## The Total Synthesis of Destomic Acid

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1-Benzoyloxy-2-t-butyldimethylsilyloxy-4-ethoxybuta-1,3-diene (3) reacts with *N*-benzyloxycarbonyl-*O*-t-butyldiphenylsilyl-L-serinal (4) to give, with high selectivity, compound (5a) which was subsequently transformed into derivatives of destomic acid (7) and (8).

6-Amino-6-deoxy-L-glycero-D-galacto-heptonic acid (1), commonly named destomic acid, is one of the three components of a new type of aminocyclitol antibiotic: destomycin A,<sup>1,2</sup> B,<sup>2,3</sup> and hygromycin B.<sup>4</sup> The syntheses of destomic acid and its 4-epimer, starting from D-galactose and D-glucose, respectively, have recently been published.<sup>5</sup>

During our studies on applications of chiral *N*-protected  $\alpha$ -amino aldehydes in organic synthesis<sup>6</sup> we found that they are very convenient and versatile heterodienophiles. High-pressure<sup>7—9</sup> or Lewis acid-mediated<sup>10—12</sup> (4 + 2)cycloaddition of 1,3-dienes to *N*-protected  $\alpha$ -amino aldehydes offers easy access to the respective optically pure adducts which were readily transformed into several natural products.<sup>6</sup>

Now we report on a new application of this methodology to the total synthesis of destomic acid (1). Restrosynthetic analysis shown in Scheme 1 suggested that 1-benzoyloxy-2-tbutyldimethylsilyloxy-4-ethoxybuta-1,3-diene (3)<sup>13</sup> and *N*-benzyloxycarbonyl-*O*-t-butyldiphenylsilyl-L-serinal (4)<sup>†</sup> could serve as starting materials.

The cyclocondensation reaction of diene (3) with aldehyde (4) in the presence of ZnBr<sub>2</sub>, followed by treatment with trifluoroacetic acid, led to a mixture of four possible diastereoisomeric adducts (5); as can be expected 10,12,14 (4*S*,5*R*)-diastereoisomer (5a) was formed as a major product‡ (Scheme 2).

The Luche-type reduction<sup>15</sup> of the chromatographically pure adduct (**5a**),§ followed by basic debenzoylation, afforded diol (**6**) which was subjected to the *cis*-hydroxylation reaction<sup>16</sup> to yield tetraol (**7**). Compound (**7**) can be easily transformed into destomic acid (**1**), but for direct comparison we converted it into the previously described<sup>5</sup> compound (**8**) and its 7-O-acetyl derivative. The latter compound has, after chromatographic purification,  $[\alpha]_D^{25} - 43.3^\circ$  (*c* 0.9, chloroform); lit.<sup>5</sup>  $[\alpha]_D^{RT} - 47.8^\circ$  (*c* 2.0, chloroform). The <sup>1</sup>H n.m.r. and i.r. spectra of compound (**8**) and of its 7-O-acetyl

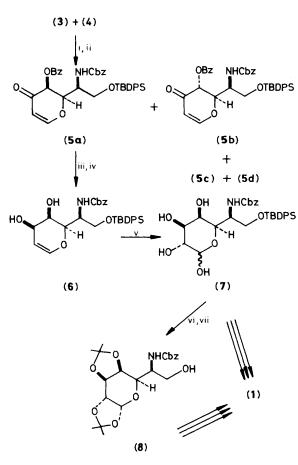
CO2H OH 0Bz NHCbz HC HO - H ЭН TBDMSO HO HO н ١H OH н HO ÓΕΙ H,N - H ÓН OH (2) (3) (1)NHCbz OTBDPS (4) Scheme 1

§ Satisfactory analyses and spectral data were obtained for all new compounds.



<sup>&</sup>lt;sup>+</sup> Compound (4) was prepared in the following reaction sequence starting from  $\iota$ -serine: i, SOCl<sub>2</sub>, MeOH; ii, CbzCl, NaHCO<sub>3</sub>, AcOEt; iii, TBDPSCl, imidazole, DMF; iv, DIBAL, Et<sub>2</sub>O, -78 °C.

<sup>&</sup>lt;sup>‡</sup> The diastereoisomeric proportion was (5a):(5b):(5c):(5d) = 87:8:4:1. The direction of asymmetric induction on the C-5 carbon atom can be rationalized on the ground of  $\alpha$ -chelation-controlled cycloaddition.



Scheme 2. Reagents and conditions: i,  $ZnBr_2$ , THF, rt; ii, TFA,  $CH_2Cl_2$ , rt; iii,  $NaBH_4$ ,  $CeCl_3 \cdot 7H_2O$ , MeOH, -78 °C; iv,  $K_2CO_3$ , MeOH, rt; v,  $OsO_4$ , NMO, Bu'OH,  $H_2O$ , rt; vi, DMP, *p*-TsOH,  $Me_2CO$ , rt; vii,  $Bu_4NF$ , THF, rt.

derivative obtained by us were identical with those measured by Hashimoto *et al.*<sup>5</sup> Compound (8) can be transformed into destomic acid (1) according to the known procedure.<sup>5</sup>

The presented total synthesis of destomic acid proves to be a practical alternative to the known approach of Hashimoto *et al.*<sup>5</sup> Moreover, it exemplifies the usefulness of *N*-protected  $\alpha$ -amino aldehydes in the synthesis of natural products.

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## References

- 1 S. Kondo, E. Akita, and M. Koike, J. Antibiot., Ser. A., 1966, 19, 139.
- 2 S. Kondo, K. Iinuma, H. Naganawa, M. Shimura, and Y. Sekizawa, J. Antibiot., 1975, 28, 79.
- 3 M. Shimura, Y. Sekizawa, K. Iinuma, H. Naganawa, and S. Kondo, Agric. Biol. Chem., 1976, 40, 611.
- 4 N. Neuss, K. F. Koch, B. B. Molloy, W. Day, L. L. Hickstep, D. E. Dorman, and J. D. Roberts, *Helv. Chim. Acta*, 1970, **53**, 2314.
- 5 H. Hashimoto, K. Asano, F. Fujii, and J. Yoshimura, *Carbohydr. Res.*, 1982, **104**, 87.
- 6 J. Jurczak and A. Gołebiowski, Chem. Rev., in the press.
- 7 A. Gołebiowski, J. Izdebski, U. Jacobsson, and J. Jurczak, *Heterocycles*, 1986, 24, 1205.
- 8 A. Gołebiowski, U. Jacobsson, M. Chmielewski, and J. Jurczak, *Tetrahedron*, 1987, **43**, 599.
- 9 A. Gołebiowski, U. Jacobsson, and J. Jurczak, *Tetrahedron*, 1987, 43, 3063.
- 10 S. Danishefsky, S. Kobayashi, and J. F. Kerwin, Jr., J. Org. Chem., 1982, 47, 1983.
- 11 P. Garner, Tetrahedron Lett., 1984, 25, 5855.
- 12 P. Garner and S. Ramakanth, J. Org. Chem., 1986, 51, 2609.
- 13 S. Danishefsky and C. J. Maring, J. Am. Chem. Soc., 1985, 107, 1269.
- 14 J. Jurczak, A. Gołebiowski, and J. Raczko, *Tetrahedron Lett.*, 1988, **29**, 5975.
- 15 J.-L. Luche and A. L. Gemal, J. Am. Chem. Soc., 1979, 101, 5842.
- 16 S. Danishefsky, E. Larson, and J. P. Springer, J. Am. Chem. Soc., 1985, 107, 1274.