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> LETTERS TO THE EDITOR

Unexpected Reaction of Secondary Phosphine Chalcogenides with Acridine

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Abstract—Secondary phosphine chalcogenides reacted with acridine under mild conditions according to the nucleophilic addition scheme to form 9-chalcogenophosphoryl-9,10-dihydroacridines.

Keywords: acridine, secondary phosphine chalcogenides, nucleophilic addition, 9-chalcogenophosphoryl-9,10dihydroacridines

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We have recently reported on an original threecomponent reaction of azines, secondary phosphine chalcogenides, and alkyl propiolates exemplified by pyridines and quinolines [1–3]. The reaction proceeds under mild conditions (20–72°C, 3–19 h, MeCN) with regio- and stereoselective formation of the *C*-phosphorylated (*E*)-*N*-ethenyl-1,4(or 1,2)-dihydropyridines (or quinolines) [1–3] (Scheme 1).

In this study, we attempted to realize that threecomponent reaction using acridine as the azine. However, heating (50–52°C) of acridine, secondary phosphine chalcogenides **1a–1c**, and methyl propiolate in acetonitrile for 4–7 h did not give the expected C⁹-phosphorylated *N*-ethenyl-9,10-dihydroacridines (Scheme 2). Under those conditions, methyl propiolate was not involved in the reaction, but 1 : 1 adducts of phosphine chalcogenides with acridine – 9-chalcogenophosphoryl-9,10-dihydroacridines 2a-2c were formed in 60–65% yield. Such addition of secondary phosphine chalcogenides to acridine readily proceeded also in the absence of methyl propiolate, the yield of dihydroacridines 2a-2c being 64–69%. That meant that the electron-deficient acetylene did not participate in the reaction even at the intermediate stages (Scheme 2).

At the same time, according to the reference data, acridine can enter three-component reactions with methyl propiolate [4] or dimethyl acetylenedicarboxylate [5] and some CH- or OH-acids (nitromethane [4] and methanol [5]) to form the corresponding functionalized *N*-ethenyl adducts.

Therefore, phosphorylation of acridine by secondary phosphine chalcogenides easily prepared





 $R = Ph, X = O(a); R = Ph(CH_2)_2, X = S(b), X = Se(c).$

from elemental phosphorus, styrene, and chalcogens [6] opens a convenient way to the earlier unknown phosphorylated dihydroacridines, promising precursors for the design of drugs [7–9], reagents for preparation of innovative materials [10, 11], ligands for the synthesis of metal complexes [12–14], and building blocks for organic and organoelement synthesis [15–17].

The mentioned experiments were performed under inert atmosphere (argon). The reactions were monitored by ³¹P NMR spectroscopy.

Nucleophilic addition of secondary phosphine chalcogenides 1a-1c to acridine. 1.0 mmol of acridine was added to a solution of secondary phosphine chalcogenide 1a-1c (1.0 mmol) in 3 mL of acetonitrile. The reaction mixture was stirred at 50-52°C during 4 h (in the case of phosphine chalcogenide 1c) or 7 h (in the cases of phosphine chalcogenides 1a, 1b) until disappearance of the signal of the starting phosphine chalcogenide (2–23 ppm) in the ³¹P NMR spectrum and appearance of the signal of compounds 2a-2c at 30-60 ppm. The solvent was removed under reduced pressure, the residue was washed with Et₂O $(5 \times 1 \text{ mL})$ via decantation (in the case of dihydroacridine 2a) or resedimented from acetone to hexane (in the cases of dihvdroacridines **2b**, **2c**).

9-(Diphenylphosphoryl)-9,10-dihydroacridine (2a). Yield 263 mg (69%), white powder, mp 218–219°C. IR spectrum (KBr), v, cm⁻¹: 3391, 3055, 2903, 1631, 1473, 1435, 1301, 1252, 1181, 1106, 1031, 752, 697, 560, 533. ¹H NMR spectrum (DMSO- d_6), δ , ppm: 5.40 d (1H, CHP, ${}^{2}J_{PH} = 11.3$ Hz); 6.46 d. d [2H, H_{2.7}, ${}^{3}J_{2(7)-1(8)} \approx {}^{3}J_{2(7)-3(6)} = 7.2 \text{ Hz}]; 6.62 \text{ d} [2\text{H}, \text{H}_{1,8}, {}^{3}J_{1(8)-2(7)} =$ 7.2 Hz]; 6.68 d [2H, H_{4,5}, ${}^{3}J_{4(5)-3(6)} = 8.2$ Hz]; 6.98 d. d [2H, H_{3,6}, ${}^{3}J_{3(6)-4(5)} \approx {}^{3}J_{3(6)-2(7)} = 7.5$ Hz]; 7.41 m (4H, H_m ; 7.54 m (2H, H_p); 7.72 m (4H, H_o); 8.56 s (1H, NH). ¹³C NMR spectrum (DMSO- d_6), δ_C , ppm: 45.7 d (CHP, ${}^{1}J_{CP} = 64.0$ Hz); 113.6 d (C_{4.5}, ${}^{4}J_{CP} = 2.1$ Hz);

114.2 d (C_{8a} , ${}^{2}J_{CP}$ = 5.6 Hz); 119.0 d ($C_{2,7}$, ${}^{4}J_{CP}$ = 2.3 Hz); 127.6 d ($C_{3,6}$, ${}^{5}J_{CP}$ = 2.7 Hz); 128.1 d (C_{m} , ${}^{3}J_{CP}$ = 10.7 Hz); 129.9 d ($C_{1,8}$, ${}^{3}J_{CP}$ = 3.8 Hz); 131.5 d (C_{pso} , ${}^{1}J_{CP}$ = 91.5 Hz); 131.6 d (C_{p} , ${}^{4}J_{CP}$ = 2.7 Hz); 131.9 d (C_o , ${}^2J_{CP} = 8.7 \text{ Hz}$); 141.8 d (C_{4a} , ${}^3J_{CP} = 3.4 \text{ Hz}$). ¹⁵N NMR spectrum (DMSO- d_6), δ_N : –279.6 ppm. ³¹P NMR spectrum (DMSO- d_6), δ_P : 30.3 ppm. Found, %: C 78.89; H 5.43; N 3.75; P 7.93. C₂₅H₂₀NOP. Calculated, %: C 78.73; H 5.29; N 3.67; P 8.12.

9-[Bis(2-phenylethyl)phosphorothioyl]-9,10-dihydroacridine (2b). Yield 290 mg (64%), yellow powder, mp 133–135°C. IR spectrum (film), v, cm⁻¹: 3388, 3289, 3204, 3057, 3029, 2921, 1602, 1484, 1450, 1406, 1301, 1214, 1072, 1033, 906, 835, 737, 703, 646, 603, 470. ¹H NMR spectrum (CDCl₃), δ , ppm: 1.86, 2.10 m (4H, CH₂P); 2.76 m (4H, CH₂Ph); ppm: 1.66, 2.16 m (11, CH₂, 1), 2.76 m (11, CH₂), 2.76 m (11, CH₂), 4.60 d (1H, CHP, ${}^{2}J_{PH} = 15.1 \text{ Hz})$; 6.14 s (1H, NH); 6.75 d [2H, H_{4,5}, ${}^{3}J_{4(5)-3(6)} = 7.8 \text{ Hz}]$; 7.04 d. d [2H, H_{2,7}, ${}^{3}J_{2(7)-1(8)} \approx {}^{3}J_{2(7)-3(6)} = 7.3 \text{ Hz}]$; 7.15 m (4H, H_o); 7.21 m (2H, H_{3,6}); 7.24 m (2H, H_p); 7.29 m (4H, H_m); 7.37 d [(2H, H_{1,8}, ${}^{3}J_{1(8)-2(7)} = 7.3$ Hz]. ${}^{13}C$ NMR spectrum (CDCl₃), $\delta_{\rm C}$, ppm: 28.7 d (<u>C</u>H₂Ph, ²J_{CP} = 2.9 Hz); 29.2 d (CH₂P, ¹J_{CP} = 43.3 Hz); 49.9 d (CHP, ${}^{1}J_{CP} = 42.2$ Hz); 114.3 (C_{4,5}); 115.4 d (C_{8a}, ${}^{2}J_{CP} =$ 3.6 Hz); 121.0 (C_{2,7}); 126.4 (C_p); 128.2 (C_o); 128.7 (C_m); 128.9 d (C_{3,6}, ${}^{5}J_{CP} = 3.4 \text{ Hz}$); 130.6 d (C_{1,8}, ${}^{3}J_{CP} = 3.2 \text{ Hz}$); 140.0 d (C_{4a}, ${}^{3}J_{CP} = 3.1 \text{ Hz}$); 141.3 d (C_{ipso}, ${}^{3}J_{CP} =$ 14.6 Hz). ¹⁵N NMR spectrum (CDCl₃), δ_N : -284.1 ppm. ³¹P NMR spectrum (CDCl₃), δ_P : 59.3 ppm. Found, %: C 76.53; H 6.39; N 3.24; P 6.62; S 6.79. C₂₉H₂₈NPS. Calculated, %: C 76.79; H 6.22; N 3.09; P 6.83; S 7.07.

9-[Bis(2-phenylethyl)phosphoroselenoyl]-9,10dihvdroacridine (2c). Yield 330 mg (66%), beige powder, mp 128–131°C. IR spectrum (film), v, cm⁻¹: 3392, 3264, 3188, 3058, 3027, 2928, 1607, 1480, 1455, 1405, 1304, 1211, 1069, 1031, 909, 853, 742, 702, 651, 485. ¹H NMR spectrum (CDCl₃), δ , ppm: 1.95, 2.19 m (4H, CH₂P); 2.76 m (4H, C<u>H</u>₂Ph); 4.74 d (1H, CHP, ${}^{2}J_{PH} = 13.9$ Hz); 6.09 br. s (1H, NH); 6.76 d [2H, H_{4,5}, ${}^{3}J_{4(5)-3(6)} = 7.7$ Hz]; 6.97 d. d [(2H, H_{2,7}, ${}^{3}J_{2(7)-1(8)} \approx {}^{3}J_{2(7)-3(6)} = 7.3$ Hz]; 7.09 m (4H, H_o); 7.15 m (4H, Ph, Ar); 7.21 m (4H, Ph, Ar); 7.44 br. d [2H, H_{1,8}, ${}^{3}J_{1(8)-2(7)} = 7.3$ Hz]. 13 C NMR spectrum (CDCl₃), δ_{C} , ppm: 29.0 d (CH₂P, ${}^{1}J_{CP} = 36.0$ Hz); 29.7 d (d, <u>C</u>H₂Ph, ${}^{2}J_{CP} = 2.1$ Hz); 49.6 d (CHP, ${}^{1}J_{CP} = 34.7$ Hz); 114.2 d (C_{4,5}, ${}^{4}J_{CP} = 2.8$ Hz); 115.4 d (C_{8a}, ${}^{2}J_{CP} = 4.1$ Hz); 121.7 d (C_{2,7}, ${}^{4}J_{CP} = 2.8$ Hz); 126.1 (C_p); 128.0 (C_o); 128.3 (C_m); 128.7 d (C_{3,6}, ${}^{5}J_{CP} = 2.8$ Hz); 130.7 d (C_{1,8}, ${}^{3}J_{CP} = 3.4$ Hz); 141.3 d (C_{*ipso*, ${}^{3}J_{CP} = 15.3$ Hz). 15 N NMR spectrum (CDCl₃), δ_{P} : 53.5 ppm (+d-satellites, ${}^{1}J_{PSe} = 709.8$ Hz). 77 Se NMR spectrum (CDCl₃), δ_{Se} : -379.9 ppm, d (${}^{1}J_{PSe} = 709.8$ Hz). Found, %: C 69.43; H 5.82; N 2.64; P 6.04; Se 15.56. C₂₉H₂₈NPSe. Calculated, %: C 69.60; H 5.64; N 2.80; P 6.19; Se 15.78.}

IR spectra were recorded using a Varian 3100 FT-IR spectrometer (KBr pellet or thin layer). ¹H, ¹³C, ¹⁵N, ³¹P, and ⁷⁷Se NMR spectra were recorded using Bruker DPX-400 and Bruker AV-400 spectrometers (400.13, 10.62, 40.56, 161.98 and 76.31 MHz, respectively) in CDCl₃ or DMSO- d_6 , with the following internal [HMDS (¹H, ¹³C), MeNO₂ (¹⁵N), Me₂Se (⁷⁷Se)] or external [85% H₃PO₄ (³¹P)] references. The signals in the proton spectra were assigned using 2D homonuclear correlation method COSY. The ¹³C signals were assigned basing on the analysis of the 2D heteronuclear correlation spectra HSQC and HMBC.

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CONFLICT OF INTEREST

No conflict of interest was declared by authors.

REFERENCES

 Gusarova, N.K., Volkov, P.A., Ivanova, N.I., Arbuzova, S.N., Khrapova, K.O., Albanov, A.I., Smirnov, V.I., Borodina, T.N., and Trofimov, B.A., *Tetrahedron Lett.*, 2015, vol. 56, p. 4804. doi 10.1016/j.tetlet.2015.06.062

- Gusarova, N.K., Volkov, P.A., Ivanova, N.I., Khrapova, K.O., Albanov, A.I., Afonin, A.V., Borodina, T.N., and Trofimov, B.A., *Tetrahedron Lett.*, 2016, vol. 57, p. 3776. doi 10.1016/j.tetlet.2016.07.024
- Volkov, P.A., Telezhkin, A.A., Ivanova, N.I., Khrapova, K.O., Albanov, A.I., Gusarova, N.K., and Trofimov, B.A., *Russ. J. Gen. Chem.*, 2018, vol. 88, p. 912. doi 10.1134/S1070363218050122
- Acheson, R.M. and Woollard, J., J. Chem. Soc. Perkin Trans. 1, 1975, p. 438. doi 10.1039/P19750000438
- Acheson, R.M. and Burstall, M.L., J. Chem. Soc., 1954, p. 3240. doi 10.1039/JR9540003240
- Gusarova, N.K., Arbuzova, S.N., and Trofimov, B.A., *Pure Appl. Chem.*, 2012, vol. 84, no. 3, p. 439. doi 10.1351/PAC-CON-11-07-11
- Ramesh, K.B. and Pasha, M.A., *Bioorg. Med. Chem. Lett.*, 2014, vol. 24, p. 3907. doi 10.1016/ j.bmcl.2014.06.047
- Pérez, S.A., de Haro, C., Vicente, C., Donaire, A., Zamora, A., Zajac, J., Kostrhunova, H., Brabec, V., Bautista, D., and Ruiz, J., *ACS Chem. Biol.*, 2017, vol. 12, p. 1524. doi 10.1021/acschembio.7b00090
- Kudryavtseva, T.N., Lamanov, A.Yu., Klimova, L.G., and Nazarov, G.V., *Russ. J. Gen. Chem.*, 2018, vol. 88, p. 676. doi 10.1134/S1070363218040102
- Li, Z., Liu, R., Tan, Z., He, L., Lu, Z., and Gong, B., ACS Sensors, 2017, vol. 2, p. 501. doi 10.1021/ acssensors.7b00139
- Zhao, B., Miao, Y., Wang, Z., Wang, K., Wang, H., Hao, Y., Xu, B., and Li, W., *Nanophotonics*, 2017, vol. 6, p. 1133. doi 10.1515/nanoph-2016-0177
- Srimani, D., Diskin-Posner, Y., Ben-David, Y., and Milstein, D., Angew. Chem. Int. Ed., 2013, vol. 52, p. 14131. doi 10.1002/anie.201306629
- Zhu, R.-Y., He, J., Wang, X.-C., and Yu, J.-Q., J. Am. Chem. Soc., 2014, vol. 136, p. 13194. doi 10.1021/ ja508165a
- Chowdhury, M.A.H., Rahman, M.S., Islam, M.R., Rajbangshi, S., Ghosh, S., Hogarth, G., Tocher, D.A., Yang, L., Richmond, M.G., and Kabir, S.E., *J. Organomet. Chem.*, 2016, vol. 805, p. 34. doi 10.1016/j.jorganchem.2015.12.023
- Mironovich, L.M., Ageeva, L.S., and Podol'nikova, A.Yu., *Russ. J. Gen. Chem.*, 2016, vol. 86, p. 420. doi 10.1134/ S1070363216020390
- Cho, A.-N., Chakravarthi, N., Kranthiraja, K., Reddy, S.S., Kim, H.-S., Jin, S.-H., and Park, N.-G., *J. Mater. Chem.* (*A*), 2017, vol. 5, p. 7603. doi 10.1039/C7TA01248A
- 17. Wang, M., Fan, Q., and Jiang, X., *Org. Lett.*, 2018, vol. 20, p. 216. doi 10.1021/acs.orglett.7b03564