



Synthesis of Bicyclic *Cis* Fused Dihydrofuran Derivatives via the Intramolecular Mitsunobu Reaction

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Abstract: Stereoselective dehydrative alkylation/annulation of hydroxy β -diketones, β -ketoesters or β -diesters using Mitsunobu conditions ($\text{Bu}_3\text{P}/\text{DEAD}$) proceeds under mild and essentially neutral conditions affording *cis* fused bicyclic dihydrofurans in good yields.

Furans and partially reduced furans are important structural units in naturally occurring compounds.¹ In connexion with the synthesis of polyoxacyclic compounds, we needed an easy entry into the heteroring system of pyranoids derived from sugars ("off-template"). Although a wide range of synthetic routes to 2,3-dihydrofurans is known,² the most straightforward synthesis of such derivatives involves the manganese(III) acetate-promoted oxidative cyclization of β -dicarbonyl compounds with electron rich olefins.³ Since this process involves monoelectron transfer and oxidative steps which may not be suitable for elaborated and sensitive substrates, we looked for a milder route under neutral conditions. We report here such a method for the construction of *cis* fused bicyclic dihydrofuran derivatives using α -(1,3-dicarbonyl)-substituted allylic alcohols.

First we examined a series of *cis* 4-alkyl cyclohexen-2-ols **1-5**, readily available via the well-known palladium-catalyzed alkylation of cyclic vinyl epoxides with soft carbon nucleophiles under neutral conditions.⁴ Since these compounds contain a nucleophilic center and an electrophilic counterpart, a dehydrative alkylation/annulation process was anticipated under conditions enhancing the displacement of an activated allylic leaving group. Therefore, the reaction was carried out following Mitsunobu's protocol⁵ which provides activation under mild and essentially neutral conditions. The best results were obtained when DEAD and triphenylphosphine (2.2 eq. each) were mixed first⁶ before the dropwise addition of the substrate (Table I).

Thus, *cis* 4-alkyl cyclohexen-2-ols **1** and **2** smoothly cyclize (C_6H_6 , rt), respectively, into bicyclic tetrahydrobenzofuran derivatives⁷ **6** (64%) and **7** (49%). In sharp contrast, **4** was recovered unchanged under the same conditions⁸ (Table I, entry 3). As expected, when switching to tributylphosphine as the phosphane component,⁹ **10** was obtained in a satisfactory 64% yield from **4**. Using the latter conditions, which were definitely more adapted for our purposes, dehydrative cyclizations of **1** and **2** are dramatically improved (Table I, entry 1-2). The regioselectivity of cyclization was briefly examined with the unsymmetrical β -diketone **3** (Table I, entry 4) which afforded an unseparable 1:1 mixture of bicyclic isomers **8** and **9** in 83% yield.

Table I. Cyclisation of *cis*-4-Alkyl cyclohexen-2-ols under Mitsunobu conditions ^a

Entry	<i>cis</i> -4-Alkyl cyclohexen-2-ol	Product(s)	Yield (%) ^b	
			c	d
1	1 R ¹ = Me, R ² = COMe	6	64	83
2	2 R ¹ = Me, R ² = CO ₂ Me	7	49	74
3	3 R ¹ = Me, R ² = C(=O)Ph	$\begin{cases} 8 \text{ R}^1 = \text{Me, R}^2 = \text{C(=O)Ph} \\ 9 \text{ R}^1 = \text{Ph, R}^2 = \text{COMe} \end{cases}$	-	83 (1:1)
4	4 R ¹ = OMe, R ² = CO ₂ Me	10	0	64
5	5 R ¹ = OEt, R ² = NO ₂		-	82

a) DEAD : PR₃ (2.2 molar eq. each), THF, rt, 2-6h; b) isolated yield; c) PPh₃; d) PBu₃

In all cases the expected *O*-alkylation leads to the exclusive formation of a single *cis* fused bicyclic 4,5-dihydrofuran isomer,¹⁰ except when the nucleophilic counterpart is an α -nitroester group, such as in **5**, which affords a *cis* fused bicyclic nitronate¹¹ **11** in 82% yield. No trace of cyclopropanation resulting from *C*-alkylation was detected. This result is predicted on the basis of stereoelectronic arguments.¹² The *cis* fused ring junction was deduced from NMR NOESY experiments and by comparison with literature data.¹³ The resonances of H-3a of compounds **6** and **10** are observed as a dd at δ 2.96 (*J* 4.1, 8.3, 12.1 Hz) and 3.05 (*J* 4.4, 8.1, 11.4 Hz) respectively. In addition, the allylic bridgehead H-7a in **7** and **10** appears as a doublet of doublet at δ 4.70 (*J* 1.6, 8.6 Hz) and as a broad doublet at 4.70 (*J* 8.0 Hz), respectively.

A more general applicability of this method was then tested with β -C-glycosides **12-15**¹⁴ which afforded *cis*-fused chiral dihydrofurans¹⁰ branched on the 4-hexenopyranoside moiety¹⁵ (Table II).

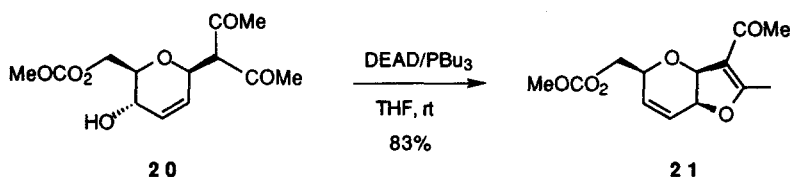
The dehydrative alkylation/annulation process presumably involves a 5-enol *endo-exo*-trig ring closure¹⁶ and delivers the dihydrofuran derivatives rather than an unfavourable 3-enol *exo-exo*-trig pathway giving rise to the cyclopropanated isomer. We assume that this easy heterocyclization is in agreement with a *syn* process which generally prevails in the SN2' reactions.^{12, 17}

Table II. Cyclisation of β -C-glycosides under Mitsunobu conditions ^a

Entry	C-glycoside	Product	Yield (%) ^b
1	12 R ¹ = CH ₂ OTBDMS, R ² = COMe	16	71
2	13 R ¹ = Me, R ² = COMe	17	80
3	14 R ¹ = H, R ² = COMe	18	60
4	15 R ¹ = H, R ² = CO ₂ Me	19	72

a) DEAD : PBu₃ (2.2 molar eq. each), THF, rt, 2-6h; b) isolated yield

This question was examined with the 1,4-*trans*-disubstituted hex-2-enopyranoside derivative **20**,¹⁴ which on treatment under the same conditions cyclized to **21** as a single stereomer in 83% yield.



In this specific case, an *anti* S_N2' process¹⁷ is likely to proceed rather than an unfavourable S_N2 displacement on a distal carbinol center. On the other hand, the *anti* stereochemistry of the nucleophilic attack could be well rationalized *via* a S_N1-type mechanism. The occurrence of a contact ion pair or an allylic cation that is not entirely free originating from the solvolysis of the intermediate alkoxyphosphonium ion has been discussed, respectively, in intermolecular Mitsunobu amination¹⁸ and esterification¹⁹ of allylic alcohols. A more detailed investigation is now in progress to clarify this point.

References and notes

1. a) Westley, J. W. *Polyether Antibiotics: Naturally Occurring Acid Ionophores*; Marcel Dekker: New York, 1982, vols 1 and 2; b) Gokel, G. W.; Korzeniowski, S. H. *Macrocyclic Polyether Syntheses*; Springer Verlag: Berlin, 1982.
2. a) Faller, J. W.; Murray, H. H.; White, D. L.; Chao, K. H. *Organometallics* **1983**, 2, 400-409; b) Pearson, A. J.; Khan, M. N. I.; Clardy, J. C.; Cun-heng, H. *J. Am. Chem. Soc.* **1985**, 107, 2748-2757; c) L. M. Barinelli, K. M. Nicholas *J. Org. Chem.* **1988**, 53, 2114-2117; d) McDonald, F. E.; Connolly, C. B.; Gleason, M. M.; Towne, T. B.; Treibek, K. D. *J. Org. Chem.* **1993**, 58, 6952-6953.
3. a) Heiba, E. I.; Desau R. M. *J. Org. Chem.* **1974**, 39, 3456-3457; b) Iqbal, J.; Bathia, B.; Nayyar, N. K. *Chem. Rev.* **1994**, 94, 519-564.
4. a) Trost, B. M.; Molander, G. A. *J. Am. Chem. Soc.* **1981**, 103, 5969-5972; b) Tsuji, J.; Kataoka, H.; Kobayashi, Y. *Tetrahedron Lett.* **1981**, 22, 2575-2578.
5. a) Mitsunobu, O. *Synthesis* **1981**, 1-28; b) Hughes, D. L. *Org. React.* **1992**, 42, 335-656.

6. As emphasized earlier, the order mixing of the reagents is crucial for the efficiency . See ref. 5.
7. Intermolecular version of this reaction using β -diketones or β -ketoesters with alcohols was first reported by Mitsunobu and proceeds mainly or exclusively with *O*-alkylation. See: a) Wada, M.; Mitsunobu, O. *Tetrahedron Lett.* **1972**, 22, 1279-1282; b) Kurihara, T.; Sugizaki, M.; Kime, I.; Wada, M.; Mitsunobu, O. *Bull. Chem. Soc. Jpn.* **1981**, 54, 2107-2112.
8. O. Mitsunobu has pointed out that ethyl malonate (pK 13.3) was unreactive in the reaction with alcohols, presumably due to low acidity. See ref. 5a.
9. a) For a discussion on the role of the phosphane in the course of the Mitsunobu reaction, see: Tsunoda, T.; Yamamiya, Y.; Itô, S. *Tetrahedron Lett.*, **1993**, 34, 7639-7642; b) Tributylphosphine generates an alkoxide as intermediate which might be beneficial for the annulation. See: Camp, D.; Jenkins, I. *Aust. J. Chem.* **1992**, 45, 47-55.
10. All new compounds gave IR, NMR, and elemental analyses consistent with the assigned structures. Selected spectral data. **10**: IR (neat) ν 2956, 1708, 1622, 1470, 1387, 1329, 1271, 1165, 1087 cm^{-1} ; RMN ^1H (CDCl_3) δ 1.20 (m, 1H), 1.83 (m, 1H), 2.09 (m, 2H), 3.05 (ddd, $J = 4.4, 8.1, 11.4$ Hz, 1H, H-3a), 3.67 (s, 3H), 3.89 (s, 3H), 4.80 (br d, $J = 8.0$ Hz, 1H, H-7a), 5.87 (m, 1H), 6.22 (m, 1H); RMN ^{13}C (CDCl_3) δ 22.88, 25.15, 38.97, 50.32, 56.47, 77.63, 81.45, 122.17, 135.93, 165.85, 167.52. **16**: mp 63°C; $[\alpha]_D^{25} -96^\circ$ (c 0.4, CH_2Cl_2); IR (neat) 2928, 1671, 1593, 1387, 1361, 1254, 1079, 941, 836 cm^{-1} ; ^1H RMN (CDCl_3) δ 0.03 (s, 6H), 0.86 (s, 9H), 2.27 (s, 3H), 2.28 (s, 3H), 3.53 (dd, 1H, $J = 7.1, 10.0$ Hz), 3.76 (dd, 1H, $J = 6.0, 10.0$ Hz), 3.98 (m, 1H), 4.44 (m, 1H), 4.81 (d, 1H, $J = 5.4$ Hz), 6.11 (ddd, 1H, $J = 2.1, 4.0, 10.3$ Hz), 6.30 (bd, 1H, $J = 10.3$ Hz); ^{13}C RMN (CDCl_3) δ -5.45, -5.36, 15.28, 18.17, 25.72, 28.87, 64.80, 72.05, 76.26, 76.50, 114.16, 120.71, 134.93, 173.31, 195.08.
11. a) Falck, J. R.; Yu, J. *Tetrahedron Lett.* **1992**, 33, 6723-6726; b) When reacted with ethyl nitroacetate under Mitsunobu conditions, alcohols are usually oxidized into carbonyl compounds *via* the acinitroesters. See: Mitsunobu, O.; Yoshida, N. *Tetrahedron Lett.* **1981**, 22, 2295-2296.
12. Assuming a chair-like (rather than a twist-boat-like) transition state, incoming nucleophiles prefer attack on γ -position of the allylic moiety from the more "axial" direction, in order to maintain maximum orbital overlap. See: P. Deslongchamps *Stereoelectronic Effects in Organic Chemistry*; Pergamon Press: Oxford, 1983; pp. 174-178.
13. Vicinal coupling constants of 5-9 Hz have been recorded in related *cis* fused bicyclic compounds, see: a) Anastassiou, A. G.; Cellura, R. P. *J. Chem. Soc., Chem. Commun.*, **1969**, 1521-1522; b) Holovka, J. M.; Grabbe, R. R.; Gardner, P. D.; Stroro, L. B.; Hill, M. L.; Van Auker, T. V. *J. Chem. Soc., Chem. Commun.*, **1969**, 1522-1523; c) ref. 2.
14. The β -C-glycosides were prepared using a palladium-catalyzed alkylation of 3,4-carbonate glycals. Submitted for publication.
15. For a related approach involving *cis*-oriented epoxy pyranose triflates and 1,3-dicarbonyl compounds, see: Al-Tel, T. H.; Al-Abed, Y.; Voelter, W. *J. Chem. Soc., Chem. Commun.*, **1994**, 1735-1736.
16. a) Baldwin, J. E.; Lusch, M.J. *Tetrahedron*, **1982**, 38, 2939-2947; b) Baldwin, J. E. *J. Chem. Soc., Chem. Commun.*, **1976**, 734-736.
17. Magid, R. M. *Tetrahedron* **1980**, 36, 1901-1930.
18. Mulzer, J.; Funk, G. *Synthesis* **1995**, 101-112 and references cited therein.
19. Farina, V. *Tetrahedron Lett.* **1989**, 30, 6645-6648.

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