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Synthesis and Acaricidal/Insecticidal Activities Evaluation of Novel Oxazolines Containing Sulfiliminyl Moieties and Derivatives

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1 ABSTRACT

Sulfimides and sulfoximines are highly relevant for medicinal chemistry and crop 2 3 protection, as the resulting products can reveal interesting bioactivities. Herein, we report the design and synthesis of a series of novel 2,4-diphenyl-1,3-oxazolines 4 containing sulfiliminyl and sulfoximinyl moieties. The acaricidal and insecticidal 5 activities of the new compounds were evaluated and indicated that these compounds 6 exhibited excellent acaricidal activities against spider mite larvae and eggs. The LC₅₀ 7 values of 6a-7, 6b-3, 6b-4, 6c-2 and 6c-4 against larvae of spider mite were about 4 to 8 9 6 times lower than commercial insecticide etoxazole (0.0221 mg L⁻¹), and the LC₅₀ value of **6a-4** against eggs of spider mite was 0.0006 mg L⁻¹, which was ten times 10 lower than etoxazole (0.0063 mg L^{-1}). At the same time, most of the compounds 11 12 showed insecticidal activity though their structure-activity relationships were different. Oxazolines containing N-cyano sulfiliminyl moiety at the *para* position of 4-phenyl 13 group exhibited better insecticidal activities against cotton bollworm and corn borer 14 15 than etoxazole, while the compounds containing groups derived from sulfiliminyl and sulfoximinyl had weak insecticidal activities. This research again proved that the 16 substituent type at the para site of 4-phenyl moiety have a decisive role on its 17 biological activity and insecticidal spectrum. 18

19 KEYWORDS: 2,4-diphenyl-1,3-oxazoline, etoxazole, sulfimides, sulfoximines
20 acaricidal/insecticidal activity, structure-activity relationship

- 21
- 22

2

23 INTRODUCTION

24 Phytophagous mites are common pests on agricultural production. There are about 25 dozens of phytophagous mites in our country, including the most destructive 26 *Tetranychus urticae (Koch)*, and *Panonychus citri (Mcgregor)*; they mainly cause 27 damage to crops such as cottons, vegetables, fruit trees, tea trees and wheats. Mites 28 not only feed on plant juices, but also spread plant pathogens and viruses thus 29 resulting in huge decrease in agricultural production.¹

Now, the application of acaricides is still one of the effective methods to prevent and control agricultural spider mites. But the frequent use and misuse of acaricide caused more and more resistance of the mites.² In order to deal with the growing insect resistance problem, low toxic, highly efficient and environment friendly acaricides with unique structure and novel mechanism of action are still needed to continuously introduce to the market, which also is a difficult task and huge challenge for pesticide chemists.

37 Sulfimides ($R^1R^2S=NR$) and sulfoximines ($R^1R^2S(O)=NR$) are mono-aza analogues of sulfoxides and sulfones, respectively. Sulfoximines have been applied with great 38 success in asymmetric synthesis,³ and more recently, much attention has been focused 39 on their bioactive profiles in crop protection and drug development.³⁻⁷ For example, 40 sulfoximines can be modified at the nitrogen atom, which allows improving the 41 solubility of the corresponding molecules.^{8,9} Sulfoxaflor (Figure 1) aroused our 42 43 interests in that it has a sulfoximine group linked to N-cyano and exhibited excellent insecticidal activities.¹⁰ Sulfilimines, possessing a nitrogen substituent and a free 44

electron pair at the sulfur atom, are intermediates in the sulfoximine synthesis and 45 have shown interesting insecticidal activities.^{6b,11} Furthermore, sulfilimines should 46 also be similar in properties to the corresponding sulfoximines but they have not been 47 extensively examined for use in agrochemical research, except that some sulfiliminyl 48 derivatives were reported to exhibit herbicidal properties.¹² The sulfilimine moiety 49 was postulated to have a small hydrophilic core as well as a hydrogen bond acceptor 50 site, thus when it was introduced into compounds, the novel structures might bring us 51 unknown, interesting bioactivities. 52

53 Etoxazole (Figure 1) is the only commercial acaricides/insecticides belonging to the chemical class of 2,4-diphenyl-1,3-oxazolines. The acaricidal and insecticidal 54 structure-activity relationships of etoxazole analogues showed that the substituent 55 56 type at the *para* site of 4-phenyl moiety has a decisive role on their biological spectrum.^{13,14} activities and insecticidal In 2015, we reported 57 2,4-diphenyl-1,3-oxazolines containing sulfur ether moiety which exhibited excellent 58 acaricidal activities.¹⁵ In consideration of the above viewpoints, a series of 59 2,4-diphenyl-1,3-oxazolines were designed by introducing the sulfiliminyl and 60 sulfoximinyl group into the para position of 4-phenyl moiety, and the synthesis are 61 shown in Figure 1. Their acaricidal/insecticidal activities were evaluated accordingly, 62 and the structure-activity relationships were discussed. 63

64 MATERIALS AND METHODS

Instruments. Reaction progress was monitored by thin-layer chromatography on
 silica gel GF254 with ultraviolet (UV) detection. Melting points were obtained using

67	an X-4 binocular microscope melting point (mp) apparatus and are uncorrected.
68	Yields were not optimized. ¹ H-NMR spectra and ¹³ C-NMR spectra were recorded
69	utilizing a Bruker AV400 spectrometer with CDCl3 as solvent and tetramethylsilane
70	as internal standard. Chemical shifts (δ) were given in parts per million (ppm).
71	High-resolution mass spectra (HRMS) data were obtained with a Fourier transform
72	ion cyclotron resonance mass spectrometry (FTICR-MS) spectrometer (ionspec, 7.0T)
73	InfraRed (IR) was conducted with a MAGNA-560 FTIR (Nicolet).
74	General Synthesis. The reagents were all analytically or chemically pure
75	purchased from commercial sources and were used as received. All anhydrous
76	solvents were dried and distilled by standard techniques just before use. All the
77	sulfides 5 were synthesized according to the method reported in our previous
78	publication. ¹³ The synthetic route is given in Figure 2.
79	General Synthetic Procedure for the Target Compounds 6a, 6b, 6c, 6d,
80	and 6e (Figure 2).
81	Synthesis of
82	2-(2,6-difluorophenyl)-4-(4-((naphthalen-2-yl-N-cyanosulfilimidoyl)methyl)phenyl)-
83	4,5-dihydrooxazole (6a-1). A mixture of

2-(2,6-difluorophenyl)-4-(4-((naphthalen-2-ylthio)methyl)phenyl)-4,5-dihydrooxazole
(0.2 g, 0.46 mmol) and cyanamide (0.02 g, 0.51 mmol) in 20 mL of acetonitrile was
cooled below 0 °C. To this solution was added iodobenzene diacetate (0.15 g, 0.46
mmol) all at once. The reaction mixture was allowed to stir below 0 °C for 10 min
and slowly warmed to room temperature; the progress of the reaction was monitored

by TLC. Excess oxidant was destroyed by adding 5 mL of 2.5% aqueous sodium hydrogen sulfite, and then the solution was added dichloromethane and water. The aqueous phase was separated and then extracted with dichloromethane twice. The combined organic layer was washed with saturated brine, dried over anhydrous sodium sulfate, and filtered. After the solvent was removed in vacuo, the residue was purified by column chromatography on silica gel to give the target compound **6a-1**.¹⁶⁻¹⁹

Compounds **6a-2–6a-10** were prepared according to the method used for compound **6a-1**. The physical data in detail are included in the Supporting Information.

100 2-(2,6-difluorophenyl)-4-(4-(p-tolyl-N-trifluoroacetylsulfilimidoylmethyl)phenyl)-4,5 -dihydrooxazole То solution of (**6b-2**). 101 а 2-(2,6-difluorophenyl)-4-(4-(p-tolyl-N-cyanosulfilimidoylmethyl)phenyl)-4,5-dihydro 102 oxazole (6a-2) (0.50 g, 1.15 mmol) in 20 mL dichloromethane, trifluoroacetic 103 anhydride (TFAA) (0.72 g, 3.44 mmol) was added dropwise at room temperature. 104 Then the reaction mixture was stirred continuously until the raw material disappeared 105 (monitored by TLC). The resulting mixture was washed with water, the organic phase 106 was separated, and the aqueous phase was extracted by dichloromethane (2 ×20 mL). 107 The combined organic layer was washed with saturated brine and dried with 108 anhydrous sodium sulfate, then filtered and concentrated under reduced pressure. The 109 residue was purified by column chromatography on silica gel and eluted with 110

111	petroleum ether and ethyl acetate (v/v = 3:1) to give the target compound 6b-2 . ²⁰
112	Compounds 6b-1, 6b-3-6b-7, 6b-9 were synthesized by a method similar to that
113	used for compound 6b-2. Their physical data are included in the Supporting
114	Information.
115	Synthesis of 2-(2,6-difluorophenyl)-4-(4-(p-tolyl-NH-sulfoximidoyl
116	methyl)phenyl)-4,5-dihydrooxazole (6c-2). A solution of
117	2-(2,6-difluorophenyl)-4-(4-(p-tolyl-N-trifluoroacetylsulfilimidoylmethyl)phenyl)-4,5
118	-dihydrooxazole (6b-2) (0.42 g, 0.83 mmol) in 20 mL ethanol was added
119	3-chloroperoxybenzoic acid (m-CPBA) at 0 °C. Potassium carbonate (0.34 g, 2.49
120	mmol) aqueous solution was added dropwise to the above mixture. The progress of
121	the reaction was monitored by TLC until the reaction was complete. The reaction
122	mixture was concentrated under reduced pressure. The resulting mixture was added
123	dichloromethane and water, the organic phase was separated, and the aqueous phase
124	was extracted by dichloromethane (2 \times 20 mL). The combined organic layer was
125	washed with saturated brine, dried with anhydrous sodium sulfate, and then filtered.
126	The solvent was removed in vacuo, and the residue was purified by silica gel column
127	chromatography to give the target compound $6c-2$. ²¹
128	The synthesis procedure of 6c-1, 6c-4 and 6c-5 was similar to that of compound
129	6c-2. The physical data in detail are included in the Supporting Information.
130	Synthesis of 2-(2,6-difluorophenyl)-4-(4-(N-cyanosulfoximidoyl
131	methyl)phenyl)-4,5-dihydrooxazole (6d). The synthesis procedure of compounds
132	6d-1-6d-5, 6d-8, 6d-9 was similar to that of compound 6c-2 using intermediate

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6a-1-6a-5, 6a-8, 6a-9, respectively, as materials. Their physical data are included in
the Supporting Information.

Synthesis of 2-(2,6-difluorophenyl)-4-(4-(p-tolyl-N-carbamoylsulfoximidoyl 135 methyl)phenyl)-4,5-dihydrooxazole (**6e-2**). То solution of 136 а 2-(2,6-difluorophenyl)-4-(4-(p-tolyl-N-cyanosulfoximidoylmethyl)phenyl)-4,5-dihydr 137 ooxazole (6d-2) (0.44 g, 0.97 mmol) in 20 mL dichloromethane, trifluoroacetic acid 138 (0.33 g, 2.91 mmol) was added dropwise at 0 °C. The reaction mixture was stirred at 139 0 °C until the reactant disappeared (monitored by TLC). The resulting mixture was 140 141 washed with water. The organic phase was separated and then the aqueous phase was extracted with dichloromethane twice. The combined organic layer was washed with 142 saturated brine, dried with anhydrous sodium sulfate, and then filtered. The solvent 143 144 was concentrated in vacuo, and the residue was purified by silica gel column chromatography to give the target compound 6e-2.²¹ 145

The synthesis procedure of 6e-1, 6e-3–6e-5 was similar to that of compound 6e–2.
The physical data in detail can be found in the Supporting Information.

Biological Assay. Detailed bioassay procedures for spider mites²²⁻²⁴ and various insects^{25,26} were carried out according to the published literature and can also be found in the Supporting Information. According to statistical requirements each bioassay was repeated at least three times. The error of the experimments was 5%. Assessments were made on a dead/alive basis, and mortality rates were corrected using Abbott's formula. Evaluation were based on a percentage scale of 0–100, where 0 equals no activity and 100 equals total kill. For comparative purpose, etoxazole was tested under the same conditions.

156 RESULT AND DISCUSSION

157 Chemistry

Synthesis. A series of sulfides 5 were initially prepared according to the 158 previously reported literature in moderate yields. The treatment of the sulfides with 159 cyanamine as a nitrogen source and iodobenzene diacetate as an oxidant in 160 acetonitrile led efficiently to corresponding N-cyanosulfilimines 6a in considerable 161 yields. However, when there is a heterocyclic ring connected with the sulfur atom in 5, 162 163 most of the oxidation step could not occur; only one compound 6a-10 was obtained, in which the sulfur atom was connected with a thiadiazole ring. It is worth mentioning 164 that these novel compounds are much stable, as they are easily handled and stored for 165 166 weeks at room temperature without degradation.

Then the N-cyanosulfilimidoyl group of **6a** was derived to provide more diverse groups. The replacement of cyano of the sulfiliminyl moiety with trifluoroacetyl group could bring changes in physical, chemical, and biological properties of the whole molecule.²⁵⁻³⁰ So firstly N-trifluoroacetyl sulfilimines **6b** were synthesized by treating **6a** with trifluoroacetic anhydride (TFAA) at room temperature. But the product **6b-8** was not obtained using this method. This is probably because the steric hindrance of tertiary butyl is unfavorable factor for the synthesis.

Subsequently, we conducted the oxidation reaction of **6b**. Surprisingly, the oxidation of **6b** with m-CPBA didn't afford the desired N-trifluoroacetylsulfoximines, but afforded sulfoximine compounds **6c**. The compounds **6b** bearing a 4-phenyl group at the sulfur atom all converted to the corresponding sulfoximines (6c-1, 6c-2, 6c-4
and 6c-5), while both compounds 6b-6 (bearing a 2-methoxylphenyl group) and 6b-7
(bearing a 2-bromophenyl group) were not oxidized under the synthetic procedure
because of the ortho steric hindrance.

Direct oxidation of **6a** with m-CPBA could afford N-cyanosulfoximines **6d**. Interestingly, the oxidation of **6a-6** and **6a-7** gave the same compound: **4-(2-(2,6-difluorophenyl)-4,5-dihydrooxazol-4-yl)benzaldehyde**; it was reported that the oxidation at sulfur is hindered by the adjacent ortho substituent and the removal of the NCN group occurs in an unusual reductive transformation.²¹ Finally, N-cyanosulfoximines **6d** were treated with trifluoroacetic acid (TFA) in dichloromethane under an ice bath to produce N-acylaminosulfilimines **6e**.

Structure. The novel structures of the target compounds have been identified by ¹H NMR, ¹³C NMR and HRMS. Since there are four different groups connected with sulfur atom in each of the structures, the two hydrogen atoms on the methylene linked to sulfur atom are chemical non-equivalent atoms and therefore give different chemical shift in ¹H-NMR spectrum. Correspondingly, these compounds also exhibit two sets of peaks in ¹³C-NMR spectrum.

Biological Activity and Structure-Activity Relationship. *Activities against Spider Mite (T. cinnabarinus) Eggs and Larvae*. The acaricidal activities of the compounds **6a-6e** and etoxazole (as control) against the eggs and larvae of spider mite (*T. cinnabarinus*) were tested and the data are listed in Table 1. Most oxazolines **6a** (containing N-cyanosulfilimidoyl group) exhibited good to

199	excellent ovicidal and larvicidal activities. Compounds 6a-1 (containing
200	N-cyano-S-naphthylsulfilimidoyl group), 6a-2-6a-5 (containing
201	N-cyano-S-(4-substituted phenyl)sulfilimidoyl group), and 6a-8 (containing
202	N-cyano-S-t-butylsulfilimidoyl group) showed comparable ovicidal activities to that
203	of etoxazole. Especially, the mortality of 6a-4 (containing a
204	N-cyano-S-(4-fluorophenyl)sulfilimidoyl group) was 86% at the concentration of 0.01
205	mg/L, but etoxazole showed only a mortality of 67% at the same concentration.
206	Further test showed the LC_{50} value of 6a-4 (0.0006 mg L ⁻¹) was about ten times lower
207	than etoxazole (0.0063 mg L ⁻¹) (Table 2). The ovicidal activities of compound 6a-7
208	(containing N-cyano-S-2-brominephenyl sulfilimidoyl) were much better than that of
209	compound 6a-6 (containing N-cyano-S-2-methoxylphenyl sulfilimidoyl), and both of
210	them had higher larvicidal activities than etoxazole.
211	Other oxazolines (6b-6e) derived from corresponding 6a also exhibited excellent
212	acaricidal activities, but with no obvious structure-activity relationships.
213	In N-trifluoroacetylsulfilimines 6b, the larvicidal activities of compounds 6b-3
214	(containing 4-chlorophenyl) on sulfur atom and 6b-4 (containing 4-flluorophenyl)

(containing 4-chlorophenyl) on sulfur atom and 6b-4 (containing 4-flluorophenyl)
were much better than etoxazole, and 6b-2, 6b-5, 6b-7 and 6b-9 also displayed good
larvicidal activity. Meanwhile, the ovicidal activities of 6b-4 and 6b-7 (containing
2-brominephenyl) on sulfur atom were as good as etoxazole. The activities of
NH-sulfoximines 6c against mite eggs were not as good as etoxazole, but the
activities against mite larvae of compounds 6c-2 (containing 4-methylphenyl on sulfur
atom) and 6c-4 (containing 4-flluorophenyl) were higher than etoxazole.

11

N-cyanosulfoximine 6d-4 containing 4-flluorophenyl on sulfur atom had good 221 activities against both mite eggs and larvae. The acaricidal activities of 222 223 N-carbamoyl-sulfoximines 6e-4 and 6e-5 pretty were good. N-trifluoroacetylsulfilimine 6b-4, NH-sulfoximine 6c-4, N-cyanosulfoximine 6d-4, 224 and N-carbamoyl-sulfoximine 6e-4 containing 4-fluorophenyl all showed wonderful 225 acaricidal activities against mite both eggs and larvae, while 6b-5, 6c-5, 6d-5 and 6e-5 226 containing 4-brominephenyl exhibited better larvicidal activities but weaker ovicidal 227 activities. The LC₅₀, LC₉₀ values of compounds 6a-7, 6b-3, 6b-4, 6c-2 and 6c-4 228 229 against larvae of spider mite are given in Table 3. These compounds showed lower concentrations than commercial etoxazole. LC_{50} of **6c-2** was 0.0039 mg L⁻¹, much 230 lower than etoxazole ($0.0221 \text{ mg } \text{L}^{-1}$). 231

Insecticidal Activities against Lepidopteran Pests and Mosquito Larvae. 232 The insecticidal activities of the target compounds 6a-6e and etoxazole against 233 mosquito larvae and lepidopteran pests larvae (oriental armyworm, cotton bollworm 234 and corn borer) are listed in Tables 4 and 5. As shown in Tables, most compounds 235 exhibited more or less insecticidal activities at the preliminary screening 236 concentration. The inhibitory activities of 6b, 6c, 6d, 6e series were superior to 6a 237 series and etoxazole for mosquito larvae, but were weaker than 6a series compounds 238 and etoxazole for cotton bollworm and corn borer. Compounds 6c-1, 6d-1, 6e-1, 6d-3 239 and 6c-4 exhibited excellent activities against mosquito larvae; among these 240 compounds, 6d-1 and 6e-1 showed a morality of 60% and 20% at 0.5 mg/L, 241 respectively; the moralities of 6d-3 and 6c-4 were 20% and 60% at 0.25 mg/L 242

respectively; etoxazole gave a death rate of only 20% at 2 mg/L. Compounds 6a-4
and 6a-5 displayed 50% and 40% insecticidal activities against cotton bollworm, 40%
and 60% insecticidal activities against corn borer at the concentration of 100 mg/L,
but etoxazole only showed insecticidal activities of 40% and 35% against cotton
bollworm and corn borer respectively.

In summary, by introducing N-cyanosulfiliminyl moieties into 248 2,4-diphenyl-1,3-oxazolines and further derivation steps, a series of novel oxazoline 249 derivatives with sulfilimine functional groups were designed and synthesized. Their 250 251 acaricidal activities against spider eggs and larvae, insecticidal activities against oriental armyworm, cotton bollworm, corn borer and mosquito larvae were evaluated. 252 The bioassay results of N-cyano sulfilimine oxazolines indicated that compounds 253 254 6a-1-6a-10 exhibited good to excellent acaricidal activities, among them compound 6a-4 (containing a N-cyano-S-(4-fluorophenyl)sulfilimidoyl group) displayed highest 255 morality against eggs, the LC_{50} value (0.0006 mg L⁻¹) was about ten times lower than 256 etoxazole (0.0063 mg L^{-1}); the LC₅₀ of the acaricidal activity against spider larvae of 257 compound 6a-7 (containing 2-brominephenyl on sulfur atom) was 0.0056 mg L⁻¹, 258 also much lower than etoxazole (0.0221 mg L^{-1}); especially compounds 6a-5, 6a-6, 259 6a-7 gave outstanding acaricidal activities against both spider eggs and larvae. In 6b, 260 6c, 6d, 6e series, compounds 6b-3, 6b-4, 6c-2 and 6c-4 had much better larvicidal 261 activities than etoxazole; their LC₅₀ values were 0.0058 mg L⁻¹, 0.0055 mg L⁻¹, 0.0039 262 mg L^{-1} , 0.0041 mg L^{-1} respectively. The insecticidal activities of **6a** series compounds 263 showed that the target compounds displayed excellent activities against cotton 264

265	bollworm and corn borer, but the activities against oriental armyworm and mosquito
266	larvae were weak. 6b, 6c, 6d, 6e series compounds all showed wonderful insecticidal
267	activities against mosquito larvae. It was further concluded that the present research
268	indicated that the introduction of sulfiliminyl moieties into the
269	2,4-diphenyl-1,3-oxazolines maintained or improved biological activities. Some of the
270	target compounds might be potential candidates for further investigation.
271	ASSOCIATED CONTENT
272	Supporting Information. The physical data of target compounds 6a-1-6a-10,
273	6b-1-6b-7, 6b-9, 6c-1-6c-2, 6c-4-6c-5, 6d-1-6d-5, 6d-8-6d-9, 6e-1-6e-5 can be
274	found in Supporting Information. This material is available free of charge via the
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Figure Captions

Figure 1 The Design of the Target Compounds.

Figure 2 General Synthetic Procedure for the Target Compounds 6a-6e. Reagent

and conditions: (a) PhI(OAc)₂, NH₂CN, CH₃CN, 0 °C to rt, (b) (CF₃CO)₂O, CH₂Cl₂, 0

°C to rt, (c) m-CPBA, K₂CO₃, EtOH, 0 °C to rt, (d) CF₃COOH, CH₂Cl₂, ice bath.

Figure 1





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Table 1 Acaricidal Activities of the Target Compounds against Spider Mite

comp	activities (%) against eggs at concentration	activities (%) against larvae at concentration
d	(mg L ⁻¹)	(mg L ⁻¹)

	100	50	25	2.5	1	0.1	0.01	100	50	25	2.5	1	0.1	0.01
6a-1	100	100	100	100	91	83	52	100	100	100	94	88	72	65
6a-2	100	100	100	90	83	72	61	100	100	91	84	64	48	21
6a-3	100	100	100	96	88	79	51	100	100	100	95	89	76	50
6a-4	100	100	100	100	100	100	86	100	100	100	83	67	52	_
6a-5	100	100	100	100	100	89	50	100	100	100	100	81	69	55
6a-6	100	100	90	79	67	54	0	100	100	100	100	96	88	66
6a-7	100	100	100	100	100	90	48	100	100	100	100	100	81	69
6a-8	100	100	100	100	95	88	56	100	100	100	86	78	69	54
6a-9	100	92	81	73	67	49	0	100	100	90	79	63	48	27
6a-10	100	100	92	79	66	54	36	100	100	100	92	87	75	63
6b-1	100	100	100	100	93	82	79	100	100	100	93	88	67	59
6b-2	100	100	100	95	89	77	30	100	100	100	100	86	79	61
6b-3	100	100	100	100	93	67	49	100	100	100	100	100	84	74
6b-4	100	100	100	100	100	82	71	100	100	100	100	100	81	77
6b-5	100	100	93	85	76	62	51	100	100	100	100	96	86	74
6b-6	100	100	100	90	77	61	51	100	100	91	84	75	69	49
6b-7	100	100	100	100	100	85	71	100	100	100	100	91	78	53
6b-9	100	100	100	93	85	70	43	100	100	100	100	92	84	79
6c-1	100	100	100	92	85	75	64	100	100	100	91	79	62	47
6c-2	100	100	100	93	80	76	66	100	100	100	100	100	87	70
6c-4	100	100	100	100	87	76	69	100	100	100	100	100	89	66
6c-5	100	100	100	100	92	87	63	100	100	100	100	96	81	55
6d-1	100	100	100	100	93	88	61	100	100	100	92	78	61	50
6d-2	100	100	88	72	61	49	17	100	94	86	77	61	44	31
6 d-3	100	100	100	91	82	71	64	100	100	92	80	71	63	58
6d-4	100	100	100	100	100	86	76	100	100	100	100	90	77	72
6d-5	100	100	100	100	89	74	56	100	100	100	100	93	79	59
6 d- 8	100	100	100	100	93	81	62	100	100	100	90	83	75	68
6d-9	100	100	100	87	79	60	39	100	100	100	90	83	72	60
6e-1	100	100	94	82	63	46	30	100	100	100	89	83	76	61
6e-2	100	100	100	95	83	72	59	100	100	100	93	81	72	60
6e-3	100	100	94	78	61	53	48	100	100	100	100	89	76	68
6e-4	100	100	100	100	100	85	63	100	100	100	100	89	76	54
6e-5	100	100	100	100	100	88	79	100	100	100	100	95	82	70
Etox- azole	100	100	100	100	100	84	67	100	100	100	100	85	67	49

Table 2 LC50, LC90 Values of 6a-4 and Etoxazole against Eggs of Spider Mitempdv=ax + b $LC50(mg L^{-1})$ $LC00(mg L^{-1})$ correlation

compd	y=ax+b	$LC_{50}(mg L^{-1})$	$LC_{90}(mg L^{-1})$	correlation	
				coefficient	

6a-4	y=1.47x+9.77	0.0006	0.0043	0.9655	
Etoxazole	y=2.43x+5.32	0.0063	0.0455	0.9437	

Table 3 LC₅₀, LC₉₀ Values of **6a-7**, **6b-3**, **6b-4**, **6c-2**, **6c-4**, and Etoxazole against Larvae of Spider Mite

compd	y=ax+b	LC ₅₀ (mg L ⁻¹)	LC ₉₀ (mg L ⁻¹)	correlation
				coefficient
6a-7	y=1.40x+8.17	0.0056	0.0452	0.9466
6b-3	y=1.46x+8.27	0.0058	0.0434	0.9580
6 b -4	y=1.42x+8.22	0.0055	0.0434	0.9447
6c-2	y=1.33x+8.21	0.0039	0.0357	0.9533
6 c -4	y=1.35x+8.24	0.0041	0.0358	0.9606
Etoxazole	y=0.66x+6.10	0.0221	1.8293	0.9855

Table 4 Insecticidal	Activity of the	Target Compo	ounds against Mc	osauito
				~ ~ ~ ~ ~ ~ ~

	activities (%) against mosquito at concentration (mg L ⁻¹)					
10	5	2	1	0.5	0.25	
		26				

compd						
6a-1	30			—		
6a-2	100	100	40	_		
6a-3	45			_		
6a-4	10		_	_		
6a-5	30		_	_		
6a-6	5			—		
6a-7	70		—	—		
6a-8	25		—	—		
6a-9	5			—		—
6a-10	15			—		
6b-1	40			—		—
6c-1	100	100	20	_	—	
6d-1	100	100	100	100	60	
6e-1	100	100	100	100	20	
6b-2	65		—	_	—	
6c-2	20					
6 d-2	50					
6e-2	50	60	75			
6b-3	50	—	50	—		
6d-3	100	100	100	100	100	20
6e-3	100	0	_	—		_
6b-4	20	—	—	—		—
6c-4	100	100	100	100	100	60
6d-4	100	60		—		—
6e-4	50		—	—		—
6b-5	100	100	0	—		—
6c-5	50			_		
6d-5	100	20		_		
6e-5	20		—	—		—
6b-6	40			_		
6b-7	30			_	—	—
6 d- 8	50			_	—	—
6b-9	60			_	—	
6d-9	100	20		—	—	—
Etoxazole	100	100	20	_	_	

Table 5 Insecticidal Activities of the Target Compounds (Mortality, Percent)compdcotton bollwormcorn boreroriental armyworm(600 mg L⁻¹)(600 mg L⁻¹)(600 mg L⁻¹)

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6a-1	55	45	30
6a-2	80	85	45
6a-3	85	80	75
6a-4	$100/100^{a}/50^{b}$	$100/100^{a}/40^{b}$	50
6a-5	$100/100^{a}/40^{b}$	$100/100^{a}/60^{b}$	20
6a-6	65	60	75
6a-7	100/60 ^a	100/80 ^a	30
6a-8	75	70	50
6a-9	60	45	45
6a-10	85	80	20
6b-1	0	0	25
6c-1	20	15	10
6d-1	20	5	10
6e-1	50	50	45
6b-2	50	45	75
6c-2	0	10	5
6d-2	70	65	40
6e-2	15	25	20
6b-3	0	5	20
6d-3	15	20	25
6e-3	0	10	50
6b-4	30	15	5
6c-4	35	45	70
6d-4	10	25	20
6e-4	5	10	20
6b-5	0	5	20
6c-5	0	0	5
6d-5	0	15	30
6e-5	5	10	20
6b-6	0	10	5
6b-7	15	15	15
6d-8	0	0	5
6b-9	25	35	50
6d-9	25	10	20
Etoxazole	$100/100^{a}/40^{b}$	100/100ª/35 ^b	100/100ª/60b

^a Activities at 200 mg L⁻¹. ^b Activities at 100 mg L⁻¹.

TOC graphic

Agrochemical Bioregulators

