v-Triazolines. Part 42.1 Study on the reactivity of 4,5-dihydro-1-(6-methyl-2-oxo-2H-pyran-4-yl)-5-morpholino-v-triazoles. Synthetic approach to pyrano[4,3-b]pyrrol-4(1H)-ones

Emanuela Erba, Donato Pocar and Pasqualina Trimarco*

Istituto di Chimica Organica, Facoltà di Farmacia e Centro Interuniversitario di Ricerca sulle Reazioni Pericicliche e Sintesi di Sistemi Etero e Carbociclici, Università di Milano, Via Venezian 21, I-20133 Milano, Italy. E-mail: trimarco@mailserver.unimi.it

Received (in Cambridge, UK) 23rd October 2000, Accepted 29th May 2001 First published as an Advance Article on the web 28th June 2001

Pyrolysis of 4-aryl-5-morpholino-4,5-dihydrotriazoles **3** affords two products: pyrano[4,3-b]pyrrol-4(1H)-ones **4** and arylacetamidines 5. The reaction mechanism of this transformation is discussed and reaction conditions optimized to enhance the formation of pyrrole-fused pyran-2-one derivatives 4. 2-Aminoaziridines are considered to be key intermediates in this transformation.

2H-Pyran-2-ones are of interest as synthons for organic synthesis, particularly since many derivatives, isolated from natural sources, exhibit remarkable biological properties.² Renewed interest comes from the discovery of a new type of non-peptidic HIV-protease inhibitor containing this nucleus.³

In 2H-pyranones the C-3 position is nucleophilic, presenting the characteristic reactivity of enols, and electrophiles react at C-3 with conservation of the pyrone ring.⁴

Our research group has been studying transformations and uses of 5-amino-4,5-dihydro-v-triazoles, mainly as precursors of substituted amidines, synthons from which N-heterocycles can be derived.⁵ It is well known that 5-amino-4,5-dihydro-vtriazoles readily aromatize to triazoles^{6a} by heat or under the action of acid or base by elimination of a molecule of amine. The 5-amino-4,5-dihydro-v-triazoles give rise to amidines when the R group on N-1 is highly electron withdrawing and the 5amino group is tertiary.^{6b} In a previous paper⁷ we pointed out that thermolysis can induce 4,5-dihydro-v-triazole ring cleavage (Scheme 1). The zwitterionic intermediate, produced by ringchain tautomerism of the triazoline ring, undergoes two different reactions: direct formation of the amidine by N2 loss and simultaneous hydride transfer, or alternatively ring contraction into the 2-aminoaziridine, associated with N2 loss. We also observed the formation of labile diaminoethylenic derivatives, whose formation has been rationalized by invoking the presence of an aziridine intermediate. Cleavage of the aziridine bond between N and the C atom bearing the amino residue, through its iminium form, explained the ethylenic intermediates.

These results prompted us to synthesize 5-amino-4,5-dihydro-v-triazoles bearing a 2H-pyran-2-one group at N-1, and to explore the possibility of linking both the reactivity of 2-aminoaziridines and the nucleophilic features of C-3 in the 2*H*-pyran-2-one nucleus.

Suitable conditions were sought to achieve transformation of the intermediate aminoaziridine pyrrole-fused pyran-2-one derivatives. Little chemistry of this ring system has been investigated⁸ although some substituted derivatives have been prepared from pyrrol-2-ylacetic acid derivatives⁹ by a multistep sequence. A synthetic pathway involving 5-amino-4,5dihydro-v-triazoles has never been described.

DOI: 10.1039/b008531f

Scheme 1

Results and discussion

4,5-Dihydro-v-triazoles 3a-d,f,g were easily obtained from cycloaddition of azide 1 and appropriate enamines 2a-d,f,g in benzene while **3e** was prepared by one-pot procedure¹⁰ from propionaldehyde, 4-azido-6-methyl-2H-pyran-2-one 1 and morpholine without isolation of the intermediate enamine 2e (Scheme 2). Satisfactory yields were obtained and the structures of products, including their *trans* configuration,¹¹ were established from ¹H NMR data (see Experimental section).

Compounds 3a-g were thermolyzed by heating them in boiling toluene until complete consumption of the starting

J. Chem. Soc., Perkin Trans. 1, 2001, 1723-1728 1723

This journal is © The Royal Society of Chemistry 2001





d: R = 4-CIC₆H₄ **e**: R = Me^a **f**: R = 4-BrC₆H₄

Scheme 2 ^{*a*} 2e not isolated, but obtained *in situ*: see Experimental section, preparation of compound 3e.

compound (about 1–2 hours). Starting from **3a–d,f,g** a mixture of pyrrole-fused pyran-2-ones **4a–d,f,g** and amidines **5a–d,f,g**, respectively, was formed. Dihydro-v-triazole **3e** gave only amidine **5e** (Scheme 3).



Scheme 3 ^{*a*} 4e was obtained in toluene but only in the presence of BF_3 ·Et₂O (Tables 1 and 2).

Spectroscopic data established the structure of tertiary amidines **5a**–g and of 2-aryl-6-methylpyrano[4,3-*b*]pyrol-4(1*H*)-ones **4a–d,f,g**. Assignment of protons and carbons in ¹H and ¹³C spectra of amidines **5a–g** was made on the basis of HETCOR and COLOC experiments, and data related to the pyran-2-one nucleus were in agreement with the literature.¹² All amidines **5** (in CDCl₃) show a typical pattern for the signal of the C-3 proton in the range δ 5.15–5.29 and for the C-5 proton in the range δ 5.58–5.66. A NOESY experiment, performed for amidine **5a**, demonstrated correlation between the singlet at δ 3.8 (CH₂Ph) and the singlet at δ 5.29 (pyran-2-one 3-H), supporting the assigned (*E*) configuration of the amidine double bond.

All pyrano[4,3-*b*]pyrrol-4-ones **4a–d**,**f**,**g** show typical proton signals (in DMSO-d₆) due to the C-7 and C-3 protons at $\delta \approx 6.5$ and $\delta 6.87-7.07$, respectively. The complete assignment of the peaks in the ¹³C NMR spectrum was facilitated by 2D NMR spectroscopy performed on the pyrano[4,3-*b*]pyrrol-4(1*H*)-one **4a**: HETCOR experiments allowed attribution of the CH-7 signals ($\delta 6.48$ and δ_c 96) as well as the CH-3 signal ($\delta 7.01$ and $\delta_c \approx 103$), quaternary carbons being assigned from the results of a COLOC experiment. For all compounds 4 proton/carbon resonances are reported in the Experimental section.

Conclusive structural assignment of compounds $4a-d_1f_2$ was obtained by NOESY experiments (DMSO-d₆) on 4a, which displayed clear correlations between H-7 and the CH₃ linked to C-6. A positive Overhauser effect confirmed also the spatial proximity of the NH group to H-7 as well as to the *ortho* hydrogens of the C-2 phenyl group.

The formation of amidines **5** is expected ^{6b} according to the general rearrangement path of 5-amino-4,5-dihydrotriazoles as indicated in Scheme 1. The production of fused heterocycles **4** can be rationalized as occurring *via* aminoaziridine **C** (Scheme 4). The previously observed⁷ bond cleavage between



the N atom and the C atom linked to the morpholine residue forms **D**, where nucleophilic attack of C-3 of the pyran-2-one on the iminium carbon occurs to give a dihydropyrano[4,3-*b*]-pyrrol-4-one intermediate, which, by amine loss, turns into the corresponding pyrano[4,3-*b*]pyrrol-4(1*H*)-one **4**.

The reaction was studied in more detail, both to confirm that bicycles 4 do not arise from the amidines 5, and to determine those factors that would enhance the yield of the pyrrole-fused pyran-2-ones 4. In a first experiment amidine 5a was refluxed in toluene for 60 min, under the same conditions adopted for the decomposition of 4,5-dihydro- ν -triazole 3a. The ¹H NMR spectrum of the crude mixture showed only the signals of amidine 5a, thus confirming the independence of paths leading to 4 and 5, respectively.

In order to increase the yield of compounds 4, 4,5-dihydro-*v*-triazoles **3a–g** were treated under two different conditions: i) refluxing in propan-1-ol until disappearance of starting material was complete; ii) boiling in dry toluene with an equimolar amount of $BF_3 \cdot Et_2O$ for the same time. Propan-1-ol was selected on account of its suitable boiling point. It was

 Table 1
 Ratios of 4 and 5 estimated by ¹H NMR spectroscopy

Compoun 3	d Toluene	Pr ⁿ OH	Toluene, BF₃∙Et₂O
3a	4a : 5 a (33 : 67)	4a : 5a (18 : 82)	4a : 5a (70 : 30)
3b	4b : 5b (38 : 62)	4b : 5b (34 : 66)	4b : 5b (88 : 12)
3c	4c : 5c (37 : 63)	4c : 5c (66 : 34)	4c : 5c (82 : 18)
3d	4d : 5d (44 : 56)	4d : 5d (19 : 81)	4d : 5d (94 : 6)
3e	4e : 5e (0 : 100)	4e : 5e (0 : 100)	4e : 5e (25 : 75)
3f	4f : 5f (48 : 52)	4f : 5f (28 : 72)	4f : 5f (83 : 17)
3g	4g : 5g (32 : 68)	4g : 5g (22 : 78)	4g : 5g (63 : 37)

expected that this polar and protic solvent might influence the equilibrium between the conformers A and B previously seen in Scheme 1. In this reaction setting we were expecting the dipolar form A to prevail and, from N₂ loss, the amount of amidines 5 to increase.

Dry toluene and $BF_3 \cdot Et_2O$ were used in attempts to stabilize the iminium intermediate **D** in Scheme 4. It is well known that aziridines with electron-withdrawing groups,¹³ such as alkylsulfonyl, acyl, alkoxycarbonyl, and in our case the unsaturated lactone pyran-2-one linked to the N atom, conjugatively stabilize the negative charge that develops on the nitrogen in the ring-opening transition state. In this context (see Scheme 4) BF_3 could bond to the anionic nitrogen, in iminium form **D**, or conjugatively to the enolate oxygen atom and hence promote the formation of **4**. Table 1 collects the results obtained under three different conditions. The ratio **4** : **5** in the product mixture was determined by ¹H NMR analysis.

The data of Table 1 confirm the correctness of the above choices and give support to the formulated mechanistic hypothesis, that the achievement of heterocycles 4 and the amidines 5 comes from two competitive pathways. An effective increase in the yield of amidines 5 was seen when the reaction was performed propan-1-ol, while the addition of the Lewis acid catalyst brought about an increase of the yield of pyrrole-fused pyran-2-one derivatives 4, suggesting that the participation of polar solvents and/or complexing catalyst can act on the equilibrium between A and B, favouring the hydride transfer or the formation of the aziridine intermediate.

In conclusion, variations in the reaction medium controlled the transformation of 4,5-dihydro- ν -triazoles **3** and facilitated our aim of linking the reactivity of 2-aminoaziridines and the nucleophilic reactivity of C-3 in the 2*H*-pyranone nucleus.

Experimental

Mps were determined using a Büchi 510 (capillary) or an Electrothermal 9100 apparatus and are uncorrected. IR spectra were measured using a JASCO IR Report 100 instrument. ¹H and ¹³C NMR spectra (tetramethylsilane as internal standard) were recorded with EM Varian Gemini 200, Bruker AC 200 and Bruker Avance 300 Spectrometers. *J*-Values are given in Hz for solutions in CDCl₃ or DMSO-*d*₆. Mass spectral data were obtained on a Varian MAT 1H COS 50 instrument using electron-impact ionization techniques at 70 eV. Column chromatography was performed on Kieselgel 60 (Merck), 0.063–0.200 mm, and elution with cyclohexane–ethyl acetate (ratios indicated in Experimental section).

Materials

Azide 1^{14} and enamines 2a-c, ¹⁵ 2f, ¹⁵ 2d, ¹⁶ 2g ¹⁷ have already been described.

4,5-Dihydro-1-(6-methyl-2-oxo-2*H*-pyran-4-yl)-5-morpholino-1,2,3-triazoles 3a-d,f,g

Typical procedure. Azide **1** (3.0 g, 20 mmol) was dissolved in benzene (20 ml) and an equimolar amount of an enamine **2a**, **b**,

d, **f**, **g**, dissolved in benzene (20 ml), was added under stirring at room temperature. Enamine **2c** solution in benzene was added dropwise to the stirred azide solution at 0-5 °C. As soon as the starting azide had disappeared (TLC, cyclohexane–ethyl acetate 3 : 7), the solvent was removed under reduced pressure and the crude product was crystallized from an appropriate solvent.

4-(4,5-Dihydro-5-morpholino-4-phenyl-1*H***-1,2,3-triazol-1-yl)-6-methyl-2***H***-pyran-2-one 3a.** Reaction time 2 h; yield 6.5 g (96%); white plates from Prⁱ₂O; mp 85 °C (decomp.); IR (Nujol) ν_{max} 1700 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 2.30 (3H, s, Me), 2.39–2.47 (4H, m, CH₂NCH₂), 3.64–3.72 (4H, m, CH₂OCH₂), 4.62 (1H, d, *J* 3.4, 5-H), 5.61 (1H, d, *J* 3.4, 4-H), 5.72 (1H, s, 3-H pyranone), 6.78 (1H, s, 5-H pyranone), 6.95–7.42 (5H, m, ArH) (Calc. for C₁₈H₂₀N₄O₃: C, 63.52; H, 5.92; N, 16.46. Found: C, 63.69; H, 6.03; N, 16.18%).

4-(4,5-Dihydro-5-morpholino-4-*p*-tolyl-4-1*H*-1,2,3-triazol-1-yl)-6-methyl-2*H*-pyran-2-one 3b. Reaction time 2 h; yield 5.7 g (81%); white plates from Pr_2^iO ; mp 108 °C (decomp.); IR (Nujol) v_{max} 1700 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 2.29 (3H, s, Me), 2.33 (3H, s, *p*-tolyl), 2.35–2.44 (4H, m, CH₂NCH₂), 3.63–3.69 (4H, m, CH₂OCH₂), 4.58 (1H, d, *J* 3.4, 5-H), 5.57 (1H, d, *J* 3.4, 4-H), 5.71 (1H, s, 3-H pyranone), 6.78 (1H, s, 5-H pyranone), 6.87 and 7.17 (4H, dd, AB system, *J* 8.0, ArH (Calc. for C₁₉H₂₂N₄O₃: C, 64.39; H, 6.26; N, 15.81. Found: C, 64.6; H, 6.31; N, 15.55%).

4-[4,5-Dihydro-4-(4-Methoxyphenyl)-5-morpholino-1*H***-1,2,3-triazol-1-yl]-6-methyl-2***H***-pyran-2-one 3c.** Reaction time 2 h; yield 4.7 g (64%); white plates from $Pr_{2}^{i}O$; mp 65–66 °C (decomp.); IR (Nujol) v_{max} 1700 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 2.30 (3H, s, Me), 2.38–2.47 (4H, m, CH₂NCH₂), 3.63–3.70 (4H, m, CH₂OCH₂), 3.80 (3H, s, OMe), 4.58 (1H, d, *J* 3.4, 5-H), 5.56 (1H, d, *J* 3.4, 4-H), 5.71 (1H, s, 3-H pyranone), 6.79 (1H, s, 5-H pyranone), 6.88–6.98 (4H, m, ArH) (Calc. for C₁₉H₂₂N₄O₄: C, 61.61; H, 5.99; N, 15.13. Found: C, 61.84; H, 6.12; N, 14.97%).

4-[4-(4-Chlorophenyl)-4,5-dihydro-5-morpholino-1*H*-1,2,3-

triazol-1-yl]-6-methyl-2H-pyran-2-one 3d. Reaction time 24 h; yield 5.5 g (73%); white plates from benzene; mp 107 °C (decomp.); IR (Nujol) ν_{max} 1700 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 2.30 (3H, s, Me), 2.38–2.48 (4H, m, CH₂NCH₂), 3.64–3.75 (4H, m, CH₂OCH₂), 4.58 (1H, d, J 3.5, 5-H), 5.58 (1H, d, J 3.5, 4-H), 5.71 (1H, s, 3-H pyranone), 6.75 (1H, s, 5-H pyranone), 6.94 and 7.36 (4H, dd, AB system, J 8.5, ArH) (Calc. for C₁₈H₁₉ClN₄O₃: C, 57.68; H, 5.11; N, 14.95. Found: C, 57.89; H, 5.18; N, 14.68%).

4-[4-(4-Bromophenyl)-4,5-dihydro-5-morpholino-1H-1,2,3-

triazol-1-yl]-6-methyl-2*H***-pyran-2-one 3f.** Reaction time 2 h; yield 5.4 g (64%); cream plates from ethanol; mp 115 °C (decomp.); IR (Nujol) v_{max} 1700 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 2.32 (3H, s, Me), 2.36–2.47 (4H, m, CH₂NCH₂), 3.64–3.75 (4H, m, CH₂OCH₂), 4.60 (1H, d, *J* 3.6, 5-H), 5.58 (1H, d, *J* 3.6, 4-H), 5.73 (1H, s, 3-H pyranone), 6.79 (1H, s, 5-H pyranone), 6.89 and 7.53 (4H, dd, AB system, *J* 8.4, ArH) (Calc. for C₁₈H₁₉BrN₄O₃: C, 51.56; H, 4.57; N, 13.36. Found: C, 51.63; H, 4.81; N, 13.07%).

4-[4-(4-Fluorophenyl)-4,5-dihydro-5-morpholino-1*H*-1,2,3-

triazol-1-yl]-6-methyl-2H-pyran-2-one 3g. Reaction time 3 h; yield 4.1 g (57%); cream plates from ethanol; mp 118 °C (decomp.); IR (Nujol) ν_{max} 1690 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 2.33 (3H, s, Me), 2.40–2.47 (4H, m, CH₂NCH₂), 3.68–3.75 (4H, m, CH₂OCH₂), 4.60 (1H, d, J 3.6, 5-H), 5.60 (1H, d, J 3.6, 4-H), 5.74 (1H, s, 3-H pyranone), 6.80 (1H, s, 5-H pyranone), 6.90–7.16 (4H, m, ArH) (Calc. for

 Table 2
 Preparation of pyranol[4,3-b]pyrrol-4(1H)-ones 4 or amidines 5 from 5-amino-4,5-dihydro-v-triazoles 3

E	ntry	R	Method ^a	Reaction time (t/min)	<i>T</i> /°C	Yield of isolated products (%)		
38	a	Ph	A	60	111	4a (21) 5a (55)		
38	a	Ph	В	120	97	4a (13) 5a (70)		
38	a	Ph	С	60	111	4a (58) 5a (18)		
31	b	4-MeC ₆ H ₄	А	90	111	4b (24) 5b (46)		
31	b	4-MeC ₆ H ₄	В	90	97	4b (22) 5b (56)		
31	b	4-MeC ₆ H ₄	С	90	111	4b (77) 5b (7)		
30	c	4-MeOC ₆ H ₄	А	90	111	4c (25) 5c (48)		
30	c	4-MeOC ₆ H ₄	В	120	97	4c (53) 5c (20)		
30	c	4-MeOC ₆ H ₄	С	90	111	4c (70) 5c (13)		
30	d	4-ClC ₆ H ₄	A	100	111	4d (36) 5d (47)		
30	d	4-ClC ₆ H ₄	В	80	97	4d (11) 5d (68)		
30	d	4-ClC ₆ H ₄	С	120	111	4d (88) 5d (2)		
36	e	Me	А	120	111	5e (86)		
36	e	Me	В	120	97	5e (88)		
36	e	Me	С	120	111	4e (16) 5e (58)		
31	f	4-BrC ₆ H ₄	А	120	111	4f (36) 5f (43)		
31	f	$4-BrC_6H_4$	В	120	97	4f (15) 5f (60)		
31	f	$4-BrC_6H_4$	С	120	111	4f (65) 5f (10)		
39	g	$4-FC_6H_4$	А	120	111	4g (22) 5g (56)		
39	g	$4 - FC_6H_4$	В	120	97	4g (13) 5g (64)		
3	g	$4-FC_6H_4$	С	120	111	4g (45) 5g (30)		
^{<i>a</i>} Solvents: A, toluene; B, propan-1-ol; C, toluene, BF ₃ ·Et ₂ O.								

C₁₈H₁₉FN₄O₃: C, 60.33; H, 5.34; N, 15.63. Found: C, 60.46; H, 5.41; N, 15.35%).

Preparation of 4-(4,5-dihydro-4-methyl-5-morpholino-1*H*-1,2,3-triazol-1-yl)-6-methyl-2*H*-pyran-2-one 3e

A benzene solution (30 ml) of morpholine (2.61 g, 30 mmol) was added dropwise to a stirred solution of azide 1 (4.5 g, 30 mmol) and propanal (1.74 g, 30 mmol) in benzene (30 ml) at rt until the starting azide had disappeared (*ca.* 20 h) (TLC cyclohexane–EtOAc, 1 : 9). The solution was dried with Na₂SO₄, filtered, and the filtrate was evaporated under reduced pressure. The residue was crystallized from Pr¹₂O to afford pure **3e**. (5.6 g, 67%), as white needles; mp 113 °C (decomp.); IR (Nujol) v_{max} 1700 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 1.28 (3H, d, *J* 7.2, 4-Me triazoline), 2.28 (3H, s, Me), 2.30–2.38 (4H, m, CH₂NCH₂), 3.61–3.68 (4H, m, CH₂OCH₂), 4.31 (1H, d, *J* 3.1, 5-H), 4.58 (1H, dq, *J* 3.1 and 7.2, 4-H), 5.70 (1H, s, 3-H pyranone), 6.72 (1H, s, 5-H pyranone) (Calc. for C₁₃H₁₈N₄O₃: C, 56.10; H, 6.52; N, 20.13. Found: C, 56.22; H, 6.64; N, 20.01%).

Thermal behaviour of dihydrotriazoles 3a-g. Method A

Typical procedure for the synthesis of compounds 4a-d,f,g and 5a-g. Dihydrotriazoles 3a-g (15 mmol) dissolved in toluene (70 ml) were heated under reflux for times indicated in Table 2, progress of the reaction being followed by TLC. After disappearance of the starting material the solvent was removed in vacuo and a small amount of the residue was dissolved in DMSO- d_6 and the ratio of 4 to 5 was determined by ¹H NMR analysis (see Table 1). The crude residue was taken up with CH₂Cl₂ and slowly and partially deposited a compound 4ad,f,g as a crystalline product, which was filtered off and recrystallized as indicated below. The left over mixture was chromatographed (cyclohexane-ethyl acetate 3:7) to give a first fraction containing a pyrano[4,3-b]pyrrol-4-one 4a-d,f,g and a second fraction containing an amidine 5a-d,f,g. Dihydrotriazole 3e afforded only amidine 5e, isolated by crystallization from Prⁱ₂O. Pyrrole-fused pyran-2-one 4e was isolated only from reaction with BF₃·Et₂O (see below). Reaction times of dihydrotriazoles 3 and yields of isolated products 4 and 5 are collected in Table 2.

6-Methyl-2-phenylpyrano[4,3-b]pyrrol-4(1H)-one 4a. White

plates from $Pr_{2}^{i}O$; mp 223 °C; IR (Nujol) ν_{max} 3250 (NH), 1675 (C=O) cm⁻¹; ¹H NMR (300 MHz; DMSO-*d*₆) δ 2.26 (3H, s, Me), 6.48 (1H, s, 7-H), 7.01 (1H, s, 3-H), 7.19–7.82 (5H, m, ArH), 12.2 (1H, br s, NH exchangeable); ¹³C NMR (75 MHz; DMSO-*d*₆) δ 20.2 (Me), 96.0 (C-7), 103.2 (C-3), 108.3 (C-3a), 125.3, 128.1, 129.7 (ArCH), 131.9 (ArC), 135.98 (C-2), 141.4 (C-7a), 156.5 (C-6), 160.4 (C=O); MS *m*/*z* 225 (M⁺, 100%), 197 (35) (Calc. for C₁₄H₁₁NO₂: C, 74.65; H, 4.92; N, 6.22. Found: C, 74.39; H, 5.01; N, 6.12%).

6-*Methyl*-4-(1-morpholino-2-phenylethylideneamino)-2*H*pyran-2-one **5a**. White plates from Pr¹₂O; mp 135 °C; IR (Nujol) v_{max} 1705 (C=O) cm⁻¹; ¹H NMR (300 MHz; CDCl₃) δ 2.15 (3H, s, Me), 3.45–3.47 (8H, 2m, morpholino), 3.80 (2H, s, CH₂), 5.29 (1H, d, *J* 1.7, 3-H), 5.66 (1H, d, *J* 1.7, 5-H), 7.10–7.38 (5H, m, ArH); ¹³C NMR (75 MHz; CDCl₃) δ 20.3 (Me), 34.7 (CH₂), 45.8 (CH₂NCH₂), 66.8 (CH₂OCH₂), 96.4 (C-3), 105.0 (C-5), 127.6, 128.0, 129.6 (ArCH), 135.3 (ArC), 156.8 (N-C=N), 161.8 (C-6), 165.1 (C-4), 165.4 (C=O) (Calc. for C₁₈H₂₀N₂O₃: C, 69.21; H, 6.45; N, 8.97. Found: C, 69.33; H, 6.53; N, 8.88%).

6-*Methyl*-2-*p*-tolylpyrano[4,3-b]pyrrol-4(1H)-one **4b**. White plates from Pr¹₂O; mp 268 °C; IR (Nujol) v_{max} 3150 (NH), 1680 (C=O) cm⁻¹; ¹H NMR (200 MHz; DMSO-d₆) δ 2.27 (3H, s, Me), 2.32 (3H, s, *p*-tolyl), 6.48 (1H, s, 7-H), 6.93 (1H, s, 3-H), 7.24 and 7.66 (4H, dd, AB system, *J* 8.4, ArH), 12.10 (1H, br s, NH exchangeable); ¹³C NMR (50 MHz; DMSO-d₆) δ 19.6 (Me), 20.9 (Me), 95.4 (C-7), 101.8 (C-3), 107.7 (C-3a), 124.6, 129.7 (ArCH), 128.6 (ArC), 135.5 (C-2), 136.9 (ArC), 140.6 (C-7a), 155.6 (C-6), 159.8 (C=O) (Calc. for C₁₅H₁₃NO₂: C, 75.30; H, 5.48; N, 5.85. Found: C, 75.06; H, 5.57; N, 5.63%).

6-Methyl-4-(1-morpholino-2-p-tolylethylideneamino)-2Hpyran-2-one **5b**. White plates from Pr¹₂O; mp 191 °C; IR (Nujol) v_{max} 1700 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 2.15 (3H, s, Me), 2.32 (3H, s, p-tolyl), 3.44–3.57 (8H, m, morpholine), 3.74 (2H, s, CH₂), 5.28 (1H, s, 3-H), 5.65 (1H, s, 5-H), 6.99 and 7.12 (4H, dd, AB system, J 8.4, ArH); ¹³C (50 MHz; CDCl₃) δ 19.9 (Me), 21.0 (Me), 33.9 (CH₂), 45.4 (CH₂NCH₂), 66.5 (CH₂OCH₂), 96.2 (C-3), 104.7 (C-5), 127.5, 129.9 (ArCH), 131.8 (ArC), 136.8 (ArC), 156.6 (N-C=N), 161.4 (C-6), 164.8 (C-4), 165.0 (C=O) (Calc. for C₁₉H₂₂N₂O₃: C, 69.92; H, 6.79; N, 8.58. Found: C, 70.20; H, 6.72; N, 8.45%).

2-(4-Methoxyphenyl)-6-methylpyrano[4,3-b]pyrrol-4(1H)one 4c. Cream plates from CH₂Cl₂; mp 242 °C; IR (Nujol) v_{max} 3230 (NH), 1670 (C=O) cm⁻¹; ¹H NMR (200 MHz; DMSO- d_6) δ 2.28 (3H, s, Me), 3.80 (3H, s, OMe), 6.48 (1H, s, 7-H), 6.87 (1H, s, 3-H), 7.02 and 7.71 (4H, dd, AB system, *J* 8.7, ArH), 12.05 (1H, br s, NH exchangeable); ¹³C NMR (50 MHz; DMSO-*d*₆) δ 19.6 (Me), 55.4 (OMe), 95.4 (C-7), 101.1 (C-3), 107.7 (C-3a), 114.6, 126.1 (ArCH), 124.2 (ArC), 135.5 (C-2), 140.49 (C-7a), 155.5 (C-6), 158.9 (OMe), 159.9 (C=O) (Calc. for C₁₅H₁₃NO₃: C, 70.58; H, 5.13; N, 5.49. Found: C, 70.37; H, 4.96; N, 5.35%).

4-[2-(4-Methoxyphenyl)-1-morpholinoethylideneamino]-6-

methyl-2H-pyran-2-one 5c. Orange plates from $Pr_{2}^{i}O$; mp 106 °C; IR (Nujol) v_{max} 1700 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 2.16 (3H, s, Me), 3.40–3.50 (4H, m, CH₂NCH₂), 3.50–3.60 (4H, m, CH₂OCH₂), 3.72 (2H, s, CH₂), 3.80 (3H, s, OMe), 5.28 (1H, s, 3-H), 5.65 (1H, s, 5-H), 6.85 and 7.03 (4H, dd, AB system, J 8.7, ArH) (Calc. for C₁₉H₂₂N₂O₄: C, 66.65; H, 6.48; N, 8.18. Found: C, 66.88; H, 6.71; N, 6.22%).

2-(4-Chlorophenyl)-6-methylpyrano[4,3-b]pyrrol-4(1H)-one 4d. White plates from. CH₂Cl₂; mp 291 °C; IR (Nujol) v_{max} 3230 (NH), 1670 (C=O) cm⁻¹; ¹H NMR (200 MHz; DMSO-d₆) δ 2.28 (3H, s, Me), 6.50 (1H, s, 7-H), 7.07 (1H, s, 3-H), 7.50 and 7.80 (4H, dd, AB system, J 8.5, ArH), 12.22 (1H, br s, NH exchangeable); ¹³C NMR (50 MHz; DMSO-d₆) δ 19.6 (Me), 95.3 (C-7), 103.3 (C-3), 107.8 (C-3a), 126.3, 129.1 (ArCH), 130.3 and 131.9 (2 × ArC), 134.2 (C-2), 141.0 (C-7a), 156.1 (C-6), 159.7 (C=O) (Calc. for C₁₄H₁₀CINO₂: C, 64.75; H, 3.88; N, 5.39. Found: C, 64.54; H, 3.83; N, 5.31%).

4-[2-(4-Chlorophenyl)-1-morpholinoethylideneamino]-6methyl-2H-pyran-2-one 5d. Pale yellow plates from MeOH; mp 158 °C; IR (Nujol) v_{max} 1690 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 2.16 (3H, s, Me), 3.38–3.50 (4H, m, CH₂NCH₂), 3.50– 3.63 (4H, m, CH₂OCH₂), 3.76 (2H, s, CH₂), 5.24 (1H, s, 3-H), 5.64 (1H, s, 5-H), 7.06 and 7.31 (4H, dd, AB system, J 8.4, ArH) (Calc. for C₁₈H₁₉ClN₂O₃: C, 62.34; H, 5.52; N, 8.08. Found: C, 62.08; H, 5.43; N, 7.95%).

6-Methyl-4-(1-morpholinopropylideneamino)-2H-pyran-2-

one 5e. White plates from $Pr_{2}^{i}O$; mp 85–86 °C; IR (Nujol) v_{max} 1700 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 1.11 (3H, t, J 7.6, Me), 2.19 (3H, s, pyranone-Me), 2.36 (2H, dd, J 7.6, CH₂), 3.45–3.51 and 3.70–3.76 (8H, 2m, morpholine), 5.24 (1H, d, J 1.7, 3-H), 5.62 (1H, d, J 1.7, 5-H); ¹³C NMR (50 MHz; DMSO- d_{6}) δ 12.4 (Me linked to CH₂), 19.6 (6-Me), 21.7 (CH₂), 45.2 (CH₂NCH₂), 66.2 (CH₂OCH₂), 95.0 (C-3), 104.5 (C-5), 160.6 (N-C=N), 161.5 (C-6), 163.7 (C-4), 165.39 (C=O) (Calc. for C₁₃H₁₈N₂O₃: C, 62.38; H, 7.25; N, 11.19. Found: C, 62.13; H, 7.46; N, 11.03%).

2-(4-Bromophenyl)-6-methylpyrano[4,3-b]pyrrol-4(1H)-one 4f. White plates from CH₂Cl₂: mp >300 °C (decomp.); IR (Nujol) ν_{max} 3200 (NH), 1660 (C=O) cm⁻¹; ¹H NMR (200 MHz; DMSO-d₆) δ 2.27 (3H, s, Me), 6.5 (1H, s, 7-H), 7.07 (1H, s, 3-H), 7.60–7.80 (4H, m, ArH), 12.25 (1H, br s, NH exchangeable); ¹³C NMR (50 MHz; DMSO-d₆) δ 19.32 (Me), 95.04 (C-7), 103.00 (C-3), 107.44 (C-3a), 126.23, 131.67 (ArCH), 120.02 and 130.26 (2 × ArC), 133.84 (C-2), 140.72 (C-7a), 155.80 (C-6), 159.38 (C=O) (Calc. for C₁₄H₁₀BrNO₂: C, 55.29; H, 3.31; N, 4.61. Found: C, 55.03; H, 3.34; N, 4.55%).

4-[2-(4-Bromophenyl)-1-morpholinoethylideneamino]-6methyl-2H-pyran-2-one **5f**. Chestnut plates from EtOH; mp 158 °C; IR (Nujol) v_{max} 1695 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 2.08 (3H, s, Me), 3.35–3.55 (8H, m, morpholine), 3.69 (2H, s, CH₂), 5.15 (1H, s, 3-H), 5.58 (1H, s, 5-H), 6.94 and 7.37 (4H, dd, AB system, J 8.4, ArH) (Calc. for C₁₈H₁₉BrN₂O₃: C, 55.38; H, 4.91; N, 7.18. Found: C, 55.17; H, 4.98; N, 6.93%).

2-(4-Fluorophenyl)-6-methylpyrano[4,3-b]pyrrol-4(1H)-one 4g. White plates from CH₂Cl₂; mp 248 °C; IR (Nujol) v_{max} 3100 (NH), 1680 (C=O) cm⁻¹; ¹H NMR (200 MHz; DMSO- d_6) δ 2.27 (3H, s, Me), 6.49 (1H, s, 7-H), 7.00 (1H, s, 3-H), 7.20–7.40 and 7.85–7.95 (4H, 2m, ArH), 12.18 (1H, br s, NH exchangeable); ¹³C NMR (50 MHz; DMSO- d_6) δ 19.31 (Me), 95.04 (C-7), 102.25 (C-3), 107.35 (C-3a), 115.54 and 115.97 (ArCH *ortho* to F, J 21.6), 126.33 and 126.48 (ArCH *meta* to F, J 7.6), 127.70 (ArC), 134.11 (C-2), 140.46 (C-7a), 155.62 (C-6), 158.9 and 159.46 (CF, *J* 245.0), 159.46 (C=O) (Calc. for C₁₄H₁₀FNO₂: C, 69.12; H, 4.15; N, 5.76. Found: C, 68.97; H, 4.05; N, 5.64%). *4-[2-(4-Fluorophenyl)-1-morpholinoethylideneamino]-6-*

methyl-2H-pyran-2-one **5***g*. White needles from PrⁱOH; mp 126 °C; IR (Nujol) ν_{max} 1690 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) δ 2.13 (3H, s, CH₃), 3.40–3.58 (8H, m, morpholine), 3.75 (2H, s, CH₂), 5.22 (1H, s, 3-H), 5.63 (1H, s, 5-H), 6.95–7.15 (4H, m, ArH) (Calc. for C₁₈H₁₉FN₂O₃: C, 65.43; H, 5.80; N, 8.48. Found: C, 65.29; H, 5.87; N, 8.33%).

Method B

A solution of a dihydrotriazole 3a-g (10 mmol) in propan-1-ol (60 ml) was boiled under reflux, progress of the reaction being followed by TLC. After disappearance of the starting material the solvent was removed *in vacuo*, a small amount of residue was dissolved in DMSO- d_6 and the ratio of 4 and 5 was determined by ¹H NMR analysis (see Table 1). Compounds 4 and 5 were isolated after column chromatography (cyclohexane–ethyl acetate 3 : 7). The dihydrotriazole 3e yielded only the amidine **5e**, which was crystallized. Analytical data were in agreement with those previously reported. The reaction times and isolated yields of compounds are collected in Table 2.

Method C. Effect of BF₃·Et₂O with respect to pyrolysis of dihydrotriazoles 3a–e

BF₃·Et₂O (1.9 ml, 15 mmol) in dry toluene (30 ml) was added to a stirred solution of a dihydrotriazole 3a-e (4.2 g, 15 mmol) in dry toluene (25 ml) at rt. The resulting mixture was refluxed for the same time used in the toluene thermal decomposition, after which the reaction was quenched with 10 ml of saturated aq. NaHCO₃ (10 ml). The organic phase was separated, and the aqueous phase was extracted with ethyl acetate (4×50 ml). The combined organic phases were washed with brine and dried (Na₂SO₄), filtered, and evaporated to dryness. Ratio of 4 and 5 determined by ¹H NMR analysis (DMSO-d₆) (see Table 1) of the residue. The crude mixtures containing 4a-g and 5a-g were chromatographed (cyclohexane-ethyl acetate 2:8) to afford first the pyrano[4,3-*b*]pyrrol-4-ones **4** and then the amidines **5**. For analytical data of the products previously obtained see above. The reaction times and isolated yields of compounds are collected in Table 2.

2,6-Dimethylpyrano[4,3-*b***]pyrrol-4(1***H***)-one 4e. White needles from Pr_{2}^{i}O; mp 157 °C; IR (Nujol) v_{max} 3090 (NH), 1670 (C=O) cm⁻¹; ¹H NMR (200 MHz; CDCl₃) \delta 2.26 (3H, s, 6-Me), 2.34 (3H, s, 2-Me), 6.27 (1H, s, 7-H), 6.33 (1H, s, 3H), 9.20 (1H, br s, NH exchangeable); ¹³C NMR (50 MHz; CDCl₃) \delta 13.5 (2-Me), 20.2 (6-Me), 96.3 (C-7), 103.4 (C-3), 107.8 (C-3a), 133.1 (C-2), 140.64 (C-7a), 155.2 (C-6), 162.4 (C=O) (Calc. for C₉H₉NO₂: C, 66.25; H, 5.56; N; 8.58. Found: C, 66.06; H, 5.74; N; 8.73%).**

Acknowledgements

We are indebted to Mrs Donatella Nava for recording NOESY, HETCOR and COLOC NMR experiments and discussing her experimental data, and to the M.U.R.S.T. (Italian Ministry of University, Science and Technology) for financial support.

References

- 1 Part 41. E. Erba, D. Pocar and M. Valle, J. Chem. Soc., Perkin Trans. 1, 1999, 421.
- 2 J. D. Hepworth, *Comprehensive Heterocyclic Chemistry*, ed. A. J. Boulton and A. McKillop, Pergamon Press, Oxford, 1984, vol. 3, pp. 874–883.
- 3 S. Thaisrivongs, K. D. Watenpaugh, W. J. Howe, P. K. Tomich, L. A. Dolak, K. T. Chong, C. S. C. Tomich, A. G. Tomasselli, S. R. Turner, J. W. Strohbach, A. M. Mulichak, M. N. Janakiraman, J. B. Moon, J. C. Linn, M. M. Horng, R. R. Hinshaw, K. H. Curry and D. J. Rothrock, *J. Med. Chem.*, 1995, **38**, 3624.

- 4 (a) M. Moreno-Mañas and R. Pleixats, *Adv. Heterocycl. Chem.*, 1992, **53**, 35; (b) G. P. Ellis, *Comprehensive Heterocyclic Chemistry*, ed. A. J. Boulton and A. McKillop, Pergamon Press, Oxford, 1984, vol. 3, pp. 679–681.
- 5 E. Erba and D. Sporchia, J. Chem. Soc., Perkin Trans. 1, 1997, 3021.
- 6 (a) P. Kadaba, B. Stanovnik and M. Tišler, Adv. Heterocycl. Chem., 1984, 37, 306 and references cited therein; (b) P. Kadaba, B. Stanovnik and M. Tišler, Adv. Heterocycl. Chem., 1984, 37, 339 and references cited therein.
- 7 D. Pocar, E. Roversi, P. Trimarco and G. Valgattarri, *Liebigs Ann.*, 1995, 487.
- 8 J. D. Hepworth, C. D. Gabbutt and B. M. Heron, *Comprehensive Heterocyclic Chemistry II*, ed. A. R. Katritzky, C. W. Rees and E. F. V. Scriven, Pergamon Press, Oxford, 1996, vol. 5, pp. 351–468.
 9 T. Gizur and K. Harsanyi, *J. Heterocycl. Chem.*, 1994, 31, 361.
- 10 R. Stradi and D. Pocar, *Gazz. Chim. Ital.*, 1969, **99**, 1131.

- 11 (a) D. Pocar, R. Stradi and L. M. Rossi, J. Chem. Soc., Perkin Trans. 1, 1972, 619; (b) E. Erba, G. Mai and D. Pocar, J. Chem. Soc., Perkin Trans. 1, 1992, 2709.
- (a) T. Pelter and M. Ayoub, J. Chem. Soc., Perkin Trans. 1, 1981, 1173; (b) T. D. Cyr and A. G. Poulton, Can. J. Chem., 1982, 60, 133;
 (c) J. Cervello, J. Marquet and M. Moreno-Mañas, Tetrahedron, 1990, 46, 2035 and references cited therein.
- (a) D. Tanner, Angew. Chem., Int. Ed. Engl., 1994, 33, 599;
 (b) L. Dubois, A. Metha, E. Tourette and R. H. Dodd, J. Org. Chem., 1994, 59, 433.
- 14 M. Cervera, M. Moreno-Mañas and R. Pleixats, *Tetrahedron*, 1990, 46, 7885.
- 15 N. L. J. M. Broekhof, F. L. Jonkers and A. van der Gen, *Tetrahedron Lett.*, 1979, 2433.
- 16 K. Stamos, Tetrahedron Lett., 1982, 23, 459.
- 17 G. Crispi, P. Giacconi, E. Rossi and R. Stradi, Synthesis, 1982, 787.