chromatography (ether/hexane, 1:1), 5 as colourless crystals (9.0 g, 85%), mp 97–99°C; [α]_D +40.9° (c 1.12, CHCl₃); ¹H NMR δ 1.79–2.05 (s, 15H, 5 × CH₃), 3.21–3.43 (m, 4H, H-1'a, b and H-6'a, b), 3.85 (t, 1H, $J_{3,4} = J_{4,5}$ 10.8

Hz, H-4), 4.08–4.35 (m, 4H, H-5, 5′, 6a, b), 4.63 (dd, 1H, $J_{1,2}$ 4.0 $J_{2,3}$ 10.0 Hz, H-2), 5.38–5.49 (m, 3H, H-1, 3, 4′), 5.79 (d, 1H, $J_{3',4'}$ 6.0 Hz, H-3′) and 7.19–7.46 (m, 30H, Ar-H). ¹³C NMR δ 170.2, 169.8, 169.7, 169.5, 169.3 (*C*OCH₃), 143.6, 143.3 (*C*Ph₃), 129.5–126.5 (Ar-C), 104.6 (C-2′), 89.5 (C-1), 79.7 (C-5′), 76.0, 75.2 (C-3′, 4′), 71.3, 70.4, 70.2 (C-2, 3, 5), 63.7, 63.5, 63.1 (C-1′, 6, 6′), 46.1 (C-4) and 20.8, 20.6, 20.3 (COCH₃). EI-MS m/z (%): 351, 353 [M - C₄₈H₄₃O₈]⁺ (1:1, 8), 291, 293 (1:1, 12), 244 (83), 243 (88), 242 (79), 241 (80), 229 (26), 215 (19), 183 (54), 169 (21), 165, 167 (1:1, 79), 43 (100).

Anal. Calcd for $C_{60}H_{59}BrO_{15}$: C, 65.50; H, 5.41; Br, 7.62. Found: C, 67.89; H, 5.41; Br, 6.99.

3,4-Di-*O*-acetyl-1,6-di-*O*-trityl-β-D-fructofuranosyl 2,3,6-tri-*O*-acetyl-4-deoxy-4-iodo-α-D-glucopyranoside (6). Treatment of crude **3** (1.51 g, 1.29 mmol) with potassium iodide (0.65 g) as described for **4** gave after flash chromatography (ether/hexane, 1:1), crystalline **6** (1.20g, 81%), mp 95–97°C; [α]_D +38.6° (*c* 1.01, CHCl₃); ¹H NMR δ 1.79–2.07 (s, 15H, 5 × CH₃), 3.21–3.44 (m, 4H, H-1'a, b and H-6'a, b), 3.92 (t, 1H, $J_{3,4} = J_{4,5}$ 11.0 Hz, H-4), 4.09–4.42 (m, 4H, H-5, 5', 6a, b), 4.63 (dd, 1H, $J_{1,2}$ 4.0 $J_{2,3}$ 10.0 Hz, H-2), 5.41–5.50 (m, 3H, H-1, 3, 4'), 5.79 (d, 1H, $J_{3',4'}$ 6.0 Hz, H-3') and 7.19–7.46 (m, 30H, Ar-H). ¹³C NMR δ 170.1, 169.7, 169.6, 169.4, 169.0 (COCH₃), 143.6, 143.3 (*C*Ph₃), 129.0–126.9 (Ar-C), 104.6 (C-2'), 89.6 (C-1), 79.6 (C-5'), 76.0, 75.2 (C-3', 4'), 72.2, 71.0, 70.5 (C-2, 3, 5), 64.6, 63.8, 63.4 (C-1', 6, 6'), 24.8 (C-4) and 20.7, 20.6, 20.5, 20.3 (COCH₃). EI-MS m/z (%): 399 [M - C₄₈H₄₃O₈]⁺ (9), 279 (13), 244 (92), 243 (94), 242 (62), 241 (60), 165 (90), 43 (100); exact mass calcd for [C₁₂H₁₆IO₇]⁺ 398.9941, found 398.99285.

Anal. Calcd for $C_{60}H_{59}IO_{15}$: C, 62.83; H, 5.18; I, 11.06. Found: C, 62.68; H, 5.61; I, 10.70.

3,4-Di-*O*-acetyl-β-D-fructofuranosyl **2,3,6-tri-***O*-acetyl-**4-chloro-4-de-oxy-**α-**D-glucopyranoside** (**7**). To a solution of **4** (1.20 g, 1.14 mmol) in ice-cold CH₂Cl₂-AcOH (1:1, 20 mL) was added dropwise, concentrated hydrochloric acid (0.15 mL). After all starting material had reacted (~2 h), the solution was neutralized with NaHCO₃, filtered and concentrated under reduced pressure. The crude product was subject to column chromatography (ethyl acetate/hexane, 1:1) to afford **7** as colourless crystals (0.52 g, 80%), mp 99–101°C; [α]_D +25.0° (c 1.12, CHCl₃); ¹H NMR δ 2.07–2.20 (s, 15H, 5 × CH₃), 3.55–3.84 (m, 4H, H-1'a, b and H-6'a, b), 3.91 (t, 1H, $J_{3,4} = J_{4,5}$ 10.4 Hz, H-4), 4.06 (m, 1H, H-5'), 4.33–4.55 (m, 3H, H-5, 6a, b), 4.84 (dd, 1H, $J_{1,2}$ 3.5 $J_{2,3}$ 10.4 Hz, H-2), 5.43–5.55 (m, 3H, H-3, 3', 4')and 5.68 (d, 1H, $J_{1,2}$ 3.5 Hz, H-1). ¹³C NMR δ 170.6, 170.5, 170.2, 169.6 (COCH₃), 104.8 (C-2'), 90.0 (C-1), 81.8 (C-5'), 76.7, 74.2 (C-3', 4'), 71.0, 70.9, 70.7 (C-2, 3, 5), 64.1, 62.3, 61.5 (C-1', 6, 6'), 54.9 (C-4) and 20.7, 20.6, 20.5 (COCH₃). EI-MS m/z (%): 307, 309 [M - C₁₀H₁₅O₈]⁺ (3:1, 24), 247 [M - C₁₂H₁₆ClO₈]⁺ (51), 225, 207 (3:1, 24), 187 (41), 43 (100).

Anal. Calcd for $C_{22}H_{31}ClO_{15}$: C, 46.28; H, 5.47; Cl, 6.21. Found: C, 46.57; H, 5.63; Cl, 6.33. HRMS-ESI (positive mode): calcd for $[M + Na]^+$ 593.1294, 595.1219. Found: 593.1263, 595.1247 (3:1).



3,4-Di-O-acetyl-β-D-fructofuranosyl 2,3,6-tri-O-acetyl-4-bromo-4-deoxy- α -D-glucopyranoside (8). Treatment of 5 (5.97 g, 5.44 mmol) as for 7 gave 8 as colourless crystals (2.8 g, 84%), mp 77–80°C; $[\alpha]_D$ +26.9° (c 1.01, CHCl₃); ¹H NMR δ 2.06–2.21 (s, 15H, 5 × CH₃), 3.56–3.86 (m, 4H, H-1'a, b and H-6'a, b), 3.95 (t, 1H, $J_{3,4} = J_{4,5}$ 10.3 Hz, H-4), 4.05–4.57 (m, 4H, H-5, 5', 6a, b), 4.82 (dd, 1H, $J_{1,2}$ 3.5 $J_{2,3}$ 10.3 Hz, H-2), 5.31–5.48 (m, 2H, H-3', 4'), 5.56 (t, 1H, $J_{2,3}$ = $J_{3,4}$ 10.3 Hz, H-3) and 5.70 (d, 1H, $J_{1,2}$ 3.5 Hz, H-1). ¹³C NMR δ 170.6, 170.4, 170.2, 170.1, 169.4 (COCH₃), 104.8 (C-2'), 90.1 (C-1), 81.8 (C-5'), 76.7, 74.3 (C-1) 3', 4'), 71.1, 70.7 (C-2, 3, 5), 64.1, 63.0, 61.5 (C-1', 6, 6'), 45.8 (C-4) and 20.7, $20.6, 20.5, 19.6 \text{ (COCH}_3)$. EI-MS m/z (%): 351, 353 [M - C₁₀H₁₅O₈]⁺ (1:1, 7), 291, 293 (1:1, 12), 247 [M - $C_{12}H_{16}BrO_8$] (21), 231, 233 (1:1, 9), 189, 191 (1:1, 10), 187 (13), 169 (29), 43 (100); exact mass calcd for $[C_{12}H_{16}BrO_7]^+$ 351.0079, 353.0059, found 351.0059, 353.0038 (1:1); calcd for $[C_{10}H_{15}O_7]^+$ 247.0817, found 247.0813.

REPRINTS

Anal. Calcd for C₂₂H₃₁BrO₁₅: C, 42.94; H, 5.08; Br, 12.98. Found: C, 42.67; H, 5.19; Br, 13.32.

3,4-Di-O-acetyl-β-D-fructofuranosyl 2,3,6-tri-O-acetyl-4-deoxy-4-iodo- α -D-glucopyranoside (9). Treatment of 6 (1.29 g, 1.12 mmol) as for 7 gave 9 as colourless crystals (0.56 g, 75%), mp 93–95°C; $[\alpha]_D + 15.5^\circ$ (c 1.08, CHCl₃); ¹H NMR δ 2.06–2.21 (s, 15H, 5 × CH₃), 3.56–3.86 (m, 4H, H-1'a, b and H-6'a, b), 3.95 (t, 1H, $J_{3,4} = J_{4,5}$ 10.3 Hz, H-4), 4.05–4.57 (m, 4H, H-5, 5', 6a, b), 4.82 (dd, 1H, $J_{1,2}$ 3.5 $J_{2,3}$ 10.3 Hz, H-2), 5.31–5.48 (m, 2H, H-3', 4'), 5.56 (t, 1H, $J_{2,3} = J_{3,4}$ 10.3 Hz, H-3) and 5.70 (d, 1H, $J_{1.2}$ 3.5 Hz, H-1). ¹³C NMR δ 170.6, 170.4, 170.2, 170.1, 169.4 (COCH₃), 104.8 (C-2'), 90.3 (C-1), 81.9 (C-5'), 76.8, 74.4 (C-3', 4'), 72.0, 71.6, 71.0 (C-2, 3, 5), 64.5, 64.2, 61.5 (C-1', 6, 6'), 24.3 (C-4) and 20.8, 20.7, 20.6 (COCH₃). EI-MS m/z (%): 399 [M - C₁₂H₁₆IO₈]⁺ (21), 289 (16), 279 (27), $247 \left[M - C_{10} H_{15} O_8 \right]^+$ (15), 211 (25), 187 (19), 169 (41), 45 (100); exact mass calcd for $[C_{12}H_{16}IO_7]^+$ 398.9941, found 398.9957; calcd for $[C_{10}H_{15}O_7]^+$ 247.0817, found 247.0812.

Anal. Calcd for C₂₂H₃₁IO₁₅: C, 39.89; H, 4.72; I, 19.16. Found: C, 40.14; H, 5.11; I, 19.22.

β-D-Fructofuranosyl 4-chloro-4-deoxy-α-**D-glucopyranoside** (10). A solution of 7 (0.23 g, 0.40 mmol) in dry methanol (10 mL) was adjusted to pH \sim 10 with methanolic sodium methoxide (0.5 M). It was stirred at room temperature for 4 h. It was then neutralized with Amberlite IR120 (H⁺) ion exchange resin and concentrated. Flash chromatography (chloroform/methanol, 4:1) gave 10 as a colourless syrup (0.12 g, 82%), $[\alpha]_D + 53.5^\circ$ (c 0.52, H₂O); ¹H NMR (D₂O) δ 4.05 (dd, 1H, $J_{1,2}$ 3.8 $J_{2,3}$ 9.8 Hz, H-2), 4.66 (d, 1H, $J_{3',4'}$ 8.7 Hz, H-3') and 5.91 (d, 1H, $J_{1,2}$ 3.8 Hz, H-1). ¹³C NMR (D₂O) δ 106.4 (C-2'), 94.8 (C-1), 84.0 (C-5'), 79.1, 76.7 (C-3', 4'), 75.3, 75.2, 74.2 (C-2, 3, 5), 65.0, 64.0, 63.0 (C-1', 6, 6') and 60.5 (C-4).

Anal. Calcd for C₁₂H₂₁ClO₁₀: C, 39.93; H, 5.87; Cl, 9.83. Found: C, 40.27; H, 5.69; Cl, 10.02. HRMS-ESI (positive mode): calcd for [M + Na]⁺ 383.0721, 385.0691. Found: 383.0733, 385.0720 (3:1).



β-**D-Fructofuranosyl 4-bromo-4-deoxy-α-D-glucopyranoside** (11). Treatment of **8** (0.24 g, 0.39 mmol) as for **10** gave **11** as a colourless syrup (0.14 g, 89%), $[\alpha]_D$ +34.7° (c 0.49, H_2O); 1H NMR (D_2O) δ 4.11 (dd, 1H, $J_{1,2}$ 3.8 $J_{2,3}$ 9.7 Hz, H-2), 4.73 (d, 1H, $J_{3',4'}$ 8.7 Hz, H-3') and 6.01 (d, 1H, $J_{1,2}$ 3.8 Hz, H-1). ^{13}C NMR (D_2O) δ 106.4 (C-2'), 94.8 (C-1), 84.0 (C-5'), 79.0, 76.6 (C-3', 4'), 75.3, 75.2, 74.4 (C-2, 3, 5), 64.9, 63.9, 63.7 (C-1', 6, 6') and 52.9 (C-4).

Anal. Calcd for $C_{12}H_{21}BrO_{10}$: C, 35.57; H, 5.22; Br, 19.72. Found: C, 35.42; H, 5.69; Br, 19.52. HRMS-ESI (positive mode): calcd for $[M + Na]^+$ 427.0215, 429.0195. Found: 427.0232, 429.0219 (1:1).

β-D-Fructofuranoside 4-deoxy-4-iodo-α-D-glucopyranosyl (12). Deacetylation of **9** (0.23 g, 0.35 mmol) as for **10** gave **12** as a colourless syrup (0.13 g, 83%): $[\alpha]_D$ +39.1° (c 1.62, H₂O); ¹H NMR (D₂O) δ 4.06 (dd, 1H, $J_{1,2}$ 3.8 $J_{2,3}$ 9.1 Hz, H-2), 4.71 (d, 1H, $J_{3',4'}$ 8.7 Hz, H-3') and 5.98 (d, 1H, $J_{1,2}$ 3.8 Hz, H-1). ¹³C NMR (D₂O) δ 106.4 (C-2'), 95.0 (C-1), 84.0 (C-5'), 79.1, 76.6 (C-3', 4'), 76.2, 75.9, 74.3 (C-2, 3, 5), 65.4, 65.0, 63.9 (C-1', 6, 6') and 33.5 (C-4).

Anal. Calcd for $C_{12}H_{21}IO_{10}$: C, 31.87; H, 4.68; I, 28.06. Found: C, 31.49; H, 4.69; I, 27.92. HRMS-ESI (positive mode): calcd for $[M + Na]^+$ 475.0077. Found: 475.0077.

3,4-Di-O-acetyl-1,6-dideoxy-1,6-dichloro-β-D-fructofuranosyl 2,3,6-tri-**O-acetyl-4-bromo-4-deoxy-\alpha-D-glucopyranoside (14).** To a solution of **8** (0.79) g, 1.29 mmol) in dry CH₂Cl₂-pyridine (15:1, 32 mL) at -78°C, was added trifluoromethanesulfonic anhydride (0.86 mL, 5.24 mmol). The mixture was stirred at -78°C for 15 min and then at 0°C for 2 h. It was diluted with dichloromethane and the organic solution was washed successively with aq KHSO₄ (10%), satd NaHCO₃, and water. After drying (Na₂SO₄) and concentration, the crude product 13 (0.47 g, 0.53 mmol) was dissolved in dry acetone (20 mL) and lithium chloride (0.10 g) was added to the solution. The mixture was then allowed to stir overnight at room temperature, and concentrated. The residue was taken up in dichloromethane, filtered, and the organic layer was washed with satd NaHCO₃, water, dried and concentrated. Flash chromatography (ether/hexane, 3:1) gave 14 as colourless crystals (0.30 g, 86%): 50–52°C; $[\alpha]_D$ +22.9° (c 1.11, CHCl₃); ¹H NMR δ 2.07–2.19 (s, 15H, 5 \times CH₃), 3.56–3.79 (m, 4H, H-1'a, b and H-6'a, b), 3.92 (t, 1H, $J_{3,4} = J_{4,5}$ 10.4 Hz, H-4), 4.22–4.60 (m, 4H, H-5, 5', 6a, b), 4.83 (dd, 1H, $J_{1,2}$ 3.5 $J_{2,3}$ 10.1 Hz, H-2), 5.38 $J_{1,2}$ 3.5 Hz, H-1) and 5.67 (d, 1H, $J_{3',4'}$ 5.9 Hz, H-3'). ¹³C NMR δ 170.4, 170.0, 169.7, 169.5, 169.4 (COCH₃), 104.4 (C-2'), 90.3 (C-1), 81.4 (C-5'), 77.0, 75.9 (C-1) 3', 4'), 70.9, 70.8 (C-2, 3, 5), 63.3 (C-6), 45.8 (C-4), 44.8 (C-1'), 43.8 (C-6') and 20.7, 20.6, 20.5, 20.4 (COCH₃). EI-MS m/z (%): 351, 353 [M - C₁₀H₁₃Cl₂O₆]⁺ (1:1, 39), 347 (43), 292, 294 (1:1, 52), 285 (46), 284, 286, 288 $[M - C_{12}H_{16}BrO_8]^+$ (9:6:1, 14), 283 (61), 249, 251 (15), 231, 233 (1:1, 59), 225 (55), 224, 226, 228 (16), 223 (69), 189, 191 (1:1, 50), 43 (100), exact mass calcd for $(C_{12}H_{16}BrO_7)$ 351.0079, 353.0058, found 351.0076, 353.0054 (1:1); calcd for $(C_{10}H_{13}Cl_2O_5)$ 283.0141, 285.0114, 287.0081, found 283.0136, 285.0114, 287.0077 (8:5:1)

Anal. Calcd for $C_{22}H_{29}BrCl_2O_{13}$: C, 40.51; H, 4.48; Br, 12.25; Cl, 10.87. Found: C, 40.32; H, 4.58; Br, 11.19.

3,4-Di-O-acetyl-1,6-dibromo-1,6-dideoxy-β-D-fructofuranosyl 2,3,6-tri-*O*-acetyl-4-bromo-4-deoxy- α -D-glucopyranoside (15). Treatment of a solution of 13 (6.54 g, 7.45 mmol) in acetone (60 mL) with lithium bromide (3.0 g) as for **14** gave, after column chromatography, crystalline **15** (4.5 g, 82%), mp 72–73°C; $[\alpha]_D$ +24.6° (c 2.79, CHCl₃); ¹H NMR δ 2.07–2.21 (s, 15H, 5 × CH₃), 3.47–3.64 (m, 4H, H-1'a, b and H-6'a, b), 3.92 (t, 1H, $J_{3,4} = J_{4,5}$ 10.4 Hz, H-4), 4.26–4.60 (m, 4H, H-5, 5', 6a, b), 4.84 (dd, 1H, $J_{1,2}$ 3.5 $J_{2,3}$ 10.4 Hz, H-2), 5.35 (t, 1H, $J_{3',4'}$ = $J_{4',5'}$ 6.0 Hz, H-4'), 5.56 (t, 1H, $J_{2,3} = J_{3,4}$ 10.4 Hz, H-3), 5.66 (d, 1H, $J_{1,2}$ 3.5 Hz, H-1) and 5.71 (d, 1H, $J_{3',4'}$ 6.0 Hz, H-3'). ¹³C NMR δ 170.3, 170.0, 169.7, 169.4, 169.3 (COCH₃), 103.9 (C-2'), 90.4 (C-1), 81.2 (C-5'), 77.2 (C-3', 4'), 71.0, 70.8, 70.7, (C-2, 3, 5), 63.3 (C-6), 45.8 (C-4), 32.6 (C-1'), 31.2 (C-6') and 20.7, 20.6, 20.4 (COCH₃). EI-MS m/z (%): 371, 373, 375 [M - C₁₂H₁₆O₈]⁺ (1:2:1, 39), 351, 353 $[M - C_{10}H_{13}Br_2O_6]^+$ (1:1, 44), 291, 293 (1:1, 50), 269, 271, 273 (1:2:1, 15), 249, 252 (1:1, 10), 231, 233 (1:1, 55), 189, 191 (1:1, 56), 169 (46), 43 (100); exact mass calcd for $[C_{10}H_{13}Br_2O_5]^+$ 370.9129, 372.9109, 374.9089, found 370.9128, 372.9111, 374.9090 (1:2:1); calcd for $[C_{12}H_{16}BrO_7]^+$ 351.0079, 353.0058, found 351.0076, 353.0061 (1:1).

Anal. Calcd for $C_{22}H_{29}Br_3O_{13}$: C, 35.65; H, 3.94; Br, 32.34. Found: C, 35.42; H, 3.51; Br, 32.32.

3,4-Di-O-acetyl-1,6-dideoxy-1,6-diiodo-β-D-fructofuranosyl 2,3,6-tri-Oacetyl-4-bromo-4-deoxy-α-D-glucopyranoside (16). A solution of 13 (0.49 g, 0.56 mmol) in acetone (20 mL) was stirred with potassium iodide (0.40 g) for ~15 h at room temperature. The solvent was removed and the residue taken up in dichloromethane. The suspension was filtered and the organic solution was washed with aq Na₂S₂O₃ (10%), satd NaHCO₃, water, dried (Na₂SO₄) and concentrated. Flash chromatography (ether/hexane, 3:1) gave 16 as colourless crystals (0.38 g, 82%), mp 75–77°C; $[\alpha]_D$ +21.9° (c 1.81, CHCl₃); ¹H NMR δ 2.10-2.20 (s, 15H, 5 × CH₃), 3.41-3.43 (m, 4H, H-1'a, b and H-6'a, b), 3.93 (t, 1H, $J_{3,4} = J_{4,5}$ 10.4 Hz, H-4), 4.21–4.59 (m, 4H, H-5, 5', 6a, b), 4.84 (dd, 1H, $J_{1,2}$ 3.6 $J_{2,3}$ 10.4 Hz, H-2), 5.30 (t, 1H, $J_{3',4'} = J_{4',5'}$ 6.0 Hz, H-4'), 5.58 (t, 1H, $J_{2,3} = J_{3,4}$ 10.4 Hz, H-3), 5.64 (d, 1H, $J_{1,2}$ 3.6 Hz, H-1) and 5.70 (d, 1H, $J_{3',4'}$ 6.0 Hz, H-3'). ¹³C NMR δ 170.4, 169.6, 169.4 (COCH₃), 103.3 (C-2'), 90.7 (C-1), 80.9 (C-5'), 77.8, 77.6 (C-3', 4'), 70.8, 70.6 (C-2, 3, 5), 63.3 (C-6), 45.8 (C-4), 20.9, 20.8, 20.6, 20.5 (COCH₃), 6.3 (C-1'), and 3.4 (C-6'). EI-MS m/z(%): 467 $[M - C_{12}H_{16}BrO_8]^+$ (43), 351, 353 $[M - C_{10}H_{13}I_2O_6]^+$ (1:1, 44), 291, 293 (1:1, 65), 249, 251 (1:1, 14), 237 (51%), 231, 233 (1:1, 65), 189, 191 (1:1, 48), 169 (72), 43 (100); exact mass calcd for $[C_{10}H_{13}I_2O_5]^+$ 466.8853, found 466.8858; exact mass calcd for $[C_{12}H_{16}BrO_7]^+$ 351.0079, 353.0058, found 351.0073, 353.0055 (1:1).

Anal. Calcd for $C_{22}H_{29}BrI_2O_{13}$: C, 31.64; H, 3.50; Br, 9.57, I, 30.39. Found: C, 31.54; H, 3.57; Br, 9.32, I, 29.99.



1,6-Dichloro-1,6-dideoxy-β-**D-fructofuranosyl 4-bromo-4-deoxy-**α-**D-glucopyranoside** (**17**). Compound **14** (0.33 g, 0.51 mmol) was treated with sodium methoxide (0.566 M) as described for **10** to afford **17** as a colourless syrup (0.18 g, 80%): [α]_D +54.5° (c 1.44, H₂O); ¹H NMR (D₂O) δ 4.12 (dd, 1H, $J_{1,2}$ 3.8 $J_{2,3}$ 9.4 Hz, H-2), 4.91 (d, 1H, $J_{3',4'}$ 8.7 Hz, H-3') and 6.01 (d, 1H, $J_{1,2}$ 3.8 Hz, H-1). ¹³C NMR (D₂O) δ 106.1 (C-2'), 95.5 (C-1), 83.7 (C-5'), 79.0, 77.9 (C-3', 4'), 75.6, 75.2, 74.4 (C-2, 3, 5), 63.9 (C-6), 53.0 (C-4), 47.5 (C-6') and 46.0 (C-1').

Anal. Calcd for $C_{12}H_{19}BrCl_2O_8$: C, 32.60; H, 4.32; Br, 18.07, Cl, 16.04. Found: C, 32.45; H, 4.73; Br, 18.29, Cl, 16.09. HRMS-ESI (positive mode): calcd for $[M + Na]^+$ 462.9539, 464.9519, 466.9489, 468.9459. Found: 462.9540, 464.9514, 466.9491, 468.9433 (10:16:7:1).

1,6-Dibromo-1,6-dideoxy-β-**D-fructofuranosyl 4-bromo-4-deoxy-**α-**D-glucopyranoside** (**18**). Compound **15** (0.16 g, 0.22 mmol) was deacetylated as for **10** to afford **18** as a colourless syrup (0.10 g, 87%), [α]_D +47.4° (c 1.90, H₂O); ¹H NMR (D₂O) δ 4.12 (dd, 1H, J_{1,2} 3.8 J_{2,3} 9.8 Hz, H-2), 4.17–4.31 (m, 4H), 4.35–4.77 (m, 7H), 4.97 (d, 1H, J_{3',4'} 8.0 Hz, H-3') and 6.01 (d, 1H, J_{1,2} 3.8 Hz, H-1). ¹³C NMR (D₂O) δ 103.6 (C-2'), 93.5 (C-1), 81.5 (C-5'), 77.8, 77.0 (C-3',4'), 73.6, 73.2, 72.4 (C-2, 3, 5), 61.9 (C-6), 51.0 (C-4), 33.5, 32.4 (C-1', 6').

Anal. Calcd for $C_{12}H_{19}Br_3O_8$: C, 27.14; H, 3.61; Br, 45.14. Found: C, 26.84; H, 3.77; Br, 44.89. HRMS-ESI (positive mode): calcd for $[M + Na]^+$ 550.8519, 552.8507, 554.8487, 556.8467. Found: 550.8519, 552.8526, 554.8486, 556.8468 (1:3:3:1).

1,6-Dideoxy-1,6-diiodo-β-**D-fructofuranosyl 4-bromo-4-deoxy-**α-**D-glucopyranoside** (**19**). Compound **16** (0.43 g, 0.51 mmol) was deacetylated as for **10** to afford **19** as a colourless syrup (0.26 g, 81%), $[\alpha]_D$ +43.5° (c 0.83, H_2O); 1H NMR (D_2O) δ 3.86–4.74 (m, 11H), 4.97 (d, 1H, $J_{3',4'}$ 8.4 Hz, H-3') and 5.99 (d, 1H, $J_{1,2}$ 3.8 Hz, H-1) 13 C NMR (D_2O) δ 102.9 (C-2'), 93.4 (C-1), 81.4 (C-5'), 79.1, 78.3 (C-3',4'), 73.4, 73.0, 72.2 (C-2, 3, 5), 61.8 (C-6), 50.8 (C-4), 6.2, 5.6 (C-1', 6').

Anal. Calcd for $C_{12}H_{19}BrI_2O_8$: C, 23.06; H, 3.06; Br, 12.79, I, 40.61. Found: C, 23.34; H, 3.27; Br, 12.92, I, 40.99. HRMS-ESI (positive mode): calcd for [M + Na]⁺ 646.8251, 648.8231. Found 646.8250, 648.8229 (1:1).

3,4-Di-*O*-acetyl-1-chloro-1,6-dideoxy-6-iodo-β-D-fructofuranosyl 2,3,6-tri-*O*-acetyl-4-bromo-4-deoxy-α-D-glucopyranoside (20). A solution of 13 (0.93 g, 1.06 mmol) in acetone (25 mL) was stirred with potassium iodide (0.35 g) for 30 min at 0°C. The mixture was concentrated and the residue was taken up in dichloromethane. Work-up in the usual manner gave a syrupy residue, which was dissolved (0.49 g, 0.57 mmol) in acetone (15 mL) and treated with lithium chloride (0.10 g) at room temperature overnight and worked up in the usual way. Flash chromatography (ethyl acetate/hexane, 1:2) gave 20 as colourless crystals (0.35 g, 63%), mp 49–51°C; [α]_D +24.8° (*c* 1.08, CHCl₃); ¹H NMR δ 2.08–2.18 (s, 15H, 5 × CH₃), 3.39–3.77 (m, 4H, H-1'a, b and H-6'a, b), 3.93 (t, 1H, $J_{3.4} = J_{4.5}$ 10.3

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Hz, H-4), 4.22-4.59 (m, 4H, H-5, 5′, 6a, b), 4.84 (dd, 1H, $J_{1,2}$ 3.5 $J_{2,3}$ 10.3 Hz, H-2), 5.31 (t, 1H, $J_{3',4'} = J_{4',5'}$ 6.0 Hz, H-4′), 5.56 (t, 1H, $J_{2,3} = J_{3,4}$ 10.3 Hz, H-3) and 5.65–5.69 (m, 2H, H-1, 3′) ¹³C NMR δ 170.4, 170.0, 169.4 (COCH₃), 104.3 (C-2′), 90.4 (C-1), 81.5 (C-5′), 77.3, 76.3 (C-3′, 4′), 71.0, 70.9, 70.8 (C-2, 3, 5), 63.3 (C-6), 45.8 (C-4), 44.7 (C-1′), 20.8, 20.7, 20.6, 20.5 (COCH₃) and 3.3 (C-6′). EI-MS m/z(%): 375, 377 [M - C₁₂H₁₆BrO₈]⁺ (3:1, 63), 351, 353 [(C₁₀H₁₃CIIO₆]⁺ (1:1, 64), 317 (75), 315 (63), 291, 293 (1:1, 71), 275 (45), 249, 251 (1:1, 24), 231, 233 (1:1, 74), 189, 191 (1:1, 78), 169 (74), 43 (100), exact mass calcd for [C₁₀H₁₃CIIO₆]⁺ 374.9497, 376.9467, found 374.9489, 376.9471; calcd for [C₁₂H₁₆BrO₇]⁺ 351.0079, 353.0058, found 351.0078, 353.0061 (1:1).

Anal. Calcd for C₂₂H₂₉BrClIO₁₃: C, 35.53; H, 3.93; Br, 10.74, Cl, 4.77; I, 17.02. Found: C, 35.41; H, 3.73; Br, 11.12; Cl, 4.78; I, 16.87.

3,4-Di-O-acetyl-1-bromo-1,6-dideoxy-6-iodo-β-D-fructofuranosyl 2,3,6tri-O-acetyl-4-bromo-4-deoxy- α -D-glucopyranoside (21). Treatment of 13 (0.46 g, 0.52 mmol) with potassium iodide (0.18 g) and then lithium bromide (0.12 g) as for **20** gave crystalline **21** (0.30 g, 67%), mp 58–61°C; $[\alpha]_D$ +22.3° (c 1.27, CHCl₃); ¹H NMR δ 2.08–2.20 (s, 15H, 5 × CH₃), 3.37–3.63 (m, 4H, H-1'a, b and H-6'a, b), 3.93 (t, 1H, $J_{3,4} = J_{4,5}$ 10.3 Hz, H-4), 4.23–4.59 (m, 4H, H-5, 5', 6a, b), 4.85 (dd, 1H, $J_{1,2}$ 3.5 $J_{2,3}$ 10.3 Hz, H-2), 5.30 (t, 1H, $J_{3',4'} = J_{4',5'}$ 6.0 Hz, H-4'), 5.56 (t, 1H, $J_{2,3} = J_{3,4}$ 10.3 Hz, H-3), 5.65 (d, 1H, $J_{1,2}$ 3.5 Hz, H-1) and 5.72 (d, 1H, $J_{3',4'}$ 6.0 Hz, H-3'). ¹³C NMR δ 170.4, 170.1, 169.8, 169.5, 169.4 (COCH₃), 103.8 (C-2'), 90.5 (C-1), 81.4 (C-5'), 77.6, 77.1 (C-3', 4'), 71.0, 70.9, 70.8 (C-2, 3, 5), 63.3 (C-6), 45.8 (C-4), 32.6 (C-1'), 20.8, 20.7, 20.6, 20.5 (COCH₃) and 3.3 (C-6'). EI-MS m/z(%): 419, 421 [M -C₁₀H₁₃BrIO₆]⁺ (1:1, 68), 359, 361 (1:1, 75), $351, 353 [M - C_{12}H_{16}BrO_8]^+$ (1:1, 55), 317, 319 (1:1, 40), 291, 293 (1:1, 63), 249, 251 (1:1, 21), 231, 233 (1:1, 71), 189, 191 (1:1, 58), 169 (60), 43 (100); exact mass calcd for $[C_{10}H_{13}BrIO_5]^+$ 418.8991, 420.8971, found 418.8987, 420.8968 (1:1); exact mass calcd for $[C_{12}H_{16}BrO_7]^+$ 351.0079, 353.0058, found 351.0079, 353.0056 (1:1).

Anal. Calcd for $C_{22}H_{29}Br_2IO_{13}$: C, 33.53; H, 3.71; Br, 20.28; I, 16.10. Found: C, 33.44; H, 3.37; Br, 20.12, I, 16.37.

3,4-di-*O*-acetyl-6-chloro-1,6-dideoxy-1-iodo-β-D-fructofuranosyl 2,3,6-tri-*O*-acetyl-4-bromo-4-deoxy-α-D-glucopyranoside (22). Compound 13 (0.33 g, 0.37 mmol) in acetone (10 mL) was stirred with lithium chloride (0.05 g) at 0°C for 30 minutes. The mixture was concentrated, then taken up in dichloromethane, filtered, and the organic solution was washed with satd NaHCO₃, water, dried (Na₂SO₄) and concentrated. The residue (0.24 g, 0.31 mmol) was dissolved in acetone (10 mL) and stirred with potassium iodide (0.15 g) at room temperature overnight, and then worked-up in the usual method. Flash chromatography (ethyl acetate/hexane, 1:2) on the crude product gave 22 as colourless crystals (0.19 g, 60%), 50–52°C; [α]_D +36.3° (*c* 0.52, CHCl₃); ¹H NMR δ 2.03–2.11 (s, 15H, 5 × CH₃), 3.40 (s, 2H, H-1'a, b), 3.70 (d, 2H, $J_{5',6'a,b}$ 6.3 Hz, H-6'a, b), 3.86 (t, 1H, $J_{3,4} = J_{4,5}$ 10.4 Hz, H-4), 4.17 (q, 1H, $J_{4',5'}$ 5.9 $J_{5',6'a,b}$ 6.3 Hz, H-5'),

4.29 (dd, 1H, $J_{5,6a}$ 4.5 $J_{6a,6b}$ 12.2 Hz, H-6a), 4.40 (m, 1H, H-5), 4.49 (dd, 1H, $J_{5,6b}$ 1.7 $J_{6a,6b}$ 12.2 Hz, H-6b), 4.77 (dd, 1H, $J_{1,2}$ 3.5 $J_{2,3}$ 10.4 Hz, H-2), 5.29 (t, 1H, $J_{3',4'}$ = $J_{4',5'}$ 5.9 Hz, H-4'), 5.50 (t, 1H, $J_{2,3}$ = $J_{3,4}$ 10.4 Hz, H-3), 5.57 (d, 1H, $J_{1,2}$ 3.5 Hz, H-1) and 5.63 (d, 1H, $J_{3',4'}$ 5.9 Hz, H-3'). ¹³C NMR δ 170.3, 170.0, 169.6, 169.5, 169.2 (COCH₃), 103.3 (C-2'), 90.4 (C-1), 80.9 (C-5'), 77.3, 76.1 (C-3', 4'), 70.7, 70.6, 70.5 (C-2, 3, 5), 63.2 (C-6), 45.7 (C-4), 43.8 (C-6'), 20.6, 20.5, 20.4 (COCH₃) and 6.4 (C-1'). EI-MS m/z (%): 375, 377 [M - $C_{12}H_{16}BrO_{8}$] + (3:1, 64), 351, 353 [M - $C_{10}H_{13}CIIO_{6}$] + (1:1, 65), 291, 293 (1:1, 66), 249, 251 (1:1, 24) 231, 233 (1:1, 74), 189, 191 (1:1, 78), 169 (100), 43 (90); exact mass calcd for $[C_{10}H_{13}CIIO_{5}]$ + 374.9497, 376.9470, found 374.9489, 376.9471 (3:1); exact mass calcd for $[C_{12}H_{16}BrO_{7}]$ + 351.0079, 353.0058, found 351.0078, 353.0061 (1:1).

Anal. Calcd for C₂₂H₂₉BrClIO₁₃: C, 35.53; H, 3.93; Br, 10.74, Cl, 4.77; I, 17.02. Found: C, 34.41; H, 3.73; Br, 11.12, I, 16.87.

3,4-Di-O-acetyl-6-bromo-1,6-dideoxy-1-iodo-β-D-fructofuranosyl 2,3,6tri-O-acetyl-4-bromo-4-deoxy- α -D-glucopyranoside (23). Treatment of 13 (0.45 g, 0.51 mmol) in acetone (15 mL) with lithium bromide (0.08 g) and then potassium iodide (0.22 g) gave 23 as colourless crystals (0.29 g, 67%), mp 50–52°C; [α]_D +30.9° (c 1.07, CHCl₃); ¹H NMR δ 2.03–2.10 (s, 15H, 5 × CH₃), 3.40 (s, 2H, H-1'a, b), 3.55 (d, 2H, $J_{5',6'a,b}$ 6.4 Hz, H-6'a, b), 3.87 (t, 1H, $J_{3,4} = J_{4,5}$ 10.4 Hz, H-4), 4.21 (q, 1H, $J_{4',5'}$ 6.0 $J_{5',6'a,b}$ 6.4 Hz, H-5'), 4.29 (dd, 1H, $J_{5,6a}$ 4.5 $J_{6a,6b}$ 12.2 Hz, H-6a), 4.42 (m, 1H, H-5), 4.49 (dd, 1H, $J_{5,6b}$ 1.7 $J_{6a,6b}$ 12.2 Hz, H-6b), 4.77 (dd, 1H, $J_{1,2}$ 3.5 $J_{2,3}$ 10.4 Hz, H-2), 5.28 (t, 1H, $J_{3',4'} = J_{4',5'}$ 5.9 Hz, H-4'), 5.50 (t, 1H, $J_{2,3} = J_{3,4}$ 10.4 Hz, H-3), 5.57 (d, 1H, $J_{1,2}$ 3.5 Hz, H-1) and 5.63 (d, 1H, $J_{3',4'}$ 6.0 Hz, H-3'). ¹³C NMR δ 170.3, 170.1, 169.7, 169.5, 169.3 (COCH₃), 103.4 (C-2'), 90.5 (C-1), 80.7 (C-5'), 77.7, 77.0 (C-3', 4'), 70.9, 70.8, 70.7 (C-2, 3, 5), 63.3 (C-6), 45.8 (C-4), 31.3 (C-6'), 20.7, 20.6, 20.5 (COCH₃) and 6.5 (C-1'). EI-MS m/z (%): 419, 421 [M - $C_{12}H_{16}BrO_{8}$] + (1:1, 55), 359, 361 (1:1, 67), 351, $353 [M - C_{10}H_{13}BrIO_6]^+ (1:1, 56), 317, 319 (1:1, 28), 291, 293 (1:1, 53), 249, 251$ (1:1, 12), 231, 233 (1:1, 56), 189, 191 (1:1, 78), 169 (66), 43 (100); exact mass calcd for $[C_{10}H_{13}BrIO_6]^+$ 418.8991, 420.8971, found 418.9000, 420.8980 (1:1); exact mass calcd for $[C_{12}H_{16}BrO_7]^+$ 351.0079, 353.0058, found 351.0080, 353.0060 (1:1).

Anal. Calcd for $C_{22}H_{29}Br_2IO_{13}$: C, 33.53; H, 3.71; Br, 20.28; I, 16.10. Found: C, 33.86; H, 3.97; Br, 20.54; I, 15.92.

1-Chloro-1,6-dideoxy-6-iodo-β-**D-fructofuranosyl 4-bromo-4-deoxy-**α-**D-glucopyranoside (24).** Deacetylation of **20** (0.20 g, 0.27 mmol) as described for **10** afforded **24** as a colourless syrup (0.12 g, 84%), [α]_D +55.6° (c 1.34, H₂O); ¹H NMR (D₂O) δ 4.12 (dd, 1H, $J_{1,2}$ 3.8 $J_{2,3}$ 9.4 Hz, H-2), 4.91 (d, 1H, $J_{3',4'}$ 8.7 Hz, H-3') and 6.02 (d, 1H, $J_{1,2}$ 3.8 Hz, H-1). ¹³C NMR (D₂O) δ 106.0 (C-2'), 95.5 (C-1), 83.8 (C-5'), 80.3, 79.5 (C-3', 4'), 75.6, 75.2, 74.4 (C-2, 3, 5), 64.0 (C-6), 53.0 (C-4), 45.9 (C-1') and 8.1 (C-6').

Anal. Calcd for $C_{12}H_{19}BrClIO_8$: C, 27.01; H, 3.59; Br, 14.98; Cl, 6.64; I, 23.79. Found: C, 36.97; H, 3.72; Br, 14.61; Cl, 6.33; I, 24.09. HRMS-ESI (positive



mode): calcd for $[M + Na]^+$ 554.8895, 556.8875, 558.8845. Found: 554.8906, 556.8868, 556.8856 (3.7:4.7:1).

REPRINTS

1-Bromo-1,6-dideoxy-6-iodo-β-D-fructofuranosyl 4-bromo-4-deoxy-α-D**glucopyranoside** (25). Deacetylation of 22 (0.29 g, 0.37 mmol) as above gave **25** as a colourless syrup (0.18 g, 85%), $[\alpha]_D$ +49.7° (c 1.53, H₂O); ¹H NMR (D₂O) δ 4.10 (dd, 1H, $J_{1,2}$ 3.8 $J_{2,3}$ 9.7 Hz, H-2), 4.96 (d, 1H, $J_{3',4'}$ 8.4 Hz, H-3') and 6.00 (d, 1H, $J_{1,2}$ 3.8 Hz, H-1). ¹³C NMR (D₂O) δ 103.3 (C-2'), 93.3 (C-1), 81.5 (C-5'), 78.1, 77.9 (C-3', 4'), 73.4, 73.0, 72.2 (C-2, 3, 5), 61.8 (C-6), 50.7 (C-4), 32.1 (C-1') and 5.6 (C-6').

Anal. Calcd for C₁₂H₁₉Br₂IO₈: C, 24.94; H, 3.62; Br, 27.65; I, 21.96. Found: C, 27.39; H, 3.72; Br, 14.61; I, 24.09. HRMS-ESI (positive mode): calcd for [M + Na]⁺ 598.8389, 600.8369, 602.8349. Found: 598.8391, 600.8372, 602.8354 (1:2:1).

6-Chloro-1,6-dideoxy-1-iodo-β-D-fructofuranosyl 4-bromo-4-deoxy-α-**D-glucopyranoside (26).** Deacetylation of **21** (0.21 g, 0.28 mmol) as above gave **26** as a colourless syrup (0.12 g, 80%), $[\alpha]_D$ +45.1° (c 0.92, H₂O); ¹H NMR (D₂O) δ 4.12 (dd, 1H, $J_{1,2}$ 3.8 $J_{2,3}$ 9.8 Hz, H-2), 4.99 (d, 1H, $J_{3',4'}$ 8.4 Hz, H-3') and 6.00 (d, 1H, $J_{1.2}$ 3.8 Hz, H-1). ¹³C NMR (D₂O) δ 105.3 (C-2'), 95.6 (C-1), 83.6 (C-5'), 80.9, 78.0 (C-3', 4'), 75.6, 75.2, 74.5 (C-2, 3, 5), 63.9 (C-6), 53.0 (C-4), 47.4 (C-6') and 8.5 (C-1').

Anal. Calcd for C₁₂H₁₉BrClIO₈: C, 27.01; H, 3.59; Br, 14.98; Cl, 6.64; I, 23.79. Found: C, 26.87; H, 3.27; Br, 14.57; Cl, 7.01; I, 23.60. HRMS-ESI (positive mode): calcd for $[M + Na]^+$ 554.8895, 556.8875, 558.8845. Found: 554.8901, 556.8879, 558.8841 (3.8:4.9:1)

6-Bromo-1,6-dideoxy-1-iodo-β-D-fructofuranosyl 4-bromo-4-deoxy-α-Dglucopyranoside (27). Deacetylation of 23 (0.29 g, 0.37 mmol) as above gave **27** as a colourless syrup (0.17 g, 80%), $[\alpha]_D$ +41.7° (c 1.53, H₂O); ¹H NMR (D₂O) δ 4.13 (dd, 1H, $J_{1,2}$ 3.8 $J_{2,3}$ 9.6 Hz, H-2), 5.01 (d, 1H, $J_{3',4'}$ 8.4 Hz, H-3') and 6.02 (d, 1H, $J_{1,2}$ 3.8 Hz, H-1). ¹³C NMR (D₂O) δ 105.4 (C-2'), 95.7 (C-1), 83.5 (C-5'), 81.2, 79.2 (C-3', 4'), 75.6, 75.2, 74.4 (C-2, 3, 5), 64.0 (C-6), 53.1 (C-4), 35.6 (C-6') and 8.6 (C-1').

Anal. Calcd for C₁₂H₁₉Br₂IO₈: C, 24.94; H, 3.31; Br, 27.65; I, 21.96. Found: C, 25.35; H, 3.23; Br, 26.61; I, 21.70. HRMS-ESI (positive mode): calcd for [M + Na] + 598.8389, 600.8369, 602.8349. Found: 598.8386, 600.8389, 602.8352 (1:2:1).

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REFERENCES

- 1. Synthesis and Reactions of Halodeoxy Sucrose. Part 2. For Part 1, see ref. 13.
- 2. In line with commonly used format, primed numbers are used for the furanosyl ring, and unprimed numbers are used for pyranosyl ring.
- Hough. L.; Phadnis, S.P. Enhancement in the Sweetness of Sucrose. Nature 1976, 263, 800–801.
- Shallenberger, R.S.; Acree, T.E. Molecular Theory of Sweet Taste. Nature 1967, 216, 480–482.
- 5. Kier, L.B. A Molecular Theory of Sweet Taste. J. Pharm. Sci. 1972, 61, 1394–1397.
- Hough, L.; Khan, R. Intensification of Sweetness. Trends Biochem. Sci. 1978, (March), 61–63; van der Heidjen, A.; van der Wel, H.; Peer, H.G. Structure-activity Relationships in Sweeteners. II. Saccharins, Acesulfames, Chlorosugars, Tryptophans and Ureas. Chem. Senses 1985, 10(1), 73–88.
- 7. Hough, L. Sucrose, Sweetness and Sucralose. Int. Sugar J. 1989, 91(1082), 23–37.
- 8. Hooft, R.W.W.; Kanters J.A.; Kroon, J. Molecular Mechanics and Dynamics. Calculations on Sucrose and Some Derived Artificial Sweeteners. In *Sweet Taste Chemore-ception*; Mathlouthi, M., Kanters, J.A., Birch, G.G., Eds.; Elsevier Applied Science: London, 1993; 11–19.
- 9. Lichtenthaler F.W.; Immel, S. Sucrose, Sucralose and Fructose: Correlation Between Hydrophobicity Potential Profiles and AH-B-X Assignments. In *Sweet Taste Chemoreception*; Mathlouthi, M., Kanters J.A., Birch, G.G., Eds.; Elsevier Applied Science: London, 1993; 21–53.
- 10. Hough L.; Khan, R. The Relationship Between Sweetness and the Molecular Structure of Sucrose and its Derivatives. In *Sweet-taste Chemoreception*; Mathlouthi, M., Kanters, J.A., Birch, G.G., Eds.; Elsevier Applied Science: London, 1993; 91–102.
- Mathlouthi, M.; Bressan, C.; Potmann, M.O.; Serghat, S. Role of Water Structure in Sweet-Taste Chemoreception. In *Sweet Taste Chemoreception*; Mathlouthi, M., Kanters J.A., Birch, G.G., Eds.; Elsevier Applied Science: London, 1993; 141–174.
- 12. Tsuami, T.; Hough, L.; Tsuboi, M.; Machinami, T. Molecular Mechanisms of Sweet Taste. V. Sucralose and its Derivatives. J. Carbohydr. Chem. **1994**, *13*(8), 1079–1092.
- 13. Lee, C.K.; Kang H.C.; Linden, A. Synthesis, Structure and Sweetness of 1,4,6-Trichloro-1,4,6-Trideoxy-β-D-Tagatofuranosyl 4-Chloro-4-Deoxy-α-D-Galactopyranoside. J. Carbohydr. Chem. **1999**, *18*(2), 241–253.
- Hough, L.; Phadnis, S.P.; Khan R.; Jenner, M.R. Sweeteners. *Br. Pat.* 1 543 167, 1979; Chloroderivatives of Sucrose. *Br. Pat.* 1 543 168, 1979.
- 15. McKeown, G.G.; Serenius R.S.; Hayward, L.D. Selective Substitution in Sucrose. I. The Synthesis of 1',4,6'-Tri-O-Methyl Sucrose and C₄ to C₆ Acetyl Migration in Sucrose. Can. J. Chem. 1957, 35, 28–36; Bredereck, H.; Zinner, H.; Wagner, A.; Faber, G.; Grenier W.; Huber, W. Darstellung und Konstitution Zweier Pentaacetyl-saccharosen. Chem. Ber. 1958, 91, 2824-29; Fairclough, P.H.; Hough L.; Richardson, A.C. Sucrochemistry: Part XVI. Derivatives of β-D-Fructofuranosyl α-D-Galactopyranoside. Carbohydr. Res. 1975, 40, 285–298.
- Simiand C.; Driguez, H. Synthesis of Sucrose Analogs Modified at Position 4. J. Carbohydr. Chem. 1995, 14(7), 977–83; Blanc-Muesser, M.; Driguez, H. A Convenient Method for the S-Glycosidic Bond Formation. Synthesis of p-Iodophenyl 4'-Thiomaltotrioside and its 2",3"-Unsaturated Analog. J. Chem. Soc., Perkin Trans. 1 1988,





- 1(12), 3345–3351; Albert, R.; Dax, K.; Link, R.W.; Stütz, A.E. Carbohydrate Triflates: Reaction with Nitrite, Leading Directly to *epi*-Hydroxy Compounds. Carbohydr. Res. **1983**, *118*, C5–C6.
- 17. Kakinuma, H.; Yuasa H.; Hashimoto, H. Synthesis of 1',6'-Disubstituted Sucroses and Their Behaviour as Glucosyl Donors for a Microbial α-Glucosyltransferase. Carbohydr. Res. **1996**, *284*, 61–72.
- Shallenberger, R.S. Chemical Clues to the Perception of Sweetness. In Sensory Properties of Foods; Birch, G.G., Brennan J.G., Parker, K.J. Eds.; Applied Science: London, 1977; 91–110; Shallenberger R.S.; Lindley, M.G. A Lipophilic-hydrophlic Attribute and Component in the Stereochemistry of Sweetness. Food Chem. 1977, 2, 145–154.
- 19. Daniel, J.D. Sweeteners: Theory and Design. In *Frontiers in Carbohydrate Research 1. Food Applications*, Millane, R.P., BeMiller J.N., Chandrasekaran, R., Eds.; Elsevier Applied Science: London, 1989; 34–65.

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