

Pd-Catalyzed Regioselective Iminothiolation of Alkynes: A Remarkable Effect of the CF₃ Group of Iminosulfides

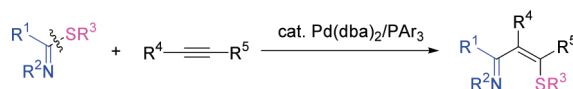
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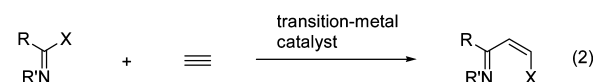
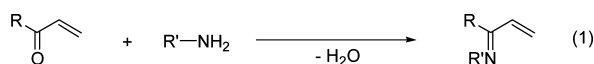
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ABSTRACT



Pd-catalyzed iminothiolation of alkynes took place to afford 4-SR substituted 1-azadienes regioselectively.

1-Azadienes have been employed as versatile synthetic building blocks: electrophiles in 1,2-addition as well as Michael-type 1,4-addition, nucleophiles by a nitrogen atom, and heterodienes in hetero-Diels–Alder reactions.¹ The general method of preparation is the condensation of α,β -unsaturated ketones with primary amines (eq 1). In addition, the transition-metal-catalyzed iminocarbonvinylcarbon bond-formation reaction is a promising alternative. Although some catalytic reactions such as the Pd-catalyzed cross-coupling of imidoyl chlorides with vinyl stannanes and the Pd-catalyzed Mizoroki–Heck-type reaction of imidoyl iodides with alkenes have been reported,² the catalytic introduction of imino groups is still limited. In particular, to the best of our knowledge, an addition reaction to alkynes is unprecedented (eq 2).³



We have devoted our attention to the study of transition-metal-catalyzed carbochalcogenation of carbon–carbon un-

saturated bonds.^{4,5} The Pt(PPh₃)₄-catalyzed intermolecular decarbonylative arylthiolation of alkynes using thioesters is among them (Scheme 1, upper left).^{4b,5a} Recently, CO-retained arylthiolation (Scheme 1, lower left) and trifluoroacetylthiolation (Scheme 1, upper right) were realized by Pd/dppe and Pt(PPh₃)₄ catalyzes, respectively.^{5c} Herein is reported a new synthetic method of 4-sulfur functionalized 1-azadienes by the iminothiolation of alkynes using imino-sulfide (Scheme 1, lower right).

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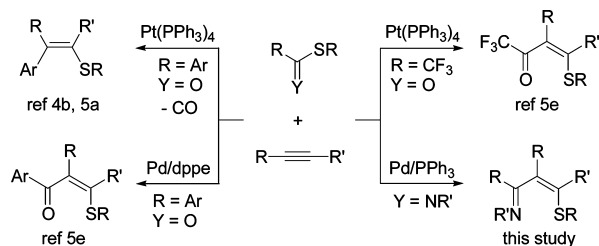
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Scheme 1. A Series of Carbothiolations of Alkyne



First, the success of trifluoroacetylthiolation of alkynes by $\text{CF}_3\text{C}(\text{O})\text{SR}^{5e}$ prompted scrutiny of reactions that use an iminosulfide [**1a**; $\text{CF}_3\text{C}(\text{=NPh})\text{S}(p\text{-tolyl})$]. Gratifyingly, it was found that the reaction of **1a** (0.5 mmol) with ethyl phenylpropiolate (**2a**, 1.0 mmol), which exhibited high activity for the Pt-catalyzed decarbonylative arylthiolation by thioesters,^{5a,6} was catalyzed by $\text{Pd}(\text{dba})_2$ (0.025 mmol) and PPh_3 (0.5 mmol) in 1,2-dichloroethane at 100 °C for 3 h under microwave irradiation using a sealed glass vessel. The desired adduct, *E*- $\text{CF}_3\text{C}(\text{=NPh})\text{C}(\text{CO}_2\text{Et})=\text{C}(\text{Ph})[\text{S}(p\text{-tolyl})]$ (**E-3a**), was isolated in 91% yield regio- and stereo-selectively (entry 1, Table 1).⁷ The structure of **E-3a** was unambiguously determined by X-ray crystallography, demonstrating that $\text{CF}_3\text{C}(\text{=NPh})$ of **1a** was bound at the α -carbon

of the $\text{EtOC}(\text{O})$ of **2a** and $\text{S}(p\text{-tolyl})$ of **1a** at the α -carbon of the Ph of **2a** as the *E*-isomer with *anti* conformation regarding the stereochemistry of lone pair of N (Figure 1).^{8,9}

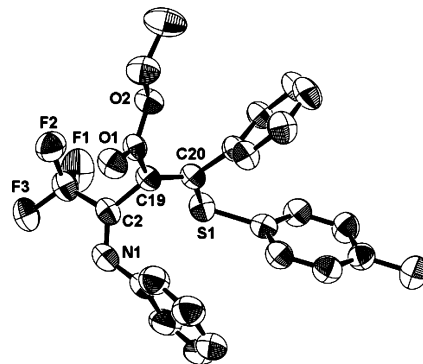


Figure 1. ORTEP diagram of **E-3a**.

The azadiene moiety was not arranged in a planar fashion ($\text{N1}-\text{C2}-\text{C19}-\text{C20}$ torsion angle is 68.7°), showing that the imine and vinyl moieties are not conjugated in the solid state.

In stark contrast, Pd/dppe and $\text{Pt}(\text{PPh}_3)_4$, which work as good catalysts for arylthiolation and trifluoroacetylthiolation, respectively (Scheme 1), hardly promoted the reaction (entries 2 and 3, Table 1).

The results of the $\text{Pd}(\text{dba})_2/\text{PPh}_3$ catalyzed addition of various iminosulfides [**1**; $\text{R}^1\text{C}(\text{=NR}^2)\text{SR}^3$] to **2a** are summarized in Table 1. The reactions of **1b** and **1c** having SPh and $\text{S}(p\text{-ClC}_6\text{H}_4)$ as S-aryl groups took place to furnish the corresponding adducts **3b** and **3c** in good yields (entries 4 and 5, Table 1). Neither electron-donating nor electron-withdrawing groups at Y interfered with the reactions (entries 6 and 7, Table 1). A similar transformation was also accomplished by using **1f** with a benzyl group as R^3 to produce **3f** in 91% yield (entry 8, Table 1). In the case of **1g** ($\text{R}^1 = \text{Ph}$), the desired adduct **3g** was obtained only in 44% yield even in the presence of 10 mol % of $\text{Pd}(\text{dba})_2/2\text{P}(p\text{-tolyl})_3$ (entry 9, Table 1). Phenethyl substituted iminosulfide (**1h**) was not effective; **3h** was not synthesized (entry 10, Table 1). These results demonstrated that CF_3 groups as R^1 of **1** is quite essential for the successful transformation.

The results of the reactions of **1a** with a variety of alkynes are summarized in Table 2. It was found that even the addition to 4-octyne (**2b**) took place at 160 °C to give **3i**, albeit in low yield (entry 1, Table 2). The treatment with 3-methoxy-1-phenylpropyne (**2c**) afforded **3j** in 51% yield (*E*:*Z* = 67:33) at 180 °C (entry 2, Table 2). On the other hand, terminal alkynes underwent the addition under much milder conditions (80 °C) than internal alkynes (entries 3–12, Table 2). Both electron-rich and electron-poor arylalkynes (**2d–g**) reacted with **1a** to form high yields of **3** (entries 3–6, Table 2). The product of iminothiolation was obtained in 89% yield (*E*:*Z* = 18:82) when 1-octyne was used (**2h**) (entry 7, Table 2). Cyclohexylacetylene (**2i**) was also applied

Table 1. Pd-Catalyzed Iminothiolation of **2a** Using Various Iminosulfides (**1**)^a

		$\text{R}^1\text{C}(\text{=NR}^2)\text{SR}^3 + \text{EtOC}(\text{O})-\text{C}\equiv\text{C}-\text{Ph}$ 1 2a		$\xrightarrow[\text{1,2-dichloroethane, 100 }^\circ\text{C}]{\text{cat. Pd}(\text{dba})_2/\text{PPh}_3}$ 3	
entry	1		time (h)	3	(%) ^b (<i>E</i> : <i>Z</i>)
		$\text{F}_3\text{C}-\text{C}(\text{=N}-\text{S}(p\text{-X-C}_6\text{H}_4))=\text{C}(\text{Ph})-\text{C}(\text{CO}_2\text{Et})=\text{C}(\text{Ph})-\text{C}(\text{SR}^3)$ $(p\text{-YC}_6\text{H}_4)$			
1	1a	X = Me, Y = H	3	3a	91 (>99:1)
2 ^c	1a		3	3a	4 ^d (95:5)
3 ^e	1a		3	3a	n.d.
4	1b	X = Y = H	3	3b	82 (>99:1)
5	1c	X = Cl, Y = H	1	3c	71 (>99:1)
6	1d	X = Me, Y = OMe	1	3d	88 (>99:1)
7	1e	X = Me, Y = Cl	1	3e	85 (79:21)
8	1f	$\text{F}_3\text{C}-\text{C}(\text{=N}-\text{S}-\text{CH}_2\text{Ph})=\text{C}(\text{Ph})-\text{C}(\text{CO}_2\text{Et})=\text{C}(\text{Ph})-\text{C}(\text{SR}^3)$	2	3f	91 (97:3)
9 ^f	1g	$\text{Ph}-\text{C}(\text{=N}-\text{S}(p\text{-tolyl}))=\text{C}(\text{Ph})-\text{C}(\text{CO}_2\text{Et})=\text{C}(\text{Ph})-\text{C}(\text{SR}^3)$	3	3g	44 (81:19)
10 ^f	1h	$\text{Ph}-\text{CH}_2-\text{CH}_2-\text{C}(\text{=N}-\text{S}(p\text{-tolyl}))=\text{C}(\text{Ph})-\text{C}(\text{CO}_2\text{Et})=\text{C}(\text{Ph})-\text{C}(\text{SR}^3)$	3	3h	n.d.

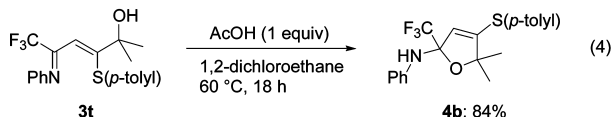
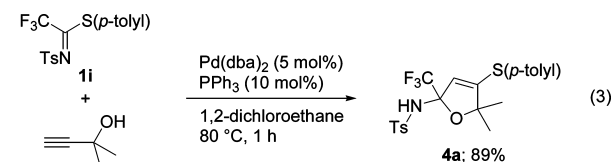
^a Unless otherwise noted, **1** (0.5 mmol), **2a** (1.0 mmol), $\text{Pd}(\text{dba})_2$ (0.025 mmol), PPh_3 (0.05 mmol), and 1,2-dichloroethane (0.25–0.5 mL) at 100 °C under microwave irradiation for 1–3 h. All reactions were performed in sealed glass vessels. ^b Isolated yield. ^c $\text{Pd}(\text{dba})_2$ (0.025 mmol) and dppe (0.025 mmol). ^d NMR yield. ^e $\text{Pt}(\text{PPh}_3)_4$ (0.025 mmol). ^f $\text{Pd}(\text{dba})_2$ (0.05 mmol) and $\text{P}(p\text{-tolyl})_3$ (0.1 mmol). dba = dibenzylideneacetone.

Table 2. Pd-Catalyzed Iminothiolation of Various Alkynes (**2**) Using **1a**^a

entry	2	temp (°C)	time (h)	3	(%) ^b (<i>E</i> : <i>Z</i>)
1	2b	160 ^c	1	3i	14 (5:95)
2	2c	180 ^c	3	3j	51 (67:33)
3	2d	80	1	3k	92 (2:98)
4	2e	80	1	3l	95 (14:86)
5	2f	80	1	3m	74 (2:98)
6	2g	80	1	3n	83 (1:~99)
7	2h	80	1	3o	89 (18:82)
8	2i	80	3	3p	76 (14:86)
9	2j	80	1	3q	81 (13:87)
10	2k	80	2	3r	81 (14:86)
11	2l	80	2	3s	87 (13:87)
12 ^d	2m	80	1	3t	89 (1:~99)

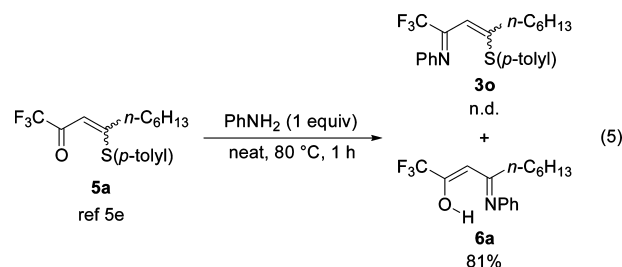
^a Unless otherwise noted, **1a** (0.5 mmol), **2** (0.6–3.0 mmol), Pd(dba)₂ (0.025 mmol), PPh₃ (0.5 mmol) and 1,2-dichloroethane (0.25–0.5 mL) for 1–3 h. In the case of runs 1 and 2, reactions were performed in sealed glass vessels. ^b Isolated yield. ^c Microwave irradiation. ^d Pd(PPh₃)₄ (0.025 mmol). Pr = propyl, MOM = methoxymethyl, THP = 2-tetrahydro-pyranyl.

to this reaction (entry 8, Table 2). Functional groups such as chlorine (**2j**), methoxy carbonyl (**2k**), and 2-tetrahydro-pyranyl (**2l**) aptly furnished the corresponding adducts **3p–s** in high yields as *cis*-rich stereoisomers (entries 9–11, Table 2). In the case of the reaction with 2-methyl-3-butyne-2-ol (**2m**), *Z*-**3t** was selectively obtained in 89% yield (entry 12, Table 2).



Interestingly, the reaction of **1i** containing an N-tosyl group with **2m** selectively produced furan derivative **4a** in 89% yield, indicating that the *cis*-to-*trans* isomerization of CF₃C(=N-Ts)-C(H)=C(C(Me)₂OH)(S-*p*-tolyl) (**3u**) and the following cyclization facily occurred (eq 3).¹⁰ A similar transformation was realized by the treatment of **3t** with AcOH to form **4b** in 84% yield (eq 4).

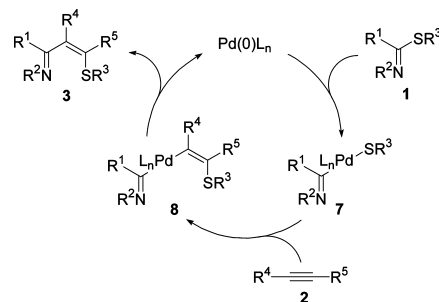
(6) The high reactivity and regioselectivity would be attributed to the oxygen atom at the propargyl moiety in alkynes; see ref 5a.



As mentioned, for the synthesis of **3**, the conversion of CF₃C(O)C(R⁴)=C(R⁵)(SR³) (**5**)^{5c} into **3** is conceivable. Thus, the reaction of **5a** (R³ = *p*-tolyl, R⁴ = H, R⁵ = *n*-C₆H₁₃) with PhNH₂ was attempted (eq 5). The reaction selectively furnished CF₃C(OH)=C(H)C(*n*-C₆H₁₃) (=NC₆H₅) (**6a**) in 81% yield, elucidating that the present Pd-catalyzed iminothiolation of **2** by **1** is complementary to the formation of **6** via **5**.

A plausible mechanism of the present reaction of imino-sulfides (**1**) with alkynes (**2**) is depicted in Scheme 2. The

Scheme 2. Plausible Mechanism for the Pd-Catalyzed Iminothiolation of Alkynes (**2**) Using Iminosulfides (**1**)



oxidative addition of **1** to the Pd(0)L_n complex triggers the reaction to afford PdL_n[C(=NR²)R¹](SR³) (**7**).¹¹ Subsequent regio- and stereoselective insertion of **2** into the Pd–S bond of **7** generates *cis*-PdL_n[C(=NR²)R¹][*Z*-C(R⁴)=C(R⁵)(SR³)] (**Z-8**).¹² Finally, the reductive elimination of **3** from **8** with

(7) When the reaction of **1a** and **2a** was performed at 100 °C in a sealed vessel without a microwave irradiation, **3a** was obtained in a *E* to *Z* ratio of 93:7 with the same yield, indicating that following isomerization of the adducts was partly prevented by the microwave irradiation.

(8) Crystal data for *E*-**3a**: space group *Pbca* (No. 61) with *a* = 15.2284(7) Å, *b* = 17.6487(8) Å, *c* = 17.8733(9) Å, β = 96.385(2)°, *Z* = 8, ρ = 1.298 g/cm³, *R* = 0.0661, and *R*_w = 0.189. See Supporting Information for crystal data for *E*-**3a**.

(9) The NMR spectra of **3** showed temperature dependence due to the fluxional movement of the molecules. The ¹H NMR spectra of *E*-**3a** in toluene-*d*₈ solution at –40 °C appeared as a mixture of two stereoisomers by slow inversion of the N lone pair on the imino group in a ratio of 75:25 (stereochemistry undetermined).

(10) For the utility of furan derivatives, see: Eicher, T., Hauptmann, S. In *The Chemistry of Heterocycles: Structure, Reactions, Syntheses, and Applications*; Wiley-VCH: Weinheim, 2003 and references therein.

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regeneration of Pd(0)L_n completes the catalytic cycle.¹³ The reason that iminosulfide **1** with a CF₃ group exhibited high activity for iminothiolation is presumably due to the facile oxidative addition of a carbon–sulfur bond to the Pd(0) complex.

In conclusion, the present study substantiated the iminothiolation of alkynes with iminosulfides giving rise to the formation of 1-azadiene derivatives, where the introduction of a CF₃ group to the imino-carbon moiety is key to achieving the reaction. Furthermore, the facile syntheses of furan derivatives through 1-azadienes were also presented.

(13) Monitoring the reaction of **1a** with **2h** at room temperature showed that the *cis:trans* ratio of **3o** stayed the same during the course of the reaction, implicating that *trans*-**3** was produced through isomerisation of *cis*-**8** to *trans*-**8** prior to the reductive elimination.

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Supporting Information Available: Experimental procedures and characterization data of new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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