Photoswitching off the Antiproliferative Activity of Combretastatin A-4 Analogues

Anton V. Yadykov,[†] Alexander M. Scherbakov,[‡] Victoria V. Trofimova,[§] Andrey G. Lvov,[†] Ashot I. Markosyan,^{||} Igor V. Zavarzin,[†] and Valerii Z. Shirinian^{*,†}

[†]N. D. Zelinsky Institute of Organic Chemistry, RAS, 47, Leninsky prosp., 119991 Moscow, Russian Federation [‡]N. N. Blokhin National Medical Research Center of Oncology, Kashirskoye sh. 24, 115522 Moscow, Russian Federation [§]D.I. Mendeleev University of Chemical Technology of Russia, Moscow 125047, Russian Federation Scientific Technological Center of Organic and Pharmaceutical Chemistry, NAS RA, Yerevan 0014, Republic of Armenia

S Supporting Information

ABSTRACT: The photostability and antiproliferative activity of combretastatin A-4 (CA-4) analogues against human epidermoid carcinoma cells A-431 were studied. For the first time, it was shown that UV or sunlight irradiation of furanone analogues of CA-4 results in a photorearrangement giving products with relatively low antiproliferative activity. The observed ability of this series CA-4 to the photodegradation can be used for the design of a new class of drug candidates with high selectivity to cancer cells.

Diarylethenes occupy an important place among the photoswitchable compounds, which are extensively studied for the design of smart materials and devices.^{1,2} On the other hand, CA-4 analogues, being diarylethene derivatives, are of interest as biologically active compounds acting as inhibitors of tubulin polymerization and suppressing micro-tubule formation.³⁻⁵ CA-4 analogues occupy a special place among the ligands capable of binding to the colchicine-binding site of tubulin and inhibiting proliferation of cancer cells.⁶ Due to their simple structures and high biological activity, these compounds are extensively studied as effective anticancer drug candidates.⁹⁻¹¹ Some CA-4 derivatives, such as CA-4P, ombrabulin, and podophyllotoxin, and also several compounds of the colchicine and allocolchicine series, are currently in clinical trials.4,12

Despite considerable research on the anticancer activity of CA-4 analogues, $^{13-15}$ the photostability of these compounds (resistance to UV light or sunlight) is poorly known,^{16,17} although this property is an important factor in drug design. Since heterocyclic analogues of CA-4 have attracted great interest and are potential drug candidates, the aim of this work is to study the photodegradation of new heterocyclic CA-4 analogues of the furanone series $^{18-20}$ and to evaluate their antiproliferative activity before and after exposure to UV light or sunlight.

Furanone analogues of CA-4 (1a-g) comprising a 3,4,5trimethoxyphenyl moiety as the ring A and various fivemembered heterocycles or 4-methoxyphenyl residue as the ring B were synthesized according to the known method²¹ in

lity of CA-4 analogues · Photo-switching off antiproliferative activity Solar chemical transformations

good yields (Scheme S1, Table 1). Photochemical studies of these compounds showed that as opposed to photochromic diarylethenes,²² they undergo an irreversible photorearrangement giving naphthalene derivatives (Scheme 1).²³ The photoirradiation of CA-4 analogues was carried out in Nmethyl-2-pyrrolidone (NMP) (see Scheme S2). The photorearrangement of these diarylethenes provides naphthalene derivatives with different functional groups (Table 1).

Evaluation of the effectiveness of this phototransformation was carried out using UV and NMR spectroscopy. The spectral properties of the initial analogues of CA-4 and the products of photochemical reactions were studied in acetonitrile upon irradiation at 365 nm (for UV-vis spectra of 1a-g and 2a-g, see Figures S1-S15). The dynamics of spectral alterations under irradiation of diarylethenes solutions are characterized by the presence of two isosbestic points that is indicative of a high rate of a 1,9-sigmatropic rearrangement with subsequent oxazole ring opening and the absence of the contribution of intermediates to the measured stationary absorption spectra.² Therefore, the absorption spectra measured under irradiation of diarylethenes 1a-g are superpositions of solely two forms, namely, initial 1 and the product 2 (Scheme 1).

The high efficiency of the photorearrangement was demonstrated by ¹H NMR monitoring. Figure 1 presents the results of monitoring for diarylethene 1c. The formation of other byproducts associated with the photocyclization/

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Table 1. Structures and Yields of CA-4 Analogues and Their Photoproducts



Scheme 1. Photorearrangement of CA-4 Analogues



oxidation or photocyclization/elimination was virtually not observed, as can be clearly seen by following the changes in the position of a signal of methoxy groups. As can be seen in Figure 1, the signal of methyl group in the naphthalene system



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Figure 1. ¹H NMR monitoring of the photoirradiation of diarylethene **1c**.

appears at δ 2.80 ppm with the disappearance of the starting diarylethene at δ 1.92 ppm.

The ¹H NMR monitoring of **1a** upon irradiation with UV light was also carried out (see Figure S16). As opposed to the heterocyclic analogues, furanone **1a**, comprising a 4-methoxyphenyl residue, gives three photocyclization products upon UV irradiation: phenanthrene derivative **2a**' and two E/Z isomers of photorearrangement product **2a**'' (Scheme 2).²⁵

Scheme 2. Photoreaction of Diarylethene 1a



The photoreaction affords phenanthrene derivative 2a' as the major product that is formed via the Mallory reaction (tandem photocyclization/oxidation reaction, Figure S16). The formation of the phenanthrene derivative is confirmed by the presence of a characteristic doublet at δ 9.03 ppm with a small spin-spin coupling constant (2.4 Hz). Two isomeric products E/Z-2a'' are generated via a tandem reaction involving the photocyclization of the hexatriene system followed by a rearrangement and benzene ring opening. These compounds are characterized by spin-spin coupling constants of olefinic protons (16.0 and 12.5 Hz for the E and Zisomers, respectively, Figure 3). It should be noted that the maximum formation of butadiene derivatives 2a'' (Z- and Eisomers) is observed when the reaction mixture is irradiated for 2 h; further irradiation leads to the disappearance of these compounds and only phenanthrene 2a' remains (Figure S16).

Thus, the photochemical studies of CA-4 analogues of the furanone series showed that the compounds containing fivemembered heterocycles as the ring B are very sensitive to UV light and undergo a rearrangement giving naphthalene derivatives. Diarylethenes comprising the 4-methoxyphenyl moiety as the ring B undergo either a tandem photocyclization/oxidation reaction (Mallory reaction) giving a phenanthrene 2a' or a tandem photocyclization/rearrangement reaction accompanied by the benzene ring opening to form an E/Z isomer mixture of butadienes 2a''. For a comparative evaluation of the antiproliferative activity of CA-4 analogues and their photoproducts, analytically pure samples were prepared.

Combretastatin A-4 analogues and their photoproducts were examined for their effect on viability of human epidermoid carcinoma cells A-431. The activity was evaluated in the micromolar concentration range from 0.7 to 15 μ M. Six pairs of compounds (CA-4 analogues and their photoproducts) were tested (Table 2). Two CA-4 analogues, compounds 1a

Table 2. Effect of the Synthesized Compounds on A-431Cell Viability (72 h Growth in the Presence of Compounds)

entry	compd	IC ₅₀ (µM)	cell viability (15 μM treatment) (%)
	DMSO (vehicle control)		99
1	1a	<0.7	9
2	1b	>15	60
3	2b	>15	81
4	1c	<0.7	10
5	2c	>15	77
6	1d	>15	41
7	2d	>15	77
8	1e	>15	44
9	2e	>15	11
10	1f	>15	22
11	2f	>15	82
12	1g	>15	74
13	2g	>15	57
14	cisplatin	6.3 ± 0.9	25

and 1c, exhibited high activity against A-431 cells (IC₅₀ of these compounds were less than 0.7 μ M). These compounds were selected for more detailed investigation. Both compounds contain the 4-methoxyphenyl moiety as the ring B, which is apparently responsible for their high activity.²⁶ The antiproliferative properties of 1a and 1c are much better compared to the reference cisplatin, and the latter shows an IC₅₀ of about 6 μ M. Naphthalene 2c obtained from compound 1c via a photoreaction exerted a weak effect on cell viability and, when used at a concentration of 15 μ M, insignificantly inhibited the cell growth. A similar, but less pronounced, decrease in activity was observed for photoproduct 2d as compared to starting 1d.

On the contrary, photoproduct **2e** proved to be more active that the starting compound. CA-4 analogue **1f** exhibited high activity, and the cell viability in the presence of this compound at a concentration of 15 μ M was about 20%. Its photoproduct **2f** showed much lower antiproliferative activity. Interesting results were obtained for **1b**. Thus, the replacement of the peripheral 4-methoxyphenyl substituent in compound **1c** by the phenyl group led to a significant loss in the activity of both diarylethene **1b** and its photoproduct **2b**. These compounds did not show a pronounced antiproliferative effect on A-431 cells; at a concentration of 15 μ M, they induced less than 40% cell death. Compounds **1g** and **2g** also did not show significant antiproliferative effects. In this pair, photoproduct **2g** is more active than the starting diarylethene **2g**. Additional assays were performed for more active compounds.

Thus, the photobiological studies demonstrated that CA-4 analogues containing the 3,4,5-trimethylphenyl group as the ring A and different five-membered heterocycles as the ring B are highly susceptible to UV irradiation and undergo photodegradation. Most photodegradation products exhibit relatively low activity compared to the starting diarylethenes.

For a more detailed study and dynamic evaluation of biological activity, we selected two compounds (1a and 1c) displaying activity in the submicromolar concentration range. To determine the dependence of antiproliferative activity of these compounds on UV irradiation, we evaluated the activity before and after irradiation for 2 and 4 h. Irradiation was performed before treatment of A-431 cells. As can be seen in Figure 2, in the concentration range under study, these



Figure 2. Effect of compounds **1a** (A) and **1c** (B) on the viability of the A-431 cells (72 h cell growth in the presence of compounds), w/o UV, with UV (λ = 365 nm) for 2 or 4 h.

compounds did not exhibit inhibitory activity against A-431 cell growth after 2 h irradiation. The UV irradiation of the compounds 1a and 1c for 4 h also led to the loss of antiproliferative activity. As can be seen in Figure 2 and Table S1, these compounds did not exhibit inhibitory activity against A-431 cell growth after 2 h irradiation. The UV irradiation of the compounds 1a and 1c for 4 h also led to the loss of antiproliferative activity.

Since light sensitivity of biologically active compounds is an important factor in the efficient drug design, 16,27,28 we have studied the sunlight-induced photodegradation of two diary-lethenes. Solutions of CA-4 analogues in DMSO- d_6 were monitored by ¹H NMR spectroscopy upon exposure to sunlight (Figure 3, the experiment was performed at the latitude of Moscow at room temperature in April 2019).

The exposure to sunlight was found to induce processes similar to those caused by UV irradiation. Furanone 1a gives three products, including the phenanthrene analogue. The only difference is the reaction time. As expected, upon exposure to sunlight, the reaction is not completed even within 4 days, whereas under UV irradiation, the reaction proceeds within 4 h (compare Figure 3 with Figure S16). A similar pattern was observed for diarylethene 1c (Figure S17). The complete conversion giving a photocyclization/rearrangement product was achieved also after irradiation for 4 days.

To summarize, we have synthesized combretastatin A-4 analogues of the furanone series comprising a 3,4,5-trimethoxyphenyl derivative as the ring A and various fivemembered heterocycles as the ring B. Photochemical and antiproliferative properties of the synthesized compounds and





their photoproducts were studied. When exposed to UV light, these compounds undergo a photorearrangement giving photoproducts with relatively low antiproliferative activity. The antiproliferative activity of the most active compounds (1a and 1c) against A-431 cells was evaluated before and after UV irradiation at λ = 365 nm. The irradiation was found to cause a significant decrease in antiproliferative activity. For the first time, it was demonstrated that these compounds undergo similar sunlight-induced photodegradation. To conclude, it can be noted that, despite relatively high antiproliferative activity, CA-4 analogues undergo photodegradation, with loss of activity, under exposure to sunlight (UV light). Investigations of the effect of sunlight (UV light) on the photodegradation and antiproliferative activity of other CA-4 analogues will be continued. The observed ability of this series compounds to undergo photodegradation can be used for the design of a new class of drug candidates with high selectivity to cancer cells, which are particularly interesting for the treatment of skin tumors. The neutralization of toxic (adverse) effects on the surrounding normal tissues by means of irradiation will help in designing new drugs with improved therapeutic profiles (high selectivity/low overall and organ-specific toxicity).

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.9b03780.

Detailed experimental procedures, compound characterization data, UV-vis, HRMS, and NMR spectra, and biological assays (PDF)

AUTHOR INFORMATION

Corresponding Author

*E-mail: shir@ioc.ac.ru

ORCID 💿

Andrey G. Lvov: 0000-0003-2951-2651 Valerii Z. Shirinian: 0000-0001-9480-3565

Notes

The authors declare no competing financial interest.

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