

**NEW SYNTHETIC APPROACHES TO $Mn_2(CO)_{10-n}L_n$
DERIVATIVES (L = PHOSPHINES): THEIR CATALYTIC
ACTIVITY IN THE HOMOGENEOUS HYDROGENATION OF
DIPHENYLACETYLENE AND THE CRYSTAL STRUCTURE OF
 $Mn_2(CO)_9(PPh_2H)$**

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Abstract—Novel $Mn_2(CO)_{10-n}L_n$ complexes have been investigated as hydrogenation catalysts and a representative member has been structurally characterized.

We are investigating the behaviour of carbonyl clusters, and in particular that of phosphine-substituted and phosphido-bridged derivatives as homogeneous hydrogenation catalysis.¹ We have studied the behaviour of $Mn_2(CO)_{10}$ (1), $Mn_2(CO)_9(PPh_2H)$ (2), $Mn_2(\mu-H)(\mu-PPhH)(CO)_8$ (3), $Mn_2(\mu-H)(\mu-PPh_2)(CO)$ (4) and $Mn_2(\mu-PPh_2)_2(CO)_8$ (5) in the hydrogenation of diphenylacetylene.

Phosphine-substituted dimanganese derivatives are generally obtained by thermal or photochemical activation of the CO's, generally in non-selective processes; indeed it is believed that Mn—Mn bond cleavage occurs and that the phosphine-substituted fragments recombine in a random fashion.³

We have attempted the synthesis of $Mn_2(CO)_{10-n}L_n$ derivatives by labilization of CO's with Me_3NO^4 in relatively mild conditions (L = PPh_3 , PPh_2H , $PPh_2CH=CH_2$, $PPh_2PC\equiv CPh_2$). Heating $Mn_2(CO)_{10}$ with a 1–5 molar excess of ligand in hexane or heptane, in the presence of dry Me_3NO for 2 min, leads to mixtures containing 30–40% of $Mn_2(CO)_9L$, about 5–10% of $Mn_2(CO)_8L_2$ (one or two isomers) and about 50% of unreacted carbonyl (which was re-used).

The complexes were identified by spectroscopy and, in particular, by ³¹P NMR; for 2 a sharp singlet at δ 51.12 was observed. However, for the other monosubstituted complexes sharp singlets centred at δ 50–60 and a broad peak at δ 20–30 was found (but for PPh_3 , a sharp singlet at δ 29.52, a broad peak at δ 74.62 ppm). This should indicate

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Table 1

Complex	Reaction time ^a	Conversion ^a	Selectivity to:		
			<i>cis</i> -stilbene	<i>trans</i> -stilbene	<i>trans-cis</i> ratio ^b
1	1–27	5.0–12.5	24.0–22.4	76.0–77.6	3.2–3.5
2	1–48	5.1–11.9	15.7–21.0	84.3–79.0	5.4–3.8
3	1–76	14.2–21.3	28.9–27.2	71.1–72.8	2.5–2.7
4 ^c	48	50.4	20.2	79.8	3.9
5	1–48	5.8–15.4	25.9–27.3	74.1–72.7	2.9–2.7

^a Minimum and maximum value; 1 atm H₂.

^b Initial and final value.

^c Under 34 atm pressure of H₂.

that mostly equatorial, monosubstituted complexes are obtained.

Thus, apparently, the use of Me₃NO gives, in most cases, stereospecific substitution of one equatorial CO, a statistically favoured process. In order to confirm the structure of **2**, a suitable crystal of **2** was selected for an X-ray diffraction study. Crystal data: C₂₁H₁₁Mn₂O₉P, *M* = 548.16, monoclinic space group, *C*2/*c*, *a* = 24.038(8), *b* = 14.645(6), *c* = 16.328(6) Å, β = 125.58(2)°, *Z* = 8, *d*_{calc} = 1.559 g cm⁻³, μ(Mo-*K*_α) = 11.50 cm⁻¹. Of 4409 reflections collected on an Enraf-Nonius CAD-4 diffractometer (Mo-*K*_α) (3 < θ < 24°), 1548, having *I* > 2σ(*I*), were used in the refinement. All non-hydrogen atoms were refined anisotropically. Hydrogen atoms were clearly located and refined isotropically. Final *R* and *R*_w values were 0.0363 and 0.0436, respectively. Atomic coordinates have been deposited with the Director, Cambridge Crystallographic Data Centre. The structure of **2** is shown in Fig. 1. The Mn—Mn distance

of 2.910(2) Å, is practically identical to that found in Mn₂(CO)₁₀.⁵

Monosubstituted derivatives, characterized as axial, were obtained with PPh₃, PBu₃, P(OPh)₃, AsPh₃, PPhMe₂ (X-ray structure),⁶ AsPhMe₂, PPh₂Me and PF₃; only with PH₃ was an equatorial derivative observed.⁷ A few X-ray structures for axial,⁶ diaxial⁸ and diequatorial⁸ complexes have been reported; thus **2** is the first confirmed example of equatorial substitution.

Some preliminary results in homogeneous hydrogenation of diphenylacetylene are given in Table 1.

Worthy of note are the high *trans-cis* stilbene ratios observed; identification of the organometallic species in the hydrogenation solutions is in progress in order to gain better insight into the reaction mechanisms.

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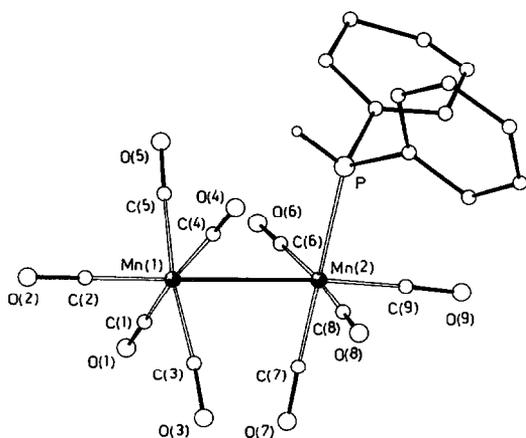


Fig. 1. View of the molecular structure of Mn₂(CO)₉(PPh₂H).