Transannular reactions in the dibenzo [a,d] cycloheptene series. III. Preparation of 11-substituted-10,11-dihydro-10,5-(iminomethano)-5*H*-dibenzo [a,d] cycloheptenes

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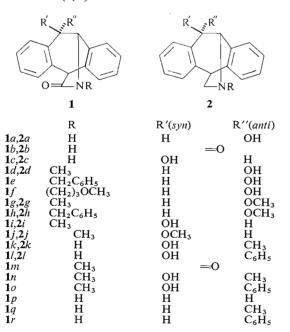
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Treatment of the *syn*-epoxyamide 3 with either ammonium hydroxide or sodium hydride gives 10,11dihydro-*anti*-11-hydroxy-10,5-(iminomethano)-5*H*-dibenzo[*a*,*d*]cyclohepten-13-one (1*a*). This compound is readily converted to the *sym*-epimer 1*c* by oxidation to 1*b* and subsequent hydrogenation. The ketone 1*b* reacts with Grignard reagents to give the tertiary alcohols 1k, *l* which undergo hydrogenolysis to give the 11-substituted lactams 1q, *r*. Reduction of the lactams with lithium aluminium hydride gives the corresponding amines.

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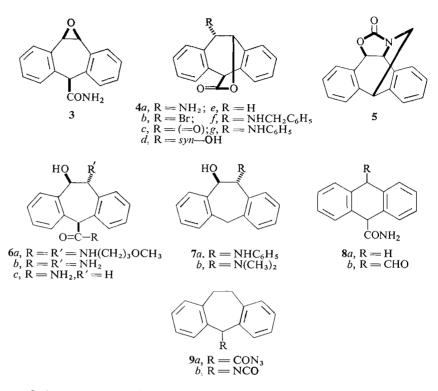
We have recently described the preparation of a number of 10,11-dihydro-10,5-(epoxymethano)-5H-dibenzo [a,d]cyclohepten-13-ones (1, 2). The present paper outlines a general route to 11-substituted-10,11-dihydro-10,5-(iminomethano)-5Hdibenzo [a,d]cycloheptene derivatives 1 and 2. A different approach has been used by other workers (3-6) to prepare other derivatives and the ring system has been shown to occur naturally in the alkaloids amurensine, amurensinine, and roemfrine (7, 8).



We have previously found that the synepoxyamide **3** reacts with secondary amines to give anti-11-dialkylaminolactones. In attempts

to prepare the unsubstituted anti-11-aminolactone 4a both the syn-epoxyamide 3 and the bromolactone 4b were treated with ammonium hydroxide at 140°. The product, obtained in 70% yield in both cases, was not the expected lactone but the anti-11-hydroxylactam 1a. Oxidation of 1a gave the ketolactam 1b which regenerated 1aupon reduction with sodium borohydride. In contrast, catalytic hydrogenation of 1b gave a mixture of 20% 1a and 80% of the syn-hydroxylactam 1c. This hydrogenation parallels that of the ketolactone 4c which gives the syn-hydroxylactone 4d as the major product (2). Reduction of the two epimers 1a and 1c with lithium aluminium hydride in dimethoxyethane gave the epimeric aminoalcohols 2a and 2c respectively. Other solvents (see Experimental) were markedly inferior for these reductions in accordance with experience in the preparation of 8-hydroxy-5phenylbenzomorphan derivatives (9). The infrared (i.r.) hydroxyl absorptions of 2a and 2c were concentration-dependent, and concentration-independent respectively, indicating that 2a was the anti-epimer and that 2c was the syn-epimer (10). These assignments were confirmed by the reaction of the syn-aminoalcohol 2c with phosgene. The product was the cyclic carbamate 5, the structure of which was fully supported by its analytical and spectral properties. The geometry of 2a and 2c, as shown by inspection of molecular models, indicates that only 2c can form 5. Cristol and Bly (11) have cited an analogous cyclic carbonate formation as a structure proof of dibenzobicyclo [3.2.1] octadiene-exo-4-syn-8-diol. In accord with these assignments only intractable mixtures were obtained when 2a was treated with either phosgene or diethyl oxalate (12).

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Treatment of the syn-epoxyamide 3 with a variety of primary amines at 140° gave the Nsubstituted-anti-hydroxylactams 1d-f. The same compounds were prepared by monoalkylating 1a with sodium hydride and the appropriate alkyl halide confirming that 1a, d-f belonged to the same epimeric series. Subsequent methylation of 1*d*,*e* gave the ethers 1*g*,*h*. The alkylated derivatives 1i, j of the syn-hydroxylactam were prepared in the same manner. Reduction of 1d,g,h and 1*i*, *j* with lithium aluminium hydride gave the two epimeric series of amines 2d,g,h and 2i,jrespectively. Unlike the nuclear magnetic resonance (n.m.r.) spectra of the previously described lactones (2), the n.m.r. spectra of the lactams and amines described herein cannot be used to distinguish between syn- and anti-substitution.

Under milder conditions two exceptions to the generality of the amine-epoxyamide reaction were observed. First, treatment of **3** with benzylamine at 100° gave the *anti*-11-benzylaminolactone **4**f. The geometry of this compound was confirmed by its n.m.r. spectrum (2). Subsequent treatment of **4**f with either benzylamine or ammonium hydroxide at 140° gave the *N*-benzyl-*anti*-hydroxylactam **1**e. Second, treatment of **3** with

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3-methoxypropylamine at 80° gave a small yield of the secondary amide 6a.

These observations suggest that 1a arises by normal *trans*-opening of the epoxide function (13) with, most probably, simultaneous lactonization to give the initially expected aminolactone 4a. Epimerization of 4a at the benzhydrilic carbon atom (C-5) via an equilibrium concentration of the amide 6b followed by lactam formation would give 1a. It is improbable that 6b epimerizes without prior lactonization. For example, the synhydroxyamide 6c lactonizes at room temperature (2). The ready lactam formation of the epimerized form of 6b is unremarkable in view of the classical methods for forming lactam bridges (14).

It was not possible to convert the *anti*-anilinolactone 4g to the corresponding lactam by treatment with ammonium hydroxide since starting material or the aminoalcohol 7a were obtained. Presumably the anilino residue was insufficiently basic to undergo lactam formation. Decarboxylation also occurred when the *syn*-epoxyamide **3** was heated with aqueous dimethylamine at 140° since the product was the aminoalcohol 7b.

The present studies bear a formal similarity to the preparation of isoquinuclidones recently

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reported by Huffman and co-workers (15). Their procedure requires, however, a *trans*epoxyester as starting material and subsequent pyrolysis of the intermediate aminoester.

Treatment of the syn-epoxyamide 3 with sodium hydride in dry dioxane also gave 1a. Presumably this reaction proceeds via epimerization at C-5 and subsequent 'N-6' participation (16). Many examples are known of nucleophilic participation by amide nitrogen under basic conditions (17). A by-product from this reaction was 9,10-dihydroanthracene-9-carboxamide 8a. This side reaction was undoubtedly a basepromoted conversion of 3 since it was unaffected by both boiling dioxane and boiling water, and the anti-hydroxylactam 1a was unaffected by sodium hydride in boiling dioxane. Possibly 8a arose by initial rearrangement¹ of 3 to the carboxaldehyde 8b with subsequent base-catalyzed elimination of the formyl group (11, 19).

Attempts to ammonolyze the lactones 4d, e to the corresponding lactams were abortive since starting materials or decarboxylated products were obtained. In contrast, the closely related isochroman-3-one and its 4-phenyl derivative have been successfully ammonolyzed to the corresponding lactams (20, 21).

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The ketolactam 1b showed normal ketonic properties. Enolization, which sometimes masks the ketonic character of 10,11-dihydro-5*H*dibenzo[*a,d*]cyclohepten-10-ones (22), is precluded by the presence of the 10,5 bridge system. Thus, 1b formed an oxime and a ketal and readily underwent Grignard reactions to give the carbinols 1k,l.² In these Grignard reactions the lactam function was presumably protected from attack by complex formation (23, p. 876). Grignard reactions with the *N*-methyl-ketolactam Im gave very poor yields of the carbinols 1n, o and these compounds were better prepared by methylating 1k,l. Lithium aluminium hydride reduction of 1k,l gave the aminoalcohols 2k,l. The i.r.

anthraldehyde on either pyrolysis or acid-treatment (18). ²The physical properties and chemical behavior of lk_i confirmed that they were single cpimers. The reaction of lb with allyl magnesium bromide appeared to give a mixture of epimers which is being investigated further. Ethyl magnesium bromide reduced lb to the *anti*hydroxylactam (23, p. 147–158). hydroxyl absorptions of these compounds were concentration-independent indicating that they were *syn*-aminoalcohols. This geometry would be expected assuming that Grignard reagents, like hydrogen, tend to attack the less hindered side of 1b.

Hydrogenolysis of the epimeric hydroxylactams 1a and 1c with sodium in liquid ammonia gave the unsubstituted lactam 1p in good yield. Similar hydrogenolyses of the carbinols 1k, l gave 1q,r. Catalytic hydrogenolyses of 1a and 1k were unsuccessful, as is the case with similarly substituted benzylic alcohols (9).

The unsubstituted lactam 1p was also obtained, in very poor yield, by photolysis of the acyl azide 9a. The major product was the isocyanate 9b in accord with experience in the photolysis of carbocylic acyl azides (24, 25).

The more important pharmacological properties of some of the compounds described in this paper, particularly those of the amines 2a-m, will be described elsewhere.

Experimental

10,11-Dihydro-anti-11-hydroxy-10,5-(iminomethano)-5H-dibenzo[a,d]cyclohepten-13-one (1a)

Method a

A mixture of the *syn*-epoxyamide **3** (50 g) (2) and ammonium hydroxide (200 ml) was kept at 140° for 6 h. The solid product was collected and washed with water, chloroform and ether, and then crystallized from ethanol to give the title product (35 g) as needles, m.p. $260-262^{\circ}$ (decomp.); v_{max} (Nujol) 1670 cm⁻¹.

Anal. Calcd. for $C_{16}H_{13}NO_2$: C, 76.47; H, 5.22; N, 5.57. Found: C, 76.50; H, 5.08; N, 5.57.

Method b

Following the procedure of method a, the bromolactone 4b (315 g) (2) and ammonium hydroxide (2 l) gave 1a (178 g).

Method c

A mixture of the *syn*-epoxyamide 3 (2.51 g), sodium hydride (460 mg of a 52% suspension in mineral oil), and dioxane (50 ml) was heated under reflux for 1 h. The reaction mixture was diluted with water and then concentrated *in vacuo*. The residue was washed with water, and then with hexane, and the residue was crystallized from ethanol to give 1a (1.5 g), m.p. 260–262° (decomp.). The infrared (i.r.) spectrum of this material was identical with that of the product from method a.

The mother liquors from the above crystallization yielded 9,10-dihydroanthracene-9-carboxamide (8*a*), m.p. and mixture m.p. with an authentic sample $151-152^{\circ}$ (lit. (26), m.p. 151-152°).

10,11-Dihydro-10,5-(iminomethano)-5H-dibenzo[a,d]-

cyclohepten-11,13-dione (1b)

Jones reagent (20 ml) (27) was added dropwise to an ice-cold stirred suspension of 1a (10.0 g) in acetone (150

¹The influence of the 5-proton is manifest in this rearrangement. Thus, 10,11-epoxy-10,11-dihydro-5H-dibenzo[a,d]cyclohepten-5-one gave the corresponding 5,10-dione on base treatment, and gave 10-hydroxy-9-anthraldehyde on either pyrolysis or acid-treatment (18).

ml). The mixture was kept at 0° for 0.5 h and then diluted with isopropanol (5.0 ml). The mixture was concentrated *in vacuo*, diluted with water, and then extracted with chloroform. The combined extracts were washed with water, dried, evaporated, and the residue was crystallized from ethanol to give 1b (9.0 g), m.p. 197–199°; v_{max} (CHCl₃) 3425, 3200, and 1685 cm⁻¹.

Anal. Calcd. for C₁₆H₁₁NO₂: C, 77.09; H, 4.45; N, 5.62. Found: C, 77.34; H, 4.36; N, 5.78.

The oxime derivative of 1b was prepared in the usual manner and recrystallized from ethanol, m.p. $290-295^{\circ}$ (decomp.).

Anal. Calcd. for C₁₆H₁₂N₂O₂: C, 72.71; H, 4.58; N, 10.60. Found: C, 72.40; H, 4.69; N, 10.75.

Treatment of 1b (1.0 g) with sodium borohydride (0.5 g) in ethanol (30 ml) at room temperature gave 1a (0.93 g) identical in all respects with 1a prepared by method a above.

10,11-Dihydro-syn-11-hydroxy-10,5-(inninomethano)-5H-dibenzo[a,d]cyclohepten-13-one (1c)

A solution of the ketolactam 1b (100 g) in ethanol (3.0 l) was hydrogenated at 50° and 7 atm in the presence of 10% palladium-on-charcoal catalyst (0.5 g) until hydrogen uptake ceased. The catalyst was removed and the solution was concentrated to half-volume whereupon 1c (60 g) crystallized as fine needles, m.p. 260–263° (decomp.); v_{max} (Nujol) 1665 cm⁻¹.

Anal. Calcd. for $C_{16}H_{13}NO_2$: C, 76.47; H, 5.22; N, 5.57. Found: C, 76.35; H, 5.13; N, 5.21.

Fractional crystallization of the residue from the mother liquors gave a further 19.0 g of 1c and 17.0 g of 1a.

10,11-Dihydro-10,5-(iminomethano)-5H-dibenzo[a,d]cyclohepten-anti-11-ol (2a)

Lithium aluminium hydride (7.25 g) was added portionwise to a stirred suspension of 1*a* (25.1 g) in 1,2-dimethoxyethane (150 ml). The mixture was stirred and heated under reflux for 10 h. The excess hydride was destroyed with water and the basic material was isolated in the usual manner. This material was recrystallized from benzene-hexane to give 2*a* (19.0 g), m.p. 130-134°; v_{max} (CHCl₃) 3570, 3325, and 1020 cm⁻¹.

Anal. Calcd. for C₁₆H₁₅NO: C, 80.98; H, 6.37; N, 5.90. Found: C, 80.97; H, 6.65; N, 5.71.

The hydrochloride salt of 2a was recrystallized from ethanol, m.p. 220–225° (decomp.).

Anal. Calcd. for $C_{16}H_{16}$ ClNO: C, 70.30; H, 5.85; Cl, 12.99; N, 5.12. Found: C, 70.52; H, 6.04; Cl, 12.70; N, 4.81.

Reduction of 1*a* with lithium aluminium hydride in ether gave mainly starting material; similar reductions of 1*a* in tetrahydrofuran and in dioxane gave 40% and 22% yields respectively of 2a.

10,11-Dihydro-10,5-(iminomethano)-5H-dibenzo[a,d]cyclohepten-syn-11-ol (2c)

The syn-hydroxylactam 1c (30 g) was treated with lithium aluminium hydride (6.0 g) in dimethoxyethane (200 ml) and then processed as described above. The crude basic material was recrystallized from methanol to give 2c (24.1 g), m.p. 191–193°; v_{max} (CHCl₃) 3300, 1488, and 1308 cm⁻¹.

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Anal. Calcd. for C₁₆H₁₅NO: C, 80.98; H, 6.37; N, 5.90. Found: C, 80.78; H, 6.42; N, 5.69.

3a,12b-Dihydro-3,8-methano-2H,8H-dibenzo[3,4:6,7]cyclohepta[1,2-d]oxazol-2-one (5)

A 4% solution of phosgene in benzene (30 ml) was added dropwise during 3 h to a solution of 2c (2.63 g) in dioxane (100 ml) and pyridine (4 drops). The mixture was kept at room temperature for 1 h and then evaporated to dryness. The residue was partitioned between chloroform and water and the organic phase was evaporated to leave the crude product. This material was recrystallized from methanol to give 5 (1.0 g), m.p. 226-228° (decomp.); v_{max} (CHCl₃) 1758 cm⁻¹.

Anal. Calcd. for $C_{17}H_{13}NO_2$: C, 77.55; H, 4.98; N, 5.32. Found: C, 77.59; H, 5.00; N, 5.17.

Similar treatment of the *anti*-aminoalcohol 2a with phosgene gave a solid product, v_{max} (CHCl₃) 1700 cm⁻¹. This material was shown to be inhomogeneous by thin-layer chromatography and it could not be purified.

Preparation of the anti-Hydroxylactams 1d-f Method a

Under the conditions described for the preparation of 1a (method a) the following compounds were obtained.

10,11-Dihydro-*anti*-11-hydroxy-12-methyl-10,5-(iminomethano)-5*H*-dibenzo [*a*,*d*]cyclohepten-13-one (1*d*) (66.0 g), m.p. 267–270° (from ethanol); v_{max} (Nujol) 3415, 3260, and 1650 cm⁻¹ was obtained from 3 (100 g) and aqueous methylamine (600 ml of a 30% solution).

Anal. Calcd. for C₁₇H₁₅NO₂: C, 76.96; H, 5.70; N, 5.28. Found: C, 76.96; H, 5.53; N, 4.99.

12-Benzyl-10, 11-dihydro-*anti*-11-hydroxy-10, 5-(iminomethano)-5*H*-dibenzo[*a*,*d*]cyclohepten-13-one (1*e*) (10.0 g), m.p. 224-226° (from ethanol); v_{max} (CHCl₃) 3570, and 1660 cm⁻¹ was obtained from 3 (30 g), benzylamine (30 g) and water (80 ml).

Anal. Calcd. for C₂₃H₁₉NO₂: C, 80.91; H, 5.61; N, 4.10. Found: C, 80.99; H, 5.48; N, 4.25.

10,11-Dihydro-*anti*-11-hydroxy-12-(3'-methoxy-propyl)-10,5-(iminomethano)-5*H*-dibenzo[*a*,*d*]cyclo-hepten-13-one (1*f*) (3.0 g), m.p. 150–152° (from ethyl acetate); v_{max} (CHCl₃) 3570, 1660, and 1100 cm⁻¹ was obtained from 3 (5.0 g), 3-methoxypropylamine (9.0 g), and water (40 ml).

Anal. Calcd. for $C_{20}H_{21}NO_3$: C, 74.28; H, 6.55; N, 4.33. Found: C, 74.30; H, 6.48; N, 4.39.

Method b

A mixture of the *anti*-hydroxylactam 1a (5.02 g, 0.02 mole), sodium hydride (920 mg of a 52% suspension in mineral oil, 0.02 mole), methyl iodide (3.12 g), and dioxane (60 ml) was stirred and heated under reflux for 4 h. The mixture was evaporated and the residue was washed with hexane and then with water and then recrystallized from ethanol to give 1d (4.8 g), m.p. $265-269^{\circ}$. The i.r. spectrum of this material was identical with that of 1d prepared by method a.

anti-11-Benzylamino-10,11-dihydro-10,5-(epoxymethano)-5H-dibenzo[a,d]cyclohepten-13-one (4f)

A mixture of the *syn*-epoxyamide **3** (25.0 g), benzylamine (30.0 g), and water (50 ml) was stirred and heated under reflux for 12 h. The reaction mixture was extracted

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Anal. Calcd. for $C_{23}H_{19}NO_2$: C, 80.91; H, 5.61; N, 4.10. Found: C, 81.16; H, 5.83; N, 3.98.

From the mother liquors of the above crystallization there was obtained 1e (11.0 g), m.p. and mixture m.p. 224-226°.

10,11-Dihydro-10-hydroxy-N-(3-methoxypropyl)-11-[N-(3-methoxypropylamino)]-5H-dibenzo[a,d]cyclo-

hepten-5-carboxamide (6a)

A solution of the *syn*-epoxyamide 3 (10.0 g) and 3methoxypropylamine (10.0 g) in benzene (100 ml) was stirred and heated under reflux for 18 h. The resulting solution was concentrated and the concentrate was chromatographed upon alumina. The first material eluted was the lactam 1 f (3.0 g). The second material eluted was the title product 6a (2.5 g) which was recrystallized from isopropanol to m.p. $123-125^\circ$; v_{max} (CHCl₃) 3420, 3370, 1655, and 1110 cm⁻¹.

Anal. Calcd. for $C_{24}H_{32}N_2O_4$: C, 69.88; H, 7.82; N, 6.79. Found: C, 69.98; H, 8.02; N, 6.67.

10,11-Dihydro-anti-11-methoxy-12-methyl-10,5-

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(iminomethano)-5H-dibenzo[a,d]cyclohepten-13-one (1g)

A suspension of 1d (5.30 g, 0.02 mole), sodium hydride (1.38 g of a 52% suspension in mineral oil), methyl iodide (4.26 g, 0.03 mole), and dioxane (50 ml) was stirred and heated under reflux for 18 h. The mixture was evaporated and the residue was washed with water and then with hexane and then recrystallized from methanol to give 1g (5.0 g), m.p. 212–215°; v_{max} (CHCl₃) 1660 cm⁻¹.

Anal. Calcd. for $C_{18}H_{17}NO_2$: C, 77.39; H, 6.13; N, 5.01. Found: C, 77.29; H, 6.05; N, 5.12.

12-Benzyl-10,11-dihydro-anti-11-methoxy-10,5-(imino-

methano)-5*H*-dibenzo[a,d]cyclohepten-13-one (1h) This compound was obtained from 1e and methyl iodide following the procedure described above. The product crystallized from methanol, m.p. $152-154^{\circ}$; v_{max} (CHCl₃) 1652 cm⁻¹.

Anal. Calcd. for $C_{24}H_{21}NO_2$: C, 81.10; H, 5.96; N, 3.94. Found: C, 80.99; H, 6.34; N, 4.24.

10,11-Dihydro-syn-11-hydroxy-12-methyl-10,5-(iminomethano)-5H-dibenzo[a,d]cyclohepten-13-one (1i)

This compound was obtained from the *syn*-hydroxylactam 1c (5.02 g) and methyl iodide (3.12 g) under the conditions described for the preparation of 1d (method b). It was purified from methanol as needles (4.0 g), m.p. 218-220°; v_{max} (Nujol) 1668 cm⁻¹.

Anal. Calcd. for $C_{17}H_{15}NO_2$: C, 76.96; H, 5.70; N, 5.28. Found: C, 76.76; H, 5.76; N, 5.20.

10,11-Dihydro-syn-11-methoxy-12-methyl-10,5-(imino-

methano)-5*H*-dibenzo[a,d]cyclohepten-13-one (1j) This compound was obtained from 1*i* under the conditions described for the preparation of 1g. It was recrystallized from benzene as needles, m.p. 189–191°; v_{max} (CHCl₃) 1666 cm⁻¹. Anal. Calcd. for C₁₈H₁₇NO₂: C, 77.39; H, 6.13; N, 5.01. Found: C, 77.13; H, 6.03; N, 4.99.

10,11-Dihydro-syn-11-hydroxy-anti-11-methyl-10,5-

(iminomethano)-5H-dibenzo[a,d]cyclohepten-13-one (1k)

A 3 *M* ethereal solution of methyl magnesium bromide (80 ml) was added dropwise to a solution of the ketolactam 1*b* (24.9 g, 0.1 mole) in anhydrous tetrahydrofuran (200 ml). The reaction mixture was stirred and heated under reflux for 6 h and then treated with saturated ammonium chloride solution (300 ml). The organic phase was collected, dried, and evaporated. The residue was recrystallized from methanol to give 1*k* (21.2 g), m.p. 278–280°; v_{max} (Nujol) 3590, 3210, and 1690 cm⁻¹.

Anal. Calcd. for $C_{17}H_{15}NO_2$: C, 76.97; H, 5.70; N, 5.28. Found: C, 76.88; H, 5.64; N, 5.29.

10,11-Dihydro-syn-11-hydroxy-anti-11-phenyl-10,5-

(iminomethano)-5H-dibenzo[a,d]cyclohepten-13-one (11)

This compound was obtained from 1b (10 g) and phenyl magnesium bromide (0.16 mole) under the conditions used for the preparation of 1k. The crude product was recrystallized from ethanol to give 1l (8.0 g), m.p. 280–285° (decomp.); v_{max} (Nujol) 1652 cm⁻¹.

Anal. Calcd. for C₂₂H₁₇NO₂: C, 80.71; H, 5.23; N, 4.28. Found: C, 80.72; H, 5.30; N, 4.11.

10,11-Dihydro-12-methyl-10,5-(iminomethano)-514dibenzofa,d/cycloheptene-11,13-dione (1m)

Oxidation of 1d (10 g) with Jones reagent (20 ml) (27) gave the above product. It was purified from methanol,

m.p. 153–155°. Anal. Calcd. for $C_{17}H_{13}NO_2$: C, 77.55; H, 4.98; N, 5.32. Found: C, 77.78; H, 4.82; N, 5.30.

Preparation of the Lactams 1n and 10

Method a

Under the conditions described for the preparation of 1k the following compounds were obtained.

10,11-Dihydro-syn-11-hydroxy-anti-11-methyl-12methyl-10,5-(iminomethano)-5*H*-dibenzo[a,d]cyclohepten-13-one (1n) (1.3 g), m.p. 224-227° (from methanol); v_{max} (CHCl₃) 3600, 3430, and 1667 cm⁻¹ from the *N*-methyl-ketolactam 1m (10.0 g) and methyl magnesium bromide (0.1 mole).

Anal. Calcd. for C₁₈H₁₇NO₂: C, 77.39; H, 6.13; N, 5.01. Found: C, 77.25; H, 6.08; N, 5.18.

10,11-Dihydro-*syn*-11-hydroxy-12-methyl-*anti*-11phenyl-10,5-(iminomethano)-5*H*-dibenzo [*a*,*d*]cyclohepten-13-one (1*o*) (0.3 g), m.p. 197–199° (from benzenehexane); v_{max} (Nujol) 1650 cm⁻¹ from 1*m* (5.0 g) and phenyl magnesium bromide (0.05 mole).

Anal. Calcd. for $C_{23}H_{19}NO_2$: C, 80.91; H, 5.61; N, 4.10. Found: C, 80.83; H, 5.54; N, 3.80.

Method b

Methylation of 1k (5.3 g, 0.02 mole) with sodium hydride (920 mg of a 52% suspension in mineral oil, 0.02 mole) and methyl iodide (3.32 g, 0.22 mole) in anhydrous dioxane gave 1n (5.0 g), m.p. and mixture m.p. 224-227°.

A similar methylation of 1/(3.27 g, 0.01 mole) gave 1o (3.0 g), m.p. and mixture m.p. $197-199^{\circ}$.

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10,11-Dihydro-10,5-(iminomethano)-5H-dibenzo[a,d]cyclohepten-13-one (1p)

Method a

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Small pieces of sodium were added to a mechanically stirred suspension of the *anti*-hydroxylactam 1*a* (50.0 g) in freshly distilled liquid ammonia (500 ml) until the mixture remained blue for 5 min. An excess of ammonium chloride was then added and the ammonia was allowed to evaporate. The residue was extracted with hot water and the insoluble material was recrystallized from ethanol to give 1*p* (45.0 g), m.p. 242–245°; v_{max} (Nujol) 1675 cm⁻¹.

to give 1p (45.0 g), m.p. $242-245^{\circ}$; γ_{max} (Nujol) 1675 cm⁻¹. Anal. Calcd. for C₁₆H₁₃NO: C, 81.68; H, 5.57; N, 5.95. Found: C, 81.67; H, 5.62; N, 5.91.

The same product was obtained when the *syn*-hydroxylactam 1c was used as starting material.

Method b

A solution of sodium azide (0.7 g) in water (5 ml) was added to a solution of 10,11-dihydro-5*H*-dibenzo [*a*,*d*]cycloheptene-5-carbonyl chloride (2.56 g) (26) in dioxane (10 ml) at 0°. The mixture was stirred at 0° for 45 min, diluted with iced water, and extracted with hexane (3 × 20 ml portion). The combined extracts were washed with cold water and then dried at 0° with molecular sieves. The solution of the acid azide 9*a* so obtained was photolyzed at 0° until an aliquot no longer absorbed energy at 2140 em⁻¹. The precipitate which had formed was crystallized from benzene–hexane and then from ethanol to give 1*p* (200 mg). Concentration of the hexane filtrate gave the isocyanate 9*b* (1.5 g), m.p. 60–62° (lit. (26), m.p. 62–63°).

Preparation of the Lactams 1q and 1r

Under the conditions used for the preparation of 1p (method a) the following compounds were obtained.

10,11-Dihydro-anti-11-methyl-10,5-(iminomethano)-5*H*-dibenzo [*a*, *d*]cyclohepten-13-one (1*q*) (10.7 g), m.p. 243-246° (from methanol); v_{max} (CHCl₃) 3410, 3200, and 1675 cm⁻¹ was obtained from 1*k* (12.0 g).

Anal. Calcd. for $C_{17}H_{15}NO$: C, 81.90; H, 6.06; N, 5.62. Found: C, 81.87; H, 5.94; N, 5.70.

10,11-Dihydro-*anti*-11-phenyl-10,5-(iminomethano)-5*H*-dibenzo[*a,d*]cyclohepten-13-one (1*r*) (2.8 g), m.p. 250– 252° (from methanol); v_{max} (CHCl₃) 1670 cm⁻¹ was obtained from 1/(3.3 g).

Anal. Calcd. for $\tilde{C}_{22}H_{17}NO$: C, 84.86; H, 5.50; N, 4.50. Found: C, 84.68; H, 5.71; N, 4.36.

Treatment of the Lactones 4d, 4e, and 4g with Ammonium Hydroxide

(a) A solution of the syn-hydroxylactone 4d (300 mg) in ammonium hydroxide (10 ml) was heated at 120° for 5 h in an autoclave. The i.r. spectrum of the solid product was devoid of carbonyl absorption. A small amount of the starting material was recovered from the ammoniacal solution.

(b) A solution of the unsubstituted lactone 4e (10 g) in ammonium hydroxide (200 ml) was kept at 160° for 5 h. The reaction mixture was extracted with ether. Evaporation of the washed and dried extracts and recrystallization of the residue from hexane gave 10,11-dihydro-5*H*-dibenzo[*a*,*d*]cyclohepten-10-ol (7.0 g), m.p. 63-65° (lit. (28), m.p. 64-66°).

Treatment of 4e with liquid ammonia at 200° gave an intractable resin.

(c) A suspension of the *anti*-anilinolactone 4g (4.0 g)

in animonium hydroxide (80 ml) was heated at 170° for 6 h. The mixture was processed as described above to give 11-anilino-10,11-dihydro-5*H*-dibenzo[a,d]cyclohepten-10-ol (7a) (2.4 g), m.p. 177–179° (from ethanol).

Anal. Calcd. for $C_{21}H_{19}NO$: C, 83.69; H, 6.35; N, 4.65. Found: C, 83.99; H, 6.20; N, 4.86.

Similar treatment of the *syn*-epoxyamide 3 (10 g) with dimethylamine (10 g) in water (70 ml) gave 10,11-dihydro-11-dimethylamino-5*H*-dibenzo[a,d]cyclohepten-10-ol (7*b*) (7.0 g), m.p. 114–116° (lit. (29), m.p. 117–118°).

Preparations of the Amines 2d,g-1

Under the conditions described for the preparation of 2*a*, the following compounds were prepared.

10,11-Dihydro-*anti*-11-hydroxy-12-methyl-10,5-(iminomethano)-5*H*-dibenzo[*a,d*]cycloheptene (2*d*) hydrochloride (3.8 g), m.p. 220° (decomp.) (from isopropanolether); v_{max} (Nujol) 3460 and 3300 cm⁻¹ was obtained from 1*d* (5.0 g).

Anal. Calcd. for $C_{17}H_{18}CINO$: C, 70.94; H, 6.30; Cl, 12.35; N, 4.87. Found: C, 70.64; H, 6.58; Cl, 12.69; N, 4.73.

10,11-Dihydro-*anti*-11-methoxy-12-methyl-10,5-(iminomethano)-5*H*-dibenzo[a,d]cycloheptene (2g) (9.7 g), m.p. 122–125° (from benzene-hexane) was obtained from 1g (12.0 g).

Anal. Calcd. for C₁₈H₁₉NO: C, 81.47; H, 7.22; N, 5.28. Found: C, 81.28; H, 7.25; N, 5.14.

12-Benzyl-10,11-dihydro-*anti*-11-methoxy-10,5-(iminomethano)-5*H*-dibenzo[*a*,*d*]cycloheptene (2*h*) hydrochloride (3.0 g), m.p. 167–170° (decomp.) (from isopropanol) was obtained from 1*h* (5.0 g).

Anal. Calcd. for $C_{24}H_{24}CINO$: C, 76.30; H, 6.40; Cl, 9.38; N, 3.70. Found: C, 76.33; H, 6.06; Cl, 9.32; N, 3.90.

10,11-Dihydro-syn-11-hydroxy-12-methyl-10,5-(iminomethano)-5*H*-dibenzo[a,d]cycloheptene (2*i*) oxalate (6.4 g), m.p. 183–185° (decomp.) (from isopropanolether) was obtained from 1*i* (8.0 g).

Anal. Calcd. for $C_{19}H_{19}NO_5$: C, 66.85; H, 5.61; N, 4.10. Found: C, 67.61; H, 5.63; N, 4.24.

10,11-Dihydro-syn-11-methoxy-12-methyl-10,5-(iminomethano)-5*H*-dibenzo[a,d]cycloheptene (2*j*) oxalate (3.2 g), m.p. 183-186° (decomp.) (from acetonitrileether) was obtained from 1*j* (5.4 g).

Anal. Calcd. for $C_{20}H_{21}NO_5$: C, 67.59; H, 5.96; N, 3.94. Found: C, 67.29; H, 6.20; N, 3.83.

10,11-Dihydro-*syn*-11-hydroxy-*anti*-11-methyl-10,5-(iminomethano)-5*H*-dibenzo[*a*,*d*]cycloheptene (2*k*) hydrochloride (4.7 g), m.p. 233–235° (from ethanol) was obtained from 1*k* (7.0 g).

Anal. Calcd. for $C_{17}H_{19}CINO$: Cl, 12.27; N, 4.11. Found: Cl, 11.99; N, 4.60.

10,11-Dihydro-syn-11-hydroxy-anti-11-phenyl-10,5-(iminomethano)-5*H*-dibenzo[a,d]cycloheptene (2*l*) hydrochloride (4.1 g), m.p. 219–224° (decomp.) (from ethanol) was obtained from 1*l* (6.0 g).

Anal. Calcd. for $C_{22}H_{20}$ ClNO: C, 75.53; H, 5.76; Cl, 10.13; N, 4.00. Found: C, 75.68; H, 5.73; Cl, 10.21; N, 3.88.

10,11-Dihydro-10,5-(iminomethano)-5H-dibenzo[a,d]cyclohepten-11-one (2b)

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A mixture of the ketolactam 1a (25.0 g), ethylene glycol

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(20 ml), p-toluenesulfonic acid (250 mg) and benzene was stirred and heated under reflux (Dean-Stark) for 24 h. The resulting suspension was washed with 2NNaOH and the solids were collected, washed with water, and dried to give the ethylene ketal of 1a (27.0 g), m.p. 265-269° (decomp.). A mixture of the ketal (26.5 g) and lithium aluminium hydride (4.5 g) in tetrahydrofuran was heated under reflux for 18 h. The mixture was treated with water, filtered, and the filtrate was evaporated. The residue was dissolved in 2N HCl (200 ml) and kept overnight. The resulting precipitate was collected and recrystallized from ethanol to give 2b hydrochloride (17.0 g), m.p. 285° (decomp.); v_{max} (Nujol) 1685 cm⁻¹.

Anal. Calcd. for C16H14CINO: C, 70.76; H, 5.16; Cl, 13.08; N, 5.16. Found: C, 70.85; H, 5.08; Cl, 13.23; N, 5.22.

Treatment of 2b (3.0 g) with a mixture of formic acid (10 ml) and formaldehyde (10 ml) at room temperature overnight gave, after processing in the usual manner, 10,11-dihydro-12-methyl-10,5-(iminomethano)-5Hdibenzo[a,d]cyclohepten-11-one (2m), m.p. 149–151° (from ethanol); v_{max} (CHCl₃) 2800, and 1665 cm⁻¹

Anal. Calcd. for C17H15NO: C, 81.90; H, 6.06; N, 5.62. Found: C, 81.80; H, 6.09; N, 5.56.

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- 1. (a) T. A. DOBSON, M. A. DAVIS, and A. M. HARTUNG. First Intern. Congr. Heterocyclic Chem. Albuquerque, N.M. June 12th, 1967. Abstr. No. 32. (b) T.A. DOBSON, M. A. DAVIS, A. M. HARTUNG, and J. MANSON. Tetrahedron Letters, 4139 (1967).
- 2. T. A. DOBSON, M. A. DAVIS, A. M. HARTUNG, and
- I. A. DOBSON, MI. A. DAVIS, A. M. HARTUNG, and J. MANSON. Can. J. Chem. 46, 2843 (1968).
 D. A. GUTHRIE, A. W. FRANK, and C. B. PURVES. Can. J. Chem. 33, 729 (1955).
 VON E WALDMANN and C. CHWALA. Ann. Chem. 609, 125 (1957).
- A. R. BATTERSBY and D. A. YOEWELL. J. Chem. 5. Soc. 1988 (1965).
- South African Patent No. 67/2148; Derwent Farm-6. doc Abstracts, 30, No. 29,127.

- F. SĂNTAVÝ, M. MATUROVÁ, and L. HRUBAN. Chem. Commun. 36, 144 (1966). F. SĂNTAVÝ, L. HRUBAN, and M. MATUROVÁ. Collection Czech. Chem. Commun. 31, 4286 (1966).
- K. M. S. YUNUSOV, S. T. SKRAMOV, and S. YU. YUNUSOV, Khim. Prirodn. Soedin. Akad Nauk Uz. SSR. 3, 68 (1967); Chem. Abstr. 67, 11625s (1967).
 G. N. WALKER and D. ALKALAY. J. Org. Chem. 31, 2007 (1977)
- 905 (1966).
- 10. L. J. BELLAMY. The infrared spectra of complex molecules. 2nd ed. Methuen and Co., Ltd., London. 1958. p. 95ff.
- 11. S. J. CRISTOL and R. K. BLY. J. Am. Chem. Soc. 82, 6155 (1960). 12. G. DREFAHL, M. HARTMANN, and A. SKURK. Ber.
- 99, 1168 (1966).
 13. E. L. ELIEL. Stereochemistry of carbon compounds. McGraw-Hill Book Co., Inc., New York. 1962.
- MCGTaW-HIII BOOK CO., HIC., New YOR, 1962.
 p. 230ff.
 14. W. SCHNEIDER and R. DILLMAN. Ber. 96, 2377 (1963). L. H. WERNER and S. RICCA. J. Am. Chem. Soc. 80, 2733 (1958).
 15. J. W. HUFFMAN, C. B. S. RAO, and T. KAMIYA. J. Am. Chem. Soc. 87, 2288 (1965); J. Org. Chem. 32, 607 (200 (1067)).
- 697, 700 (1967).
- S. WINSTEIN, R. HECK, S. LAPPORTE, and R. BAIRD. Experientia, 12, 138 (1956).
 B. CAPON. Quart. Rev. London, 18, 71 (1964), and references therein.
- J. RIGAUDY and L. NÉDÉLEC. Bull. Soc. Chim. France, 400 (1960). 18.
- M. SHEMYAKIN and L. A. SHCHUKINA. Quart. Rev. London, 10, 261 (1956).
- 20. E. HOEFT and H. SCHULTZE. J. Prakt. Chem. 32, 12 (1966).
- J. GARDENT and M. HAMON. Bull. Soc. Chim. France, 556 (1966).
 J. O. JILEK, V. SEIDLOVÁ, E. SVÁTEK, and M. PROTIVA. Monatsh. Chem. 96, 182 (1965). M. PROTIVA. I. Editional Sciencific (Double) 21 76 (1066). Il Farmaco; Edizione Scientifica (Pavia) 21, 76 (1966).
- 23. M. KHARASCH and O. REINMUTH. Grignard reaction of non-metallic substances. Prentice-Hall, Inc., New York. 1954.
- 24. J. W. APSIMON and O. E. EDWARDS. Can. J. Chem. 40, 896 (1962).
- 25. W. L. MEYER and A. S. LEVINSON. J. Org. Chem.
- 28, 2859 (1963).
 26. M. A. DAVIS, S. O. WINTHROP, R. A. THOMAS, F. HERR, M-P. CHAREST, and R. GAUDRY. J. Med. Chem. 7, 88 (1964).
- C. DJERASSI, R. R. ENGLE, and A. BOWERS. J. Org. Chem. 21, 1547 (1956).
 F. J. VILLANI, C. A. ELLIS, C. TEICHMAN, and C. BIGOS, J. Med. Pharm. Chem. 5, 381 (1962).
 S. KULLANI, C. M. ELLIS, C. TEICHMAN, and C.
- S. KIMOTO and S. OHTA. Yakugaku Zasshi, 87, 861 29. (1967).

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