AN UNUSUAL TRANSFORMATION OF 2-PENTAFLUOROPHENOXYETHYL-N, N-DIETHYLAMINE

WITH ALKYL BROMIDES AND ALKYL IODIDES TO FORM PIPERAZINIUM SALTS

AND ALKYL PENTAFLUOROPHENYL ETHERS

V. E. Platonov, O. I. Osina,

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D. Prescher, and G. Engler

The reaction of tertiary amines with alkyl halides give quarternary ammonium salts [1]. In contrast, heating 2-(pentafluorophenoxy)ethyl-N,N-diethylamine (I) with decyl bromide (II), 2-(pentafluorophenoxy)ethyl bromide (III) or 2-(pentafluorophenoxy)ethyl iodide (IV) leads to an unusual transformation giving piperazinium dibromide (V) or diiodide (VI) along with the decyl ether of pentafluorophenol (VII) of the bispentafluorophenyl ether of ethyleneglycol (VIII)

 $R = CH_0(CH_0)_0CH_0$ (VII); $CH_0CH_0OF_5C_6$ (VIII); X = Br(V), I (VI)

The reaction of 8 g (I) and 6.28 g bromide (II) gave 3.3 g (65%) salt (V), mp \sim 280°C (dec., from acetone) and 4.5 g (47%) ether (VII).

Analogously, 5 g (I) and 5 g bromide (III) gave 2.5 g (79%) salt (V) and 4.8 g (69%) ether (VII), while 6.5 g (I) and 7.48 g iodide (VI) gave 3.3 g (66%) salt (VI), mp $\sim\!280^{\circ}\text{C}$ (dec.) and 5.7 g (63%) ether (VIII).

The PMR spectra of salts (V) (in CF3CO2D with HMS as the internal standard) and of (VI) (in water with HMS as the external standard) (δ , ppm, J, Hz): 1.11 t (CH₃, J = 7), 1.48 g (CH₃, J = 7), 3.50 q (CH₂, J = 7), 3.78 q (CH₂, J = 7), 3.76 s (CH₂), 4.00 s (CH₂).

The elemental analysis data of salts (V) and (VI) corresponded to the theoretical The properties of ethers (VII) and (VIII) corresponded to those of authentic samples.

LITERATURE CITED

1. K. V. Vatsuro and G. L. Mishchenko, Name Reactions in Organic Chemistry [in Russian], Izd. Khimiya, Moscow (1976), p. 278.

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