

SHORT
COMMUNICATIONS

Pyrolysis of Enaminone Derivatives of 1,2-Ethylenediamine

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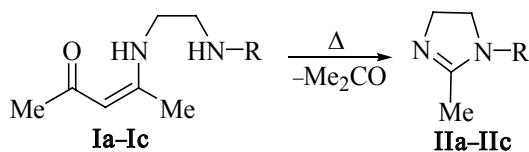
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Enaminones are applied as synthons in the preparation of heterocyclic compounds [1, 2]. Their properties depend essentially on the character of the amine fragment of the molecule. In this study we showed that the enaminones obtained from acetylacetone and monosubstituted 1,2-ethylenediamines at 200–250°C suffer a thermal cleavage to give acetone and N-substituted imidazolines.

The reaction apparently proceeds along a mechanism similar to that of the ester cleavage of β -dicarbonyl compounds.



Enaminones **Ia–Ic** were heated at 200–250°C and pressure of 15 mm Hg collecting a wide fraction of the decomposition products. Imidazolines **IIa–IIc** were separated by subsequent distillation.

2-(2-Methylimidolin-1-yl)ethanol (IIa). Yield 65%, bp 166–170°C (10 mm Hg), mp 40–41°C, n_D^{20} 1.514 (of supercooled liquid). IR spectrum, ν , cm^{−1}: 3290, 3079, 2958, 2932, 2776, 1667, 1605, 1499, 1454, 1433, 1362, 1316, 1268, 1203, 1177, 1114, 1078, 1054, 1013, 976, 940, 866, 814, 679, 602, 590, 512, 418. ¹H NMR spectrum, δ , ppm: 1.84 s (3H, CH₃), 3.15 t (2H, NCH₂CH₂O), 3J 5.4 Hz), 3.30 m (2H, NCH₂), 3.56 m (4H, CH₂N=, CH₂O), 5.78 br.s (1H, OH). ¹³C NMR spectrum (CDCl₃)

δ , ppm: 14.10 (CH₃), 49.29 (NCH₂CH₂O), 49.89 (NCH₂), 51.46 (CH₂N=), 59.17 (OCH₂), 165.10 (C=N). Found, %: C 56.34; H 9.53; N 21.71. C₆H₁₂N₂O. Calculated, %: C 56.22; H 9.44; N 21.86.

1-(2-Vinyloxyethyl)-2-methylimidazoline (IIb).

Yield 74%, oily substance, bp 90–92°C (3 mm Hg), d_4^{20} 0.9840, n_D^{20} 1.4942. IR spectrum, ν , cm^{−1}: 3105, 3030, 2915, 2855, 1620, 1605, 1475, 1410, 1350, 1300, 1250, 1180, 1105, 1065, 995, 950, 920, 805, 685, 575, 475. ¹H NMR spectrum, δ , ppm: 1.89 s (3H, CH₃), 3.33 m (4H, CH₂NHCH₂), 3.62 m (2H, CH₂N=), 3.72 t (2H, CH₂O, 3J 5.4 Hz), 3.99 d.d (1H, *cis*-HC=CO, $^2J_{\text{gem}}$ 2.2, $^3J_{\text{cis}}$ 6.7 Hz), 4.14 d.d (1H, *trans*-HC=CO, $^2J_{\text{gem}}$ 2.2, $^3J_{\text{trans}}$ 14.3 Hz), 6.41 d.d (1H, OCH=C, $^3J_{\text{cis}}$ 6.7, $^3J_{\text{trans}}$ 14.3 Hz). ¹³C NMR spectrum (CDCl₃), δ , ppm: 13.97 (CH₃), 45.67 (NCH₂CH₂O), 49.86 (NCH₂), 51.87 (CH₂N=), 65.37 (OCH₂), 86.53 (=CH₂), 150.99 (OCH=), 163.37 (C=N). Found, %: C 62.47; H 9.23; N 18.05. C₈H₁₄N₂O. Calculated, %: C 62.31; H 9.15; N 18.17.

3-Vinyloxy-1-(2-methylimidolin-1-yl)-2-propanol (IIc). Yield 69%, oily substance, bp 155–158°C (2 mm Hg), n_D^{20} 1.5117. IR spectrum, ν , cm^{−1}: 3309, 3116, 3080, 2928, 2868, 2745, 2582, 1664, 1613, 1494, 1431, 1376, 1321, 1269, 1201, 1117, 1077, 1021, 1004, 975, 943, 819, 746, 704, 647, 595, 554, 470. ¹H NMR spectrum, δ , ppm: 1.89 s (3H, Me), 3.19–3.44 m (4H, NCH₂CH₂N=), 3.60 m (2H, CHCH₂N), 3.67 m (2H, OCH₂), 3.95 m (2H, CHOH), 4.01 d.d (1H, *cis*-HC=CO, $^2J_{\text{gem}}$ 2.1, $^3J_{\text{cis}}$ 6.7 Hz), 4.17 d.d (1H, *trans*-HC=CO,

$^{2}J_{\text{gem}}$ 2.1, $^{3}J_{\text{trans}}$ 14.2 Hz), 6.46 d.d (1H, OCH=C, $^{3}J_{\text{cis}}$ 6.7, $^{3}J_{\text{trans}}$ 14.2 Hz). ^{13}C NMR spectrum (CDCl_3), δ , ppm: 13.89 (Me), 49.80 (CHCH_2N), 50.62 ($\text{NCH}_2\text{CH}_2\text{N=}$), 51.32 ($\text{NCH}_2\text{CH}_2\text{N=}$), 67.04 (OCH₂), 69.25 (CHOH), 86.76 (=CH₂), 151.29 (OCH=), 164.87 (NC=N). Found, %: C 58.97; H 8.18; N 15.27. $\text{C}_9\text{H}_{16}\text{N}_2\text{O}_2$. Calculated, %: C 58.67; H 8.75; N 15.21.

REFERENCES

1. *The Chemistry of Enamines*, Rappoport, Z., Ed., Chichester: Wiley, 1994, part, 1, 1683, p.
2. Freimanis, Ya.F. *Khimiya enaminoketonov, enaminoiminov, enaminotionov* (Chemistry of Enaminoketones, Enaminoimines, Enaminotiones), Riga: Zinatne, 1974.