DOI: 10.1002/ejoc.201100049

Chemoselectivity in the Reduction of (2*S*,4*S*)-4-Amino-5-oxoproline Derivatives with Borane Complexes

Alexey Yu. Vigorov,^[a] Irina A. Nizova,^[a] Liliya Sh. Sadretdinova,^[a] Marina A. Ezhikova,^[a] Mikhail I. Kodess,^[a] Iliya N. Ganebnykh,^[a] and Victor P. Krasnov^{*[a]}

Keywords: Amino acids / Protecting groups / Lactams / Reduction / Chemoselectivity

Reduction of the carbonyl groups in *N*-protected (2S,4S)-4amino-5-oxo-1-phenylprolinates with BH₃ complexes resulted in (2S,4S)-4-aminoproline or (2S,4S)-4-aminoprolinol derivatives depending on the reaction conditions and the type of protecting groups used.

Introduction

Optically active prolines including 4-aminoproline derivatives have been widely used in organic synthesis as chiral synthons,^[1] fragments of conformationally constrained peptides,^[2] substrates of glutamate neuroreceptors,^[3] and the components of catalysts for asymmetric synthesis.^[4] One of the efficient synthetic approaches for producing various substituted prolines involves the reduction of the lactam carbonyl in the appropriate 5-oxoproline derivatives. BH₃ complexes have been used for selective reduction of *N*-Boc protected 5-oxoprolinates.^[5]

In the present work we explored the reduction of various *N*- and *C*-protected derivatives of (2*S*,4*S*)-4-amino-5-oxo-1phenylproline^[6] (1) with BH₃ complexes (Figure 1). 5-Oxo-1-phenylprolines were chosen as the substrates for our study because their reduction results in chiral 1-phenylprolines and 1-phenylprolinols. It should be noted that 1-arylprolines are of special interest for preparing catalysts for asymmetric synthesis,^[7] and they are also promising biologically active compounds.^[8] The probable reduction mechanism involves the coordination of the borane with the carbonyl oxygen atom.^[9] Accordingly, we consider that the number and type of protecting groups used can influence the spatial accessibility and the reactivity of the different carbonyl groups to thereby influence the chemoselectivity of the reduction of the lactam and ester groups.



Figure 1. (2S,4S)-4-Amino-5-oxo-1-phenylproline.

 [a] I. Ya. Postovsky Institute of Organic Synthesis of RAS, 22/20, S. Kovalevskoy/Akademicheskaya St., 620990 Ekaterinburg, Russian Federation Fax: +7-343-3741189 E-mail: ca@ios.uran.ru

Results and Discussion

N-Protected derivatives of 4-amino-5-oxo-1-phenylproline (1), such as 4-tosylamino- (2), 4-phthalimido- (3), and 4-Boc-amino- (4) (2S,4S)-5-oxo-1-phenylprolines were prepared as described previously.^[6]

Methyl esters 5 and 6 were prepared in high yields according to the standard procedure of treating *N*-tosylated 2 and *N*-phthaloyl-protected 3 with MeOH and thionyl chloride, respectively (Scheme 1). *tert*-Butyl ester 7 was obtained by the action of $Boc_2O/DMAP$ on *N*-Boc derivative 4 in *t*BuOH. When we used Boc_2O in *t*BuOH for esterification of compound 2, we obtained the *tert*-butyl ester of *N*-dipro-



Scheme 1. Synthesis of alkyl esters of *N*-protected 4-amino-5-oxo-1-phenylprolines.

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tected 4-amino-5-oxo-1-phenylproline (8) in a low yield due to the formation of the byproduct 4-tosyl-5-phenyl-2,5-diazabicyclo[2.2.1]heptan-3,6-dione (9) obtained by intermolecular cyclization as described previously.^[6] Thus, we first prepared *tert*-butyl ester 10 by esterification of *N*-tosyl derivative 2 with 2-methylpropene^[10] in dichloromethane in the presence of sulfuric acid. We then introduced *N*-Boc protection using Boc₂O/DMAP in *t*BuOH to obtain compound 8 in 81% overall yield. The structure of compound 8 was confirmed by X-ray crystallographic analysis (Figure 2).



Figure 2. X-ray crystal structure of compound 8.

Reduction of compounds 5–8 and 10 was carried out using borane complexes, such as $BH_3 \cdot Me_2S$ and $BH_3 \cdot THF$ in THF under an atmosphere of argon at various temperatures (Scheme 2). We used different amounts of reducing agent, from 1.5 to 4 equiv. relative to the substrate. The duration of the reduction was varied from 3 to 35 d. Each reaction was quenched by addition of MeOH. After evaporation to dryness under reduced pressure each residue was separated by flash column chromatography on silica gel, and the major reaction products were isolated. The structures of the isolated compounds were confirmed by ¹H and ¹³C NMR spectroscopy, elemental analysis, and LC– MS data.

The composition of the mixtures of the reaction products was quantitatively estimated based on isolated yields (Table 1). When we failed to separate the reaction products we applied ¹H NMR spectroscopy and HPLC data.

It was found that reaction of *N*-tosyl compounds **5** and **10** with 4 equiv. of BH_3 complex led to reduction of not only the lactam carbonyl groups affording esters **11** and **12** but also the ester carbonyl groups affording ethers **14** and **15**, prolinol **13**, and aldehyde **16** (Scheme 2). Decreasing the amount of reducing agent did not result in a considerable increase in the chemoselectivity, but diminished the conversion of the starting compounds.

In the case of the reduction of compound **5**, conducting the reaction at low temperature (-5 °C) made it possible to increase the yield of ester **11** to 44% (Table 1, Entry 2), whereas when the reaction was carried out at 40 °C the major product of reduction with BH₃·Me₂S was prolinol **13** (Table 1, Entry 9; 72% yield). The structure of compound **13** was established on the basis of X-ray crystallographic data (Figure 3).

In the case of *tert*-butyl ester **10**, together with the lactam carbonyl, the ester carbonyl group also underwent significant reduction. Thus, when the reduction was carried out at -5 °C for 12 d the major reduction products included not



Scheme 2. Reduction of compounds 5-8 and 10.

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Table 1. Reaction conditions and the results of reduction of compounds 5-8 and 10.

Entry		Reaction conditions				Reduction products						
	Substrate	Temp.	Borane	Time	Ester	Yield ^[a]	Prolinol	Yield ^[a]	Êther	Yield ^[a]	Other	Yield ^[a]
		[°C]	(equiv.)	[d]		[%]		[%]		[%]	products	[%]
1	5	-5	$BH_3 \cdot Me_2S(2)$	3	11	38	13	8			16	3
2	5	-5	$BH_3 \cdot Me_2S$ (4)	3	11	44	13	22			16	4
3	5	-5	$BH_3 \cdot Me_2S$ (4)	12	11	20	13	59			16	5
4	5	-5	$BH_3 \cdot Me_2S$ (4)	35	11	7	13	70	14	2	16	8
5	5	+25	$BH_3 \cdot Me_2S$ (2)	7	11	4	13	63				
6	5	+25	$BH_3 \cdot Me_2S$ (4)	7			13	72	14	2	16	3
7	5	+25	BH ₃ ·THF (1.5)	5	11	21						
8	5	+25	BH ₃ ·THF (4)	8			13	52	14	12		
9	5	+40	$BH_3 \cdot Me_2S$ (4)	7			13	72	14	4		
10	5	+40	BH_3 ·THF (4)	7			13	55	14	20		
11	10	-5	$BH_3 \cdot Me_2S$ (4)	3	12	43	13	9			16	11
12	10	-5	$BH_3 \cdot Me_2S$ (4)	7	12	48	13	12	15	2	16	14
13	10	-5	$BH_3 \cdot Me_2S$ (4)	12	12	42	13	14	15	3	16	20
14	10	-5	$BH_3 \cdot Me_2S$ (4)	35	12	21	13	23	15	8	16	25
15	10	+25	$BH_3 \cdot Me_2S$ (4)	7			13	37	15	38	16	12
16	6	-5	$BH_3 \cdot Me_2S$ (4)	35	17	24	18 ^[b]	11			19	22
17	6	+25	BH ₃ ·THF (1.5)	7	17	21	18 ^[b]	20				
18	6	+25	$BH_3 \cdot Me_2S$ (4)	7	17	26	18 ^[b]	10			19	28
											20	11
19	7	-5	$BH_3 \cdot Me_2S$ (4)	35	21	77						
20	7	+25	$BH_3 \cdot Me_2S$ (4)	7	21	43						
21	8	+25	$BH_3 \cdot Me_2S$ (4)	35					23	11	24 ^[b]	18
											10	2
22	8	+40	$BH_3 \cdot Me_2S$ (4)	22			22 ^[b]	12	23	48	13	14

[a] Isolated yield. [b] Compounds 18, 22, and 24 were not isolated in pure form; yields were determined on the basis of ¹H NMR spectroscopy.



Figure 3. X-ray crystal structure of compound 13.

only ester 12 (42% yield), but prolinol 13 (14% yield) and aldehyde 16 (20% yield) (Table 1, Entry 13) as well.

Treatment of *N*-phthaloyl derivative **6** with borane complexes resulted not only in the reduction of the lactam and ester carbonyl groups affording ester **17** and prolinol **18** but also in the partial reduction of the phthalimido group affording compounds **19** and **20**. The amount of **19** and **20** produced decreased when the reaction temperature was lowered, as did the conversion of starting compound **6** (Table 1, Entries 16–18). The yield of prolinol **18**, the product of reduction of the ester carbonyl in *N*-phthaloyl compound **6**, was lower than the yield of prolinol **13**, the product of *N*-tosyl-protected **5** (Table 1 Entries 3–8 and 16–18).

The only isolated reduction product for *N*-Boc derivative 7 with BH₃·Me₂S was ester 21, whose structure was confirmed by X-ray crystallographic data (Figure 4). When reductions was carried out at -5 °C for 35 d or at +25 °C for 7 d the yield of compound 21 was 77 or 43%, respectively (Table 1, Entries 19 and 20).



Figure 4. X-ray crystal structure of compound 21.

The reduction of sterically constrained compound 8 with both N-Boc and N-tosyl groups at the same nitrogen atom proceeded more slowly. When the reaction was carried out with BH₃·Me₂S at +25 °C for 35 d the conversion was 46% and the composition of reduction products was distinctive (Scheme 2). The product mixtures at both +40 and +25 °C contained alcohol 22, whose structure was assigned on the basis of ¹H NMR spectroscopy and LC-MS data, ether 23 with the lactam carbonyl reduced, and ether 24, in which the lactam carbonyl was untouched. The relative configuration of cis-23 was confirmed by X-ray crystallographic data (Figure 5) and a 2D NOESY spectrum, in which crosspeaks between 3-H^A and both 2-H and 4-H were observed. Compounds 10 and 13 without the N-Boc group were also detected among the reaction products. Thus, an increase in the steric hindrance caused by the presence of both N-tosyl



and *N*-Boc protecting groups at C-4 led to an increase in the number of the reduction products.



Figure 5. X-ray crystal structure of compound 23.

Thus, it has been found that the chemoselectivity of reduction of N-protected (2S,4S)-4-amino-5-oxo-1-phenylprolinates is determined to a large degree by the nature of the protecting groups, and to a smaller degree on the type of borane complex and the reagent ratios used (Table 1).

Conclusions

In sum, it has been shown that, depending on the reaction conditions and the type of protecting groups attached to the substrate, one can obtain both (2S,4S)-4-aminoproline and (2S,4S)-4-aminoprolinol derivatives as a result of the borane-mediated reduction of the carbonyl moieties in *N*-protected (2S,4S)-4-amino-5-oxo-1-phenylprolinates. In particular, we obtained certain (2S,4S)-4-amino-1-phenylproline and (2S,4S)-4-amino-1-phenylprolinol derivatives in approximately 70% yields.

Experimental Section

General: (2S,4S)-4-[(4-Methylbenzenesulfonyl)amino]-5-oxo-1phenylproline (2), (2S,4S)-5-oxo-1-phenyl-4-phthalimidoproline (3), and (2S,4S)-4-(tert-butyloxycarbonyl)amino-5-oxo-1-phenylproline (4) were prepared as described previously.^[6] 2-Methylpropene was obtained according to a literature procedure^[10] and was used as such in the subsequent step. All other reagents were of commercial quality. Solvents were dried and purified by standard methods. Routine monitoring of reaction mixtures was carried out by using Sorbfil UV 254 (Russia) TLC aluminum-plated silica gel. Silica gel 60 (230-400 mesh) was used for flash chromatography. Analytical HPLC was performed with a LiChrosorb Si-60 $(4 \times 250 \text{ mm}, 5 \mu\text{m})$ column, with a flow rate of 1 mL/min, and by using a tuneable UV detector set at 230 nm. Mixtures of hexane (solvent A) and *i*PrOH (solvent B) were used as the mobile phases. ¹H and ¹³C NMR spectra were recorded at 400 and 100 MHz, respectively, using TMS as a reference. The assignment of chemical ¹H and ¹³C shifts for compounds 11, 13, and 23 was based on standard 2D NMR techniques (COSY, 1H-1H NOESY, 1H-13C HSQC, and HMBC). Optical rotation values were measured with a Perkin-Elmer M 341 polarimeter. All optical rotations were obtained at room temperature. LC-MS data were collected by LCMS solution software with a LCMS-2010 instrument (Shimadzu) operating in positive and negative ion mode with ESI or APCI probe installed at N2 flow rate of 4.5 or 2.5 L/min, correspondingly. The probe voltage was set to 4.5 kV, and the APCI probe temperature was set to 400 °C. The curved desolvation line (CDL) and block heater were set at 250 and 200 °C, respectively. Quadrupole array (Q-array) and CDL were used in scan mode according to the autotuning parameters. LC separation was performed using ODS columns made by Phenomenex [Luna C18(2), 150×2.0 mm, 3 µm] or Supelco (Supelcosil LC-18 250 × 4.6 mm, 5 µm) at 60 °C. MeCN/ H₂O or MeOH/H₂O mixtures were used as solvents.

Methyl (2S,4S)-4-(4-Methylbenzenesulfonyl)amino-5-oxo-1-phenylprolinate (5): SOCl₂ (0.50 mL, 6.88 mmol) was added dropwise to a stirred solution of compound 2 (1.00 g, 2.67 mmol) in dry MeOH (20 mL) at -5 °C. The reaction mixture was kept for 0.5 h at that temperature and for 1 d at r.t. and then evaporated to dryness under reduced pressure. The residue was recrystallized from benzene and dried at 80 °C in vacuo to give compound 5 (0.920 g, 89%) as a white solid. M.p. 190–193 °C. $[a]_{D} = -2.7$ (c = 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ = 2.22 (ddd, J = 13.1, 8.8, 8.0 Hz, 1 H, 3-H^B), 2.42 (s, 3 H, Me-Ts), 2.98 (ddd, J = 13.1, 8.6, 7.2 Hz, 1 H, 3-H^A), 3.67 (s, 3 H, OMe), 3.94 (td, J = 8.6, 3.9 Hz, 1 H, 4-H), 4.71 (dd, J = 8.0, 7.2 Hz, 1 H, 2-H), 5.47 (d, J = 3.9 Hz, 1 H, NH), 7.20 (m, 1 H, Hp-Ph), 7.31-7.37 (m, 6 H, Ph, Ts), 7.81 (m, 2 H, Ts) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 21.53 (Me-Ts), 32.22 (C-3), 52.93, 53.74 (OMe, C-4), 58.58 (C-2), 121.71 (Co-Ph), 126.43 (Cp-Ph), 127.36 (Co-Ts), 129.11 and 129.86 (Cm-Ts, Cm-Ph), 135.89 and 137.00 (Ci-Ph, Ci-Ts), 144.02 (Cp-Ts), 169.78 and 170.84 (C-5, CO2Me) ppm. C19H20N2O5S (388.44): calcd. C 58.75, H 5.19, N 7.21, S 8.26; found C 58.79, H 5.14, N 7.10, S 8.18. HPLC (A/B, 40:1): $t_{\rm R} = 21.2$ min.

Methyl (2S,4S)-5-Oxo-1-phenyl-4-phthalimidoprolinate (6): SOCl₂ (0.25 mL, 3.4 mmol) was added dropwise to a stirred suspension of compound 3 (0.500 g, 1.43 mmol) in dry MeOH (10 mL) at 0 °C. The reaction mixture was kept for 0.5 h at that temperature and for 1 d at r.t. and then evaporated to dryness under reduced pressure. The residue was recrystallized from MeOH to give compound **6** (0.50 g, 96%) as white crystals. M.p. 175–177 °C. $[a]_{\rm D}$ = +30.7 (c = 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ = 2.65 (ddd, J = 12.5, 10.5, 8.9 Hz, 1 H, $3-H^B$), 2.93 (ddd, J = 12.5, 9.5, 7.6 Hz, 1 H, 3-H^A), 3.67 (s, 3 H, OMe), 4.92 (dd, J = 8.9, 7.6 Hz, 1 H, 2-H), 5.14 (dd, J = 10.5, 9.5 Hz, 1 H, 4-H), 7.22 (m, 1 H, Hp-Ph), 7.39 (m, 2 H, Hm-Ph), 7.48 (m, 2 H, Ho-Ph), 7.75 (m, 2 H, Phth), 7.88 (m, 2 H, Phth) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 27.54 (C-3), 49.49 (C-4), 52.73 (OMe), 57.94 (C-2), 122.24 (Co-Ph), 123.62 (Co-Phth), 126.19 (Cp-Ph), 128.96 (Cm-Ph), 131.82 (Cm-Phth), 134.32 (Ci-Phth), 137.61 (Ci-Ph), 167.20 (CO-Phth), 168.29 (C-5), 170.32 (CO2Me) ppm. C20H16N2O5 (364.36): calcd. C 65.93, H 4.43, N 7.69; found C 65.84, H 4.31, N 7.52. HPLC (A/B, 20:1): $t_{\rm R} = 26.4$ min.

tert-Butyl (2S,4S)-4-(tert-Butyloxycarbonyl)amino-5-oxo-1-phenylprolinate (7): Py (0.25 mL, 3.11 mmol), DMAP (0.038 g, 0.31 mmol) and Boc₂O (1.06 g, 4.83 mmol) were added to a stirred suspension of compound 4 (1.00 g, 3.12 mmol) in tBuOH (25 mL). The reaction mixture was stirred for 12 h at r.t., poured into cooled water (125 mL), and kept for 0.5 h at 10 °C. Filtration of the precipitate gave compound 7 (1.02 g, 87%) as colorless crystals. M.p. 171.5–172.5 °C (subl.) $[a]_D = -14.9$ (c = 1.0, CHCl₃). ¹H NMR (400 MHz, [D₆]DMSO): δ = 1.25 (s, 9 H, *t*Bu), 1.41 (s, 9 H, *t*Bu), 1.89 (ddd, J = 11.5, 11.0, 9.6 Hz, 1 H, 3-H^B), 2.68 (ddd, J = 11.5, 8.8, 7.0 Hz, 1 H, 3-H^A), 4.42 (ddd, J = 11.0, 8.9, 8.8 Hz, 1 H, 4-H), 4.81 (dd, J = 9.6, 7.0 Hz, 1 H, 2-H), 7.17 (m, 1 H, Hp-Ph), 7.34 (d, J = 8.9 Hz, 1 H, NH), 7.36–7.40 (m, 4 H, Ph) ppm. ¹³C NMR (100 MHz, $[D_6]DMSO$): $\delta = 27.25$ and 28.15 (both Me-*t*Bu), 29.43 (C-3), 51.20 (C-4), 57.31 (C-2), 78.16 and 81.42 (both C-tBu), 121.36 (Co-Ph), 124.95 (Cp-Ph), 128.44 (Cm-Ph), 138.47 (Ci-Ph), 155.29 (NCO₂-tBu), 169.65 and 171.46 (CO₂-tBu, C-5) ppm. C₂₀H₂₈N₂O₅ (376.45): calcd. C 63.81, H 7.50, N 7.44; found C 63.52, H 7.60, N 7.35. HPLC (A/B, 40:1): t_R = 14.0 min.

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tert-Butyl (2S,4S)-4-(4-Methylbenzenesulfonyl)amino-5-oxo-1-phenylprolinate (10): Concentrated H_2SO_4 (0.1 mL) was added to a stirred suspension of compound 2 (0.75 g, 2.0 mmol) in dry CH₂Cl₂ (6.5 mL). The reaction mixture was cooled to -18 °C. Freshly prepared 2-methylpropene was bubbled into the reaction mixture. The reaction bottle was tightly sealed, and the reaction mixture was stirred for 1 d until the solution clarified. The reaction mixture was then poured into a 1:1 mixture of EtOAc and 5% aqueous Na₂CO₃ (140 mL). The organic layer was separated, washed with aqueous Na₂CO₃ (5%) and brine to neutral pH, dried with Na₂SO₄, and then evaporated to dryness under reduced pressure. The residue was reprecipitated from EtOAc solution with hexane to give compound 10 (0.76 g, 88%) as colorless crystals. M.p. 179-183 °C (decomp.). [*a*]_D = +1.4 (*c* = 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃): $\delta = 1.28$ (s, 9 H, *t*Bu), 2.19 (ddd, J = 13.0, 9.0, 8.3 Hz, 1 H, 3-H^B), 2.41 (s, 3 H, Me-Ts), 2.97 (ddd, J = 13.0, 8.3, 7.0 Hz, 1 H, 3-H^A), 3.89 (ddd, J = 9.0, 8.3, 3.8 Hz, 1 H, 4-H), 4.57 (dd, J = 8.3, 7.0 Hz, 1 H, 2-H), 5.45 (d, J = 3.8 Hz, 1 H, NH), 7.19 (m, 1 H, Hp-Ph), 7.30-7.38 (m, 6 H, Ph, Ts), 7.82 (m, 2 H, Ts) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 21.52 (Me-Ts), 27.57 (Me-*t*Bu), 32.32 (C-3), 53.85 (C-4), 59.59 (C-2), 83.22 (C-tBu), 121.69 (Co-Ph), 126.19 (Cp-Ph), 127.38 (Co-Ts), 128.92 (Cm-Ph), 129.86 (Cm-Ts), 135.82 and 137.25 (Ci-Ts, Ci-Ph), 143.99 (Cp-Ts), 169.19 and 169.64 (C-5, CO₂-tBu) ppm. C₂₂H₂₆N₂O₅S (430.52): calcd. C 61.38, H 6.09, N 6.51, S 7.45; found C 60.97, H 5.97, N 6.19, S 7.51. HPLC (A/ B, 40:1): $t_{\rm R} = 8.9$ min.

tert-Butyl (2S,4S)-4-[(tert-Butyloxycarbonyl)(4-methylbenzenesulfonyl)amino]-5-oxo-1-phenylprolinate (8): Py (0.12 mL, 1.49 mmol), DMAP (10 mg, 0.08 mmol), and Boc₂O (300 mg, 1.38 mmol) were added to a stirred solution of compound 10 (352 mg, 0.82 mmol) in tBuOH (9 mL). The reaction mixture was stirred for 18 h at r.t., poured into water (40 mL), and kept for 2 h at 0 °C. Filtration of the precipitate gave compound 8 (398 mg, 92%) as colorless crystals. M.p. 166–172 °C (decomp.). $[a]_D = -45.4$ (c = 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃): *δ* = 1.29 (s, 9 H, *t*Bu), 1.37 (s, 9 H, *t*Bu), 2.42 (s, 3 H, Me-Ts), 2.63 (ddd, J = 11.8, 11.0, 9.6 Hz, 1 H, 3-H^B), 2.95 (ddd, J = 11.8, 9.0, 7.4 Hz, 1 H, 3-H^A), 4.70 (dd, J = 9.6, 7.4 Hz, 1 H, 2-H), 5.33 (dd, J = 11.0, 9.0 Hz, 1 H, 4-H), 7.19 (m, 1 H, Hp-Ph), 7.31 (m, 2 H, Ho-Ph), 7.35–7.43 (m, 4 H, Hm-Ph, Ts), 8.02 (d, J = 8.1 Hz, 2 H, Ts) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 21.63 (Me-Ts), 27.62 and 27.95 (both Me-*t*Bu), 28.95 (C-3), 57.28 and 58.05 (C-4, C-2), 82.59 and 85.45 (both C-tBu), 121.87 (Co-Ph), 125.76 (Cp-Ph), 128.16 (Co-Ts), 128.79 (Cm-Ph), 129.31 (Cm-Ts), 136.77 and 137.98 (Ci-Ts, Ci-Ph), 144.26 (Cp-Ts), 149.57 (NCO₂-tBu), 168.69 and 169.14 (CO₂-tBu, C-5) ppm. C27H34N2O7S (530.64): calcd. C 61.11, H 6.46, N 5.28, S 6.04; found C 61.17, H 6.63, N 4.74, S 6.21. HPLC (A/B, 80:1): $t_{\rm R}$ = 6.90 min.

General Procedure for Reduction of Compounds 5–8 and 10: BH₃ complex was added dropwise to a stirred solution of substrate 5–8 or 10 in THF at 0 °C under an atmosphere of argon. The reaction mixture was kept for 0.5 h at 0 °C under an argon atmosphere, then for 3–35 d at the appropriate temperature. MeOH was then added dropwise to the reaction mixture until foaming stopped. The reaction mixture was evaporated to dryness. The residue was subjected to column flash chromatography (benzene/EtOAc) to isolate the reaction products.

Methyl (2*S*,4*S*)-4-(4-Methylbenzenesulfonyl)amino-1-phenylprolinate (11): BH₃·Me₂S (0.2 mL, 2.09 mmol) was added to a solution of compound 5 (200 mg, 0.52 mmol) in dry THF (2 mL). The reaction mixture was kept for 3 d at -5 °C. Column flash chromatography gave compound 11 (85 mg, 44%) as a colorless amorphous solid. $[a]_{\rm D} = -65.6 \ (c = 2.0, \text{ CHCl}_3)$. ¹H NMR (400 MHz, CDCl₃): $\delta =$ 1.96 (ddd, J = 14.0, 2.8, 1.6 Hz, 1 H, 3-H^B), 2.38 (ddd, J = 14.0, 9.9, 6.3 Hz, 1 H, 3-H^A), 2.44 (s, 3 H, Me-Ts), 3.39 (ddd, J = 10.2, 2.2, 1.2 Hz, 1 H, 5-H^B), 3.43 (dd, J = 10.2, 5.0 Hz, 1 H, 5-H^A), 3.79 (s, 3 H, OMe), 4.14 (dd, J = 9.9, 1.6 Hz, 1 H, 2-H), 4.19 (m, 1 H, 4-H), 6.06 (d, J = 10.2 Hz, 1 H, NH), 6.43 (dd, ${}^{3}J = 8.8$, 1.0 Hz, 2 H, Ho-Ph), 6.77 (tt, ${}^{3}J$ = 7.3, 1.0 Hz, 1 H, Cp-Ph), 7.21 $(dd, {}^{3}J = 8.8, 7.3 Hz, 2 H, Hm-Ph), 7.31 (d, J = 8.3, Hz, 2 H, Hm-Ph)$ Ts), 7.77 (d, J = 8.1 Hz, 2 H, Ho-Ts) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 21.53$ (Me-Ts), 36.42 (C-3), 52.79 (C-4 and CO₂Me), 55.18 (C-5), 59.73 (C-2), 112.30 (Co-Ph), 117.95 (Cp-Ph), 126.95 (Co-Ts), 129.38 (Cm-Ph), 129.79 (Cm-Ts), 138.21 (Ci-Ts), 143.46 (Cp-Ts), 145.94 (Ci-Ph), 175.77 (CO2Me) ppm. C19H22N2O4S (374.46): calcd. C 60.94, H 5.92, N 7.48; found C 61.30, H 6.19, N 7.34. RP-HPLC-MS [Supelcosil LC-18; MeCN/H2O, 1:1; 1.0 mL/ min; $t_{\rm R} = 11.6$ min; APCI]: m/z (%) = 375 (100) [M + H]⁺; 373 (100) $[M - H]^{-}$. HPLC (A/B, 40:1): $t_{R} = 7.9$ min.

tert-Butyl (2S,4S)-4-(4-Methylbenzenesulfonyl)amino-1-phenylprolinate (12): BH₃·Me₂S (0.18 mL, 1.87 mmol) was added to a solution of compound 10 (200 mg, 0.46 mmol) in dry THF (2 mL). The reaction mixture was kept for 7 d at -5 °C. Column flash chromatography gave a 96:4 mixture (100 mg) of ester 12 and ether 15, respectively, according to ¹H NMR spectroscopy and HPLC data. Column flash chromatography (hexane/EtOAc) of that mixture gave compound 12 (80 mg, 43%) as a colorless amorphous solid. $[a]_{D} = -66.8$ (c = 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃): $\delta = 1.48$ (s, 9 H, tBu), 1.90 (dm, J = 14.0 Hz, 1 H, 3-H^B), 2.34 (ddd, J = 14.0, 9.9, 6.3 Hz, 1 H, 3-H^A), 2.44 (s, 3 H, Me-Ts), 3.37 and 3.40 (ABX system, J_{AB} = 10.1 Hz, J_{AX} = 4.4 Hz, J_{BX} = 2.6 Hz, 2 H, 5-H^B and 5-H^A), 3.98 (dd, J = 9.9, 1.6 Hz, 1 H, 2-H), 4.17 (m, 1 H, 4-H), 6.29 (d, J = 10.4 Hz, 1 H, NH), 6.45 (dd, J = 8.8, 1.0 Hz, 2 H, Ho-Ph), 6.75 (tt, ${}^{3}J$ = 7.3, 1.0 Hz, 1 H, Hp-Ph), 7.20 (dd, J = 8.8, 7.3 Hz, 2 H, Hm-Ph), 7.30 (d, J = 8.4 Hz, 2 H, Hm-Ts), 7.76 (d, J = 8.4 Hz, 2 H, Ho-Ts) ppm. ¹³C NMR (100 MHz, $CDCl_3$): $\delta = 21.51$ (Me), 27.88 (Me-*t*Bu), 36.39 (C-3), 52.81 (C-4), 55.29 (C-5), 60.99 (C-2), 82.96 (C-tBu), 112.29 (Co-Ph), 117.71 (Cp-Ph), 126.94 (Co-Ts), 129.27 and 129.77 (Cm-Ts, Cm-Ph), 138.32 (Ci-Ts), 143.33 (Cp-Ts), 146.09 (Ci-Ph), 174.66 (CO₂-tBu) ppm. C₂₂H₂₈N₂O₄S (416.54): calcd. C 63.44, H 6.78, N 6.72; found C 63.56, H 6.88, N 6.76. RP-HPLC-MS [Luna C18(2); MeCN/ H₂O, 75:25; 0.25 mL/min; $t_{\rm R}$ = 3.9 min; ESI]: m/z (%) = 439 (45) $[M + Na]^+$, 417 (100) $[M + H]^+$, 361 (89) $[M - tBu + 2H]^+$; 415 (100) $[M - H]^{-}$. HPLC (A/B, 120:1): $t_{R} = 8.1$ min.

(2S,4S)-2-Hydroxymethyl-4-(4-methylbenzenesulfonyl)amino-1-phenylpyrrolidine (13): BH₃·Me₂S (0.68 mL, 7.21 mmol) was added to a solution of compound 5 (0.70 g, 1.80 mmol) in dry THF (10 mL). The reaction mixture was kept for 7 d at 40 °C. Column flash chromatography gave compound 13 (0.45 g, 72%) as colorless crystals. M.p. 142–144 °C. $[a]_D = -73.8$ (c = 1.0, acetone). ¹H NMR (400 MHz, CDCl₃): δ = 1.72 (ddd, J = 13.9, 2.6, 1.3 Hz, 1 H, 3- H^{B}), 2.23 (br. m, 1 H, OH), 2.33 (ddd, J = 13.9, 9.9, 7.3 Hz, 1 H, $3-H^{A}$), 2.44 (s, 3 H, Me-Ts), 3.36 (dd, J = 10.3, 6.0 Hz, 1 H, $5-H^{B}$), 3.48 (dd, J = 10.3, 1.2 Hz, 1 H, 5-H^A), 3.50 (dm, J = 11.3 Hz, 1 H, 6-H^B), 3.85 (dm, J = 9.9, Hz, 1 H, 2-H), 4.03 (m, 1 H, 4-H), 4.12 (dt, J = 11.3, 2.4 Hz, 1 H, 6-H^A), 6.54 (d, J = 8.7 Hz, 2 H, Ho-Ph), 6.74 (t, J = 7.3 Hz, 1 H, Hp-Ph), 6.76 (d, J = 9.4 Hz, 1 H, NH), 7.21 (dd, ³*J* = 8.7, 7.3 Hz, 2 H, H*m*-Ph), 7.31 (d, *J* = 8.1 Hz, 2 H, H*m*-Ts), 7.78 (d, J = 8.1 Hz, 2 H, H*o*-Ts) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 21.52 (Me-Ts), 35.85 (C-3), 51.68 (C-4), 57.40 (C-5), 58.74 (C-2), 62.62 (C-6), 112.73 (Co-Ph), 117.34 (Cp-Ph), 126.96 (Co-Ts), 129.37 (Cm-Ph), 129.75 (Cm-Ts), 138.35 (Ci-Ts), 143.24 (Cp-Ts), 146.90 (Ci-Ph) ppm. C₁₈H₂₂N₂O₃S (346.45):



calcd. C 62.40, H 6.40, N 8.09, S 9.26; found C 62.42, H 6.47, N 7.94, S 9.52. HPLC (A/B, 40:1): $t_{\rm R}$ = 16.5 min.

(2S,4S)-2-Methoxymethyl-4-(4-methylbenzenesulfonyl)amino-1-phenylpyrrolidine (14): A 1 M solution (1.0 mL) of BH₃ in THF was added to a solution of compound 5 (100 mg, 0.26 mmol) in dry THF (1 mL). The reaction mixture was kept for 8 d at 25 °C. Column flash chromatography gave compound 14 (11 mg, 12%) as a pale-yellow amorphous solid. $[a]_D = -56.5$ (c = 1.0, acetone). ¹H NMR (400 MHz, CDCl₃): δ = 1.72 (dm, J = 13.9 Hz, 1 H, 3-H^B), 2.30 (ddd, J = 13.9, 10.1, 7.3 Hz, 1 H, 3-H^A), 2.44 (s, 3 H, Me-Ts), 3.30 (dd, J = 10.3, 6.1 Hz, 1 H, 5-H^B), 3.36 (dd, J = 9.8, 2.1 Hz, 1 H, 6-H^B), 3.39 (s, 3 H, OMe), 3.41 (dt, J = 10.3, 1.4 Hz, 1 H, 5-H^A), 3.80 (dd, J = 9.8, 2.4 Hz, 1 H, 6-H^A), 3.87 (dm, J = 10.1 Hz, 1 H, 2-H), 4.05 (dddt, J = 9.9, 7.3, 6.1, 1.4 Hz, 1 H, 4-H), 6.55 (d, J = 8.7 Hz, 2 H, Ho-Ph), 6.75 (t, J = 7.3 Hz, 1 H, Hp-Ph), 6.78 (d, J = 9.9 Hz, 1 H, NH), 7.22 (dd, J = 8.7, 7.3 Hz, 2 H, Hm-Ph), 7.32 (d, J = 8.3, Hz, 2 H, Hm-Ts), 7.77 (d, J = 8.3 Hz, 2 H, Ho-Ts) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 21.50 (Me-Ts), 35.41 (C-3), 51.67 (C-4), 56.94, 57.94 and 59.28 (OMe, C-5, C-2), 72.86 (C-6), 112.70 (Co-Ph), 117.20 (Cp-Ph), 126.93 (Co-Ts), 129.28 and 129.68 (Cm-Ph, Cm-Ts), 138.55 (Ci-Ts), 143.10 (Cp-Ts), 146.71 (Ci-Ph) ppm. MS [direct liquid introduction; MeOH, 0.25 mL/min. ESI]: m/z (%) = 383 (100) [M + Na]⁺, 329 (14) [M - OMe]⁻, 359 (100) $[M - H]^{-}$. $C_{19}H_{24}N_2O_3S$ (360.48). HPLC (A/B, 40:1): $t_R =$ 7.1 min.

(2S,4S)-2-tert-Butyloxymethyl-4-(4-methylbenzenesulfonyl)amino-1phenylpyrrolidine (15): BH₃·Me₂S (0.17 mL, 1.83 mmol) was added to a solution of compound 10 (197 mg, 0.46 mmol) in dry THF (2 mL). The reaction mixture was kept for 7 d at 25 °C. Column flash chromatography gave compound 15 (70 mg, 38%) as a colorless amorphous solid. $[a]_{578} = -46.5$ (c = 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ = 1.19 (s, 9 H, *t*Bu), 1.70 (dm, *J* = 14.0 Hz, 1 H, 3-H^B), 2.30 (ddd, J = 14.0, 10.1, 7.7 Hz, 1 H, 3-H^A), 2.44 (s, 3 H, Me-Ts), 3.26-3.30 (m, 2 H, $5-H^{B}$, $6-H^{B}$), 3.36 (dm, J =10.4 Hz, 1 H, 5-H^A), 3.86–3.93 (m, 2 H, 2-H, 6-H^A), 4.06 (m, 1 H, 4-H), 6.49 (dd, J = 8.8, 1.0 Hz, 2 H, Ho-Ph), 6.70 (tt, J = 7.3, 1.0 Hz, 1 H, Hp-Ph), 7.19 (dd, J = 8.8, 7.3 Hz, 2 H, Hm-Ph), 7.27 (d, J = 9.7 Hz, 1 H, NH), 7.31 (d, J = 8.4 Hz, 2 H, Hm-Ts), 7.77 (d, J = 8.4 Hz, 2 H, Ho-Ts) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 21.52 (Me-Ts), 27.40 (Me-tBu), 35.67 (C-3), 51.66 (C-4), 57.03 and 57.90 (C-5, C-2), 61.75 (C-6), 74.72 (C-tBu), 112.45 (Co-Ph), 116.66 (Cp-Ph), 126.90 (Co-Ts), 129.24 and 129.68 (Cm-Ph, Cm-Ts), 139.01 (Ci-Ts), 142.94 (Cp-Ts), 146.95 (Ci-Ph) ppm. C₂₂H₃₀N₂O₃S (402.56): calcd. C 65.64, H 7.51, N 6.96; found C 65.84, H 7.54, N 6.92. RP-HPLC-MS [Luna C18(2); MeCN/H₂O, 75:25; 0.25 mL/min; $t_{\rm R}$ = 4.6 min; ESI]: m/z (%) = 403 (100) [M + $H]^+$, 347 (16) $[M - tBu + 2H]^+$, 401 (100) $[M - H]^-$. HPLC (A/B, 120:1): $t_{\rm R} = 5.9$ min.

(2*S*,4*S*)-4-(4-Methylbenzenesulfonyl)amino-1-phenylpyrrolidine-2-carbaldehyde (16): BH₃·Me₂S (0.18 mL, 1.89 mmol) was added to a solution of compound **10** (200 mg, 0.47 mmol) in dry THF (2 mL). The reaction mixture was kept for 35 d at -5 °C. Column flash chromatography gave compound **16** (41 mg, 25%) as an amorphous solid. [a]_D = -122.7 (c = 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ = 2.11 (dt, J = 13.6, 4.1 Hz, 1 H, 3-H^B), 2.31 (ddd, J = 13.6, 9.4, 5.8 Hz, 1 H, 3-H^A), 2.44 (s, 3 H, Me), 3.47 (d, J = 4.5 Hz, 2 H, 5-H), 3.97 (m, 1 H, 4-H), 4.07 (ddd, J = 9.4, 4.1, 2.5 Hz, 1 H, 2-H), 5.08 (d, J = 6.4 Hz, 1 H, NH), 6.47 (d, J = 7.8 Hz, 2 H, Ho-Ph), 6.79 (t, J = 7.4 Hz, 1 H, Hp-Ph), 7.21 (dd, J = 8.7, 7.4 Hz, 2 H, Hm-Ph), 7.32 (d, J = 8.4 Hz, 2 H, Hm-Ts), 7.76 (d, J = 8.4 Hz, 2 H, Ho-Ts), 9.59 (d, J = 2.5 Hz, 1 H, CHO) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 21.51 (Me), 34.42 (C-3), 52.19 (C-4), 54.92

(C-5), 65.71 (C-2), 112.39 (Co-Ph), 118.17 (Cp-Ph), 127.10 (Co-Ts), 129.44 and 129.90 (Cm-Ph, Cm-Ts), 136.87 (Ci-Ts), 143.88 (Cp-Ts), 146.39 (Ci-Ph), 203.24 (CHO) ppm. $C_{18}H_{20}N_2O_3S$ (344.43): calcd. C 62.77, H 5.85, N 8.13; found C 62.78, H 6.21, N 7.75. HPLC (A/B, 40:1): $t_R = 8.2$ min.

Methyl (2*S*,4*S*)-1-Phenyl-4-phthalimidoprolinate (17), (2*S*,4*S*)-2-Hydroxymethyl-1-phenyl-4-phthalimidopyrrolidine (18), Methyl (2*S*,4*S*)-4-(2,3-Dihydro-3-oxo-1*H*-isoindol-2-yl)-1-phenylprolinate (19), and (2*S*,4*S*)-4-(2,3-Dihydro-3-oxo-1*H*-isoindol-2-yl)-2-hydroxymethyl-1-phenylpyrrolidine (20): BH₃·Me₂S (0.49 mL, 5.17 mmol) was added to a solution of compound 6 (0.47 g, 1.29 mmol) in dry THF (10 mL). The reaction mixture was kept for 7 d at 25 °C. Column flash chromatography gave ester 17 (118 mg, 26%), a 26:74 (according to ¹H NMR spectroscopy and HPLC and LCMS data, 163 mg) mixture of prolinol 18 and ester 19, and compound 20 (44 mg, 11%).

Methyl (2S,4S)-1-Phenyl-4-phthalimidoprolinate (17): Pale-yellow crystals. M.p. 122–124 °C. $[a]_D = -40.3$ (c = 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃): $\delta = 2.73$ (dt, $J = 12.6, 8.3, \text{Hz}, 1 \text{ H}, 3\text{-H}^{\text{B}}$), 2.99 (ddd, J = 12.6, 9.9, 7.5 Hz, 1 H, 3-H^A), 3.74 (s, 3 H, CO₂Me), 3.78 (dd, J = 8.6, 8.3 Hz, 1 H, 5-H^B), 4.02 (dd, J = 9.0, 8.6 Hz, 1 H, 5-H^A), 4.51 (dd, J = 8.2, 7.5 Hz, 1 H, 2-H), 4.94 (ddt, J = 9.9, 9.0, 8.3 Hz, 1 H, 4-H), 6.57 (dd, J = 8.8, 1.0 Hz, 2 H, Ho-Ph), 6.76 (tt, J = 7.3, 1.0 Hz, 1 H, Hp-Ph), 7.23 (dd, J = 8.8, 7.3 Hz, 2 H, H*m*-Ph), 7.74 (m, 2 H, Phth), 7.86 (m, 2 H, Phth) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 32.17 (C-3), 47.81 and 49.58 (C-4, C-5), 52.30 (CO₂Me), 59.51 (C-2), 112.33 (Co-Ph), 117.58 (Cp-Ph), 123.36 (Co-Phth), 129.21 (Cm-Ph), 131.74 (Cm-Phth), 134.21 (Ci-Phth), 146.25 (Ci-Ph), 167.96 (CO-Phth), 173.68 (CO₂Me) ppm. RP-HPLC-MS [Supelcosil LC-18; MeOH/H₂O, 6:4; 1.0 mL/min; $t_{\rm R} = 12.3 \text{ min; APCI}: m/z \ (\%) = 351 \ (100) \ [M + H]^+, 367 \ (100)$ $[M - H + H_2O]^-$. $C_{20}H_{18}N_2O_4$ (350.37): calcd. C 68.56, H 5.18, N 7.99; found C 68.34, H 5.06, N 7.79. HPLC (A/B, 20:1): $t_{\rm R}$ = 5.5 min.

(2*S*,4*S*)-2-Hydroxymethyl-1-phenyl-4-phthalimidopyrrolidime (18): ¹H NMR (400 MHz, CDCl₃): $\delta = 2.11$ (br. m, 1 H, OH), 2.50 (ddd, J = 12.7, 9.0, 7.4 Hz, 1 H, 3-H^B), 2.86 (ddd, J = 12.7, 10.0, 7.3 Hz, 1 H, 3-H^A), 3.67 (t, J = 9.0 Hz, 1 H, 5-H^B), 3.76–3.79 (m, 1 H, 6-H^B), 3.99–4.03 (m, 2 H, 6-H^A), 4.01 (t, J = 9.0 Hz, 1 H, 5-H^A), 4.16 (m, 1 H, 2-H), 4.92 (m, 1 H, 4-H), 6.71 (d, J = 8.7 Hz, 2 H, Ho-Ph), 6.76 (t, J = 7.4 Hz, 1 H, Hp-Ph), 7.24 (dd, J = 8.7, 7.4 Hz 2 H, Hm-Ph), 7.74 (m, 2 H, Phth), 7.86 (m, 2 H, Phth) ppm. RP-HPLC–MS [Supelcosil LC-18; MeOH/H₂O, 7:3; 1.0 mL/min; $t_R =$ 5.8 min; APCI]: m/z (%) = 323 (100) [M + H]⁺, 339 (100) [M – H + H₂O]⁻. HPLC (A/B, 20:1): $t_R = 13.0$ min.

(2S,4S)-4-(2,3-Dihydro-3-oxo-1H-isoindol-2-yl)-1-phenyl-Methyl prolinate (19): 3 M NaOH (0.05 mL) was added to a solution of a mixture of compounds 18 and 19 (163 mg) in acetone (1 mL). The reaction mixture was kept for 1 h at 5 °C and diluted with water. The precipitate was filtered off, washed with water, dried, and reprecipitated from EtOAc solution with hexane to give compound **19** (79 mg, 28%) as colorless needles. M.p. 160–162 °C. $[a]_{D} = -31.2$ $(c = 0.8, \text{CHCl}_3)$. ¹H NMR (400 MHz, CDCl₃): $\delta = 2.37$ (ddd, J =13.4, 6.6, 5.8 Hz, 1 H, 3-H^B), 2.85 (ddd, J = 13.4, 9.1, 7.8 Hz, 1 H, $3-H^{A}$), 3.70 (s, 3 H, CO₂Me), 3.74 (dd, J = 9.8, 6.5 Hz, 1 H, $5-H^{B}$), $3.77 (dd, J = 9.8, 7.0 Hz, 1 H, 5-H^A), 4.38 (dd, J = 9.1, 5.8 Hz, 1$ H, 2-H), 4.50 (d, J = 17.0 Hz, 1 H, CH^B-Ar), 4.58 (d, J = 17.0 Hz, 1 H, CH^A-Ar), 5.18 (dddd, J = 7.8, 7.0, 6.6, 6.4 Hz, 1 H, 4-H), 6.59 (dd, J = 8.8, 1.0 Hz, 2 H, Ho-Ph), 6.80 (tt, J = 7.4, 1.0 Hz, 1 H, Hp-Ph), 7.25 (dd, J = 8.8, 7.4 Hz, 2 H, Hm-Ph), 7.47 (m, 2 H, Ar), 7.55 (m, 1 H, Ar), 7.86 (m, 1 H, 4'-H) ppm. ¹³C NMR $(100 \text{ MHz}, \text{ CDCl}_3)$: $\delta = 34.85 \text{ (C-3)}, 46.68 \text{ and } 49.62 \text{ (C-4, C-5)},$

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51.31 and 52.31 (CO₂Me, CH₂-Ar), 59.83 (C-2), 112.76 (Co-Ph), 118.07 (Cp-Ph), 122.84, 123.61 and 128.07 (Ar), 129.31 (Cm-Ph), 131.55, 132.35 and 141.29 (Ar), 146.40 (C*i*-Ph), 168.91 (CO-Ar), 174.22 (CO₂Me) ppm. RP-HPLC–MS [Supelcosil LC–18, MeCN/ H₂O, 97:3, 1.2 mL/min; $t_{\rm R}$ = 2.5 min; APCI]: m/z (%) = 337 (100) [M + H]⁺, 303 (100) [M - H - MeOH]⁻, 335 (7) [M - H]⁻. C₂₀H₂₀N₂O₃ (336.39): calcd. C 71.41, H 5.99, N 8.33; found C 71.42, H 5.84, N 8.52. HPLC (A/B, 20:1): $t_{\rm R}$ = 15.2 min.

(2*S*,4*S*)-4-(2,3-Dihydro-3-oxo-1*H*-isoindol-2-yl)-2-hydroxymethyl-1-phenylpyrrolidine (20): Pale-yellow amorphous solid. ¹H NMR (400 MHz, CDCl₃): δ = 1.89 (br. s, 1 H, OH), 2.50 (dd, *J* = 8.9, 7.4 Hz, 2 H, 3-H), 3.59 (dd, *J* = 9.5, 8.6 Hz, 1 H, 5-H^B), 3.67 (dm, *J* = 11.3, Hz, 1 H, 6-H^B), 3.76 (dd, *J* = 9.5, 7.9 Hz, 1 H, 5-H^A), 4.02 (dd, *J* = 11.3, 4.4 Hz, 1 H, 6-H^A), 4.09 (tdd, *J* = 7.4, 4.4, 2.2 Hz, 1 H, 2-H), 4.50 (d, *J* = 16.8 Hz, 1 H, CH^B-Ar), 4.56 (d, *J* = 16.8 Hz, 1 H, CH^A-Ar), 4.97 (tdd, *J* = 8.9, 8.6, 7.9 Hz, 1 H, 4-H), 6.71 (d, *J* = 8.7 Hz, 2 H, Ho-Ph), 6.77 (t, *J* = 7.4 Hz, 1 H, Hp-Ph), 7.25 (dd, *J* = 8.7, 7.4 Hz, 2 H, Hm-Ph), 7.45–7.50 (m, 2 H, Ar), 7.56 (m, 1 H, Ar), 7.86 (m, 1 H, Ar) ppm. RP-HPLC–MS [Supelcosil LC-18; MeCN/H₂O, 8:2; 1.0 mL/min; *t*_R = 3.1 min; APCI]: *m/z* (%) = 309 (100) [M + H]⁺. HPLC (A/B, 20:3): *t*_R = 11.3 min.

tert-Butyl (2S,4S)-4-(tert-Butyloxycarbonyl)amino-1-phenylprolinate (21): BH₃·Me₂S (0.5 mL, 5.31 mmol) was added to a solution of compound 7 (0.5 g, 1.33 mmol) in dry THF (5 mL). The reaction mixture was kept for 35 d at -5 °C. Column flash chromatography gave compound 21 (370 mg, 77%) as colorless crystals. M.p. 112-114 °C. $[a]_{578} = -84.7$ (c = 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ = 1.44 (s, 9 H, *t*Bu), 1.49 (s, 9 H, *t*Bu), 2.08 (dq, *J* = 13.8, 1.5 Hz, 1 H, 3-H^B), 2.51 (ddd, J = 13.8, 10.0, 6.3 Hz, 1 H, 3- H^{A}), 3.47 (dt, J = 9.8, 1.4 Hz, 1 H, 5- H^{B}), 3.57 (dd, J = 9.8, 5.5 Hz, 1 H, 5-H^A), 4.08 (dd, J = 10.0, 1.5 Hz, 1 H, 2-H), 4.46 (m, 1 H, 4-H), 5.77 (d, J = 9.4 Hz, 1 H, NH), 6.54 (dd, J = 8.7, 1.0 Hz, 2 H, Ho-Ph), 6.76 (tt, J = 7.3, 1.0 Hz, 1 H, Hp-Ph), 7.23 (dd, J = 8.7, 7.3 Hz, 2 H, H*m*-Ph) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 27.86 and 28.40 (both Me-tBu), 37.20 (C-3), 49.77 (C-4), 55.61 (C-5), 61.28 (C-2), 79.31 and 82.26 (both C-tBu), 112.30 (Co-Ph), 117.38 (Cp-Ph), 129.24 (Cm-Ph), 146.52 (Ci-Ph), 155.39 (NCO₂-tBu), 174.25 (CO₂-tBu) ppm. C₂₀H₃₀N₂O₄ (362.47): calcd. C 66.27, H 8.34, N 7.73; found C 66.20, H 8.57, N 7.45. HPLC (A/B, 40:1): $t_{\rm R} = 3.5 \, {\rm min.}$

(2*S*,4*S*)-4-[(*tert*-Butyloxycarbonyl)(4-methylbenzenesulfonyl)amino]-2-hydroxymethyl-1-phenylpyrrolidine (22) and (2*S*,4*S*)-4-[(*tert*-Butyloxycarbonyl)(4-methylbenzenesulfonyl)amino]-2-*tert*-butyloxymethyl-1-phenylpyrrolidine (23): BH₃·Me₂S (0.13 mL, 1.41 mmol) was added to a solution of compound **8** (188 mg, 0.35 mmol) in dry THF (2 mL). The reaction mixture was kept for 24 d at 40 °C. After evaporation to dryness under reduced pressure the residue was subjected to column flash chromatography (hexane/*i*BuOH) to give compound **23** (85 mg, 48%), a 4:1 (according to ¹H NMR spectroscopy) mixture of compounds **22** and **13** (23 mg), and compound **13** (10 mg, 14%).

(2*S*,4*S*)-4-[(*tert*-Butyloxycarbonyl)(4-methylbenzenesulfonyl)amino]-2-hydroxymethyl-1-phenylpyrrolidine (22): RP-HPLC–MS [Luna C18(2); MeCN/H₂O, 7:3; 0.3 mL/min; $t_{\rm R}$ = 3.5 min; ESI]: m/z (%) = 469 (3.6) [M + Na]⁺, 447 (100) [M + H]⁺, 391 (4.9) [M - *t*Bu + 2H]⁺, 347 (39) [M - Boc + 2H]⁺. HPLC (A/B, 40:1): $t_{\rm R}$ = 10.4 min.

(2*S*,4*S*)-4-[(*tert*-Butyloxycarbonyl)(4-methylbenzenesulfonyl)amino]-2-*tert*-butyloxymethyl-1-phenylpyrrolidine (23): ¹H NMR (400 MHz, CDCl₃): δ = 1.16 (s, 9 H, *t*Bu), 1.37 (s, 9 H, Me-Boc), 2.45 (s, 3 H, Me-Ts), 2.51 (ddd, *J* = 12.3, 10.4, 6.9 Hz, 1 H, 3-H^B), 2.58 (ddd, *J* = 12.3, 8.9, 7.8 Hz, 1 H, 3-H^A), 3.23 (dd, *J* = 8.9, 8.7 Hz, 1 H, 6-H^B), 3.62 (t, J = 8.6 Hz, 1 H, 5-H^B), 3.78 (dd, J =8.6, 3.6 Hz, 1 H, 6-H^A), 3.88 (t, J = 8.9 Hz, 1 H, 5-H^A), 4.01 (dddd, J = 8.9, 7.8, 6.9, 3.6 Hz, 1 H, 2-H), 5.08 (dq, J = 10.4, 8.9 Hz, 1 H, 4-H), 6.67 (d, J = 8.6 Hz, 2 H, Ho-Ph), 6.71 (t, J = 7.3 Hz, 1 H, Hp-Ph), 7.23 (dd, J = 8.6, 7.3 Hz, 2 H, Hm-Ph), 7.33 (d, J =8.1 Hz, 2 H, H*m*-Ts), 7.76 (d, J = 8.1 Hz, 2 H, H*o*-Ts) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 21.58$ (Me-Ts), 27.54 (Me-*t*Bu), 27.84 (Me-Boc), 33.44 (C-3), 50.68 (C-5), 54.67 (C-4), 56.70 (C-2), 63.08 (C-6), 73.02 (C-tBu), 84.69 (C-Boc), 113.05 (Co-Ph), 116.60 (Cp-Ph), 127.45 (Co-Ts), 129.20 (Cm-Ph), 129.37 (Cm-Ts), 137.65 (Ci-Ts), 144.10 (Cp-Ts), 146.99 (Ci-Ph), 150.14 (CO-Boc) ppm. RP-HPLC-MS [Luna C18(2); MeOH/H₂O, 8:2; 0.25 mL/min; $t_{\rm R}$ = 13.6 min; ESI]: m/z (%) = 525 (5) [M + Na]⁺, 503 (100) [M + H]⁺, 447 (5) $[M - Boc + 2Na]^+$, 425 (1) $[M - Boc + H + Na]^+$, 403 (10) $[M - Boc + 2H]^+$, 347 (18) $[M - Ts]^+$. HPLC (A/B, 80:1): $t_R =$ 3.3 min.

(2S,4S)-4-[(tert-Butyloxycarbonyl)(4-methylbenzenesulfonyl)amino]-2-tert-butyloxymethyl-5-oxo-1-phenylpyrrolidine (24): BH₃·Me₂S (0.12 mL, 1.31 mmol) was added to a solution of compound 8 (173 mg, 0.33 mmol) in dry THF (2 mL). The reaction mixture was kept for 35 d at 25 °C. After evaporation to dryness under reduced pressure the residue was subjected to column flash chromatography (hexane/iBuOH) to give 0.65:1 mixture (49 mg) of compounds 8 and 24. ¹H NMR (400 MHz, CDCl₃): $\delta = 1.03$ (s, 9 H, *t*Bu), 1.33 (s, 9 H, Me-Boc), 2.38 (m, 1 H, 3-H^B), 2.41 (s, 3 H, Me-Ts), 2.80 $(ddd, J = 12.3, 9.8, 7.2 Hz, 1 H, 3-H^{A}), 3.34 (dd, J = 8.9, 7.2 Hz, 1 H, 3-H^{A})$ 1 H, 6-H^B), 3.55 (dd, J = 8.9, 4.3 Hz, 1 H, 6-H^A), 4.29 (m, 1 H, 2-H), 5.33 (dd, J = 10.2, 9.8 Hz, 1 H, 4-H), 7.22–7.39 (m, 7 H, Ph, Ts), 8.06 (d, J = 7.7 Hz, 2 H, Ts) ppm. RP-HPLC-MS [Luna C18(2), MeOH/H₂O, 8:2, 0.25 mL/min; $t_{\rm R}$ = 4.6 min; APCI]: *m*/*z* $(\%) = 517 (6) [M + H]^+, 417 (69) [M - Boc]^+, 361 (100) [M - M]^+$ Ts]⁺. HPLC (A/B, 80:1): $t_{\rm R} = 6.4$ min.

X-ray Analysis: Data for compounds 8, 13, 21, and 23 were collected with an XCALIBUR-3 diffractometer (CCD) by using graphite-monochromated $Mo-K_a$ radiation. Absorption and anomalous dispersion effects were not taken into consideration due to its insignificance. The structures were solved by direct methods and expanded by using Fourier techniques. Refinement of the structure was accomplished with the SHELXL-97 program. Non-hydrogen atoms were refined anisotropically. Hydrogen atoms involved in hydrogen bonding were located in electron density maps. The remainder of the hydrogen atoms were placed in idealized positions and allowed to ride on the C atoms to which they are bonded. CCDC-803760 (for 8), -803761 (for 13), -803762 (for 21), and -803763 (for 23) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/ data_request/cif.

Crystal Data for 8: $M_r = 530.62$; $0.34 \times 0.21 \times 0.06$ mm; colorless plate; T = 295 K; orthorhombic; space group $P2_12_12_1$; Z = 4; a = 10.5475(7) Å, b = 12.1445(9) Å, c = 22.4485(13) Å, $a = \beta = \gamma = 90.00^\circ$, V = 2875.5(3) Å³; $\rho_{calcd.} = 1.226$ g cm⁻³; $\Theta_{max} = 26.38^\circ$; $R_1 = 3.79\%$ [$I > \sigma(I)$], $wR_2 = 4.54\%$, S = 0.984.

Crystal Data for 13: $M_r = 346.44; 0.38 \times 0.18 \times 0.07$ mm; colorless needle; T = 295 K; monoclinic, space group $P2_1$; Z = 2; a = 9.718(3) Å, b = 7.974(3) Å, c = 11.531(3) Å, $a = \gamma = 90.0^{\circ}$ $\beta = 104.19(2)^{\circ}$, V = 866.3(4) Å³; $\rho_{calcd.} = 1.328$ gcm⁻³; $\Theta_{max} = 26.39^{\circ}$; $R_1 = 4.18\%$ [$I > \sigma(I)$], $wR_2 = 2.61\%$, S = 0.813.

Crystal Data for 21: $M_r = 362.46$; $0.43 \times 0.27 \times 0.14$ mm; colorless prism; T = 295 K; orthorhombic, space group $P2_12_12_1$; Z = 4; a = 6.3189(4) Å, b = 9.0566(5) Å, c = 35.412(2) Å, $a = \beta = \gamma = 90.0^\circ$,



 $V = 2026.5(2) \text{ Å}^3$; $\rho_{\text{calcd.}} = 1.188 \text{ g cm}^{-3}$; $\Theta_{\text{max}} = 26.36^\circ$; $R_1 = 3.05\%$ $[I > \sigma(I)]$, $wR_2 = 2.90\%$, S = 1.006.

Crystal Data for 23: $M_r = 502.65$; $0.34 \times 0.18 \times 0.09$ mm; colorless prism; T = 120 K; monoclinic, space group $P2_1$; Z = 4; a = 10.5799(8) Å, b = 10.0648(9) Å, c = 25.178(2) Å, $a = \gamma = 90.0^{\circ}$, $\beta = 90.146(6)^{\circ}$, V = 2681.1(4) Å³; $\rho_{calcd.} = 1.245$ g cm⁻³; $\Theta_{max} = 26.37^{\circ}$; $R_1 = 3.95\%$ [$I > \sigma(I)$], $wR_2 = 3.43\%$, S = 0.849.

Acknowledgments

The work was financially supported by the Ural Division of the Russian Academy of Sciences (Ural Div. of RAS), grants 09-P-3-2001 and 09-I-3-2001, the State Program for Supporting of Leading Scientific Schools of the Russian Federation (Grant NSh 65261.2010.3), and the Russian Foundation for Basic Research (Grant 11-03-00548). The authors thank the Ural Div. of RAS for financial support for A.Yu.V. The authors thank Dr. P. Slepukhin for performing X-ray crystallography analyses.

- a) M. I. Calaza, C. Cativiela, *Eur. J. Org. Chem.* 2008, 3427– 3448; b) H. Sakashita, F. Akahoshi, H. Kitajima, R. Tsutsumiuchi, Y. Hayashi, *Bioorg. Med. Chem.* 2006, 14, 3662–3671; c) C. E. Schafmeister, Z. Z. Brown, S. Gupta, *Acc. Chem. Res.* 2008, 41, 1387–1398.
- [2] a) D. Torino, A. Mollica, F. Pinnen, F. Feliciani, S. Spisani, G. Lucente, *Bioorg. Med. Chem.* 2009, *17*, 251–259; b) S. Flemer, A. Wurthmann, A. Mamai, J. S. Madalengoitia, *J. Org. Chem.* 2008, *73*, 7593–7602; c) A. Flores-Ortega, J. Casanovas, X. Assfeld, C. Alemán, *J. Org. Chem.* 2009, *74*, 3101–3108.

- [3] a) M. G. Moloney, Nat. Prod. Rep. 2002, 19, 597–616; b) D. D. Schoepp, B. G. Johnson, C. R. Salhoff, M. J. Valli, M. A. Desai, J. P. Burnett, N. G. Mayne, J. A. Monn, Neuropharmacology 1995, 34, 843–850; c) V. Bruno, G. Battaglia, A. Copani, M. D'Onofrio, P. Di Iorio, A. De Blasi, D. Melchiorri, P. J. Flor, F. Nicoletti, J. Cerebral Blood Flow Metabol. 2001, 21, 1013–1033.
- [4] a) T. E. Kristensen, T. Hansen, *Eur. J. Org. Chem.* 2010, 3179–3204; b) S. B. Tsogoeva, S. B. Jagtap, Z. A. Ardemasova, *Tetrahedron: Asymmetry* 2006, *17*, 989–992; c) S. G. Zlotin, A. S. Kucherenko, I. P. Beletskaya, *Russ. Chem. Rev.* 2009, *78*, 737–784.
- [5] a) C. Nájera, M. Yus, *Tetrahedron: Asymmetry* 1999, 10, 2245–2303; b) S. K. Panday, J. Prasad, D. K. Dikshit, *Tetrahedron: Asymmetry* 2009, 20, 1581–1632; c) F. Lenda, F. Guenoun, J. Martinez, F. Lamaty, *Tetrahedron Lett.* 2007, 48, 805–808.
- [6] V. P. Krasnov, I. A. Nizova, A. Yu. Vigorov, T. V. Matveeva, G. L. Levit, P. A. Slepukhin, M. A. Ezhikova, M. I. Kodess, *Eur. J. Org. Chem.* 2008, 1802–1810.
- [7] B. Jiang, Z.-G. Huang, K.-J. Cheng, *Tetrahedron: Asymmetry* **2006**, *17*, 942–951 and references cited therein.
- [8] a) T. Hosokami, M. Kuretani, K. Higashi, M. Asano, K. Ohya, N. Takasugi, E. Mafune, T. Miki, *Chem. Pharm. Bull.* 1992, 40, 2712–2719; b) T. M. Kamenecka, T. Lanza Jr., S. E. de Laszlo, B. Li, E. D. McCauley, G. Van Riper, L. A. Egger, U. Kidambi, R. A. Mumford, S. Tong, M. MacCoss, J. A. Schmidt, W. K. Hagmann, *Bioorg. Med. Chem. Lett.* 2002, 12, 2205–2208.
- [9] A. F. C. Alcântara, H. S. Barroso, D. Piló-Veloso, *Quim. Nova* 2002, 25, 300–311.
- [10] P. Sieber, B. Iselin, Helv. Chim. Acta 1969, 52, 1525-1531.

Received: January 13, 2011 Published Online: March 22, 2011