

Experimental Section<sup>†</sup>

**Condensations with Indole-3-acetic Acid Hydrazide.**—Equimolar quantities of indole-3-acetic acid hydrazide and the appropriate carbonyl compounds were dissolved in a minimum of EtOH and heated on a steam bath for 30 min. After cooling, and in some cases standing for several days the products described in Table I were obtained by filtration.

**Condensations with Amines.**—In a similar manner equimolar quantities of isatin, indole-3-carboxaldehyde, or 1-benzylindole-3-carboxaldehyde were allowed to react in EtOH with the appropriate amines to give the compounds in Table II.

**Indole-3-acetic Acid Hydrazide and Succinic Anhydride.**—A mixture of 1.89 g (0.01 mole) of indole-3-acetic acid hydrazide and 1.00 g (0.01 mole) of succinic anhydride in Me<sub>2</sub>CO (5 ml) was refluxed for 15 min and allowed to stand overnight at room temperature. Filtration gave 2.30 g (80%) of I, mp 203–204° from EtOH; ir(KBr): 3400, 3240, 2945, 1700 (broad), 1610 cm<sup>-1</sup>. *Anal.* (C<sub>14</sub>H<sub>15</sub>N<sub>3</sub>O<sub>4</sub>): C, H.

**Tolualdehyde Mustard and 3-Aminocarbazole.**—A mixture of 1.82 g (0.01 mole) of 3-aminocarbazole and 2.60 g (0.01 mole) of 4-[bis-(2-chloroethyl)amino]-*o*-tolualdehyde was refluxed in EtOH to give 3.22 g (76%) of imine, mp 188° from EtOH. *Anal.* (C<sub>24</sub>H<sub>23</sub>Cl<sub>2</sub>N<sub>3</sub>): N.

(†) Analyses by Spang Microanalytical Laboratory, Ann Arbor, Mich. All melting points were taken in capillaries and are corrected.

## Studies of the Chemistry of Azole Derivatives.

## XII. Possible Anticonvulsant

Thiazolo[3,2-*a*]benzimidazoles

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In view of the potent pharmacological activity of a large number of heterocyclic thioureas<sup>1–3</sup> additional thioureidothiazolo[3,2-*a*]benzimidazoles were synthe-

## Experimental Section

**2-Aminothiazolo[3,2-*a*]benzimidazol-3-(2*H*)-one.**—A solution of thiazolo[3,2-*a*]benzimidazol-3(2*H*)-one<sup>4</sup> (5 g) in AcOH (20 ml) was slowly added at 0° to a solution of PhN<sub>2</sub>Cl with stirring. The mixture was kept for 1 hr at (0–5°) and the product obtained was crystallized from EtOH. The azo compound (5 g) was dissolved in hot EtOH (25 ml). A solution of Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub> (25 g) in H<sub>2</sub>O (50 ml) was added and the mixture was refluxed for 30 min and then cooled. The amino compound obtained was recrystallized from EtOH, yield 57%, mp 185°. *Anal.* (C<sub>9</sub>H<sub>7</sub>N<sub>3</sub>S): N, S.

**Synthesis of Thioureas.**—Equimolecular quantities of 2-aminothiazolo[3,2-*a*]benzimidazol-3(2*H*)-one and an aryl isothiocyanate were refluxed in abs EtOH for 5 hr and cooled. The precipitated thioureas were crystallized (C<sub>6</sub>H<sub>6</sub>). The hydrochlorides were prepared in Et<sub>2</sub>O solution.

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(4) Part XI, J. M. Singh, *J. Med. Chem.*, **12**, 962 (1969).

## A Reinvestigation of the Reaction of Monosodium Urea with Various Substituted Pvrzinecarboxylate Esters

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We have reinvestigated the reaction of monosodium urea with methyl 3,5-diamino-6-chloropyrazinecarboxylate<sup>1</sup> as well as the 5-methylamino analog and found that a small amount of the desired *N*-carbamoylpyr-

TABLE I  
2-AMINOTHIAZOLO[3,2-*a*]BENZIMIDAZOL-3(2*H*)-ONETHIOUREA HYDROCHLORIDES

No.	R	Formula	Mp, °C	Yield, %	Activity	LD <sub>50</sub> (toxicity)
1	Ph	C <sub>6</sub> H <sub>13</sub> ClN <sub>4</sub> S <sub>2</sub>	220–221	60	++	200
2	<i>o</i> -MePh	C <sub>17</sub> H <sub>15</sub> ClN <sub>4</sub> S <sub>2</sub>	195–197	65	++	260
3	<i>p</i> -MePh	C <sub>17</sub> H <sub>15</sub> ClN <sub>4</sub> S <sub>2</sub>	175–177	59	++	240
4	<i>m</i> -MePh	C <sub>17</sub> H <sub>15</sub> ClN <sub>4</sub> S <sub>2</sub>	198–200	58	++	280
5	<i>o</i> -BrPh	C <sub>16</sub> H <sub>12</sub> ClBrN <sub>4</sub> S <sub>2</sub>	210	60	+++	300
6	<i>p</i> -BrPh	C <sub>16</sub> H <sub>12</sub> ClBrN <sub>4</sub> S <sub>2</sub>	165–166	65	+++	280
7	<i>m</i> -BrPh	C <sub>16</sub> H <sub>12</sub> ClBrN <sub>4</sub> S <sub>2</sub>	189	58	+++	290
8	<i>o</i> -ClPh	C <sub>16</sub> H <sub>12</sub> Cl <sub>2</sub> N <sub>4</sub> S <sub>2</sub>	202–204	62	++++	300
9	<i>p</i> -ClPh	C <sub>16</sub> H <sub>12</sub> Cl <sub>2</sub> N <sub>4</sub> S <sub>2</sub>	211	59	++++	350
10	<i>m</i> -ClPh	C <sub>16</sub> H <sub>12</sub> Cl <sub>2</sub> N <sub>4</sub> S <sub>2</sub>	215	61	++++	330

<sup>a</sup> All new compounds were analyzed for N, S and the analytical values were within ± 0.4% of the calculated values. <sup>b</sup> Mice were used for the experiments for anticonvulsant activity following the method in Putnam and H. H. Merritt, *Science*, **85**, 525 (1937). A +++ rating was given if the convulsive threshold is elevated more than 60 ma, +++ is raised to 60 ma, +++ is raised by 40 ma, ++ is raised by 15–20 ma and + is raised by 10–15 ma, 3.5 hr after treatment.

sized. These compounds have been tested for anticonvulsant activity (Table I).

(1) L. Goldman, U. S. Patent 2,617,804, *Chem. Abstr.*, **48**, 2124 (1954).

(2) T. N. Ghosh and A. R. Chaudhuri, *J. Indian Chem. Soc.*, **28**, 268 (1951).

(3) H. P. Kautmann and P. Schultz, *Arch. Pharm.*, **273**, 22 (1935).

azinecarboxamide is produced in each case. The products were isolated by liquid-liquid partition chromatography (llpc) and shown to be the desired compounds by comparison of ir, mass spectrum, tlc,

(4) J. William Hanifin, Rosemary Capuzzi, and Elliott Cohen, *J. Med. Chem.*, **12**, 1102 (1969).