Synthesis and Properties of Novel *C*₃-Symmetric Coplanar Chromophores

Jia-Hong Chen, Shao-An Wang, Yi-Hung Liu, and Ken-Tsung Wong*

Department of Chemistry, National Taiwan University, Taipei 10617, Taiwan kenwong@ntu.edu.tw

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ABSTRACT



New synthetic pathways for novel C_3 -symmetric molecules featured with a thiophene-fused *six-five-five* carbon-bridged coplanar core structure have been established. The incorporation of thiophene as the constituent of a C_3 -symmetric core provides effective modulation of the physical properties and imparts extra flexibility for π -conjugated functionalization stemming from either the embedded thiophene or pendant aryl substitutions.

 C_3 -symmetric molecules intrigue organic chemistry scientists for their new synthetic pathways and optoelectronic applications.¹ A major goal for C_3 -symmetric material chemistry is to design new families of π -conjugated star-shaped molecules with novel core structures, and to investigate their structure-property relationships. Among the reported C_3 -symmetric polycyclic aromatics, truxene (Figure 1) is one of the most promising and versatile building blocks for the synthesis of fascinating molecules with various applications such as organic field-effect transistors (OFETs), solar cells, and electroluminescent devices.² In addition to truxene, C_3 -symmetric coplanar analogues with bridging heteroatoms (N, O, S; Figure 1) have also been explored,³ making possible direct manipulations on the electronic structure (HOMO/LUMO levels) and consequent physical properties of the truxene analogues. For example, triazatruxene-based materials have been exploited as the precursor of a triaza-analogue of crushed fullerenes⁴ and have been utilized as liquid crystals,⁵ as two-photon absorption materials,⁶ and in OLED applications.⁷ To the best of our knowledge, the reported C_3 -symmetric coplanar structures are generally composed of *six-five-six* fused rings. To date, an analogous C_3 -symmetric coplanar structure featuring six-five-five fused rings has not been reported. In this communication, we report for the first time the synthetic routes and physical properties of a novel class of thiophene-embedded C_3 symmetric star-shaped coplanar molecules. The incorporation of electron-rich thiophene rings as the π -conjugated constituent of this new C3-symmetric coplanar structure drastically perturbs the electronic properties of

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the core, namely raising the HOMO level as compared to the pure hydrocarbon homologues.



Figure 1. Chemical structures of truxene and its heterohomologues.

Scheme 1 depicts our synthetic approaches of the target C_3 -symmetric chromophores. Based on our previously established synthesis of sp³-carbon bridge coplanar ring systems,⁸ a double addition of aryl organometallics onto an ortho ester group of biaryls followed by intramolecular Friedel–Crafts cyclization is generally involved.

Scheme 1. Synthesis of the Thiophene-Fused *C*₃-Symmetric Coplanar Molecules



The crucial triester intermediate 1 was furnished in 42% yield by the Negishi cross-coupling reaction of

1,3,5-tribromobenzene with 3-(ethoxycarbonyl)-2-thienyl zinc cholride. Before carrying out the nucleophilic addition on the ester groups of 1 with aryl lithium, a triisopropylsilyl (TIPS) group was introduced to block the reactive sites of the α -thienyl groups. Thus, deprotonation of 1 with LDA followed by quenching with TIPS-Cl afforded compound 2 in good yield (76%). Subsequent addition of phenyllithium, followed by intramolecular Friedel-Crafts cyclization, gave the desired product **3a** in 60% yield. Finally, the hexaphenyl substituted coplanar molecule 4 was isolated with an excellent yield of 90% after protodesilylation with tetrabutylammonium fluoride (TBAF). Instead of using TIPS as the reactive site blocking group, bromination of triester 1 with NBS gave the versatile tribromide 5 in high yield. The regioselective bromination indicates that π conjugated derivatives stemming from the new thiophenefused coplanar core structure could be feasibly achieved. Along this line, a Pd-catalyzed Suzuki coupling reaction of intermediate 5 with 4-tert-butylphenyl boronic acid gave triester 6 with an isolated yield of 82%. The addition of an excess amount of phenyllithium onto the ester groups of 6 gave the corresponding triaryl alcohol. Without further purification, the crude triaryl alcohol was subjected to acid-mediated triple cyclization to afford the coplanar C_3 -symmetric product 7 with an isolated yield of 63%. Scheme 2 depicts an alternative pathway for the synthesis of thiophene-fused C_3 -symmetric coplanar molecules.

Scheme 2. Alternative Synthetic Route of the Thiophene-Fused C_3 -Symmetric Coplanar Molecules



For this approach, the lithiated core was utilized to add onto the diarylketones for generating the tertiary alcohols used for the proton-promoted intramolecular cyclization. The synthesis started from the bromination of 1,3,5trithienylbenzene 8 to yield hexabromo intermediate 9 with a yield of 78%. Treatment of compound 9 with *n*-BuLi at -78 °C regioselectively gave the trilithiated intermediate,

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which was quenched with TIPS-Cl and 1-iodohexane, giving TIPS and hexyl-substituted tribromo compounds 10a and 10b, respectively. A subsequent Li/Br exchange reaction of tribromo compound 10 with n-BuLi gave the reactive lithiated intermediate, which was then quenched with various diarylketones to afford the corresponding tertiary alcohols. Without further purifications, these alcohols underwent acid-mediated intramolecular cvclization to give the coplanar cores 3a - e in moderate yields. After protodesilylation of 3a with TBAF followed by regioselective NBS bromination, the resulting tribromine intermediate can be directly exploited for a Suzuki coupling reaction with 4-tert-butylphenyl boronic acid to give molecule 7 in 68% yield (three steps). This synthetic strategy provides extra flexibility for the functionalization of the aryl substitutions, as we can envision that further decoration on the peripheral bromo-substituted fluorene moieties of 3e would give rise to new structural features arranging perpendicularly relative to the coplanar C_3 symmetric core.

The molecular structure of thiophene-fused C₃-symmetric molecules **3a** and **3d** can be verified by X-ray crystallography (Figure 2). The averaged dihedral angles of the pivotal phenylene ring and peripheral thiophene rings of molecules **3a** and **3d** were calculated to be 6.67° and 6.15° , respectively, suggesting slight deviations from a perfect coplanar plane due to the manner of fused *six-five-five* rings. The aryl substitutions on the saturated carbon bridge position on the top and bottom faces of the molecular plane imparts a low tendency for interchromophore interactions. In addition, the rigid and coplanar spirofluorene substituents are orthogonal relative to the molecular plane of **3d**, giving an interesting tripod-like molecule.



Figure 2. (a) Top view of molecular structure of 3a. (b) C_3 -symmetric core structure of 3a. (c) The tripod-like 3d with orthogonal spirofluorene substituents; the alkyl side chains are omitted for clarity.

The introduction of aryl substitutions on the top and bottom faces of the thiophene-fused C_3 -symmetric coplanar core makes the resulting molecules rather bulky and rigid. Differential scanning calorimetry (DSC) analysis indicated that these new thiophene-fused C_3 -symmetry materials exhibit no obvious glass transition temperatures (T_g) except for **3d** (120 °C). Thermogravimetric analysis (TGA) revealed the high heat tolerance of these new compounds with decomposition temperatures (T_d) (5% weight loss) ranging from 365 to 433 °C. The data are summarized in Table 1.

The absorption (UV–vis) and photoluminescent (PL) sepecta of molecules 4 and 7 are compared to those of pure hydrcarbon parent counterpart T1 (Figure 3). The UV-vis and PL spectra of the thiophene-fused coplanar molecule 4 exhibited substantial red shifts relative to those of truxene derivative T1. This result clearly indicates the effect of reducing the optical gap when thiophene was introduced as the constitutent of the C_3 -symmetric coplanar core. An evident Stokes shift (6832 cm^{-1}) of **4** was observed which may be a result of the first two excitations being symmetryforbidden. If so, the main absorption peak could result from the electronic transition to higher states, followed by internal conversion to the S1 state where emission originates.⁹ The PL quantum yields (QYs) of these C_3 symmetric molecules are comparable as compared to that of the parent truxene T1 (Table 1). In addition, the extension of the π -conjugation from the thiophene-fused core can pronouncedly alter the photophysical behavior as evidenced by the significant red shifts in UV-vis and PL spectra and improved PL efficiency of compound 7.



Figure 3. UV–vis absorption spectra (solid symbol) and photoluminescence spectra (open symbol) of T1, 4, and 7.

Cyclic voltammetry (CV) was used to probe the electrochemical behavior of thiophene-fused C_3 -symmetry coplanar molecules and parent molecule **T1** (Figure 4). The electrochemical data are summarized in Table 1. The oxidation potentials of the thiophene-embedded molecules are significantly lower than that of truxene **T1**. For example, the oxidation onset of **4** is 1.28 V, which is evidently lower than that of **T1** (1.44 V). This result clearly indicates the considerable HOMO-raising

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Table 1. Properties of Truxene T1 and the Thiophene-Embedded Truxene Analogues								
compd	$abs^{a}\lambda_{max} \left(\log \epsilon\right) [nm]$	$\mathrm{PL}^b \lambda_{\max} \left[\mathrm{nm} \right]$	Q.Y. c [%]	$T_{\rm g}{}^d$ [°C]	$T_{d}^{f}[^{\circ}C]$	$E_{1/2 \text{ oxd}}^g$ [V]	HOMO^i [eV]	$LUMO^i$ [eV]
T 1	292, 308 (4.85)	377	12	$\mathrm{n.d.}^{e}$	409	1.52	-5.85	-2.10
3a	326 (4.79), 342 (4.80)	410	12	n.d.	339	1.25	-5.58	-2.11
3b	324 (5.00), 341 (4.99)	408	9	n.d.	365	1.22	-5.55	-2.06
3c	324 (4.86), 339 (4.86)	411	11	n.d.	365	1.16	-5.49	-2.00
3d	321 (4.92), 337 (4.85)	409	6	120	380	1.14	-5.47	-1.93
4	316 (4.48), 330 (4.44)	403	10	n.d.	373	1.36^{h}	-5.69	-2.07
7	367~(4.95)	423	30	n.d.	433	1.15	-5.48	-2.39

^{*a*} In CHCl₃. ^{*b*} In CHCl₃, upon excitation at the absorption maximum. ^{*c*} In CH₂Cl₂, measured by calibrated integrating sphere system (Hamamatsu C9920) ^{*d*} Analyzed using DSC. ^{*e*} n.d. = not detected. ^{*f*} Analyzed using TGA (5% weight loss). ^{*g*} In CH₂Cl₂ containing 0.1 M *n*Bu₄NPF₆ as a supporting electrolyte; potentials were recorded vs Ag/AgCl. ^{*h*} E_p^a , due to irreversible CV. ^{*i*} HOMO and LUMO were estimated from $E_{1/2 \text{ oxd}}$ and absorption onset.

effect upon the implantation of electron-rich thiophenes as the constitutional moieties of the C_3 -symmetric coplanar core. It should be noted that the reduction peak observed at 0.89 V of **4** is due to the electrochemically dimerized species as the electroactive sites are unprotected (Figure S-2, Supporting Information). The ex-



Figure 4. Cyclic voltammograms of compounds T1, 4, and 7.

tended π -conjugation is beneficial for giving a higher HOMO level and quasi-reversible oxidation and reduction potentials (Figure S-3).

In summary, for the first time, inventive synthetic pathways for novel C_3 -symmetric molecules featuring

a thiophene-fused six-five-five carbon-bridged coplanar structure have been successfully established. The versatile synthetic pathways provide the flexibility for further π conjugated functionalization stemming from either the constituted thiophene ring or the pendant aryl substitutions. The molecular core structures were confirmed to be nearly coplanar by X-ray crystallography. The incorporation of thiophene as the constituent of a C_3 -symmetric core provides effective modulation of the physical properties, particularly the HOMO-raisng effect, which was evidenced by the lower oxidation potential in CV and red-shifted absorption and emission in photophysics. With great thermal stability as a property, we could anticipate the great opportunity of these novel C_3 -symmetric chromophores to serve as potential materials for optoelectronic devices.

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Supporting Information Available. Detailed experimental procedures; spectroscopic characterization; the crystal data and cif files of compounds **1**, **3a**, and **3d**; the X-ray crystal structure of compound **1**; cyclic voltammograms of compounds **T1**, **4**, and **7**; UV–visible absorption and photoluminescence spectra of **3a**–d; ¹H, ¹³C NMR spectra of new compounds. This material is available free of charge via the Internet at http:// pubs.acs.org.