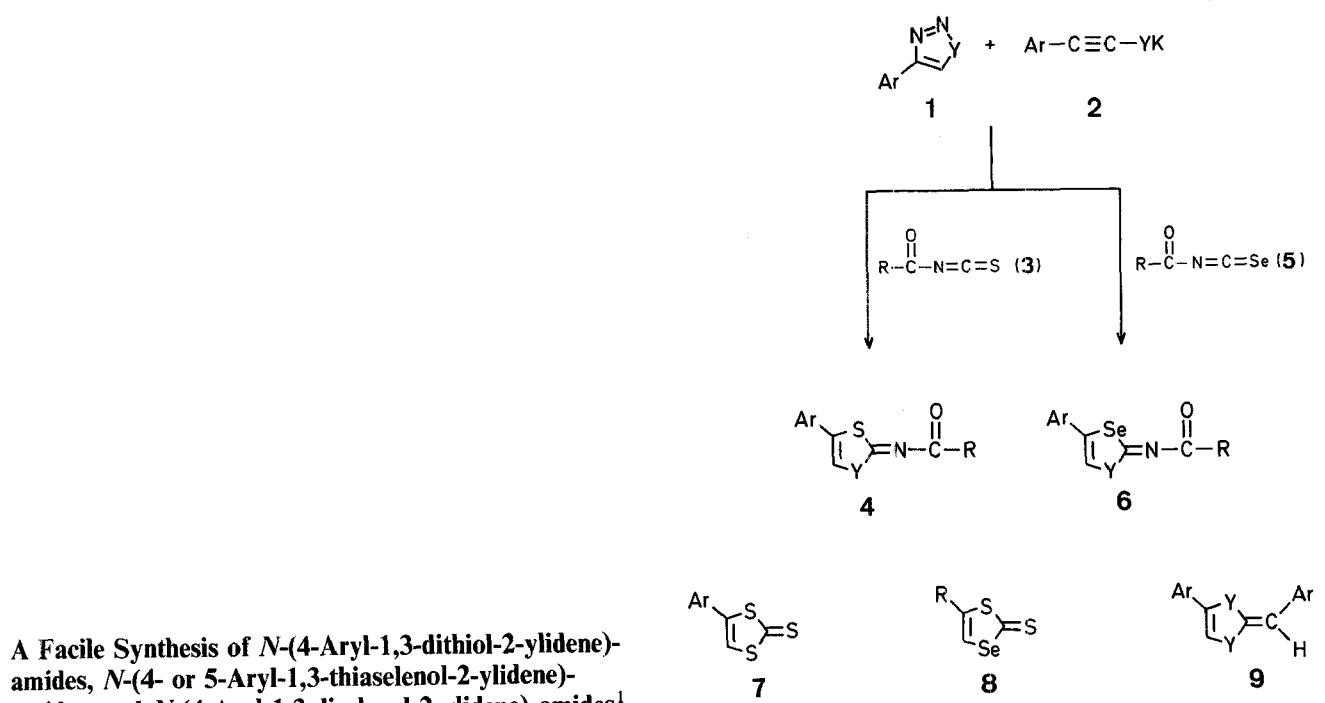


(8)<sup>4</sup> with ethyl azidoformate in 30–45% yield<sup>5</sup> has been reported. We describe here a new simple and high yield synthesis of the title compounds.

During our previous work we had observed that 4-aryl-1,2,3-thiadiazoles (**1**; Y = S) and 4-aryl-1,2,3-selenadiazoles (**1**, Y = Se)<sup>6</sup> are unstable in basic medium. In fact with alcoholic solution of potassium hydroxide an immediate effervescence of nitrogen gas occurred and the corresponding fulvenes, namely, *cis*-2,  $\omega$ -diaryl-1,4-dithiafulvenes (**9**; Y = S) and *cis*-2,  $\omega$ -diaryl-1,4-diselenafulvenes (**9**; Y = Se) were formed respectively<sup>7,8</sup>. It was shown that potassium arylethyne thiolates (**2**; Y = S) and potassium arylethyne selenolates (**2**; Y = Se) are intermediates in these reactions, respectively<sup>7,8,9</sup>. It appeared possible that under certain conditions the bimolecular reaction of **2** leading to the fulvene **9** could be suppressed in favor of cycloaddition of **2** with isothiocyanates (**3**) or isoselenocyanates (**5**)<sup>10</sup>. In fact, reaction of potassium 2-arylethyne thiolates (**2**; Y = S) with **3** in tetrahydrofuran gave *N*-(4-aryl-1,3-dithiol-2-ylidene)-amides (**4**; Y = S) and the reaction of **2** (Y = Se) with **3** afforded *N*-(5-aryl-1,3-thiaselenol-2-ylidene)-amides (**4**, Y = Se). Reaction of **2** (Y = S) with **5**, which was prepared *in situ* from the reaction of acetyl or aroyl chloride with potassium selenocyanate in tetrahydrofuran, yielded *N*-(4-aryl-1,3-thiaselenol-2-ylidene)-amides (**6**; Y = S) and the reaction of compound **2** (X = Se) with **5** gave *N*-(4-aryl-1,3-diselenol-2-ylidene)-amides (**6**; Y = Se).



#### A Facile Synthesis of *N*-(4-Aryl-1,3-dithiol-2-ylidene)-amides, *N*-(4- or 5-Aryl-1,3-thiaselenol-2-ylidene)-amides, and *N*-(4-Aryl-1,3-diselenol-2-ylidene)-amides<sup>1</sup>

A. SHAFIEE\*, G. FANAI

Department of Chemistry, College of Pharmacy, Tehran University,  
Tehran, Iran

The synthesis of ethyl (4-phenyl-1,3-dithiol-2-ylidene)-carbamate (**4**; Y = S, Ar = C<sub>6</sub>H<sub>5</sub>, R = OC<sub>2</sub>H<sub>5</sub>) from the reaction of 4-phenyl-2-thioxo-1,3-dithiole (**7**; Ar = C<sub>6</sub>H<sub>5</sub>) with either ethyl azidoformate or *N,N*-dichlorourethane in 41 or 10% yield, respectively<sup>2,3</sup> and ethyl (5-substituted 1,3-thiaselenol-2-ylidene)-carbamates (**4**; Y = Se, R = OC<sub>2</sub>H<sub>5</sub>) from the reaction of 5-substituted 2-thioxo-1,3-thiaselenoles

After work up, the products **4** and **6** were sufficiently pure (T.L.C.) in most cases. Where necessary, further purification was achieved by crystallization from acetone or preparative T.L.C. on silica gel. The structure was determined by I.R., N.M.R., and M.S. spectrometry, and in some cases by comparison with an authentic material prepared by known method<sup>2,5</sup>.

Melting points were taken on a Kofler hot stage microscope. The I.R. spectra were obtained on a Perkin-Elmer 267 spectrograph. N.M.R. and M.S. spectra were run on a Varian T-60A and MS-311 spectrometers.

**Table 1.** *N*-(4-Aryl-1,3-dithiol-2-ylidene)-amides (**4**; X = S) and *N*-(4-Aryl-1,3-thiaselenol-2-ylidene)-amides (**6**; X = Se) prepared

Com- ound No.	Ar	R	Yield [%]	m.p. [°C] (acetone)	Molecular formula <sup>a</sup>	MS <i>m/e</i> (M <sup>+</sup> ) <sup>b</sup>	<sup>1</sup> H-N.M.R. (CDCl <sub>3</sub> ) <i>δ</i> [ppm]
<b>4a</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	90	131–133°	C <sub>16</sub> H <sub>11</sub> NOS <sub>2</sub> (297.4)	297	8.36 (m, 2H <sub>arom</sub> ); 7.46 (m, 8H <sub>arom</sub> ); 7.15 (s, 1H, H-4)
<b>4b</b>	C <sub>6</sub> H <sub>5</sub>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	90	123–125°	C <sub>17</sub> H <sub>13</sub> NOS <sub>2</sub> (311.4)	311	8.27 (d, 2H <sub>arom</sub> ); 7.46 (m, 5H <sub>arom</sub> ); 7.30 (d, 2H <sub>arom</sub> ); 7.13 (s, 1H, H-4)
<b>4c</b>	C <sub>6</sub> H <sub>5</sub>	CH <sub>3</sub>	60	144–146°	C <sub>11</sub> H <sub>9</sub> NOS <sub>2</sub> (235.3)	235	7.34 (m, 5H <sub>arom</sub> ); 7.10 (s, 1H, H-4); 2.38 (s, 3H, CH <sub>3</sub> )
<b>4d</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>2</sub> H <sub>5</sub> O	95	124–126° <sup>c</sup>	C <sub>12</sub> H <sub>11</sub> NO <sub>2</sub> S <sub>2</sub> (265.4)	265	7.30 (m, 5H <sub>arom</sub> ); 7.10 (s, 1H, H-4); 4.33 (q, 2H, OCH <sub>2</sub> ); 1.40 (t, 3H, CH <sub>3</sub> )
<b>4e</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	88	133–135°	C <sub>17</sub> H <sub>13</sub> NOS <sub>2</sub> (311.4)	311	8.33 (m, 2H <sub>arom</sub> ); 7.40 (m, 7H <sub>arom</sub> ); 7.06 (s, 1H, H-4); 2.37 (s, 3H, CH <sub>3</sub> )
<b>4f</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	89	131–133°	C <sub>18</sub> H <sub>15</sub> NOS <sub>2</sub> (325.5)	325	8.40 (d, 2H <sub>arom</sub> ); 7.43 (d, 2H <sub>arom</sub> ); 7.23 (dd, 4H <sub>arom</sub> ); 7.08 (s, 1H, H-4); 2.40 (s, 6H, CH <sub>3</sub> )
<b>4g</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	CH <sub>3</sub>	58	128–130°	C <sub>12</sub> H <sub>11</sub> NOS <sub>2</sub> (249.4)	249	7.96 (d, 2H <sub>arom</sub> ); 7.30 (d, 2H <sub>arom</sub> ); 7.16 (s, 1H, H-4); 2.36 (s, 6H, CH <sub>3</sub> )
<b>4h</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>5</sub> O	96	153–154°	C <sub>13</sub> H <sub>13</sub> NO <sub>2</sub> S <sub>2</sub> (279.4)	279	7.53 (d, 2H <sub>arom</sub> ); 7.23 (d, 2H <sub>arom</sub> ); 7.10 (s, 1H, H-4); 4.16 (q, 2H, OCH <sub>2</sub> ); 2.30 (s, 3H, CH <sub>3</sub> ); 1.33 (t, 3H, CH <sub>3</sub> )
<b>4i</b>	<i>p</i> -H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	90	126–128°	C <sub>17</sub> H <sub>13</sub> NO <sub>2</sub> S <sub>2</sub> (327.4)	327	8.38 (m, 2H <sub>arom</sub> ); 7.42 (m, 7H <sub>arom</sub> ); 7.08 (s, 1H, H-4); 3.83 (s, 3H, OCH <sub>3</sub> )
<b>4j</b>	<i>p</i> -H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	85	163–164°	C <sub>18</sub> H <sub>15</sub> NO <sub>2</sub> S <sub>2</sub> (341.5)	341	8.28 (d, 2H <sub>arom</sub> ); 7.60 (d, 2H <sub>arom</sub> ); 7.28 (d, 2H <sub>arom</sub> ); 7.16 (s, 1H, H-4); 6.93 (d, 2H <sub>arom</sub> ); 3.83 (s, 3H, OCH <sub>3</sub> ); 2.36 (s, 3H, CH <sub>3</sub> )
<b>4k</b>	<i>p</i> -H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub>	CH <sub>3</sub>	55	125–126°	C <sub>12</sub> H <sub>11</sub> NO <sub>2</sub> S <sub>2</sub> (265.4)	265	7.43 (d, 2H <sub>arom</sub> ); 7.02 (s, 1H, H-4); 6.90 (d, 2H <sub>arom</sub> ); 3.83 (s, 3H, OCH <sub>3</sub> ); 2.32 (s, 3H, CH <sub>3</sub> )
<b>4l</b>	<i>p</i> -H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>5</sub> O	95	163–165°	C <sub>13</sub> H <sub>13</sub> NO <sub>3</sub> S <sub>2</sub> (295.4)	295	7.40 (d, 2H <sub>arom</sub> ); 7.14 (s, 1H, H-4); 6.87 (d, 2H <sub>arom</sub> ); 4.30 (q, 2H, OCH <sub>2</sub> ); 3.80 (s, 3H, OCH <sub>3</sub> ); 1.38 (t, 3H, CH <sub>3</sub> )
<b>6a</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	90	136–137°	C <sub>16</sub> H <sub>11</sub> NOSSe (344.3)	345	8.40 (m, 2H <sub>arom</sub> ); 7.50 (m, 8H <sub>arom</sub> ); 7.16 (s, 1H, H-4)
<b>6b</b>	C <sub>6</sub> H <sub>5</sub>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	92	141–143°	C <sub>17</sub> H <sub>13</sub> NOSSe (358.3)	359	8.37 (d, 2H <sub>arom</sub> ); 7.40 (m, 7H <sub>arom</sub> ); 7.13 (s, 1H, H-4); 2.33 (s, 3H, CH <sub>3</sub> )
<b>6c</b>	C <sub>6</sub> H <sub>5</sub>	CH <sub>3</sub>	65	104–105°	C <sub>11</sub> H <sub>9</sub> NOSSe (282.2)	283	7.46 (m, 5H <sub>arom</sub> ); 7.13 (s, 1H, H-4); 2.46 (s, 3H, CH <sub>3</sub> )
<b>6d</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	85	129–131°	C <sub>17</sub> H <sub>13</sub> NOSSe (358.3)	359	8.43 (m, 2H <sub>arom</sub> ); 7.50 (m, 7H <sub>arom</sub> ); 7.13 (s, 1H, H-4); 2.40 (s, 3H, CH <sub>3</sub> )
<b>6e</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	88	132–134°	C <sub>18</sub> H <sub>15</sub> NOSSe (372.4)	373	8.40 (m, 2H <sub>arom</sub> ); 7.42 (d, 2H <sub>arom</sub> ); 7.23 (d, 2H <sub>arom</sub> ); 7.10 (s, 1H, H-4); 6.93 (d, 2H <sub>arom</sub> ); 2.36 (s, 6H, CH <sub>3</sub> )
<b>6f</b>	<i>p</i> -H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	85	126–127°	C <sub>17</sub> H <sub>13</sub> NO <sub>2</sub> SSe (374.3)	375	8.43 (m, 2H <sub>arom</sub> ); 7.33 (m, 7H <sub>arom</sub> ); 7.10 (s, 1H, H-4); 3.83 (s, 3H, OCH <sub>3</sub> )
<b>6g</b>	<i>p</i> -H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	90	151–153°	C <sub>18</sub> H <sub>15</sub> NO <sub>2</sub> SSe (388.4)	389	8.23 (d, 2H <sub>arom</sub> ); 7.20 (m, 6H <sub>arom</sub> ); 7.10 (s, 1H, H-4); 3.78 (s, 3H, OCH <sub>3</sub> ); 2.36 (s, 3H, CH <sub>3</sub> )

<sup>a</sup> Satisfactory microanalyses obtained: C, ± 0.16; H, ± 0.21; N, ± 0.19<sup>b</sup> M<sup>+</sup> based on <sup>80</sup>Se isotope peak.<sup>c</sup> Ref.<sup>2</sup>, m.p. 122–124°; Ref.<sup>3</sup>, m.p. 126°.***N*-(4-Aryl-1,3-dithiol-2-ylidene)-amides (**4**; Y = S); General Procedure:**

To a stirred solution of 4-aryl-1,2,3-thiadiazole (**1**, Y = S; 0.01 mol) in dioxan (50 ml), a solution of potassium hydroxide (0.56 g, 0.01 mol) in ethanol (5 ml) is added, whereupon the evolution of nitrogen ceases. The precipitate is filtered, washed with ether, and dissolved in tetrahydrofuran (20 ml). To a stirred solution of the latter at 0°C is added a solution of substituted isothiocyanate (3, 0.01 mol) in tetrahydrofuran (20 ml). Stirring is continued 15 min at 0°C followed by 15 min at ambient temperature. The solvent is evaporated under reduced pressure, water (50 ml) is added to the residue, and the mixture is extracted with chloroform (2 × 50 ml).

The extract is dried with sodium sulfate and filtered. The solvent distilled off, and the residual product purified by crystallization or preparative T.L.C. on silica gel using chloroform/petroleum ether (1:1) as eluent (Table 1).

***N*-(4-Aryl-1,3-thiaselenol-2-ylidene)-amides (**4**; Y = Se):**

These compounds are prepared from 4-aryl-1,2,3-selenadiazole (**1**; Y = Se) similarly (Table 2).

***N*-(4-Aryl-1,3-thiaselenol-2-ylidene)-amides (**6**; Y = Se); General Procedure:**

To a stirred solution of potassium selenocyanate (1.44 g, 0.01 mol) in tetrahydrofuran (150 ml) at 0°C, a solution of acetyl or aryl chloride (0.01 mol) in tetrahydrofuran (10 ml) is added. The stirring

**Table 2.** *N*-(5-Aryl-1,3-thiaselenol-2-ylidene)-amides (**4**; Y = Se) and *N*-(4-Aryl-1,3-diselenol-2-ylidene)-amides (**6**; Y = Se) prepared

Com- ound No.	Ar	R	Yield [%]	m.p. [°C] (acetone)	Molecular formula <sup>a</sup>	MS <i>m/e</i> (M <sup>+</sup> ) <sup>b</sup>	<sup>1</sup> H-N.M.R. (CDCl <sub>3</sub> ) <i>δ</i> [ppm]
<b>4a</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	92	146–148 <sup>c</sup>	C <sub>16</sub> H <sub>11</sub> NOSSe (344.3)	345	8.17 (m, 2H <sub>arom</sub> ); 7.87 (s, 1H, H-4); 7.50 (m, 8H <sub>arom</sub> )
<b>4b</b>	C <sub>6</sub> H <sub>5</sub>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	87	133–135 <sup>c</sup>	C <sub>17</sub> H <sub>13</sub> NOSSe (358.3)	359	8.36 (d, 2H <sub>arom</sub> ); 7.87 (s, 1H, H-4); 7.47 (m, 7H <sub>arom</sub> ); 2.41 (s, 3H, CH <sub>3</sub> )
<b>4c</b>	C <sub>6</sub> H <sub>5</sub>	CH <sub>3</sub>	55	119–120 <sup>c</sup>	C <sub>11</sub> H <sub>9</sub> NOSSe (282.2)	283	7.67 (s, 1H, H <sub>4</sub> ); 7.50 (m, 5H <sub>arom</sub> ); 2.46 (s, 3H, CH <sub>3</sub> )
<b>4d</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>2</sub> H <sub>5</sub> O	90	141–143 <sup>c</sup>	C <sub>12</sub> H <sub>11</sub> NO <sub>2</sub> SSe (312.3)	313	7.67 (s, 1H, H-4); 7.33 (m, 5H <sub>arom</sub> ); 4.33 (q, 2H, OCH <sub>2</sub> ); 1.40 (t, 3H, CH <sub>3</sub> )
<b>4e</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	90	125–127 <sup>c</sup>	C <sub>17</sub> H <sub>13</sub> NOSSe (358.3)	359	8.33 (m, 2H <sub>arom</sub> ); 7.76 (s, 1H, H-4); 7.32 (m, 7H <sub>arom</sub> ); 2.33 (s, 3H, CH <sub>3</sub> )
<b>4f</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	95	136–138 <sup>c</sup>	C <sub>18</sub> H <sub>15</sub> NOSSe (372.4)	373	8.33 (d, 2H <sub>arom</sub> ); 7.77 (s, 1H, H-4); 7.33 (m, 6H <sub>arom</sub> ); 2.43 (s, 3H, CH <sub>3</sub> ); 2.40 (s, 3H, CH <sub>3</sub> )
<b>4g</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	CH <sub>3</sub>	50	128–130 <sup>c</sup>	C <sub>12</sub> H <sub>11</sub> NOSSe (296.3)	297	7.67 (s, 1H, H-4); 7.33 (q, 4H <sub>arom</sub> ); 2.44 (s, 3H, CH <sub>3</sub> ); 2.40 (s, 3H, CH <sub>3</sub> )
<b>4h</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>5</sub> O	50	153–155 <sup>d</sup>	C <sub>13</sub> H <sub>13</sub> NO <sub>2</sub> SSe (326.3)	327	7.67 (s, 1H, H-4); 7.33 (q, 4H <sub>arom</sub> ); 4.33 (q, 2H, OCH <sub>2</sub> ); 2.43 (s, 3H, CH <sub>3</sub> ); 1.40 (t, 3H, CH <sub>3</sub> )
<b>4i</b>	<i>p</i> -H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	85	120–122 <sup>c</sup>	C <sub>17</sub> H <sub>13</sub> NO <sub>2</sub> SSe (374.3)	375	8.33 (m, 2H <sub>arom</sub> ); 7.76 (s, 1H, H-4); 7.60 (m, 5H <sub>arom</sub> ); 7.05 (d, 2H <sub>arom</sub> ); 3.93 (s, 3H, OCH <sub>3</sub> )
<b>4j</b>	<i>p</i> -H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	90	153–155 <sup>c</sup>	C <sub>18</sub> H <sub>15</sub> NO <sub>2</sub> SSe (388.4)	389	8.33 (d, 2H <sub>arom</sub> ); 7.73 (s, 1H, H-4); 7.57 (d, 2H <sub>arom</sub> ); 7.04 (q, 4H <sub>arom</sub> ); 3.93 (s, 3H, OCH <sub>3</sub> ); 2.53 (s, 3H, CH <sub>3</sub> )
<b>4k</b>	<i>p</i> -H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>5</sub> O	50	165–167 <sup>c</sup>	C <sub>13</sub> H <sub>13</sub> NO <sub>3</sub> SSe (342.3)	343	7.70 (s, 1H, H-4); 7.60 (d, 2H <sub>arom</sub> ); 7.07 (d, 2H <sub>arom</sub> ); 4.43 (d, 2H, OCH <sub>2</sub> ); 3.90 (s, 3H, OCH <sub>3</sub> ); 1.40 (t, 3H, CH <sub>3</sub> )
<b>6a</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	95	124–125 <sup>c</sup>	C <sub>16</sub> H <sub>11</sub> NOSe <sub>2</sub> (391.2)	393	8.23 (m, 2H <sub>arom</sub> ); 7.80 (s, 1H, H-4); 7.40 (m, 6H <sub>arom</sub> )
<b>6b</b>	C <sub>6</sub> H <sub>5</sub>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	90	114–115 <sup>c</sup>	C <sub>17</sub> H <sub>13</sub> NOSe <sub>2</sub> (405.2)	407	8.37 (d, 2H <sub>arom</sub> ); 7.83 (s, 1H, H-4); 7.63 (m, 7H <sub>arom</sub> ); 2.43 (s, 3H, CH <sub>3</sub> )
<b>6c</b>	C <sub>6</sub> H <sub>5</sub>	CH <sub>3</sub>	52	120–122 <sup>c</sup>	C <sub>11</sub> H <sub>9</sub> NOSe <sub>2</sub> (329.1)	331	7.77 (s, 1H, H-4); 7.43 (m, 5H <sub>arom</sub> ); 2.43 (s, 3H, CH <sub>3</sub> )
<b>6d</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	85	143–145 <sup>c</sup>	C <sub>17</sub> H <sub>13</sub> NOSe <sub>2</sub> (405.2)	407	8.43 (m, 2H <sub>arom</sub> ); 7.78 (s, 1H, H-4); 7.40 (m, 7H <sub>arom</sub> ); 2.38 (s, 3H, CH <sub>3</sub> )
<b>6e</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	80	138–140 <sup>c</sup>	C <sub>18</sub> H <sub>15</sub> NOSe <sub>2</sub> (419.3)	421	8.37 (d, 2H <sub>arom</sub> ); 7.77 (s, 1H, H-4); 7.33 (m, 6H <sub>arom</sub> ); 2.47 (s, 3H, CH <sub>3</sub> ); 2.40 (s, 3H, CH <sub>3</sub> )
<b>6f</b>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	CH <sub>3</sub>	55	122–124 <sup>c</sup>	C <sub>12</sub> H <sub>11</sub> NOSe <sub>2</sub> (343.2)	345	7.77 (s, 1H, H-4); 7.43 (m, 4H <sub>arom</sub> ); 2.43 (s, 3H, CH <sub>3</sub> ); 2.40 (s, 3H, CH <sub>3</sub> )
<b>6g</b>	<i>p</i> -H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	90	133–135 <sup>c</sup>	C <sub>17</sub> H <sub>13</sub> NO <sub>2</sub> Se <sub>2</sub> (421.2)	423	8.50 (m, 2H <sub>arom</sub> ); 7.77 (s, 1H, H-4); 7.33 (m, 7H <sub>arom</sub> ); 3.93 (s, 3H, CH <sub>3</sub> )
<b>6h</b>	<i>p</i> -H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub>	<i>p</i> -H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>	90	140–142 <sup>c</sup>	C <sub>18</sub> H <sub>15</sub> NO <sub>2</sub> Se <sub>2</sub> (435.3)	437	8.28 (d, 2H <sub>arom</sub> ); 7.73 (s, 1H, H-4); 7.49 (q, 4H <sub>arom</sub> ); 7.03 (d, 2H <sub>arom</sub> ); 3.91 (s, 3H, OCH <sub>3</sub> ); 2.40 (s, 3H, CH <sub>3</sub> )

<sup>a</sup> Satisfactory microanalyses obtained: C, ± 0.19; H, ± 0.18; N, ± 0.31.

<sup>b</sup> M<sup>+</sup> based on <sup>80</sup>Se isotope peak.

<sup>c</sup> Ref.<sup>5</sup>, m.p. 143–144°C.

<sup>d</sup> Ref.<sup>5</sup>, m.p. 154–155°C.

<sup>e</sup> Ref.<sup>5</sup>, m.p. 166–167°C.

is continued for 15 min. The mixture is added to a stirring solution of potassium arylethyne thiolate (**2**, Y = S; 0.01 mol) in tetrahydrofuran (20 ml) at 0°C. Stirring is continued for 15 min at ambient temperature and work up is continued as described in the general procedure for **4** (Y = S) (Table 1).

#### *N*-(4-Aryl-1,3-diselenol-2-ylidene)-amides (**6**; Y = Se):

These compounds are prepared from potassium arylethyne selenolates (**2**; Y = Se) similarly (Table 2).

This work was supported by a grant from the Ministry of Culture and Higher Education Research Development Council of Islamic Republic of Iran.

Received: October 21, 1983  
(Revised form: December 19, 1983)

\* Author to whom correspondence should be addressed.

<sup>1</sup> Selenium Heterocycles XXXVI. A preliminary account of this work was presented in the IXth International Congress of Heterocyclic Chemistry, Tokyo, p. 84 (1983).

<sup>2</sup> M. S. Chauhan, D. M. McKinnon, *Can. J. Chem.* **54**, 3879 (1976).

<sup>3</sup> F. Boberg, U. Puttins, G.-J. Wentrup, *Liebigs Ann. Chem.* **1979**, 689.

<sup>4</sup> A. Shafiee, I. Lalezari, F. Savabi, *Synthesis* **1977**, 764.

<sup>5</sup> A. Shafiee, M. M. Vosooghi, R. Asgharian, *J. Heterocyclic Chem.* **17**, 117 (1980).

<sup>6</sup> I. Lalezari, A. Shafiee, M. Yalpani, *Tetrahedron Lett.* **1969**, 5105

<sup>7</sup> A. Shafiee, I. Lalezari, *J. Heterocyclic Chem.* **10**, 11 (1973).

<sup>8</sup> I. Lalezari, A. Shafiee, M. Yalpani, *J. Org. Chem.* **38**, 338 (1973).

<sup>9</sup> F. Malek-Yazdi, M. Yalpani, *J. Org. Chem.* **41**, 729 (1976).

<sup>10</sup> I. B. Douglass, *J. Am. Chem. Soc.* **59**, 740 (1937).