This article was downloaded by: [University of Saskatchewan Library] On: 03 July 2012, At: 01:16 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Organic Preparations and Procedures International: The New Journal for **Organic Synthesis**

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/uopp20

PRACTICAL ROUTE TO A β-KETOPHOSPHONATE, A KEY INTERMEDIATE FOR THE TOTAL SYNTHESIS OF 20(S)-CPT AND RELATED **ANALOGUES**

Ming Huo^a, Yun-Yan Kuang^a & Fen-Er Chen^a ^a Department of Chemistry, Fudan University, Shanghai, 200433, P R CHINA

Version of record first published: 21 Feb 2009

To cite this article: Ming Huo, Yun-Yan Kuang & Fen-Er Chen (2004): PRACTICAL ROUTE TO A β-KETOPHOSPHONATE, A KEY INTERMEDIATE FOR THE TOTAL SYNTHESIS OF 20(S)-CPT AND RELATED ANALOGUES, Organic Preparations and Procedures International: The New Journal for Organic Synthesis, 36:4, 331-335

To link to this article: <u>http://dx.doi.org/10.1080/00304940409458674</u>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.tandfonline.com/page/terms-and-conditions

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

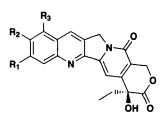
The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

PRACTICAL ROUTE TO A β-KETOPHOSPHONATE, A KEY INTERMEDIATE FOR THE TOTAL SYNTHESIS OF 20(S)-CPT AND RELATED ANALOGUES

Ming Huo, Yun-Yan Kuang and Fen-Er Chen*

Department of Chemistry, Fudan University Shanghai, 200433, P. R. CHINA

As part of a program for the total synthesis of pentacyclic alkaloids, we were interested in the 20(S)-camptothecin series, such as 20(S)-camptothecin (**A**), 20(S)-9-nitrocamptothecin (**B**) and 20(S)-10-hydroxycamptothecin (**C**)¹, in which the β -ketophosphonate (**8**) had been established as a key intermediate in the synthesis of these compounds. Ciufolini and Roschanger obtained this compound from propionanilide in 43% overall yield,² while Boger and Hong has prepared it from *o*-amino-benzaldehyde in 47% overall yield.³ Although both yields are



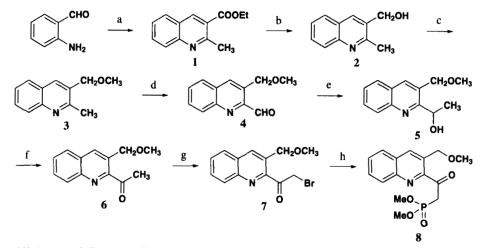
A. $R_1 = R_2 = R_3 = H$, 20(S)-CPY B. $R_1 = R_2 = H$, $R_3 = NO_2$, 9-NO₂-CPT C. $R_1 = R_3 = H$, $R_2 = OH$, 10-OH-CPT

moderate, the use of chromatographic separation, expensive catalyst $(Pd(dppp)_2Cl_2)$ and low temperatures (-78°C) makes these routes difficult for large-scale preparations. We now describe a new and efficient procedure for the synthesis of **8** from commercially available *o*-aminobenzaldehyde (*Scheme 1*).

Compound 8 was obtained as follows. The Friedlander condensation of *o*-aminobenzaldehyde with ethyl acetoacetate and subsequent reduction of the crude ester 1 with NaBH₄/CaCl₂/EtOH provided alcohol 2 in 70% yield. Treatment of 2 with MeI in DMSO led smoothly to the formation of the corresponding ether derivatives 3 in nearly quantitative yield. Oxidation of 3 with SeO₂ in dioxane at 80-85°C for 3 hrs afforded the aldehyde 4 in 92% yield. Addition of MeMgI to 4, followed by the oxidation of the resulting crude alcohol 5 with CrO₃/AcOH furnished ethyl ketone 6 in 64% yield, which was easily converted to compound 7

^{© 2004} by Organic Preparations and Procedures Inc.

through NBS bromination. Finally, protection of the carbonyl group of 7 with $PhSO_2NHNH_2$ to form the sulfonylhydrazone intermediate, followed by Arbukov reaction with trimethyl phosphite, and deprotection to afford the desired β -ketophosphonate 8 in 87% yield.⁴



a) CH₃COCH₂CO₂Et, NaOEt, EtOH, -5°C to reflux, 2 h, 80%; b) NaBH₄, CaCl₂, EtOH, r.t, 26 h, 87%; c) MeI, KOH, DMSO, r.t, 3 h, 95%; d) SeO₂, dioxane, reflux, 2 h, 92%; e) MeMgI, Et₂O, -5°C to reflux, 3 h, 85%; f) CrO₃, glacial AcOH, 45-50°C, 2 h, 75%; g) NBS, CCl₄, benzoyl peroxide, reflux, 12 h, 87%; h) PhSO₂NHNH₂, P(OMe)₃, NaOCl, THF, 20°C to reflux, 7 h, 87%

Scheme 1

In summary, an efficient synthesis of β -ketophosphonate **8** was accomplished in an overall yield of 30% from *o*-aminobenzaldehyde in eight steps. The superiority of this route stems from the fact that commercially available materials were used and that it does not require chromatographic separation, proceeds under the mild reaction conditions and avoids the use of expensive catalyst and the rigorous temperature conditions of the Ciufolini and Boger's routes.

EXPERIMENTAL SECTION

Mps were measured on a WRS-1B digital melting point apparatus and are uncorrected. Elemental analyses were performed on a Carlo-Erba 1006 elemental analyzer. ¹H NMR spectra were recorded on a Bruker DMX500 using TMS as an internal standard. Chemical shifts (δ) are expressed in ppm. GC-MS spectra were obtained on Finnigan Voyager instrument.

2-Methyl 3-quinolinecarboxylate (1).- To a stirred solution of *o*-aminobenzaldehyde (10.95 g, 90 mmol) in EtOH (12 mL) were added dropwise ethyl acetoacetate (12.5 g, 96 mmol) in EtOH (12 mL) and 10% sodium ethoxide solution (10 mL). The reaction mixture was stirred under reflux for 2 h. After removal of the solvent under vacuum, water (10 mL) was added, and the mixture was allowed to stand at 0°C overnight. The precipitate was collected and recrystallized from EtOAc-hexane to give pure 1 (15.5 g, 80%) as pale yellow needles, mp 71-72°C, *lit*.⁵ 70-72°C.

¹H NMR (DMSO-d₆): δ 8.74 (s, 1H), 8.07-8.05 (m, 1H), 7.88-7.86 (m, 1H), 7.80-7.77 (m, 1H), 7.56-7.53 (m, 1H), 4.47-4.42 (q, 2H), 3.01 (s, 3H), 1.47-1.45 (t, 3H). GC-MS (m/z): 215(M⁺), 200(M⁺-Me), 186(M⁺-Et), 171(186-Me), 142(M⁺-CO₂Et), 127(142-Me).

Anal. Calcd for C₁₃H₁₃NO₂: C, 72.55; H, 6.04; N, 6.51. Found: C, 72.75; H, 6.15; N, 6.82

2-Methyl-3-hydroxymethylquinoline (2).- To a stirred solution of NaBH₄ (5.1 g, 138 mmol) in EtOH (20 mL) were added at 0-5°C, a solution of **1** (10.75 g, 50 mmol) in EtOH (75 mL) and a solution of anhydrous CaCl₂ (7.38 g, 60 mmol) in EtOH (75 mL). The reaction mixture was allowed to warm up to room temperature and stirred for 26 h. After concentration of the solution under reduced pressure to a volume of approximately 75 mL, CHCl₃ (200 mL) and water (300 mL) were added and the organic layer was washed with saturated aq. NaCl (100 mL x 3) and water (50 mL x 3), then dried over MgSO₄. The solvent was concentrated *in vacuo* to give the crude product. Recrystallization from EtOH gave the pure **2** (7.54 g, 87%) as a pale yellow crystalline powder, mp 140-142°C, *lit.*⁶ 140-142°C.

¹H NMR (CDCl₃): δ 8.11 (s, 1H), 8.04-8.01 (m, 1H), 7.78-7.74 (m, 1H), 7.67-7.66 (m, 1H), 7.51-7.46 (m, 1H), 4.87 (s, 2H), 2.71 (s, 3H), 2.50 (bs, 1H). GC-MS (m/z): 173(M⁺), 156(M⁺-OH), 142(156-CH₂), 127(142-CH₂)

Anal. Calcd for C₁₁H₁₁NO: C, 76.30; H, 6.35; N, 8.09. Found: C, 76.53; H, 6.22; N, 8.21

2-Methyl-3-methoxymethylquinoline (3).- To a stirred suspension of KOH (4.48 g, 40 mmol) in anhydrous DMSO (50 mL) were added **2** (3.46 g, 20 mmol) and MeI (5.68 g, 40 mmol). The reaction mixture was stirred at room temperature for 3 h. Water (60 mL) was added, and the mixture was extracted with CH_2Cl_2 (40 mL x 4). The combined organic extracts were washed with water (30 mL x 5), dried over MgSO₄, The solvent was concentrated to dryness under *vacuo*. The crude product was purified over silica gel [PE(60-90°C), Rf 0.48] to gave pure **3** as a viscous yellow oil (3.55 g, 95%).

¹H NMR (CDCl₃): δ 8.04-8.01 (m, 2H), 7.78-7.74 (m, 1H), 7.66-7.63 (m, 1H), 7.49-7.44 (m, 1H), 4.58 (s, 2H), 3.47 (s, 3H), 2.70 (s, 3H). MS (m/e): 187(M⁺), 172(M⁺-CH₃), 156(M⁺-OCH₃), 141(156-CH₃).

Anal. Calcd for C₁₂H₁₃NO: C, 77.00; H, 6.95; N, 7.49. Found: C, 77.12; H, 6.89; N, 7.53

2-Formyl-3-methoxymethylquinoline (4).- To a stirred suspension of SeO₂ (3.33 g, 30 mmol) in dioxane (150 mL) was added at 50-55°C, a solution of **3** (2.91 g, 16.8 mmol) in dioxane (50 mL). The reaction mixture was stirred at 80-85°C for 3 h, and cooled to room temperature. The reaction mixture was filtered through the Celite, and the solvent was concentrated under *vacuo*. CH_2Cl_2 (120 mL) and Water (80 mL) was added to the reaction mixture. The organic layer was washed with saturated aq. NaCl (100 mL x 3) and water (50 mL x3), then dried over MgSO₄. The solvent was concentrated under *vacuo* to a crude solid which was recrystallized from EtOAc to afford **4** (3.04 g, 92%) as yellow crystals, mp. 80-81°C

¹H NMR (DMSO-d₆): δ 10.29 (s, 1H), 8.48 (s, 1H), 8.23-8.21 (m, 1H), 7.91-7.89 (m, 1H), 7.80-7.77 (m, 1H), 7.70-7.67 (m, 1H), 5.05 (s, 2H), 3.60 (s, 3H). GC-MS (m/z): 201(M⁺), 186(M⁺-

CH₃), 172(M⁺-CHO), 170(M⁺-OCH₃), 141(172-OCH₃)

Anal. Calcd for C₁₂H₁₁NO₂: C, 71.64; H, 5.47; N, 6.96. Found: C, 71.78; H, 5.35, N, 7.04.

2-(α -Hydroxyethyl)-3-methoxymethylquinoline (5).- To a stirred solution of Grignard reagent [prepared from CH₃I (3.98 g, 28 mmol) and Mg (0.68 g, 28 mmol) in anhydrous. Et₂O (20 mL) with stirring at reflux for 30 minutes under nitrogen] at 0°C, was added a solution of 4 (2.01 g, 10 mmol) in Et₂O (30 mL). The reaction mixture was stirred under reflux for 4 h, and cooled to room temperature. After addition of saturated aq. NH₄Cl (7.5 mL) and 2N HCl (20 mL), the mixture was extracted with CHCl₃ (75 mL). The organic layer was washed with saturated aq. NaHCO₃ (50 mL x3) and water (50 mL x 3), then dried over Na₂SO₄. The solvent was concentrated under *vacuo* and the oily residue was chromatographed on silica gel (AcOEt:cyclohexane, 2:5) to give the pure 5 (1.85 g, 85.2%) as a pale yellow oil.

¹H NMR (DMSO-d₆): δ 8.18 (s, 1H), 8.07-8.05 (m, 1H), 7.83-7.82 (m, 1H), 7.72-7.69 (m, 1H), 7.55-7.52 (m, 1H), 5.17-5.13 (q, 1H), 4.95 (bs, 1H), 4.66-4.60 (s, 2H), 3.49 (s, 3H), 1.52-1.50 (d, 3H). GC-MS (m/z): 217(M⁺), 202(M⁺-CH₃), 199(M⁺-OH), 186(M⁺-OCH₃), 184(199-CH₃), 172(186-CH₂), 141(184-OCH₃)

Anal. Calcd for C₁₃H₁₅NO₂: C, 71.89; H, 6.91; N, 6.45. Found: C, 71.99; H, 6.84; N, 6.38.

2-Acetyl-3-methoxymethylquinoline (6).- To a stirred solution of **5** (2.73 g, 12.7 mmol) in glacial AcOH (10 mL) was added CrO_3 (1.0 g, 10 mmol) in portions at 45-50°C over a period of 2 h. Stirring was continued at 50°C for an additional 3 h and then after cooling to room temperature, CHCl₃ (75 mL) and water (50 mL) were added to the reaction mixture. The organic layer was washed with saturated aq. NaCl (15 mL x 3) and water (15 mL x 3), then dried over Na₂SO₄. The solvent was concentrated *in vacuo* to give the crude product. Recrystallization from EtOAchexane gave pure **6** (2.04 g, 75%) as pale yellow crystals, mp 56-57°C.

¹H NMR (DMSO-d₆): δ 8.40 (s, 1H), 8.13-8.11 (m, 1H), 7.83-7.81 (m, 1H), 7.72-7.70 (m, 1H), 7.61-7.59 (m, 1H), 4.93 (s, 2H), 3.55 (s, 3H), 2.84 (s, 3H). GC-MS (m/z): 215(M⁺), 200(M⁺-CH₃), 184(M⁺-OCH₃), 172(200-CO), 169(184-CH₃), 141(169-CO)

Anal. Calcd for C₁₃H₁₃NO₅: C, 72.56; H, 6.05; N, 6.51. Found: C, 72.46; H, 5.99; N, 6.57.

2-(Bromoacetyl)-3-methoxymethylquinoline (7).- A mixture of **6** (1.08 g, 5.0 mmol), NBS (0.89 g, 5.0 mmol), benzoyl peroxide (0.1 g) and anhydrous CCl_4 (20 mL) was stirred under reflux for 12 h, and then cooled to room temperature. After removal of succinimide, the filtrate was concentrated under reduced pressure to give the crude product. Recrystallization from EtOAc-hexane gave **7** (1.27 g, 87%) as a pale yellow solid, mp 74-75°C.

¹H NMR (DMSO-d₆): δ 8.63 (s, 1H), 8.14-8.13 (m, 1H), 7.85-7.84 (m, 1H), 7.74-7.73 (m, 1H), 7.58-7.55 (m, 1H), 4.97 (s, 2H), 3.95 (s, 3H), 2.88 (s, 2H).GC-MS (m/z): 294(M⁺), 214(M⁺-Br), 200(214-CH₂), 183(214-OCH₃), 172(200-CO), 169(183-CH₂), 155(169-CH₂), 127(155-CO)

Anal. Calcd for C₁₃H₁₂BrNO₂: C, 53.06; H, 4.08; N, 4.76. Found: C, 53.14; H, 4.01; N, 4.89.

Dimethyl[2-[3-(Methoxymethyl)quinolin-2-yl]-2-oxoethyl]phosphonate (8).- To a stirred solution of PhSO₂NHNH₂ (0.55 g, 3.0 mmol) in THF (20 mL) was added 7 (0.88 g, 3.0 mmol).

The reaction mixture was stirred at room temperature for 3 h, then $(MeO)_3P$ (0.6 g, 4.8 mmol) was added and the mixture, and heated under reflux for a further 4 h with stirring. Then 5% aq. NaOCl (20 mL) was added and the mixture was stirred at room temperature for 30 min. After addition of 5% aq. NaHSO₃ (15 mL) and AcOEt (75 mL), the organic layer was washed with saturated aq. NaCl (100 mL x 3) and water (50 mL x 3), then dried over MgSO₄. The solvent was concentrated in *vacuo* to give the crude product. Recrystallization from EtOAc-hexane gave pure **8** (0.85 g, 87%) as a white solid, mp. 47-48°C, *lit.*³ 46-47°C.

¹H NMR (CDCl₃): δ 8.30 (s,1H), 7.98-7.96 (m, 1H), 7.69-7.68 (m, 1H), 7.58-7.56 (m, 1H), 7.48-7.47 (m, 1H), 4.79 (s, 2H), 4.07 (d, 2H), 3.61 (d, 6H), 3.40 (s, 3H); MS (m/s): 324(M⁺), 214[M⁺-P(O)(OMe)₂], 200(214-CH₂), 183(214-OCH₃),172(200-CO)

Anal. Calcd for C₁₅H₁₈NO₅P: C, 55.73; H, 5.61; N, 4.33. Found: C, 55.68; H, 5.93; N, 4.05

REFERENCES

- (a) M. E. Wall, M. C. Wani, C. E. Cook, K. H. Palmer, A. T. McPhail and G. A. Sim, J. Am. Chem., Soc., 88, 3888 (1966). (b) M. E. Wall and M. C. Wani, J. Ethnopharmacol., 51, 239 (1996). (c) M. E. Wall and M. C. Wani, Alkaloids, 50, 509 (1998).
- 2. M. A. Ciufolini and F. Roschanger. Angew. Chem. Ed. Engl., 35, 1693 (1996).
- 3. D. L. Boger and J. Y. Hong, J. Am. Chem. Soc., 120, 1218 (1998).
- (a) C. Y. Yuan and R. Y. Xie, *Phosphorus Sulfur and Silicon and The Related Elements*, 90, 47 (1994).
 (b) To avoid the competition of Perkow reaction with Arbukov rearrangement, we first protect the carbonyl group with PhSO₂NHNH₂. After Arbukov reaction, the sulfonyl hydrazone intermediate could be easily deprotected by 5% aqueous NaOCl.
- S. X. Liu, Y. J. Gao and G. J. Jiang. Chem. J. Chin. Univ., 7, 1104 (1986). Chem. Abstr.: 107, 236554n (1987).
- 6. G. K. Jnaneshwara, N. S. Shaikh, N. V. Bapat and V. H. Deshpande, J. Chem. Res. (S), 34 (2000).

(Received May 5, 2004; in final form June 7, 2004)