



# Polysubstituted Pyrimidines as mPGES-1 Inhibitors: Discovery of Potent Inhibitors of PGE<sub>2</sub> Production with a Strong Antiinflammatory Effects in Carrageenan-Induced Rat Paw Edema

Filip Kalčic,<sup>[a,b]</sup> Dr. Viktor Kolman,<sup>[a]</sup> Dr. Haresh Ajani,<sup>[a]</sup> Dr. Zdeněk Zídek,\*<sup>[c]</sup> and Dr. Zlatko Janeba\*<sup>[a]</sup>

[a]	F. Kalčic, Dr. V. Kolman, Dr. H. Ajani, and Dr. Z. Janeba							
	Institute of Organic Chemistry and Biochemistry of the Ozech Academy of Sciences							
	Flemingovo nám. 2, 166 10 Prague 6, Czech Republic							
	E-mail: janeba@uochb.cas.cz							
[b]	F. Kalčic							
	Department of Organic Chemistry, Faculty of Science							
	Charles University							
	Hlavova 8, 128 43 Prague 2, Czech Republic							
[C]	Dr. Zdeněk Zídek							
	Institute of Experimental Medicine of the Czech Academy of Sciences							
	Vídeňská 1083, 142 20 Prague 4, Czech Republic							
	E mail: zdanak zidak@iam cas cz							

Supporting information for this article is given via a link at the end of the document.

Abstract: We report an extensive structure-activity relationship optimization of polysubstituted pyrimidines that led to a discovery of 5-butyl-4-(4-benzyloxyphenyl)-6-phenylpyrimidin-2-amine, and its difluorinated analogue. These compounds are submicromolar inhibitors of PGE<sub>2</sub> production (IC<sub>50</sub> as low as 12 nM). In order to identify the molecular target of anti-inflammatory pyrimidines, we performed extensive studies including enzymatic assays, homology modeling and docking. The difluorinated analogue simultaneously inhibits two key enzymes of the arachidonic acid cascade, namely mPGES-1 and COX-2, where the mPGES-1 inhibition represents the principal mechanism of action. Other pyrimidines studied are potent mPGES-1 inhibitors with no observed inhibition of COX-1/2 enzymes. Moreover, two most potent compounds proved to be significantly effective in vivo in a model of acute inflammation, suppressing the carrageenan-induced rat paw edema by 36% and 46%. Promising results of this study warrant further preclinical evaluation of selected anti-inflammatory candidates.

#### Introduction

Prostaglandin  $E_2$  (PGE<sub>2</sub>) is a naturally occurring lipid mediator, which plays an important role in various inflammatory processes, fever, pain, and cancer. PGE<sub>2</sub> belongs to the most predominant pro-inflammatory prostanoids. PGE<sub>2</sub> biosynthesis consists of the release of arachidonic acid (AA) from membrane phospholipids by phospholipases (PLAs), oxygenation of AA to prostaglandin G<sub>2</sub> (PGG<sub>2</sub>) and subsequent reduction of PGG<sub>2</sub> to prostaglandin H<sub>2</sub> (PGH<sub>2</sub>) by cyclooxygenase (COX) enzymes, and conversion of PGH<sub>2</sub> to PGE<sub>2</sub> by inducible microsomal PGE<sub>2</sub> synthase-1 (mPGES-1).

Non-steroidal anti-inflammatory drugs (NSAIDs),<sup>[1]</sup> which today belong to the most widely used therapeutic agents to treat inflammation, fever, and pain, exert their biological activity *via* non-selective COX enzymes inhibition, i.e. inhibition of both constitutive COX-1 and inducible COX-2.<sup>[2-4]</sup> The major

drawback of traditional NSAIDs (t-NSAIDs) is the occurrence of serious gastrointestinal (GI) complications,<sup>[5]</sup> namely gastric bleeding, ulceration, perforation and dyspepsia. Thus, selective COX-2 inhibitors (coxibs) were developed and commercialized as promising anti-inflammatory agents in order to avoid the GI adverse events,<sup>[6,7]</sup> However, coxibs were later related to the increased risk of cardiovascular events like myocardial infarction, stroke, and systemic and pulmonary hypertension.<sup>[8]</sup>

Due to a wide spectrum of adverse events of t-NSAIDs as well as cardiovascular toxicity of coxibs, researchers intensively searched for new approaches to treat inflammation, pain, and fever. As most t-NSAIDs contained a free carboxylic acid group in their molecule, the GI toxicity seemed to be closely related to their acidic nature. Thus, structural modification of some NSAIDs like indomethacin and meclofenamic acid (mostly by conversion of their free carboxylic acid moiety into ester or amide derivatives) led to a generation of highly selective COX-2 inhibitors with eliminated GI side effects compared to the parent compounds.<sup>[9-12]</sup> Also synthesis of analogues with lower membrane permeabilization activity produced fewer gastric lesions after their oral administration.<sup>[13,14]</sup> Other attempts how to enhance gastric safety profile of anti-inflammatory drugs were based on their association with cytoprotective mediators (for see<sup>[1,15,16]</sup>). comprehensive reviews such as phosphatidylcholine,[17,18] dialkylphosphate,[19] nitric oxide[20-34] or nitroxyl,[35] and hydrogen sulfide.[36-42] Nevertheless, it has been speculated,<sup>[43]</sup> that the NO release is not required to exert the cytoprotective effects of modified NSAIDs, while the simple formation of NSAIDs prodrugs is a sufficient condition to develop a safer alternative to unprotected NSAIDs.

Relatively recent strategy in development of new antiinflammatory agents is a design of compounds targeting downstream and/or multiple enzymes of the AA cascade. In that matter, selective inhibitors of mPGES-1 have been recently identified,<sup>[44-61]</sup> as well as selective inhibitors of 5-lipoxygenase (5-LOX),<sup>[62]</sup> dual COX/5-LOX inhibitors,<sup>[34,63-67]</sup> dual mPGES-1/5-LOX inhibitors,<sup>[53,68-73]</sup> and dual thromboxane antagonists–COX-

2 inhibitors.<sup>[74]</sup> Similarly, dual inhibition of fatty acid amide hydrolase (FAAH) and the COX enzymes led to enhanced analgesic effects of NSAIDs with decreased GI side effects.<sup>[75,76]</sup> All these findings have raised high expectations for development of novel and safer anti-inflammatory drugs.

We previously reported polysubstituted pyrimidines as potent inhibitors of prostaglandin  $E_2$  (PGE<sub>2</sub>) production with potential anti-inflammatory properties.<sup>[77-81]</sup> Out of them, lead compound 5-butyl-4-(4-methoxyphenyl)-6-phenylpyrimidin-2-amine (**1**, Figure 1) was selected for further evaluation as a preclinical candidate for treatment of inflammation. Moreover, compound **1** served as a starting point for further structure-activity relationship (SAR) studies with the aim to identify even more potent inhibitors of PGE<sub>2</sub> production and, eventually, superior anti-inflammatory agents.



Figure 1. Structure of preclinical anti-inflammatory candidate 1 and a general structure of target compounds prepared within this study.

Herein, we describe a systematic structure-activity relationship optimization of compound **1** *via* preferential introduction of electron-donating groups (e.g. alkyls, amines or ethers) at the phenyl moiety in C4 position of the pyrimidine ring. Other substituents (at positions C2, C5, and C6 of pyrimidine) were kept intact in order not to introduce too many variables. Based on our previous research,<sup>[80-81]</sup> such compounds were expected to exhibit an increased potential to inhibit PGE<sub>2</sub> production. The potency of prepared compounds to inhibit PGE<sub>2</sub> production was evaluated *in vitro* on mouse peritoneal cells with induced immune response provoked by lipopolysaccharide (LPS) from *Escherichia coli*. The extensive SAR study led to the discovery of very potent inhibitors of PGE<sub>2</sub> production and their efficacy was verified *in vivo* using carrageenan-induced rat paw edema experiments.

#### **Results and Discussion**

*Chemistry.* In our laboratory, Suzuki-Miyaura cross-coupling represents a major tool to obtain target polysubstituted pyrimidines bearing two aromatic moleties in the C4 and C6 positions of the pyrimidine ring.<sup>[80-81]</sup> Various arylboronic acids (or their pinacol esters) can be nowadays purchased from commercial suppliers as a starting material for the Suzuki-Miyaura cross-coupling reactions. Nevertheless, two arylboronic acid pinacol esters had to be prepared in order to be able to synthesize some of the desired final compounds.

Firstly, synthesis of (4-methoxynaphthalen-1-yl)boronic acid pinacol ester (4, Scheme 1) started with a selective bromination of 1-methoxynaphthalene using *N*bromosuccinimide (NBS) to give bromo derivative **3**  quantitatively.<sup>[82]</sup> Subsequent lithiation of intermediate **3**, followed by transmetallation using pinacol diboronate,<sup>[83]</sup> afforded pinacol ester **4** (Scheme 1) in a 12% yield. However, the desired product **4** was obtained in a 66% yield when we employed conditions of catalytic borylation reported by Harada et al.<sup>[84]</sup>



Scheme 1. Synthesis of boronic acid pinacol ester 4. Reagents and conditions: (i) NBS, MeCN, 98%; (ii) BuLi,  $B_2pin_2$ , THF, -78 °C, 12%; (iii)  $B_2pin_2$ ,  $K_3PO_4$ , Pd(PPh<sub>3</sub>)<sub>4</sub>, DMF, 120 °C, 66%.

Secondly, commercially available o-vanillin (2-hydroxy-3methoxybenzaldehyde, **5**) was used as a starting material in order to prepare novel (7-methoxybenzofuran-4-yl)boronic acid pinacol ester (**10**, Scheme 2). The condensation of *o*-vanillin with ethyl 2-chloroacetate afforded ethyl carboxylate **6** in an 84% yield,<sup>[85,86]</sup> although Yamaguchi et al.<sup>[87]</sup> have reported the carboxylic acid derivative as the only product of the aforementioned condensation. Intermediate **6** was selectively brominated with NBS to yield quantitatively bromo derivative **7**, which was subsequently hydrolyzed to give carboxylic acid **8** in a 94% yield. Microwave-assisted decarboxylation of acid **8** was performed according to Musser et al.<sup>[88]</sup> to afford intermediate **9** (Scheme 2). Finally, catalytic borylation<sup>[84]</sup> of bromo derivative **9** gave the novel pinacol ester **10** in a 75% yield.



Having all the desired arylboronic acids and pinacol boronates in hand, the previously reported synthetic methodology<sup>[80,81]</sup> was employed for the synthesis of target 2-amino-5-butyl-6-phenylpyrimidines bearing a substituted/modified phenyl group in the C4 position of the pyrimidine moiety. Firstly, 2-amino-5-butyl-4-chloro-6-phenylpyrimidine <sup>[81]</sup> (**11**, Scheme 3) was treated with the selected arylboronic acids (and prepared pinacol boronates **4** and **10**), Pd(PPh<sub>3</sub>)<sub>4</sub> and Cs<sub>2</sub>CO<sub>3</sub> in a 1,4-dioxane-H<sub>2</sub>O (4:1) mixture at 110 °C to obtain compounds **12–28** in 26–85% yields (Table 1).

#### WILEY-VCH



**Scheme 3.** Synthesis of target compounds **12–39**. Reagents and conditions: (i) arylboronic acid or pinacol boronate,  $Pd(PPh_3)_4$ ,  $Cs_2CO_3$ , 1,4-dioxane-H<sub>2</sub>O (4:1), 110 °C, yields are summarized in Table 1 and Table 2. Being encouraged by the promising data generated by biological evaluation of the first series of compounds (**12–28**, Table 1, see the discussion below), the second series of compounds was designed based on the most potent 4-(benzyloxy)phenyl derivative **18**. Compounds **29–39** (Scheme 3, Table 2) were then prepared in 36–99% yields starting from 4-chloropyrimidine derivative **11** and from various commercially available arylboronic acids under the above described Suzuki-Miyaura cross-coupling conditions. It should be noted that the determined yields were usually based on a single experiment and the reactions were not optimized.

Table 1. Structures and yields of prepared compounds 12–28 (the 1st series) and their effect on *in vitro* production of PGE<sub>2</sub> and viability of mouse peritoneal cells.

		۵٢	vield	inhibition o	viability [%] <sup>[b]</sup>	
entry	compd	Scheme 1	[%]	remaining production (%) <sup>[a]</sup>	IC <sub>50</sub> (μM)	
1	control untreat	NA	NA	2.58±1.96 <sup>[c]</sup>	NA	100.00±1.99 <sup>[c]</sup>
2	control LPS	NA	NA	100.00±4.41	NA	NA
3	1	p <sup>d</sup> 0	NA	12.25±4.48*	<b>4.83</b> <sup>[c]</sup> /3.70-6.32/	102.00±0.89
4	12	Provide the second seco	68	18.60±1.37*	ND	85.00±14.4
5	13	s of the second se	76	18.09±2.63*	ND	104.10±1.90
6	14	3 de la companya de l	63	26.35±0.12*	ND	104.65±1.30
7	15	st of the second	83	12.28±0.50*	<b>9.05</b> /5.53-14.80/	71.17±1.91*
8	16		81	16.36±3.77*	ND	105.15±0.49
9	17		74	10.07±2.62*	<b>6.58</b> /3.90-11.11/	102.58±0.67
10	18		71	0.33±0.17*	<b>0.031</b> /0.018-0.056/	105.40±0.23

WILEY-VCH

## ChemMedChem

# FULL PAPER

.

				<u> 11.</u>		
20	28	st st	71	7.92±1.05	<b>2.70</b> /1.32-5.55/	88.87±0.83
19	27	st Ph	26	53.70±7.34	ND	101.00±0.46
18	26		50	64.32±6.14*	ND	97.63±1.21
17	25	<sup>s<sup>d</sup></sup> ↓ N ↓ O	28	56.11±12.69*	ND	99.25±1.61
16	24	Provide the second seco	77	2.69±0.24*	<b>1.09</b> /0.26-4.67/	102.91±0.32
15	23	Street CF3	68	32.51 ±2.48*	ND	36.68±2.42*
14	22	s <sup>st</sup> O	30	34.62±1.24*	ND	35.59±0.84*
13	21	s <sup>d</sup>	57	15.35±0.34*	ND	45.16±1.25*
12	20	p <sup>d</sup> C O	77	25.60±0.11*	ND	103.99±0.92
11	19	5 <sup>4</sup>	85	30.55±9.77*	ND	99.92±1.02

[a] Effects recorded for 50  $\mu$ M concentration of compounds and expressed in % of control LPS response; [b] Cell viability expressed in % of control untreated cells; [c] ± S.E.M. Results represent means of three to four experiments. Statistical significance: \* P < 0.001, not significant in other values. NA: not applicable. ND: not determined.

Structure–activity relationship study and structural optimization. All prepared polysubstituted pyrimidines were evaluated for their ability to inhibit *in vitro*  $PGE_2$  production using C57BL6 mouse peritoneal cells. The effects of the pyrimidines on the  $PGE_2$ production are expressed as a percentage change (remaining production of  $PGE_2$ ) relative to the response of LPS stimulated cells (positive control, 100%, Table 1 and 2, entry 2) or unstimulated cells (negative control, 2.78%, Table 1 and 2, entry 1).

The first series of tested compounds (12–28, Table 1, entries 4–20) are direct analogues of preclinical candidate 1 (Figure 1, Table 1, entry 3). It was found that majority of new derivatives exhibited similar potency to inhibit PGE<sub>2</sub> production compared to parent compound 1 (with 12% remaining production of PGE<sub>2</sub>). Moreover, the compounds did not affect the viability of the mouse peritoneal cells, with the exception of compounds 21–23 (Table 1, entries 13–15) which decreased the viability of the cells significantly.

However, compound **18** (Table 1, entry 10), the derivative bearing 4-(benzyloxy)phenyl moiety in the C4 position of pyrimidine, excelled in the first series of compounds as the most potent inhibitor of  $PGE_2$  production. Compound **18** almost completely (less than 1% of remaining production of  $PGE_2$ ) inhibited  $PGE_2$  production with no apparent toxicity. This promising result encouraged us to design and synthesize another series of polysubstituted pyrimidines derived from compound **18** and bearing various benzyloxyphenyl-like moieties in the C4 position of the pyrimidine ring.

*In vitro* evaluation of the second series of compounds (**29–39**, Table 2, entries 3–13) revealed compound **32** (Table 2, entry 6), the difluorinated derivative of compound **18**, as the most potent inhibitor of PGE<sub>2</sub> production from the entire SAR study. Compound **32** exhibited a remarkable inhibitory activity with only about 0.04% of remaining production of PGE<sub>2</sub> *in vitro*.

### WILEY-VCH

Accepted Manuscript

# **FULL PAPER**

		Ar	vield	inhibition o		
entry	compd	Scheme 1	[%]	remaining production (%) <sup>[a]</sup>	IC <sub>50</sub> (µM)	viability [%] <sup>ioj</sup>
1	control untreat	NA	NA	2.78±1.02 <sup>[c]</sup>	NA	100.00±2.83 <sup>[c]</sup>
2	control LPS	NA	NA	100.00±3.15	NA	NA
3	29		78	4.01±1.53*	<b>1.64</b> <sup>[c]</sup> /1.38-1.94/	104.24±0.35
4	30	p <sup>d</sup>	71	0.73±0.53*	<b>0.131</b> /0.073-0.235/	101.25±0.84
5	31	2 <sup>nd</sup>	85	51.13±0.83*	ND	103.20±0.52
6	32	F F	70	0.04±0.20*	<b>0.012</b> /0.010-0.017/	103.66±0.44
7	33		98	11.74±2.27*	<b>7.03</b> /4.27-11.56/	101.58±0.74
8	34		99	20.44±1.30*	ND	103.57±0.70
9	35		97	7.12±1.95*	<b>1.65</b> /1.17-2.33/	103.57±0.32
10	36		94	5.54±0.41*	<b>1.31</b> /0.97-1.77/	102.82±0.46
11	37		36	26.29±5.33*	ND	104.82±0.44

Table 2. Structures of prepared compounds 29-39 (the 2nd series) and their effect on in vitro production of PGE2 and viability of mouse peritoneal cells.

This article is protected by copyright. All rights reserved.

### WILEY-VCH



[a] Effects recorded for 50  $\mu$ M concentration of compounds and expressed in % of control LPS response; [b] Cell viability expressed in % of control untreated cells; [c] ± S.E.M. Results represent means of three to four experiments. Statistical significance: \* P < 0.001, not significant in other values. NA: not applicable. ND: not determined.

Changes in production of PGE<sub>2</sub> may be associated with changes in production of another important mediator of inflammation, nitric oxide (NO), although the crosstalk between them is controversial showing both positive and negative influence.<sup>[89,90]</sup> Our previously reported data suggested that some pyrimidine derivatives might act as dual inhibitors of PGE2 and NO production.<sup>[80,91]</sup> Thus, the ability of studied pyrimidines to inhibit in vitro production of NO was also evaluated (data not shown). Mostly insignificant NO suppressive potential was observed in the present set of pyrimidine analogues (including strong PGE<sub>2</sub> inhibitors 18 and 32) compared to the previous series of compounds,<sup>[80]</sup> with the exception of compounds **19**, **22**, and **31**, which mildly inhibited NO production. Thus, we have concentrated on the mode of a predominant effect of pyrimidines, i.e. inhibition of PGE<sub>2</sub> production, although some synergistic anti-inflammatory effects in vivo can be expected and these will be addressed in our ongoing research.

Several important conclusions can be deducted from the overall SAR study: a) an introduction of 4-(benzyloxy)phenyl moiety at the C4 position of pyrimidine ring was beneficial for the increased inhibition of PGE<sub>2</sub> production (compound **18**, Table 1, entry 10); b) further substitution of the benzyloxy moiety in the *para* position of **18** was well-tolerated (compounds **29** and **30**, Table 2, entries 3 and 4); c) an introduction of the methyl group into *ortho* position of the C4 phenyl ring of **18** resulted in a substantial decrease of biological potency (compound **31**, Table 2, entry 5); d) an introduction of fluorine atoms into the phenyl ring of the 4-(benzyloxy)phenyl moiety in compound **18** resulted in almost one-order of magnitude more potent inhibitor **32** (Table

2, entry 6); e) moving the benzyloxy moiety from *para* to *meta* position of the C4 phenyl ring led to a decreased inhibition of  $PGE_2$  production (compounds **33** and **34**, Table 2, entries 7 and 8); f) a replacement of an oxygen atom in compound **18** for sulfur resulted in a decrease of biological potency (compound **39**, Table 2, entry 13).

Finally, for the set of selected compounds, namely compounds **1**, **15**, **17**, **18**, **24** and **28** (Table 1) and **29**, **30**, **32**, **33**, **35**, **36**, **38** and **39** (Table 2), the IC<sub>50</sub> values were determined. The data showed that most potent compounds **18** (IC<sub>50</sub> = 0.031  $\mu$ M) and **32** (IC<sub>50</sub> = 0.012  $\mu$ M) were approx. 150-fold and 400-fold more potent inhibitors of PGE<sub>2</sub> production, respectively, compared to the parent compound **1** (IC<sub>50</sub> = 4.83  $\mu$ M).

Evaluation of the mechanism of action. The mechanism of the PGE<sub>2</sub>-inhibitory effects of pyrimidines was studied using the most potent compound **32** at first. Applied at the concentration of 20  $\mu$ M, **32** did not influence the activity of COX-1, whereas it significantly suppressed (by 47%, P < 0.001) the activity of COX-2 (Figure 2A). Interestingly, other compounds from this study (including **1**, **18**, and **28**) did not exhibit any noticeable activity against COX-1/2 enzymes (data not shown). On the other hand, compound **32** (at 20  $\mu$ M) virtually completely inhibited (by 98%) the activity of human mPGES-1 (Figure 2A). It was inhibited in a dose-dependent manner (Figure 2B), the IC<sub>50</sub> estimate being 0.066  $\mu$ M, with 0.047– 0.082  $\mu$ M 95% limits of confidence. The data indicate that **32** is a dual inhibitor of COX-2 and mPGES-1, although its interaction with mPGES-1 can be considered as the major mechanism underlying the inhibition of PGE<sub>2</sub> production.



Figure 2. A) Interaction of compound 32 (20  $\mu$ M) with COX-1/2 and human mPGES-1. The effects are compared to those of reference standards, *i.e.* indomethacin (5  $\mu$ M) for COX enzymes and CAY10678 (2  $\mu$ M) for mPGES-1. The bars are means ± S.E.M. obtained from two COX and six mPGES-1 experiments, each run in duplicate. Statistical significance against the control groups: \*\* *P* < 0.01, \*\*\* *P* < 0.001. B) Inhibition of mPGES-1 activity is dependent on the concentration of 32. The curve represents amalgamated effects of two experiments, each done in duplicate. The points are means ± S.E.M.

anuscr

Table	3.	Direct	inhibition	of	mouse	and	human	mPGES-1	with	selected
compo	unc	ls.								

		mPGES-1 inhibition <sup>[a]</sup> IC <sub>50</sub> ( $\mu$ M) (95% limits of confidence)				
Entry	Comp	mouse	human			
1	1	12.780 (9.600 - 54.120)	5.065 (4.071 - 6.303)			
2	18	0.060 (0.045 - 0.307)	0.117 (0.111 - 0.259)			
3	28	6.280 (1.760 - 15.010)	2.423 (1.341 - 4.378)			
4	30	0.700 (0.360 - 2.130)	0.248 (0.160 - 0.385)			
5	32	2.730 (1.760 - 7.400)	0.066 (0.047 - 0.082)			

[a] The enzyme activity reflects changes in transformation of PGH2 to of PGE2. Effects of pyrimidine derivatives were recorded for recombinant enzymes as described in Experimental Procedures. The results represent two independent experiments.

In order to verify that mPGES-1 inhibition is the key mechanism of action common to the whole series of pyrimidine analogues studied (since only **32** partially inhibits also COX-2), we decided to evaluate activity in the mPGES-1 assay with other compounds, namely **1**, **18**, **28**, and **30**. Interestingly, previously described inhibitors of human mPGES-1 were not, in general, potent inhibitors of the mouse or rat enzyme, and only recently, dual inhibitors, active against both human and mouse enzymes, have been reported.<sup>[92]</sup> In order to correlate our data obtained on mouse peritoneal cells (Table 1 and 2) with those obtained with human mPGES-1 (Fig. 2), selected compounds **1**, **18**, **28**, **30**, and **32** were evaluated on both mouse and human enzymes side by side (Table 3).

All tested compounds are potent inhibitors of both mouse and human mPGES-1 (Table 3), though the human enzymes seems to be more susceptible to the inhibition with polysubstituted pyrimidines. Coumpound **18** (IC<sub>50</sub> = 60 nM) is the most potent inhibitor of the mouse mPGES-1, while compound **32** (IC<sub>50</sub> = 66 nM) is the most potent inhibitor of the human enzyme. The obtained data support the notion that mPGES-1 inhibition represents the main mechanism of action common to anti-inflammatory pyrimidines studied in this work.

In vivo evaluation of selected compounds in the model of acute inflammation. For the most potent compounds, namely 18, 30, and 32, we carried out the carrageenan-induced rat paw edema experiments to verify whether the compounds are active also in vivo when administered orally (gastric intubation). The rat model is preferred to mouse one for its higher accuracy and reproducibility. Since the aqueous solubility of most of the studied polysubstituted pyrimidines is low, the compounds were solubilized using a mixture of DMSO (1.5%) in TWEEN® 80 before dilution in the physiological saline solution. The results are depicted in Figure 3. It was shown that solvent alone did not have any significant effect on the development of edema induced by carrageenan while solubilized compounds 18, 30, and 32 reduced the carrageenan-enhanced paw volumes by 36.3%, 20.7%, and 45.5%, respectively. It can be concluded, that the compounds are bioavailable after oral administration. The most potent compound 32 substantially suppresses the development of rat paw edema validating its potential to become





**Figure 3.** Effect of compounds **18, 30**, and **32** on the development of the acute carrageenan-induced rat paw edema. P, statistical significance against the positive control group (carrageenan alone). Solvent: DMSO (1.5 wt%)/TWEEN<sup>®</sup> 80 mixture (100 mg) in sterile 1% saline solution (1 mL). Compound samples: 5 mg of compound in 1 mL of solvent.

Computational Studies. The detailed structure-activity relationship (SAR) of anti-inflammatory pyrimidines was discussed above. In order to elucidate and better understand the binding mode of mPGES-1 inhibitors, we decided to perform molecular docking studies with some of our compounds. High resolution X-ray crystal structure (PDB: 4BPM) of human mPGES-1 with potent inhibitor is available,<sup>[93,94]</sup> while mouse mPGES-1 structure is so far unavailable. Human mPGES-1 is, however, highly homologous to the mouse envzme, with sequence identity being 79% and the sequence similarity being 84%. Thus, docking experiments of compounds 1 and 32 were carried out using Glide based on the X-ray crystal structure of human mPGES-1 (PDB: 4BPM) and on homology model of mouse mPGES-1 developed by using the human mPGES-1 structure as template (Figure 4). The molecular docking revealed that both compounds 1 and 32 can favorably bind in conserved region of the active site of human and mouse mPGES-1 enzymes. The conserved region around S127 has mainly hydrophobic pocket surrounded by Y28, I32 (V32 in mouse), G35, L39, Y130, T131 (V131 in mouse), Q134, L135 and A138 (F138 in mouse) for human (mouse) mPGES-1.[95] As shown in Figure 4, the phenyl ring in C6 position of pyrimidine ring stays at bottom of the substrate-binding pocket of mPGES-1 enzymes. 4-Methoxyphenyl (compound 1) and 4-benzyloxy(3,2difluoro)phenyl (compound 32) moiety in C4 position of pyrimidine are involved in a  $\pi$ - $\pi$  stacking interaction with the side chain of Tyr130 of human subunit A and the side chain of Tyr130 of mouse subunit C, respectively. The computational studies support strong binding of studied derivatives to both mouse and human mPGES-1 enzymes.

ChemMedChem

#### **FULL PAPER**

Scepte



Figure 4. Docked binding pose of A) compound 1 and B) compound 32 in human mPGES-1. Docked binding pose of C) compound 1 and D) compound 32 in modeled mouse mPGES-1. The trimer protein structure of human and modeled mouse mPGES-1 is depicted in ribbon diagram. Human and modeled mouse mPGES-1 interacting residues are represented green stick form and ligand structure in yellow stick form (colored by atom type: C: green or yellow, N: blue, O: red, F: cyan, H: white). The cyan dashed lines indicate  $\pi$ - $\pi$  interaction. Figure was prepared with Maestro, Schrodinger, LLC.

#### Conclusion

Compound 1 was identified previously as a potent inhibitor of prostaglandin E2 (PGE2) production with confirmed antiinflammatory properties and is being evaluated as a preclinical candidate for the treatment of ulcerative colitis and rheumatoid arthritis. The goal of the current extensive SAR study was further optimization of polysubstituted structural pyrimidines, identification of more potent anti-inflammatory agents, as well as elucidation of their mechanism of action. Synthesis of 28 new analogues derived from compound 1 bearing substituted and/or modified phenyl moiety in the C4 position of the pyrimidine ring, led to the discovery of several superior inhibitors of PGE2 production. Firstly, compound 18 (IC<sub>50</sub> = 31 nM) with the 4-(4benzyloxy)phenyl arm in C4 position of pyrimidine was found to be a potent inhibitor of PGE2 production. Subsequently, compound 32, the difluorinated derivative of compound 18, was identified as the most potent inhibitor from the whole SAR series. Compound 32 (IC<sub>50</sub> = 12 nM) was 400-fold more potent inhibitor of PGE<sub>2</sub> production compared to parent compound 1 (IC<sub>50</sub> =  $4.83 \mu$ M). Moreover, after oral administration, compounds 18 and 32 significantly (by 36% and 46%, respectively) suppressed the development of carrageenan-induced rat paw edema, confirming in vivo potency in the treatment of acute inflammation. An extensive study of mechanism of action revealed that the studied compounds (1, 18, 28, 30 and 32) are potent inhibitors of mPGES-1, both mouse and human. Therefore, these inhibitors can be used as tool compounds to study the effects of mPGES-1 inhibition in mice and rats, with potential applications in humans. Moreover, it has been suggested, that the specific inhibition of mPGES-1 is extremely interesting for drug development with respect to the side-effects observed with COX-1/2 inhibitors. The pyrimidines studied did not inhibit COX-1/2 enzymes, only derivative 32 weakly inhibited COX-2. Although the overall mode of anti-inflammatory action of the studied compounds seems to be relatively complex, mPGES-1 evidently is their principal anti-inflammatory target. These exciting data strongly encourage future evaluation of the most potent candidates in various chronic inflammation models (e.g. ulcerative colitis and rheumatoid arthritis) as well as subsequent lead optimization in order to develop novel potent anti-inflammatory and anti-cancer therapeutic agents.

### **Experimental Section**

Experimental details are given in the Supporting information. All protocols were approved by the institutional ethics committee.

#### Acknowledgements

This work was supported by the subvention for development of research organization (Institute of Organic Chemistry and Biochemistry, Czech Academy of Sciences, RVO: 61388963), by the Technology Agency of the Czech Republic (TE01020028 and IT4Innovations National Supercomputing Center – LM2015070). We are grateful to Vladimír Král (Faculty of Chemical Engineering, University of Chemistry and Technology, Prague) for help with selection and preparation of suitable formulation used for the in vivo experiments.

Abbreviations arachidonic COX, used. AA. acid: cyclooxygenase; GI, gastrointestinal; LDH, lactate dehydrogenase; LPS, lipopolysaccharide; NBS, Nbromosuccinimide; NO, nitric oxid, NSAIDs, non-steroidal antiinflammatory drugs; PGE<sub>2</sub>, prostaglandin E<sub>2</sub>; mPGES-1, microsomal prostaglandin E2 synthase; t-NSAIDs, traditional non-steroidal anti-inflammatory drugs.

**Keywords:** pyrimidines • Suzuki Miyaura reaction • mPGES-1 • prostaglandin  $E_2$  • anti-inflammatory

- C. Pereira-Leite, C. Nunes, S. K. Jamal, I. M. Cuccovia, S. Reis, *Med. Res. Rev.* 2017, 37, 802–859.
- [2] D. L. Simmons, *Pharmacol. Rev.* **2004**, *56*, 387–437.
- [3] A. L. Blobaum, L. J. Marnett, J. Med. Chem. 2007, 50, 1425–1441.
- [4] W. L. Smith, Y. Urade, P. J. Jakobsson, *Chem. Rev.* 2011, 111, 5821– 5865.
- [5] B. J. R. Whittle, *Fundam. Clin. Pharmacol.* **2003**, *17*, 301–313.
- [6] G. Coruzzi, N. Venturi, S. Spaggiari, Acta Biomed. l'Ateneo Parm. 2007, 78, 96–110.
- [7] P. N. P. Rao, E. E. Knaus, T. P. Road, L. Jolla, J. Pharm. Pharm. Sci. 2008, 11, 81–110.
- [8] J.-M. Dogné, C. T. Supuran, D. Pratico, J. Med. Chem. 2005, 48, 2251– 2257.
- [9] A. S. Kalgutkar, A. B. Marnett, B. C. Crews, R. P. Remmel, L. J. Marnett, J. Med. Chem. 2000, 43, 2860–2870.
- [10] A. S. Kalgutkar, B. C. Crews, S. W. Rowlinson, A. B. Marnett, K. R. Kozak, R. P. Remmel, L. J. Marnett, *Proc. Natl. Acad. Sci.* 2000, *97*, 925–930.
- [11] A. Palomer, F. Cabré, J. Pascual, J. Campos, M. A. Trujillo, A. Entrena, M. A. Gallo, L. García, D. Mauleón, A. Espinosa, *J. Med. Chem.* 2002, 45, 1402–1411.
- [12] A. S. Kalgutkar, B. C. Crews, S. Saleh, D. Prudhomme, L. J. Marnett, *Bioorg. Med. Chem.* **2005**, *13*, 6810–6822.
- [13] N. Yamakawa, S. Suemasu, M. Matoyama, A. Kimoto, M. Takeda, K. I. Tanaka, T. Ishihara, T. Katsu, Y. Okamoto, M. Otsuka, T. Mizushima, J. Med. Chem. 2010, 53, 7879–7882.
- [14] N. Yamakawa, S. Suemasu, Y. Okamoto, K. I. Tanaka, T. Ishihara, T. Asano, K. Miyata, M. Otsuka, T. Mizushima, *J. Med. Chem.* 2012, 55, 5143–5150.
- [15] S. K. Suthar, M. Sharma, Med. Res. Rev. 2015, 35, 341-407.
- [16] S. Fiorucci, L. Santucci, E. Distrutti, *Dig. Liver Dis.* 2007, 39, 1043– 1051.
- [17] L. M. Lichtenberger, J. J. Romero, E. J. Dial, J. E. Moore, Inflammopharmacology 2009, 17, 1–5.

- [18] L. M. Lichtenberger, J. J. Romero, E. J. Dial, Br. J. Pharmacol. 2009, 157, 252–257.
- [19] L. Huang, G. G. MacKenzie, N. Ouyang, Y. Sun, G. Xie, F. Johnson, D. Komninou, B. Rigas, Br. J. Pharmacol. 2011, 162, 1521–1533.
- [20] A. H. Abadi, G. H. Hegazy, A. A. El-Zaher, *Bioorg. Med. Chem.* 2005, 13, 5759–5765.
- [21] U. K. Bandarage, L. Chen, X. Fang, D. S. Garvey, A. Glavin, D. R. Janero, L. G. Letts, G. J. Mercer, J. K. Saha, J. D. Schroeder, M. J. Shumway, S. W. Tam, *J. Med. Chem.* **2000**, *43*, 4005–4016.
- [22] C. Cena, P. Tosco, E. Marini, L. Lazzarato, M. Piccinini, C. Ramondetti, E. Lupino, R. Fruttero, A. Gasco, *ChemMedChem* 2011, 6, 523–530.
- [23] C. Cena, M. L. Lolli, L. Lazzarato, E. Guaita, G. Morini, G. Coruzzi, S. P. McElroy, I. L. Megson, R. Fruttero, A. Gasco, J. Med. Chem. 2003, 46 (5), 747–754.
- [24] K. Chegaev, L. Lazzarato, P. Tosco, C. Cena, E. Marini, B. Rolando, P. A. Carrupt, R. Fruttero, A. Gasco, J. Med. Chem. 2007, 50, 1449–1457.
- [25] S. Consalvi, G. Poce, R. Ragno, M. Sabatino, C. La Motta, S. Sartini, V. Calderone, A. Martelli, C. Ghelardini, L. Di Cesare Mannelli, M. Biava, *ChemMedChem* **2016**, *11*, 1804–1811.
- [26] Z. Huang, C. A. Velázquez, K. R. A. Abdellatif, M. A. Chowdhury, J. A. Reisz, J. F. Dumond, S. B. King, E. E. Knaus, *J. Med. Chem.* 2011, *54*, 1356–1364.
- [27] M. L. Lolli, C. Cena, C. Medana, L. Lazzarato, G. Morini, G. Coruzzi, S. Manarini, R. Fruttero, A. Gasco, J. Med. Chem. 2001, 44, 3463–3468.
- [28] R. R. Ranatunge, M. Augustyniak, U. K. Bandarage, R. A. Earl, J. L. Ellis, D. S. Garvey, D. R. Janero, L. G. Letts, A. M. Martino, M. G. Murty, S. K. Richardson, J. D. Schroeder, M. J. Shumway, S. W. Tam, A. M. Trocha, D. V. Young, *J. Med. Chem.* **2004**, *47*, 2180–2193.
- [29] R. R. Ranatunge, M. E. Augustyniak, V. Dhawan, J. L. Ellis, D. S. Garvey, D. R. Janero, L. G. Letts, S. K. Richardson, M. J. Shumway, A. M. Trocha, D. V. Young, I. S. Zemtseva, *Bioorg. Med. Chem.* 2006, *14*, 2589–2599.
- [30] I. T. Schiefer, S. Abdul-Hay, H. Wang, M. Vanni, Z. Qin, G. R. J. Thatcher, J. Med. Chem. 2011, 54, 2293–2306.
- [31] C. A. Velázquez, P. N. P. Rao, E. E. Knaus, J. Med. Chem. 2005, 48, 4061–4067.
- [32] C. A Velázquez, P. N. Praveen Rao, M. L. Citro, L. K. Keefer, E. E. Knaus, *Bioorg. Med. Chem.* 2007, 15, 4767–4774.
- [33] C. A. Velázquez, Q.-H. Chen, M. L. Citro, L. K. Keefer, E. E. Knaus, J. Med. Chem. 2008, 51, 1954–1961.
- [34] J. Kaur, A. Bhardwaj, Z. Huang, E. E. Knaus, *ChemMedChem* 2012, 7, 144–150.
- [35] D. Basudhar, G. Bharadwaj, R. Y. Cheng, S. Jain, S. Shi, J. L. Heinecke, R. J. Holland, L. A. Ridnour, V. M. Caceres, R. C. Spadari-Bratfisch, N. Paolocci, C. A. Valázquez-Martínez, D. A. Wink, K. M. Miranda, J. Med. Chem. 2013, 56, 7804–7820.
- [36] L. Lazzarato, K. Chegaev, E. Marini, B. Rolando, E. Borretto, S. Guglielmo, S. Joseph, A. Di Stilo, R. Fruttero, A. Gasco, *J. Med. Chem.* 2011, 54, 5478–5484.
- [37] J. Sidhapuriwala, L. Li, A. Sparatore, M. Bhatia, P. K. Moore, *Eur. J. Pharmacol.* 2007, 569, 149–154.
- [38] M. Magierowski, K. Magierowska, M. Surmiak, M. Hubalewska-Mazgaj, S. Kwiecien, J. L. Wallace, T. Brzozowski, *J. Physiol. Pharmacol.* 2017, 68, 749–756.
- [39] A. E. Dief, D. K. Mostafa, G. M. Sharara, T. H. Zeitoun, *Eur. Rev. Med. Pharmacol. Sci.* 2015, *19*, 1537–1546.
- [40] M. V. Chan, J. L. Wallace, AJP Gastrointest. Liver Physiol. 2013, 305, G467–G473.
- [41] S. Fiorucci, S. Orlandi, A. Mencarelli, G. Caliendo, V. Santagada, E. Distrutti, L. Santucci, G. Cirino, J. L. Wallace, *Br. J. Pharmacol.* 2007, 150, 996–1002.
- [42] J. L. Wallace, G. Caliendo, V. Santagada, G. Cirino, Br. J. Pharmacol. 2010, 159, 1236–1246.
- [43] S. Jain, S. Tran, M. A. M. El Gendy, K. Kashfi, P. Jurasz, C. A. Velázquez-Martínez, J. Med. Chem. 2012, 55, 688–696.
- [44] B. Samuelsson, R. Morgenstern, P. Jakobsson, *Pharmacol. Rev.* 2007, 59, 207–224.

- [45] J. Bauer, B. Waltenberger, S. M. Noha, D. Schuster, J. M. Rollinger, J.
- Boustie, M. Chollet, H. Stuppner, O. Werz, *ChemMedChem* 2012, 7, 2077–2081.
  [46] F. Rörsch, I. Wobst, H. Zettl, M. Schubert-Zsilavecz, S. Grösch, G.
- Geisslinger, G. Schneider, E. Proschak, J. Med. Chem. 2010, 53, 911– 915.
- [47] F. Rörsch, E. La Buscató, K. Deckmann, G. Schneider, M. Schubert-Zsilavecz, G. Geisslinger, E. Proschak, S. Grösch, J. Med. Chem. 2012, 55, 3792–3803.
- [48] A. Koeberle, E. M. Haberl, A. Rossi, C. Pergola, F. Dehm, H. Northoff, R. Troschuetz, L. Sautebin, O. Werz, *Bioorg. Med. Chem.* 2009, 17, 7924–7932.
- [49] G. Lauro, P. Tortorella, A. Bertamino, C. Ostacolo, A. Koeberle, K. Fischer, I. Bruno, S. Terracciano, I. M. Gomez-Monterrey, M. Tauro, F. Loiodice, E. Novellino, R. Riccio, O. Werz, P. Campiglia, G. Bifulco, *ChemMedChem* **2016**, *11*, 612–619.
- [50] S. He, C. Li, Y. Liu, L. Lai, J. Med. Chem. 2013, 56, 3296–3309.
- [51] J. G. Luz, S. Antonysamy, S. L. Kuklish, B. Condon, M. R. Lee, D. Allison, X. P. Yu, S. Chandrasekhar, R. Backer, A. Zhang, M. Russell, S. S. Chang, A. Harvey, A. V. Sloan, M. J. Fisher, *J. Med. Chem.* 2015, 58, 4727–4737.
- [52] K. M. Partridge, S. Antonysamy, S. N. Bhattachar, S. Chandrasekhar, M. J. Fisher, A. Fretland, K. Gooding, A. Harvey, N. E. Hughes, S. L. Kuklish, J. G. Luz, P. R. Manninen, J. E. McGee, D. R. Mudra, A. Navarro, B. H. Norman, S. J. Quimby, M. A. Schiffler, A. V. Sloan, A. M. Warshawsky, J. M. Weller, J. S. York, X.-P. Yu. *Bioorg. Med. Chem. Lett.* 2017, *27*, 1478–1483.
- [53] A. J. Liedtke, P. R. W. E. F. Keck, F. Lehmann, A. Koeberle, O. Werz, S. A. Laufer, *J. Med. Chem.* **2009**, *52*, 4968–4972.
- [54] R. W. Friesen, J. A. Mancini, J. Med. Chem. 2008, 51, 4059–4067.
- [55] A. Kats, T. Båge, P. Georgsson, J. Jönsson, H. C. Quezada, A. Gustafsson, L. Jansson, C. Lindberg, K. Näsström, T. Yucel-Lindberg, *FASEB J.* 2013, 27, 2328–2341.
- [56] A. Wiegard, W. Hanekamp, K. Griessbach, J. Fabian, M. Lehr, *Eur. J. Med. Chem.* 2012, 48, 153–163.
- [57] S. Chandrasekhar, A. K. Harvey, X.-P. Yu, M. G. Chambers, J. L. Oskins, C. Lin, T. W. Seng, S. J. Thibodeaux, B. H. Norman, N. E. Hughes, M. A. Schiffler, M. J. Fisher, *J. Pharmacol. Exp. Ther.* **2016**, 356, 635–644.
- [58] A. Koeberle, O. Werz, Biochem. Pharmacol. 2015, 98, 1–15.
- [59] S. Di Micco, S. Terracciano, V. Cantone, K. Fischer, A. Koeberle, A. Foglia, R. Riccio, O. Werz, I. Bruno, G. Bifulco, *Eur. J. Med. Chem.* 2018, 143, 1419–1427.
- [60] K. Ding, Z. Zhou, S. Zhou, Y. Yuan, K. Kim, T. Zhang, X. Zheng, F. Zheng, C.-G. Zhan, *Bioorg. Med. Chem. Lett.* **2018**, *28*, 858–862.
- [61] S. L. Kuklish, S. Antonysamy, S. N. Bhattachar, S. Chandrasekhar, M. J. Fisher, A. J. Fretland, K. Gooding, A. Harvey, N. E. Hughes, J. G. Luz, P. R. Manninen, J. E. McGee, A. Navarro, B. H. Norman, K. M. Partridge, S. J. Quimby, M. A. Schiffler, A. V. Sloan, A. M. Warshawsky, J. S. York, X.-P. Yu, *Bioorg. Med. Chem. Lett.* **2016**, *26*, 4824–4828.
- [62] A. Koeberle, E. Muñoz, G. B. Appendino, A. Minassi, S. Pace, A. Rossi, C. Weinigel, D. Barz, L. Sautebin, D. Caprioglio, J. A. Collado, O. Werz, *J. Med. Chem.* 2014, 57, 5638–5648.
- [63] D. L. Flynn, H. Capiris, W. J. Cetenko, D. T. Connor, R. D. Dyer, C. R. Kostlan, D. E. Nies, D. J. Schrier, J. C. Sircar, *J. Med. Chem.* **1990**, *33*, 2070–2072.
- [64] M. Inagaki, T. Tsuri, H. Jyoyama, T. Ono, K. Yamada, M. Kobayashi, Y. Hori, A. Arimura, K. Yasui, K. Ohno, S. Kakudo, K. Koizumi, R. Suzuki, M. Kato, S. Kawai, S. Matsamuto, *J. Med. Chem.* **2000**, *43*, 2040–2048.
- [65] J. M. Janusz, P. A. Young, J. M. Ridgeway, M. W. Scherz, K. Enzweiler, L. I. Wu, L. Gan, R. Darolia, R. S. Matthews, D. Hennes, D. E. Kellstein, S. A. Green, J. L. Tullich, T. Rosario-Jansen, I. J. Magrisso, K. R. Wehmeyer, D. L. Kuhlenbeck, T. H. Eichhold, R. L. M. Dobson, S. P. Sirko, R. W. Farmer, J. Med. Chem. **1998**, *41*, 1112–1123.
- [66] D. Lokwani, R. Azad, A. Sarkate, P. Reddanna, D. Shinde, *Bioorg. Med. Chem.* 2015, 23, 4533–4543.
- [67] M. Zheng, Z. Zhang, W. Zhu, H. Liu, X. Luo, K. Chen, H. Jiang, *Bioorg. Med. Chem.* 2006, *14*, 3428–3437.

- [68] M. Elkady, R. Nieß, A. M. Schaible, J. Bauer, S. Luderer, G. Ambrosi, O. Werz, S. A. Laufer, *J. Med. Chem.* **2012**, *55*, 8958–8962.
- [69] T. Hanke, F. Dehm, S. Liening, S. D. Popella, J. MacZewsky, M. Pillong, J. Kunze, C. Weinigel, D. Barz, A. Kaiser, M. Wurglick, M. Lämmerhofer, G. Schneider, L. Sautebin, M. Schubert-Zsilavecz, O. Werz, *J. Med. Chem.* **2013**, *56*, 9031–9044.
- [70] A. Koeberle, H. Zettl, C. Greiner, M. Wurglics, M. Schubert-Zsilavecz, O. Werz, J. Med. Chem. 2008, 51, 8068–8076.
- [71] F. Bruno, S. Errico, S. Pace, M. B. Nawrozkij, A. S. Mkrtchyan, F. Guida, R. Maisto, A. Olgaç, M. D'Amico, S. Maione, M. De Rosa, E. Banoglu, O. Werz, A. Fiorentino, R. Filosa, *Eur. J. Med. Chem.* **2018**, 155, 946–960.
- [72] S.-Y. Cheung, M. Werner, L. Esposito, F. Troisi, V. Cantone, S. Liening, S. König, J. Gerstmeier, A. Koeberle, R. Bilancia, R. Rizza, A. Rossi, F. Roviezzo, V. Temml, D. Schuster, H. Stuppner, M. Schubert-Zsilavecz, O. Werz, T. Hanke, S. Pace, *Eur. J. Med. Chem.* **2018**, 156, 815–830.
- [73] G. Lauro, S. Terracciano, V. Cantone, D. Ruggiero, K. Fischer, S. Pace, O. Werz, I. Bruno, G. Bifulco, *ChemMedChem* **2020**, *15*, 481–489.
- [74] M. Bertinaria, M. A. A. G. Shaikh, C. Buccellati, C. Cena, B. Rolando, L. Lazzarato, R. Fruttero, A. Gasco, M. Hoxha, V. Capra, A. Sala, G. E. Rovati, *ChemMedChem* 2012, 7, 1647–1660.
- [75] G. Palermo, A. D. Favia, M. Convertino, M. De Vivo, *ChemMedChem* 2016, *11*, 1252–1258.
- [76] R. Scarpelli, O. Sasso, D. Piomelli, ChemMedChem 2016, 11, 1242– 1251.
- [77] P. Jansa, A. Holý, Z. Zídek, E. Kmoníčková, Z. Janeba, Pyrimidine Compounds Inhibiting the Formation of Nitric Oxide and Prostaglandin E2, Method of Production Thereof and Use Thereof. 2012, WO2012/116666.
- [78] P. Jansa, A. Holý, M. Dračínský, V. Kolman, Z. Janeba, P. Kostecká, E. Kmoníčková, Z. Zídek, *Med. Chem. Res.* 2014, 23, 4482–4490.
- [79] P. Jansa, A. Holý, M. Dračínský, V. Kolman, Z. Janeba, E. Kmoníčková, Z. Zídek, *Med. Chem. Res.* 2015, *24*, 2154–2166.
- [80] V. Kolman, P. Jansa, F. Kalčic, Z. Janeba, Z. Zídek, Nitric Oxide 2017, 67, 53–57.
- [81] V. Kolman, F. Kalčic, P. Jansa, Z. Zídek, Z. Janeba, Eur. J. Med. Chem. 2018, 156, 295–301.
- [82] M. C. Carreno, J. L. Garcia Ruano, G. Sanz, M. A. Toledo, A. Urbano, J. Org. Chem. 1995, 60, 5328–5331.
- [83] W. Li, D. P. Nelson, M. S. Jensen, R. S. Hoerrner, D. Cai, R. D. Larsen, P. J. Reider, J. Org. Chem. 2002, 67, 5394–5397.
- [84] K. Harada, K. Makino, N. Shima, H. Okuyama, T. Esumi, M. Kubo, H. Hioki, Y. Asakawa, Y. Fukuyama, *Tetrahedron* 2013, *69*, 6959–6968.
- [85] O. Saku, M. Saki, M. Kurokawa, K. Ikeda, T. Takizawa, N. Uesaka, Bioorg. Med. Chem. Lett. 2010, 20, 1090–1093.
- [86] D. Bogdal, M. Warzala, *Tetrahedron* **2000**, *56*, 8769–8773.
- [87] Y. Yamaguchi, I. Akimoto, K. Motegi, T. Yoshimura, K. Wada, N. Nishizono, K. Oda, *Chem. Pharm. Bull.* **2013**, *61*, 997–1001.
- [88] J. H. Musser, U. Chakraborty, K. Bailey, S. Sciortino, C. Whyzmuzis, D. Amin, C. A. Sutherland, J. Med. Chem. 1987, 30, 62–67.
- [89] K. Raddassi, J. F. Petit, G. Lemaire, Cell. Immunol. 1993, 149, 50-64.
- [90] M. G. Schwacha, S. D. Somers, *J. Leuko. Biol.* **1998**, 63, 51–58.
- [91] Z.Zídek, M. Kverka, A. Dusilová, E. Kmoníčková, P. Jansa, Nitric oxide 2016, 57, 48–56.
- [92] K. Ding, Z. Zhou, S. Hou, Y. Yuan, S. Zhou, X. Zheng, J. Chen, C. Loftin, F. Zheng, C.-G. Zhan, Sci. Rep. 2018, 8, 5205.
- [93] T. Sjögren, J. Nord, M. Ek, P. Johansson, G. Liu, S. Geschwindner, Proc. Natl. Acad. Sci. U. S. A. 2013, 110, 3806–3811.
- [94] D. Li, N. Howe, A. Dukkipati, S. T. Shah, B. D. Bax, C. Edge, A. Bridges, P. Hardwicke, O. M. Singh, G. Giblin, A. Pautsch, R. Pfau, G. Schnapp, M. Wang, V. Olieric, M. Caffrey, *Cryst. Growth Des.* **2014**, *14*, 2034– 2047.
- [95] A. Hamza, X. Zhao, M. Tong, H. H. Tai, C. G. Zhan, *Bioorg. Med. Chem.* 2011, 19, 6077–6086.

### WILEY-VCH

### Entry for the Table of Contents



Anti-inflammatory drugs belong to the mostly used therapeutic agents, but both traditional NSAIDs and coxibs suffer from serious side effects. Thus, development of anti-inflammatory drugs with novel mechanisms of actions is desirable and inhibitors of mPGES-1 may be good approach. We designed and synthesized potent mPGES-1 inhibitors based on polysubstituted pyrimidines, which showed strong anti-inflammatory effects *in vivo*.