

Total Synthesis of (+)-Polyzonimine

Tsutomu Sugahara, Yuuki Komatsu, and Seiichi Takano*

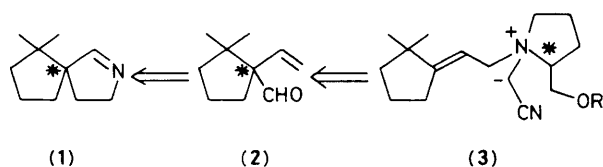
Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980, Japan

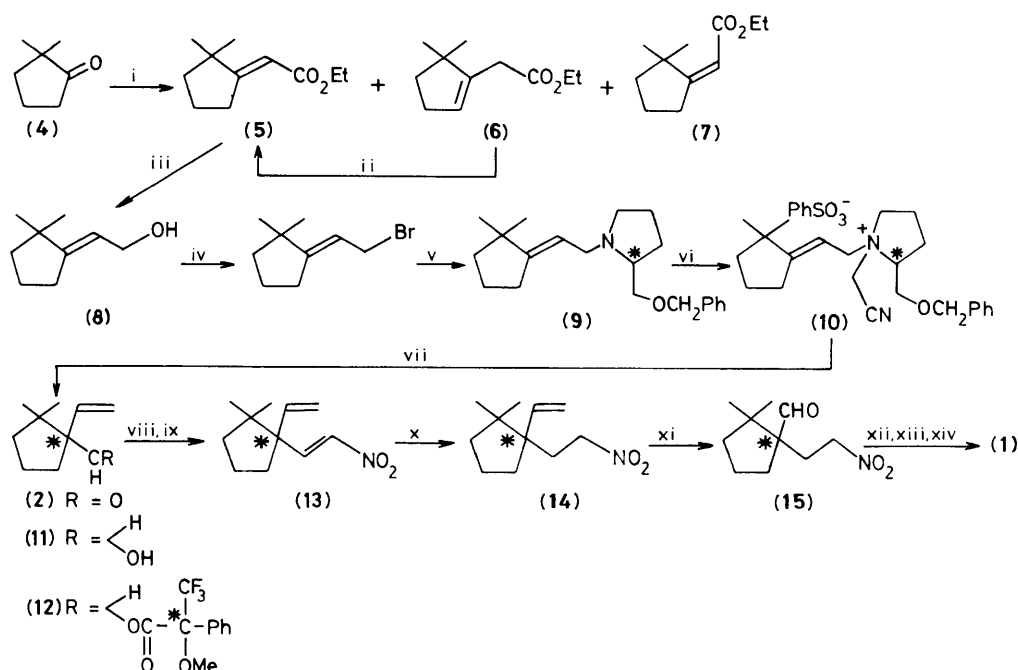
(+)-Polyzonimine (**1**), a terpenoid insect repellent produced by a millipede, was synthesized by a reaction sequence which utilizes the asymmetric [2,3] sigmatropic rearrangement of the ammonium ylide formed from the salt (**10**) to generate the chiral intermediate (**2**).

In 1975 Smolanoff *et al.*¹ reported the isolation of a novel nitrogenous monoterpene, the insect repellent (+)-6,6-dimethyl-2-azaspiro[4.4]non-1-ene [(+)-polyzonimine] (**1**)

from the millipede *Polyzonium rosalbum*. Assignment of structure (**1**) for this natural product was based on the spectral analysis and the synthesis of (±)-polyzonimine.

We report here the total synthesis of (+)-polyzonimine (**1**) in which the key synthetic transformation leading to (**1**) is the asymmetric [2,3] sigmatropic rearrangement² of the ammonium ylide (**3**) to give the chiral intermediate (**2**). The first step in the synthesis (Scheme 1) was the Horner–Emmons reaction of the ketone (**4**)³ with triethyl phosphonoacetate and NaH in refluxing dimethoxyethane (DME) to give a mixture of (**5**) and (**6**) (ratio 3:2, 85%) together with the Z-isomer (**7**)





Scheme 1. Reagents and conditions: i, NaH (1.6 equiv.), (EtO)₂POCH₂CO₂Et (1.8 equiv.), DME, reflux, 15 h; ii, PPA (1 equiv.) on silica gel, CH₂Cl₂ (100%); iii, LiAlH₄ (1.2 equiv.), AlCl₃ (0.4 equiv.), Et₂O, 0°C, 1 h (94%); iv, PBr₃, pyridine, Et₂O, 0–25°C, 12 h; v, L-benzyloxypyrrolinol (1.1 equiv.), K₂CO₃, DMSO, 25°C, 15 h [60% from (8)]; vi, PhSO₃CH₂CN (3 equiv.), acetonitrile, 25–60°C, 24 h; vii, (a) Bu^tOK, THF, DMSO, –78°C, 24 h, (b) CuSO₄·5H₂O, EtOH, 10 min [61% from (9)]; viii, nitromethane, KOH, MeOH, 25°C, 1 h; ix, methanesulphonyl chloride (6 equiv.), triethylamine, 25°C, 1.5 h [72% from (11)]; x, NaBH₄, MeOH, 0–25°C (87%); xi, (a) O₃, CH₂Cl₂, PrOH, –78°C, (b) Me₂S, –78°C (88.2%); xii, ethylene glycol, triethyl orthoformate, *p*-MeC₆H₄SO₂OH, 25°C, 0.5 h; xiii, H₂, PtO₂ (cat.), EtOH, 25°C, 5 h; xiv, 10% HCl, THF, 25°C, 10 h [50% from (15)].

(9%).[†] The mixture of (5) and (6) was treated with polyphosphoric acid (PPA) on silica gel in refluxing CH₂Cl₂ for 24 h to give the desired *exo*-ester (5) exclusively by isomerization of (6).

Reduction of the *exo*-ester (5) with LiAlH₄ and AlCl₃ in Et₂O at 0°C provided (8). Bromination of the allylic alcohol (8) with PBr₃ and pyridine in Et₂O followed by amination with L-benzyloxypyrrolinol and K₂CO₃ in dimethyl sulphoxide (DMSO) afforded, after alumina column chromatography, (9) in 60% overall yield. The pyrrolidine derivative (9) was converted into the quaternary salt (10) with cyanomethyl benzenesulphonate⁴ in acetonitrile. Treatment of (10) with KOBu[†] in tetrahydrofuran (THF)–DMSO at –78°C followed by hydrolysis with CuSO₄·5H₂O in refluxing EtOH for 10 min afforded the optically active olefin-aldehyde (2) { $[\alpha]_D -7.06^\circ$ (CHCl₃)} in 61% overall yield.

[†] All new compounds were fully characterized by spectroscopic methods (¹H n.m.r., i.r., mass). Representative spectral properties of the key compounds are as follows: compound (5), i.r. ν_{\max} (neat) 1720, 1650 cm^{–1}; ¹H n.m.r. δ (CDCl₃) 1.1 (s, 6H, CH₃ × 2), 1.29 (t, *J* 7.6 Hz, 3H, –CO₂CH₂CH₃), 2.97 (tt, *J* 2.6 and 7.2 Hz, 2H, ring CH₂), 4.20 (q, *J* 7.6 Hz, 2H, –CO₂CH₂CH₃), 5.72 (t, *J* 2.6 Hz, 1H, olefinic H); *m/z* 182 (*M*⁺), 167, 137, 109. Compound (9), ¹H n.m.r. δ (CDCl₃) 1.00 (s, 6H, CH₃ × 2), 4.49 (s, 2H, –OCH₂Ar), 5.30 (tt, *J* 2.6 and 6.8 Hz, 1H, olefinic H), 7.26 (s, 5H, ArH); *m/z* 313.2402 (C₂₁H₃₁NO requires 313.2404). Compound (2), i.r. ν_{\max} (neat) 1708, 1630 cm^{–1}; ¹H n.m.r. δ (CDCl₃) 0.97 (s, 3H, CH₃), 1.07 (s, 3H, CH₃), 5.01 (dd, *J* 1.6 and 17.4 Hz, 1H, olefinic H), 5.25 (dd, *J* 1.6 and 11 Hz, 1H, olefinic H), 6.10 (dd, *J* 11 and 17.4 Hz, 1H, olefinic H), 9.65 (s, 1H, CHO); *m/z* 152.1208 (C₁₀H₁₆O requires 152.1200). Compound (15), i.r. ν_{\max} (neat) 1705, 1550 cm^{–1}; ¹H n.m.r. δ (CDCl₃) 1.0 (br. s, 6H, CH₃ × 2), 4.30 (m, 2H, –CH₂NO₂), 9.62 (s, 1H, CHO); *m/z* (field desorption) 199 (*M*⁺). Compound (1), i.r. ν_{\max} (neat) 1620 cm^{–1}; ¹H n.m.r. δ (CDCl₃) 0.88 (s, 3H, CH₃), 0.91 (s, 3H, CH₃), 1.70 (m, 8H), 3.79 (br. t, *J* 6.5 Hz, 2H, –CH₂N=), 7.37 (br. s, 1H, –CH=N–); *m/z* 151.1333 (C₁₀H₁₇N requires 151.1360).

The optical purity of (2) was determined by applying the methoxy(trifluoromethyl)phenylacetyl (MTPA) method.⁵ Thus, reduction of (2) with NaBH₄ in MeOH followed by treatment with (+)-MTPA chloride in CCl₄, pyridine, and 4-*N,N*-dimethylaminopyridine (4-DMAP) gave the corresponding (+)-MTPA ester (12). The optical purity of (12) was estimated to be 68% enantiomeric excess from ¹H and ¹⁹F n.m.r. spectra (300 MHz).

Condensation of (2) with nitromethane and KOH in MeOH followed by treatment with methanesulphonyl chloride and triethylamine in Et₂O afforded (13) in 72% overall yield. Reduction of the α,β -unsaturated nitro compound (13) with NaBH₄ in MeOH gave (14). Ozonolysis of the olefinic function of (14) by O₃ in CH₂Cl₂–PrOH at –78°C followed by treatment with Me₂S at –78 to 25°C provided (15). Protection of the formyl group of (15) as the acetal (ethylene glycol and triethyl orthoformate) followed by reduction of the nitro group with H₂ and PtO₂ in EtOH gave the corresponding amino acetal which was immediately treated with 10% HCl in THF at 25°C to give (1) by means of preparative g.c.

The synthetic polyzonimine (1) exhibited i.r. and n.m.r. spectra identical with those of the reported natural product.¹ The optical rotation observed for synthetic polyzonimine (1) was +1.95° (CHCl₃) {lit.,¹ $[\alpha]_D +3.26^\circ$ (CHCl₃)}.

Received, 17th November 1983; Com. 1506

References

- J. Smolanoff, A. F. Kluge, J. Meinwald, A. McPhail, R. W. Miller, K. Hicks, and T. Eisner, *Science*, 1975, **188**, 734.
- K. Hiroi and K. Nakazawa, *Chem. Lett.*, 1980, 1077.
- M. Larcheveque, A. Debal, and Th. Cureigny, *J. Organomet. Chem.*, 1975, **87**, 25.
- E. B. Sanders, H. V. Secor, and J. I. Seeman, *J. Org. Chem.*, 1978, **43**, 324, and references cited therein.
- J. A. Dale, D. L. Dull, and H. S. Mosher, *J. Org. Chem.*, 1969, **34**, 2543.