



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/lcyc20>

Synthesis of Optically Active (E)- γ -Hydroxy- α , β -unsaturated Nitriles

Suk-Ku Kang^a, Dong-Ha Lee^a, Yun-Sik Kim^a & Sin-Cheol Kang^a

^a Department of Chemistry, Sung Kyun Kwan University, Natural Science Campus, Suwon, 440-746, Korea

Published online: 23 Sep 2006.

To cite this article: Suk-Ku Kang, Dong-Ha Lee, Yun-Sik Kim & Sin-Cheol Kang (1992) Synthesis of Optically Active (E)- γ -Hydroxy- α , β -unsaturated Nitriles, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 22:8, 1109-1113, DOI: [10.1080/00397919208021094](https://doi.org/10.1080/00397919208021094)

To link to this article: <http://dx.doi.org/10.1080/00397919208021094>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with

primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at <http://www.tandfonline.com/page/terms-and-conditions>

SYNTHESIS OF OPTICALLY ACTIVE (*E*)- γ -HYDROXY
 α , β -UNSATURATED NITRILES

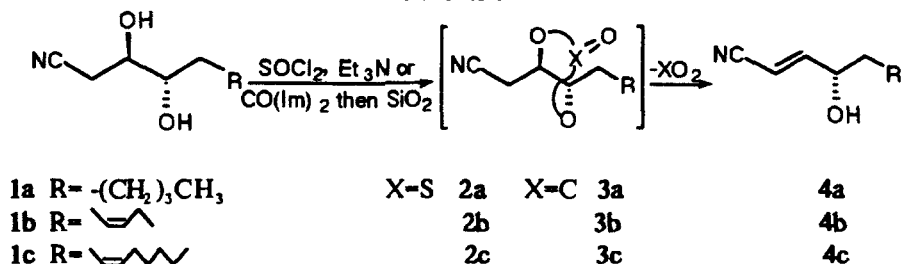
Suk-Ku Kang,* Dong-Ha Lee, Yun-Sik Kim, and Sin-Cheol Kang
Department of Chemistry, Sung Kyun Kwan University,
Natural Science Campus, Suwon 440-746, Korea

ABSTRACT : Treatment of (2*S*, 3*S*)- or (2*R*, 3*S*)-1-cyano-2, 3-alkanediols with carbonyldiimidazole or thionyl chloride resulted in one-pot preparation of (*E*, *S*)- γ -hydroxy- α , β -unsaturated nitriles based on the elimination of the cyclic carbonates or sulfites formed *in situ*.

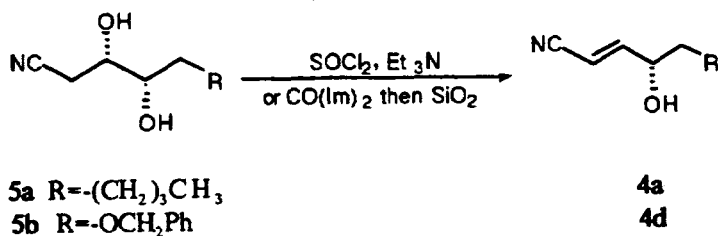
Optically active (*E*)- γ -hydroxy- α , β -unsaturated nitrile moiety was recently prepared and functionalized to chiral synthons.¹ The synthesis of (*E*, *S*)- γ -hydroxy- α , β -unsaturated nitriles by Sharpless kinetic resolution of γ -iodoallyl alcohol followed by reaction with CuCN was known.² Recently, optically active (*E*, *S*)- γ -hydroxy- α , β -unsaturated nitriles were synthesized by reaction of chiral sulfoxides with aldehydes,³ which in turn were asymmetrically amplified by lipase-mediated resolution.⁴ We wish to report here a convenient and a highly enantioselective route to (*E*, *S*)- γ -hydroxy- α , β -unsaturated nitriles from (2*S*, 3*S*)- or (2*R*, 3*S*)-1-cyano-2, 3-alkanediols by elimination *via* cyclic sulfites and carbonates prepared *in situ* (Scheme 1 and 2).

The (2*R*, 3*S*)-1-cyano-2, 3-alkanediols **1** were prepared from (2*R*, 3*S*)-2, 3-O-isopropylidenedioxyalkanols⁵ by one-pot conversion⁶ to nitrile (Tf₂O, pyridine, -50 °C, then NaCN, HMPA, rt) followed by deprotection (70 % HOAc, rt). On treatment of **1a** with thionyl chloride (1.2 equiv) in the presence of 5 equiv triethylamine at 0 °C rt, (*E*)- α , β -unsaturated nitrile **4a**, [α]_D²⁵ + 37.5° (*c* 2.1, CHCl₃) [lit.² [α]_D²⁵ + 36.8° (*c* 0.99, CHCl₃)], was obtained directly as the only product without formation of (*Z*)-isomer in 86 % yield (run 1, Table). The (*E*)-stereochemistry of **4a** was inferred from ¹H NMR (300 MHz) coupling constants of the two olefinic protons and capillary GLC. It is presumed that the cyclic sulfite **2a** is the intermediate in this conversion.⁷ Alternatively, reaction of **1a**

Scheme 1



Scheme 2



with carbonyldiimidazole (1.2 equiv) in dry benzene at room temperature gave the cyclic carbonate 3a, which on exposure to silica gel afforded 4a in 80 % overall yield (run 2). When 1a was treated with 3 equiv carbonyldiimidazole for 12 h, 1a was obtained directly (run 3).⁸

In these one-pot conversions, it is noteworthy that the cyclic sulfites and carbonates are good leaving groups in these β -elimination reactions. Alternatively, (2*S*, 3*S*)-1-cyano-2, 3-alkanediols 5a-b were prepared easily from L-tartaric acid. (2*S*, 3*S*)-2, 3-O-Isopropylidenedioxyoctanol⁹ derived from L-tartaric acid was converted to 5a by conversion to nitrile followed by deprotection. Treatment of 5a with thionyl chloride or carbonyldiimidazole afforded 4a (runs 8 and 9). The reaction sequence was also applied to prepare 4d (Table).

The hydroxy nitrile 4a thus synthesized was protected with silyl group and then reduced with DIBAH to afford the unsaturated aldehyde 6, $[\alpha]_D^{25} -19^\circ$ (*c* 0.4, CHCl_3) [lit.^{10, 11} -18.9° (*c* 1.67, CHCl_3)], which is the key chiral synthon in the synthesis¹¹ of (+)-coriolic acid, a self-defensive substance against the rice blast disease (Scheme 3).

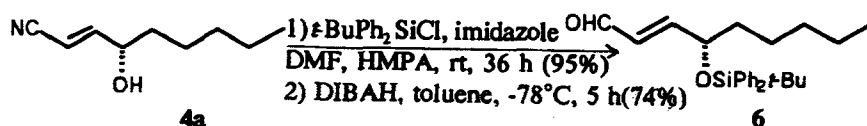
In summary, optically active (*E*)- γ -hydroxy- α , β -unsaturated nitriles were synthesized by mild and one-pot dehydration¹² of γ , β -dihydroxy nitriles.

Table Optically active (E)- γ -hydroxy- α,β -unsaturated nitriles prepared

Run	Diols	Products ^d	Reaction Conditions ^b	Yield(%) ^c
1	1a	4a	A	86
2	1a	4a	B	85
3	1a	4a	C	82
4	1b	4b	A	88
5	1b	4b	B	86
6	1c	4c	A	88
7	1c	4c	B	88
8	5a	4a	A	82
9	5a	4a	B	84
10	5b	4d	A	91
11	5b	4d	B	90

^aThe products 4a-4d were pure by capillary GLC analysis. ^bA: SOCl_2 (1.2 equiv), Et_3N (5 equiv), CH_2Cl_2 , rt, 1 h. B: $\text{CO}(\text{Im})_2$ (1.2 equiv), PhH , rt, 2 h then SiO_2 . C: $\text{CO}(\text{Im})_2$ (3.0 equiv), PhH , rt, 12 h. ^cThe yields are isolated yields.

Scheme 3



EXPERIMENTAL

(1E, 3S)-1-Cyano-1-octen-3-ol (4a): General Procedures :

Method A : To a stirred solution of the diol 1a (500 mg, 2.92 mmol) in dry CH_2Cl_2 (50 ml) at 0°C was added Et_3N (2.0 ml). The reaction mixture was warmed to room temperature and stirred for 1 h and then concentrated *in vacuo*. The crude product was dissolved in EtOAc (10 ml) and then filtered through Celite pad and then concentrated. The product was purified by column chromatography on silica gel using EtOAc /hexanes 1:3 as eluent to afford 4a (385 mg, 86%). TLC; SiO_2 , EtOAc /hexanes 1:3, $R_f=0.26$. $[\alpha]_D^{25} +37.5^\circ$ (c 2.1, CHCl_3) [$\text{lit}^2 +36.8^\circ$

(*c* 0.99, CHCl₃)] ¹H NMR (270 MHz, CDCl₃) δ 0.90 (t, 3H, *J*=7.5 Hz, CH₃), 1.20–1.78 (m, 8H, 4 CH₂), 2.12 (bs, 1H, OH), 4.18 (m, 1H, CHO), 5.73 (dd, 1H, *J*=15.5, 1.9 Hz, =CH), 6.82 (dd, 1H, *J*=15.5, 3.7 Hz, =CH). IR (neat) 3500, 2250, 1640 cm⁻¹. MS (*m/e*): 153 (*M*⁺).

Method B: To a stirred solution of the diol **1a** (500 mg, 2.92 mmol) in dry benzene (5.0 ml) was added carbonyldiimidazole (570 mg, 3.50 mmol) in dry benzene (5.0 ml) at room temperature. The reaction mixture was stirred at room temperature for 2 h. The mixture was washed with H₂O (10 ml) and then brine (10 ml). The organic layer was dried over anhydrous MgSO₄ and then concentrated *in vacuo*. The crude product (TLC; SiO₂, ether, *R*_f=0.65) was passed through a short pad of silica gel and then concentrated. The crude product (TLC; SiO₂, ether, *R*_f=0.93) was separated by column chromatography using EtOAc/hexanes 1:3 as eluent to afford **4a** (381 mg, 85 %).

(1*E*, 3*S*, 5*Z*)-1-Cyano-1,5-octadien-3-ol (4b)

TLC; SiO₂, EtOAc/hexanes 1:3, *R*_f=0.24. [α]_D²⁵ -8.24 ° (*c* 0.3, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ 0.97 (t, 3H, *J*=7.5 Hz, CH₃), 2.04 (m, 2H, CH₂), 2.36 (m, 2H, CH₂), 4.33 (m, 1H, CHO), 5.30–5.40 (m, 1H, =CH), 5.68 (m, 1H, =CH), 5.70 (dd, 1H, *J*=15.5, 2.0 Hz, =CH), 6.70 (dd, 1H, *J*=15.5, 3.8 Hz, =CH). IR (neat) 3450, 2250, 1640 cm⁻¹. MS (*m/e*): 151 (*M*⁺).

(1*E*, 3*S*, 5*Z*)-1-Cyano-1,5-undecadien-3-ol (4c)

TLC; SiO₂, EtOAc/hexanes 1:3, *R*_f=0.30. [α]_D²⁵ -1.57 ° (*c* 3.0, CHCl₃). ¹H NMR (80 MHz, CDCl₃) δ 0.91 (t, 3H, *J*=7.5 Hz, CH₃), 1.27 (m, 6H, 3 CH₂), 2.05 (m, 2H, CH₂), 2.37 (m, 2H, CH₂), 4.23 (m, 1H, CHO), 5.45 (m, 2H, CH=CH), 5.70 (dd, 1H, *J*=15.5, 1.9 Hz, =CH), 6.76 (dd, 1H, *J*=15.5, 3.9 Hz, =CH). IR (neat) 3450, 2250, 1640 cm⁻¹. MS (*m/e*): 192 (*M*⁺).

(1*E*, 3*S*)-4-Benzyloxy-1-cyano-1-buten-3-ol (4d)

TLC; SiO₂, EtOAc/hexanes 1:3, *R*_f=0.14, [α]_D²⁵ +34.1 ° (*c* 2.1, CHCl₃). ¹H NMR (80 MHz, CDCl₃) δ 3.37–3.67 (m, 2H, OCH₂), 4.35 (m, 1H, CHO), 4.50 (s, 2H, PhCH₂O), 5.75 (dd, 1H, *J*=15.5, 3.8 Hz, =CH), 6.72 (dd, 1H, *J*=15.5, 2.0 Hz, =CH), 7.38 (s, 5H, Ar). IR (neat) 3450, 2250, 1640 cm⁻¹. MS (*m/e*): 193 (*M*⁺), 91 (base peak).

Acknowledgment. Generous financial support by Korea Science and Engineering Foundation (KOSEF) and the Organic Chemistry Research Center - KOSEF is gratefully acknowledged. We thank Dr. Kwan-Ung Kim for helpful discussion.

REFERENCES AND NOTES

1. I. Yamakawa, H. Urabe, Y. Kobayashi, and F. Sato, *Tetrahedron Lett.*, **32**, 2045 (1991).
2. Y. Kitano, T. Matsumoto, T. Wakasa, S. Okamoto, T. Shimazaki, Y. Kobayashi, F. Sato, K. Miyaji, and K. Arai, *Tetrahedron Lett.*, **28**, 6351 (1987).
3. J. Nakami, T. Manda, A. Nishimura, T. Takeda, and S. Wakabayashi, *Tetrahedron Lett.*, **27**, 5109 (1986).
4. K. Burgess, J. Cassidy, and I. Henderson, *J. Org. Chem.*, **56**, 2050 (1991).
5. S-K. Kang, Y-S. Kim, J-S. Lim, K-S. Kim, and S-G. Kim, *Tetrahedron Lett.*, **32**, 363 (1991).
6. A. P. Kozikowski, and J. Lee, *J. Org. Chem.*, **55**, 863 (1990).
7. Treatment of **1a** with thionyl chloride (1.2 equiv) in the presence of 2 equiv triethylamine at 0 °C for 10 min provided the cyclic sulfite **2a**, which on exposure to 2 equiv LDA in THF at -78 °C for 30 min afforded **4a** in 80 % overall yield.
8. When the compound **1a** was treated with 3 equiv NaH and excess dimethyl carbonate, **4a** was obtained directly.
9. H. Suemune, T. Harabe, and K. Sakai, *Chem. Pharm. Bull.*, **36**, 3632 (1988).
10. Y. Le Merrer, C. Gravier-Pelletier, J. Dumas, and J. C. Pepezay, *Tetrahedron Lett.*, **31**, 1003 (1990).
11. (a) L. De Montarby, P. Mosset, and R. Gree, *Tetrahedron Lett.*, **29**, 3937 (1988). (b) I. Tranchepain, F. Le Berre, A. Dureault, Y. Le Merrer, and J. C. Depezay, *Tetrahedron.*, **45**, 2057 (1989).
12. Dehydrations are normally carried out in two steps by first activating as acetate or sulfonate of the hydroxyl and then effecting elimination with base. In the case of β -hydroxy lactone, one-pot dehydration by treatment with mesyl chloride in the presence of triethylamine is known, see : (a) J. E. Worbel and B. Ganem, *J. Org. Chem.*, **48**, 3761 (1983). (b) S. Takano, M. Morimoto, and K. Ogasawara, *Synthesis*, 834 (1984).

(Received in UK 23 October, 1991)