



Synthesis and Characterization of Salamo-type Bisoximes Based on 3-Ethoxysalicylaldehyde and *Bis*(aminooxy)alkane

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A series of ethoxy-substituted Salamo-type bisoximes $H_2L^1-H_2L^4$ have been synthesized from 3-ethoxysalicylaldehyde and 1,3-*bis*(aminooxy)propane, 1,4-*bis*(aminooxy)butane, 1,5-*bis*(aminooxy)pentane or 1,6-*bis*(aminooxy)hexane in hot ethanol medium, respectively and characterized by elemental analyses, IR, UV-visible spectra and 1H NMR spectroscopy. Salamo-type compounds are conducive to coordinate experiments and construct supramolecular complexes.

Keywords: Salamo-type bisoxime, 3-Ethoxysalicylaldehyde, *bis*(Aminooxy)alkane.

INTRODUCTION

Salen (N,N'-disalicylideneethylenediamine) and its analogues are most versatile chelate ligands in inorganic and organometallic chemistry¹⁻⁴. Simultaneously, as bisoxime (-C=N-OR) ligands, they are usually synthesized by the condensation of aldehydes or ketones with hydroxylamines. Bisoxime compounds are a kind of good multidentate N_2O_2 -donating ligands with large π -conjugating system and often showing a diversity of structure. The chemical characteristics of the oxime compounds have been well known, since they have been widely used in insecticide, anti-virus, especially in medicine, analytical chemistry, metal corrosion and functional material fields⁵⁻¹⁰. There contain two hydroxylamine (-C=N-OR) and more than two hydroxyl (-OH) among each Salamo-type bisoxime ligand, which provide coordination site to combine metal atoms and have advantage to construct supramolecular complexes¹¹. So the study on Salamo-type bisoxime compounds has good prospects. Therefore, we have research on synthesis and characterization of those ethoxy-substituted Salamo-type bisoxime compounds, *e.g.*, 6,6'-diethoxy-2,2'-[(propylene-1,3-diyl-dioxy)-*bis*(nitrilomethylidyne)]diphenol (H_2L^1), 6,6'-diethoxy-2,2'-[(butylene-1,4-diyl-dioxy)-*bis*(nitrilomethylidyne)]diphenol (H_2L^2), 6,6'-diethoxy-2,2'-[(propane-1,5-diyl-dioxy)-*bis*(nitrilomethylidyne)]diphenol (H_2L^3) and 6,6'-diethoxy-2,2'-[(hexane-1,6-diyl-dioxy)-*bis*(nitrilomethylidyne)]diphenol (H_2L^4), which provided a reference for the further investigation.

EXPERIMENTAL

3-Ethoxysalicylaldehyde ($\geq 99\%$) was purchased and used without further purification. The others reagents used are the same as reported earlier¹¹. 1,3-*bis*(Aminooxy)propane, 1,4-*bis*(aminooxy)butane, 1,5-*bis*(aminooxy)pentane and 1,6-*bis*(aminooxy)hexane were synthesized according to an analogous method reported earlier¹²⁻¹⁴.

Preparation of 6,6'-diethoxy-2,2'-[(propylene-1,3-diyl-dioxy)-*bis*(nitrilomethylidyne)]diphenol (H_2L^1): To an ethanolic solution (2 mL) of 1,3-*bis*(aminooxy)propane (53.10 mg, 0.50 mmol) was added an ethanol solution (4 mL) of 3-ethoxy-salicylaldehyde (155.2 mg, 1 mmol). The reaction mixture was stirred at 331 K for 4 h, a white precipitation was obtained immediately. The formed precipitate was separated by filtration and washed successively with ethanol and ethanol/hexane (1:4), respectively. The product was dried under reduced pressure to obtain white compound H_2L^1 (Fig. 1).

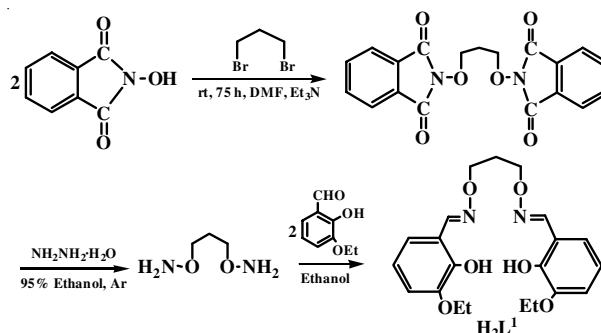


Fig. 1. Synthetic route to the Salamo-type bisoxime H_2L^1

Preparation of 6,6'-diethoxy-2,2'-[(butylene-1,4-diylldioxy)bis(nitrilomethylidyne)]diphenol (H_2L^2): To an ethanol solution (5 mL) of 1,4-*bis*(aminooxy)butane (122.8 mg, 1.02 mmol) was added an ethanolic solution (10 mL) of 3-ethoxysalicylaldehyde (362.2 mg, 2.18 mmol). After the solution had been stirred at 328 K for 6 h, a pale-yellow precipitation was obtained immediately. The formed precipitate was separated by filtration and washed successively with ethanol and ethanol/hexane (1:4), respectively. The product was dried under reduced pressure to obtain pale-yellow compound H_2L^2 (Fig. 2).

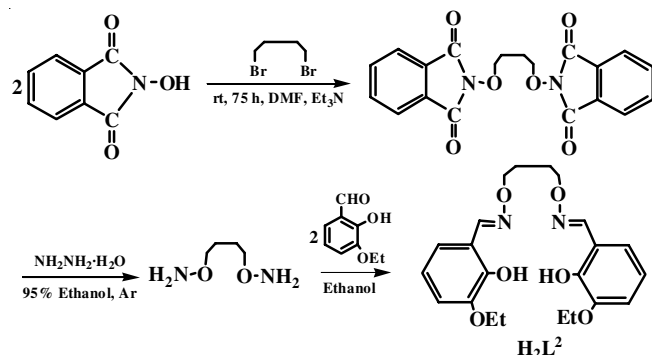


Fig. 2. Synthetic route to the Salamo-type bisoxime H_2L^2

Preparation of 6,6'-diethoxy-2,2'-[(propane-1,5-diylldioxy)bis(nitrilomethylidyne)]diphenol (H_2L^3): To an ethanolic solution (5 mL) of 1,5-*bis*(aminooxy)pentane (133.4 mg, 1 mmol) was added an ethanol solution (8 mL) of 3-ethoxysalicylaldehyde (359.2 mg, 2.16 mmol). After the solution had been stirred at 325 K for 3 h, a white precipitation was obtained immediately. The formed precipitate was separated by filtration and washed successively with ethanol and ethanol/hexane (1:4), respectively. The product was dried under reduced pressure to obtain white compound H_2L^3 (Fig. 3).

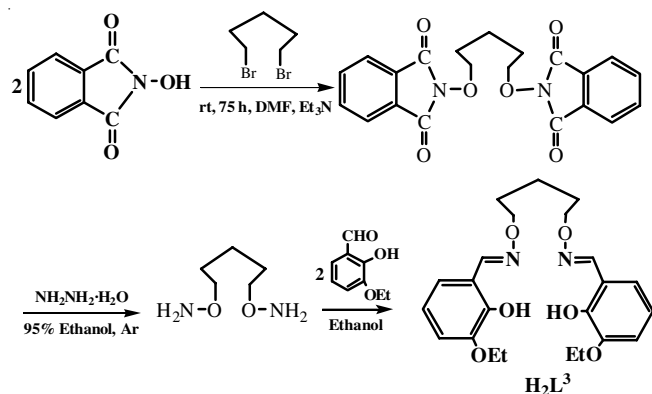


Fig. 3. Synthetic route to the Salamo-type bisoxime H_2L^3

Preparation of 6,6'-diethoxy-2,2'-[(hexane-1,6-diylldioxy)bis(nitrilomethylidyne)]diphenol (H_2L^4): To an ethanolic solution (5 mL) of 1,6-*bis*(aminooxy)hexane (151.1 mg, 1.02 mmol) was added an ethanolic solution (10 mL) of 3-ethoxysalicylaldehyde (347.8 mg, 2.09 mmol). After the solution had been stirred at 328 K for 6 h, a pale-yellow precipitation was obtained. The formed precipitate was separated by filtration and washed successively with ethanol and ethanol/hexane (1:4), respectively. The product was dried under reduced pressure to obtain pale-yellow compound H_2L^4 .

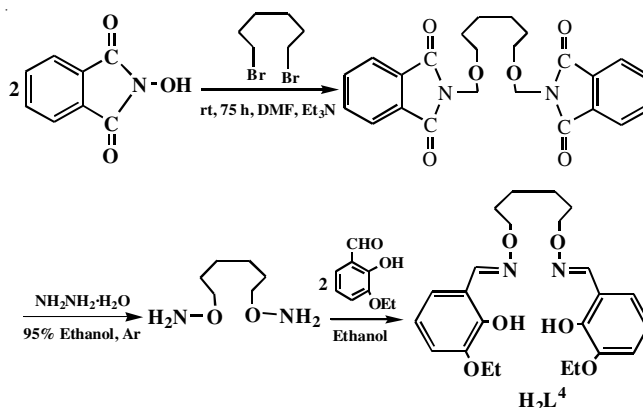


Fig. 4. Synthetic route to the Salamo-type bisoxime H_2L^4

hexane (1:4), respectively. The product was dried under reduced pressure to obtain pale-yellow compound H_2L^4 .

RESULTS AND DISCUSSION

Four ethoxy-substituted Salamo-type bisoximes H_2L^1 , H_2L^2 , H_2L^3 and H_2L^4 have been synthesized with good yields and the compositions are confirmed by elemental analyses, IR, UV-visible spectra and 1H NMR spectroscopy.

The colour, yields, melting points and elemental analytical results of the synthesized Salamo-type bisoximes H_2L^1 , H_2L^2 , H_2L^3 and H_2L^4 are summarized in Table-1.

The bisoxime compounds H_2L^1 , H_2L^2 , H_2L^3 and H_2L^4 are stable in air and soluble in majority of organic solvents that conclude to coordinated experiments, not soluble in hexane that they can be recrystallized from *n*-hexane. The elemental analytical data and the compositions of the Salamo-type bisoximes show that the elemental analytical data of Salamo-type bisoximes close to the theoretical value which demonstrate the accuracy of the results.

IR spectra: IR spectral data of the Salamo-type bisoximes H_2L^1 , H_2L^2 , H_2L^3 and H_2L^4 exhibit various bands from 4000 to 400 cm^{-1} . The IR spectral details of the Salamo-type bisoximes H_2L^1 , H_2L^2 , H_2L^3 and H_2L^4 are presented in Table-2.

The characteristic C=N stretching bands of the Salamo-type bisoximes H_2L^1 - H_2L^4 appear at 1605-1611 cm^{-1} , respectively¹⁵. And the Ar-O stretching bands occur at 1247, 1252, 1261 and 1256 cm^{-1} for the Salamo-type bisoximes H_2L^1 - H_2L^4 , respectively indicating that 3-ethoxysalicylaldehyde has been condensed with 1,3-*bis*(aminooxy)propane, 1,4-*bis*(aminooxy)butane, 1,5-*bis*(aminooxy)pentane and 1,6-*bis*(aminooxy)hexane, respectively and formed new Salamo-type bisoximes¹⁶. In the 1479-1446 cm^{-1} region, the observed bands were attributed to aromatic C=C vibrations. In addition, the O-H stretching bands of the Salamo-type bisoximes at 3600 cm^{-1} region disappear, the strong absorption bands of the Salamo-type bisoximes appear at 3435-3410 cm^{-1} region, which are the evidence for the existence of associating hydroxyl group in the Salamo-type bisoximes. IR spectral results of the Salamo-type bisoximes further confirmed the accuracy of the consequence.

UV-visible and 1H NMR spectra: The UV-visible spectra of the Salamo-type bisoximes H_2L^1 , H_2L^2 , H_2L^3 and H_2L^4 in 5×10^{-5} mol L^{-1} dichloromethane solution are given in Table-3. The Salamo-type bisoximes H_2L^1 , H_2L^2 , H_2L^3 and H_2L^4 exhibit

TABLE-1
COLOUR, YIELDS, MELTING POINTS, ELEMENTARY ANALYTICAL RESULTS
AND COMPOSITIONS OF THE BISOXIMES H₂L¹, H₂L², H₂L³ AND H₂L⁴

Compound	m.f. (m.w.)	Colour	m.p. (K)	Yield (%)	Elemental analysis (%) Found (Calcd.)		
					C	H	N
H ₂ L ¹	C ₂₁ H ₂₆ N ₂ O ₆ 402.44	White	462.5-463.5	76.5	62.61 (62.67)	6.40 (6.41)	7.01 (6.95)
H ₂ L ²	C ₂₂ H ₂₈ N ₂ O ₆ 416.47	Pale-yellow	413-414	85.8	63.47 (63.45)	6.75 (6.78)	6.66 (6.73)
H ₂ L ³	C ₂₃ H ₃₀ N ₂ O ₆ 430.49	White	355.5-356.5	67.7	64.20 (64.17)	7.01 (7.02)	6.48 (6.51)
H ₂ L ⁴	C ₂₄ H ₃₂ N ₂ O ₆ 444.52	Pale-yellow	354.5-355.5	57.2	64.97 (64.85)	7.05 (7.26)	6.12 (6.30)

TABLE-2
IR SPECTRAL DATA FOR THE BISOXIMES H₂L¹, H₂L², H₂L³ AND H₂L⁴ (cm⁻¹)

Compound	ν(O-H)	ν (CH ₂)	ν (C=N)	ν (Ar-O)	ν (C-C) _{benzene ring}
H ₂ L ¹	3410	2970, 2885	1605	1247	1446
H ₂ L ²	3427	2977, 2882	1607	1252	1475
H ₂ L ³	3441	2941, 2880	1611	1261	1481
H ₂ L ⁴	3435	2975, 2879	1608	1256	1479

TABLE-3
UV-VISIBLE SPECTRA AND ¹H NMR DATA FOR THE SYNTHESIZED BISOXIMES H₂L¹, H₂L², H₂L³ AND H₂L⁴

Compound	π-π* (nm)	¹ H NMR (400 MHz, DMSO-d ₆ , δ/ppm)
H ₂ L ¹	272, 310	2.18 (m, <i>J</i> = 6.5 Hz, 2H, CH ₂), 3.92 (s, 6H, CH ₃), 4.17 (m, <i>J</i> = 7.2 Hz, 4H, CH ₂ -O), 4.36 (m, <i>J</i> = 6.2, 1.4 Hz, 4H, CH ₂ -O), 6.78 (s, 2H, PhH), 6.85 (m, 2H, PhH), 6.92 (dd, <i>J</i> = 7.6, 2.2 Hz, 2H, PhH), 8.25 (s, 2H, N=CH), 9.89 (s, 2H, OH).
H ₂ L ²	274, 315	2.22 (m, <i>J</i> = 6.2 Hz, 4H, CH ₂), 3.90 (s, 6H, CH ₃), 4.16 (m, <i>J</i> = 7.0 Hz, 4H, CH ₂ -O), 4.37 (m, <i>J</i> = 6.0, 1.4 Hz, 4H, CH ₂ -O), 6.78 (s, 2H, PhH), 6.85 (m, 2H, PhH), 6.92 (dd, <i>J</i> = 7.2, 2.3 Hz, 2H, PhH), 8.23 (s, 2H, N=CH), 9.87 (s, 2H, OH).
H ₂ L ³	272, 312	2.19 (m, <i>J</i> = 6.2 Hz, 6H, CH ₂), 3.91 (s, 6H, CH ₃), 4.18 (m, <i>J</i> = 7.4 Hz, 4H, CH ₂ -O), 4.35 (m, <i>J</i> = 6.3, 1.4 Hz, 4H, CH ₂ -O), 6.78 (s, 2H, PhH), 6.85 (m, 2H, PhH), 6.92 (dd, <i>J</i> = 7.4, 2.2 Hz, 2H, PhH), 8.26 (s, 2H, N=CH), 9.88 (s, 2H, OH).
H ₂ L ⁴	273, 321	2.20 (m, <i>J</i> = 6.0 Hz, 8H, CH ₂), 3.93 (s, 6H, CH ₃), 4.18 (m, <i>J</i> = 7.0 Hz, 4H, CH ₂ -O), 4.35 (m, <i>J</i> = 6.4, 1.4 Hz, 4H, CH ₂ -O), 6.78 (s, 2H, PhH), 6.85 (m, 2H, PhH), 6.92 (dd, <i>J</i> = 7.5, 2.0 Hz, 2H, PhH), 8.21 (s, 2H, N=CH), 9.86 (s, 2H, OH).

two intense peaks at around 272 and 310 nm. The former absorption peaks at about 270 nm can be assigned to the π-π* transition of the naphthalene rings, while the latter can be attributed to the intra-ligand π-π* transition of the C=N bonds¹⁷. It is of note that there was no absorption around 400 nm, which is seen in the corresponding Salen derivatives. The absorption is ascribed to the quinoid form of H₂salen¹⁷.

The ¹H NMR spectra of the Salamo-type bisoximes H₂L¹, H₂L², H₂L³ and H₂L⁴ in DMSO-*d*₆ are shown in Table-3. The ¹H NMR spectra showed a singlet at about 8.21-8.26 ppm indicating the existence of oxime bonds^{17,18}.

Conclusion

Four Salamo-type bisoximes H₂L¹-H₂L⁴ have been designed and successfully synthesized by the reaction of 2 equivalents of 3-ethoxysalicylaldehyde with 1,3-bis(aminooxy)propane, 1,4-bis(aminooxy)butane, 1,5-bis(aminooxy)pentane or 1,6-bis(aminooxy)hexane under comfortable conditions, respectively.

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REFERENCES

1. T. Katsuki, *Coord. Chem. Rev.*, **140**, 189 (1995).
2. N. Hoshino, *Coord. Chem. Rev.*, **174**, 77 (1998).
3. J.-P. Costes, F. Dahan and A. Dupuis, *Inorg. Chem.*, **39**, 165 (2000).
4. H. Chen and J. Rhodes, *J. Mol. Med.*, **74**, 497 (1996).
5. S. Goyal and K. Lal, *J. Indian Chem. Soc.*, **66**, 477 (1989).
6. H.L. Amjid, I. Siddiqui and M.A. Chaudhary, *Chem. Pharm. Bull. (Tokyo)*, **7**, 1070 (2007).
7. H. Chen and J. Rhodes, *J. Mol. Med.*, **74**, 497 (1996).
8. S.T. Girousi, E.E. Golia, A.N. Voulgaropoulos and A.J. Maroulis, *Fresenius J. Anal. Chem.*, **358**, 667 (1997).
9. M.N. Desai, P.O. Chauhan and N. Shah, *Proceeding of 7th European Symposium on "Corrosion Inhibitors" Sez V. Ann. Univ. Ferrara Italy*, p. 1199 (1990).
10. S. Di Bella and I. Fragalà, *Synth. Met.*, **115**, 191 (2000).
11. W.K. Dong, Y.X. Sun, Y.P. Zhang, L. Li, X.-N. He and X.-L. Tang, *Inorg. Chim. Acta*, **362**, 117 (2009).
12. W.K. Dong, J.H. Feng and X.Q. Yang, *Synth. React. Inorg. Met.-Org. Nano-Met. Chem.*, **37**, 189 (2007).
13. W.K. Dong and J.G. Duan, *J. Coord. Chem.*, **61**, 781 (2008).
14. W.K. Dong, J.G. Duan and G.L. Liu, *Transition Met. Chem.*, **32**, 702 (2007).
15. S. Akine, T. Taniguchi, W. Dong, S. Masubuchi and T. Nabeshima, *J. Org. Chem.*, **70**, 1704 (2005).
16. J.A. Faniran, K.S. Patel and J.C. Bailar Jr., *J. Inorg. Nucl. Chem.*, **36**, 1547 (1974).
17. H.E. Smith, *Chem. Rev.*, **83**, 359 (1983).
18. S. Akine, T. Taniguchi and T. Nabeshima, *Chem. Lett.*, **30**, 682 (2001).