

SHORT COMMUNICATIONS

Three-Component Synthesis of 2-Chloropyridine-3,4-dicarbonitriles

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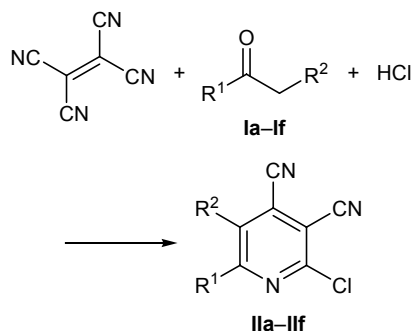
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According to published data, tetracyanoethylene reacts with ketones in the presence of a catalytic amount of hydrochloric acid to give 4-oxoalkane-1,1,2,2-tetracarbonitriles [1, 2]. It is also known that concentrated hydrochloric acid reacts with 4-oxoalkane-1,1,2,2-tetracarbonitriles to produce 2-chloropyridine-3,4-dicarbonitriles [3]. These data suggest that 2-chloropyridine-3,4-dicarbonitriles could be prepared via one-pot procedure, i.e., without isolation of 4-oxoalkane-1,1,2,2-tetracarbonitriles, which should make their preparation simpler and less expensive.

In fact, three-component reaction of tetracyanoethylene with the corresponding ketone **Ia–If** (aliphatic, aromatic, or heterocyclic) and hydrochloric acid in 1,4-dioxane afforded substituted 2-chloropyridine-3,4-dicarbonitriles **Ila–IIIf** in 86–97% yield. The structure of compounds **Ila–IIIf** was confirmed by IR, ¹H NMR, and mass spectra.



R¹ = R² = Me (**a**); R¹ = Me, R² = Et (**b**); R¹R² = (CH₂)₄, (CH₂)₆ (**d**); R¹ = Ph, R² = H (**e**); R¹ = 2-thienyl, R² = H (**f**).

Thus we proposed a one-pot procedure for the synthesis of 2-chloropyridine-3,4-dicarbonitriles, which

ensured higher yield and shorter reaction time. Compounds **IId** and **IIf** were not reported previously.

2-Chloro-5,6-dimethylpyridine-3,4-dicarbonitrile (IIa). Tetracyanoethylene, 0.64 g (0.005 mol), was added to a solution of 0.45 g (0.006 mol) of butan-2-one in 10 ml of 1,4-dioxane, 5 ml of concentrated hydrochloric acid was then added, and the mixture was stirred for 1–2 h at 60–70°C. When the reaction was complete (TLC), the mixture was diluted with water, and the precipitate was filtered off, washed with water and propan-2-ol, and recrystallized from propan-2-ol. Yield 0.85 g (89%), mp 72–74°C [3]. IR spectrum, ν, cm⁻¹: 2233 (C≡N), 1561, 1535 (C=C). ¹H NMR spectrum, δ, ppm: 2.56 s (3H, CH₃), 2.68 s (3H, CH₃). Mass spectrum: *m/z* 191 (*I*_{rel} 57%). Found, %: C 55.98; H 3.15; N 21.58. C₉H₆ClN₃. Calculated, %: C 56.41; H 3.16; N 21.93. *M* 191.62.

Compounds **IId–IIIf** were synthesized in a similar way.

2-Chloro-5-ethyl-6-methylpyridine-3,4-dicarbonitrile (IIb). Yield 0.95 g (93%), mp 46–48°C. IR spectrum, ν, cm⁻¹: 2234 (C≡N), 1554 (C=C). ¹H NMR spectrum, δ, ppm: 1.19 t (3H, CH₃, *J* = 8 Hz), 2.66 s (3H, CH₃), 2.86 q (2H, CH₂, *J* = 8 Hz). Mass spectrum: *m/z* 205 (*I*_{rel} 45%). Found, %: C 57.98; H 3.15; N 20.02. C₁₀H₈ClN₃. Calculated, %: C 58.41; H 3.92; N 20.43. *M* 205.65.

2-Chloro-5,6,7,8-tetrahydroquinoline-3,4-dicarbonitrile (IIc). Yield 1.05 g (97%), mp 95–96°C [3]. IR spectrum, ν, cm⁻¹: 2222 (C≡N), 1539 (C=C). ¹H NMR spectrum, δ, ppm: 1.90 m (4H, CH₂), 2.89 t (2H, CH₂, *J* = 6 Hz), 3.09 t (2H, CH₂, *J* = 6 Hz). Mass spectrum: *m/z* 217 (*I*_{rel} 100%). Found, %: C 59.96;

H 3.56; N 20.28. $C_{11}H_8ClN_3$. Calculated, %: C 60.70; H 3.70; N 19.31. M 217.66.

2-Chloro-5,6,7,8,9,10-hexahydrocycloocta[b]pyridine-3,4-dicarbonitrile (II d). Yield 1.15 g (94%), mp 72–74°C. IR spectrum, ν , cm^{-1} : 2234 ($C\equiv N$), 1552, 1538 ($C=C$). 1H NMR spectrum, δ , ppm: 1.31 m (4H, CH_2), 1.74 m (4H, CH_2), 3.01 t (2H, CH_2 , $J = 6$ Hz), 3.03 t (2H, CH_2 , $J = 6$ Hz). Mass spectrum: m/z 245 (I_{rel} 29%). Found, %: C 62.98; H 4.75; N 17.01. $C_{13}H_{12}ClN_3$. Calculated, %: C 63.55; H 4.92; N 17.10. M 245.71.

2-Chloro-6-phenylpyridine-3,4-dicarbonitrile (II e). Yield 1.04 g (87%), mp 171–172°C. IR spectrum, ν , cm^{-1} : 2232 ($C\equiv N$), 1601, 1573, 1526 ($C=C$). 1H NMR spectrum, δ , ppm: 7.61 m (3H, H_{arom}), 8.23 t (2H, H_{arom} , $J = 5$ Hz), 8.89 s (1H, CH). Mass spectrum: m/z 239 (I_{rel} 100%). Found, %: C 65.35; H 2.99; N 17.78. $C_{13}H_6ClN_3$. Calculated, %: C 65.15; H 2.52; N 17.53. M 239.66.

2-Chloro-6-(2-thienyl)pyridine-3,4-dicarbonitrile (II f). Yield 1.06 g (86%), mp 155–156°C. IR spectrum, ν , cm^{-1} : 2234 ($C\equiv N$), 1542, 1518 ($C=C$). 1H NMR spectrum, δ , ppm: 7.33 t (1H, thiophene, $J = 4$ Hz), 8.03 d (1H, thiophene, $J = 4$ Hz), 8.21 d (1H, thiophene, $J = 4$ Hz), 8.80 s (1H, CH). Mass spectrum: m/z 245 (I_{rel} 100%). Found, %: C 53.63; H 1.62;

N 17.23. $C_{11}H_4ClN_3S$. Calculated, %: C 53.77; H 1.64; N 17.10. M 245.69.

The purity of the isolated compounds was checked by TLC on Silufol UV-254 plates; development with UV light, by treatment with iodine vapor, or by heating. The IR spectra were recorded on an FSM-1202 spectrometer with Fourier transform from samples dispersed in mineral oil. The 1H NMR spectra were measured on a Bruker DRX-500 instrument operating at 500.13 MHz; DMSO- d_6 was used as solvent, and tetramethylsilane, as internal reference. The mass spectra (electron impact, 70 eV) were obtained on a Finnigan MAT Incos 50 mass spectrometer.

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