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A ruthenium(II) complex-cyanine energy transfer scaffold based luminescence probe for ratiometric detection and imaging of mitochondrial peroxynitrite[†]

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A novel ruthenium(II) complex-cyanine energy transfer scaffold has been established for the development of a ratiometric luminescence probe for $ONOO^-$ detection. The probe, Ru-Cy5, is localized in mitochondria of live cells, allowing ratiometric sensing and imaging of $ONOO^-$ therein.

As a member of reactive nitrogen species (RNS), peroxynitrite (ONOO⁻) is endogenously generated by the diffusion-controlled reaction of endogenous nitric oxide (NO) and superoxide $(O_2^{\bullet-})$ mainly in mitochondria.¹ In living organisms, ONOO⁻ at a reasonable concentration acts as a signalling transduction molecule to modulate a series of cellular functions, such as selective nitration of tyrosine residues.² On the other hand, an elevated level of ONOO⁻ can directly oxidize or decompose lipids, proteins, and nucleic acids, leading to the apoptosis and death of cells.² Growing evidence has established that the level of ONOO⁻ is highly correlated with the progress of various diseases and disorders, such as cardiovascular and metabolic diseases, shock, arteriosclerosis, neurodegenerative diseases and even cancers.³ Therefore, an effective method for ONOO⁻ detection is required that might benefit disease early diagnosis and treatment monitoring.

Towards this end, the development of molecular probes that are capable of sensing and imaging ONOO⁻ has attracted considerable attention due to the advantages of high sensitivity, selectivity, and simplicity in data collection of sensing technology.^{4–7} In the past few years, a number of probes have been engineered, allowing ONOO⁻ detection to be performed by recording their emission intensity changes during the sensing process.^{8–16} In comparison with the

intensity on/off approach,^{17–20} ratiometric probes that have a self-calibration of external interference on emission intensity are desirable for effective detection of ONOO[–] in mitochondria of live cells.^{21–23}

The ratiometric changes in luminescence intensities at two emission wavelengths mainly arise from the factor of Förster resonance energy transfer (FRET) where generally organic dyes act as energy donors and acceptors. Energy transfer (ET) scaffolds established by exploiting luminescent metal complexes, especially ruthenium(II) complexes, are scarcely reported,^{24,25} but such a scaffold is important due to the abundant photophysics and photochemistry properties of these luminophores, such as strong visible absorptions and emissions, large Stokes shifts, and good water solubility.^{26–33} Furthermore, previous research of metal complex probes is mainly focused on the modulation of their excited states through photoinduced electron transfer (PET) and intramolecular charge transfer (ICT), leaving the ET scaffold strategy yet to be well demonstrated.

In this contribution, we report a Ru(II) complex-cyanine based ET scaffold and the development of a luminescence probe for ratiometric detection of ONOO⁻. Based on this ET system, the ratiometric luminescence probe, **Ru–Cy5**, was engineered by employing a Ru(II) complex as the energy donor and Cy5 as the energy acceptor. The absorption of Cy5 (λ_{abs} = 630 nm) is diminished due to the ONOO--mediated oxidative cleavage of the polymethine bridge (Scheme 1A).²² The ET from the Ru(n)complex to Cy5 is thus significantly interrupted (Scheme 1B), leading to the decrease of Cy5 emission (λ_{em} = 660 nm) and the increase of Ru(II) emission (λ_{em} = 610 nm) for ratiometric (I_{610}/I_{660}) detection of ONOO⁻. Furthermore, owing to the mitochondria targeting feature of Cy5 after cellular internalization,²⁰ Ru-Cy5 is expected to be localized in mitochondria, to enable ONOO⁻ in this organelle to be visualized (Scheme 1C) under a ratiometric mode. To the best of our knowledge, this is the first report on constructing a Ru(II) complex-Cy5 ET probe for ratiometric luminescent sensing and imaging of biomolecules.

To synthesize **Ru–Cy5**, a Cy5 derivative, compound **8**, was first synthesized (Scheme S1, ESI†). Then, **Ru–Cy5** was obtained

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Scheme 1 (A) ONOO⁻-mediated oxidative cleavage of the polymethine bridge of Cy5; (B) FRET scaffold of **Ru–Cy5** and its response mechanism to ONOO⁻; (C) **Ru–Cy5** in the sensing and imaging of ONOO⁻ in mitochondria of live cells.

by reacting compound **8** with a Ru(π)-bipyridine complex derivative (complex **10**) that was synthesized through chloridization of the carboxyl acid in complex **9** (Scheme S1, ESI[†]). The chemical structures of all intermediates and **Ru–Cy5** were confirmed by NMR, ESI-MS, HRMS, and elemental analysis (Fig. S1–S17, ESI[†]). To validate the reaction mechanism of **Ru–Cy5** with ONOO⁻, their reaction products were investigated by ESI-MS analysis. As shown in Fig. S18 (ESI[†]), the peak at *m*/*z* 632.20 was absent, and a new peak at *m*/*z* 448.70 that can be assigned to the species of [**Ru–Cy**]²⁺ was found, which confirmed the reaction of **Ru–Cy5** with ONOO⁻, with the formation of [**Ru–Cy**]²⁺ as proposed in Scheme 1B.

In the absence and presence of ONOO⁻, changes of UV-vis spectra in **Ru–Cy5** were initially examined in 25 mM PBS buffer of pH 7.4. As shown in Fig. 1A, the absorption spectrum of **Ru–Cy5** showed two intense absorption bands centred at 450 nm and 630 nm, which can be ascribed to the metal-to-ligand charge transfer (MLCT) absorption of the Ru(n) complex and the typical absorption of Cy5. Upon the reaction with ONOO⁻, the absorption band at 630 nm was remarkably decreased due to the oxidization of Cy5, while no obvious changes at 450 nm were noticed, indicating the tolerance of the Ru(n) complex to ONOO⁻. On increasing the concentration of ONOO⁻, the color of the solution was clearly changed from blue to yellow (Fig. 1A inset). These facts indicate that **Ru–Cy5** can act as a "naked-eye" probe for ONOO⁻ detection.

Significant changes in the emission spectrum of **Ru–Cy5** were also observed in the presence of different concentrations of ONOO⁻ in 25 mM PBS buffer of pH 7.4 (Fig. 1B). Under excitation at



Fig. 1 (A) Absorption spectrum changes of **Ru–Cy5** (10 μ M) in the presence of different concentrations of ONOO⁻ (0, 30, 80, 100, 150 and 200 μ M) in 25 mM PBS buffer of pH 7.4 (inset: color changes of the **Ru–Cy5** solution upon reaction with different concentrations of ONOO⁻); (B) emission spectra ($\lambda_{ex} = 450$ nm) of **Ru–Cy5** (10 μ M) upon reaction with different concentrations of ONOO⁻ (0–220 μ M); (C) the I_{610}/I_{660} ratio as a function of the ONOO⁻ concentration; (D) calibration curve for the ratiometric luminescence detection of ONOO⁻. The absorption and emission spectra were recorded after incubating **Ru–Cy5** with ONOO⁻ for 30 min.

450 nm, Ru-Cy5 exhibited a strong emission band centred at 660 nm, the characteristic emission of Cv5, which suggested the effective ET from the Ru(II) complex to the Cy5 moiety. In the presence of ONOO⁻, owing to the oxidation of Cy5 and the corruption of ET (Scheme 1B), the emission band at 660 nm was remarkably decreased with the increase of the emission intensity at 610 nm. The ratio of the intensity at 610 nm to that at 660 nm (I_{610}/I_{660}) was found to be more than 46-fold enhanced in the presence of 200 µM ONOO- (Fig. 1C). Furthermore, the enhancement in the I_{610}/I_{660} ratio showed an excellent linear correlation to the concentration of $ONOO^-$ in the range from 0-30 μ M. Based on the concentration corresponding to three standard deviations of the background signal, the detection limit $(3\sigma/k)$ was calculated to be 0.28 µM. This result suggests that Ru-Cy5 could be used as a ratiometric luminescence probe for highly sensitive quantification of ONOO- in aqueous media.

The specificity of **Ru–Cy5** for responding to ONOO⁻ over other interfering species, including ROS, RNS, amino acids and metal ions, was investigated. As shown in Fig. 2A, more than 46-fold enhancement in the I_{610}/I_{660} ratio was noticed after **Ru–Cy5** was reacted with 20 equiv. of ONOO⁻, while the reactions of **Ru–Cy5** with 100 equiv. of other species, including HOCl (20 equiv.), H_2O_2 , •OH, $O_2^{\bullet-}$, 1O_2 , NO, NO_2^- , HS⁻, Cys, Hcy, GSH (5 mM), Ca²⁺, Zn²⁺, K⁺, Na⁺, Mg²⁺, Fe²⁺ and Fe³⁺, could not induce the change in the I_{610}/I_{660} ratio. These results reveal that the luminescence response of **Ru–Cy5** towards ONOO⁻ is highly specific. The effect of pH on the ratiometric luminescence response of **Ru–Cy5** towards ONOO⁻ was then evaluated in 25 mM PBS buffer with different pH values (pH 4–10). As shown in Fig. 2B, in the absence of ONOO⁻, **Ru–Cy5** exhibited



Fig. 2 (A) The I_{610}/I_{660} ratios of **Ru–Cy5** (10 µM) upon reactions with various ROS, RNS, amino acids and ions (a, control; b, ONOO⁻; c, HOCl; d, NO₂⁻; e, NO; f, HS⁻; g, H₂O₂; h, •OH; i, O₂^{•-}; j, ¹O₂; k, Cys; l, Hcy; m, GSH (5 mM); n, Ca²⁺; o, Zn²⁺; p, K⁺; q, Na⁺; r, Mg²⁺; s, Fe²⁺; and t, Fe³⁺) in 25 mM PBS buffer of pH 7.4. (B) Effects of pH on the I_{610}/I_{660} ratios of **Ru–Cy5** (10 µM) before and after the reaction with SIN-1 (200 µM). Excitation was performed at 450 nm.

small and stable I_{610}/I_{660} ratios in the pH range from 4 to 10. Upon reaction with 3-morpholinosydnonimine (SIN-1, a ONOO⁻ donor, 200 μ M), the I_{610}/I_{660} ratio remained low at pH below 6.5, and then rapidly increased to the maximum level at pH 7.5. Considering the weakly basic pH in mitochondria (pH ~8),^{34,35} the above result indicates that **Ru–Cy5** is favorable to be used for the detection of ONOO⁻ in the mitochondria of live cells.

Fig. S19 (ESI[†]) illustrated the time profile changes of the I_{610}/I_{660} ratio of **Ru–Cy5** in the absence and presence of 200 µM ONOO⁻. The I_{610}/I_{660} ratio of **Ru–Cy5** kept at a low and stable value during the period of 25 min. However, upon the addition of ONOO⁻, the I_{610}/I_{660} ratio was gradually increased, and reached a maximum after 20 min incubation, which reveals that the reaction between **Ru–Cy5** and ONOO⁻ could be completed within 20 min. Upon the addition of ONOO⁻, a good linearity between ONOO⁻ concentration and the corresponding initial reaction rate is obtained (Fig. S20, ESI[†]). According to the reported methods, ${}^{36,37}_{36,37}$ the total reaction rate constant (k_{tot}) and reaction half-life time ($t_{1/2}$) were estimated to be 6.48×10^4 M⁻¹ min⁻¹ and 1.07×10^{-5} min, respectively.

On the basis of the above results, the feasibility of **Ru–Cy5** for sensing and imaging of mitochondrial ONOO[–] was evaluated. MTT assay showed that **Ru–Cy5** has low toxicity towards live cells. As shown in Fig. S21 (ESI†), no obvious changes in cell proliferation were noticed when HeLa cells were incubated with increased concentrations of **Ru–Cy5**. The cell viability remained more than 84% even after incubating HeLa cells with 50 µM of **Ru–Cy5** for 24 h, suggesting the low cytotoxicity of **Ru–Cy5** for loading live cells.

The distribution of **Ru–Cy5** in mitochondria of live cells was confirmed by colocalization analysis with a commercially available mitochondria labeling agent, MitoTrackerTM Green FM. HeLa cells were incubated with 50 μ M **Ru–Cy5** for 2 h, and then stained with MitoTrackerTM Green FM for the colocalization analysis. As shown in Fig. 3, the mitochondrial distribution was supported by the synchronous luminescence signals in the linear region of interest across the cells. The excellent overlapping of red (**Ru–Cy5**) and green (MitoTrackerTM Green FM) channels was observed. The Pearson's correlation coefficient (0.94) is close to 1, revealing the mitochondrial localization of the internalized **Ru–Cy5** molecules.

With the mitochondria-localization feature of **Ru–Cy5**, mitochondrial ONOO[–] in live cells was visualized *via* a ratiometric



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Fig. 3 Colocalization analysis of internalized **Ru–Cy5** and MitoTracker[™] Green FM in HeLa cells. (A) Green channel of MitoTracker[™] Green FM; (B) red channel of **Ru–Cy5**; (C) merged image of A, B, and a bright-field image; (D) intensity correlation plot of **Ru–Cy5** and MitoTracker[™] Green FM; (E) intensity profiles of the linear regions of interest (ROIs) across HeLa cells. Excitation at 488 nm was used for both the red channel (640–680 nm) and the green channel (500–540 nm).

luminescence imaging approach. As shown in Fig. 4, under laser excitation at 488 nm, the control group of HeLa cells (incubated with 25 μ M **Ru–Cy5** for 2 h only) showed a weak signal in the Ru(π) channel (590–630 nm) and an intense signal in the Cy5 channel (640–680 nm). After supplying with 500 μ M SIN-1 for 30 min, significant decrease of the Cy5 channel signal and increase of the Ru(π) channel signal were observed, with a more than 7-fold change of the luminescence intensity ratio. These results demonstrated the feasibility of **Ru–Cy5** for the visualization of mitochondrial ONOO[–] in live cells under ratiometric mode.

In summary, a Ru(II) complex-cyanine energy transfer based luminescence probe, **Ru-Cy5**, has been developed for the ratiometric



Fig. 4 Confocal luminescence images of ONOO⁻ in HeLa cells using **Ru-Cy5** as a probe. (A) HeLa cells were incubated with **Ru-Cy5** (25μ M) for 2 h at 37 °C and then treated with 500 μ M SIN-1 for 30 min (control group: HeLa cells were stained with **Ru-Cy5** only). Image analysis: luminescence intensity analysis (B) and changes of the luminescence intensity ratio (C). Scale bar: 10 μ m.

detection of ONOO⁻. This probe showed high response specificity and sensitivity to ONOO⁻, as well as low cytotoxicity, good cell membrane permeability and mitochondria-localization feature, which enabled it to be successfully used for the ratiometric sensing and imaging of ONOO⁻ in the mitochondria of live cells, to provide a useful tool for investigating the functions of ONOO⁻ in live cells. In addition, the ET scaffold strategy described in this work also contributes to the future development of luminescent transition metal complex-based ratiometric probes.

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Conflicts of interest

There are no conflicts to declare.

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