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## Tetramic acid derivatives via Ugi-Dieckmann-reaction

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Tetramic acid derivatives represent an important class of nitrogen-containing heterocycles with a pyrrolidine-2,4-dione moiety. They are key structural motifs in many natural products of terrestrial and marine origin exhibiting a wide range of biological activities including antibiotic,<sup>1</sup> antiviral,<sup>2</sup> antifungal,<sup>3</sup> phyto-,<sup>4</sup> cytotoxic<sup>5</sup> and enzyme inhibitory activities against bacterial DNA-directed RNA polymerase.<sup>6</sup> Tetramic acids are also found in the agrochemical field, therein they have been patented for fungicidal and herbicidal use. The interesting biological and structural diversity of this compound class makes it a particularly interesting template for the design of compound libraries in search of small molecules that effect cellular signalling pathways. Therefore, several classical synthetic procedures<sup>7</sup> were developed, but with a lack of diversity that is required for an effective lead discovery and optimization. In contrast to the classical organic synthesis, the combinatorial synthesis of 'drug-like' compounds permits the fast preparation of compound libraries suitable for lead finding and optimization.<sup>8–18</sup> Thus, multi-component reactions (MCRs) represent a powerful tool for the high-throughput screening strategy.<sup>19,20</sup> Especially, the Ugi-reaction has generated much interest due to its synthetic potential, and the capacity to generate molecular diversity. In the Ugi-four component reaction,<sup>21</sup> an amine, an

## ABSTRACT

Tetramic acid derivatives constitute an important class of nitrogen containing heterocycles, and are key structural motifs in many natural products of terrestrial and marine origin. The interesting biological and structural diversity of this class of substances makes it a particularly interesting template for the design of compound libraries in search of small molecules that effect cellular signalling pathways. Therefore, a novel combinatorial synthesis of tetramic acids by an Ugi/Dieckmann condensation is described.

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aldehyde, a carboxylic acid and an isocyanide react simultaneously to afford peptide-like structures.

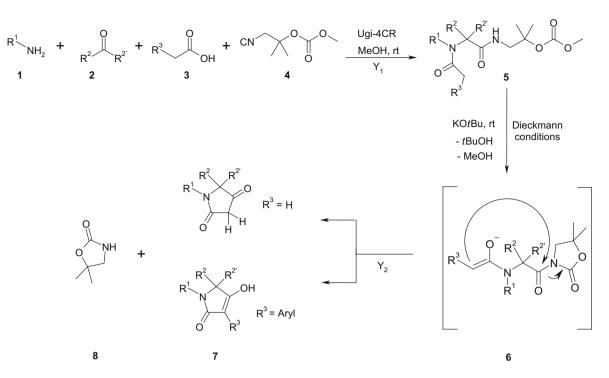
In order to reach a maximum of diversity, several research groups have successfully combined different classical methods with multi-component reactions.<sup>22–26</sup> In this context, we wish to introduce a novel reaction type (Ugi–Dieckmann) based on the concept of a 'universal isocyanide'<sup>27</sup> that enables the required post condensation modification after the Ugi-four-component reaction.

1,1,-Dimethyl-2-isocyano-ethyl-methylcarbonate<sup>28</sup> is used as cleavable isocyanide for the Ugi-4CR. In the following post condensation modification the deprotonation of amide 5 initiates the cyclization to the N-acyl-5,5-dimethyloxazolidin-2-one 6. Upon attack by the enolized carboxylic acid moiety 5,5-dimethyloxazolidin-2-one 8 acts as leaving group and a Dieckmann-like cyclization to pyrrolidine-2,4-dione or hydroxy-dihydropyrrolidone structures 7 takes place (Scheme 1). The Ugi-reaction is generally initiated by the condensation of amine 1 with aldehyde 2 leading to an intermediate imine, which subsequently reacts with  $\alpha$ -CH-acidic carboxylic acid/acetic acid **3** and isocyanide **4** to afford the desired product 5. Herein, MeOH turned out to be the best solvent for the MCR, step. After completion of the MCR the solvent was removed in vacuo. The conversions of the MCR products 5a**ae** determined by HPLC–MS<sup>29</sup> were generally good. The clean-up of the crude products was done by column chromatography.<sup>3</sup> The subsequent intramolecular cyclization required a strong base KOtBu in THF (dry) at room temperature. In the case of acetic acid,



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Scheme 1. Combinatorial synthesis of tetramic acids by an Ugi/Dieckmann condensation.

2.0 equiv KOtBu is used to obtain exclusively the cyclization product, whereby for  $\alpha$ -CH-acidic carboxylic acids 1.2 equiv was sufficient.

Conversions were monitored by HPLC–MS, generally after 1 h of reaction time a maximum of conversion was reached, and the reaction mixture was neutralized with 6 N HCl (pH 6–7). All com-

pounds were isolated by column chromatography on silica in good yields  $(Y_2)$  and purities.<sup>32</sup>

Table 1 shows the results for the synthesized tetramic acids<sup>33–38</sup> **7a–ae** with specific yields for each step ( $Y_1$  = MCR,  $Y_2$  = cyclization). Aliphatic, aromatic and benzylic amines, aldehydes as well as ketones, aliphatic as well as  $\alpha$ -CH-acidic carbox-

Table 1	1
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Synthesized (	tetramic	acid	derivatives
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R <sup>1</sup>	R <sup>2</sup>	R <sup>2'</sup>	R <sup>3</sup>	Y <sub>1</sub> (%)	MCR	Y <sub>2</sub> (%)	Cyclization
4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub>	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub>	Н	C <sub>6</sub> H <sub>5</sub>	64	5a	13	7a
$4-CH_{3}O-C_{6}H_{4}$	CH <sub>3</sub>	CH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	51	5b	21	7b
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	$2-F-C_6H_4$	Н	$4-Cl-C_6H_4$	65	5c	74	7c
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub>	Н	2-Thienyl	64	5d	26	7d
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	$4-CH_{3}O-C_{6}H_{4}$	Н	$4-Cl-C_6H_4$	43	5e	78	7e
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	Н	Н	$4-Cl-C_6H_4$	60	5f	45	7f
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	$2-F-C_6H_4$	28	5g	48	7g
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	2-Thienyl	54	5h	43	7h
$C_6H_5-CH_2$	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	$4-Cl-C_6H_4$	75	5i	81	7i
$C_6H_5-CH_2$	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	$4-CH_{3}O-C_{6}H_{4}$	79	5j	81	7j
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	$4 - NO_2 - C_6 H_4$	79	5k	82	7k
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	COOC <sub>2</sub> H <sub>5</sub>	20	51	73	71
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	Н	79	5m	68	7m
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	$O-C_6H_5$	74	5n	24	7n
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	C <sub>6</sub> H <sub>5</sub>	85	50	99	70
$C_6H_5-CH_2$	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	$PO(OC_2H_5)_2$	72	5p	99	7p
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	S-CH <sub>2</sub> -C <sub>6</sub> H <sub>5</sub>	73	5q	55	7q
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	CH <sub>3</sub>	CH₃	2-Thienyl	45	5r	18	7 <b>r</b>
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	CH <sub>3</sub>	CH₃	$4-Cl-C_6H_4$	47	5s	69	7s
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	CH <sub>3</sub>	CH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	71	5t	32	7t
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	C <sub>6</sub> H <sub>5</sub>	Н	$4-Cl-C_6H_4$	42	5u	68	7u
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	$(CCH_3)_3$	Н	$4-Cl-C_6H_4$	50	5v	46	7v
$C_6H_5-CH_2$	$(CH_3)_2 - CH$	Н	$4-Cl-C_6H_4$	53	5w	35	7w
$(CH_2)_2 - OCH_3$	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub>	Н	2-Thienyl	76	5x	99	7x
$(CH_2)_2 - OCH_3$	$4-CH_{3}O-C_{6}H_{4}$	Н	$4-Cl-C_6H_4$	58	5y	93	7y
$(CH_2)_2 - OCH_3$	$4-CH_{3}O-C_{6}H_{4}$	Н	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub>	61	5z	13	7z
$(CH_2)_2 - OCH_3$	$4-CH_{3}O-C_{6}H_{4}$	Н	C <sub>6</sub> H <sub>5</sub>	73	5aa	80	7aa
$(CH_2)_2 - OCH_3$	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	2-Thienyl	57	5ab	85	7ab
$(CH_2)_2 - OCH_3$	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	$4-Cl-C_6H_4$	67	5ac	72	7ac
$(CH_2)_2 - OCH_3$	$(CH_3)_2$ -CH	Н	$4-CH_{3}O-C_{6}H_{4}$	57	5ad	99	7ad
$(CH_2)_2 - OCH_3$	(CH <sub>3</sub> ) <sub>2</sub> -CH	Н	C <sub>6</sub> H <sub>5</sub>	89	5ae	99	7ae

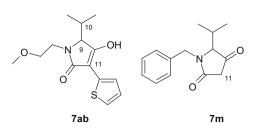


Figure 1. Compounds 7ab and 7m.

ylic acid could successfully be involved in the reaction. The reaction time (rt) for the cyclization is generally short, and the conversions are moderate to good for all compounds. The chromatographic methods used allow the isolation of products with high purity (>95%). All compounds were characterized by NMR<sup>30</sup> and HPLC–MS data.

The observable tautomer (Fig. 1) in <sup>1</sup>H NMR spectra (DMSO- $d_6$ ) depends on the substituent at **C-11**. If the starting material is acetic acid there is no substituent at **C-11** of the resulting product, and so only the pyrrolidine-2,4-dione-tautomer is observed (CH<sub>2</sub>-group at 3.02 ppm, carbonyl-carbon at 206 ppm). If an  $\alpha$ -CH-acidic carboxylic acid is used, the resulting products contain a phenyl moiety at **C-11**, and therefore the hydroxy-dihydropyrrolidone tautomer is the exclusively observable structure in <sup>1</sup>H NMR spectra (DMSO- $d_6$ ). The enolic structure of compound **7ab** was confirmed by NOE-experiments. In DPFGSE-NOE-spectra of compound **7ab** NOEs with methine-proton 9 ( $\delta$  4.11 ppm) and methine-proton 10 ( $\delta$  2.36 ppm) of the isopropyl-group were observed upon excitation of enol-proton at ( $\delta$  11.47 ppm) and vice versa.

In summary, a novel two-step synthetic procedure for the preparation of substituted tetramic acid derivatives has been described. Amines, carbonyls and  $\alpha$ -CH-acidic carboxylic acid can be varied broadly, leading to compounds with three potential points of diversity.

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- HPLC-MS/MS spectra (Varian 1200), Polaris, RP C18 column, 3 × 150 mm, 5 μm, ProStar 325 (254 nm), 1 ml/min, 3 min gradient from 10% ACN to 90% ACN (0.1% HCOOH) versus H<sub>2</sub>O, coupled with a Quadrupol MS/MS mass spectrometer using electrospray ionization (ESI).
  NMR<sup>1</sup>H/<sup>13</sup>C: Bruker AV 250: 250.13 MHz, <sup>1</sup>H-<sup>1</sup>H NOESY: Jeol ECP500:
- NMR:<sup>1</sup>H/<sup>13</sup>C: Bruker AV 250: 250.13 MHz, <sup>1</sup>H-<sup>1</sup>H NOESY: Jeol ECP500: 500.16 MHz.
- 31. General procedure (GP 1) for the synthesis of MCR products 5a-ae: Amine 1 (1 mmol) and aldehyde 2 (1 mmol) were stirred in 3 mL methanol for 2 h. Then, carboxylic acid 3 (1 mmol) and isocyanide 4 (1 mmol) were added, and the reaction mixture was stirred for 16 h at room temperature. The solvent was removed in vacuo. The resulting crude product was purified by flash chromatography on silica (ethyl cetate/hexane).
- 32. General procedure (GP 2) for the synthesis of tetramic acid derivatives 6a-ae: 0.2 mmol of MCR product 5a-ae was dissolved in 4 mL THF (dry), and 0.24 mmol KOtBu were added under nitrogen atmosphere. The reaction mixture was stirred at room temperature for 1 h. After 1 h of reaction time a maximum of conversion was reached, and the reaction mixture was neutralized with 6 N HCl (pH 6-7). The solvent was removed in vacuo. The resulting crude product was purified by flash chromatography on silica (ethyl acetate/hexane).
- 33. Compound **70** was prepared according to **GP 2**, and purified by flash chromatography on silica gel with eluent ethyl acetate/hexane = 1:2 (61 mg of a colourless solid, 99%). *m*/*z* = 308 [M+H]<sup>+</sup>, *m*/*z* = 330 [M+N]<sup>+</sup>. <sup>1</sup>H NMR (250.13 MHz, CDCl<sub>3</sub>):  $\delta$  11.25 (br s, 1H, OH), 7.92–7.90 (m, 2H, CH<sub>ar</sub>), 7.36–7.20 (m, 8H, CH<sub>ar</sub>), 5.00 (d, *J* = 15.7 Hz, 1H, PhCH<sub>2</sub>), 4.20 (d, *J* = 15.7 Hz, 1H, PhCH<sub>2</sub>), 3.73 (d, *J* = 2.3 Hz, 1H, NCH), 2.23–2.29 (m, 1H, (CH<sub>3</sub>)<sub>2</sub>CH), 0.96 (d, *J* = 6.9 Hz, 3H, CH(CH<sub>3</sub>)<sub>2</sub>), 0.81 (d, *J* = 6.9 Hz, 3H, CH(CH<sub>3</sub>)<sub>2</sub>). <sup>13</sup>C NMR (62.90 MHz, CDCl<sub>3</sub>):  $\delta$  172.8 (C<sub>q</sub>, CON), 170.0 (C<sub>q</sub>, HOC<sub>q</sub>), 139.2 (C<sub>q</sub>, CH<sub>2</sub>C<sub>ar</sub>), 132.7 (C<sub>q</sub>, c<sub>ar</sub>), 129.5 (CH, C<sub>ar</sub>H), 128.9 (CH, C<sub>ar</sub>H), 128.5 (CH, C<sub>ar</sub>H), 128.3 (CH, C<sub>ar</sub>H), 127.8 (CH, C<sub>ar</sub>H), 128.5 (CH, C<sub>ar</sub>H), 128.9 (CH, C<sub>ar</sub>H), 128.9 (CH, C<sub>ar</sub>H), 20, 63.7 (CH, NCH), 45.2 (CH<sub>2</sub>, PhCH<sub>2</sub>), 2.9.6 (CH, (CH<sub>3</sub>)<sub>2</sub>CH), 19.9 (CH<sub>3</sub>, CH(CH<sub>3</sub>)<sub>2</sub>), 16.4 (CH<sub>3</sub>, CH(CH<sub>3</sub>)<sub>2</sub>).
- 34. Compound **7s** was prepared according to **GP 2** and purified by flash chromatography on silica gel with eluent ethyl acetate/hexane = 1:1 (19 mg of a white solid, 69%). m/z = 294 [M+H]<sup>+</sup>, m/z = 316 [M+Na]<sup>\*</sup>. <sup>1</sup>H NMR (250.13 MHz, DMSO): δ 8.12–8.06 (m, 2H, CH<sub>ar</sub>), 8.01 (br s, 1H, OH), 7.19–7.41 (m, 7H, CH<sub>ar</sub>), 4.53 (s, 2H, NCH<sub>2</sub>), 1.32 (s, 6H, CH<sub>3</sub>).
- 35. Compound **7**t was prepared according to **GP 2** and purified by flash chromatography on silica gel with eluent ethyl acetate/hexane = 1:2 (45 mg of a colourless oil, 69%).  $m/z = 328 [M+H]^+$ ,  $m/z = 350 [M+Na]^+$ . <sup>1</sup>H NMR (250.13 MHz, CDCl<sub>3</sub>):  $\delta$  7.98 (d, J = 7.4 Hz, 1H, CH<sub>ar</sub>), 7.40–7.18 (m, 9H, CH<sub>ar</sub>), 4.53 (s, 2H, PhCH<sub>2</sub>), 1.26–1.08 (m, 6H, CH(CH<sub>3</sub>)<sub>2</sub>). <sup>13</sup>C NMR (62.90 MHz, CDCl<sub>3</sub>):  $\delta$  174.0 (Cq, CON), 169.5 (Cq, HOCq), 140.1 (Cq, CH<sub>2</sub>Car), 128.4 (CH, CarH), 128.2 (Cq, Car), 127.7 (CH, CarH), 127.4 (CH, CarH), 126.7 (CH, CarH), 125.7 (CH, CarH), 101.2 (Cq, PhCar), 60.6 (Cq, NCq(CH<sub>3</sub>)<sub>2</sub>), 41.1 (CH<sub>2</sub>, PhCH<sub>2</sub>), 23.5 (CH<sub>3</sub>, Cq(CH<sub>3</sub>)<sub>2</sub>), 23.0 (CH<sub>3</sub>, Cq(CH<sub>3</sub>)<sub>2</sub>).
- 36. Compound **7y** was prepared according to **GP 2** and purified by flash chromatography on silica gel with eluent ethyl acetate/hexane = 9:1 (68 mg of a colourless oil, 93%). m/z = 374 [M+H]<sup>+</sup>, m/z = 396 [M+Na]<sup>+</sup>. <sup>1</sup>H NMR (250.13 MHz, CDCl<sub>3</sub>):  $\delta$  7.96 (d, <sup>3</sup>J = 8.1 Hz, 2H,  $CH_{ar}$ ), 7.21 (d, <sup>3</sup>J = 8.1 Hz, 2H,  $CH_{ar}$ ), 7.08 (d, <sup>3</sup>J = 8.6 Hz, 2H, CH<sub>ar</sub>), 6.77 (d, <sup>3</sup>J = 8.6 Hz, 2H, CH<sub>ar</sub>), 4.74 (s, 1H, NCH), 3.74 (s, 3H, OCH<sub>3</sub>), 3.53–3.71 (m, 2H, CH<sub>2</sub>), 3.34–3.38 (m, 1H, CH<sub>2</sub>), 3.20 (s, 3H, CH<sub>2</sub>OCH<sub>3</sub>), 2.77–2.86 (m, 1H, CH<sub>2</sub>).
- 37. Compound **7ab** was prepared according to **GP 2** and purified by flash chromatography on silica gel with eluent ethyl acetate/hexane = 9:1 (48 mg of a white solid, 85%). m/z = 282 [M+H]<sup>\*</sup>, m/z = 304 [M+Na]<sup>\*</sup>. <sup>1</sup>H NMR (500.16 MHz,  $d_6$ -DMSO):  $\delta$  11.47 (br s, 1H, OH), 7.47 (dd, J = 1.1, 2.5 Hz, 1H,  $CH_{ar}$ ), 7.24 (dd, J = 1.1, 2.5 Hz, 1H,  $CH_{ar}$ ), 6.92 (dd, J = 1.6, 8.7 Hz, 1H,  $CH_{ar}$ ), 4.00 (s, 1H, H<sub>3</sub>COCH<sub>2</sub>), 3.85–3.75 (m, 1H, H<sub>3</sub>COCH<sub>2</sub>), 3.47–3.43 (m, 2H, NCH<sub>2</sub>), 3.13 (s, 3H, OCH<sub>3</sub>), 3.08–2.98 (m, 1H, NCH), 2.35 (m, 1H, (CH<sub>3</sub>)<sub>2</sub>CH), 0.90 (d, J = 7.0 Hz, 3H, CH(CH<sub>3</sub>)<sub>2</sub>), 0.67 (d, J = 7.0 Hz, 3H, CH(CH<sub>3</sub>)<sub>2</sub>).<sup>13</sup>C NMR (62.90 MHz, DMSO- $d_6$ ):  $\delta$  171.0 (Cq, HOCq), 167.2 (Cq, CON), 143.7 (Cq, SCar), 133.3 (CH,  $C_{ar}$ H), 126.9 (CH,  $C_{ar}$ H), 124.0 (CH,  $C_{ar}$ H), 102.3 (Cq,  $c_{ar}$ ), 78.1 (CH, NCH), 70.5 (CH<sub>2</sub>, H<sub>3</sub>COCH<sub>2</sub>), 64.1 (CH<sub>3</sub>, OCH<sub>3</sub>), 58.6 (CH<sub>2</sub>, NCH<sub>2</sub>), 29.5 (CH, (CH<sub>3</sub>)<sub>2</sub>/LH), 19.8 (CH<sub>3</sub>, CH(CH<sub>3</sub>)<sub>2</sub>), 15. (CH<sub>3</sub>, CH(CH<sub>3</sub>)<sub>2</sub>).
- 38. Compound **7ae** was prepared according to **GP 2** and purified by flash chromatography on silica gel with eluent ethyl acetate/hexane = 9:1 (54 mg of a white solid, 99%).  $m/z = 276 [M+H]^+$ ,  $m/z = 298 [M+Na]^+$ . <sup>1</sup>H NMR (250.13 MHz, CDCl<sub>3</sub>):  $\delta$  7.73 (d, J = 8.5 Hz, 2H,  $CH_{ar}$ ), 7.23 (t, J = 7.5 Hz, 2H,  $CH_{ar}$ ), 7.99 (d, J = 7.4 Hz, 1H,  $CH_{ar}$ ), 3.93 (d, J = 2.2 Hz, 1H, NCH), 3.85–3.75 (m, 2H, H<sub>3</sub>COCH<sub>2</sub>), 3.14 (s, 3H, OCH<sub>3</sub>), 3.10–2.98 (m, 2H, NCH<sub>2</sub>), 2.19 (m, 1H,  $(CH_3)_2CH$ ), 0.89 (d, J = 7.0 Hz, 3H,  $CH(CH_3)_2$ ), 0.75 (d, J = 7.0 Hz, 3H,  $CH(CH_3)_2$ ). <sup>13</sup>C NMR (62.90 MHz, CDCl<sub>3</sub>):  $\delta$  171.6 (C<sub>q</sub>. CON), 169.2 (C<sub>q</sub>. HOC<sub>q</sub>), 132.1 (C<sub>q</sub>. C<sub>ar</sub>), 128.0 (CH, C<sub>ar</sub>H), 127.7 (CH, C<sub>ar</sub>H), 126.1 (CH, C<sub>ar</sub>H), 105.1 (Cq, PhC<sub>ar</sub>), 70.3 (CH, NCH), 63.5 (CH<sub>2</sub>, H<sub>3</sub>COCH<sub>2</sub>), 58.2 (CH<sub>3</sub>, OCH<sub>3</sub>), 38.7 (CH<sub>2</sub>, NCH<sub>2</sub>), 30.0 (CH, (CH<sub>3</sub>)<sub>2</sub>CH), 18.9 (CH<sub>3</sub>, CH(CH<sub>3</sub>)<sub>2</sub>), 16.1 (CH<sub>3</sub>, CH(CH<sub>3</sub>)<sub>2</sub>).