Brønsted Acid-Catalyzed Formal [2 + 2 + 1] Annulation for the Modular Synthesis of Tetrahydroindoles and Tetrahydrocyclopenta[b]pyrroles

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Supporting Information

ABSTRACT: An expedient synthesis of tetrahydroindoles and tetrahydrocyclopenta[b]pyrroles, highlighted by Brønsted acid catalyzed formal [2 + 2 + 1] annulation reaction, is reported. Using three readily accessible reaction components, i.e., an electrophilic species in silyloxyallyl cations and two distinct nucleophiles in silylenol ethers and amines, our chemistry enables



the assembly and functionalization of these biologically important N-heterocycles in a highly modular manner.

etrahydroindoles and tetrahydrocyclopenta[b]pyrroles are a class of nitrogen-containing heterocycles of biological importance.¹ Exemplified in compounds 1a to 1c, these molecular structures are pharmacophores in molecules that serve as an anti-inflammatory agent, ^{1a} or inhibitors for cyclooxygenase 2, ^{1b} and Src tyrosine kinase. ^{1c} It has also been reported that tetrahydroindole 1d exhibits selective dopaminergic activities.^{1d} Natural products, such as microindolinone A and (+)-roseophilin,² also possess these ring motifs as a key feature to their architectures. Due to their biological contexts, the development of synthetic reactions that readily assemble tetrahydroindoles and tetrahydrocyclopenta[b]pyrroles remains an important synthetic endeavor.³ Scheme 1 illustrates recent advancements. Kurkin reported a solvent-free approach to tetrahydroindole core 2b, which was achieved via heatinduced 5-endo-dig cyclization of amino propargylic alcohols 2a.^{3c,d} Unlike tetrahydroindoles, literature precedents on the construction of tetrahydrocyclopenta[b]pyrroles are scarce.⁴ One of which was reported by Zhang who demonstrated that exposure of alkyne diols 3a to catalytic PPh₃AuCl and AgOTf in the presence of amines produced tetrahydrocyclopenta[b]pyrrole 3b, which occurred via Meyer-Schuster rearrangement and Paal-Knorr cyclization cascade.4

Combined with the use of expensive organometallic catalysts to induce ring formation, many of the elegant chemistries to construct tetrahydroindoles and tetrahydrocyclopenta[b]pyrroles rely on linear preconstruction of the substrates prior to cyclization. These strategies are at a disadvantage in terms of enabling the introduction of a vast array of functionalities to these rings in an expedient manner, a practice that is essential in drug discovery.⁵ Herein we describe our approach toward broadly substituted tetrahydroindoles and tetrahydrocyclopenta[b]pyrroles, which will be enabled by two distinct nucleophiles in silylenol ethers and primary amines, and an electrophile in silyloxyallyl cations that will be generated *in situ* upon ionization of α -hydroxy silylenol ether **4a**. Catalyzed by Brønsted acid,^{6,7} these three simple modular components Scheme 1. Synthetic Methods to Access Tetrahydroindoles and Tetrahydrocyclopenta[b]pyrroles



should readily undergo a formal [2 + 2 + 1] annulation reaction as represented in framework 4f. Our rationale is as

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follows. The regioselective interception of silyloxyallyl cations **4b** with silylenol ether should create a new C–C connectivity in the resulting γ -keto silylenol ether **4c**.⁷ The ensuing protodesilylation under the Brønsted acidic conditions will produce 1,4-diketone **4d**. In the presence of primary amine, the Paal-Knorr condensation should occur,⁸ thus completing our synthesis of tetrahydroindole and tetrahydrocyclopenta[*b*]-pyrrole motifs **4e**.

Our proof-of-concept and reaction optimization is summarized in Table 1,⁹ in which α' -hydroxy silylenol ether 5,

Tal	ole	1.	Reaction	0	ptimization
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Me-	OTBS	OTBS + Ph 6 (2.0 equiv)	Py•TfOH (MeCN (0 <i>then</i> add <i>then</i> PrNH	0.2 equiv) .5 M), rt; itive, rt; Itive, rt; I2, reflux 7	r }Ph
entry	additive	additive equiv.	PrNH ₂ equiv.	total reaction time (h)	yield ^a (%)
1	-	-	4.0	168	77
2	PPTS	0.5	4.0	96	74
3	TfOH	0.5	4.0	4	62
4	CSA	0.5	4.0	8	82
5	TsOH	0.5	4.0	6	83
6	TsOH	0.3	4.0	5	83
7	TsOH	0.2	4.0	8	77
8	TsOH	0.3	2.0	7	75
9	TsOH	0.3	4.0	192	56 ^b
^a Isola perfor	ted yield med at roo	after colun om temperatu	nn chrom re.	atography. ^b Reacti	on was

acetophenone-derived silylenol ether 6, and propylamine were employed as model substrates. As shown in entry 1, we began by subjecting compounds 5 and 6 to catalytic pyridinium triflate in acetonitrile at room temperature to allow formation of the key monosilylated 1,4-diketone construct in a regioselective manner.^{7a} Excess propylamine was then added to the mixture after complete consumption of α' -hydroxy silvlenol ether 5 and the ensuing protodesilvlation as easily monitored by TLC, and the reaction was then warmed to reflux. Interestingly, this pilot experiment indeed led to the formation tetrahydrocyclopenta-[b]pyrrole 7, which was isolated in 77% yield after 7 days of reaction time. We noted while the coupling of α' -hydroxy silylenol ether **5** with silylenol ether 6 concluded in just approximately 1 h, the ratedetermining steps in these initial conditions appeared to be controlled by protodesilylation of the forming γ -keto silylenol ether.

Following this initial result, we focused our efforts to improve the protodesilylation sequence with a notion that the efficiency of this step could be affected by an introduction of an additive, such as a secondary Brønsted acid catalyst. As shown in entries 2–5, our hypothesis was tested through the screening of pyridinium tosylate, triflic acid, camphorsulfonic acid, and tosic acid monohydrate. These additives were added in the amount of 0.5 mol equiv after completion of the preceding reaction between α' -hydroxy silylenol ether 5 and silylenol ether 6, but before the addition of propylamine. Indeed, the addition of stronger Brønsted acids substantially enhanced the rate of protodesilylation. To this end, we ultimately chose tosic acid. Entries 5–8 indicate that the molar amount of tosic acid or propylamine could be further reduced without notably affecting the robustness of the reaction.

Nonetheless, the efficiency of Paal-Knorr condensation diminished dramatically without heat (entry 9). Overall, our optimized conditions are as follows: ensuing the coupling between α' -hydroxy silylenol ether 5 and 2.0 equiv of silylenol ether 6 promoted by 0.2 equiv of pyridinium triflate in acetonitrile at room temperature, 0.3 equiv of tosic acid monohydrate was introduced to induce protodesilylation. After the addition of 4.0 equiv of propylamine, the mixture was warmed to reflux, furnishing tetrahydrocyclopenta[b]pyrrole 7 in 83% yields in just 5 h of total reaction time.

To identify the scope of our method, we began by screening the applicability of various primary amines (Scheme 2). In

Scheme 2. Scope of Amines



^aIsolated yield after column chromatography.

these studies, both 5- and 6-membered α -hydroxy silylenol ethers 5 and 8 were employed to furnish a diverse library of Nsubstituted tetrahydrocyclopenta[b]pyrrole and tetrahydroindole 7 and 9, respectively. Commencing with aniline and its derivatives, i.e., 4-trifluoromethylaniline, 4-aminoacetophenone, and 4-methoxy-aniline, our reaction successfully produced the corresponding adducts 7a-7d and 9a-9d in good yields. Interestingly, the use of ammonia as the Paal-Knorr reactant furnished unprotected N-heterocyclic core 7e and 9e in 84% and 56% yields, respectively. Cyclopropylamine, propylamine, allylamine, and benzylamine were tolerated by our reaction conditions. These aliphatic amines furnished the corresponding tetrahydrocyclopenta [b] pyrrole 7f-7i and tetrahydroindole 9f-9i again in excellent yields within several hours of reaction time. We also subjected neurotransmitter tryptamine, which produced tethered heterocyclic motifs 7j and 9j in 84% and 70% yields, respectively. Compounds 7j and 9a existed in the crystalline form, allowing us to assign their structure by X-ray diffraction.¹⁰ Interestingly, these experiments revealed formation of 6-membered tetrahydroindole 9

was generally lower yielding than that of its 5-membered counterpart 7. This phenomenon is most likely stemmed from challenges associated with the generation of 6-membered silyloxyallyl cations from the corresponding α' -hydroxy silylenol ether 5.¹¹

Our investigation continued with substituent effects at the α carbon. In this study, we employed a series of secondary α' hydroxy silylenol ethers **10** and **11** in the presence of silylenol ether **6** and aniline as coupling partners. Schemes 3 and 4





^{*a*}Isolated yield after column chromatography. ^{*b*}The reaction was performed in 0.2 M concentration using 4.0 equiv of silylenol ether 6. ^{*c*}3.0 equiv of silylenol ether 6 was employed.



^{*a*[a]}Isolated yield after column chromatography. ^[b]3.0 equiv of silylenol ether 7 was employed.

depict our results, starting with the use of unsubstituted α' -hydroxy silylenol ethers, which led to products **12a** and **13a**, however in poor yields due to decomposition of materials under the reaction conditions. Nonetheless, our method was tolerated by both aromatic and aliphatic substituents at the α -carbon. For instance, arene-bearing substrates, such as phenyl, toluoyl, and *p*-chlorophenyl, readily afforded tetrahydro-cyclopenta[*b*]pyrrole **12b**-**12d** and tetrahydroindole **13b**-**13d** in high yields. A slower rate of reaction and diminishing yield of the products involving chlorine-containing starting materials was noted. With respect to the aliphatic variants, we

explored linear *n*-octyl chain as well as branched isobutyl and cyclohexyl groups. Interestingly, steric effects did not play significant role, as the resulting 5-membered products 12e-12g and 6-membered products 13e and 13g were all isolated in high yields. We also observed that several of the 6-membered substrates required a higher loading of silylenol ether 6 to improve the rate of the initial coupling reaction. Due to their crystalline form, we were able to confirm the structures of 13b and 13g by X-ray diffraction.¹⁰

Our previous studies revealed that unsymmetrical silyloxyallyl cations could be effectively generated from tertiary α hydroxy silylenol ethers.⁶ To examine the utility of this alternate substrate motif in this chemistry, we subjected a series of 5-membered starting materials 14 that were elaborated with identical aromatic and aliphatic substituents at the α -carbon to those of substrates 10. Indeed, exposure of these compounds to our method furnished target products 12b-12g. Nonetheless, a substantial attrition in the product yields was observed, as these substrates could not be fully consumed despite the prolonged reaction time. We suspected that the underlying cause for this phenomenon lay in the steric barrier imposed by the tertiary alcohols, therefore slowing the rate of ionization and ultimately prompting competitive quenching of the catalyst by either silvlenol ether 6 or α hydroxy silylenol ethers 14 via protodesilylation.

Using both 5-membered and 6-membered substrates 5 and 8 and propylamine as the coupling partners, we then varied silylenol ethers 15 to demonstrate our ability to expediently diversify the C2 and C3 positions of tetrahydrocyclopenta[b]pyrrole 16 and tetrahydroindole 17 (Table 2). Interestingly, we identified in some 6-membered systems that protodesilylation of the resulting γ -keto silylenol ether constructs unexpectedly led to concomitant intramolecular cyclization, prior to addition of propylamine, therefore affording tetrahydrobenzofuran byproducts. To circumvent this undesired furan annulation, we produced an alternate set of protodesilylation conditions to unmask the 1,4-diketone motif using CsF at reflux (Conditions B).⁹ This protocol was followed by addition of TsOH and propylamine to affect the Paal-Knorr condensation.

With the two complementary conditions in hand, we subjected acetophenone-derived silylenol ethers bearing electronically opposing trifluoromethyl and methoxy groups (entries 1 and 2), revealing that the electron-deficient group appeared to be less robust. Our method allows for simultaneous incorporation of functional groups at the C2 and C3 positions of the forming heterocycles. For instance, α tetralone and cyclohexanone derived silylenolates were found to be compatible to furnish their respective polycyclic adducts 16c and 16d as well as 17c and 17d in good yields. As shown in entry 5, we employed a highly substituted silvlenol ether that was prepared from 3-pentanone. This nucleophile selectively introduced C2-ethyl and C3-methyl substituents in products 16e and 17e with 50% and 54% isolated yields, respectively. A similar observation was noted when 2-phenylacetophenone derived silylenol ether nucleophile was applied to our reaction (entry 6). In this case, the resulting tetrahydrocyclopenta [b]pyrrole 16f and tetrahydroindole 17f, both of which were decorated with two phenyl groups at the pyrrole moiety, were isolated in good yields. Finally, we attempted to elaborate exclusively on the C3 position using phenylacetaldehyde derived nucleophile (entry 7). While the resulting annulation product 16g was accessible from 5-membered substrate 5 in

Table 2. Scope of Silylenol Ethers



Conditions B: Py-TfOH (0.2 equiv), MeCN (0.5M), rt; then CsF (2.0 equiv), reflux; then TsOH (1.0 equiv), propylamine (4.0 equiv), reflux



^{*a*}Isolated yield after column chromatography. ^{*b*}Substrate 8 was not fully consumed despite the prolonged reaction time.

36% yield, an application to starting material **8** was not fruitful, as such a reaction under either condition only led to decomposition of materials.

In summary, we have developed a robust one-pot chemistry to construct highly substituted tetrahydroindoles and tetrahydrocyclopenta[b]pyrroles under mild catalytic conditions. Highlighted by a formal [2 + 2 + 1] annulation reaction that involves three reaction partners in silyloxyallyl cations, silylenolates, and primary amines, our method successfully produced these biologically significant *N*-heterocycles in good yields. Applications of our strategy to access structurally related heterocycles and complex molecules are ongoing in our laboratory.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.9b01032.

Experimental procedures and spectral data of new compounds (PDF)

Accession Codes

CCDC 1894940–1894943 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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(9) See Supporting Information for detailed reaction optimizations. (10) CCDC 1894940–1894943 contain the supplementary crystallographic data for compounds 7j, 9a, 13b, and 13g, respectively.

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