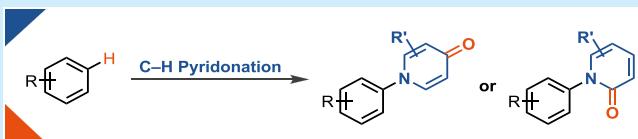


C–H Pyridonation of (Hetero-)Arenes by Pyridinium Radical Cations

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S Supporting Information

ABSTRACT: Pyridones are important heteroaromatic scaffolds found in natural products and pharmaceuticals and are, therefore, of major interest in organic synthetic chemistry. Here we report the first C–H pyridonation of unactivated (hetero-)arenes, providing a methodology to directly access N-aryl-2- and 4-pyridones. Generation of pyridinium radical cations through single-electron reduction allows for the synthesis of pyridones on structurally complex molecules.



Nitrogen heterocycles represent one of the most important structural scaffolds in pharmaceuticals and are found in 59% of all small-molecule drugs approved by the FDA.¹ Pyridine is the most commonly used aromatic heterocycle and has been extensively studied in terms of metabolism *in vivo*.^{2,3} Oxidation or methylation pathways can produce highly toxic metabolites of pyridine.^{2,3} N-Arylpyridones are less prone to oxidation or methylation pathways and are well recognized in medicinal chemistry for their diverse biological activities.^{4–7} A large set of N-arylpyridone structures can be found in marketed drugs such as anti-inflammatory drug pirfenidone for the treatment of idiopathic pulmonary fibrosis (IPF).^{8–11} However, significant challenges are encountered for installing pyridones in the presence of sensitive functional groups through traditional cross-coupling protocols. Due to the low nucleophilicity of 2- and 4-pyridones, forcing reaction conditions and an excess of the arene coupling partner are required.^{12–15} C–H functionalization has been extensively used to functionalize pyridones via C–C bond formation (Figure 1),^{16–25} but direct substitution of an aryl C–H bond with the nitrogen atom of a pyridone is unprecedented.

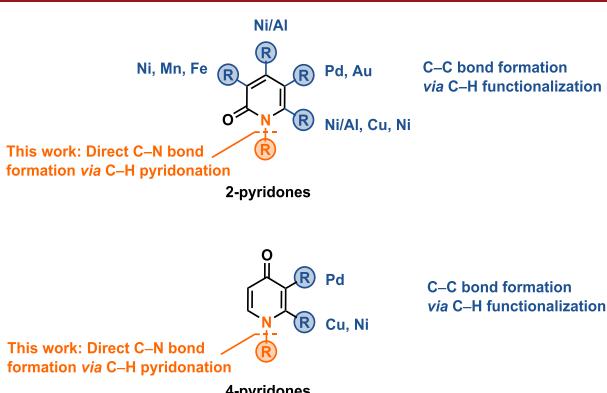


Figure 1. C–H functionalization on pyridones.

Here, we present the first direct C–H pyridonation reaction of (hetero-)arenes as the limiting reagent.

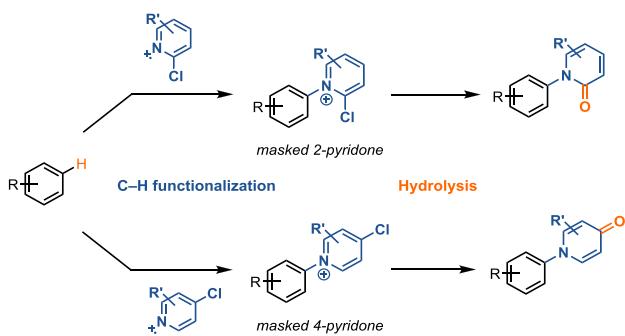
The traditional approach for the synthesis of N-arylpyridones involves copper-catalyzed coupling between 2- or 4-pyridones and a coupling partner such as an arylboronic acid,¹² an aryl stannane,¹³ an aryl bismuth,¹⁴ an aryl iodonium salt,¹⁵ or an aryl halide.²⁶ However, due to the low nucleophilicity and the ambidentate nature of pyridones, such methods often require high temperatures^{14,26} or excess of the arene coupling partner^{12–15} or give mixtures of N-/O-arylated products.^{14,15,26} N-Arylpyridones can also be prepared through ring construction including condensation reactions,^{27,28} oxidative annulation,²⁹ metal-mediated cycloaddition,³⁰ or Vilsmeier–Haack cyclization.³¹ Harsh reaction conditions and/or multistep procedures of the currently utilized methods limit the synthetic availability of N-arylpyridones.^{27–31}

On the basis of our previous work on direct C–H amination of arenes,^{32,33} we have evaluated different nitrogen electrophiles for addition to arenes. Earlier this year, Carreira and Togni as well as our laboratory have independently demonstrated the ability of pyridinium radical cations to react with arenes to afford N-arylpyridinium salts.^{34,35} Here we show that 2- or 4-chloropyridinium radical cations can function as surrogates for the installation of 2- and 4-pyridones on (hetero-)arenes, respectively. Although we anticipated that chlorine-substituted pyridinium radical cations would display different spin density distributions when compared to the parent counterparts,³⁶ we identified novel reaction chemistry to access aryl chloropyridiniums, from which the corresponding N-aryl-2- or 4-pyridone compounds can be accessed upon hydrolysis (Scheme 1).^{37,38} The method represents the first direct N-pyridonation of arenes.

We investigated the C–H functionalization step using 4-chloropyridine N-OTf under visible-light irradiation in the

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Scheme 1. C–H Pyridonation to Access N-Aryl-2- and 4-Pyridones



presence of a photoredox catalyst. The pyridine-N-OTf reagent can be prepared *in situ* from commercially available 4-chloropyridine *N*-oxide and trifluoromethanesulfonic anhydride or synthesized on a multigram scale for isolation and storage under an inert atmosphere for at least six months. The optimum reaction condition for C–H functionalization of an arene as the limiting reagent requires 1.5 equiv of the *N*-OTf reagent and 5 mol % of Ru(bpy)₃(PF₆)₂ in acetonitrile at room temperature (Table S1). Irradiation with a 23 W household compact fluorescent lamp (CFL) led to complete substrate conversion within 24 h. Irradiation with a 70 W blue LED reduced the reaction time to 10 min but afforded the desired product in a lower yield (52%). Control experiments indicated that the photocatalyst and visible-light irradiation are crucial for the formation of the desired product. Subsequently, hydrolysis of the arylpyridinium salt to the desired *N*-aryl-4-pyridone can be accomplished (Table 1).³⁹

Table 1. Optimization of the Hydrolysis Conditions for *N*-Aryl-4-pyridones

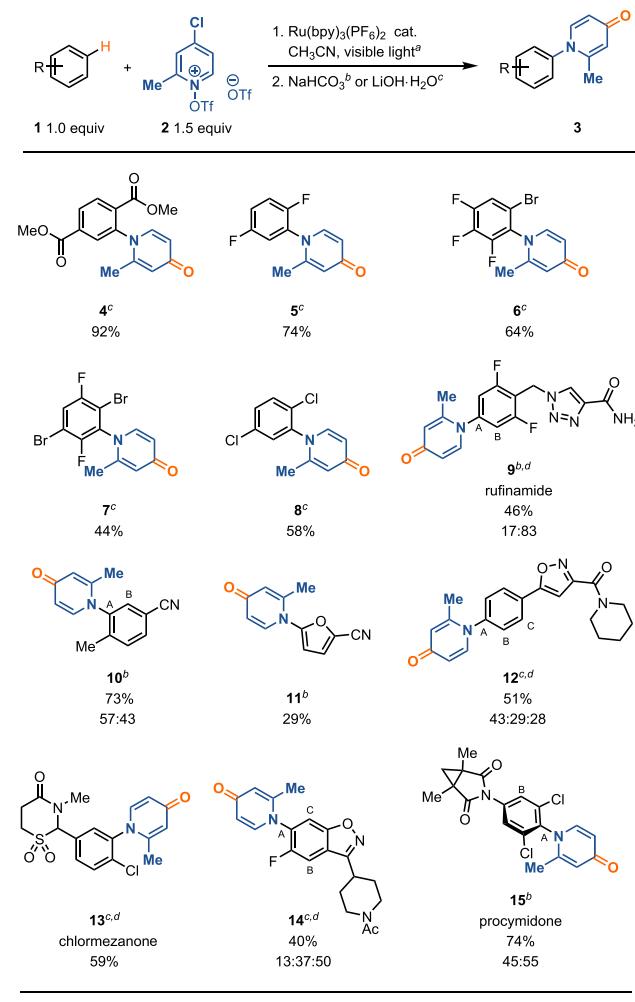
entry	reagents	equiv	solvent	T (°C)	time (h)	yield (%)
1	TMSOK	5	THF	23	4	–
2	KOH	10	MeOH	50	4	–
3	NaOH	10	CH ₃ CN:H ₂ O (1:1)	50	4	15
4	NaHCO ₃	10	CH ₃ CN:H ₂ O (1:1)	50	4	30
5	LiOH/H ₂ O ₂	5:10	THF:H ₂ O (3:1)	0–25	0.5	54
6	LiOH/ <i>t</i> -BuOOH	5:10	THF:H ₂ O (3:1)	0–25	0.5	92
7	NaHCO ₃ / <i>t</i> -BuOOH	5:10	THF:H ₂ O (3:1)	0–25	4	90

In the subsequent hydrolysis, oxygen nucleophiles such as potassium trimethylsilanolate (TMSOK), KOH, NaOH, and NaHCO₃ resulted in either decomposition (entries 1 and 2) or low product yields (entries 3 and 4). As confirmed by LC–MS analysis, methyl esters are hydrolyzed under these reaction conditions, and competitive nucleophilic attack on the 2-position of the pyridinium salt provides the free aniline.^{40,41} Inspired by the hydrolysis conditions for cleavage of chiral oxazolidone heterocycles (Evans auxiliaries), we used lithium hydroperoxide to provide the desired product in 54% yield (entry 5).^{42,43} Substituting hydrogen peroxide with *tert*-butyl

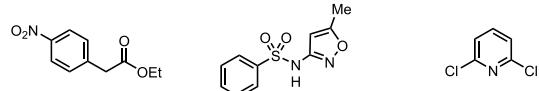
hydroperoxide gave the *N*-aryl-4-pyridone in 92% yield (entry 6).⁴⁴ Using NaHCO₃ instead of LiOH base increased the reaction time (4 h) but gave the product in comparable yield (90%) (entry 7). We reasoned that the bulky oxygen nucleophile allows for the regioselective attack on the less hindered γ -position of the pyridinium salt.

The optimized reaction conditions proved effective for introduction of the 4-pyridone motif on a range of diverse aromatics (Scheme 2, top), while the attempted hydrolysis reactions for certain substrates (Scheme 2, bottom) led to low conversion or compound decomposition. Disubstituted arenes afforded the corresponding *N*-aryl-4-pyridones 4, 5, 8, and 10 in 58–92% yields. Highly electron-deficient aromatics (such as 1,4-dimethyl terephthalate) can be efficiently functionalized,

Scheme 2. C–H Pyridonation of 2-Methyl-4-pyridone with Various (Hetero-)arenes



substrates for which hydrolysis failed after successful C–H functionalization^a:



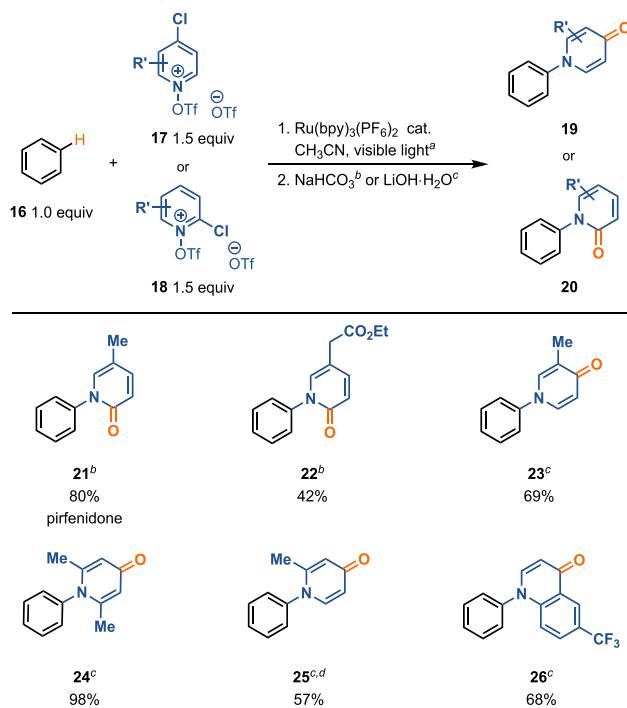
^aReaction conditions: arene (0.4 mmol), Ru(bpy)₃(PF₆)₂ (5 mol %), 23 W CFL, CH₃CN (0.2 M), 23–25 °C, 24 h. ^bHydrolysis conditions with NaHCO₃ (5–10 equiv), *t*-BuOOH (10 equiv), THF:H₂O (3:1), 0 to 25 °C. ^cHydrolysis with LiOH·H₂O (4–5 equiv) and *t*-BuOOH (10 equiv) in THF:H₂O (3:1), 0 to 25 °C. ^dTfOH (1.1 equiv). ^eSee Supporting Information (S40–S46).

whereas many other modern C–H amination reactions can only functionalize electron-rich carboarenes.^{45–59} Pyridonation of halogen-substituted arenes allowed efficient syntheses of **6** and **7**, the access to which would be challenging via traditional cross-coupling protocols. The reaction conditions also proved to be compatible with structurally complex substrates, such as rufinamide, procymidone, and chloromezanone, which are marketed drugs or agrochemicals. Late-stage functionalization of those molecules furnished the pyridone analogues **9**, **12**, **13**, **14**, and **15** in 40–74% yields, demonstrating that our methodology efficiently complements traditional pyridonation reactions.

Aryl C–H pyridonation of drug-like molecules that generates a broad range of constitutional isomers can be beneficial for drug discovery, and it allows for rapid exploration of structure–activity relationships.^{60,61} Our reaction is capable of functionalizing all available aryl C–H bonds with the following exceptions: (1) Basic functional groups were protonated in situ with a stoichiometric amount of trifluoromethanesulfonic acid to suppress reagent decomposition. Therefore, basic N-heterocycles become deactivated toward functionalization (**9** and **12**). (2) Five-membered heteroarenes often exhibit a pronounced electronic bias favoring selective C5-functionalization (**11**). (3) Aryl C–H bonds *ortho* to substituents with high steric hindrance are not functionalized (**13**).

We then evaluated several 2- and 4-chloropyridine N-OTf reagents for C–H pyridonation reactions (Scheme 3). Hydrolysis of the corresponding *N*-phenyl-chloropyridinium salts can be achieved by simply adding NaHCO₃ (10 equiv) and

Scheme 3. C–H Pyridonation of Benzene with Various 2- and 4-Pyridones

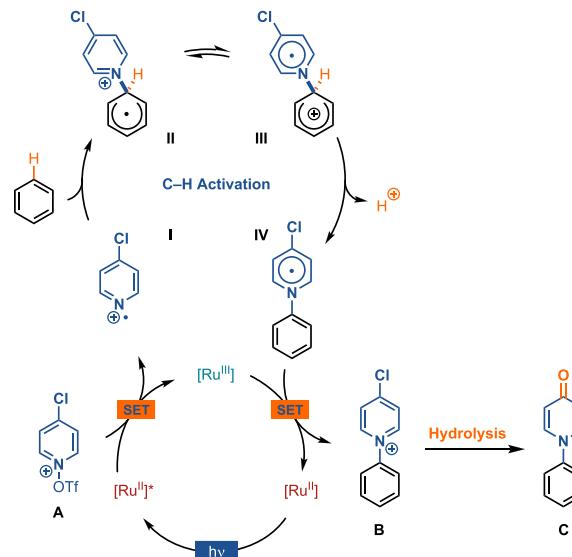


^aReaction conditions: arene (0.4 mmol), Ru(bpy)₃(PF₆)₂ (5 mol %), 23 W CFL, CH₃CN (0.2 M), 23–25 °C, 24 h. ^bHydrolysis with NaHCO₃ (10 equiv) in CH₃CN:H₂O (1:1), 50 °C. ^cHydrolysis with NaHCO₃ (5–10 equiv) or LiOH·H₂O (5 equiv) and *t*-BuOOH (10 equiv) in THF:H₂O (3:1), 0 to 25 °C. ^d1.0 mmol scale.

H₂O (1 equiv) in a one-pot sequence. 5-Methyl-2-pyridonation of benzene afforded an antifibrotic agent pirfenidone (**21**) in 80% yield. An ester functionality tolerated the basic hydrolysis conditions for the synthesis of *N*-aryl-2-pyridone **22**. Mono- and dimethyl-substituted 4-pyridonation reagents generally proved efficient (**23**, **24**, and **25**). Extension of this methodology to install quinolones directly can be achieved with electron-deficient quinolinium reagents. The C–H functionalization provides the corresponding arylquinolinium salt, which upon hydrolysis furnishes the *N*-phenyl-4-quinolone **26** in 68% yield.^{62,63} It is particularly noteworthy that our direct C–H pyridonation is a fragment coupling, in which both partners can be varied, enabling fast structural diversification.

The proposed mechanism for the C–H pyridonation is outlined in Scheme 4.^{34,35,64} Excitation of the photocatalyst

Scheme 4. Proposed Mechanism for C–H Pyridonation to Access *N*-Aryl-4-pyridones^a



Ru(bpy)₃(PF₆)₂ under visible-light irradiation produces a long-lived (1100 ns) photoexcited state, [Ru^{II}]^{*}.⁶⁵ Single-electron reduction of reagent A (onset reduction potential = 0.23 V vs saturated calomel electrode) is hypothesized to generate the pyridinium σ-radical cation (I) and [Ru^{III}]. C–N bond formation by the addition of the pyridinium radical cation (I) to an arene would afford the distonic radical II and/or III. Facile deprotonation and subsequent single-electron oxidation by the oxidized photocatalyst would then lead to the positively charged arylpyridinium salt B,⁶⁶ which upon hydrolysis can be converted into the corresponding *N*-arylpypyridone C.

In summary, we have developed the first C–H pyridonation reaction that can introduce 2- or 4-pyridones to (hetero-)arenes. Generation of chloropyridinium radical cations through photo-redox catalysis allows for the functionalization of structurally complex molecules and therefore renders our method useful for late-stage modification.

■ ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.9b02054.

Detailed experimental procedures and spectroscopic characterization (PDF)

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Notes

The authors declare no competing financial interest.

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