

# Doebner-type pyrazolopyridine carboxylic acids in an Ugi four-component reaction

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# Abstract

Substituted 1*H*-pyrazolo[3,4-*b*]pyridine-4- and 1*H*-pyrazolo[3,4-*b*]pyridine-6-carboxamides have been synthetized through a Doebner–Ugi multicomponent reaction sequence in a convergent and versatile manner using diversity generation strategies: combination of two multicomponent reactions and conditions-based divergence strategy. The target products contain as pharmacophores pyrazolopyridine and peptidomimetic moieties with four points of diversity introduced from readily available starting materials including scaffold diversity. A small focused compound library of 23 Ugi products was created and screened for antibacterial activity.

# Introduction

Modern medicinal chemistry is faced with the task of quick and effective screening a variety of organic molecules in order to identify new active pharmaceutical ingredients among them. Therefore, in turn, organic chemistry has to solve an equally important task of the rapid generating focused libraries of druglike compounds characterized by several important features, e.g., molecular complexity and diversity at different levels, high variability and easy accessibility from relatively simple reagents. These challenges can be overcome by using multi-component reactions (MCRs) but also other strategies can be applied in addition to MCRs for generating diversity, e.g., build/couple/pair- (BCP), single reactant replacement- (SRR), modular reaction sequences- (MRS), conditions-based divergence- (CBD) and combination of multicomponent reactions (MCR<sup>2</sup>) strategies (for more details and examples see [1-3] and Scheme 1). A synergetic application of several diversity-oriented synthesis (DOS) instruments allows an effective decoration of the privileged scaffolds for creating collections of unique, highly potent bioactive compounds [4,5].

The pyrazolopyridine scaffold can be regarded as a privileged motif as it exhibits various biological actions: antiproliferative [6-9], antimicrobial [10,11], anxiolytic [12], analgesic [13], hypnotic [13], antiviral [13], anti-HIV [13] activities, etc. Soural et al. [14] explored different data and showed the relevance of compounds composed of two or more heterocyclic rings for drug discovery. The target products containing a heterocyclic core bound to a peptide-like chain also showed a

broad spectrum of biological activity: β-secretase (BACE1) inhibitory activity [15]; inducing apoptosis in colorectal cancer cells [16]; antimalarial activity against a chloroquine (CQ) nonresistant Plasmodium falciparum 3D7 strain [17]; antagonists of p53-Mdm2 interaction [18]; antiproliferative activity in the human solid tumor cell lines A549 (lung), HBL-100 (breast), HeLa (cervix), SW1573 (lung), T-47D (breast), and WiDr(colon) [19]; cyclophilin A inhibitory activity for the treatment of hepatitis C virus infections [20], etc.

Among the variety of heterocyclic acids used in Ugi-4CR [15-19,21-42] only a few of them in addition to bearing a simple pharmacophore core (group I, Scheme 1) are also characterized by the complexity and diversity of the skeleton itself gained through multi-step transformations (group II) [18,19,34-36] or allow for generating additional diversity through post-cyclization reactions (group III) [18,35-42]. Meanwhile the complexity of the acid skeleton can be achieved by MCR. Several publications illustrated this principle: synthesis of heterocyclic acids [26,43] or enols [44] in a first multicomponent step, followed by subjecting them to a subsequent Ugi process, thus, applying the MCR<sup>2</sup> approach (group IV, Scheme 1).



Actually, there was no example for the combined application of Doebner and Ugi-type MCRs although the former condensation easily affords the azoloazine pharmacophore that is able to participate as an acid component in the latter reaction. It should be noted, that Cowen et al. [6] reported N-substituted-1*H*-pyrazolo[3,4-*b*]pyridine-4-carboxamides being SMYD2 inhibitors (an oncogenic methyltransferase that represses the functional activity of the tumor suppressor proteins p53 and RB); the similar structures can be obtained using the methodology of sequential Doebner- and Ugi-type MCRs.

In the present work we combined several diversity-oriented synthetic (DOS) approaches. First, by using CBD and MCR strategies in a Doebner-type reaction we synthesized pyrazolopyridine carboxylic acids which were subsequently applied in the Ugi reaction, thus, combining two multicomponent procedures.

#### Results and Discussion

As mentioned above, the diversification of the privileged scaffold using different DOS strategies allowed to significantly increase the diversity of the final products. In our study pyrazolo[3,4-*b*]pyridine scaffold was chosen as a privileged one and based on this, we combined two MCRs: the previously wellstudied three-component Doebner-type condensation of aminopyrazoles, aldehydes and pyruvic acid [45,46] with the isocyanide-based four-component Ugi reaction.

As we have shown before [47,48] the application of the CBD strategy to multicomponent Doebner-type condensations involving aminoazoles allowed the synthesis of several chemo-

types of structurally complex products from a limited number of relatively simple starting materials just by varying the reaction conditions (temperature, solvent–catalyst system, activation method, forced realization of one of the cascades of multicomponent treatment). We decided to use this strategy and to synthetize heteroaromatic carboxylic acids **4** and **7** starting from the same reactants but using a multicomponent and a sequential protocol. We chose these heterocyclic acids to be subjected to the further Ugi transformation based on their higher stability compared to other azoloazine carboxylic acids (e.g., tetrahydro-[49,50] and dihydroazoloazines that may undergo oxidation during the Ugi 4CR) and as they do not contain additional functional groups that may influence the subsequent Ugi reaction (e.g., hydroxy group in tetrahydro- [46,51], dihydro- [51] or aromatic derivatives [51]).

Two different reaction pathways were applied based on known synthetic procedures (Scheme 2): the three-component reaction between pyruvic acid (1), aromatic aldehydes **2a**,**b** and 5-amino-3-methylpyrazole (**3**) (HOAc,  $\Delta$ , 30 min) [45] and a two-component condensation of the preliminary synthetized 4-(4-methoxyphenyl)-2-oxobut-3-enoic acid (**5b**) [52,53] with 5-amino-3-methyl-*N*-phenylpyrazole (**6**) (HOAc,  $\Delta$ , 5 h). As a result, two different types of pyrazolo[3,4-*b*]pyridines containing the carboxylic group either at C4 position (compounds **4a**,**b**) or at the C6 position (compound **7b**) were synthetized (Scheme 2). We modified the earlier described methodology for the synthesis of pyrazolo[3,4-*b*]pyridine-6-carboxylic acid (**7b**) [45]: the solvent was changed from DMF to HOAc and the reaction time was increased from 30 min to 5 hours. Despite of



the longer reaction time the whole procedure became more efficient due to the easier work-up stage as well as due to avoiding the formation of impurities of the dihydropyrazolo[3,4*b*]pyridines.

Thus, starting from the same set of reactants two different types of heterocyclic acids **4** and **7** containing two diversity points were obtained. Afterwards compounds **4a,b** were introduced into the Ugi four-component reaction to create 3 additional points of diversity. However, due to the low solubility of the pyrazolopyridine acids **4a,b** under the literature standard reaction conditions for the Ugi transformation (stirring in methanol at rt and similar procedures) the reaction did not take place. Under these conditions, the pyrazolopyridine carboxylic acids **4a,b** did not dissolve and remained unreacted even after prolonged stirring and heating. Consequently, the solvent was changed to DMF that allowed us to isolate the Ugi products **11** after long stirring (48–72 h) at rt. It must be noted, that in many cases the pyrazolopyridine acids **4a,b** did not fully dissolve in DMF at rt that resulted in considerably decreased yields.

In an attempt to increase the yield of the products the reaction was repeated at different temperatures ranging from rt to 80 °C and it was found that heating at 70 °C afforded the best results. At this temperature not only the yields increased but also the reaction time could be reduced to 48 hours. When applying a solvent mixture of DMF and MeOH the yields further increased, with the best results obtained using a ratio of 1:2. We presume that methanol provides the optimal acidity to the reaction medium needed for successful protonation of the intermediate azomethine, formed between the aromatic aldehyde **8** and aniline **9**, to the corresponding iminium cation and its further transformation involving carboxylic acid **4** and isocyanide **10**.

As a result, we developed an efficient procedure for the synthesis of compounds **11a–q** through reaction of aromatic aldehydes **8a–d**, amines **9a–f**, *tert*-butylisocyanide (**10**) and heteroaromatic carboxylic acids **4a,b** in a 2:1 mixture of methanol and DMF at 70 °C. Following this procedure, a small library of seventeen Ugi products was obtained (Table 1).

Next, we applied pyrazolo[3,4-*b*]pyridine-6-carboxylic acid 7**b** with another positional location of the substituents in comparison with compounds 4 in the Ugi reaction with the same reagents 8, 9 and 10 using the optimized procedure (Table 1). This expanded the library of Ugi products by adding compounds 12a-f.

The purity and structures of the obtained heterocyclic products were established by means of NMR spectroscopy, mass spectrometry, and elemental analysis. The final assignment of the structures **11** and **12** was made by X-ray analysis for the structure **11n** (Figure 1).



phenyl)-2-oxoethyl)-6-(4-methoxyphenyl)-3-methyl-*N-p*-tolyl-1*H*-pyrazolo[3,4-*b*]pyridine-4-carboxamide (**11n**) according to X-ray diffraction data. Non-hydrogen atoms are presented as thermal ellipsoids with 50% probability.

#### Antibacterial activity

It is worth mentioning that the modification of the pyrazolo[3,4*b*]pyridine scaffold through Ugi reaction allowed not only to introduce three additional diversity points but also to increase significantly the solubility of the products **11** and **12** compared to the starting acids **4** and **7**. Compounds **11** are soluble in MeOH, EtOH, iPrOH, acetone, EtOAc, CH<sub>3</sub>CN, DCM, CHCl<sub>3</sub> and compounds **12** are soluble in acetone, CH<sub>3</sub>CN, DCM, CHCl<sub>3</sub> and partially soluble when heated in EtOAc, MeOH, EtOH, showing the advantages of this protocol for investigating the activity of pyrazolo[3,4-*b*]pyridine moiety in biological experiments. Particularly, the evaluation of the antibacterial activity of the small library of new compounds **11** and **12** was carried out.

We next screened some selected compounds for their antibacterial activity (Table 2, Supporting Information File 1) against the reference bacterial strains *Bacillus subtilis* (strain 1211), *Staphylococcus aureus* (strain 2231) (gram-positive) and *Escherichia coli* (strain 1257), *Pseudomonas aeruginosa* (strain 1111) (gram-negative).

Generally, the compounds were found to be less active than nitroxoline being the reference substance. The results obtained



	Starting materials						Products	
Entry	Acid	R <sup>1</sup>	8	R <sup>2</sup>	9	R <sup>3</sup>	11,12	yield, %
1	4a	Н	8a	Н	9a	Н	11a	39
2	4a	Н	8a	Н	9b	4-CH <sub>3</sub>	11b	40
3	4a	Н	8a	Н	9c	4-Br	11c	30
4	4a	Н	8a	Н	9d	2-CH <sub>3</sub> O	11d	28
5	4a	Н	8a	Н	9e	3-CH <sub>3</sub> O	11e	30
6	4a	Н	8a	Н	9f	4-CH <sub>3</sub> O	11f	42
7	4b	CH <sub>3</sub> O	8a	н	9a	Н	11g	43
8	4b	CH <sub>3</sub> O	8a	н	9b	4-CH <sub>3</sub>	11h	53
9	4b	CH <sub>3</sub> O	8a	Н	9c	4-Br	11i	37
10	4b	CH <sub>3</sub> O	8a	н	9d	2-CH <sub>3</sub> O	11j	37
11	4b	CH <sub>3</sub> O	8a	Н	9e	3-CH <sub>3</sub> O	11k	35
12	4b	CH <sub>3</sub> O	8a	н	9f	4-CH <sub>3</sub> O	111	42
13	4b	CH <sub>3</sub> O	8b	Cl	9a	Н	11m	44
14	4b	CH <sub>3</sub> O	8b	Cl	9b	4-CH <sub>3</sub>	11n	49
15	4b	CH <sub>3</sub> O	8b	Cl	9c	4-Br	11o	34
16	4b	CH <sub>3</sub> O	8b	Cl	9f	4-CH <sub>3</sub> O	11p	37
17	4b	CH <sub>3</sub> O	8c	NO <sub>2</sub>	9a	Н	11q	20
18	7b	CH <sub>3</sub> O	8a	Н	9a	Н	12a	50
19	7b	CH₃O	8a	н	9b	4-CH <sub>3</sub>	12b	51
20	7b	CH₃O	8b	CI	9a	Н	12c	34
21	7b	CH₃O	8b	CI	9b	4-CH <sub>3</sub>	12d	36
22	7b	CH₃O	8d	CH <sub>3</sub> O	9a	Н	12e	46
23	7b	CH <sub>3</sub> O	8d	CH <sub>3</sub> O	9b	4-CH <sub>3</sub>	12f	25

indicate that some substances inhibited the growth of the test microorganisms demonstrating weak antimicrobial effect (Table 2). The growth of gram-positive bacteria (strains of *S. aureus* and *B. subtilis*) was inhibited in a more effective way. Particularly, compound **11b** inhibited the growth of *B. subtilis* at a concentration of 125 mg/L. A bacteriostatic activity against *S. aureus* was observed only at the higher concentrations of 250 and 500 mg/L. The same situation was found for the tested *E.* 

*coli* strain. The gram-negative bacterium *P. aeruginosa* showed resistance to all compounds tested in the given concentration range. The observed low level of antibacterial activity of the synthesized heterocycles is a good prerequisite for screening them for other types of activity, e.g., anticancer, antidiabetic, etc., because in these cases a negative influence on the microflora of the organism would be decreased [54].

Table 2: Antibacterial activity results.										
Entry	Compound	MIC <sup>a</sup> /MBC <sup>b</sup> , mg/L	Strains of test cultures							
			Escherichia coli	Pseudomonas aeruginosa	Staphylococcus aureus	Bacillus subtilis				
1	11a	MIC	250	_c	250	250				
		MBC	-	-	-	_				
2	11b	MIC	500	_	500	125				
		MBC	-	_	_	_				
3	11f	MIC	_	_	_	_				
		MBC	_	_	_	_				
4	11g	MIC	250	_	_	250				
		MBC	_	_	_	_				
5	111	MIC	_	_	_	_				
		MBC	_	_	_	_				
6	11m	MIC	500	_	250	250				
		MBC	_	_	_	_				
7	nitroxoline	MIC	15.6	62.5	31.25	1.9				
		MBC	15.6	62.5	31.25	1.9				

<sup>a</sup>MIC – minimum inhibitory concentration; <sup>b</sup>MBC – minimum bactericidal concentration; <sup>c</sup>the substance at concentration ≤ 500 mg/L does not inhibit culture growth.

# Conclusion

In summary, two multicomponent reactions of Doebner and Ugi-type were combined in a convergent and versatile manner giving substituted 1*H*-pyrazolo[3,4-*b*]pyridine-4- and 1*H*-pyrazolo[3,4-*b*]pyridine-6-carboxamides. The use of a conditions-based divergence strategy allowed introducing the scaffold diversity and obtaining two types of structures with different orientation of substituents (containing a carboxylic group either at C4 or C6 position of the pyrazolopyridine core). The optimal methodology for the synthesis of target products was elaborated (mixture of methanol and DMF (2:1) and heating to 70 °C) and a small focused library of 23 Ugi products was created. The target compounds containing two pharmacophore pyrazolopyridine and peptidomimetic moieties were screened for their antibacterial activity and demonstrated weak antibacterial effect.

# Supporting Information

#### Supporting Information File 1

Experimental and analytical data. [https://www.beilstein-journals.org/bjoc/content/ supplementary/1860-5397-15-126-S1.pdf]

# Supporting Information File 2

NMR spectra. [https://www.beilstein-journals.org/bjoc/content/ supplementary/1860-5397-15-126-S2.pdf]

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# References

- Ruijter, E.; Scheffelaar, R.; Orru, R. V. A. Angew. Chem., Int. Ed. 2011, 50, 6234–6246. doi:10.1002/anie.201006515
- Dömling, A.; Ugi, I. Angew. Chem., Int. Ed. Engl. 1993, 32, 563–564. doi:10.1002/anie.199305631
- Elders, N.; van der Born, D.; Hendrickx, L. J. D.; Timmer, B. J. J.; Krause, A.; Janssen, E.; de Kanter, F. J. J.; Ruijter, E.; Orru, R. V. A. *Angew. Chem., Int. Ed.* **2009**, *48*, 5856–5859. doi:10.1002/anie.200902683

- Welsch, M. E.; Snyder, S. A.; Stockwell, B. R. Curr. Opin. Chem. Biol. 2010, 14, 347–361. doi:10.1016/j.cbpa.2010.02.018
- Marcaurelle, L. A.; Foley, M. A. Curr. Opin. Chem. Biol. 2010, 14, 285–288. doi:10.1016/j.cbpa.2010.05.001
- Cowen, S. D.; Russell, D.; Dakin, L. A.; Chen, H.; Larsen, N. A.; Godin, R.; Throner, S.; Zheng, X.; Molina, A.; Wu, J.; Cheung, T.; Howard, T.; Garcia-Arenas, R.; Keen, N.; Pendleton, C. S.; Pietenpol, J. A.; Ferguson, A. D. J. Med. Chem. 2016, 59, 11079–11097. doi:10.1021/acs.jmedchem.6b01303
- Giannouli, V.; Lougiakis, N.; Kostakis, I. K.; Pouli, N.; Marakos, P.; Skaltsounis, A.-L.; Nam, S.; Jove, R.; Horne, D.; Tenta, R.; Pratsinis, H.; Kletsas, D. *Bioorg. Med. Chem. Lett.* **2016**, *26*, 5229–5233. doi:10.1016/j.bmcl.2016.09.056
- Wenglowsky, S. Expert Opin. Ther. Pat. 2013, 23, 281–298. doi:10.1517/13543776.2013.749861
- Ahmed, O. M.; Mohamed, M. A.; Ahmed, R. R.; Ahmed, S. A. *Eur. J. Med. Chem.* 2009, 44, 3519–3523. doi:10.1016/j.ejmech.2009.03.042
- 10. Abdel-Mohsen, S. A.; El-Emary, T. I. Pharma Chem. 2018, 10, 44-51.
- Foks, H.; Pancechowska-Ksepko, D.; Kędzia, A.; Zwolska, Z.; Janowiec, M.; Augustynowicz-Kopeć, E. *Farmaco* 2005, *60*, 513–517. doi:10.1016/j.farmac.2005.05.002
- Patel, J. B.; Malick, J. B.; Salama, A. I.; Goldberg, M. E. *Pharmacol., Biochem. Behav.* **1985**, *23*, 675–680. doi:10.1016/0091-3057(85)90436-8
- Dodiya, D. K.; Trivedi, A. R.; Kataria, V. P.; Shah, V. H. *Curr. Org. Chem.* **2012**, *16*, 400–417. doi:10.2174/138527212799499912
- Soural, M.; Bouillon, I.; Krchnák, V. J. Comb. Chem. 2008, 10, 923–933. doi:10.1021/cc8001074
- Edraki, N.; Firuzi, O.; Fatahi, Y.; Mahdavi, M.; Asadi, M.; Emami, S.; Divsalar, K.; Miri, R.; Iraji, A.; Khoshneviszadeh, M.; Firoozpour, L.; Shafiee, A.; Foroumadi, A. Arch. Pharm. (Weinheim, Ger.) 2015, 348, 330–337. doi:10.1002/ardp.201400322
- 16. He, L.-J.; Yang, D.-L.; Li, S.-Q.; Zhang, Y.-J.; Tang, Y.; Lei, J.; Frett, B.; Lin, H.-k.; Li, H.-y.; Chen, Z.-Z.; Xu, Z.-G. *Bioorg. Med. Chem.* **2018**, 26, 3899–3908. doi:10.1016/j.bmc.2018.06.010
- 17. Avilés, E.; Prudhomme, J.; Le Roch, K. G.; Franzblau, S. G.; Chandrasena, K.; Mayer, A. M. S.; Rodríguez, A. D. *Bioorg. Med. Chem. Lett.* **2015**, *25*, 5339–5343. doi:10.1016/j.bmcl.2015.09.033
- 18. Huang, Y.; Wolf, S.; Bista, M.; Meireles, L.; Camacho, C.; Holak, T. A.; Dömling, A. *Chem. Biol. Drug Des.* **2010**, *76*, 116–129. doi:10.1111/j.1747-0285.2010.00989.x
- Ingold, M.; Dapueto, R.; Victoria, S.; Galliusi, G.; Batthyàny, C.; Bollati-Fogolín, M.; Tejedor, D.; García-Tellado, F.; Padrón, J. M.; Porcal, W.; López, G. V. *Eur. J. Med. Chem.* **2018**, *143*, 1888–1902. doi:10.1016/j.ejmech.2017.11.003
- 20. Yang, S.; Jyothi, K. R.; Lim, S.; Choi, T. G.; Kim, J.-H.; Akter, S.; Jang, M.; Ahn, H.-J.; Kim, H.-Y.; Windisch, M. P.; Khadka, D. B.; Zhao, C.; Jin, Y.; Kang, I.; Ha, J.; Oh, B.-C.; Kim, M.; Kim, S. S.; Cho, W.-J. *J. Med. Chem.* **2015**, *58*, 9546–9561. doi:10.1021/acs.jmedchem.5b01064
- Azuaje, J.; El Maatougui, A.; García-Mera, X.; Sotelo, E. ACS Comb. Sci. 2014, 16, 403–411. doi:10.1021/co500036n
- 22. Che, C.; Li, S.; Yu, Z.; Li, F.; Xin, S.; Zhou, L.; Lin, S.; Yang, Z. ACS Comb. Sci. 2013, 15, 202–207. doi:10.1021/co400001h
- Ghandi, M.; Zarezadeh, N.; Abbasi, A. Org. Biomol. Chem. 2015, 13, 8211–8220. doi:10.1039/c5ob01095k

- 24. Shaw, A. Y.; Xu, Z.; Hulme, C. *Tetrahedron Lett.* **2012**, *53*, 1998–2000. doi:10.1016/j.tetlet.2012.02.030
- 25. Xu, Z.; Martinez-Ariza, G.; Cappelli, A. P.; Roberts, S. A.; Hulme, C. J. Org. Chem. 2015, 80, 9007–9015. doi:10.1021/acs.joc.5b00955
- Balalaie, S.; Saeedi, S.; Ramezanpour, S. *Helv. Chim. Acta* 2016, 99, 138–142. doi:10.1002/hlca.201500187
- Konstantinidou, M.; Kurpiewska, K.; Kalinowska-Tłuscik, J.; Dömling, A. *Eur. J. Org. Chem.* **2018**, 6714–6719. doi:10.1002/ejoc.201801276
- 28. Liao, W.-L.; Li, S.-Q.; Wang, J.; Zhang, Z.-Y.; Yang, Z.-W.; Xu, D.; Xu, C.; Lan, H.-T.; Chen, Z.-Z.; Xu, Z.-G. ACS Comb. Sci. 2016, 18, 65–69. doi:10.1021/acscombsci.5b00145
- Caputo, S.; Basso, A.; Moni, L.; Riva, R.; Rocca, V.; Banfi, L. Beilstein J. Org. Chem. 2016, 12, 139–143. doi:10.3762/bjoc.12.15
- 30. Zheng, Q.; Mi, N.; Fan, Z.; Zuo, X.; Zhang, H.; Wang, H.; Yang, Z. J. Agric. Food Chem. 2010, 58, 7846–7855. doi:10.1021/jf1006193
- Sheikhhosseini, E.; Balalaie, S.; Bigdeli, M. A.; Habibi, A.; Moghaddam, H. P. *J. Korean Chem. Soc.* **2014**, *58*, 186–192. doi:10.5012/jkcs.2014.58.2.186
- Kumar, S.; Mukesh, K.; Harjai, K.; Singh, V. Tetrahedron Lett. 2019, 60, 8–12. doi:10.1016/j.tetlet.2018.11.030
- Alavijeh, N. S.; Ramezanpour, S.; Alavijeh, M. S.; Balalaie, S.; Rominger, F.; Misra, A.; Bijanzadeh, H. R. *Monatsh. Chem.* 2014, *145*, 349–356. doi:10.1007/s00706-013-1098-0
- 34. Plant, A.; Thompson, P.; Williams, D. M. J. Org. Chem. 2009, 74, 4870–4873. doi:10.1021/jo900244m
- Beaumont, S.; Retailleau, P.; Dauban, P.; Dodd, R. H. Eur. J. Org. Chem. 2008, 5162–5175. doi:10.1002/ejoc.200800643
- 36. Van den Bogaert, A. M.; Nelissen, J.; Ovaere, M.; Van Meervelt, L.; Compernolle, F.; De Borggraeve, W. M. *Eur. J. Org. Chem.* **2010**, 5397–5401. doi:10.1002/ejoc.201000549
- Mahdavi, M.; Hassanzadeh-Soureshjan, R.; Saeedi, M.; Ariafard, A.; BabaAhmadi, R.; Ranjbar, P. R.; Shafiee, A. RSC Adv. 2015, 5, 101353–101361. doi:10.1039/c5ra17056g
- 38. Purohit, P.; Pandey, A. K.; Kumar, B.; Chauhan, P. M. S. RSC Adv. 2016, 6, 21165–21186. doi:10.1039/c5ra27090a
- Madhavachary, R.; Naveen, N.; Wang, Y.; Wang, Q.; Konstantinidou, M.; Dömling, A. *Eur. J. Org. Chem.* **2018**, 3139–3143. doi:10.1002/ejoc.201800557
- 40. Pandey, S.; Khan, S.; Singh, A.; Gauniyal, H. M.; Kumar, B.; Chauhan, P. M. S. J. Org. Chem. 2012, 77, 10211–10227. doi:10.1021/jo3018704
- 41. Balalaie, S.; Bararjanian, M.; Hosseinzadeh, S.; Rominger, F.; Bijanzadeh, H. R.; Wolf, E. *Tetrahedron* **2011**, *67*, 7294–7300. doi:10.1016/j.tet.2011.07.052
- Rasouli, M. A.; Mahdavi, M.; Firoozpour, L.; Shafiee, A.; Foroumadi, A. Tetrahedron 2014, 70, 3931–3934. doi:10.1016/j.tet.2014.03.079
- 43. Al-Tel, T. H.; Al-Qawasmeh, R. A.; Voelter, W. Eur. J. Org. Chem. 2010, 5586–5593. doi:10.1002/ejoc.201000808
- 44. Castellano, T. G.; Neo, A. G.; Marcaccini, S.; Marcos, C. F. Org. Lett. 2012, 14, 6218–6221. doi:10.1021/ol302976g
- Chebanov, V. A.; Sakhno, Y. I.; Desenko, S. M.; Chernenko, V. N.; Musatov, V. I.; Shishkina, S. V.; Shishkin, O. V.; Kappe, C. O. *Tetrahedron* **2007**, *63*, 1229–1242. doi:10.1016/j.tet.2006.11.048
- 46. Murlykina, M. V.; Sakhno, Y. I.; Desenko, S. M.; Konovalova, I. S.; Shishkin, O. V.; Sysoiev, D. A.; Kornet, M. N.; Chebanov, V. A. *Tetrahedron* **2013**, *69*, 9261–9269. doi:10.1016/j.tet.2013.08.055
- Sakhno, Y. I.; Murlykina, M. V.; Morozova, A. D.; Kozyryev, A. V.; Chebanov, V. A. *Fr.-Ukr. J. Chem.* **2015**, *3*, 1–20. doi:10.17721/fujcv3i2p1-20

- Murlykina, M. V.; Morozova, A. D.; Zviagin, I. M.; Sakhno, Y. I.; Desenko, S. M.; Chebanov, V. A. *Front. Chem. (Lausanne, Switz.)* 2018, 6, 527–569. doi:10.3389/fchem.2018.00527
- Sakhno, Y. I.; Kozyryev, A. V.; Desenko, S. M.; Shishkina, S. V.; Musatov, V. I.; Sysoiev, D. O.; Chebanov, V. A. *Tetrahedron* 2018, *74*, 564–571. doi:10.1016/j.tet.2017.12.031
- Sakhno, Y. I.; Desenko, S. M.; Shishkina, S. V.; Shishkin, O. V.;
  Sysoyev, D. O.; Groth, U.; Kappe, C. O.; Chebanov, V. A. *Tetrahedron* 2008, *64*, 11041–11049. doi:10.1016/j.tet.2008.09.089
- Murlykina, M. V.; Sakhno, Y. I.; Desenko, S. M.; Shishkina, S. V.; Shishkin, O. V.; Sysoiev, D. O.; Kornet, M. N.; Schols, D.; Goeman, J. L.; Van der Eycken, J.; Van der Eycken, E. V.; Chebanov, V. A. *Eur. J. Org. Chem.* **2015**, 4481–4492. doi:10.1002/ejoc.201500469
- 52. Annan, N.; Paris, R.; Jordan, F. J. Am. Chem. Soc. 1989, 111, 8895–8901. doi:10.1021/ja00206a019
- Meng, Q.; Zhu, L.; Zhang, Z. J. Org. Chem. 2008, 73, 7209–7212. doi:10.1021/jo801140j
- Albert, A., Ed. Selective Toxicity: the physico-chemical basis of therapy, 7th ed.; Springer: Netherlands, 1985. doi:10.1007/978-94-009-4846-4

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