## Synthesis and coordination behaviors of P-stereogenic polymers<sup>†</sup>

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Optically active polymers containing P-stereogenic bisphosphine and 1,2,3-triazole units in the main chain were obtained by copper-catalyzed Huisgen cycloaddition. The repeating unit of the polymer adopted bidentate as well as tetradentate coordination, leading to the change in the conformations of the polymer *via* complexation with transition metals.

The functions of polymers are highly influenced by the higher-ordered structures of molecules and of the polymers themselves. Among the different higher-ordered structures, the one-handed helix, which is observed in naturally occurring macromolecules such as DNA molecules<sup>1</sup> (right-handed double helix) and polypeptides<sup>2</sup> ( $\alpha$ -helix), is a representative example of a second-ordered structure. Their chiral higherordered structures are essential for maintaining biological activity in various organisms. The design and artificial construction of structurally analogous polymers<sup>3</sup> have attracted attention in polymer science; in particular, the control of the higher-ordered structures of polymers by external stimuli has been extensively investigated.<sup>4</sup> Yashima and coworkers found that optically inactive cis-transoid poly(phenylacetylene)s formed one-handed helix structures under the dynamic influence of chiral additives.5 The chiral helix structure was maintained even after the replacement of chiral additives with achiral ones.<sup>5c,d</sup> It is reported that some foldamers enclosed chiral molecules within their cylindrical cavity through hydrophobic interactions to form one-handed helix structures.<sup>6</sup>

Metal ions can serve as a trigger for the conformational change in polymers possessing metal receptors such as crown ether<sup>7</sup> and bipyridine<sup>8</sup> moieties. Recently, we synthesized optically active polymers<sup>9</sup> and oligomers<sup>10</sup> containing phosphorus atoms, which can potentially capture transition metals, by using P-stereogenic bisphosphine compounds as monomers. The addition of transition metals to P-stereogenic polymers causes the polymers to undergo stimulus-induced conformation changes and thus form chiral higher-ordered structures. In this study, we investigate optically active polymers comprising P-stereogenic bisphosphine and 1,2,3-triazole units in the main chain, which exhibit drastic conformational changes in response to transition metal addition.

We incorporated the 1,2,3-triazole skeleton and P-stereogenic bisphosphine into a repeating unit of the polymer to ensure that the repeating unit adopts multidentate coordination when transition metals are added. We designed and synthesized

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Scheme 1 Synthesis of monomer (S,S)-3-BH<sub>3</sub>.

monomer (S,S)-**3**-BH<sub>3</sub> from P-stereogenic bisphosphine (S,S)-**1**-BH<sub>3</sub>,<sup>11</sup> as shown in Scheme 1.

It is noted that the P-stereocenters of bisphosphine retain their configuration during the reaction as well as during the next polymerization reaction. The Huisgen cycloaddition<sup>12</sup> polymerization of (*S*,*S*)-**3**–BH<sub>3</sub> with 9,9-didodecyl-2,7-diethynylfluorene **4** was carried out in the presence of a copper catalyst; this reaction is known as the click reaction.<sup>13</sup> The effect of the catalysts was examined (results listed in Table 1). For successful polymerization, it was important to use an appropriate catalyst combined with a solvent. The combination of the Cu(MeCN)<sub>4</sub>PF<sub>6</sub> complex with CHCl<sub>3</sub> solvent was ineffective even in the presence of tridentate nitrogen ligands, and the low molecular weights of the obtained polymer resulted in the loss





Entry	Catalyst	Ligand	Solvent	$\mathbf{Y}^{b}\left(\%\right)$	$M_n^c$	$M_{ m w}{}^c$
1	Cu(MeCN) <sub>4</sub> PF <sub>6</sub>	$(BimH)_3^d$	CHCl <sub>3</sub>	Trace	3300	3700
2	Cu(MeCN) <sub>4</sub> PF <sub>6</sub>	TBTA <sup>e</sup>	CHCl <sub>3</sub>	21	2200	4300
3	Cu(PPh <sub>3</sub> ) <sub>3</sub> Br		CHCl <sub>3</sub>	45	6100	8000
4	Cu(MeCN) <sub>4</sub> PF <sub>6</sub>		DMF	69	7900	14800
$5^{f}$	Cu(MeCN) <sub>4</sub> PF <sub>6</sub>	_	DMF	75	16 500	42 000

<sup>*a*</sup> Condition: concentration of each monomer = 0.05 M, catalyst (10 mol%), ligand (10 mol%), 50 °C, 6 h. <sup>*b*</sup> Isolated yield. <sup>*c*</sup> Determined by GPC (polystyrene standards, CHCl<sub>3</sub> eluent). <sup>*d*</sup> Tris(2-benzimidazolyl-methyl)amine. <sup>*e*</sup> Tris(benzyltriazolylmethyl)amine. <sup>*f*</sup> Concentration of each monomer = 0.20 M.



Scheme 2 Coordination behaviors of the P-stereogenic polymer.

of the polymer during the purification procedure (entries 1 and 2). The combination of Cu(PPh<sub>3</sub>)<sub>3</sub>Br with CHCl<sub>3</sub> showed good catalytic activity (entry 3) to obtain polymer **5**–BH<sub>3</sub> with a yield of 45% and number-average molecular weight ( $M_n$ ) of 6100 as estimated by gel permeation chromatography (GPC, polystyrene standards, CHCl<sub>3</sub> eluent). The Cu(MeCN)<sub>4</sub>PF<sub>6</sub>/DMF system enhanced the catalytic activity (entries 4 and 5), and a relatively concentrated solution (0.20 M) provided the best result (entry 5).

Coordinated boranes were completely removed by the reaction of **5**–BH<sub>3</sub> with 1,4-diazabicyclo[2.2.2]octane (DABCO) in toluene (Scheme 2). Successive treatment of **5** with PtCl<sub>2</sub>(cod) (cod = 1,5-cyclooctadiene) and Cu(MeCN)<sub>4</sub>PF<sub>6</sub> in DMF afforded the corresponding polymer complexes **5**–Pt and **5**–Cu with yields of 63% and 73%, respectively. The complexations of **5** with Pt and Cu were confirmed by <sup>31</sup>P and <sup>1</sup>H NMR spectra (Fig. 1). In the <sup>31</sup>P NMR spectra (Fig. 1a), the signal of the chiral phosphine in **5**–BH<sub>3</sub> appeared at + 16.1 ppm. This signal was shifted downfield to +42.2 ppm by the complexation with PtCl<sub>2</sub>, and the satellite peak assigned to phosphorus–platinum coupling ( $J_{P-Pt} = 3577$  Hz) was clearly observed. This  $J_{P-Pt}$  value suggests the bidentate coordination of the bisphosphine unit to *cis*-Pt(II)Cl<sub>2</sub>.<sup>14</sup> The broad signal in 5–Cu was observed at –8.1 ppm. The upfield shift of the <sup>31</sup>P NMR signals by the coordination of phosphines to Cu<sup>+</sup>PF<sub>6</sub><sup>-</sup> species was reported,<sup>15</sup> indicating the complexation of P-stereogenic phosphine units to Cu<sup>+</sup>PF<sub>6</sub><sup>-</sup>.

In the <sup>1</sup>H NMR spectra (Fig. 1b), the signal of the 1,2,3triazole proton in **5**–BH<sub>3</sub> appeared at 7.66 ppm. This peak was slightly shifted to 7.78 ppm by the complexation with Pt, while the peak of **5**–Cu was observed at around 8.94 ppm.<sup>16</sup> The <sup>1</sup>H NMR chemical shift of the 1,2,3-triazole proton at around 9 ppm indicates the coordination of the 1,2,3-triazole moiety to  $Cu(1)^+$ .<sup>17</sup> Thus, the P-stereogenic phosphine unit of the polymer can undergo bidentate and tetradentate coordination depending on the metal ion used, *i.e.*, only bisphosphine coordinates to Pt( $\pi$ )Cl<sub>2</sub> and both bisphosphine and 1,2,3triazoles coordinate to Cu(1)<sup>+</sup>. These NMR behaviors were also observed in the case of model compounds (Fig. S22 in ESI†).

Fig. 2 shows the circular dichroism (CD) and ultravioletvisible (UV-vis) absorption spectra of 5-BH<sub>3</sub>, 5-Pt, and 5-Cu recorded in DMF (1.0  $\times$  10<sup>-5</sup> M) at room temperature. The UV-vis absorption spectra of 5-BH<sub>3</sub>, 5-Pt, and 5-Cu exhibited the same absorption band at 280-350 nm; this band was assigned to the  $\pi$ -conjugated triazolyl-fluorenylenetriazolvl unit in the polymer main chain. A relatively weak Cotton effect was observed in the CD spectrum of optically active 5-BH<sub>3</sub> because of the flexible polymer backbone. In the CD spectrum of 5-Pt, the coordination of the P-stereogenic bisphosphine units to Pt(II)Cl<sub>2</sub> resulted in the appearance of a strong Cotton effect in the range 280-350 nm. Coordination to  $Cu(I)^+$  caused a drastic change in the chiroptical property, and the CD profile of 5-Cu was distinct from those of 5-BH<sub>3</sub> and 5-Pt, as shown in Fig. 2. The coordination of the 1,2,3-triazole units in addition to the P-stereogenic centers is expected to contribute to the tetrahedral coordination, which triggers the chiral conformational change. The specific rotations of 5-Pt  $([\alpha]^{25}{}_{\rm D}$  40.2, c 0.5 in CHCl<sub>3</sub>) and **5**-Cu  $([\alpha]^{25}{}_{\rm D}$  8.4, c 0.5 in



Fig. 1 (a) <sup>31</sup>P and (b) <sup>1</sup>H NMR spectra of 5–BH<sub>3</sub> (in CDCl<sub>3</sub>), 5–Pt (in CDCl<sub>3</sub>), and 5–Cu (in DMF-d<sub>7</sub>).



Fig. 2 CD (upper) and UV-vis absorption (lower) spectra of 5–BH<sub>3</sub>, 5–Pt, and 5–Cu in DMF  $(1.0 \times 10^{-5} \text{ M})$  at room temperature.

CHCl<sub>3</sub>) were relatively small, implying that the Cotton effect of the  $\pi$ -conjugated unit in the polymer arises from the chirality around the metal-complex moiety and the conformational change does not induce the chiral higher-ordered structure of the whole polymer chain due to the free rotation of the triazolyl-fluorenylene-triazolyl unit.

In summary, we synthesized optically active polymers containing P-stereogenic bisphosphine and 1,2,3-triazole moieties in the repeating unit by Huisgen cycloaddition polymerization. We demonstrated the drastic conformational change in the polymer *via* bidentate and tetradentate coordination to Pt(II) and Cu(I), respectively. Further investigations on the higher-ordered-structure changes of P-stereogenic polymers by the metal coordination that give rise to the colorimetric and/or fluorometric detection of metal ions are currently underway.

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