

## Low-valent Niobium-mediated Synthesis of Indenes: Intramolecular Coupling Reaction of CF<sub>3</sub> Group with Alkene C–H Bond

Kohei Fuchibe, Ken Mitomi, and Takahiko Akiyama\*

*Department of Chemistry, Faculty of Science, Gakushuin University, 1-5-1 Mejiro, Toshima-ku, Tokyo 171-8588*

(Received September 7, 2006; CL-061037; E-mail: takahiko.akiyama@gakushuin.ac.jp)

CF<sub>3</sub> group of *o*-alkenyl- $\alpha,\alpha,\alpha$ -trifluorotoluenes underwent intramolecular coupling reaction with the alkene C–H bond under NbCl<sub>5</sub>/LiAlH<sub>4</sub> system. Substituted indenes were obtained in good yields.

Indene is a ubiquitous component of organic molecules and plays important roles in various fields of chemistry. For instance, substituted indene derivatives frequently occur in natural products and exhibit biological activity.<sup>1</sup> Anions prepared from indene derivatives and appropriate bases are representative variations of Cp ligands, and the  $\pi$ -coordinated indenyl complexes sometimes exhibit remarkable catalytic activities.<sup>2</sup> In recent years, indene skeletons have attracted much attention as core parts of functional materials such as conducting polymers<sup>3</sup> and discotic liquid crystals.<sup>4</sup> Development of novel synthetic method of the indene derivatives is, thus, extremely important to these areas of research.

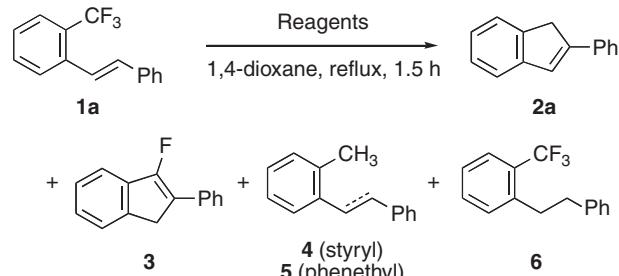
A number of methods have been reported for the construction of the five-membered carbocycles. Cation-mediated reactions such as the Nazarov-type  $\pi$ -cyclization and intramolecular electrophilic aromatic substitution are one of the most familiar methods.<sup>5</sup> Intramolecular nucleophilic attack of carbanion equivalents to carbonyl groups affords the indene or its related compounds.<sup>6</sup> Insertions of alkynes to aryl metals also give indene derivatives.<sup>7,8</sup>

On the other hand, we have previously reported a low-valent niobium-mediated synthesis of substituted fluorenes from *o*-aryl- $\alpha,\alpha,\alpha$ -trifluorotoluenes.<sup>9</sup> In this protocol, the CF<sub>3</sub> group is activated by the low-valent niobium,<sup>10</sup> and coupled with the neighboring aromatic C–H bond.<sup>11</sup> Keeping this success in mind, we planned to synthesize indene derivatives from *o*-alkenyl- $\alpha,\alpha,\alpha$ -trifluorotoluenes (Scheme 1).

At the outset, we attempted a reaction of *o*-styryl- $\alpha,\alpha,\alpha$ -trifluorotoluene (**1a**) (Table 1). To a dioxane solution of **1a** and an equimolar amount of NbCl<sub>5</sub>, dioxane suspension of LiAlH<sub>4</sub> was added over a period of 1.5 h. During the addition, the reaction mixture was allowed to reflux. When the addition was completed, the reaction was quenched with water. The desired indene **2a** was obtained in 64% yield and conventional reduction products (hydrodefluorination products) **4** and **5** were obtained in 22% yield (Entry 1).

Use of other reagent in place of NbCl<sub>5</sub> or LiAlH<sub>4</sub> gave inferior results: TaCl<sub>5</sub> partially gave fluoroindene **3**,<sup>12</sup> in which one fluorine atom survived (Entry 2). WCl<sub>6</sub> gave significant amount

**Table 1.** Optimization of reagents



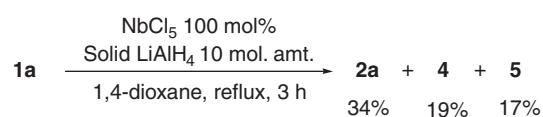
Entry	Reagents <sup>a</sup>	Yield/%				
		<b>2a</b>	<b>3</b>	<b>4 + 5</b> (ratio)	<b>6</b>	<b>1a</b>
1	NbCl <sub>5</sub> , LiAlH <sub>4</sub> <sup>b</sup>	64	—	22 (55:45)	—	—
2	TaCl <sub>5</sub> , LiAlH <sub>4</sub> <sup>b</sup>	42	11	23 (56:44)	—	—
3	WCl <sub>6</sub> , LiAlH <sub>4</sub> <sup>b</sup>	16	18	12 (83:17)	—	39
4	PdCl <sub>2</sub> , LiAlH <sub>4</sub> <sup>b</sup>	4	7	16 (76:24)	44	7
5	NbCl <sub>5</sub> , Red-Al	11	6	71 (18:82)	—	—
6	NbCl <sub>5</sub> , DIBAL	—	—	—	—	86
7	NbCl <sub>5</sub> , NaBH <sub>4</sub>	—	—	—	—	97

<sup>a</sup>100 mol % of metal salt and 10 molar amounts of reducing agent were used for each entries. <sup>b</sup>Dioxane suspension of LiAlH<sub>4</sub> was added over 1.5 h.

of the starting material **1a** (Entry 3). In the case of PdCl<sub>2</sub>, the alkene moiety was affected prior to the CF<sub>3</sub> group (Entry 4) to give **6**.<sup>13</sup> Use of Red-Al [sodium bis(methoxyethoxy)aluminum hydride], DIBAL, and NaBH<sub>4</sub> resulted in formation of the conventional reduction products **4** and **5** (Entry 5), or recovery of **1a** (Entries 6 and 7). The intramolecular coupling reaction was thus accomplished by the NbCl<sub>5</sub>/LiAlH<sub>4</sub> system.

Solid LiAlH<sub>4</sub>, in place of the dioxane suspension of the reducing agent, decreased the yield of **2a** and increased the yield of **4** and **5** (Scheme 2). Excess amount of the reducing agent in the reaction medium likely promoted conventional reduction process, leading to the formation of **4** and **5**.<sup>9a</sup>

The use of a catalytic amount (30 mol %) of NbCl<sub>5</sub> was found to be effective for the present coupling reactions. As shown in Table 2, arylvinyl- $\alpha,\alpha,\alpha$ -trifluorotoluenes gave arylindenes in good yields (Entries 1–11). It is worth noting that this reaction smoothly gave aminoindene **2i** and fused **2k**, which are potentially inaccessible by existing acid-promoted cyclizations<sup>5a,5b</sup> or the Heck-type arylations of parent indenes.<sup>14</sup>



**Scheme 1.** Our approach to indene skeletons.

**Scheme 2.**

**Table 2.** Low-valent niobium-mediated synthesis of substituted indenes<sup>a</sup>

Entry	Alkenyltrifluorotoluene	(R)	Product(s)	Yield/% <sup>b</sup>
1		<b>1a</b> (H)		64, <b>2a</b>
2		<b>1b</b> ( <i>p</i> -Me)		61, <b>2b</b>
3		<b>1c</b> ( <i>p</i> -NMe <sub>2</sub> )		64, <b>2c</b>
4		<b>1d</b> ( <i>p</i> -OMe)		66, <b>2d</b>
5		<b>1e</b> ( <i>m</i> -OMe)		62, <b>2e</b>
6		<b>1f</b> ( <i>o</i> -OMe)		57, <b>2f</b>
7		<b>1g</b>		62, <b>2g</b>
8		<b>1h</b>		64, <b>2h</b>
9 <sup>c</sup>		<b>1i</b>		61, <b>2i</b>
10		<b>1j</b>		61, <b>2j</b>
11		<b>1k</b>		68, <b>2k</b>

<sup>a</sup>Conditions: NbCl<sub>5</sub> 30 mol %, LiAlH<sub>4</sub> 5 mol. amt. (dioxane suspension, over 1.5 h), 3 h. <sup>b</sup>NMR yield. Conventional reduction products corresponding to **4** and **5** were obtained in 19–24% yields. <sup>c</sup>NbCl<sub>5</sub> 100 mol %, LiAlH<sub>4</sub> 10 mol. amt., 10 h.

We now surmise that this reaction proceeds as we have previously described for our fluorene synthesis.<sup>9a</sup> Further studies on mechanistic aspects are in progress.

In summary, we have developed a novel low-valent niobium-mediated synthetic method of indenes. We could obtain various substituted indenes in good yields, starting from *o*-alkenyl- $\alpha,\alpha,\alpha$ -trifluorotoluenes via intramolecular coupling reaction of the CF<sub>3</sub> group and the alkene C–H bond.

This work was supported in part by Grant-in-Aid for Scientific Research from the Ministry of Education, Culture, Sports, Science and Technology, Japan (No. 18750036).

Dedicated to Professor Teruaki Mukaiyama on the occasion of his 80th birthday.

## References and Notes

- 1 a) A. Korte, J. Legros, C. Bolm, *Synlett* **2004**, 2397. b) D. C. Harrowven, N. A. Newman, C. A. Knight, *Tetrahedron Lett.* **1998**, 39, 6757. c) Y. Ishiguro, K. Okamoto, F. Ojima, Y. Sododa, *Chem. Lett.* **1993**, 1139. d) Y. Wang, S.-Y. Mo, S.-J. Wang, S. Li, Y.-C. Yang, J.-G. Shi, *Org. Lett.* **2005**, 7, 1675.
- 2 a) R. Leino, P. Lehmus, A. Lehtonen, *Eur. J. Inorg. Chem.* **2004**, 3201. b) V. Cadierno, J. Díez, M. P. Gamasa, J. Gimeno, E. Lastra, *Coord. Chem. Rev.* **1999**, 193–195, 147. c) D. Zargarian, *Coord. Chem. Rev.* **2002**, 233–234, 157.
- 3 J. Yang, M. V. LakshmiKantham, M. P. Cava, *J. Org. Chem.* **2000**, 65, 6739.
- 4 J. Barberá, O. A. Rakitin, M. B. Ros, T. Torroba, *Angew. Chem., Ind. Ed.* **1998**, 37, 296.
- 5 a) M. K. Seery, S. M. Draper, J. M. Kelly, T. McCabe, T. B. H. McMurry, *Synthesis* **2005**, 470. b) Z. Xi, R. Guo, S. Mito, H. Yan, K.-i. Kanno, K. Nakajima, T. Takahashi, *J. Org. Chem.* **2003**, 68, 1252. c) P. G. Karmarkar, V. R. Chinchore, M. S. Wadia, *Synthesis* **1981**, 228. d) K. R. Romines, K. D. Lovasz, S. A. Mitzak, J. K. Morris, E. P. Seest, F. Han, J. Tulinsky, T. M. Judge, R. B. Gammill, *J. Org. Chem.* **1999**, 64, 1733.
- 6 a) R. L. Halterman, C. Zhu, *Tetrahedron Lett.* **1999**, 40, 7445. b) V. G. Nenajdenko, D. I. Denisenko, E. S. Balenkova, *Russ. Chem. Bull.* **2002**, 51, 2090.
- 7 a) L. G. Quan, V. Gevorgyan, Y. Yamamoto, *J. Am. Chem. Soc.* **1999**, 121, 3545. b) K.-J. Chang, D. K. Rayabarapu, C.-H. Cheng, *J. Org. Chem.* **2004**, 69, 4781. c) A. A. Pletnev, Q. Tian, R. C. Larock, *J. Org. Chem.* **2002**, 67, 9276. d) Y. Kuniinobu, A. Kawata, K. Takai, *J. Am. Chem. Soc.* **2005**, 127, 13498.
- 8 See also: a) H. Yamabe, A. Mizuno, H. Kusama, N. Iwasawa, *J. Am. Chem. Soc.* **2005**, 127, 3248. b) R. Shintani, K. Okamoto, T. Hayashi, *J. Am. Chem. Soc.* **2005**, 127, 2872. c) M. F. Semmelhack, S. Ho, D. Cohen, M. Steigerwald, M. C. Lee, G. Lee, A. M. Gilbert, W. D. Wulff, R. G. Ball, *J. Am. Chem. Soc.* **1994**, 116, 7108. d) T. Fukuyama, N. Chatani, F. Kakiuchi, S. Murai, *J. Org. Chem.* **1997**, 62, 5647. e) T. Jobashi, K. Maeyama, K. Noguchi, Y. Yoshida, N. Yonezawa, *Bull. Chem. Soc. Jpn.* **2006**, 79, 627. f) W. A. L. van Otterlo, E. M. Coyanis, J.-L. Panayides, C. B. de Koning, M. A. Fernandes, *Synlett* **2005**, 501.
- 9 a) K. Fuchibe, T. Akiyama, *J. Am. Chem. Soc.* **2006**, 128, 1434. See also: b) K. Fuchibe, T. Akiyama, *Synlett* **2004**, 1282.
- 10 Reports on activation of CF<sub>3</sub> group: a) V. J. Scott, R. Çelenligil-Çetin, O. V. Ozerov, *J. Am. Chem. Soc.* **2005**, 127, 2852. b) H. Amii, Y. Hatamoto, M. Seo, K. Uneyama, *J. Org. Chem.* **2001**, 66, 7216. c) C. Sabouret, M. Troupel, S. Sible, J. Périchon, *J. Chem. Soc., Chem. Commun.* **1989**, 1138. d) P. Clavel, M.-P. Léger-Lambert, C. Biran, F. Serein-Spirau, M. Bordeau, N. Roques, H. Marzouk, *Synthesis* **1999**, 829. General review on activation of inactive bonds: e) Activation of Unreactive Bonds and Organic Synthesis, ed. by S. Murai, Topics in Organometallic Chemistry, Springer, Berlin, **1999**, Vol. 3.
- 11 Other reports on C–C bond formation via alkene C–H activation: a) M. D. K. Boele, G. P. F. van Strijdonck, A. H. M. de Vries, P. C. J. Kamer, J. G. de Vries, P. W. N. M. van Leeuwen, *J. Am. Chem. Soc.* **2002**, 124, 1586. b) Y.-G. Lim, J.-B. Kang, Y. H. Kim, *Chem. Commun.* **1996**, 585. For general review, see reference 10e.
- 12 1-Fluoro-2-phenylindene was not detected at all.
- 13 This selectivity is of great interest, but the reason is not clear at this moment.
- 14 For instance: I. E. Nifant'ev, A. A. Sitnikov, N. V. Andriukhova, I. P. Laishevsev, Y. N. Luzikov, *Tetrahedron Lett.* **2002**, 43, 3213.