Chiral Diene-Phosphine Tridentate Ligands for Rhodium-Catalyzed Asymmetric Cycloisomerization of 1,6-Enynes

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ABSTRACT



Asymmetric cycloisomerization of nitrogen-bridged 1,6-envnes proceeded in the presence of a cationic rhodium complex coordinated with a chiral diene/phosphine tridentate ligand to give high yields of chiral 3-azabicyclo[4.1.0]heptenes with high enantioselectivity.

The recent development of transition-metal-catalyzed cycloisomerization of 1,n-envnes provides a useful methodology for the preparation of diverse polycyclic compounds in a single step.¹ Cycloisomerization of heteroatombridged 1,6-envnes is one of the most straightforward methods for the synthesis of bicyclo[4.1.0]heptene derivatives containing heteroatoms, such as oxygen and nitrogen, which have potential biological activities.² Although there have been several reports on the cycloisomerization catalyzed by π -acidic metals, such as Pt,³ Au,⁴ Rh,⁵ and Ir,^{6,7} asymmetric variants have not been well developed.⁸ Shibata and co-workers reported the first asymmetric cycloisomerization of nitrogen-bridged 1,6-envnes catalyzed by an iridium/bisphosphine complex under CO.⁶ A chiral bisphoshine/NHC- or a chiral monophosphine/cyclometalated NHC-platinum complex has been developed by Marinetti and co-workers.9 Michelet and co-workers reported that the asymmetric cycloisomerization with high enantioselectivity is catalyzed by chiral gold/bisphoshine complexes.^{10,11} Although high enantioselectivity is attained

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in some catalytic systems, they are still limited in terms of substrate scope and catalyst efficiency, and thus development of a new catalytic system is desirable. In this context, we recently reported that a rhodium(I) complex coordinated with triphenylphosphine and a chiral diene ligand¹² based on a tetrafluorobenzobarrelene (tfb) skeleton is a good catalyst for asymmetric cycloisomerization of nitrogen- and oxygen-bridged 1,6-envnes, where the active cationic rhodium species has a stereochemically controlled

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Scheme 1. A Rh/Chiral Diene-Phosphine Catalyst in Asymmetric Cycloisomerization of 1,6-Enynes



single coordination site¹³ on the rhodium center for electrophilic activation of the alkyne moiety (Scheme 1).¹⁴ The catalytic system, however, has some drawbacks as follows: (i) The oligomerization of enynes is sometimes observed, probably due to the dissociation of nonchelating triphenylphosphine. (ii) The applicable substrates are limited to enynes substituted with a methyl group at the alkyne terminus, and limited substituents of alkene moieties can

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only be applied to high yielding reactions with high enantioselectivity. (iii) The chiral tfb ligand is not readily available in an enantiopure form. To establish a more efficient and general catalytic system for cycloisomerization of 1,6-envnes, we designed new tridentate ligands for rhodium (Scheme 2).^{15,16} The designed rhodium catalysts involve characteristic features as follows: (i) A tridentate ligand, which has chelating one phosphorus atom and a chiral diene moiety, strongly coordinates to a rhodium center, and the in situ generated cationic complex provides a single vacant site on the square planar geometry of the rhodium(I) center. (ii) An electron-withdrawing character of an alkene moiety substituted with an ester group, which locates trans to the single vacant site of the cationic complex, is expected to enhance the π -acidity of rhodium toward electrophilic alkyne activation. (iii) The chiral diene framework is readily obtained from a natural product (R)- α -phellandrene. Here we report the development of new chiral diene-phosphine tridentate ligands for rhodium in asymmetric cycloisomerization of nitrogen-bridged 1,6-envnes giving 3-azabicyclo[4.1.0]heptene derivatives with high enantioselectivity.

Scheme 2. Concept of New Rh/Chiral Diene-Phosphine Catalysts



We focused on carboxylic acid (1R,4R,7R)-**2**,¹⁷ which is readily prepared from (R)- α -phellandrene, as a chiral diene framework for the synthesis of new chiral diene-monophosphine tridentate ligands (Scheme 3). The ligands were simply prepared by esterification of **2** with 2-(diarylphosphino)phenols. Thus, carboxylic acid **2** was treated with oxalyl chloride, and the resulting acid chloride was reacted

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with 2-(diarylphosphino)phenols in the presence of triethylamine to give diene-phosphine ligands $3\mathbf{a}-\mathbf{d}$ in high yields (90–95%) bearing several substituents on the two benzene rings of the phosphino group. Rhodium complexes coordinated with the ligands $3\mathbf{a}-\mathbf{d}$ were also prepared by the reactions with [RhCl(C₂H₄)₂]₂, and they were isolated in high yields (95–99%) by column chromatography on silica gel.

¹H NMR (CDCl₃) of the complex [RhCl((R)-3a)] displayed two nonequivalent alkenic protons at 3.81 and 4.61 ppm, which are shifted from 3.28 and 7.15 ppm, respectively, of ligand (R)-3a. Four alkenic carbons showing the



coupling with the rhodium center (50.4, 55.4, 84.1, and 109.1 ppm) were also observed in the ¹³C NMR spectrum, and ³¹P NMR displayed a doublet peak at 34.1 ppm (${}^{1}J_{Rh-P} = 173$ Hz). These results indicate that the rhodium center is coordinated with both the diene moiety and the phosphorus atom in solution.

To evaluate the designed rhodium catalysts, the reaction of 1,6-envne 4a was carried out in the presence of $[RhCl((R)-L^*)](2 \mod \%)$ and $NaBAr_4^F(4 \mod \%)(Ar^F =$ 3,5-bis(trifluoromethyl)phenyl) in 1,2-dichloroethane at 40 °C for 24 h (Table 1). The use of a rhodium complex coordinated with (R)-3a gave a 73% yield of the cycloisomerization product 5a, whose enantiomeric excess was 81% (entry 1). The substituent of the phosphorus atom on the ligand had a significant effect on the catalytic activity and enantioselectivity (entries 2-5). Thus, chiral ligand **3b** substituted with *p*-tolyl groups on the phosphorus atom improved both the yield and enantioselectivity of 5a (87%) yield, 82% ee) (entry 2). Ligand 3c having a bulky tertbutyl group at the *para*-position displayed the highest catalytic activity and enantioselectivity to give 5a in 90% vield with 91% ee (entry 3). High enantioselectivity was also observed by use of ligand 3d bearing bulkier aromatic groups (3,5-di-tert-butyl-4-methoxyphenyl), although the reaction was slow (entry 4), and a prolonged reaction time (72 h) was required for the complete conversion of 4a giving 5a in 90% yield with 89% ee (entry 5). Both the yield

Table 1. Rhodium-Catalyzed Cycloisomerization of 4a^a



| ontry | ligand (L*) | conversion $(\%)^b$ | wield $(\%)^b$ | $PP(\%)^c$ |
|--------|-----------------|---------------------|----------------|------------|
| entry | ligaliu (L) | conversion (70) | yieiu (70) | ee (70) |
| 1 | 3a | 82 | 73 | 81 |
| 2 | 3b | 98 | 87 | 82 |
| 3 | 3c | 100 | 90^d | 91 |
| 4 | 3d | 56 | 52 | 90 |
| 5^e | 3d | 100 | 90^d | 89 |
| 6 | $1/PPh_3^f$ | 33 | 27 | 83 |
| 7 | 6^{g} | 0 | 0 | _ |
| 8 | $6/PPh_3^{g,h}$ | 25 | 11 | 41 |
| 9 | 7 | 50 | 49 | 55 |
| 10^i | 3c | 0 | 0 | _ |

^{*a*} For detailed reaction conditions, see Supporting Information. ^{*b*} Determined by ¹H NMR. ^{*c*} Determined by HPLC. ^{*d*} Isolated yield. ^{*e*} For 72 h. ^{*f*} [RhCl(PPh₃)((*S*,*S*)-1)] (2 mol %). ^{*g*} [RhCl((*R*)-6)]₂ (2 mol %) of Rh). ^{*h*} PPh₃ (2 mol %). ^{*i*} Without NaBAr^F₄.

and the ee value of **5a** obtained here are higher than those obtained with ligand 1 shown in Scheme $1.^{14}$ Thus, the reaction catalyzed by $[RhCl(PPh_3)((S,S)-1)]$ (2 mol %) gave a 27% yield of 5a with 83% ee under the same reaction conditions (entry 6). Ligand 6^{17b} which lacks a phosphorus group, displayed no catalytic activity (entry 7).¹⁸ The use of triphenylphosphine as a second ligand combined with ligand 6 displayed low catalytic activity and enantioselectivity (entry 8). The use of diene-phosphine ligand 7, where an *o*-(diphenylphosphino)phenyl group is tethered by an ether functionality instead of the ester one of 3a, gave 5a in 49% yield 55% ee (entry 9). This result indicates that high catalytic activity of the rhodium/3a complex (entry 1 vs entry 9) is due to its high π -accepting ability caused by the electron-deficient alkene moiety located trans to the coordination site toward 4a. Facile formation of the cationic rhodium species with the aid of NaBAr^F₄ was also essential in the present reaction (entry 10). The absolute configuration of 5a obtained with (R)-3 was determined to be (1S, 6R, 7R)-(+) by comparison of its specific rotation with the value reported previously.¹⁴

The substrate scope of the present rhodium-catalyzed asymmetric cycloisomerization of nitrogen-bridged 1,6enynes **4** was fairly broad as shown in Scheme 4. The reaction was carried out by use of [RhCl((R)-3c)] or [RhCl((R)-3d)] as a precursor of the active cationic rhodium species,

⁽¹⁸⁾ The use of a chiral tetrafluorobenzobarrelene ligand substituted with a methyl and a 2-(diisopropylamido)phenyl group, which is an efficient tridentate ligand in the rhodium-catalyed asymmetric cyclopropanation of styrene (ref 15), gave no cycloisomerization product 5a.



Scheme 4. Rhodium-Catalyzed Cycloisomerization of

^a For detailed reaction conditions, see Table S1 in the Supporting Information. ^bPerformed with [RhCl((R)-3c)]. Performed with [RhCl((R)-3d)].

where the ligand, displaying high catalytic activity and enantioselectivity, was selected depending on the enynes. Asymmetric cycloisomerization can be applied to the enynes bearing not only a *p*-toluenesulfonyl group (Ts; **4a**) on the nitrogen atom but also a 4-nitrobenzenesulfonyl (*p*-Ns; **4b**) and 2-nitrobenzenesulfonyl group (*o*-Ns; **4c**) to give the corresponding bicyclic compounds **5a**-**5c** in high yields with 91, 87, and 76% ee, respectively. The reaction of 1,6-enynes **4d**-**4h** bearing aryl groups on the alkene moiety (\mathbf{R}^2) proceeded to give the corresponding bicyclic compounds 5d-5h in high yields, the enantioselectivity ranging between 90 and 99% ee. 1,6-Envnes substituted with a propyl group (4i), a silyloxymethyl group (4j) on the alkene moiety (\mathbb{R}^2), and unsubstituted 4k ($\mathbb{R}^2 = H$) also gave the corresponding cycloisomerization products 5i-5kin 64-97% yields over 88% ee. The envnes 4l and 4m substituted with propyl and phenyl at the alkyne terminus (\mathbf{R}^4) were also good substrates to give **5** and **5m** with 88% and 67% ee, respectively.¹⁹ In the reactions of 1,6-enynes **4n** and 40 bearing trisubstituted alkene moieties, although the vields of the cycloisomerization products were modest because of the formation of oligomeric compounds, enantioselectivities of the products were high (91% ee for 5n and 88% ee for 50). High enantioselectivities were also observed in the reactions of 1.6-envnes 4p-4t possessing an exomethylene part ($R^2 = R^3 = H$) giving the corresponding products in high yields with high enantioselectivity (86-92% ee).²⁰



The bicyclic compound 5a obtained here with 91% ee is readily converted into functionalized compounds without loss of enantiomeric purity (Scheme 5). For example, oxidative cleavage of an alkene moiety of 5a with ozone gave highly functionalized cyclopropane 8 in 55% yield. The allylation of 5a by treatment with allyltrimethylsilane in the presence of trifluoroacetic acid gave allylation product 9 in 94% yield.

In summary, we have developed a rhodium-catalyzed asymmetric cycloisomerization of nitrogen-bridged 1,6enynes giving 3-azabicyclo[4.1.0]heptenes in high yields with high enantioselectivity. The reaction was realized by use of a cationic rhodium complex coordinated with a chiral diene/phosphine tridentate ligand, which is readily prepared in an enantiopure form.

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Supporting Information Available. Experimental procedures and data for the substrates and products. This material is available free of charge via the Internet at http://pubs.acs.org.

⁽¹⁹⁾ The reaction of an oxygen-tethered analogue of 4m under the same reaction conditions did not give the corresponding oxabicyclo-[4.1.0]heptene derivative due to oligomerization of the starting 1,6enyne.

⁽²⁰⁾ The relative and absolute configurations of **5p** obtained with (R, R)-**3c** were determined to be (1*S*,6*S*) by X-ray crystallographic analysis (CCDC 816724).