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Investigation of 4-piperidinols as novel H₃ antagonists

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ABSTRACT

Compounds containing a substituted 4-piperidinol core have been found to be potent antagonists of the human H₃ receptor. The compounds exhibited up to a 60-fold preference for inhibiting the human H₃ receptor over the mouse and showed a low binding affinity for the hERG channel.

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The histamine H₃ receptor is a G-protein coupled receptor that is primarily located in the CNS and regulates the synthesis and release of the neurotransmitter histamine via a negative-feedback mechanism. H₃ receptor activation also plays a role in the release of several other neurotransmitters in the CNS, including dopamine, serotonin, GABA, and acetylcholine, and is therefore suspected to possess promising therapeutic potential. Sought after indications for H₃ modulators include treatments for Alzheimer's disease,¹ attention-deficit hyperactivity disorder (ADHD), cognition,² and obesity.³ Drug discovery efforts to modulate the histamine H₃ receptor have been ongoing in both academia and the pharmaceutical industry since its discovery in 1983.⁴ Although significant progress has been made in finding compounds that effectively target the H₃ receptor, no candidates have yet received clinical approval. However, there are presently several compounds being evaluated in early to late-stage clinical trials.⁵

Early small molecule research efforts targeting the H₃ receptor revealed that imidazole-containing compounds such as clobenpropit showed very potent antagonistic activity. However, these compounds showed metabolic liabilities and poor CNS penetration likely attributable to the imidazole moiety. Potent H₃ antagonists were eventually discovered that lacked the imidazole core and exhibited an improved CNS and metabolic profile.⁶ A common feature present in these 'non-imidazole' H₃ modulators is a basic amine group that is tethered to an aryl system via an alkyl chain

having a high degree of rotational freedom as in JNJ-5207852⁷ (Fig. 1). However, there are examples of potent H₃ modulators in the literature that have reduced conformational flexibility⁸ including a rigidified analog of JNJ-5207852.⁹ Generally, this is a desirable physico-chemical property as it has been described that reducing the number of rotatable bonds tends to improve oral bioavailability.¹⁰ After a high-throughput screening campaign, we were pleased to find that the relatively rigid and racemic compound **1**¹¹ was a potent antagonist of the human H₃ (hH₃) receptor (Table 1). Interestingly, compound **1** showed a significant difference in IC₅₀ potencies between mouse and human H₃ in the FLIPR assays.¹² However, the control H₃ antagonist clobenpropit did not

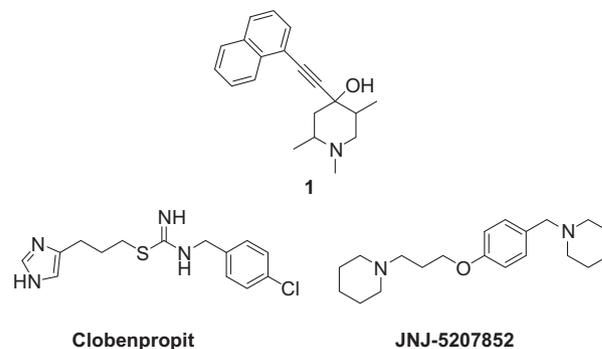


Figure 1. Structure of lead compound **1**, imidazole and non-imidazole based H₃ antagonists.

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Table 1
Potency of piperidinols against mouse and human H₃

Compound ^a	Ar	R ¹	R ²	R ⁵	FLIPR IC ₅₀ ^b (nM)		hERG
					mH ₃	hH ₃	
Clobenpropit					1	1	nd
1	A	Me	Me	Me	280 (±92) ^c	11 (±2) ^c	nd
5	A	Me	Me	Me	119 (±19) ^c	9 (±3) ^c	15% @ 50 μM
<i>ent</i> - 5	A	Me	Me	Me	1318 ^d	41 ^e	nd
1 (O-acetate)	A	Me	Me	Me	>10,000 ^e	5508 ^e	nd
6	B	Me	Me	Me	570 ^d	20 ^d	20% @ 10 μM
7	C	Me	Me	Me	480 (±300) ^c	17 ^d	15% @ 10 μM
8	C	Me	Me	Me	85 ^d	2 ^d	16% @ 10 μM
9	D	Me	Me	Me	810 ^e	84 ^e	nd
10	E	Me	Me	Me	122 ^e	2 ^e	0% @ 10 μM
13	A	(-CH ₂) ₃		Me	280 ^e	5 ^e	23% @ 10 μM
14	C	(-CH ₂) ₃		Me	345 ^e	14 ^e	nd
15	F	Me	Me	Me	690 ^e	12 ^e	3% @ 10 μM
16	G	Me	Me	Me	9730 ^e	709 ^e	nd

^a Compounds **5**, *ent*-**5**, and **8** are single enantiomers. All other compounds are racemic.

^b Measured reduction of intracellular calcium flux induced by agonist (*R*)- α -methylhistamine.

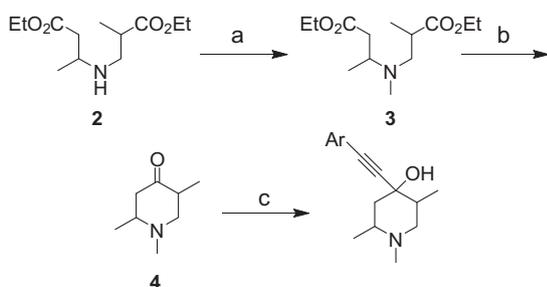
^c Mean of at least three experiments, standard error of the mean.

^d Average of two determinations.

^e Single determination.

show a bias in potency toward either species. In addition to its high potency as a lead compound against hH₃, the unique structure of **1** generated interest in further exploration of this scaffold for improved H₃ receptor antagonism.

The general synthesis of compounds **1** and **5–10** is described in Scheme 1. The piperidone intermediate **4** was synthesized using a modification of a previously reported procedure.¹³ Compound **2**, prepared from ethyl 3-aminobutyrate and ethyl methacrylate, was subjected to Eschweiler–Clarke conditions to give the N-methylated product **3**. Dieckmann cyclization of **3** and subsequent decarboxylation gave the piperidone **4** as an unresolved mixture of diastereomers. Final compounds were accessed via deprotonation of aryl acetylenes using BuLi followed by addition to piperidone **4**. There was concern that the creation of a third chiral center would yield a mixture of piperidinol diastereomers that may be difficult to purify. However, the major diastereomer



Scheme 1. Reagents and conditions: (a) (i) (CH₂O)_n, Toluene-*n*BuOH; then (ii) HCO₂H, reflux; (b) (i) Na, EtOH, xylenes, reflux; (ii) 20% aq HCl, reflux; (c) Ar-CCH, *n*BuLi, THF, 0 °C.

formed in the addition reaction was easily isolated by silica gel chromatography.

A classical resolution was used to isolate each enantiomer of pure diastereomer **1**. Re-crystallization of the salt formed from optically pure di-*p*-toluoyl-(*L*)-tartaric acid gave an enantiomer having >95% de.¹⁴ X-ray crystal structure determination of the salt (Fig. 2) provided the absolute stereochemistry of **5**.¹⁵ The crystal structure also confirmed the mode of aryl acetylene addition as being axial. The free base of **5** showed a 2- to 3-fold improvement in potency against mH₃ over the racemate **1**, but no improvement in hH₃ potency. The antipode *ent*-**5** was isolated from recrystallization of the di-*p*-toluoyl-(*D*)-tartrate salt of **1** and was found to be much less potent than **5** against both mH₃ and hH₃. Another observation was that the free 4-hydroxyl group was important for potency as the corresponding acetate of **1** was not active against mH₃ and very weakly active against hH₃ at the highest concentrations tested (Table 1).

The *trans*-alkene **6** was obtained after lithium aluminum hydride reduction of alkyne **1** and showed a loss in activity against both mH₃ and hH₃. This result was not surprising given the altered spatial arrangement of the naphthyl ring. The alkene was further reduced to alkane **7** using standard Pd-catalyzed hydrogenation conditions. We anticipated a recovery in H₃ potency for **7** due to the increased conformational flexibility of the ethyl chain, but no improvements were seen over the alkene. However, upon chiral resolution of **7**, as previously described, enantiomer **8** was found to possess excellent potency against hH₃.

Altering the position of the alkyne moiety to the 2'-position of the naphthyl ring gave compound **9** that was less potent than the corresponding 1'-substituted analog **1** in both mH₃ (~3-fold) and hH₃ (~8-fold). However, the opposite trend was observed upon reduction of alkyne **9** to the alkane **10**. Comparison of

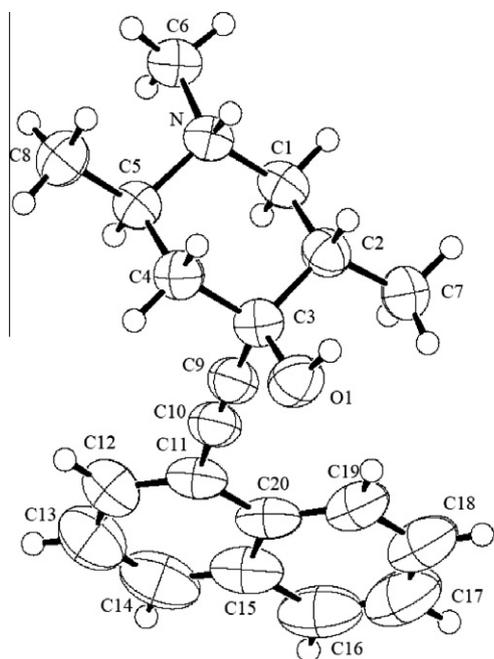


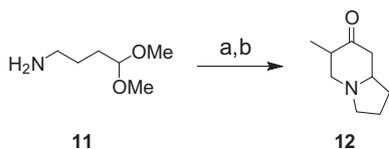
Figure 2. X-ray structure of **5**. Di-*p*-toluoyl-(*l*)-tartaric acid counter ion not shown for clarity.

reduced compounds **10** and **7** showed that 2'-substitution was more potent than the corresponding 1'-substitution. Although it was satisfying that mH₃ activity for **10** improved over compound **9**, it was not nearly as dramatic as the improvement in hH₃ potency.

A more rigid indolizidine ring system was also explored that was conveniently prepared from 4-aminobutyraldehyde dimethylacetal (**Scheme 2**) using a modified literature procedure.¹⁶ Addition of the lithium salt of **A** to **12** gave a mixture of diastereomers. The major diastereomer **13** was isolated and was found to be the most potent constituent of the mixture and showed similar potencies against mH₃ and hH₃ as its acyclic counterpart **1** with preference for hH₃. Interestingly, the isolated minor diastereomer was able to be predominantly equilibrated to **13** (~4:1) by exposure to KOH/EtOH at 50 °C.¹⁷ Following the same protocol as discussed previously, the alkyne was fully reduced to give compound **14** which showed similar potency to its counterpart **7**.

A sensitive structure–activity relationship was seen for quinoline isomers **15** and **16**. Alkyne substitution at the 5-position of the quinoline ring (**15**) showed similar activity against both mH₃ and hH₃ as the corresponding naphthyl series (**A–E**). However, alkyne substitution at the 8-position of the quinoline ring (**16**) showed a dramatic decrease in H₃ activity.

Historical H₃ modulators had showed a tendency to bind to the hERG ion channel which represents a potential safety liability.¹⁸ Recently however, this problem has been successfully addressed.¹⁹ As part of our compound safety profiling, H₃ antagonists of interest were evaluated for their potential to inhibit the hERG ion channel and all compounds tested showed minimal inhibition of hERG.



Scheme 2. Reagents and conditions: (a) 3-methyl-3-butene-2-one, MeOH, 0 °C; (b) 5% aq AcOH, reflux.

In summary, we have identified a unique piperidinol-based pharmacophore that shows potent human H₃ inhibition and a very good overall hERG profile. However, this series showed a significant disparity in potency between the human and mouse H₃ receptors as only moderate potency was achieved against the mouse. The lack of mH₃ potency was disappointing because in vivo assays were to be performed in mouse. This result is somewhat surprising given the reported high H₃ receptor homology (94%) between mouse and human.²⁰ A few antagonists have been reported that are biased toward hH₃ over mH₃ albeit to a much lesser extent than our observations with the piperidinol series.²¹ In our case, the observed differences in potency between the two species appear to result from the human H₃ receptor being much more accommodating for the piperidinol pharmacophore.

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