α-ACYLAMINORADICAL ANNELATION IN THE DIASTEREOSELECTIVE SYNTHESIS OF 1- AND 5-SUBSTITUTED TETRAHYDROPYRROLO[1,2-c]OXAZOLE AND 1-SUBSTITUTED PYRROLIZIDINE DERIVATIVES

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Diastereoselective synthesis of $1,8-\underline{\text{trans}}-1$ -substituted and $5,8-\underline{\text{trans}}-5$ -substituted tetrahydropyrrolo[1,2-c]oxazoles was achieved by an application of α -acylaminoradical cyclization at a silylated triple bond. The method was applied to an enantioselective synthesis of the 1,8-trans-oriented 1-oxygenated pyrrolizidine.

Radical cyclization is rapidly becoming an important method for the construction of bicyclic systems.^{1,2)} α -Acylaminoradical cyclization at an unsaturated component was also applied to a synthesis of N-heterocycles.¹⁾ Our interest in the efficient synthesis of functionallized heterocyclic systems led us to develop an annelation of α -acylaminoradicals, reported as a highly regiospecific reaction,²⁾ for a diastereoselective synthesis of pyrrolidine and pyrrolizidine derivatives.

First, we prepared 4-phenylthiooxazolidin-2-ones $(\underline{3a}-\underline{e})$ and 2-phenylthiopyrrolidin-5-one (9) utilized for a generation of the corresponding α -acylaminoradical species. Reduction of $\underline{1a}-\underline{e}$, obtained by condensation of 5-substituted oxazolidine-2,4-diones with silylated alcohols by Mitsunobu's method,^{3,4)} with NaBH₄ (methanol, 0 °C) afforded <u>2a-e</u>, respectively. The imide (7), obtained by starting with (<u>S</u>)-malic acid through the imide (<u>6</u>),⁵) was also reduced to <u>8</u>. Conversion of <u>2a-e</u> and <u>8</u> to <u>3a-e</u> and <u>9</u>, respectively, was carried out by an application of the modified Walker's method⁶ (PhSSPh, n-Bu₂P, benzene, room temperature, 9 h).

Benzene solution of $\underline{3a} - \underline{e}$ (0.01 M solution) was heated in the presence of trin-butyltin hydride (1.3 equiv.) and AIBN by the usual way¹) to give the corresponding exo-cyclization products⁷) as a mixture of <u>E</u>- and <u>Z</u>-isomers (<u>4a</u>, 75%; <u>4b</u>, 70%; <u>4c</u>, 32%; <u>4d</u>, 74%; and <u>4e</u>, 72% yield, respectively) as an oil in all cases. Desilylation of <u>4a-e</u> with CF₃COOH-CH₂Cl₂ (1:2; room temperature, 9 h) gave quantitative yields of <u>5a-e</u> as a single diastereomer, respectively, as an oil except <u>5a</u>, mp 63-64 °C and <u>5d</u>, mp 64-65 °C. The stereochemical assignments for <u>5a-c</u> were based on the comparison of the ¹H NMR spectrum of <u>5a</u> with that of <u>5d</u>. The ¹H NMR spectrum of <u>5d</u> showed two singlets at δ 1.27 and 1.54 due to two CH₃ groups at 1position. The higher signal was attributable to the <u>trans</u>-oriented CH₃ in regard to 8-H owing to the shielding effect of 7-exo-double bond and the lower signal was assigned to the <u>cis</u>-oriented CH₃. The ¹H NMR spectrum of <u>5a</u> showed only one doublet at δ 1.54 (J=6 Hz). These facts indicate that 1-H of <u>5a-c</u> is <u>trans</u>-oriented in regard to 8-H. The magnitude of J_{1,8}(=4.5 Hz), clearly visible in the ¹H NMR spectra of <u>5a-c</u>, would also support this assignment. The ¹H NMR spectra of <u>5a-e</u> exhibited characteristic signals due to the <u>trans</u>-oriented 5-H in regard to 8-H at around δ 3.80-4.02; these signals were lower than those due to the <u>cis</u>-oriented 5-H, appeared at δ 2.90-3.18, because of the deshielding effect of carbonyl at 3-position. The ¹H NMR spectrum of <u>5e</u> showed only lower signals due to 5-H at around δ 4.00. Based on these facts, 5-CH₃ (δ 1.22, d, J=7 Hz) of <u>5e</u> was found to take <u>cis</u> relationship with 8-H. This radical cyclization was applied to the enantioselective synthesis of <u>11</u>, which would be a useful precursor for an enantioselective synthesis of antitumor pyrrolizidine alkaloids such as hastanecine and retronecine.⁸ Treatment of <u>9</u> with tri-n-butyltin hydride as above gave <u>10</u> as an oil of a 3:1 mixture of <u>E</u>- and <u>Z</u>-isomers in 60% yield. Desilylation of <u>10</u> afforded <u>11</u> as a single oily product in 95% yield, $[\alpha]_{2}^{23}$ -36.4° (c 0.10, methanol).

Thus, these ring formations were found to proceed with remarkable diastereoselectivity in regard to the orientation of the substituent.



<u>d</u>: $R_1=R_2=Me$, $R_3=H$; <u>e</u>: $R_1=R_2=H$, $R_3=Me$



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(Received February 12, 1986)