

PREGNANE GLYCOSIDES FROM *BOUCEROSIA AUCHERIANA**

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Key Word Index—*Boucerosia aucheriana*, Asclepiadaceae, bouceroside-ANC, -ADC, -ANO, ADO, -BNO, -BDO, -BNC, -BDC, -CNO, and -CNC; D-cymarose

Abstract—Ten glycosides named bouceroside-ANC, -ADC, -ANO, -ADO, -BNO, -BDO, -BNC, -BDC, -CNO and -CNC were isolated from *Boucerosia aucheriana* and their structures deduced on the basis of the chemical and spectral evidence. The absolute configuration of cymarose, which was obtained by acidic hydrolyses of seven of the glycosides, was determined by means of HPLC analysis of its carbamoyl derivative on a chiral column as D cymarose

INTRODUCTION

In our continuing studies on the plants of Asclepiadaceae family, a large number of C/D *cis*-pregnane glycosides were isolated [1] Mitsuhashi and his co-workers [2] isolated two C/D *cis*-pregnanes, boucerin and dihydro-boucerin, from *Boucerosia aucheriana* and in a previous paper [3], we described the isolation of four glycosides named bouceroside AI, AII, BI and BII from this plant. Recently, two pregnane glycosides named caratuberside A and caratuberside B [4] were isolated from *Caralluma tuberculata*, a synonym of *B. aucheriana* by Ahmad *et al* [4]. In this paper we report the isolation and structural elucidation of 10 glycosides designated as bouceroside-ANC (1), ADC (2), -ANO (3), -ADO (4), -BNO (5), -BDO (6), -BNC (7), -BDC (8), -CNO (9), and -CNC (10) from the less polar glycoside fraction of *B. aucheriana*. The absolute configuration of cymarose was determined by using a chiral HPLC column [5] comparison with the authentic sample.

RESULTS AND DISCUSSION

A crude glycoside mixture was obtained from the chloroform-soluble portion of the methanol extract from the dried aerial part of *B. aucheriana*. From the less polar fractions of this crude glycoside, 10 glycosides were separated by means of normal and reversed phase gel, especially effective was silver nitrate impregnated silica gel for separating the Δ^5 and 5α -H types of aglycones.

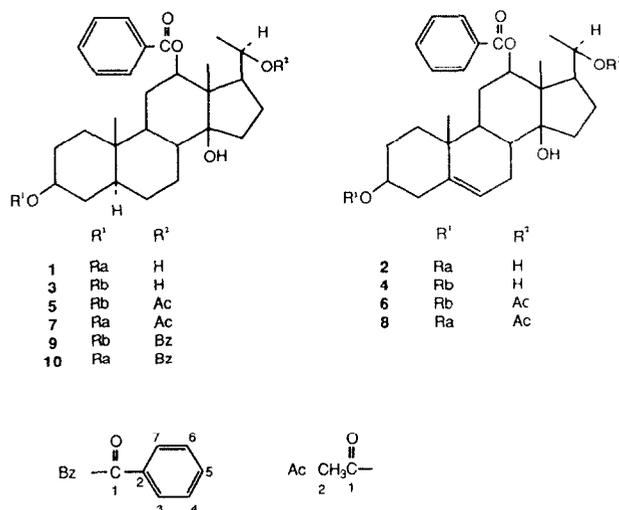
Bouceroside-ANC (1) had the molecular formula $C_{49}H_{76}O_{15}$ on the basis of ion peaks at m/z 927 $[M + Na]^+$ and 943 $[M + K]^+$ in its field-desorption mass spectrum (FDMS). The 500 MHz 1H NMR spectrum of 1 showed methyl signals of its aglycone moiety at δ 0.81 (3H, s, 19-H), 1.23 (3H, *d*, $J = 6.4$ Hz, 21-H), and 1.38 (3H, s, 18-H) and esterified methine proton at 4.72 (1H, *dd*, $J = 12.2, 4.4$ Hz). The presence of a benzoyl ester was suggested from the UV maximum (229 nm), aromatic

proton signals at δ 7.44 (2H, *t*, $J = 7.3$ Hz), 7.56 (1H, *t*, $J = 7.3$ Hz), and 8.05 (2H, *d*, $J = 7.3$ Hz) and the aromatic carbon signals (Table 1). Three anomeric protons at δ 4.59 (1H, *d*, $J = 8.3$ Hz), 4.77 and 4.83 (each 1H, *dd*, $J = 9.8, 2.0$ Hz) and anomeric carbon signals at δ 96.1, 100.4, and 104.1 were observed. The β -linkage of each sugar was revealed by the coupling constants of the respective anomeric protons in the 1H NMR spectrum. From its ^{13}C NMR spectrum (Table 2), compound 1 contained two glycosylated cymaropyranosyl and a terminal 6-deoxy-3-*O*-methyl-allopyranosyl moieties, judging from comparison of the chemical shifts with those of known asclepiadaceous glycosides. In its ^{13}C NMR spectrum, partial relaxation Fourier-transform (PRFT) measurement [6, 7] of 1 showed that 6-deoxy-3-*O*-methyl-allose was the terminal sugar. On mild acidic hydrolysis of 1, an aglycone (11), cymarose (12), and a biose (13) were obtained. The aglycone (11) had the molecular formula $C_{28}H_{40}O_5$ on the basis of ion peak at m/z 456 (M^+) in its FDMS. From its 1H and ^{13}C NMR spectra, 11 was suggested to be 12-*O*-benzoyl dihydroboucerin, a previously reported compound [3]. The NOE experiments supported the stereochemistry at C-17 and C-20 positions deduced from ^{13}C NMR chemical shifts [8]. Compound 12 was identified by spectral data and TLC comparison with the authentic sample. The biose (13) was identified as asclepiobiose (6-deoxy-3-*O*-methyl- β -D-allopyranosyl-(1 \rightarrow 4)-D-cymarose) [9-11] by specific rotation, ^{13}C NMR spectrum (Table 3), and behavior on TLC. As glycosidation shifts [12, 13] were observed at the C-2 (-2.3), C-3 (+6.2), and C-4 (-4.2) positions of 1, the sugar moiety was linked to the C-3 hydroxyl group of the aglycone. Thus, 1 was deduced to be 12-*O*-benzoyl dihydroboucerin 3-*O*-6-deoxy-3-*O*-methyl- β -D-allopyranosyl-(1 \rightarrow 4)- β -D-cymaropyranosyl-(1 \rightarrow 4)- β -D-cymaropyranoside.

Bouceroside-ADC (2) had the molecular formula $C_{49}H_{74}O_{15}$ on the basis of an ion peak at m/z 925 $[M + Na]^+$ in its FDMS. Compound 2 had the same sugar chain as 1 from its 1H and ^{13}C NMR spectra, and 6-deoxy-3-*O*-methyl-allose was confirmed as the terminal sugar by PRFT measurement. On mild acidic hydrolysis, 2 gave 12-14. The former two compounds were identified

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Table 1 ^{13}C NMR chemical shifts of the aglycone moieties of **1-4** and their aglycones **11** and **14**

C	1	3	11	2	4	14
1	37.3	37.3	37.6	37.5	37.3	37.8
2	30.0 (-2.3)	30.0 (-2.3)	32.3	30.3 (-2.2)	30.3 (-2.2)	32.5
3	76.7 (+6.2)	76.6 (+6.1)	70.5	77.4 (+6.2)	77.4 (+6.2)	71.2
4	35.0 (-4.2)	34.9 (-4.3)	39.2	39.3 (-4.0)	39.3 (-4.0)	43.3
5	44.7	44.6	45.1	139.7	139.7	140.6
6	29.2 ^a	29.2 ^a	29.2 ^a	122.4	122.5	122.0
7	28.2 ^a	28.2 ^a	28.3 ^a	27.8 ^a	27.8 ^a	27.8 ^a
8	40.4	40.3	40.4	36.9	37.0	37.6
9	46.7	46.7	46.7	43.9	44.0	43.6
10	36.1	36.1	36.1	37.5	37.6	37.5
11	27.0 ^b	27.2 ^b	27.1 ^b	26.7 ^a	26.7 ^a	26.8 ^a
12	79.8	79.7	79.9	79.3	79.3	79.3
13	53.4	53.4	53.4	53.3	53.4	53.3
14	85.0	84.9	85.0	85.1	85.2	85.2
15	32.8	32.9	32.8	33.4	33.4	33.4
16	26.5 ^b	26.6 ^b	26.6 ^b	26.4	26.4	26.5
17	53.1	53.1	53.1	53.1	53.1	53.0
18	11.5	11.5	11.6	11.5	11.5	11.6
19	12.2	12.1	12.3	19.5	19.5	19.7
20	70.8	70.8	70.8	70.9	70.9	70.8
21	23.6	23.8	23.7	23.7	23.7	23.7
1'	166.7	166.7	166.7	166.7	166.7	166.7
2'	131.7	131.6	131.7	131.6	131.6	131.6
3'	130.0	130.0	130.0	130.0	130.1	130.1
4'	128.9	128.8	128.9	128.9	128.9	129.0
5'	133.2	133.3	133.3	133.3	133.3	133.3
6'	128.9	128.8	128.9	128.9	128.9	129.0
7'	130.0	130.0	130.0	130.0	130.1	130.1

δ value (ppm) from internal TMS in pyridine- d_5

^a Values in each column may be interchangeable

(), Glycosidation shifts

by TLC comparison with the authentic samples. The molecular formula of **14** was established as $\text{C}_{28}\text{H}_{38}\text{O}_5$ by its high-resolution mass spectrum (HRMS) which was two mass units less than that of **11**. In addition to the presence of a benzoyl group (δ 7.45, 7.57, and 8.06), olefinic proton and carbon signals were observed at δ 5.44

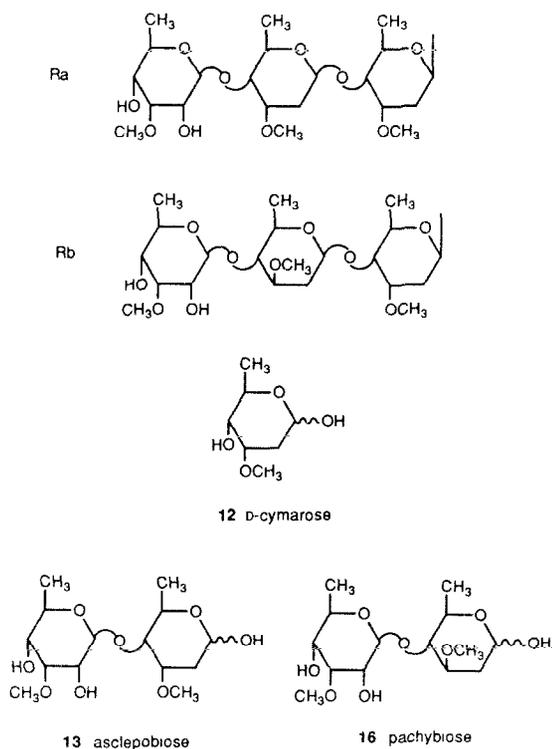
(*br s*), and 140.6 and 122.0, respectively. Moreover the 19-methyl signal at δ 0.80 in the **11** was shifted to lower field at 1.01 in the spectrum of **14**. Thus, compound **14** was assumed to be the Δ^5 derivative of **11**, 12-*O*-benzoyl boucerin [3]. Glycosidation shifts of the aglycone carbon signals were observed at the C-2, C-3 and C-4 positions

Table 2 ^{13}C NMR chemical shifts of sugar moieties of 1-7, 9 and 10

	1	2	7	10	3	4	5	6	9
cym-1	96.1	96.4	96.0	96.0	cym-1	96.0	96.4	96.0	96.1
2	36.9 ^a	36.9 ^a	37.0 ^a	36.9 ^a	2	37.3 ^a	37.3 ^a	37.3	37.2 ^a
3	78.2	78.1 ^b	78.1	78.1 ^b	3	77.8	77.9	77.8	78.0
4	83.3 ^b	83.3 ^c	83.2 ^b	83.4 ^c	4	82.7 ^b	82.8 ^b	82.8 ^a	82.8 ^b
5	69.0 ^c	69.1 ^d	69.0 ^c	69.1 ^d	5	69.0	69.0	68.9	69.0
6	18.6	18.6	18.6	18.6	6	18.6 ^c	18.6 ^c	18.6 ^b	18.6 ^c
3-OMe	58.8	58.8	58.8	58.8	3-OMe	58.8	58.8	58.8	58.8
cym'-1	100.4	100.4	100.4	100.4	ole-1	101.9	101.9	101.9	101.9
2	37.3 ^a	37.3 ^a	37.4 ^a	37.1 ^a	2	37.5 ^a	37.6 ^a	37.6	37.6 ^a
3	78.2	78.0 ^b	78.1	78.0 ^b	3	79.2	79.3	79.2	79.3
4	83.4 ^b	83.4 ^c	83.4 ^b	83.5 ^c	4	83.5 ^b	83.5 ^b	83.4 ^a	83.6 ^b
5	69.3 ^c	69.3 ^d	69.3 ^c	69.3 ^d	5	72.0	72.1	72.0	72.1
6	18.6	18.6	18.6	18.6	6	18.9 ^c	18.9 ^c	18.9 ^b	18.9 ^c
3-OMe	58.8	58.8	58.8	58.9	3-OMe	57.0	57.1	57.2	57.1
allo-1	104.2	104.2	104.2	104.2	allo-1	101.9	101.9	101.9	101.9
2	73.2	73.2	73.1	73.1	2	73.2	73.3	73.2	73.3
3	83.9	83.9	83.9	84.0	3	84.0	84.0	83.9	83.9
4	74.4	74.5	74.4	74.5	4	74.5	74.6	74.5	74.6
5	70.7	70.7	70.7	70.7	5	70.8	70.9	71.0	71.1
6	18.6	18.6	18.6	18.6	6	18.6	18.6	18.6	18.7
3-OMe	62.1	62.1	62.0	62.2	3-OMe	62.0	62.0	62.0	62.0

δ value (ppm) from internal TMS in pyridine-*d*₅ cym, β -cymaropyranose ole, β -oleandropyranose allo, 6-deoxy-3-*O*-methyl- β -allopyranose

^{a-d} Values in each column may be interchangeable.

Table 3. ^{13}C NMR chemical shifts of **13** and **16**

	13	16		16
cym-1	92.5	ole-1	ole-1	91.4
2	38.5	2	2	37.6
3	78.5	3	3	77.6
4	83.8	4	4	84.0
5	69.3	5	5	67.4
6	18.6 ^a	6	6	18.9 ^a
3-OMe	58.4	3-OMe	3-OMe	57.2
allo-1	104.2	allo-1	allo-1	102.2
2	73.0	2	2	73.3
3	83.9	3	3	84.0
4	74.4	4	4	74.6
5	70.7	5	5	70.9
6	18.9 ^a	6	6	19.2 ^a
3-OMe	62.1	3-OMe	3-OMe	62.0

δ value (ppm) from internal TMS in pyridine- d_5
 cym, β -cymaropyranose; ole, β -oleandropyranose
 allo, 6-deoxy-3-O-methyl- β -allopyranose

^aValue in each column may be interchangeable

(Table 1). Therefore, the structure of **2** was deduced to be 12-*O*-benzoyl boucerin 3-*O*-6-deoxy-3-*O*-methyl- β -D-allopyranosyl-(1 \rightarrow 4)- β -D-cymaropyranosyl-(1 \rightarrow 4)- β -D-cymaropyranoside.

Bouceroside-BNO (**5**) had the molecular formula $\text{C}_{51}\text{H}_{78}\text{O}_{16}$ on the basis of elemental analysis and FDMS (m/z 969 [$\text{M} + \text{Na}$]⁺ and 946 [M]⁻). The 500 MHz ^1H NMR spectrum of **5** showed methyl signals of its aglycone moiety at δ 0.80 (3H, *s*, 19-H), 1.12 (3H, *d*, $J = 6.3$ Hz, 21-H), 1.14 (3H, *s*, 18-H) and 1.93 (3H, *s*, 2''-H),

and two esterified methine protons at δ 4.76 (1H, *dd*, $J = 12.2, 4.4$ Hz) and 4.87 (1H, *dq*, $J = 9.4, 6.3$ Hz). Moreover, five aromatic proton signals at δ 7.44, 7.56 and 8.09 were observed. From the ^{13}C NMR spectrum (Table 4), the two esters were suggested as benzoyl and acetyl. Three anomeric proton signals at δ 4.48 (1H, *dd*, $J = 9.8, 2.0$ Hz), 4.79 (1H, *d*, $J = 7.8$ Hz) and 4.86 (1H, *dd*, $J = 9.8, 2.0$ Hz), and anomeric carbon signals at δ 96.0 and two 101.9 were observed. Thus, compound **5** was also suggested to be a triglycoside composed of terminal 6-deoxy-3-*O*-methyl- β -D-allopyranose, **12** and oleandrose from the ^{13}C NMR chemical shifts. On mild acidic hydrolysis of **5**, an aglycone (**15**), **12** and a biose (**16**) were obtained. Compound **12** was identified by TLC comparison with the authentic sample. The biose (**16**) was identified as pachybiose [6-deoxy-3-*O*-methyl- β -D-allopyranosyl-(1 \rightarrow 4)- β -D-oleandrose] [9, 14] by specific rotation, ^{13}C NMR spectrum (Table 3) and behaviour on TLC. The aglycone (**15**) was suggested to be 12-*O*-benzoyl-20-*O*-acetyl dihydroboucerin [3], which was isolated from this plant, by ^1H and ^{13}C NMR spectra (Table 4). Though the position of the two esters had been determined by chemical methods, long-range selective proton decoupling (LSPD) measurement [15–17] of **15** gave the same conclusion (Fig. 1). In pyridine- d_5 , when the proton signal at δ 5.14 (12-H) was irradiated, the carbonyl carbon signal of the benzoyl at δ 166.7 sharpened and the extent of the signal splitting was decreased simultaneously. When the proton signal at δ 5.42 (20-H) was irradiated, the carbonyl carbon signal of the acetyl at δ 170.2 sharpened and the extent of the signal splitting was decreased. As the glycosidation shifts of the aglycone carbon signals were observed at C-2 (-2.3), C-3 (+6.1) and C-4 (-4.0), the sugar moiety was linked to the C-3 hydroxyl group of the aglycone. PRFT measurement of **5** confirmed that 6-deoxy-3-*O*-methyl- β -D-allopyranose was the terminal sugar. Thus compound **5** was determined as 12-*O*-benzoyl-20-*O*-acetyl dihydroboucerin 3-*O*-6-deoxy-3-*O*-methyl- β -D-allopyranosyl-(1 \rightarrow 4)- β -D-oleandropyranosyl-(1 \rightarrow 4)- β -D-cymaropyranoside.

Bouceroside-BDO (**6**) had the molecular formula $\text{C}_{51}\text{H}_{76}\text{O}_{16}$ on the basis of the parent ion peak at m/z 944 (M^+) in its FDMS. On mild acidic hydrolysis of **6**, an aglycone (**17**), **12** and **16** were obtained. The latter two compounds were identified by TLC comparison with the authentic samples. In the ^1H NMR spectrum of **17** in addition to the benzoyl group at δ 7.45, 7.57 and 8.11, and the acetyl group at δ 1.97, an olefinic proton signal at δ 5.44 was observed. From its ^{13}C NMR spectrum (Table 4), compound **17** was deduced to be 12-*O*-benzoyl-20-*O*-acetyl boucerin [3]. PRFT measurement of **6** showed that 6-deoxy-3-*O*-methyl- β -D-allopyranose was the terminal sugar. Thus compound **6** was deduced to be 12-*O*-benzoyl-20-*O*-acetyl boucerin 3-*O*-6-deoxy-3-*O*-methyl- β -D-allopyranosyl-(1 \rightarrow 4)- β -D-oleandropyranosyl-(1 \rightarrow 4)- β -D-cymaropyranoside.

Bouceroside-ANO (**3**) and -ADO (**4**) had the molecular formulae $\text{C}_{49}\text{H}_{76}\text{O}_{15}$ and $\text{C}_{49}\text{H}_{74}\text{O}_{15}$, respectively. From their ^{13}C NMR spectra (Table 2), both glycosides have Rb type sugar chain, and 6-deoxy-3-*O*-methyl- β -D-allopyranose was confirmed as the terminal sugar by PRFT measurement. On mild acidic hydrolysis, compound **3** gave **11**, **13**, and **4** gave **14**, **12** and **16** respectively. Compounds **11**, **13** and **16** were identified by TLC comparison with the authentic samples. The glycosidation shifts were observed at the C-2, C-3 and C-4 positions of each aglycone (Table 1). Thus, compound **3** was deduced to be 12-*O*-benzoyl

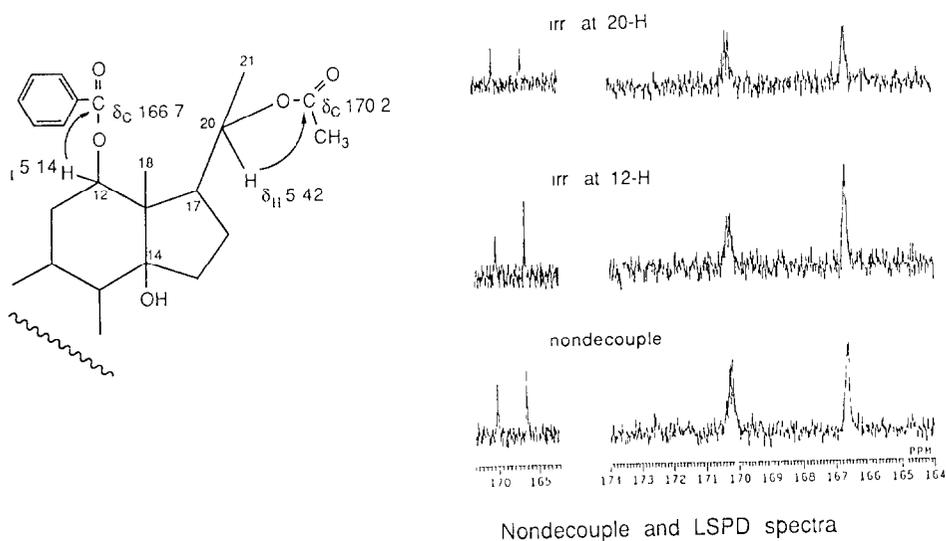
Table 4 ^{13}C NMR chemical shifts of aglycone moieties of **5**, **6**, and **7**, and their aglycones **15** and **17**

C	5	7	15	6	17
1	37.3	37.2	37.5	37.6	37.5
2	30.0 (-2.3)	30.0 (-2.3)	32.3	30.3 (-2.2)	32.5
3	76.6 (+6.1)	76.6 (+6.1)	70.5	77.2 (+6.1)	71.1
4	34.8 (-4.3)	34.8 (-4.3)	39.1	39.1 (-4.2)	43.3
5	44.6	44.5	45.0	139.7	140.7
6	29.1 ^a	29.1 ^a	29.2 ^a	122.8	122.5
7	28.1 ^a	28.1 ^a	28.2 ^a	27.7	27.8
8	40.8	40.9	40.9	37.6	37.6
9	46.5	46.5	46.6	43.9	43.8
10	36.1	36.0	36.1	37.6	37.5
11	26.8 ^b	26.8 ^b	26.9 ^b	26.7 ^a	26.7 ^a
12	79.7	79.8	79.7	79.6	79.3
13	52.5	52.5	52.5	52.3	52.4
14	85.5	85.6	85.6	85.7	85.8
15	32.2	32.2	32.3	32.7	32.5
16	25.9 ^b	26.0 ^b	26.0 ^b	25.5 ^a	25.6 ^a
17	50.7	50.7	50.7	50.6	50.7
18	10.3	10.3	10.3	10.3	10.2
19	12.1	12.1	12.3	19.4	19.4
20	73.9	74.0	74.0	74.0	73.9
21	19.3	19.4	19.4	19.4	19.4
1'	166.7	166.7	166.7	166.6	166.7
2'	131.6	131.7	131.7	131.6	131.6
3'	130.1	130.1	130.1	130.0	130.1
4'	128.9	128.9	128.9	128.9	128.9
5'	133.4	133.3	133.4	133.4	133.4
6'	128.9	128.9	128.9	128.9	128.9
7'	130.1	130.1	130.1	130.0	130.1
1''	170.2	170.3	170.2	170.2	170.2
2''	21.6	21.6	21.6	21.6	21.6

δ value (ppm) from internal TMS in pyridine- d_5

^a ^bValues in each column may be interchangeable.

(). Glycosidation shifts



Nondecouple and LSPD spectra

Fig 1

dihydroboucerin 3-*O*-6-deoxy-3-*O*-methyl- β -D-allopyranosyl-(1 \rightarrow 4)- β -D-oleandropyranosyl-(1 \rightarrow 4)- β -D-cymaropyranoside, and compound **4** to be 12-*O*-benzoyl boucerin 3-*O*-6-deoxy-3-*O*-methyl- β -D-allopyranosyl-(1 \rightarrow 4)- β -D-oleandropyranosyl-(1 \rightarrow 4)- β -D-cymaropyranoside

Bouceroside-BNC (**7**) had the same molecular formula $C_{51}H_{78}O_{16}$ as **5** on the basis of ion peak at m/z 969 $[M + Na]^+$ in its FDMS. From its ^{13}C NMR spectrum (Tables 2 and 4), compound **7** contained **16** as the aglycone moiety and an Ra type sugar chain. PRFT measurement of **7** indicated that 6-deoxy-3-*O*-methyl-allose was the terminal sugar. On mild acidic hydrolysis, **7** gave **16**, **12** and **13**, which were identified by TLC comparison with the authentic samples. The glycosidation shifts of the aglycone carbons were observed at C-2 (-2.3), C-3 (+6.1) and C-4 (-4.3). Thus compound **7** was deduced to be 12-*O*-benzoyl-20-*O*-acetyl dihydroboucerin 3-*O*-6-deoxy-3-*O*-methyl- β -D-allopyranosyl-(1 \rightarrow 4)- β -D-cymaropyranosyl-(1 \rightarrow 4)- β -D-cymaropyranoside

Bouceroside-BDC (**8**) had the molecular formula $C_{51}H_{76}O_{16}$ on the basis of elemental analysis and its fast atom bombardment mass spectrum (FABMS) (m/z 967 $[M + Na]^+$). The 500 MHz 1H NMR spectrum of **8** exhibited methyl signals of the aglycone moiety at δ 1.00 (3H, s, 19-H), 1.14 (3H, d, $J = 6.3$ Hz, 21-H), 1.17 (3H, s, 18-H) and 1.97 (3H, s, 2''-H) and two esterified methine protons at δ 4.81 (1H, dd, $J = 12.2, 4.4$ Hz, 12-H) and 4.91 (1H, dd, $J = 9.7, 6.3$ Hz, 20-H). The presence of a benzoyl group was suggested from the UV maximum (230 nm) and five aromatic protons at δ 7.45 (2H, t, $J = 7.3$ Hz), 7.58 (1H, t, $J = 7.3$ Hz) and 8.11 (2H, d, $J = 7.3$ Hz). In addition, an olefinic proton signal was seen at δ 5.44 (*br s*). Three anomeric proton signals at δ 4.59 (1H, d, $J = 7.8$ Hz), 4.76 and 4.85 (each 1H, dd, $J = 9.8, 2.0$ Hz) and methoxy methyl signals at δ 3.42, 3.45 and 3.66 (each 3H, s) were shown in its 1H NMR spectrum. On mild acidic hydrolysis, compound **8** gave **17**, **12** and **13**, which were identified by TLC comparison with the authentic samples. Thus, compound **8** was presumed to be 12-*O*-benzoyl-20-*O*-acetyl boucerin 3-*O*-6-deoxy-3-*O*-methyl- β -D-allopyranosyl-(1 \rightarrow 4)- β -D-cymaropyranosyl-(1 \rightarrow 4)- β -D-cymaropyranoside

Bouceroside-CNO (**9**) and -CNC (**10**) had the same molecular formula $C_{56}H_{80}O_{16}$. From their ^{13}C NMR spectra (Table 2 and 5), they contain the same aglycone moiety as boucerogenin II (**18**) [3], but differ from one another in the sugar moiety. On mild acidic hydrolysis, compound **9** gave **18**, **12** and **16**, while **10** gave **18**, **12** and **13**. Thus, compound **9** was deduced to be boucerogenin II 3-*O*-6-deoxy-3-*O*-methyl- β -D-allopyranosyl-(1 \rightarrow 4)- β -D-oleandropyranosyl-(1 \rightarrow 4)- β -D-cymaropyranoside, and **10** to be boucerogenin II 3-*O*-6-deoxy-3-*O*-methyl- β -D-allopyranosyl-(1 \rightarrow 4)- β -D-cymaropyranosyl-(1 \rightarrow 4)- β -D-cymaropyranoside

The chirality of cymarose, which was obtained by acidic hydrolyses of glycosides **1**, **3-7** and **9**, was deduced to be D-series by means of HPLC, using a chiral column (SUMIPAX OA-4100) [5].

EXPERIMENTAL

Mps uncorr. 1H NMR spectra were run at 270 and 500 MHz with TMS as the int. standard in $CDCl_3$ or pyridine- d_5 soln. ^{13}C NMR spectra were performed at 67.5 and 22.5 MHz in

Table 5 ^{13}C NMR chemical shifts of aglycone moieties of **9** and **10**, and their aglycone **18**

C	9	10	18
1	37.6	37.3	37.8
2	30.0 (-2.2)	30.1 (-2.1)	32.2
3	76.7 (+6.3)	76.5 (+6.1)	70.4
4	34.9 (-4.1)	34.8 (-4.2)	39.0
5	44.7	44.6	44.7
6	29.1 ^a	29.2 ^a	29.1 ^a
7	28.1 ^a	28.2 ^a	28.1 ^a
8	40.4	40.5	40.5
9	46.6	46.5	46.5
10	36.1	36.1	36.0
11	26.9 ^b	26.9 ^b	26.9 ^b
12	79.0	79.0	78.9
13	53.0	52.9	52.8
14	85.7	85.7	85.7
15	32.2	32.1	32.2
16	25.5 ^b	25.6 ^b	25.5 ^b
17	50.4	50.3	50.2
18	10.3	10.3	10.3
19	12.1	12.0	12.1
20	74.8	74.9	74.8
21	19.6	19.7	19.6
1	166.8	166.8	166.7
1''	166.1	166.1	166.0
2'	131.8	131.7	131.7
2''	131.7	131.6	131.7
3'	130.1	130.2	130.1
3''	129.9	130.0	129.8
4'	129.1	129.1	129.1
4''	128.8	128.9	128.8
5'	133.4	133.5	133.4
5''	133.0	133.1	133.4
6'	129.1	129.1	129.1
6''	128.8	128.8	128.8
7'	130.1	130.2	130.1
7''	129.9	130.0	129.8

δ value (ppm) from internal TMS in pyridine- d_5

^a ^b Values in each column may be interchangeable (), Glycosidation shifts

pyridine- d_5 soln. UV spectra were obtained in EtOH. IR spectra were recorded in $CHCl_3$. CC were carried out on Wakogel C-100 and C-200 for normal phase, and Lobar column LiChroprep RP-8 (40-63 μ m) for reversed phase. HPLC analyses were performed with a Waters A-4000 pump and 440 absorbance detector at 254 nm using SUMIPAX OA-4100 (5 μ , ϕ 4 mm 1 d \times 25 cm) (Sumitomo Chemical Co. Ltd). TLC was carried out on a precoated plate Kieselgel 60 F₂₅₄, Merck with the following solvent systems $R_{f,1}$, MeOH- $CHCl_3$ (1/9), $R_{f,2}$, hexane-Me₂CO (1/1), $R_{f,3}$, H₂O-MeOH- $CHCl_3$ (1/3/15, lower layer), $R_{f,4}$, Me₂CO-Et₂O (1/3). Abbreviations used for sugars in this section are as follows: cym, cymarose; ole, oleandrose; allo, 6-deoxy-3-*O*-methyl-allose.

Plant material Boucerosia *aucheriana* used in this research was collected and identified by Prof. N. A. Quazilbash of Peshawar University, Pakistan.

Extraction and isolation Dried and powdered aerial parts of *B. aucheriana* (420 g) were extracted with MeOH and concd to give an extract (38 g) which was re-extracted with $CHCl_3$. The $CHCl_3$

soluble fraction was dissolved in CHCl_3 (80 ml) and the soln poured into hexane (1700 ml). The insoluble portion corresponded to a crude glycoside mixture (10.9 g) which was chromatographed on Wakogel C-100 (100 g) with solvents of increasing polarity from 0 to 30% MeOH- CHCl_3 into 6 fractions. Fraction 2 (3.74 g) eluted with 2.5% MeOH in CHCl_3 was rechromatographed on Wakogel C-200 with 2% MeOH- CHCl_3 to separate fraction 2-2 (1.68 g, a mixture of 5-10) and fraction 2-3 (1.47 g, a mixture of 1-4). Rechromatography of the fraction 2-3 on silica gel with 2.5% MeOH- CHCl_3 and increasing polarity from hexane- Me_2CO (2:1) to hexane- Me_2CO (1:2), and then on reversed-phase gel with H_2O -MeOH (3:17) to separate fraction A (155.6 mg 1 and 2) and fraction B (100.8 mg 3 and 4). Fraction A and B were each chromatographed on AgNO_3 -silica gel with hexane- Me_2CO (2:1 and 3:2) to afford glycosides, 1 (34.2 mg), 2 (45.0 mg), 3 (19.4 mg) and 4 (22.5 mg), respectively. Rechromatography of fraction 2-2 on silica gel with 4% MeOH in CHCl_3 and hexane- Me_2CO (1:1 and 3:2), and then on reversed-phase gel with H_2O -MeOH (3:17 and 1:4) to separate fraction C (176.6 mg 5 and 6), fraction D (65.2 mg 7 and 8) and fraction E (76.7 mg 9 and 10). Each of fraction C, D and E was chromatographed on AgNO_3 -silica gel with hexane- Me_2CO (5:2) to afford the following glycosides 5 (65.3 mg) and 6 (30.0 mg) from fraction C, 7 (15.7 mg) and 8 (5.2 mg) from fraction D, 9 (25.5 mg) and 10 (7.0 mg) from fraction E.

R_f value: 1 (R_{f1} ; 0.56, R_{f2} ; 0.42), 2 (R_{f1} ; 0.56, R_{f2} ; 0.42), 3 (R_{f1} ; 0.56, R_{f2} ; 0.42), 4 (R_{f1} ; 0.56, R_{f2} ; 0.42), 5 (R_{f1} ; 0.70, R_{f2} ; 0.48), 6 (R_{f1} ; 0.70, R_{f2} ; 0.48), 7 (R_{f1} ; 0.70, R_{f2} ; 0.48), 8 (R_{f1} ; 0.70, R_{f2} ; 0.48), 9 (R_{f1} ; 0.74, R_{f2} ; 0.50), 10 (R_{f1} ; 0.74, R_{f2} ; 0.50).

Bouceroside-ANC (1) Amorphous powder; mp 138.5-142.5° [α]_D -3.2° (CHCl_3 , c 1.02) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 281 (2.95), 278 (3.07), 229 (4.12) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 3400, 1710, 1600, 1580, 1275. FDMS m/z 943 [$\text{M} + \text{K}$]⁺, 927 [$\text{M} + \text{Na}$]⁺ ¹H NMR (500 MHz, CDCl_3) δ 0.81 (3H, s, 19-H), 1.22 (3H, d, $J = 6.3$ Hz, 21-H), 1.38 (3H, s, 18-H), 3.42, 3.44, 3.66 (each 3H, s, 3-OMe of sugar moiety), 4.59 (1H, d, $J = 8.3$ Hz, allo-1-H), 4.72 (1H, dd, $J = 12.2, 4.4$ Hz, 12-H), 4.77, 4.86 (each 1H, dd, $J = 9.8, 2.0$ Hz, cym-1-H), 7.44 (2H, t, $J = 7.3$ Hz, 4',6'-H), 7.56 (1H, t, $J = 7.3$ Hz, 5'-H), 8.05 (2H, d, $J = 7.3$ Hz, 3',7'-H) ¹³C NMR spectrum see Tables 1 and 2.

Acidic hydrolysis of 1 A soln of 1 (28.0 mg) in MeOH (6 ml) was treated with 0.1 M H_2SO_4 (2 ml), and the mixture kept at ca 60° for 30 min, then H_2O (6 ml) was added and whole concd to 6 ml. The soln was warmed at 60° for further 30 min, then the soln was extracted with Et_2O (15 ml). The Et_2O layer was washed with 5% NaHCO_3 (10 ml) and satd NaCl (10 ml), then evapd to give a syrup, which was chromatographed on silica gel with hexane- Me_2CO (2:1) to afford 11 (4.4 mg). Compound 11 (R_{f1} ; 0.51, R_{f2} ; 0.51), an amorphous powder, mp 128.5-133.5° [α]_D -32° (CHCl_3 , c 0.30) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 281 (3.05), 273 (3.16), 229 (4.24) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 3400, 1710, 1600, 1580, 1275. FD-MS m/z 456 [M]⁺. ¹H NMR (500 MHz, CDCl_3) δ 0.83 (3H, s, 19-H), 1.23 (3H, d, $J = 6.8$ Hz, 21-H), 1.39 (3H, s, 18-H), 3.61 (1H, m, 3-H), 3.83 (1H, dq, $J = 8.4, 6.3$ Hz, 20-H), 4.72 (1H, dd, $J = 12.2, 4.4$ Hz, 12-H), 7.45 (2H, t, $J = 7.8$ Hz, 4',6'-H), 7.57 (1H, t, $J = 7.8$ Hz, 5'-H), 8.06 (2H, d, $J = 7.8$ Hz, 3',7'-H) ¹³C NMR: see Table 1. The aq. layer was neutralized with satd $\text{Ba}(\text{OH})_2$. The ppt was filtered off and the soln evapd to give a mixture of 12 and 13. The sugar mixture (12 and 13) was chromatographed on silica gel with H_2O -MeOH- CHCl_3 (1:3:18 (v/v), lower layer) to separate 12 (1.5 mg) and 13 (2.0 mg). Compound 12 (R_{f1} ; 0.46, R_{f2} ; 0.41); mp 67.0-71.0° FDMS m/z 162 [M]⁺. Compound 13 (R_{f1} ; 0.31, R_{f2} ; 0.45); [α]_D +29.8° (H_2O , c 0.20) FDMS m/z 322 (M^+). ¹³C NMR see Table 3.

Absolute configuration of cymarose (12). A soln of 12 (1.0 mg) in

MeOH (2 ml) was allowed to react with Amberlite IR-120 (H^+) at room temp. for 1 hr and then warmed at 60° for 5 min. After removal of the resin by filtration, the reaction mixture was evapd to give a mixture of methyl cymarosides. The product was dissolved in dry toluene (0.5 ml) and allowed to react with 3,5-dinitrophenylisocyanate (ca 2 mg) in the presence of dry pyridine (0.05 ml) at 60° for 1 hr, then the solvent was evapd off to afford a mixture of carbamates. Purification of the mixture was carried out by preparative TLC [solvent: hexane-EtOAc (2:1)], to collect bands with R_f 0.25 and 0.38 to yield carbamate mixtures. The combined mixtures were analysed by a chiral HPLC column SUMIPAX OA-4100 (5 μ , ϕ 4 mm i.d. \times 25 cm), mobile phase hexane-EtOH (40:1) flow rate 1 ml/min, monitored by absorption at 254 nm.

The other cymarosides obtained from 3-7 and 9, a mixture of carbamates was prepared and analysed as described above.

Bouceroside-ADC (2) Amorphous powder, mp 132.0-135.5° [α]_D -12.5° (CHCl_3 , c 1.00) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 282 (2.98), 278 (3.08), 233 (4.14) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 3450, 1700, 1600, 1580, 1280, FDMS: m/z 941 [$\text{M} + \text{K}$]⁺, 925 [$\text{M} + \text{Na}$]⁺ ¹H NMR (500 MHz, CDCl_3) δ 1.01 (3H, s, 19-H), 1.22, 1.26, 1.28 (each 3H, d, $J = 6.3$ Hz, 6-H of sugar moiety), 1.25 (3H, d, $J = 6.8$ Hz, 21-H), 1.41 (3H, s, 18-H), 3.42, 3.45, 3.66 (each 3H, s, 3-OMe of sugar moiety), 4.59 (1H, d, $J = 7.8$ Hz, allo-1-H), 4.76, 4.86 (each 1H, dd, $J = 9.8, 2.0$ Hz, cym-1-H), 4.77 (1H, dd, $J = 12.2, 4.4$ Hz, 12-H), 5.44 (1H, br s, 6-H), 7.45 (2H, t, $J = 7.3$ Hz, 4',6'-H), 7.57 (1H, t, $J = 7.3$ Hz, 5'-H), 8.06 (2H, d, $J = 7.3$ Hz, 3',7'-H) ¹³C NMR: see Tables 1 and 2.

Acidic hydrolysis of 2 A soln of 2 (32.0 mg) in MeOH (6 ml) was allowed to react in the same way as 1, and the products separated to provide 14 (10.8 mg), 12 (1.0 mg) and 13 (3.5 mg). 14 (R_{f1} ; 0.51, R_{f2} ; 0.51), an amorphous powder, mp 140.0-144.0° [α]_D -46.4° (CHCl_3 , c 1.08) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 282 (2.90), 275 (2.96), 230 (3.92) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 3400, 1715, 1600, 1585, 1270. EIMS: m/z 454 [M]⁺, 436 [$\text{M} - \text{H}_2\text{O}$]⁺, 418 [$\text{M} - \text{H}_2\text{O}$]⁺, 332 [$\text{M} - \text{C}_6\text{H}_5\text{COOH}$]⁺, 105 [$\text{C}_6\text{H}_5\text{CO}$, base peak]⁺, 77 [C_6H_5]⁺ HREIMS: Calcd for $\text{C}_{28}\text{H}_{38}\text{O}_5$, 454.6094, Found 454.2718 ¹H NMR (500 MHz, CDCl_3) δ 1.03 (3H, s, 19-H), 1.25 (3H, d, $J = 6.8$ Hz, 21-H), 1.41 (3H, s, 18-H), 3.54 (1H, m, 3-H), 3.85 (1H, dq, $J = 9.6, 6.4$ Hz, 20-H), 4.78 (1H, dd, $J = 12.2, 4.4$ Hz, 12-H), 5.44 (1H, br s, 6-H), 7.45 (2H, t, $J = 7.3$ Hz, 4',6'-H), 7.58 (1H, t, $J = 7.3$ Hz, 5'-H), 8.08 (2H, d, $J = 7.3$ Hz, 3',7'-H) ¹³C NMR: see Table 1.

Bouceroside-ANO (3) Amorphous powder, mp 113.5-116.0° [α]_D -12.4° (CHCl_3 , c 1.02) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 282 (2.95), 274 (3.05), 230 (4.12) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 3400, 1710, 1600, 1580, 1275. FDMS: m/z 927 [$\text{M} + \text{Na}$]⁺. ¹H NMR (500 MHz, CDCl_3) δ 0.82 (3H, s, 19-H), 1.22 (3H, d, $J = 6.8$ Hz, 21-H), 1.37 (3H, s, 18-H), 3.38, 3.44, 3.66 (each 3H, s, 3-OMe of sugar moiety), 4.47 (1H, dd, $J = 9.8, 2.0$ Hz, ole-1-H), 4.72 (1H, dd, $J = 12.2, 4.4$ Hz, 12-H), 4.79 (1H, d, $J = 7.8$ Hz, allo-1-H), 4.86 (1H, dd, $J = 9.8, 2.0$ Hz, cym-1-H), 7.44 (2H, t, $J = 7.3$ Hz, 4',6'-H), 7.56 (1H, t, $J = 7.3$ Hz, 5'-H), 8.05 (1H, d, $J = 7.3$ Hz, 3',7'-H) ¹³C NMR see Tables 1 and 2.

Acidic hydrolysis of 3. A soln of 3 (15.2 mg) in MeOH (6 ml) was allowed to react in the same way as 1, and products were separated to provide 11 (2.0 mg), 12 (1.0 mg) and 16 (1.7 mg).

Bouceroside-ADO (4) Amorphous powder; mp 107.5-111.0° [α]_D -11.8° (CHCl_3 , c 1.03) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 282 (2.87), 275 (2.97), 230 (3.90) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 3400, 1710, 1600, 1580, 1270. FDMS: m/z 925 [$\text{M} + \text{Na}$]⁺. ¹H NMR (500 MHz, CDCl_3) δ 1.01 (3H, s, 19-H), 1.23, 1.24, 1.26 \times 2 (each 3H, d, $J = 6.3$ Hz, 21-H and 6-H of sugar moiety), 1.41 (3H, s, 18-H), 3.38, 3.45, 3.66 (each 3H, s, 3-OMe of sugar moiety), 4.47 (1H, dd, $J = 9.8, 2.0$ Hz, allo-1-H), 4.76 (1H, dd, $J = 12.2, 4.4$ Hz, 12-H), 4.86 (1H, dd, $J = 9, 1.5$ Hz, cym-1-H), 5.44 (1H, br s, 6-H), 7.45 (2H, t, $J = 7.8$ Hz, 4',

6'-H), 7.57 (1H, *t*, *J* = 7.8 Hz, 5'-H), 8.06 (2H, *d*, *J* = 7.8 Hz, 3', 7'-H) ¹³C NMR see Tables 1 and 2

Acidic hydrolysis of 4 A soln of **4** (12.1 mg) in MeOH (6 ml) was allowed to react in the same way as **1**, and products were separated to provide **14** (2.5 mg), **12** (1.1 mg) and **16** (1.9 mg)

Bouceroside-BNO (5) Amorphous powder, mp 133.5–137.0°, [α]_D + 2.4° (CHCl₃, *c* 1.02) (Found, C, 62.68, H, 8.25 C₅₁H₇₈O₁₆ · 3/2H₂O required C, 64.69, H, 8.24) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ) 281 (2.79), 275 (2.90), 230 (3.97) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 3400, 1710, 1600, 1580, 1275 FDMS *m/z* 946 (M⁺), 969 [M + Na]⁺ ¹H NMR (500 MHz, CDCl₃) δ 0.80 (3H, *s*, 19-H), 1.12 (3H, *d*, *J* = 6.3 Hz, 21-H), 1.14 (3H, *s*, 18-H), 1.23 (3H, *d*, *J* = 6.3 Hz, 6-H of sugar moiety), 1.26 (6H, *d*, *J* = 6.3 Hz, 6-H of sugar moiety), 1.93 (3H, *s*, 2''-H), 3.39, 3.42, 3.66 (each 3H, *s*, 3-OMe of sugar moiety), 3.55, 3.56, 3.87 (each 1H, *dq*, *J* = 9.4, 6.3 Hz, 5-H of sugar moiety), 4.48 (1H, *dd*, *J* = 9.8, 2.0 Hz, ole-1-H), 4.76 (1H, *dd*, *J* = 12.2, 4.4 Hz, 12-H), 4.79 (1H, *d*, *J* = 7.8 Hz, allo-1-H), 4.86 (1H, *dd*, *J* = 9.8, 2.0 Hz, cym-1-H), 4.87 (1H, *dq*, *J* = 9.4, 6.3 Hz, 20-H), 7.44 (2H, *t*, *J* = 7.3 Hz, 4',6'-H), 7.56 (1H, *t*, *J* = 7.3 Hz, 5'-H), 8.09 (2H, *d*, *J* = 7.3 Hz, 3',7'-H) ¹³C NMR see Tables 2 and 4

Acidic hydrolysis of 5 A soln of **5** (47.1 mg) in MeOH (6 ml) was allowed to react in the same way as **1**, and the products sep'd to provide **15** (22.6 mg), **12** (3.1 mg) and **16** (5.0 mg) Compound **15** (*R*_{f1}, 0.61, *R*_{f2}, 0.52), amorphous powder, mp 118.5–123.0°, [α]_D + 8.0° (CHCl₃, *c* 0.50) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ) 281 (2.73), 275 (2.77), 229 (3.83) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 3400, 1715, 1600, 1580, 1275 FDMS *m/z* 498 (M⁺) ¹H NMR (270 MHz, CDCl₃) δ 0.82 (3H, *s*, 19-H), 1.12 (3H, *d*, *J* = 6.2 Hz, 21-H), 1.15 (3H, *s*, 18-H), 1.94 (3H, *s*, 2''-H), 3.61 (1H, *m*, 3-H), 4.76 (1H, *dd*, *J* = 12.2, 4.4 Hz, 12-H), 4.88 (1H, *dq*, *J* = 9.6, 2 Hz, 20-H), 7.45 (2H, *t*, *J* = 7.3 Hz, 4',6'-H), 7.58 (1H, *t*, *J* = 7.3 Hz, 5'-H), 8.10 (2H, *d*, *J* = 7.3 Hz, 3',7'-H), (270 MHz, pyridine) δ 0.82 (3H, *s*, 19-H), 1.16 (3H, *d*, *J* = 6.2 Hz, 21-H), 1.50 (3H, *s*, 18-H), 2.00 (3H, *s*, 2''-H), 3.83 (1H, *m*, 3-H), 5.12 (1H, *dd*, *J* = 12.2, 4.4 Hz, 12-H), 5.42 (1H, *dq*, *J* = 9.4, 6.3 Hz, 20-H), 7.54 (2H, *t*, *J* = 7.0 Hz, 4',6'-H), 7.62 (1H, *t*, *J* = 7.0 Hz, 5'-H), 8.44 (2H, *d*, *J* = 7.0 Hz, 3',7'-H) Compound **16** [α]_D - 10.0° (H₂O, *c* 0.47), FDMS *m/z* 322 [M⁺], ¹³C NMR see Table 3

Bouceroside-BDO (6) Amorphous powder, mp 135.5–139.0°, [α]_D - 21.0° (CHCl₃, *c* 1.05) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ) 278 (3.04), 272 (3.16), 230 (4.17) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 3400, 1710, 1600, 1580, 1275 FDMS *m/z* 944 [M⁺] ¹H NMR (500 MHz, CDCl₃) δ 1.00 (3H, *s*, 19-H), 1.13 (3H, *d*, *J* = 6.3 Hz, 21-H), 1.17 (3H, *s*, 18-H), 1.23 (3H, *d*, *J* = 6.3 Hz, 6-H of sugar moiety), 1.26 (6H, *d*, *J* = 6.3 Hz, 6-H of sugar moiety), 1.97 (3H, *s*, 2''-H), 3.39, 3.45, 3.66 (each 3H, *s*, 3-OMe of sugar moiety), 3.87 (1H, *dq*, *J* = 9.8, 6.3 Hz, 5-H of sugar moiety), 4.48 (1H, *dd*, *J* = 9.8, 2.0 Hz, ole-1-H), 4.79 (1H, *dd*, *J* = 7.8 Hz, allo-1-H), 4.80 (1H, *dd*, *J* = 12.2, 4.4 Hz, 12-H), 4.86 (1H, *dd*, *J* = 9.1, 5 Hz, cym-1-H), 4.91 (1H, *dq*, *J* = 9.8, 6.3 Hz, 20-H), 5.44 (1H, *br s*, 6-H), 7.45 (2H, *t*, *J* = 7.3 Hz, 4',6'-H), 7.57 (1H, *t*, *J* = 7.3 Hz, 5'-H), 8.11 (2H, *d*, *J* = 7.3 Hz, 3',7'-H) ¹³C NMR see Tables 2 and 4

Acidic hydrolysis of 6 A soln of **6** (36.5 mg) in MeOH (6 ml) was allowed to react in the same way as **1**, and the products sep'd to provide **17** (8.2 mg), **12** (2.0 mg) and **16** (4.2 mg) Compound **17** (*R*_{f1}, 0.61, *R*_{f2}, 0.52), amorphous powder, mp 120.0–123.5° [α]_D - 10.3° (CHCl₃, *c* 0.31) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ) 281 (2.68), 274 (3.01), 228 (4.14) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 3400, 1715, 1600, 1580, 1275 EIMS *m/z* 496 [M⁺], 374, 314, 296, 105 [C₆H₅CO, base peak]⁺ ¹H NMR (270 MHz, CDCl₃) δ 1.01 (3H, *s*, 19-H), 1.13 (3H, *d*, *J* = 6.2 Hz, 21-H), 1.18 (3H, *s*, 18-H), 1.98 (3H, *s*, 2''-H), 3.53 (1H, *m*, 3-H), 4.81 (1H, *dd*, *J* = 12.2, 4.4 Hz, 12-H), 4.91 (1H, *dq*, *J* = 9.6, 2 Hz, 20-H), 5.43 (1H, *br s*, 6-H), 7.45 (2H, *t*, *J* = 7.3 Hz, 4',6'-H), 7.59 (1H, *t*, *J* = 7.3 Hz, 5'-H), 8.11 (2H, *d*, *J* = 7.3 Hz, 3',7'-H) ¹³C NMR see Table 4

Bouceroside-BNC (7) Amorphous powder, mp 138.5–141.0°, [α]_D + 18.5° (CHCl₃, *c* 1.03) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ) 281 (2.69), 274

(3.04), 230 (4.14) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 3400, 1715, 1600, 1580, 1275 FDMS *m/z* 946 [M⁺], 969 [M + Na]⁺ ¹H NMR (500 MHz, CDCl₃) δ 0.80 (3H, *s*, 19-H), 1.12 (3H, *d*, *J* = 6.3 Hz, 21-H), 1.14 (3H, *s*, 18-H), 1.22, 1.27, 1.28 (each 3H, *d*, *J* = 6.3 Hz, 6-H of sugar moiety), 1.92 (3H, *s*, 2''-H), 3.42, 3.44, 3.66 (each 3H, *s*, 3-OMe of sugar moiety), 4.59 (1H, *d*, *J* = 7.8 Hz, allo-1-H), 4.76 (1H, *dd*, *J* = 12.2, 4.4 Hz, 12-H), 4.77, 4.85 (each 1H, *dd*, *J* = 9.8, 2.0 Hz, cym-1-H), 4.88 (1H, *dq*, *J* = 9.2, 6.3 Hz, 20-H), 7.44 (2H, *t*, *J* = 7.8 Hz, 4',6'-H), 7.57 (1H, *t*, *J* = 7.8 Hz, 5'-H), 8.09 (2H, *d*, *J* = 7.8 Hz, 3',7'-H) ¹³C NMR see Tables 2 and 4

Acidic hydrolysis of 7 A soln of **7** (15.1 mg) in MeOH (6 ml) was allowed to react in the same way as **1**, and the products were separated to provide **15** (3.1 mg), **12** (1.1 mg) and **13** (2.5 mg)

Bouceroside-BDC (8) Amorphous powder, mp 103.5–106.0°, [α]_D + 2.1° (CHCl₃, *c* 0.32) (Found, C, 61.18, H, 7.97 C₅₁H₇₆O₁₆ · 3/2H₂O required C, 64.89, H, 8.11) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ) 281 (2.57), 274 (2.65), 230 (3.75) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 3400, 1715, 1600, 1580, 1280 FABMS *m/z* 967 [M + Na]⁺ ¹H NMR (500 MHz, CDCl₃) δ 1.00 (3H, *s*, 19-H), 1.14 (3H, *d*, *J* = 6.3 Hz, 21-H), 1.17 (3H, *s*, 18-H), 1.22, 1.27, 1.28 (each 3H, *d*, *J* = 6.3 Hz, 6-H of sugar moiety), 1.97 (3H, *s*, 2''-H), 3.42, 3.45, 3.66 (each 3H, *s*, 3-OMe of sugar moiety), 3.85, 3.91 (each 1H, *dq*, *J* = 9.3, 6.3 Hz, 5-H of sugar moiety), 4.59 (1H, *d*, *J* = 7.8 Hz, allo-1-H), 4.76 (1H, *dd*, *J* = 9.8, 2.0 Hz, ole-1-H), 4.81 (1H, *dd*, *J* = 12.2, 4.4 Hz, 12-H), 4.85 (1H, *dd*, *J* = 9.1, 5 Hz, cym-1-H), 4.91 (1H, *dq*, *J* = 9.7, 6.3 Hz, 20-H), 5.44 (1H, *br s*, 6-H), 7.45 (2H, *t*, *J* = 7.3 Hz, 4',6'-H), 7.58 (1H, *t*, *J* = 7.3 Hz, 5'-H), 8.11 (2H, *d*, *J* = 7.3 Hz, 3',7'-H)

Acidic hydrolysis of 8 A soln of **8** (1.4 mg) in MeOH (3 ml) was allowed to react in the same way as **1**, and the products (**17**, **12** and **13**) were identified by TLC comparison with the authentic samples

Bouceroside-CNO (9) Amorphous powder, mp 143.5–147.0°, [α]_D - 7.5° (CHCl₃, *c* 0.96) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ) 282 (3.40), 275 (3.46), 228 (4.60) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 3400, 1715, 1600, 1580, 1275 FDMS *m/z* 1031 [M + Na]⁺, 1008 [M]⁺ ¹H NMR (500 MHz, CDCl₃) δ 0.78 (3H, *s*, 19-H), 1.15 (3H, *s*, 18-H), 1.23 (3H, *d*, *J* = 6.3 Hz, 21-H), 3.38, 3.44, 3.66 (each 3H, *s*, 3-OMe of sugar moiety), 3.55, 3.87 (each 1H, *dq*, *J* = 9.8, 6.3 Hz, 5-H of sugar moiety), 4.47 (1H, *dd*, *J* = 9.8, 2.0 Hz, ole-1-H), 4.79 (1H, *d*, *J* = 7.8 Hz, allo-1-H), 4.85 (1H, *dd*, *J* = 9.8, 2.0 Hz, cym-1-H), 4.91 (1H, *dd*, *J* = 12.2, 4.4 Hz, 12-H), 5.25 (1H, *dq*, *J* = 9.8, 6.3 Hz, 20-H), 7.21, 7.44 (each 2H, *t*, *J* = 7.8 Hz, 4',6' and 4'',6''-H), 7.48, 7.61 (each 1H, *t*, *J* = 7.8 Hz, 5' and 5''-H), 7.83, 8.06 (each 2H, *d*, *J* = 7.8 Hz, 3',7' and 3'',7''-H) ¹³C NMR see Tables 2 and 5

Acidic hydrolysis of 9 A soln of **9** (20.3 mg) in MeOH (6 ml) was allowed to react in the same way as **1**, and the products sep'd to provide **18** (6.5 mg), **12** (1.5 mg) and **16** (3.6 mg) Compound **18** (*R*_{f1}, 0.65, *R*_{f2}, 0.58) amorphous powder, mp 133.5–137.5°, [α]_D - 5.7° (CHCl₃, *c* 0.65) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ) 281 (3.15), 278 (3.23), 260 (3.26), 256 (3.35), 229 (4.35) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 3400, 1715, 1600, 1580, 1275 EIMS *m/z* 438 [M - C₆H₅COOH]⁺ 316 [438 - C₆H₅COOH], 298 [316 - H₂O], 105 [C₆H₅CO, base peak]⁺ ¹H NMR (270 MHz, CDCl₃) δ 0.79 (3H, *s*, 19-H), 1.16 (3H, *s*, 18-H), 1.23 (3H, *d*, *J* = 6.3 Hz, 21-H), 3.60 (1H, *m*, 3-H), 4.93 (1H, *dd*, *J* = 12.2, 4.4 Hz, 12-H), 5.26 (1H, *dq*, *J* = 9.8, 6.3 Hz, 20-H), 7.21, 7.45 (each 2H, *t*, *J* = 7.7 Hz, 4',6' and 4'',6''-H), 7.50, 7.65 (each 1H, *t*, *J* = 7.7 Hz, 5' and 5''-H), 7.84, 8.06 (each 2H, *d*, *J* = 7.7 Hz, 3',7' and 3'',7''-H) ¹³C NMR see Table 5

Bouceroside-CNC (10) Amorphous powder, mp 114.0–117.5°, [α]_D + 8.0° (CHCl₃, *c* 0.60) UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ) 281 (3.53), 273 (3.60), 228 (4.65) IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 3400, 1715, 1610, 1590, 1280 FABMS *m/z* 1031 [M + Na], ¹H NMR (270 MHz, CDCl₃) δ 0.78 (3H, *s*, 19-H), 1.14 (3H, *s*, 18-H), 1.22 (3H, *d*, *J* = 8.0 Hz, 21-H), 3.42, 3.44, 3.66 (each 3H, *s*, 3-OMe of sugar moiety), 3.84, 3.91 (each 1H, *dq*, *J* = 9.8, 6.3 Hz, 5-H of sugar moiety), 4.59 (1H, *d*, *J*

= 8 Hz, allo-1-H), 4.76, 4.85 (1H, *dd*, $J = 9.5, 2.0$ Hz, cym-1-H), 4.91 (1H, *dd*, $J = 12, 4$ Hz, 12-H), 5.25 (1H, *dq*, $J = 10, 6.4$ Hz, 20-H), 7.24, 7.45 (each 2H, *t*, $J = 8$ Hz, 4',6' and 4'',6''-H), 7.48, 7.62 (each 1H, *t*, $J = 8$ Hz, 5' and 5''-H), 7.83, 8.06 (each 2H, *d*, $J = 8$ Hz, 3',7' and 3'',7''-H) ^{13}C NMR: see Tables 2 and 5.

Acidic hydrolysis of 10. A soln of **10** (1.8 mg) in MeOH (6 ml) was allowed to react in the same way as **1**, and products (**18**, **12** and **13**) were identified by TLC comparison with the authentic samples

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