ISSN 1070-3632, Russian Journal of General Chemistry, 2011, Vol. 81, No. 2, pp. 439–441. © Pleiades Publishing, Ltd., 2011. Original Russian Text © I.E. Varshalomidze, M.G. Iglenkova, A.G. Golikov, A.P. Kriven'ko, 2011, published in Zhurnal Obshchei Khimii, 2011, Vol. 81, No. 2, pp. 341–343.

> LETTERS TO THE EDITOR

Synthesis and Reactions of Cycloheptane Dienone Derivatives with Hydrazine

I. E. Varshalomidze, M. G. Iglenkova, A. G. Golikov, and A. P. Kriven'ko

Chernyshevskii Saratov State University, ul. Astrakhanskaya 83, Saratov, 410012 Russia e-mail: vie21@mail.ru

Received May 20, 2010

DOI: 10.1134/S1070363211020320

Only few publications describe the synthesis and reactions of symmetrical cross-conjugated 2,7-diarylmethylidene cycloheptanones, their biological activity and derivatives, the cycloheptapyrazolines [1–5]. Asymmetrically built cycloheptane dienone derivatives have not been reported at all. Introduction into their structure of heterocyclic substituents creates unequal electrophilic reaction centers and expands the aspects of their possible practical applications. This paper presents the data on the synthesis of new dienone derivatives of cycloheptane containing aryl (phenyl, *m*-nitrophenyl) and heterocyclic (furyl, 5-nitrofuryl, thienyl) substituents, and the study of their reactions with hydrazine.

7-Arylmethylidene-2-hetarylmethyledene cycloheptanones **I–IV** were obtained by the condensation of the corresponding monoenone derivatives with aldehydes under alkaline (for dienones **I**, **II**, **IV**) or acidic (for dienone **III**) catalysis.



 $Z = O, R = H (II), NO_2 (III); Z = S, R = H (IV).$

Yields of the formed ketones were significantly lower than the yields of their cyclopentane and cyclohexane analogs [6] and did not exceed 30% due to the reduced activity of methylene cycloheptanone owing to the conformational features. The IR spectra of the products contain absorption bands characteristic of conjugated C=O and C=C bonds (1626–1603, 1599–1548 cm⁻¹). The presence of bands of the out-ofplane bending vibrations of C=C-H (950–935 cm⁻¹) and the chemical shifts values of vinyl protons in the ¹H NMR spectrum (7.17–7.37 ppm) indicate that the dienones have *E*,*E*-configuration [6]. In the ¹³C NMR spectra the signals appear of four sp^3 -hybridized carbon atoms of cycloheptane ring [C⁴ and C⁵ (27.0, 27.9 ppm), C³ and C⁶ (28.1–28.6 ppm)], and 14 sp^2 -hybridized carbon atoms of benzene, furan and ethylene fragments (111.9–155.8 ppm), and of carbonyl carbon atom (197.9–199.0 ppm).

Using the dienones I and III containing the pharmacophore groups (5-nitrofuryl, 3-nitrophenyl) as an example, we studied their reaction with hydrazine hydrate. Refluxing the reactants (dienone:hydrazine hydrate, 1:7) in isopropyl alcohol yields the cyclo-condensation products, the cyclohepta[c]pyrazolines V, VI:



When a nitro group is present in the benzene ring, a mixture of regioisomeric cyclohepta[*c*]pyrazolines was formed (V:Va, 3:1) as a result of azacyclization involving the carbonyl group and the β - and β '-reaction centers of the substrate. In the case of the presence of a nitro group in the furan ring the reaction proceeds regioselectively to form one of the possible regioisomers VI due to the strong electron-acceptor effect and the planar arrangement of nitrofuran substituent leading to the activation of the β -carbon atom.

The IR spectra of cycloheptapyrazolines contain the bands of stretching vibrations of the secondary amino group (3339–3340 cm⁻¹), conjugated system C=C–C=N (1558–1616 cm⁻¹), nitro group (1328, 1350, 1501, 1524 cm⁻¹), C–O–C bond (furan) (1030 cm⁻¹). In the ¹H NMR spectra the location of the protons H³ and N^{3a} (4.37–4.48 and 2.95–3.06 ppm, $J_{3,3a}$ 12 Hz) corresponds to their *trans*-configuration. There are singlets of vinyl (7.31–7.33 ppm) and NH-protons (5.56–6.02 ppm). Assignment of the signals is made on the basis of comparison with the spectra of NH-hexahydroindazoles [7], whose structure was unambiguously established by the X-ray diffraction analysis.

7-(3-Nitrophenylmethylidene)-2-furfurylidenecycloheptanone (I). To a mixture of 3.8 g (20 mmol) of 2-furfurylidenecycloheptanone and 3.0 g (20 mmol) of 3-nitrobenzaldehyde in 5 ml of ethanol was dropwise added 5 ml of 20% alkali. The formed oil was ground with hexane to give fine yellow crystals. Yield 1.3 g (16%), mp 147–149°C (2-propanol). IR spectrum, v, cm⁻¹: 1662 (C=O), 1560–1624 (C=C), 1022 [C–O–C(Fu)]. ¹H NMR spectrum, δ, ppm, (*J*, Hz): 1.94–2.03 m (4H, H⁴, H⁵), 2.64–2.88 m (4H, H³, H⁶), 6.49–6.63, 7.17 s (1H, =CH–Ar), 7.33 s (1H, =CH–Fu), 7.52–7.72 (3H, Fu), 8.14–8.30 m (4H, Ar). ¹³C NMR spectrum, δ_{C} , ppm: 27.0, 27.4 (C⁴, C⁵), 28.2, 28.3 (C³, C⁶), 113.0–154.1 (C^{2,2',7,7'}, furan, benzene ring), 197.9 (C=O). Found, %: C 70.64; H 5.36; N 4.25. C₁₉H₁₇NO. Calculated, %: C 70.59; H 5.26; N 4.33.

7-Benzylidene-2-furfurylidenecycloheptanone (II) was prepared similarly from 1.76 g (8.8 mmol) of 2benzylidenecycloheptanone and 0.85 g (8.8 mmol) of furfural. Yield 0.6 g (25%), mp 90–92°C (2-propanol). IR spectrum, v, cm⁻¹: 1626 (C=O), 1548–1603 (C=C), 1016 [C–O–C(Fu)]. ¹H NMR spectrum, δ , ppm, (*J*, Hz): 1.91–1.94 m (4H, H⁴, H⁵), 2.66–2.86 m (4H, H³, H⁶), 7.31 s (1H, =CH–Fu), 7.24 s (1H, =CH–Ar), 7.32–7.51 m (5H, Ar), 6.47–6.60, 7.32–7.51 (3H, Fu). ¹³C NMR spectrum, δ_{C} , ppm: 27.3, 27.8 (C⁴, C⁵), 28.3, 28.4 (C³, C⁶), 111.9–155.8 (C^{2,2',7,7''}, furan, benzene ring), 199.0 (C=O). Found, %: C 81.68; H 7.09. C₁₉H₁₈O₂. Calculated, %: C 82.00; H 6.47.

7-Benzylidene-2-(5-nitrofurylmethylidene)cycloheptanone (III). To a mixture of 1.7 g (12 mmol) of 2-benzylidenecycloheptanone and 2.43 g (12 mmol) of 5-nitrofurfural in 5 ml of CH₃COOH was added 1–2 drops of conc. H₂SO₄. Yield 0.47 g (12%), yellow needle crystals, mp 167–168°C (2-propanol). IR spectrum, v, cm⁻¹: 1667 (C=O), 1558–1626 (C=C), 1022 [C–O–C(Fu)]. ¹H NMR spectrum, δ , ppm, (*J*, Hz): 1.92–2.02 m (4H, H⁴, H⁵), 2.66–2.93 m (4H, H³, H⁶), 7.31 s (1H, =CH–Ar), 7.33 s [1H, =CH–(5-NO₂) Fu], 7.32–7.51 m (5H, Ar), 6.68–6.97, 7.32–7.51 [2H, (5-NO₂)Fu]. ¹³C NMR spectrum, δ_C , ppm: 27.3, 27.9 (C⁴, C⁵), 28.1, 28.6 (C³, C⁶), 112.0–152.0 (C^{2,2',7,7'}, furan, benzene ring), 199.0 (C=O). Found, %: C 71.08; H 5.56; N 4.16. C₁₉H₁₇NO. Calculated, %: C 70.59; H 5.26; N 4.33.

7-Benzylidene-2-thienylidenecycloheptanone (IV) was obtained similarly from 2 g (10 mmol) of 2benzylidenecycloheptanone and 1.12 g (10 mol) of thiophene aldehyde. Yield 0.9 g (30%), mp 132–134°C (2-propanol). IR spectrum, v, cm⁻¹: 1666 (C=O), 1570–1618 (C=C), 768 [C–S–C(Th)]. ¹H NMR spectrum, δ , ppm, (*J*, Hz): 1.93–2.12 m (4H, H⁴, H⁵), 2.63–2.78 m (4H, H³, H⁶), 7.36 s (1H, =CH–Ar), 7.37 s (1H, =CH–Th), 7.06–7.49 (3H, Th), 7.26–7.49 m (5H, Ar). Found, %: C 77.11; H 5.99; S 10.52. C₁₉H₁₈OS. Calculated, %: C 77.55; H 6.12; S 10.88.

trans-3-(3-Nitrophenyl)-8-(2-furfurylidene)-3,3adihydrocyclohepta[c]pyrazole (V) and trans-3furyl-8-(3-nitrophenylmethylidene)-3,3a-dihydrocvclohepta[c]pvrazole (Va). A mixture of 0.5 g (1.5 mmol) of 2-(3-nitrophenylmethylidene)-7-(2-furfurylidene)cycloheptanone and 0.5 ml (10 mmol) of hvdrazine hvdrate in isopropanol was refluxed for 15 min. On cooling, the yellow crystals precipitated. Yield 0.11 g (21%), mp 134–136°C (2-propanol). IR spectrum, v, cm⁻¹: 3339 (NH), 3161–3092 [CH(Ph)], 1578, 1558 (C=C-C=N), 1524, 1350 (NO₂), 1030 [C-O(Fu)]. ¹H NMR spectrum, δ , ppm (J, Hz): 1.18– 1.69 m (6H, H^{4,5,6}), 1.82–2.25 m (2H, H⁷), 2.95–3.03 m (1H, H^{3a}), 4.48 d, 4.42 d (1H, H³, J 12), 5.56 s (1H, NH), 6.96 s (1H, =CH-Fu), 7.24-7.47 m (3H, Fu), 7.47-8.40 m (4H, Ar). Found, %: C 67.19; H 5.83; N 12.78. C₁₉H₁₉NO. Calculated, %: C 67.66; H 5.64; N 12.46.

trans-3-Phenyl-8-(5-nitro-2-furylmethylidene)-3,3a-dihydrocyclohepta[c]pyrazole (VI) was obtained similarly from 0.16 g (0.5 mmol) of 2-(benzylidene)-7-(5-nitrofurylmethylidene)cycloheptanone and 0.5 ml (10 mmol) of hydrazine hydrate. Yield 0.08 g (48%), mp 113–114°C (2-propanol). IR spectrum, v, cm⁻¹: 3340 (NH), 3158–3029 [CH(Ph]], 1576, 1616 (C=C-C=N), 1501, 1328 (NO₂), 1030 [C-O-C(Fu]]. ¹H NMR spectrum, δ , ppm (*J*, Hz): 1.39–2.18 m (6H, H^{4,5,6}), 2.95–3.06 m (1H, H^{3a}), 3.01– 3.06 m (2H, H⁷), 4.37 d (1H, H³, *J* 12), 5.97 s (1H, NH), 6.48–6.49 m [2H, (5-NO₂–Fu], 7.01 s [1H, =CH– (5-NO₂–Fu]], 7.26–7.37 m (5H, Ar). Found, %: C 68.05; H 5.94; N 12.02. C₁₉H₁₉NO. Calculated, %: C 67.66; H 5.64; N 12.46.

The IR spectra were recorded on a FSM-1201 Fourier-spectrometer (from KBr pellets). The ¹H and ¹³C NMR spectra were registered on a Varian 400 spectrometer in CDCl₃ relative to internal TMS. The reaction progress and purity of the compounds obtained were monitored by TLC method on Silufol UV-254 plates eluting with hexane–2-propanol–chloroform mixture (3:1:1).

REFERENCES

- 1. Tsukerman, S.V., Kutulya, L.A., and Lavrushin, V.F., *Zh. Org. Khim.*, 1964, no. 11, p. 3597.
- Kurt, A., Hans, W., and Horst, K., *Liebigs Ann. Chem.*, 1956, vol. 601, nos. 1–3, p. 138.
- 3. Kabli, R.A., Kaddah, A.M., Khalil, A.M., and Khalaf, A.A., *Indian J. Chem. B*, 1986, vol. 25, p. 152.
- 4. Krapcho, J. and Turk, C.F., USA Patent no. 3960848, 1976.
- 5. Scanlon, W.B., USA Patent no. 3857953, 1974.
- 6. Kriven'ko, A.P., Bugaev, A.A., and Golikov, A.G., *Khim. Geterotsikl. Soed.*, 2005, no. 2, p. 191.
- Bugaev, A.A., Golikov, A.G., Fomina, Yu.A., and Egorov, S.V., *Izv. Vuzov, Ser. Khim. i Khim. Tekhnol.*, 2005, vol. 48, no. 4, p. 84.