The Absolute Stereochemistry of Pamamycin-607, an Aerial Mycelium-inducing Substance of *Streptomyces alboniger*

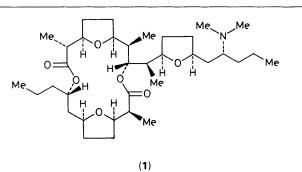
Masahiro Natsume, Satoru Kondo, and Shingo Marumo*

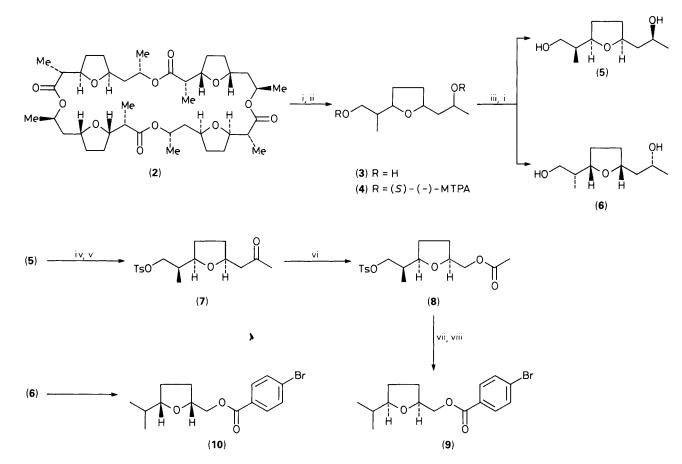
Department of Agricultural Chemistry, Nagoya University, Nagoya 464-01, Japan

The absolute stereochemistry of pamamycin-607 has been determined as (1), by derivatizing it to a furfuryl compound (14), the optical rotation of which coincided with that of an authentic compound derived chemically from nonactin antibiotic.

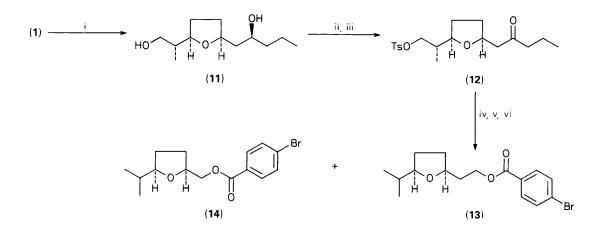
Pamamycin-607 is an aerial mycelium-inducing substance isolated from *Streptomyces alboniger* IFO 12738.¹ It also acts as an antibiotic against certain fungi and bacteria. Pamamycin-607 has the unique ability to transfer only anions (*e.g.* MnO_4^-) from the aqueous to the benzene layer at neutral and acidic pH. The structure with relative stereochemistry was elucidated by use of 500 MHz 2D ¹H–¹H and ¹H–¹³C correlation NMR and NOE difference spectroscopies.² Here we report that the absolute stereochemistry of pamamycin-607 has been determined as (1).

The smaller hydroxy-acid moiety composing one half of the macrodiolide molecule of pamamycin-607 is quite similar to





Scheme 1. Reagents: i, LiAlH₄, Et₂O; ii, (S)-(-)-MTPA-Cl, pyridine; iii, recycle HPLC; iv, TsOCl (Ts = SO₂C₆H₄Me-*p*); v, pyridinium dichromate (PDC), CH₂Cl₂; vi, CF₃CO₃H, K₂HPO₄; vii, LiAlH₄, tetrahydrofuran (THF); viii, *p*-Br-C₆H₄COCl, pyridine.



Scheme 2. Reagents: i, LiAlH₄, Et₂O; ii, TsOCl, pyridine; iii, PDC, CH₂Cl₂; iv, CF₃CO₃H, K₂HPO₄; v, LiAlH₄, THF; vi, *p*-Br-C₆H₄COCl, pyridine.

the hydroxy-acid unit of the nonactin molecule. Since the absolute stereochemistry of nonactin has already been established,³ we prepared the furfuryl compound (14) from both pamamycin-607 and nonactin through chemical degradation and derivatization, and their optical rotations were compared. Nonactin (2) (200 mg) was degraded reductively with

lithium aluminium hydride (LiAlH₄) to give a racemic diol mixture (3). The racemic mixture was converted to a diastereoisomeric mixture of (S)-(-)- α -methoxy- α -trifluoro-methylphenylacetate (MTPA) (4),⁴ and the latter mixture was separated by recycling HPLC (12 cycles; Develosil 60-10; 20 × 250 mm; Et₂O-n-hexane 2:8; 9.9 ml min⁻¹) to afford pure

diastereoisomers.⁺ Each diastereoisomer was reduced with LiAlH₄ to recover optically pure enantiomers of diols (5) and (6). The absolute configurations of (5) and (6) could be assigned by comparing their optical rotations with the data reported by Beck *et al.*³ {(5): $[\alpha]_D + 36^\circ$ (*c* 1.0, benzene), lit. +31° (*c* 2.16); (6): $[\alpha] - 37^\circ$ (*c* 1.0, benzene)}; thus, their absolute stereochemistry was established as shown in Scheme 1.

Selective tosylation of the primary alcohol in diol (5) followed by oxidation of the secondary alcohol with pyridinium dichromate yielded methylketone (7) $\{[\alpha]_D - 16^\circ (c 1.0, \text{benzene})\}$. Baeyer–Villiger oxidation of (7) with trifluoroperacetic acid and K₂HPO₄ gave the sole product (8) $\{[\alpha]_D - 23^\circ (c 1.0, \text{benzene})\}$; no reaction took place with *m*-chloroperbenzoic acid (*m*CPBA) or peracetic acid. The product (8) was reduced with LiAlH₄ to give the furfuryl alcohol derivative which, without purification, was reacted with *p*-bromobenzoyl chloride to give a non-volatile benzoate (9) (17.1 mg). Diol (6) was reacted in a similar manner to (5), and the antipodal benzoate (10) was obtained.

Pamamycin-607 (1) was reduced with LiAlH₄ to give two diols, fragments S (11) and L, the structure of which may be deduced from (1), (Scheme 2). Fragment S (42 mg) { $[\alpha]_D$ +7.2° (c 2.0, benzene)} was tosylated and oxidized to give a ketone (12). Baeyer–Villiger oxidation of (12) with trifluoroperacetic acid and K₂HPO₄ gave two products. The two products were, without separation, subjected to reductive detosylation with LiAlH₄ followed by *p*-bromobenzoylation to give a benzoate mixture (13) and (14). The mixture was separated by HPLC (Develosil 60-5; 8 × 250 mm; EtOAc–nhexane 7:93; 2.5 ml min⁻¹) to give a major benzoate (13) (11.5 mg) and a minor one (14) (0.66 mg) (Scheme 2).

⁺ The diastereoisomeric purity was analysed by HPLC (Develosil 60-5; $8 \times 250 \text{ mm} \times 2$; Et₂O-n-hexane 15:85; 2.5 ml min⁻¹), to be 96.7% for the less polar diastereoisomer and 96.3% for the more polar one.

‡ All the new compounds gave satisfactory spectral data.

The minor benzoate (14) showed the positive optical rotation, $[\alpha]_{365} + 14^{\circ}$ (c 0.078, CHCl₃), which coincided with the rotation { $[\alpha]_{365} + 14^{\circ}$ (c 1.0, CHCl₃)} of the same benzoate (9) derived from nonactin and opposed to that of the antipodal benzoate (10) { $[\alpha]_{365} - 13^{\circ}$ (c 1.0, CHCl₃)}. This excellent agreement of the optical rotations between (14) and (9) provides the evidence that the absolute configuration of the tetrahydrofuran ring of fragment S is as shown in (14).

Since the relative stereochemistry of pamamycin-607 has already been established as (1) by extensively analysing the spatial relationships among all protons in the molecule by NOE difference and 2D-J resolved spectroscopies,² and that of fragment S was also confirmed by Walcup and Park's synthesis of its racemate,⁵ the absolute stereochemistry of pamamycin-607 has thus been elucidated as shown in (1).

To confirm this result, the optical rotation of the major benzoate (13) was measured. In this case, an extra CH₂ group inserted between the asymmetric tetrahydrofuran ring and the benzoyloxy group may effect little change in the optical rotation. Actually, the positive optical rotation of benzoate (13) { $[\alpha]_{365} + 9^{\circ} (c \ 0.10, \text{CHCl}_3)$ } was similar to those of (14) and (9).

S. K. is currently at Central Research Laboratories, Hokko Chemical Industry Co., Ltd., 2165 Toda, Atsugi, Kanagawa 243, Japan.

Received, 1st August 1989; Com. 9/03244D

References

- 1 S. Kondo, K. Yasui, M. Natsume, M. Katayama, and S. Marumo, J. Antibiotics, 1988, 41, 1196.
- 2 S. Kondo, K. Yasui, M. Katayama, S. Marumo, T. Kondo, and H. Hattori, *Tetrahedron Lett.*, 1987, 28, 5861.
- 3 J. Beck, H. Gerlach, V. Prelog, and W. Voser, *Helv. Chim. Acta*, 1962, **45**, 620; H. Gerlach and V. Prelog, *Helv. Chim Acta*, 1963, **46**, 121.
- 4 J. A. Dale, D. L. Dull, and H. S. Mosher, J. Org. Chem., 1969, 34, 2543.
- 5 R. D. Walcup and G. Park, Tetrahedron Lett., 1988, 29, 5505.