

Synthesis of a Conjugation-Ready, Phosphorylated, Tetrasaccharide Fragment of the O-PS of *Vibrio cholerae* O139

Sameh E. Soliman^{†,‡} and Pavol Kováč*,[†]

[†]NIDDK, LBC, Section on Carbohydrates, National Institutes of Health, Bethesda, Maryland 20892-0815, United States [‡]Department of Chemistry, Faculty of Science, Alexandria University, Alexandria 21321, Egypt

Supporting Information

ABSTRACT: A new pathway to the tetrasaccharide α -Colp-(1 \rightarrow 2)-4,6-P- β -D-Galp-(1 \rightarrow 3)-[α -Colp-(1 \rightarrow 4)]- β -D-GlcpNAc-1-(OCH₂CH₂)₃NH₂ has been developed. Glycosylation of 8-azido-3,6-dioxaoctyl 4,6-O-benzylidene-2-deoxy-2-trichloroacetamido- β -D-glucopyranoside with 3,4,6-tri-O-acetyl-2-O-bromoacetyl- α -D-galactopyranosyl bromide afforded the β -linked disaccharide. Debromoacetylation followed by reductive opening of the benzylidene acetal afforded the disaccharide diol acceptor. Halide-assisted glycosylation with 2,4-di-O-benzyl- α -colitosyl bromide gave the 1,2-cis-coupling product. Deacetylation followed by regioselective phosphorylation gave isomeric (R,S)-(P)- Φ -R-R-R-cyclic phosphates, which were globally deprotected by one-step catalytic (Pd/C) hydrogenation/hydrogenolysis. The target tetrasaccharide, obtained in high overall yield, is amenable for conjugation to proteins.

holera is a deadly, infectious enteric disease, endemic to many countries in the Third World. The disease caused by Vibrio cholerae O1 has been around for centuries, but in 1992, a new type of cholera emerged, which manifested itself by the same symptoms, i.e., watery diarrhea, but was caused by a newly discovered pathogen Vibrio cholerae O139. This prompted the elucidation of the structure^{1,2} of the O-specific polysaccharide (O-PS) of this Gram-negative bacterium, which was the first step toward the rational development of a conjugate vaccine for the disease. This laboratory has been involved in development of conjugate vaccines for infectious diseases from synthetic and bacterial carbohydrates for a number of years. To that end, among other things, we have synthesized fragments of the protective carbohydrate antigens of the disease-causing bacterial pathogens and studied their binding with the homologous protective antibodies involved. $^{3-5}$ Determination of the minimum structural requirements in the antigen which, when transformed into immunogenic conjugates and used as vaccines elicit protective antibodies, requires assembly of the O-PS fragments in the spacer-equipped form to make them amenable for conjugation. In 2006, Oscarson's and our laboratory reported independent syntheses of the linker-equipped upstream tetrasaccharide [sequence FD(E)C, Figure 1]. Here, we describe a new, high-yielding approach to the synthesis of the same carbohydrate sequence. It provides the desired structure in

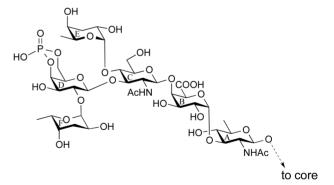


Figure 1. Structure of the O-PS of Vibrio cholerae O139.

fewer steps and in an experimentally more convenient manner than by the approaches described previously.

Each of the previous pathways to the linker-equipped sequence FD(E)C, shown in Figure 1, has its own merits. The disadvantages in the Oscarson approach⁶ are that the attachment of a linker requires chemical manipulations with the fully assembled tetrasaccharide, and the synthetic strategy requires two steps for the final deprotection. On the other hand, in the Ruttens approach,⁷ the isomers-forming phosphorylation is done in the middle of the whole reaction

Received: September 8, 2015



The Journal of Organic Chemistry

Scheme 1. Synthesis of the Disaccharide Diol Acceptor 8

Scheme 2. Synthesis of the Spacer-Equipped, Phosphorylated Tetrasaccharide 14

sequence. Consequently, the two isomeric phosphates must be resolved to eliminate complications that would arise from having to continue the sequence with a mixture. Thus, one has to either complete the sequence with only one isomer or perform all chemical transformations separately with both phosphates to increase the overall yield of the final product. The present tetrasaccharide buildup starts from the downstream⁸ end with the linker-equipped monosaccharide, and the synthesis is designed in the way allowing the phosphorylation to be done as a penultimate step. Thus, separation of isomeric phosphates is not required because the following global deprotection removes the chirality at the P atom yielding, thus, the same target product from both phosphates.

The feasible, just outlined strategy toward the tetrasaccharide fragment 14 (Scheme 2) is an extension of our preparation of the phosphorylated upstream trisaccharide sequence 9 D(E)C, Figure 1. There, and here as well, to minimize the overall number of synthetic steps, the choice of protecting groups allows final, global deprotection involving transformation of several different functional groups to be achieved in one step (catalytic hydrogenation/hydrogenolysis with Pd/C).

Accordingly (Scheme 1), 1,3,4,6-tetra-O-acetyl- α -D-galactopyranose (2)¹⁰ was bromoacetylated to give 3 (96%), which was converted into the α -glycosyl bromide 4 with 33% HBr–HOAc. Silver trifluoromethanesulfonate (triflate) mediated glycosylation of the linker-equipped glycosyl acceptor 8-azido-3,6-dioxaoctyl 4,6-O-benzylidene-2-deoxy-2-trichloroacetami-

do- β -D-glucopyranoside 11 (5) with the lpha-glycosyl donor 4 was performed using our improved protocol 11,12 (cf. ref 13) to afford the desired β -linked disaccharide 6 in 85% yield. In addition to being a key intermediate here, the foregoing substance is also a generally useful building block in oligosaccharide synthesis, as it allows orthogonal deprotection (debromoacetylation or regioselective opening of the benzylidene acetal). The new methodology using 1,1,3,3-tetramethylurea as an acid scavenger^{11,12} obviates the use of molecular sieves, whose use makes it difficult to control the mild acidity of the reaction medium required to prevent side reactions to occur. The β -configuration of the interglycosidic linkage in 6 follows from the ¹H NMR spectrum (δ 4.74, d, $J_{1,2}$ = 8.0 Hz, H-1^{II}). In addition, signals for the two anomeric carbons appeared in the 13 C NMR spectrum at almost identical chemical shifts (δ 99.4 and 99.8 for C-1^{II} and C-1^I, respectively). That the acidlabile benzylidene group was preserved under these conditions was manifested by the presence of the 13 C signal at δ 101.2 (PhCH).

Selective *O*-debromoacetylation using thiourea in the presence of a weak, non-nucleophilic organic base to prevent acyl group migration ¹⁴ gave the disaccharide acceptor 7 having O-2^{II} free, in 91% yield. The upfield shift of the signal for H-2^{II} (δ 3.79 ppm), and the COSY crosspeak between the hydroxyl group (δ 2.59 ppm) and H-2^{II} confirmed that removal of the bromoacetyl group occurred without acetyl group migration.

Subsequent regioselective reductive opening of the $4^{\rm I},6^{\rm I}$ -Obenzylidene ring in disaccharide 7 using sodium cyanoborohydride and HCl–Et₂O in tetrahydrofuran¹⁵ afforded the $6^{\rm I}$ -Obenzyl derivative 8 in 88% yield. The significant upfield shift of the signal for C-4^I (by ~10 ppm, as compared to that in the spectrum of 7), as well as the COSY crosspeak between H-4^I (δ 3.53 ppm) and the newly generated hydroxyl group (δ 4.34 ppm), confirmed that the reductive opening of the benzylidene acetal led to the $6^{\rm I}$ -benzyl ether and a free $4^{\rm I}$ -OH group.

The disaccharide diol glycosyl acceptor **8** (Scheme 2) was subjected to halide-assisted glycosylation with the α -colitosyl bromide **9** (made by treatment of the corresponding ethyl thioglycoside with bromine) to give the tetrasaccharide **10** (83%). The identification of the coupling product **10** was based on its 1 H and 13 C NMR spectra, which included signals characteristic for both the donor and the acceptor moieties. As expected, a downfield shift was observed in the spectrum of **10** for the H-4^I and H-2^{II} signals, compared to those in the spectrum of **8**, as a result of glycosylation at those positions. The signals for the two anomeric protons of colitose residues present appeared as doublets at δ 5.22 ppm (J = 3.0 Hz) and 5.07 ppm (J = 3.3 Hz) and confirmed, thus, the exclusive formation of the α -glycosidic linkages.

Subsequent de-O-acetylation of **10** with methanolic sodium methoxide afforded 8-azido-3,6-dioxaoctyl 2,4-di-O-benzyl-3,6-dioexy- α -L-xylo-hexopyranosyl- $(1\rightarrow 4)$ -[2,4-di-O-benzyl-3,6-dideoxy- α -L-xylo-hexopyranosyl- $(1\rightarrow 2)$ - β -D-galactopyranosyl- $(1\rightarrow 3)$]-6-O-benzyl-2-deoxy-2-trichloroacetamido- β -D-glucopyranoside (**11**) in virtually theoretical yield. Treatment ^{19,20} of the latter with the phosphorylating reagent 2,2,2-trichloroethyl phosphorodichloridate gave a mixture of the two isomeric 4^{II} , 6^{II} -cyclic 2,2,2-trichloroethyl phosphates **12** and **13** (\sim 3:1, ³¹P NMR) in combined 92% yield. For identification and characterization, the mixture was resolved, but the global deprotection could be done with a mixture of the isomeric phosphates.

The desired transformation of the benzyl ethers, the 2,2,2-trichloroethyl, N-trichloroacetyl, and the azide groups in 12 was accomplished in one step by hydrogenation/hydrogenolysis in the presence of Pd/C catalyst to give the title phosphorylated tetrasaccharide fragment 14 in 85% yield. Considering the lability of the α -colitosyl group to acid hydrolysis, the reaction was conducted in a pH = 7 buffer. In addition to HRMS, presence of the cyclic phosphate in 14 followed from the 31 P NMR spectrum, showing $^{3}J_{\rm P,H}$ = 21.6 Hz. 21 As expected, because the deprotection of the phosphate group canceled the chirality at that site, similar treatment of the phosphate isomer 13 gave the same product 14 (1 H NMR, 13 C NMR, 31 P NMR, TLC, HRMS) in comparable yield.

CONCLUSIONS

We have developed an improved, convenient, and high-yielding strategy for the stereoselective synthesis of the phosphorylated tetrasaccharide 14, which is one of the terminal determinants of the O-specific polysaccharide of *Vibrio cholerae* O139. The final, amino-functionalized, linker-equipped fragment 14 was obtained in a form ready for conjugation to proteins. Many features in the present, successful synthetic approach will be incorporated in the synthesis of the full O-PS, the phosphorylated hexasaccharide [sequence FD(E)CBA, Figure 1], which is in progress.

EXPERIMENTAL SECTION

General Methods. Optical rotations were measured at ambient temperature for solution in CHCl₃, unless stated otherwise. All reactions were monitored by thin-layer chromatography (TLC) on silica gel 60 coated glass slides. Column chromatography was performed by elution from prepacked columns of silica gel and monitored with the evaporative light scattering detector. Nuclear magnetic resonance (NMR) spectra were measured at 600 MHz for ¹H, 150 MHz for ¹³C, and 162 MHz for ³¹P. Solvent peaks were used as internal reference relative to TMS for ¹H and ¹³C chemical shifts (ppm); ³¹P chemical shifts (ppm) are reported relative to 85% H₃PO₄ in D₂O external reference. Assignments of NMR signals were made by homonuclear and heteronuclear two-dimensional correlation spectroscopy, run with the software supplied with the spectrometers. When reporting assignments of NMR signals, nuclei associated with the spacer are denoted with a prime; sugar residues are serially numbered, beginning with the one bearing the aglycon, and are identified by a Roman numeral superscript in listings of signal assignments, with nuclei of the colitose residues as III and IV (see Scheme 2, 10). The density of 2,2,2-trichloroethyl phosphorodichloridate (Aldrich/Sigma, $d \approx 1.7$ g/mL at 20 °C) was determined by weighing of 1 mL of the reagent. Palladium-on-charcoal catalyst (5%, Escat 103) was purchased from Engelhard Industries. 1,3,4,6-Tetra-O-acetyl-α-D-galactopyranose (2) was prepared from the commercially available 1,2,3,4,6-penta-Oacetyl- β -D-galactopyranose (1) as described, 10 except that the compound was crystallized from i-PrOH. Solutions in organic solvents were dried with anhydrous Na₂SO₄, and concentrated at 40 °C/2 kPa. 1,3,4,6-Tetra-O-acetyl-2-O-bromoacetyl- α -D-galactopyranose (3). To a solution of 1,3,4,6-tetra-O-acetyl- α -D-galactopyranose (2)¹ (5.0 g, 14.35 mmol) and 1,1,3,3-tetramethylurea (3.77 mL, 31.57 mmol) in dry CH₂Cl₂ (45 mL) was added bromoacetyl bromide (2.5 mL, 28.70 mmol) dropwise with stirring under argon at -20 °C. The cooling was removed, and with continued stirring, the mixture was allowed to warm to room temperature. After 8 h, analysis by TLC (12:1 toluene-acetone) indicated complete disappearance of the starting tetraacetate. The mixture was diluted with CH_2Cl_2 (100 mL) and washed successively with ice-water, satd aq NaHCO3, and brine.

The organic layer was dried, concentrated, and coevaporated with

toluene (twice). The residue was crystallized from isopropyl ether to

give 3 (6.45 g, 96%). Mp: 86–87 °C (*i*-Pr₂O). $[\alpha]_D^{20} = +92.4$ (*c* 1.0,

CHCl₃). ¹H NMR (600 MHz, CDCl₃): $\delta = 6.40$ (d, 1 H, $J_{1,2} = 3.2$ Hz,

H-1), 5.52 (dd, 1 H, $J_{3,4}$ = 2.9, $J_{4,5}$ = 1.2 Hz, H-4), 5.40 (dd, 1 H, $J_{2,3}$ = 10.8 Hz, $J_{3,4}$ = 3.0 Hz, H-3), 5.37 (dd, 1 H, $J_{1,2}$ = 3.3 Hz, $J_{2,3}$ = 10.8 Hz, H-2), 4.35 (ddd, 1 H, J = 0.8, J = 6.7, J = 7.6 Hz, H-5), 4.14–4.08 (m, 2 H, H-6_{a,b}), 3.80–3.75 (m, 2 H, CH₂Br), 2.18, 2.17, 2.05, 2.02 (4 s, 12 H, 4COCH₃). 13 C NMR (150 MHz, CDCl₃): δ = 170.3, 170.1, 170.0, 169.9 (4 OCOCH₃), 166.4 (COCH₂Br), 89.2 (C-1), 68.8 (C-5), 68.0 (C-2), 67.4 (C-4), 67.1 (C-3), 61.1 (C-6), 24.7 (CH₂Br), 20.9, 20.7, 20.6, 20.5 (4OCOCH₃). HRMS (ESI-TOF): m/z [M + NH₄]⁺ calcd for C₁₆H₂₅BrNO₁₁ 486.0611, found 486.0608. Anal. Calcd for C₁₆H₂₁BrO₁₁: C, 40.94; H, 4.51. Found: C, 41.05; H, 4.70.

8-Azido-3,6-dioxaoctyl 4,6-O-Benzylidene-2-deoxy-3-O-(3,4,6-tri-O-acetyl-2-O-bromoacetyl- β -D-galactopyranosyl)-2-trichloroacetamido- β -D-glucopyranoside (6). To a solution of 3 (6.0 g, 12.80 mmol) in CH₂Cl₂ (100 mL) at 0 °C (ice-water bath) was added 33% HBr-HOAc (60 mL) slowly and with stirring (during ~30 min). After 4 h at 0 °C, the solution was diluted with CH₂Cl₂ and washed subsequently with ice-water (twice) and satd aq NaHCO₃. The organic layer was dried, concentrated, and coevaporated with toluene (twice) to give 3,4,6-tri-O-acetyl-2-O-bromoacetyl-α-D-galactopyranosyl bromide (4) as a syrup in almost theoretical yield, which was used in the next step without purification. ¹H NMR (600 MHz, CDCl₃): δ = 6.36 (d, 1 H, $J_{1,2}$ = 3.9 Hz, H-1), 5.53 (br d, 1 H, J = 3.1 Hz, H-4), 5.46 (dd, 1 H, $J_{2,3} = 10.7$ Hz, $J_{3,4} = 3.3$ Hz, H-3), 5.31 (dd, 1 H, $J_{1,2} = 10.7$ Hz, $J_{3,4} = 10.7$ 3.9 Hz, $J_{2.3} = 10.6$ Hz, H-2), 4.35 (br t, 1 H, J = 6.7 Hz, H-5), 4.17 (dd, 1 H, J = 6.4, J = 11.5 Hz, H-6, 4.12 (dd, 1 H, J = 6.7, J = 11.4 Hz, H- 6_h), 3.88 (d, 1 H, J = 12.1 Hz, CHHBr), 3.84 (d, 1 H, J = 12.2 Hz, CHHBr), 2.17, 2.07, 2.02 (3 s, 9 H, 3 COCH₃). ¹³C NMR (150 MHz, $CDCl_3$): $\delta = 170.3$, 169.9, 169.7 (3 OCOCH₃), 166.6 (COCH₂Br), 90.5 (C-1), 69.5 (C-2), 69.3 (C-5), 67.2 (C-4), 66.9 (C-3), 60.9 (C-6), 24.7 (CH₂Br), 20.7, 20.6, 20.5 (3 OCOCH₃).

A solution of 4 (3.1 g, 6.32 mmol) in anhydrous CH_2Cl_2 (15 mL) was added, at -30 °C in one portion, to a mixture of glycosyl acceptor ¹¹ (5, 2.0 g, 3.51 mmol), 1,1,3,3-tetramethylurea (1.0 mL, 8.42 mmol), and powdered AgOTf (1.8 g, 7.02 mmol) in anhydrous CH_2Cl_2 (40 mL). The cooling was removed, and with continued stirring, the mixture was allowed to warm to room temperature. The stirring was continued until TLC (~8 h, 3:2 hexane—acetone) indicated that all of the acceptor was consumed. Et_3N (0.5 mL) was added, and the mixture was diluted with CH_2Cl_2 (50 mL) and filtered through a Celite pad. The filtrate was washed successively with 0.5 M aq HCl, satd aq NaHCO₃, and brine, and dried. After concentration, chromatography (2:1 hexane—acetone) gave the spacer-equipped disaccharide 6 (2.9 g, 85%).

 $[\alpha]_{\rm D}^{20} = -11.0$ (c 0.7, CHCl₃). ¹H NMR (600 MHz, CDCl₃): $\delta =$ 7.50–7.37 (m, 5 H, Ph), 7.13 (d, 1 H, $J_{2,NH}$ = 7.8 Hz, NH), 5.56 (s, 1 H, PhCH), 5.34 (d, 1 H, $J_{3.4} = 2.9$ Hz, H-4^{II}), 5.22 (dd, 1 H, $J_{1.2} = 8.1$ Hz, $J_{2,3} = 10.4$ Hz, H-2^{II}), 5.06 (d, 1 H, $J_{1,2} = 8.2$ Hz, H-1^I), 4.96 (dd, 1 H, $J_{2,3} = 10.5$ Hz, $J_{3,4} = 3.4$ Hz, H-3^{II}), 4.74 (d, 1 H, $J_{1,2} = 8.0$ Hz, H- 1^{II}), 4.47 (t, 1 H, J = 9.5 Hz, H- 3^{I}), 4.35 (dd, 1 H, J = 4.8, J = 10.6 Hz, $H-6_{b}^{I}$), 4.12 (dd, 1 H, J = 7.3, J = 11.0 Hz, $H-6_{b}^{II}$), 4.03 (dd, 1 H, J = 11.0 Hz, J = 16.4, J = 11.2 Hz, $H-6_a^{II}$), 3.93–3.89 (m, 1 H, $H-1_b'$), 3.83–3.80 (m, 2 H, H- $6_{a'}^{I}$ H- $1_{a'}^{\prime}$), 3.77 (t, 1 H, J = 6.9 Hz, H- $5_{a'}^{II}$), 3.73–3.68 (m, 3 H, CH_2Br , $H-4^{I}$), 3.67–3.58 (m, 9 H, H-2', H-3', H-4', H-5', $H-2^{I}$), 3.53-3.49 (m, 1 H, H-5^I), 3.41 (t, 2 H, J = 5.0 Hz, H-6'), 2.12, 2.01, 1.97 (3 s, 9 H, 3 COCH₃). ¹³C NMR (150 MHz, CDCl₃): $\delta = 170.3$, 170.1, 170.0 (3 OCOCH₃), 166.2 (COCH₂Br), 161.9 (NCOCCl₃), 137.0 (ipso Ph), 129.2–126.0 (Ar), 101.2 (PhCH), 99.8 (C-1¹), 99.4 (C-1^{II}), 92.5 (CCl₃), 78.8 (C-4^I), 76.3 (C-3^I), 70.8, 70.62, 70.5 (C-2', C-3', C-4'), 70.7 (C-3^{II}), 70.65 (C-5^{II}), 70.2 (C-2^{II}), 70.0 (C-5'), 68.7 (C-1'), 68.5 $(C-6^{I})$, 66.8 $(C-4^{II})$, 66.2 $(C-5^{I})$, 61.0 $(C-6^{II})$, 58.5 $(C-2^{I})$, 50.6 (C-6'), 25.3 (CH₂Br), 20.59, 20.57, 20.56 (3 OCOCH₃). HRMS (ESI-TOF): m/z [M + NH₄]⁺ calcd for $C_{35}H_{48}BrCl_3N_5O_{17}$ 994.1294, found 994.1298. Anal. Calcd for C₃₅H₄₄BrCl₃N₄O₁₇: C, 42.94; H, 4.53; N, 5.72. Found: C, 43.01; H, 4.56; N, 5.63.

8-Azido-3,6-dioxaoctyl 4,6-O-Benzylidene-2-deoxy-3-O-(3,4,6-tri-O-acetyl- β -D-galactopyranosyl)-2-trichloroacetamido- β -D-glucopyranoside (7). A solution of thiourea (1.1 g, 14.7 mmol) in methanol (50 mL) was added at 0 °C to a stirred solution of 6 (4.8 g, 4.90 mmol) and sym-collidine (0.9 mL, 7.35 mmol) in CH₂Cl₂ (100 mL). The mixture was stirred overnight at room temperature, when TLC

(6:1 dichloromethane-acetone) showed that all of the starting material was consumed and a single product was formed. The mixture was partitioned between water and CH2Cl2, the organic phase was dried and concentrated, and the residue was chromatographed (9:1 dichloromethane-acetone) to give 7 (3.8 g, 91%) as a white foam. $[\alpha]_D^{20} = -9.3$ (c 1.2, CHCl₃). ¹H NMR (600 MHz, CDCl₃): $\delta =$ 7.47–7.26 (m, 6 H, Ph, NH), 5.56 (s, 1 H, PhCH), 5.32 (dd, 1 H, J_{3,4} = 3.4, $J_{4,5}$ = 0.9 Hz, H-4^{II}), 5.08 (d, 1 H, $J_{1,2}$ = 8.3 Hz, H-1^I), 4.83 (dd, 1 H, $J_{2,3} = 10.3$ Hz, $J_{3,4} = 3.5$ Hz, H-3^{II}), 4.56 (d, 1 H, $J_{1,2} = 7.9$ Hz, H- 1^{II}), 4.48 (t, 1 H, J = 9.9 Hz, H- 3^{I}), 4.37 (dd, 1 H, J = 5.0, J = 10.6 Hz, $H-6_{b}^{I}$), 4.11 (dd, 1 H, J = 7.5, J = 11.1 Hz, $H-6_{b}^{II}$), 3.96–3.92 (m, 2 H, $H-6_{a}^{II}$, $H-1_{b}^{\prime}$), 3.84–3.80 (m, 2 H, $H-6_{a}^{I}$, $H-1_{a}^{\prime}$), 3.79 (m, 1 H, $H-2_{a}^{II}$), 3.78–3.73 (m, 2 H, H-4^I, H-5^{II}), 3.72–3.61 (m, 9 H, H-2^I, H-2^I, H-3^I, H-4', H-5'), 3.56-3.52 (m, 1 H, H-5^I), 3.41 (t, 2 H, J = 5.0 Hz, H-6'), 2.59 (br s, 1 H, 2^{II} –OH), 2.10, 2.01, 1.98 (3 s, 9 H, 3 COCH₃). 13 C NMR (150 MHz, CDCl₃): $\delta = 170.4$, 170.3, 170.1 (3 OCOCH₃), 162.3 (NCOCCl₃), 136.7 (ipso Ph), 129.3-126.0 (Ar), 102.4 (C-1^{II} 101.5 (PhCH), 100.3 (C-1¹), 92.5 (CCl₃), 79.5 (C-4¹), 76.9 (C-3¹), 72.6 $(C-3^{II})$, 71.0 $(C-5^{II})$, 70.8, 70.7, 70.5 (C-2', C-3', C-4'), 70.0 $(C-5^{II})$ 5'), 68.9 (C-1'), 68.8 (C-2^{II}), 68.6 (C-6^I), 66.9 (C-4^{II}), 66.3 (C-5^I), 61.2 (C-6^{II}), 58.2 (C-2^I), 50.6 (C-6'), 20.7–20.6 (3 OCOCH₃). HRMS (ESI-TOF): m/z [M + NH₄]⁺ calcd for $C_{33}H_{47}Cl_3N_5O_{16}$ 874.2083, found 874.2064. Anal. Calcd for C₃₃H₄₃Cl₃N₄O₁₆: C, 46.19; H, 5.05; N, 6.53. Found: C, 46.09; H, 5.14; N, 6.45.

8-Azido-3,6-dioxaoctyl 6-O-Benzyl-2-deoxy-3-O-(3,4,6-tri-O-ace $tyl-\beta-D$ -galactopyranosyl)-2-trichloroacetamido- $\beta-D$ -glucopyranoside (8). A mixture of the benzylidene acetal 7 (1.12 g, 1.31 mmol) and freshly activated, powdered molecular sieves (3 Å, 1.7 g) in dry THF (35 mL) was stirred under nitrogen for 60 min at room temperature. The solution was cooled to 0 °C, and sodium cyanoborohydride (1.0 g, 15.72 mmol) was added portionwise. After being stirred for 20 min at 0 °C, 2 M HCl-Et₂O was added dropwise at 0 °C until the effervescence ceased and the pH remained acidic. The mixture was stirred for an additional 30 min at room temperature, diluted with CH₂Cl₂ (50 mL), and filtered through Celite. The filtrate was washed with cold satd aq NaHCO3 (25 mL) and brine (25 mL), and the organic extract was dried and concentrated. Chromatography (4:1 toluene-acetone) afforded 8 (990 mg, 88%). $[\alpha]_D^{20} = +6.7$ (c 1.0, acetone). ¹H NMR (600 MHz, acetone- d_6): $\delta = 8.26$ (d, 1 H, $J_{2,NH} =$ 8.6 Hz, NH), 7.42–7.28 (m, 5 H, Ph), 5.34 (dd, 1 H, $J_{3.4} = 3.5$, $J_{4.5} =$ 1.0 Hz, H-4^{II}), 4.93 (dd, 1 H, $J_{2,3} = 10.2$ Hz, $J_{3,4} = 3.6$ Hz, H-3^{II}), 4.88 (d, 1 H, $J_{1.2} = 8.4$ Hz, H-1^I), 4.67 (d, 1 H, $J_{1.2} = 7.8$ Hz, H-1^{II}), 4.63 (bs, 1 H, PhCH₂), 4.34 (br s, 1 H, 4^I-OH), 4.33-4.29 (m, 1 H, H- 5^{II}), 4.22 (d, 1 H, J = 3.7 Hz, 2^{II} –OH), 4.18–4.12 (m, 2 H, H- 6^{II} _{a,b}), 4.07-4.04 (m, 1 H, H-3^I), 3.95-3.90 (m, 2 H, H-6^I_b, H-1'_b), 3.81 (m, 1 H, H-2^I), 3.78 (m, 1 H, H-2^{II}), 3.76-3.71 (m, 2 H, H-6^I_a, H-1[']_a), 3.70–3.60 (m, 8 H, H-2', H-3', H-4', H-5'), 3.56–3.52 (m, 2 H, H-4^I, $H-5^{I}$), 3.40 (t, 2 H, J = 5.0 Hz, H-6'), 2.13, 2.00, 1.96 (3 s, 9 H, 3 COCH₂). ¹³C NMR (150 MHz, acetone- d_6): δ = 170.8, 170.6, 170.4 (3 OCOCH₃), 162.8 (NCOCCl₃), 139.9 (ipso Ph), 129.1–128.1 (Ar), 104.7 (C-1^{II}), 101.4 (C-1^I), 93.9 (CCl₃), 85.6 (C-3^I), 76.5 (C-5^I), 73.8 (PhCH₂), 73.6 (C-3^{II}), 71.8 (C-5^{II}), 71.3, 71.1, 71.0 (C-2^I, C-3^I, C-4^I), 70.7 (C-5'), 70.5 (C-6^I), 70.3 (C-4^I), 69.5 (C-2^{II}), 69.4 (C-1'), 68.2 $(C-4^{II})$, 62.5 $(C-6^{II})$, 57.6 $(C-2^{I})$, 51.4 (C-6'), 20.7, 20.6, 20,5 (3 OCOCH₃). HRMS (ESI-TOF): m/z [M + NH₄]⁺ calcd for C₃₃H₄₉Cl₃N₅O₁₆ 876.2240, found 876.2241. Anal. Calcd for C₃₃H₄₅Cl₃N₄O₁₆: C, 46.08; H, 5.27; N, 6.51. Found: C, 46.15; H,

8-Azido-3,6-dioxaoctyl 2,4-Di-O-benzyl-3,6-dideoxy-α-L-xylo-hexopyranosyl-(1 \rightarrow 4)-[2,4-di-O-benzyl-3,6-dideoxy-α-L-xylo-hexopyranosyl-(1 \rightarrow 2)-3,4,6-tri-O-acetyl- β -D-galactopyranosyl-(1 \rightarrow 3)]-6-O-benzyl-2-deoxy-2-trichloroacetamido- β -D-glucopyranoside (10). Br₂ (360 μL, 6.96 mmol) was added to a solution of ethyl 2,4-di-O-benzyl-3,6-dideoxy-1-thio- β -L-xylo-hexopyranoside (1.13 g, 3.48 mmol) in CCl₄ (20 mL). The mixture was shaken gently, and after 5 min, hex-1-ene (1.75 mL, 13.92 mmol) was added. After concentration and coevaporation with CCl₄ (3 × 10 mL), a solution in dry CH₂Cl₂ (10 mL) of the crude α-colitosyl bromide 9 thus obtained was added to a stirred mixture of 8 (0.50 g, 0.58 mmol), Bu₄NBr (1.13 g, 3.48 mmol), and powdered molecular sieves (4 Å, 4.5

g) in 2:1 CH₂Cl₂-DMF (30 mL). After stirring under argon atmosphere for 48 h at room temperature, when TLC (6:1 toluene-acetone) indicated the glycosyl acceptor 8 was still present, more Bu₄NBr (0.75 g, 2.32 mmol) and freshly prepared bromide donor 9 (2.32 mmol/10 mL CH₂Cl₂) were added into the reaction mixture, and stirring was continued at room temperature. When TLC (6:1 toluene-acetone) showed that all glycosyl acceptor 8 had been consumed (48 h), the mixture was diluted with CH₂Cl₂ (25 mL) and filtered through a Celite pad, and the solids were washed with CH₂Cl₂ $(2 \times 10 \text{ mL})$. The combined filtrate and washings were successively washed with satd aq NaHCO₃ (25 mL) and water (25 mL), dried, and concentrated. Chromatography (6:1 chloroform-acetone) gave the title tetrasaccharide 10 (0.71 g, 83%). $[\alpha]_D^{20} = -23.6$ (c 0.6, CHCl₃). ¹H NMR (600 MHz, CDCl₃): δ = 7.34–7.22 (m, 25 H, Ph), 7.09 (d, 1 H, $J_{2,NH} = 8.0$ Hz, NH), 5.27 (d, 1 H, $J_{3,4} = 2.9$ Hz, H-4^{II}), 5.22 (d, 1 H, $J_{1,2} = 3.0$ Hz, H-1^{IV}), 5.07 (d, 1 H, $J_{1,2} = 3.3$ Hz, H-1^{III}), 4.93 (d, partial overlap, 1 H, $J_{1,2} = 7.0$ Hz, H-1¹), 4.90 (dd, partial overlap, 1 H, $\hat{J}_{2,3} = 10.1$, $\hat{J}_{3,4} = 3.5$ Hz, H-3^{II}), 4.76 (d, 1 H, $\hat{J}_{1,2} = 7.9$ Hz, H-1^{II}), 4.66-4.62 (m, 1 H, H-5^{III}), 4.57 (d, 1 H, $^2J = 12.1$ Hz, PhCHH), 4.56(d, 1 H, 2J = 12.3 Hz, PhCHH), 4.55 (d, 1 H, 2J = 12.2 Hz, PhCHH), 4.52 (d, 1 H, ^{2}J = 12.2 Hz, PhCHH), 4.46 (d, 1 H, ^{2}J = 12.1 Hz, PhCHH), 4.43 (d, 1 H, $^{2}J = 12.2$ Hz, PhCHH), 4.41–4.38 (m, 4 H, 3 x PhCHH, H-3^I), 4.28 (d, 1 H, ${}^{2}J$ = 12.4 Hz, PhCHH), 4.29–4.25 (m, 1 H, H-5^{IV}), 4.02-3.95 (m, 4 H, H-4^I, H-1'_a, H-6^{II}_{a,b}), 3.90 (dd, 1 H, $J_{1,2} = 8.1$, $J_{2,3} = 10.1$ Hz, H-2^{II}), 3.88–3.84 (m, 3 H, H-2^{III}, H-2^{IV}, H-6^I_a), 3.76 (t, 1 H, J = 6.9 Hz, H-5^{II}), 3.74–3.68 (m, 2 H, H-1'_b), H-6^I_b), 3.65-3.55 (m, 10 H, H-2^I, H-2', H-3', H-4', H-5', H-5^I), 3.51 (br s, 1 H, H-4^{IV}), 3.36 (t, 2 H, J = 5.1 Hz, H-6'), 3.33 (br s, 1 H, H-4^{III}), 2.08–2.16 (m, 2 H, H-3^{III}_{eq}, H-3^{IV}_{eq}), 2.01, 1.81, 1.68 (3 s, 9 H, 3 x COCH₃), 1.73–1.81 (m, 2 H, H-3^{III}_{ax}, H-3^{IV}_{ax}), 1.24 (d, 3 H, $J_{5,6} = 6.2$ Hz, H-6^{IV}), 1.20 (d, 3 H, $J_{5,6} = 6.2$ $CDCl_3$): $\delta = 170.4$, 169.8, 169.7 (3 OCOCH₃), 161.5 (NCOCCl₃), 138.5, 138.4, 138.3, 138.1, 138.0 (5 ipso Ph), 128.4-127.3 (Ar), 101.7 $(J_{C,H} = 161.7 \text{ Hz}, \text{ C-1}^{II}), 99.0 (J_{C,H} = 165.0 \text{ Hz}, \text{ C-1}^{I}), 97.3 (J_{C,H} = 165.0 \text{ Hz})$ 171.3 Hz, C-1^{IV}), 96.6 ($J_{C,H} = 172.1$ Hz, C-1^{III}), 92.4 (CCl₃), 75.7 (C-3¹, C-4^{IV}), 75.4 (C-4^{III}), 75.2 (C-5¹), 73.2 (PhCH₂), 73.1 (C-3^{II}), 72.8 (C-4^{II}), 72.5 (C-2^{III}), 71.4 (C-2^{III}), 71.3 (PhCH₂), 71.2 (PhCH₂), 70.9 $(C-2^{IV})$, 70.7, 70.6, 70.5 (C-2', C-3', C-4'), 70.4 $(C-5^{II})$, 70.3 (PhCH₂), 70.0 (C-5'), 68.4 (C-6^I), 67.6 (C-1'), 67.4 (C-4^{II}), 67.3 (C- \dot{S}^{IV}), 66.0 (C-5^{III}), 60.6 (C-6^{II}), 59.3 (C-2^I), 50.5 (C-6'), 26.7 (C-3^{IV}), 26.4 (C-3^{III}), 20.7, 20.6, 20.4 (3 x OCOCH₃), 16.6 (C-6^{III}), 16.3 (C- 6^{IV}). HRMS (ESI-TOF): $m/z [M + NH_4]^+$ calcd for $C_{73}H_{93}Cl_3N_5O_{22}$ 1496.5372, found 1496.5388. Anal. Calcd for C₇₃H₈₉Cl₃N₄O₂₂: C, 59.21; H, 6.06; N, 3.78. Found: C, 59.44; H, 6.17; N, 3.69.

8-Azido-3,6-dioxaoctyl 2,4-Di-O-benzyl-3,6-dideoxy-α-ι-xylo-hexopyranosyl- $(1\rightarrow 4)$ -[2,4-di-O-benzyl-3,6-dideoxy- α - ι -xylo-hexopyranosyl- $(1\rightarrow 2)$ - β -D-galactopyranosyl-(S)-(P)-4,6-cyclic 2,2,2-trichloroethyl phosphate- $(1\rightarrow 3)$]-6-O-benzyl-2-deoxy-2-trichloroacetamido-β-p-qlucopyranoside (12) and 8-Azido-3,6-dioxaoctyl 2,4-Di-Obenzyl-3,6-dideoxy- α - ι -xylo-hexopyranosyl-(1 \rightarrow 4)-[2,4-di-O-benzyl-3,6-dideoxy- α -L-xylo-hexopyranosyl- $(1\rightarrow 2)$ - β -D-galactopyranosyl-(R)-(P)-4,6-cyclic 2,2,2-trichloroethyl phosphate- $(1\rightarrow 3)$]-6-O-benzyl-2-deoxy-2-trichloroacetamido- β -D-glucopyranoside (13). A solution of sodium methoxide in methanol (1 M, ~ 1.2 mL) was added under argon to a solution of 10 (400 mg, 0.27 mmol) in 1:5 CH₂Cl₂-MeOH (30 mL), and the mixture was stirred at room temperature for 2 h. The mixture was neutralized with Amberlite IR-120 (H⁺) resin and filtered, and the solids were washed with MeOH (2×10 mL). The filtrate was concentrated and coevaporated with toluene (twice) to give compound 11 as an amorphous solid in almost theoretical yield. HRMS (ESI-TOF): m/z [M + NH₄]⁺ calcd for $C_{67}H_{87}Cl_3N_5O_{19}$ 1370.5055, found 1370.5061.

To a solution of triol 11 (330 mg, 0.24 mmol) and pyridine (200 μ L, 2.44 mmol) in CH₂Cl₂ (5 mL) was added 2,2,2-trichloroethyl phosphorodichloridate (57 μ L, 0.36 mmol) dropwise at -20 °C. When TLC (~20 min, 4:1 toluene–acetone) indicated complete conversion of 11, excess of reagent was destroyed by addition of MeOH (400 μ L). The mixture was concentrated, and EtOAc (3 mL) was added to the residue. The precipitate was filtered off and washed with EtOAc (2 × 2 mL). The combined filtrates were concentrated,

and chromatography (3:1 toluene-acetone) gave 12 (260 mg, 69%) and 13 (87 mg, 23%), in that order.

Data for 12. $[\alpha]_D^{20} = -16.5$ (c 0.5, CHCl₃). ¹H NMR (600 MHz, CDCl₃): $\delta = 7.38 - 7.23$ (m, 25 H, Ph), 7.06 (d, 1 H, $J_{2,NH} = 7.1$ Hz, NH), 5.26 (d, 1 H, $J_{1,2}$ = 3.0 Hz, H-1^{III}), 5.07 (d, 1 H, $J_{1,2}$ = 8.1 Hz, H- 11), 5.04 (d, 1 H, $J_{1,2} = 3.0$ Hz, $^{11-1}$), 5.07 (d, 1 H, $J_{1,2} = 6.1$ Hz, 11), 5.04 (d, 1 H, $J_{3,4} = 3.1$ Hz, 11), 4.76 (d, 1 H, $J_{3,4} = 3.1$ Hz, 11), 4.72 (q, 1 H, $J_{5,6} = 7.0$ Hz, 11), 4.68 –4.61 (m, 2 H, 11), 4.64 (d, 1 H, 11), 4.79 Hz, 11), 4.60 (dd, partial overlap, 1 H, 11), 11 8.2, $J_{3,4} = 2.7 \text{ Hz}$, H-3^I), 4.58–4.52 (m, 7 H, 5 x PhCHH, H-6^{II}_{a,b}), 4.51 (d, 1 H, ^{2}J = 11.6 Hz, PhCHH), 4.48 (d, 1 H, ^{2}J = 11.7 Hz, PhCHH), 4.45 (d, 1 H, ^{2}J = 12.2 Hz, PhCHH), 4.41 (d, 1 H, ^{2}J = 12.1 Hz, PhCHH), 4.36 (d, 1 H, ${}^{2}J$ = 12.2 Hz, PhCHH), 4.28 (d, 1 H, J = 3.3 Hz, 3^{II} – OH), 4.13 (q, 1 H, $J_{5,6} = 7.3$ Hz, H- 5^{III}), 3.98–3.95 (m, 2 H, H-4^I, H-2^{III}), 3.93-3.87 (m, 4 H, H-6^I_b, H-4^{IV}, H-2^{IV}, H-1'_b), 3.83 (dd, 1 H, $J_{1,2} = 8.0$, $J_{2,3} = 9.9$ Hz, H-2^{II}), 3.73–3.71 (m, 2 H, H-6^I_a, H-1'_a), 3.70–3.66 (m, 1 H, H-3^{II}), 3.63–3.57 (m, 8 H, H-2', H-3', H-4', H-5'), 3.56–3.50 (m, 2 H, H-5^I, H-2^I), 3.48 (br s, 2 H, H-4^{III}, H-5^{II}), 3.34 (t, 2 H, J = 5.1 Hz, H - 6'), 2.17 (dt, 1 H, $J_{2,3} = J_{3,4} = 3.7$, $^2J = 13.1$ Hz, 1 H, $H - 3^{II}_{eq}$), 2.06 (dt, 1 H, $J_{2,3} = J_{3,4} = 3.7$, $^2J = 12.6$ Hz, 1 H, $H - 3^{IV}_{eq}$), 1.85 (dt, 1 H, $J_{3,4} = 2.2$, $J_{2,3} = ^2J = 12.7$ Hz, 1 H, $H - 3^{IV}_{ax}$), 1.82 (dt, 1 H, $J_{3,4} = 2.3$, $J_{2,3} = ^2J = 13.0$ Hz, 1 H, $H - 3^{III}_{ax}$), 1.28 (d, 3 H, $J_{5,6} = 6.5$ Hz, $H - 6^{IV}$), 1.22 (d, 3 H, $J_{5,6} = 6.5$ Hz, $J_{5,6} =$ CDCl₃): δ = 161.3 (NCOCCl₃), 139.5, 138.3, 138.2, 138.1, 136.8 (5 ipso Ph), 128.7-127.1 (Ar), 101.9 (C-1^{II}), 98.9 (C-1^{III}), 98.1 (C-1^I), 96.7 (C-1^{IV}), 94.9 (d, $J_{C,P} = 9.9$ Hz, CH_2CCl_3), 92.6 (COCCl₃), 78.3 (d, $J_{C,P} = 6.9$ Hz, $C-4^{II}$), 77.7 (C-2^{II}), 76.9 (C-4^{IV}), 76.8 (CH₂CCl₃), 75.8 (C-3^I), 75.8 (C-5^I), 75.2 (C-4^{III}), 73.2 (PhCH₂), 72.4 (PhCH₂), 72.3 (C-4¹), 72.1 (C-2¹¹¹), 71.6 (d, $J_{C,P} = 7.1$ Hz, C-3¹¹), 71.4 (2 $PhCH_2$), 71.1 (C-2^{IV}), 70.9 (PhCH₂), 70.8 (d, $J_{C,P} = 7.7$ Hz, C-6^{II}), 70.7, 70.6, 70.5, 70.0 (C-2', C-3', C-4', C-5'), 68.6 (C-1'), 68.4 (C-5''), 67.6 (C-6'), 66.5 (C-5'V), 66.2 (d, $J_{C,P} = 6.3 \text{ Hz}, \text{C-5}^{\text{II}}$), 60.6 (C-2'), 50.6 (C-6'), 27.5 (C-3''V), 26.9 (C-3''II), 16.6 (C-6'III), 16.4 (C-6'V). ³¹P NMR (162 MHz, CDCl₃): $\delta = -10.62$. HRMS (ESI-TOF): m/z[M + Na]⁺ calcd for C₆₉H₈₃Cl₆N₄O₂₁PNa 1567.3316, found

Data for 13. $[\alpha]_D^{20} = -8.7$ (c 0.5, CHCl₃). ¹H NMR (600 MHz, CDCl₃): $\delta = 7.35 - 7.24$ (m, 25 H, Ph), 7.16 (d, 1 H, $J_{2,NH} = 7.1$ Hz, NH), 5.24 (d, 1 H, $J_{1,2} = 2.8$ Hz, H-1^{III}), 5.08 (d, 1 H, $J_{1,2} = 8.0$ Hz, H-1^I), 5.06 (d, 1 H, $J_{1,2} = 3.4$ Hz, H-1^{IV}), 4.90 (d, 1 H, $J_{3,4} = 3.2$ Hz, H-4^{II}), 4.71–4.66 (m, 2 H, H-5^{IV}, H-6^{II}_b), 4.65–4.56 (m, 2 H, CH₂CCl₃), 4.63 (d, 1 H, $J_{1,2} = 7.8$ Hz, H-1^{III}), 4.58 (d, 1 H, $J_{1,2} = 7.8$ Hz, H 4.54–4.39 (m, 9 H, 7 x PhCHH, H-3^I, H-6^{II}_a), 4.38 (d, 1 H, ${}^{2}J$ = 11.8 Hz, PhCHH), 4.36 (d, 1 H, 2J = 11.8 Hz, PhCHH), 4.27 (d, 1 H, J = 3.9 Hz, 3^{II} – OH), 4.13 (q, 1 H, $J_{5,6}$ = 7.4 Hz, H-5^{III}), 4.00 (t, 1 H, $J_{3,4}$ = $J_{4,5} = 8.8 \text{ Hz}, \text{H-4}^{\text{I}}), 3.95 - 3.92 \text{ (m, 1 H, H-2}^{\text{III}}), 3.91 - 3.86 \text{ (m, 2 H, H-2}^{\text{III}})$ 6_{b}^{I} , $H-1_{b}^{I}$), 3.85-3.83 (m, 1 H, $H-2^{IV}$), 3.79 (dd, 1 H, $J_{1,2}=7.8$, $J_{2,3}=9.9$ Hz, $H-2^{II}$), 3.75-3.72 (m, 1 H, $H-3^{II}$), 3.71-3.66 (m, 2 H, $H-6_{a}^{I}$), $H-1_{a}^{I}$), 3.62-3.57 (m, 11 H, $H-4^{IV}$, H-2', H-3', H-4', H-5', H-5 2^{I}), 3.52 (br s, 1 H, H- 5^{II}), 3.46 (br s, 1 H, H- 4^{III}), 3.34 (t, 2 H, J = 5.0Hz, H-6'), 2.16 (dt, 1 H, $I_{2,3} = I_{3,4} = 3.1$, $^{2}I_{3} = 12.2$ Hz, 1 H, H-3^{IV}_{eq}), 2.12 (dt, 1 H, $I_{2,3} = I_{3,4} = 3.6$, $^{2}I_{3} = 12.7$ Hz, 1 H, H-3^{III}_{eq}), 1.85 (dt, 1 H, $I_{3,4} = 2.2$, $I_{2,3} = ^{2}I_{3} = 12.8$ Hz, 1 H, H-3^{III}_{ax}), 1.82 (dt, 1 H, $I_{3,4} = 2.2$, $I_{2,3} = ^{2}I_{3} = 12.8$ Hz, 1 H, H-3^{III}_{ax}), 1.82 (dt, 1 H, $I_{3,4} = 2.2$, $I_{2,3} = ^{2}I_{3} = 12.8$ Hz, 1 H, H-3^{IV}_{ax}), 1.26 (dt, 3 H, $I_{5,6} = 6.3$ Hz, H-6^{IV}), 1.20 (dt, 3 H, $I_{5,6} = 6.5$ Hz, H-6^{III}). 13°C NMR (150 MHz, CDCl₃): $\delta = 161.4$ (NCOCCl₃), 138.9, 138.2, 138.1, 138.0, 137.1 (5 ipso Ph), 130.9-127.4 (Ar), 102.1 (C-1^{II}), 98.7 (C-1^{III}), 98.2 (C-1^I), 96.2 (C-1^{IV}), 94.5 (d, $J_{C,P} = 8.2 \text{ Hz}$, CH_2CCl_3), 92.6 (COCCl₃), 77.6 (d, $J_{C,P} = 2.2 \text{ Hz}$, CH₂CCl₃), 78.3 (C-4^{II}, C-2^{II}, C-3^I), 76.5 (C-4^{IV}), 75.3 (C-5^I), 75.2 (C-5^I) 4^{III}), 73.3 (PhCH₂), 72.3 (C- 4^{I}), 71.8 (d, partial overlap $J_{\text{C,P}} = 7.5 \text{ Hz}$, C-3⁽¹⁾, 71.7 (C-2⁽¹¹⁾), 71.3 (PhCH₂), 71.2 (PhCH₂), 71.1 (PhCH₂), 71.0 (PhCH₂), 70.8 (C-2^{IV}), 70.7, 70.6, 70.5, 70.0 (C-2', C-3', C-4', C-5'), 70.2 (d, $J_{C,P} = 4.4 \text{ Hz}$, C-6^{II}), 68.6 (C-1'), 68.4 (C-5^{III}), 67.9 (C-6^I), 67.1 (d, $J_{C,P} = 5.6 \text{ Hz}$, C-5^{II}), 66.4 (C-5^{IV}), 60.1 (C-2^I), 50.6 (C-6'), 27.0 (C-3^{III}, C-3^{IV}), 16.5 (C-6^{III}), 16.4 (C-6^{IV}). ³¹P NMR (162 MHz, CDCl₂): $\delta = -1.88$. HRMS (ESI-TOF): m/z [M + NH₄]⁺ calcd for C₆₉H₈₇Cl₆N₅O₂₁P 1562.3756, found 1562.3749.

8-Amino-3,6-dioxaoctyl 3,6-Dideoxy- α -L-xylo-hexopyranosyl- $(1\rightarrow 4)$ -[3,6-dideoxy- α -L-xylo-hexopyranosyl- $(1\rightarrow 2)$ - β -D-galactopyranosyl-4,6-cyclic phosphate- $(1\rightarrow 3)$]-2-deoxy-2-acetamido- β -D-glucopyranoside (14). (a) A mixture of 12 (150 mg, 0.11 mmol) and Pd/

C (150 mg) in a mixture of MeOH (6 mL) and 0.1 M potassium phosphate buffer (6 mL; pH = 7) was stirred under H₂ (1 atm) at room temperature. After 5 days, when TLC (2:1 i-PrOH-30% NH₄OH) showed complete conversion of the starting material into a more polar product, the mixture was filtered through a Celite pad, the catalyst was washed with H₂O (2 × 5 mL), and the filtrate was concentrated. Chromatography (2:1:0.1 i-PrOH-H₂O-30% NH₄OH) followed by lyophilization afforded the desired tetrasaccharide 14 (69 mg, 85%). ¹H NMR (600 MHz, D₂O): δ = 4.94 (d, 1 H, $J_{1,2}$ = 3.7 Hz, H-1^{III}), 4.83 (d, 1 H, $J_{1,2}$ = 3.5 Hz, H-1^{IV}), 4.73 (q, 1 H, $J_{5,6}$ = 6.9 Hz, H-5^{III}), 4.64 (d, 1 H, $J_{1,2} = 7.8$ Hz, H-1^{II}), 4.50 (d, 1 H, $J_{3,4} = 7.8$ Hz, H-1^{II}), 4. 3.7 Hz, H-4^{II}), 4.35 (d, partial overlap, 1 H, $J_{1,2} = 8.8$ Hz, H-1^I), 4.33-4.27 (m, 2 H, H-6^{II}_{a,b}), 4.25–4.22 (m, 1 H, H-5^{III}), 4.13 (br t, 1 H, J =3.1 Hz, H-4^{IV}), 3.84 (t, 1 H, $J_{2,3} = J_{3,4} = 9.8$ Hz, H-3^I), 3.95–3.89 (m, 4 H, H-2^{IV}, H-2^{III}, H-6¹_b, H-1'_b), 3.84–3.79 (m, 2 H, H-3^{II}, H-6¹_a), 3.77 (dd, 1 H, $J_{1,2} = 8.8$, $J_{2,3} = 10.4$ Hz, H-2^I), 3.72 (br t, 1 H, J = 3.3 Hz, $H-4^{III}$), 3.66–3.64 (m, 4 H, $H-4^{I}$, $H-1'_{b}$, H-5'), 3.62–3.58 (m, 6 H, $H-4^{III}$) 2', H-3', H-4'), 3.58-3.56 (m, 1 H, H-5^I), 3.55 (dd, partial overlap, 1 H, $J_{1,2} = 7.9$, $J_{2,3} = 9.5$ Hz, H-2^{II}), 3.44 (dt, 1 H, J = 3.1, J = 9.9 Hz, H-H, $J_{1,2} = 7.3$, $J_{2,3} = 9.5$ Hz, H-2 J_{1} , 3.44 (dt, 1 H, $J_{2} = 3.1$, $J_{2} = 9.9$ Hz, H-5¹), 3.08 (t, 2 H, $J_{2} = 5.1$ Hz, H-6 $J_{2} = 5.1$ Hz, H-6 $J_{2} = 5.1$ Hz, H-3 $J_{3,4} = 2.9$, $J_{2,3} = {}^{2}J_{2} = 12.9$ Hz, H-3 $J_{2,3} = 1.81 - 1.78$ (dt, partial overlap, 1 H, $J_{3,4} = 3.4$, $J_{2,3} = {}^{2}J_{2} = 12.9$ Hz, H-3 $J_{2} = {}^{3}J_{2} = {}^{$ (C_{-1}^{III}) , 99.4 (C-1^{III}), 97.7 (C-1^{IV}), 76.3 (d, $J_{\text{C,P}} = 4.6 \text{ Hz}$, C-4^{II}), 76.1 (C-1^{II}) 2^{II}), 75.5 (C-5^I), 75.3 (C-3^I), 72.5 (C-4^I), 72.3 (d, $J_{C,P} = 7.4 \text{ Hz}$, C-3^{II}), 69.8, 69.7, 69.5 (C-2', C-3', C-4'), 69.1 (C-1'), 68.7 (C-4^{III}), 68.6 (d, $J_{\rm C,P} = 5.0~{\rm Hz}, {\rm C}\text{-}{\rm G}^{\rm I})$, 68.3 (C-4^{IV}), 67.3 (d, $J_{\rm C,P} = 4.6~{\rm Hz}, {\rm C}\text{-}{\rm S}^{\rm II})$, 68.6 (C-5^{IV}), 66.0 (C-5^{III}), 63.5 (C-2^{III}), 63.3 (C-2^{IV}), 59.5 (C-6^I), 55.6 (C-2^I), 39.2 (C-6'), 32.7 (C-3^{III}), 32.5 (C-3^{IV}), 22.1 (COCH₃), 15.5, 15.3 (C-6^{III}, C-6^{IV}). ³¹P NMR (162 MHz, D₂O): δ = -3.74 (${}^{3}J_{P,H}$ 21.6 Hz). HRMS (ESI-TOF): m/z [M – H]⁻ calcd for C₃₂H₅₆N₂O₂₁P 835.3113, found 835.3109.

(b) Compound 13 (120 mg, 0.08 mmol) when treated with Pd/C (120 mg) and worked up, as described for 12, afforded 14 (53 mg, 81%), which was in all aspects identical with the compound described above.

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.5b02105.

¹H and ¹³C NMR spectra for all new compounds (PDF)

AUTHOR INFORMATION

Corresponding Author

*E-mail: kpn@helix.nih.gov.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

This research was supported by the Intramural Research Program of the National Institutes of Health (NIH) and the National Institute of Diabetes and Digestive and Kidney Diseases (NIDDK).

REFERENCES

- (1) Cox, A. D.; Brisson, J. R.; Varma, V.; Perry, M. B. Carbohydr. Res. **1996**, 290, 43–58.
- (2) Cox, A.; Perry, M. B. Carbohydr. Res. 1996, 290, 59-65.
- (3) Pavliak, V.; Pozsgay, V.; Kováč, P.; Karpas, A.; Chu, C.; Schneerson, R.; Robbins, J.; Glaudemans, C. P. J. *J. Biol. Chem.* **1993**, 268, 25797–25802.
- (4) Wang, J.; Zhang, J.; Miller, C. E.; Villeneuve, S.; Ogawa, Y.; Lei, P. S.; Lafaye, P.; Nato, F.; Karpas, A.; Bystricky, S.; Szu, S. C.; Robbins, J.

- B.; Kováč, P.; Fournier, J. M.; Glaudemans, C. P. J. J. Biol. Chem. 1998, 273, 2777–2783.
- (5) Liao, X.; Poirot, E.; Chang, A. H. C.; Zhang, X.; Zhang, J.; Nato, F.; Fournier, J.-M.; Kováč, P.; Glaudemans, C. P. J. *Carbohydr. Res.* **2002**, 337, 2437–2442.
- (6) Turek, D.; Sundgren, A.; Lahmann, M.; Oscarson, S. Org. Biomol. Chem. 2006, 4, 1236–1241.
- (7) Ruttens, B.; Kováč, P. Carbohydr. Res. 2006, 341, 1077-1080.
- (8) McNaught, A. D. Carbohydr. Res. 1997, 297, 1-92.
- (9) Soliman, S. E.; Kováč, P. J. Org. Chem. 2015, 80, 4851-4860.
- (10) Chittenden, G. J. F. Carbohydr. Res. 1988, 183, 140-143.
- (11) Soliman, S. E.; Kováč, P. Synthesis 2014, 46, 748-751.
- (12) Soliman, S. E.; Zoeisha, C.; Kováč, P. In Carbohydrate Chemistry: Proven Synthetic Methods; Kováč, P., Ed.; CRC/Taylor and Francis: Boca Raton, 2014.
- (13) Sherman, A. A.; Yudina, O. N.; Mironov, Y. V.; Sukhova, E. V.; Shashkov, A. S.; Menshov, V. M.; Nifantiev, N. E. *Carbohydr. Res.* **2001**, 336, 13–46.
- (14) Kováč, P.; Glaudemans, C. P. J. Carbohydr. Res. 1985, 140, 289–
- (15) Garegg, P. J.; Hultberg, H.; Wallin, S. Carbohydr. Res. 1982, 108, 97-101.
- (16) Lemieux, R. U.; Hendriks, K. B.; Stick, R. V.; James, K. J. Am. Chem. Soc. 1975, 97, 4056–4062.
- (17) Ruttens, B.; Kováč, P. Synthesis 2004, 2004, 2505-2508.
- (18) Shashkov, A. S.; Chizhov, O. S. Bioorg. Khim. 1976, 2, 437-496.
- (19) Ruttens, B.; Kováč, P. Helv. Chim. Acta 2006, 89, 320-332.
- (20) Ruttens, B.; Saksena, R.; Kováč, P. Eur. J. Org. Chem. 2007, 2007, 4366–4375.
- (21) Knirel, Y. A.; Paredes, L.; Jansson, P.-E.; Weintraub, A.; Widmalm, G.; Albert, M. J. Eur. J. Biochem. **1995**, 232, 391–396.