Organic & Biomolecular Chemistry

PAPER

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Cite this: *Org. Biomol. Chem.*, 2019, **17**, 9014

Transition-metal free C(sp²)–C(sp²) bond formation: arylation of 4-aminocoumarins using arynes as an aryl source⁺

A mild, efficient and transition-metal free synthetic strategy has been developed for the α -arylation

of 4-aminocoumarins. This synthetic strategy proceeds via C(sp²)-C(sp²) bond formation between

4-aminocoumarins and aryne precursors in a single step by simple treatment with a fluoride source in the

absence of a metal-catalyst. Moreover, this methodology affords good yields of 4-amino-3-arylcoumarin

derivatives bearing halide functionality.

Received 2nd September 2019, Accepted 25th September 2019 DOI: 10.1039/c9ob01919g

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Introduction

C-C bond formation reactions between two sp² hybridized carbon centers are one of the most fundamental reactions in organic synthesis which provide the foundation for the construction of molecular frameworks and have been in the forefront of research in organic synthesis.¹ There are several established synthetic strategies where different functionalized carbon centers have been used for the construction of such C-C bonds.² In this regard, transition-metal catalysts play a predominant role in the construction of C-C bonds.³ A variety of metal catalysts such as Mn, Cu, Ni, Pd, Ag, Ru, Rh, Ir etc. have been extensively used for this task. Among the numerous synthetic methods for the construction of C-C bonds, transition-metal free reactions have gained predominance over others and are widely accepted in industrial practice due to their environmental friendliness. Additionally most of the transition metals are toxic in nature, and the removal of trace amounts of such transition-metal residues from the desired products is quite necessary and challenging in the pharmaceutical industries.⁴ To overcome such issues, metal-free strategies have been developed for C-C bond formation via direct C-H/C-H cross coupling. In this regard, Itami and co-workers have reported a base-promoted metal-free coupling reaction of aryl halides with arenes.⁵ Later, Shirakawa and Hayashi have reported a transitionmetal free coupling reaction between aryl halides and aryl

bond.⁶ Furthermore, Wang and co-workers have reported a metal-free direct C-H arylation of guinones and naphthoquinones using diaryliodonium salts as an aryl source.⁷ Later, Li et al. have developed a photocatalyst-free oxidative C(sp²)-H thiocyanation of heteroarenes and arenes.⁸ Similarly, Xie and his group have reported C-3 alkoxycarbonylation and C-3 alkylation of quinoxalin-2(1H)-ones.^{9a,b} Recently, Xie and co-workers have reported visible lightinduced organic dve-catalyzed C-2 sulfonylation of guinoline N-oxides.^{9c} In addition to these, cross-dehydrogenative coupling (CDC) is another method which leads to the formation of C-C bonds directly between two unmodified C-H bonds.¹⁰ This CDC strategy is particularly applicable to an sp³ hybridized carbon center which is adjacent to the carbonyl group or heteroatom or at the allylic and benzylic positions. Therefore, the development of new synthetic strategies for direct C-C bond formation between two sp² hybridized carbon centers is highly desirable and necessary in organic synthesis. On the other hand, arynes are versatile transient intermedi-

Grignard reagents for the construction of a $C(sp^2)-C(sp^2)$

tions the other hand, arynes are versatile transferit intermediates and have emerged as potent electrophiles for the atom economical synthesis of various building blocks and natural products.¹¹ In this context, Kobayashi's protocol using *o*-silyl aryl triflate¹² as a versatile aryne precursor is widely accepted in the synthetic community for the development of new-aryne based reactions which include muticomponent,¹³ cycloaddition,¹⁴ and insertion reactions.¹⁵ Interestingly, various nucleophiles including N-heterocycles, urea and imines could be coupled with arynes to form their corresponding insertion products.¹⁶ They also serve as key building blocks for the construction of transition-metal-free C–C bonds *via* coupling with a suitable nucleophile. However, metal-free *C*-arylation reactions using arynes as an aryl source are rare. In this regard,





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Chartrand and Ramtohul have reported a method for direct *C*-arylation of β -enamino esters and ketones employing arynes.¹⁷ Also, Hu and co-workers have developed a nucleophilic fluoroalkylation of arvnes using a nucleophilic fluoroalkylating agent.¹⁸ Recently, Mhaske and co-workers have reported a metal-free *C*-arylation of malonamide esters employing aryne precursors.¹⁹ Furthermore, Rodriguez et al. have developed a general method for C-arylation of β -ketoamides.²⁰ Moreover, arynes have also been used for direct C-arylation reactions involving $C(sp^2)-C(sp^2)$ bond formation. In this regard Greaney et al. have reported a transition-metal-free direct arylation of anilines via C-H arylation instead of N-H arylation due to steric hindrance around the nitrogen atom (Scheme 1a).²¹ Later, they also developed a metal-free Truce-Smiles rearrangement for the synthesis of bis(hetero)aryls via coupling of two sp²-carbon centers.²² Another interesting chemistry was reported by Larock's group, where regioselective N-arylated



Scheme 1 Scope of N- vs. C-arylation

Table 1 Optimization studies^a

products were obtained when amines and sulphonamides react with arynes (Scheme 1b).²³ These interesting reactions of N-arylation as well as C-arylation of aniline encouraged us to develop a new arylation strategy for 4-aminocoumarins. In our investigation, it is interesting to note that when 4-aminocoumarins were treated with an aryne precursor in the presence of a fluoride source, a C-arylated product was formed instead of an N-arylated product without the protection of amine functionality.

In continuation of our research on the development of new synthetic strategies involving aryne chemistry,²⁴ we present here the α -arylation of 4-aminocoumarins under transitionmetal-free conditions (Scheme 1c). 4-Amino-3-arylcoumarin was previously synthesized via acylation of 2-hydroxybenzonitrile with a-ketoacid chlorides followed by treatment with TiCl₃ and Zn dust at refluxed temperature.²⁵ Besides the use of transition metals, this synthetic approach involves a multistep sequence. Therefore, a general and versatile synthetic method for efficient synthesis of 4-amino-3-arylcoumarin derivatives under mild conditions is highly desirable. To the best of our knowledge, there is no such synthetic methodology involving direct arylation of 4-aminocoumarins with arynes.

Results and discussion

We initiated our optimization studies by treating benzyne derived in situ from 2-(trimethylsilyl)phenyl trifluoromethanesulfonate 1a (0.25 mmol) and CsF (0.5 mmol) with 4-aminocoumarin 2a (0.25 mmol) in CH₃CN at room temperature for 6 h. Under these reaction conditions, the desired product 3aa was obtained in 45% yield (Table 1, entry 1). To improve the

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Entry	F ⁻ source (equiv.)	Additive (equiv.)	Solvent	Time (h)	Temp. (°C)	Yield ^{b} (%)
1	CsF (2.0)	_	CH ₃ CN	6	rt	45
2	KF (2.0)	_	CH ₃ CN	6	rt	50
3	TBAF (2.0)		CH_3CN	6	rt	Trace
4	KF (2.0)	18-Crown-6 (1.0)	CH_3CN	6	rt	65
5	KF (2.0)	18-Crown-6 (1.0)	THF	6	rt	70
6	KF (2.0)	18-Crown-6 (1.0)	DCM	6	rt	ND
7	KF (2.0)	18-Crown-6 (1.0)	Dioxane	6	rt	ND
8	KF (2.0)	18-Crown-6 (1.0)	DMF	6	rt	ND
9	KF (2.0)	18-Crown-6 (1.0)	DMSO	6	rt	ND
10	KF (2.0)	18-Crown-6 (1.0)	$CH_3CN/THF(1:1)$	6	rt	52
11	KF (3.0)	18-Crown-6 (1.0)	THF	6	rt	72
12	KF (4.0)	18-Crown-6 (1.0)	THF	6	rt	70
13	KF (3.0)	18-Crown-6 (1.5)	THF	6	rt	76
14	KF (3.0)	18-Crown-6 (1.5)	THF	3	60	67
15	_ ` ´		THF	6	rt	ND

^a Conditions: o-Silyl aryl triflate 1a (0.25 mmol), 4-aminocoumarin 2a (0.25 mmol), fluoride source (2 to 4 equiv.), additive (1 to 2 equiv.), solvent (3 mL) stirred at rt for 6 h.^b Isolated yield. ND: not detected.

yield of the reaction product we used KF instead of CsF as the fluoride source. Under these reaction conditions, the desired product **3aa** was isolated in 50% yield (Table 1, entry 2).

When TBAF was used only traces of 3aa were observed (Table 1, entry 3). The use of KF in the presence of 18-crown-6 in CH₃CN produced 3aa in 65% yield (Table 1, entry 4). During our optimization studies, we also investigated the solvent effect by screening some common solvents, such as THF, DCM, dioxane, DMF, and DMSO and a solvent mixture of CH₃CN and THF. To our delight, a significantly improved yield of 70% was observed when the reaction was performed in THF (Table 1, entry 5). The use of a mixture of CH₃CN and THF gave 3aa in 52% yield (Table 1, entry 10). The desired product 3aa was not observed when DCM, dioxane, DMF and DMSO were used as solvents (Table 1, entries 6-9). After optimizing our fluoride source and reaction media, a few more experiments were performed by changing the amount of KF and 18-crown-6 (Table 1, entries 11 to 13). Interestingly, an improved yield of 76% was observed when the reaction was performed using 3 equiv. of KF as a fluoride source and 1.5 equiv. of 18-crown-6 as an additive in THF for 6 h (Table 1, entry 13), which were considered to be the optimized reaction conditions. Moreover, increasing the reaction temperature and decreasing the reaction time lowered the yield of product 3aa (Table 1, entry 14). As a control experiment, when the reaction was performed in the absence of fluoride, product 3aa was not observed (Table 1, entry 15). The coupled product 3aa was characterized by ¹H NMR, ¹³C NMR and HRMS analyses and compared with the reported data (see the ESI[†]).

With the optimized reaction conditions in hand, we next generalized our transition-metal free C–C bond formation strategy for the synthesis of our targeted 4-amino-3-arylcoumarin derivatives (Table 2).

The cascade transition-metal free reactions of arynes with a wide range of 4-aminocoumarins for the one-pot synthesis of 4-amino-3-arylcoumarin derivatives are presented in Table 2. As shown in Table 2, o-silyl aryl triflate 1a was treated with a variety of 4-aminocoumarins bearing methyl, fluoro, chloro, bromo and methoxy substituents leading to our desired 3-aryl-4-aminocoumarins being obtained in good yields (Table 2, 3aa-3ah). It is noteworthy that both electron-rich and electrondeficient 4-aminocoumarins efficiently participated in the onepot metal-free coupling process to provide the corresponding products (3ab-3ah) in good yields. This result confirms that the electronic effects of the substituents did not play any significant role in our reaction. Importantly, the chloro- and bromo-substituents tolerated the reaction conditions and can be used for further functionalization (Table 2, 3ad and 3ae). Moreover, 6,7 disubstituted 4-aminocoumarin 2ag also provided the desired product 3ag in good yield. Other symmetrical silvltriflates such as 4,5-dimethoxy-2-(trimethylsilyl)phenyl trifluoromethanesulfonate 1b, 3-(trimethylsilyl)-2-naphthyl trifluoromethanesulfonate 1c and 5-(trimethylsilyl)benzo[d][1,3] dioxol-6-yl trifluoromethanesulphonate 1d were also explored as aryne precursors for our one-pot synthetic strategy. In this regard, symmetrical silvl triflates 1b and 1c were treated with a

Table 2 One-pot synthesis of 4-amino-3-arylcoumarins



^{*a*} Conditions: *o*-Silyl aryl triflate **1** (0.25 mmol), 4-aminocoumarin **2** (0.25 mmol), KF (0.75 mmol), 18-crown-6 (0.37 mmol), THF (3 mL) stirred at rt for 6 h.

variety of 4-aminocoumarins to obtain 4-amino-3-arylcoumarins in 72–66% yields (Table 2, **3ba–3ch**). Moreover, symmetrical aryne derived from **1d** was also treated with 4-aminocoumarin to furnish the desired product **3da** in good yield. In addition, an aryne precursor bearing an electron-withdrawing group such as 4,5-difluoro-2-(trimethylsilyl)phenyl trifluoromethanesulfonate **1e** was also used as a suitable aryne precursor, leading to our desired product **3ea** in 67% yield.

Moreover, secondary amine **2i** was also used as a substrate for our coupling process. In this investigation, *o*-silyl aryl triflate **1a** is allowed to react with **2i** under our optimized reaction conditions to obtain the desired product 3-phenyl-4-(phenylamino)-2*H*-chromen-2-one in 62% yield (Scheme 2).

Similarly, secondary amine **2j** was also examined for our arylation process where *N*-protected iso-butyl aminocoumarin **2j** gives our desired product 4-(isobutylamino)-3-phenyl-2*H*-chromen-2-one in only 19% yield, which could be due to the steric hindrance of the iso-butyl group (Scheme 3).



Scheme 2 Synthesis of 3-phenyl-4-(phenylamino)-2*H*-chromen-2-one from aryne.



Scheme 3 Synthesis of 4-(isobutylamino)-3-phenyl-2*H*-chromen-2-one from aryne.

However, the yield of the product obtained from secondary aminocoumarins 2i and 2j is lower than that from primary aminocoumarins. This suggests that 4-aminocoumarin 2a is a better substrate than secondary aminocoumarins 2i and 2j.

In order to explore the substrate scope, unsymmetrical benzyne precursors *i.e.* 4-methyl-2-(trimethylsilyl)phenyl trifluoromethanesulfonate **1f** and 4-methoxy-2-(trimethylsilyl) phenyl trifluoromethanesulfonate **1g** were treated with 4-aminocoumarin **2a** under our optimized reaction conditions (Table 3).

Benzyne precursors **1f** and **1g** gave 4-amino-3-arylcoumarin derivatives **3fa/3f'a** and **3ga/3g'a** as a mixture of regioisomers in good yields (Table 3, entries 1 and 2). The regioisomers were not separated into individual isomers and they are represented as mixtures. Interestingly, when unsymmetrical arynes derived from 2-methyl-6-(trimethylsilyl)phenyl trifluoromethanesulfonate **1h**, 3-methoxy-2-(trimethylsilyl)phenyl trifluoromethanesulfonate **1i** and 1-(trimethylsilyl)-2-naphthyl trifluoromethanesulfonate **1j** were treated with **2a**, single regioisomers **3ha**, **3ia** and **3ja** were obtained in 67%, 69% and 67% yields, respectively, with excellent regioselectivities.

To gain insight into the reaction mechanism, a few deuterium-labelling experiments were performed. As shown in Scheme 4, when 4-aminocoumarin 2a was treated with *o*-silyl aryl triflate 1a under the optimized conditions, no deuterium incorporated product 3aa was observed. Another independent experiment was performed using these two substrates in THF (dry) in the presence of D₂O as the fourth component under the same reaction conditions. Interestingly, this experiment gave the deuterium incorporated product 3aa-D in 72% yield with quantitative incorporation of deuterium at the *ortho*-position, which was characterized by ¹H and ¹³C NMR and HRMS analyses (Scheme 4).

Based on our experiments, a plausible reaction mechanism for the synthesis of 4-amino-3-arylcoumarin is proposed in

 Table 3
 Synthesis of 4-amino-3-arylcoumarins from unsymmetrical benzyne precursors^a





 a Conditions: o-Silyl aryl triflate 1 (0.25 mmol), 4-aminocoumarin 2 (0.25 mmol), KF (0.75 mmol), 18-crown-6 (0.37 mmol), THF (3 mL) stirred at rt for 6 h.



Scheme 4 Deuterium labelling experiments.

Scheme 5. As shown in Scheme 5, the carbon atom of the enamine moiety of **2a** undergoes direct nucleophilic addition to arynes generated *in situ* from aryne precursor **1a** to form a zwitterionic intermediate **A**. Subsequently, intermediate **A** undergoes deprotonation at the activated methylene position followed by protonation at the aryl ring to form our desired product **3aa**.

Scheme 5 Probable reaction mechanism for the synthesis of 4-amino-3-arylcoumarin.



Scheme 6 Gram scale experiment for 3aa.

Furthermore, to explore the practical utility of our synthetic strategy, the synthetic strategy was performed on a gram scale (Scheme 6). In our experiment, **1a** (6 mmol) was treated with **2a** (6 mmol) under our optimized conditions to give our desired product **3aa** in 72% yield.

Conclusion

In summary, we have developed a mild, efficient and transition-metal-free $C(sp^2)-C(sp^2)$ bond forming strategy for the direct synthesis of 4-amino-3-arylcoumarins using aryne as an aryl source. A series of 4-amino-3-arylcoumarins have been synthesized using this one-pot synthetic strategy in good yields with excellent functional group compatibility including halides. Remarkably, this synthetic strategy gave a *C*-arylated product instead of an *N*-arylated product without protecting amine functionality.

Experimental section

General

All reactions involving oxygen- or moisture-sensitive compounds were carried out under an argon atmosphere using oven-dried or flame-dried glassware. All other solvents and reagents were purified according to the standard procedures or were used as received from TCI, Aldrich, Merck and Spectrochem. The reactions were monitored by thin-layer chromatography (TLC) using aluminium-backed silica gel plates (0.2 mm thickness); the chromatograms were visualized with ultraviolet light (254 nm). Flash column chromatography was performed with silica gel 60 (100–200 or 200–400 mesh). HRMS data were recorded by electrospray ionization with a Q-TOF mass analyzer.

Starting materials

General procedure for the synthesis of 4-aminocoumarin. A mixture of 4-hydroxycoumarin (1 equiv.) and ammonium

acetate (20 equiv.) was stirred at 130 °C overnight. After completion of the reaction, water (50 ml) was added. The mixture was stirred for 20 min, filtered and concentrated *in vacuo*. The residue was successively washed with water and ethanol to give 4-aminocoumarin **2a**. Compounds **2b–h** were prepared using the same method.

4-Amino-2*H***-chromen-2-one** (2a).²⁶ Applying the general experimental procedure using 4-hydroxycoumarin (3 mmol, 1 equiv.) and ammonium acetate (60 mmol, 20 equiv.), 4-amino-2*H*-chromen-2-one 2a was obtained as a light brown solid (0.430 g, 89% yield); mp 233–235 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 7.98 (dd, J_1 = 1.0 Hz, J_2 = 8.3 Hz, 1H), 7.56–7.59 (m, 1H), 7.40 (br s, 2H), 7.28–7.31 (m, 2H), 5.22 (s, 1H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.6, 155.5, 153.5, 132.0, 123.1, 122.8, 116.7, 114.3, 83.7; IR (CHCl₃): 3376, 3212, 2917, 2850, 1638, 1549, 1219, 772 cm⁻¹; HRMS (+ESI) calcd for C₉H₈NO₂ [M + H]⁺: 162.0555; found: 162.0564.

4-Amino-6-methyl-2*H***-chromen-2-one (2b).²⁶** Applying the general experimental procedure using 4-hydroxy-6-methyl-coumarin (2 mmol, 1 equiv.) and ammonium acetate (40 mmol, 20 equiv.), 4-amino-6-methyl-2*H*-chromen-2-one **2b** was obtained as a light brown solid (0.287 g, 82% yield); mp 266–268 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 7.80 (d, J = 0.9 Hz, 1H), 7.39 (dd, $J_1 = 1.8$ Hz, $J_2 = 8.4$ Hz, 1H), 7.32 (br s, 2H), 7.17 (d, J = 8.4 Hz, 1H), 5.19 (s, 1H), 2.35 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.8, 155.5, 151.7, 132.8, 132.4, 122.7, 116.5, 114.0, 83.8, 20.4; IR (CHCl₃): 3376, 3207, 2922, 2845, 1621, 1570, 1218, 772 cm⁻¹; HRMS (+ESI) calcd for C₁₀H₁₀NO₂ [M + H]⁺: 176.0712; found: 176.0710.

4-Amino-6-fluoro-2*H***-chromen-2-one** (2c).²⁶ Applying the general experimental procedure using 6-fluoro-4-hydroxycoumarin (1 mmol, 1 equiv.) and ammonium acetate (20 mmol, 20 equiv.), 4-amino-6-fluoro-2*H*-chromen-2-one **2c** was obtained as a light brown solid (0.145 g, 80% yield); mp 298–300 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 7.89 (dd, J_1 = 2.9 Hz, J_2 = 9.7 Hz, 1H), 7.44–7.49 (m, 1H), 7.41 (br s, 2H), 7.34–7.37 (m, 1H), 5.24 (s, 1H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.4, 157.6 (d, J = 239.3 Hz), 154.8, 149.9, 119.4 (d, J = 24.2 Hz), 118.7 (d, J = 8.6 Hz), 115.3 (d, J = 8.7 Hz), 108.8 (d, J = 25.3 Hz), 84.2; **IR** (CHCl₃): 3368, 3203, 2922, 2835, 1635, 1551, 1220, 772 cm⁻¹; HRMS (+ESI) calcd for C₉H₇FNO₂ [M + H]⁺: 180.0461; found: 180.0461.

4-Amino-6-chloro-2*H***-chromen-2-one (2d).²⁶** Applying the general experimental procedure using 6-chloro-4-hydroxycoumarin (1 mmol, 1 equiv.) and ammonium acetate (20 mmol, 20 equiv.), 4-amino-6-chloro-2*H*-chromen-2-one **2d** was obtained as a light brown solid (0.158 g, 81% yield); mp 330–332 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.13 (d, J = 2.4 Hz, 1H), 7.61–7.63 (m, 1H), 7.46 (br s, 2H), 7.33 (d, J = 8.8 Hz, 1H), 5.23 (s, 1H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.1, 154.6, 152.3, 131.8, 127.4, 122.5, 118.8, 115.8, 84.2; IR (CHCl₃): 3367, 3198, 2917, 2845, 1626, 1570, 1219, 772 cm⁻¹; HRMS (+ESI) calcd for C₉H₇ClNO₂ [M + H]⁺: 196.0165; found: 196.0166.

4-Amino-6-bromo-2*H***-chromen-2-one** (2e).²⁶ Applying the general experimental procedure using 6-bromo-4-hydroxycou-

marin (1 mmol, 1 equiv.) and ammonium acetate (20 mmol, 20 equiv.), 4-amino-6-bromo-2*H*-chromen-2-one **2e** was obtained as a light brown solid (0.197 g, 82% yield); mp $305-307 \degree$ C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.25 (s, 1H), 7.72 (d, *J* = 8.3 Hz, 1H), 7.45 (br s, 2H), 7.26 (d, *J* = 8.8 Hz, 1H), 5.23 (s, 1H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.1, 154.5, 152.7, 134.6, 125.5, 119.1, 116.2, 115.2, 84.2; IR (CHCl₃): 3352, 3207, 2922, 2845, 1618, 1525, 1219, 772 cm⁻¹; HRMS (+ESI) calcd for C₉H₇BrNO₂ [M + H]⁺: 239.9660; found: 239.9669.

4-Amino-7-methoxy-2*H***-chromen-2-one (2f).²⁶** Applying the general experimental procedure using 7-methoxy-4-hydroxy-coumarin (2 mmol, 1 equiv.) and ammonium acetate (40 mmol, 20 equiv.), 4-amino-7-methoxy-2*H*-chromen-2-one **2f** was obtained as a light brown solid (0.322 g, 84% yield); mp 299–301 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 7.89 (d, J = 8.9 Hz, 1H), 7.29 (br s, 2H), 6.89 (dd, $J_1 = 2.5$ Hz, $J_2 = 8.9$ Hz, 1H), 6.84 (d, J = 2.5 Hz, 1H), 5.07 (s, 1H), 3.82 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 162.3, 161.9, 155.9, 155.5, 124.1, 111.1, 107.5, 100.7, 81.7, 55.7; IR (CHCl₃): 3376, 3222, 2922, 2850, 1615, 1554, 1219, 772 cm⁻¹; HRMS (+ESI) calcd for C₁₀H₁₀NO₃ [M + H]⁺: 192.0661; found: 192.0675.

4-Amino-6,7-dimethyl-2*H***-chromen-2-one (2g).²⁶** Applying the general experimental procedure using 4-hydroxy-6,7-dimethylcoumarin (1 mmol, 1 equiv.) and ammonium acetate (20 mmol, 20 equiv.), 4-amino-6,7-dimethyl-2*H*-chromen-2-one **2g** was obtained as a light brown solid (0.162 g, 86% yield); mp 299–301 °C; ¹H NMR (**500** MHz, DMSO-d₆): δ 7.73 (s, 1H), 7.27 (br s, 2H), 7.07 (s, 1H), 5.12 (s, 1H), 2.27 (s, 3H), 2.24 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 162.0, 155.7, 152.0, 141.6, 131.5, 122.9, 117.1, 111.8, 83.2, 19.5, 18.9; IR (CHCl₃): 3350, 3224, 2923, 2850, 1623, 1558, 1219, 772 cm⁻¹; HRMS (+ESI) calcd for C₁₁H₁₂NO₂ [M + H]⁺: 190.0868; found: 190.0878.

4-Amino-7-methyl-2H-chromen-2-one (2h).²⁶ Applying the general experimental procedure using 4-hydroxy-7-methylcoumarin (2 mmol, 1 equiv.) and ammonium acetate (40 mmol, 20 equiv.), 4-amino-7-methyl-2H-chromen-2-one 2h was obtained as a light brown solid (0.290 g, 83% yield); mp 262–264 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 7.85 (d, J = 7.9 Hz, 1H), 7.33 (br s, 2H), 7.09–7.11 (m, 2H), 5.16 (s, 1H), 2.36 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.8, 155.7, 153.7, 142.7, 124.2, 122.7, 116.7, 111.9, 83.1, 20.9; IR (CHCl₃): 3367, 3217, 2927, 2850, 1618, 1539, 1219, 772 cm⁻¹; HRMS (+ESI) calcd for C₁₀H₁₀NO₂ [M + H]⁺: 176.0712; found: 176.0710.

General procedure for the synthesis of 4-(phenylamino)-2*H*chromen-2-one (2i). A mixture of 4-hydroxycoumarin (1 equiv.) and aniline (3 equiv.) was stirred at 160 °C for 20 min. After completion of the reaction, the reaction mixture was dissolved in methanol (20 ml) and then aqueous NaOH solution was added dropwise. The mixture was stirred for 20 min. The precipitate so formed was washed with water and then with ethanol to obtain pure 2i.²⁵

4-(Phenylamino)-2*H***-chromen-2-one** (2i).²⁷ Applying the general experimental procedure using 4-hydroxycoumarin (1 mmol, 1 equiv.) and aniline (3 mmol, 3 equiv.), 4-(phenyl-

amino)-2*H*-chromen-2-one **2i** was obtained as a white solid (0.197 g, 83% yield); mp 258–260 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 9.33 (s, 1H), 8.24 (dd, J_1 = 1.2 Hz, J_2 = 8.1 Hz, 1H), 7.63–7.64 (m, 1H), 7.27–7.50 (m, 7H), 5.31 (s, 1H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.5, 153.3, 152.4, 138.1, 132.3, 129.5, 125.9, 125.0, 123.6, 122.7, 117.0, 114.4, 84.3; IR (CHCl₃): 3391, 3280, 2922, 2845, 1656, 1611, 1591, 1536, 1260, 1219, 772 cm⁻¹; HRMS (+ESI) calcd for C₁₅H₁₂NO₂ [M + H]⁺: 238.0868; found: 238.0880.

General procedure for the synthesis of 4-(isobutylamino)-2*H*chromen-2-one (2j).²⁶ A mixture of 4-hydroxycoumarin (1 equiv.) and *tert*-butylamine (10 equiv.) was refluxed in acetic acid overnight. After completion of the reaction, water was added until it precipitates. The mixture was stored in a fridge overnight and then filtered and concentrated *in vacuo*. The residue was successively washed with water and ethyl acetate to give 4-(isobutylamino)-2*H*-chromen-2-one (2j)²⁶

4-(Isobutylamino)-2*H***-chromen-2-one (2j).²⁶** Applying the general experimental procedure using 4-hydroxycoumarin (1 mmol, 1 equiv.) and *tert*-butylamine (10 mmol, 10 equiv.), 4-(isobutylamino)-2*H*-chromen-2-one **2j** was obtained as a light brown solid (0.175 g, 80% yield); mp 144–146 °C; ¹H NMR (**500 MHz, DMSO-d_6**): δ 8.08 (dd, $J_1 = 1.1$ Hz, $J_2 = 8.0$ Hz, 1H), 7.71 (t, J = 5.6 Hz, 1H), 7.57 (m, 1H), 7.28–7.32 (m, 2H), 5.13 (s, 1H), 3.04 (m, 2H), 1.94–2.02 (m, 1H), 0.92 (d, J = 6.6 Hz, 6H);¹³C NMR (126 MHz, DMSO-d_6): δ 161.6, 153.2, 153.0, 131.9, 123.3, 122.4, 116.9, 114.4, 81.2, 49.7, 26.5, 20.2; **IR** (CHCl₃): 3328, 3095, 2959, 2871, 1617, 1558, 1249, 767 cm⁻¹; HRMS (+ESI) calcd for C₁₃H₁₆NO₂ [M + H]⁺: 218.1181; found: 218.1190.

General procedure for the synthesis of 4-amino-3-arylcoumarin derivatives from aryne precursors. An oven-dried round bottomed flask (50 mL) equipped with a magnetic stir bar was evacuated and purged with argon. *o*-Silyl aryl triflate (0.25 mmol, 1 equiv.), 4-aminocoumarin (0.25 mmol, 1.0 equiv.), KF (0.75 mmol, 3 equiv.), 18-crown-6 (0.37 mmol, 1.5 equiv.) and THF (3 mL) were successively added at room temperature. The reaction mixture was stirred at rt for 6 h. Water (10 mL) was added to the reaction mixture and the organic layer was extracted with EtOAc (20×2 mL). The combined organic phases were washed with brine and dried over sodium sulfate. The solvent was removed, and the residue was purified by column chromatography on silica gel using hexanes/ethyl acetate as an eluent.

4-Amino-3-phenyl2*H*-chromen-2-one (3aa).²⁵ Applying the general experimental procedure using 2-(trimethylsilyl)phenyl trifluoromethanesulfonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-aminocoumarin (0.04 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-3-phenyl-2*H*-chromen-2-one **3aa** was obtained as a white solid (0.045 g, 76% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 180–182 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.28–7.58 (m, 9H), 5.00 (br s, 2H); ¹³C NMR (126 MHz, CDCl₃): δ 161.9, 153.1, 149.7, 132.7, 131.9, 130.6, 129.2, 128.1, 123.6, 121.4, 117.5, 114.2, 101.0; ¹H NMR (500 MHz, DMSO-

d₆): δ 8.15 (d, J = 7.8 Hz, 1H), 7.31–7.61 (m, 8H), 6.77 (br s, 2H); ¹³C NMR (126 MHz, DMSO-d₆): δ 160.8, 152.5, 150.8, 133.7, 131.9, 131.0, 128.6, 127.1, 123.5, 123.4, 116.5, 114.6, 97.4; **IR** (CHCl₃): 3450, 3340, 2920, 2850, 1673, 1628, 1606, 1548, 1501, 1428, 1286, 1207, 752 cm⁻¹; HRMS (+ESI) calcd for C₁₅H₁₂NO₂ [M + H]⁺: 238.0868; found: 238.0870.

4-Amino-6-methyl-3-phenyl-2H-chromen-2-one (3ab). Applying the general experimental procedure using 2-(trimethylsilyl) phenyl trifluoromethanesulfonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-amino-6-methylcoumarin (0.044 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-6methyl-3-phenyl-2H-chromen-2-one 3ab was obtained as a white solid (0.043 g, 68% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 277-279 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 7.99 (s, 1H), 7.41-7.46 (m, 3H), 7.30-7.35 (m, 3H), 7.22 (d, J = 8.4 Hz, 1H), 6.78 (br s, 2H), 2.38 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 160.9, 150.7, 150.6, 133.8, 132.6, 132.5, 130.9, 128.6, 127.0, 123.3, 116.2, 114.1, 97.4, 20.4; IR (CHCl₃): 3468, 3348, 2917, 2855, 1663, 1633, 1610, 1556, 1508, 1429, 1280, 1218, 771 cm⁻¹; HRMS (+ESI) calcd for $C_{16}H_{14}NO_2$ [M + H]⁺: 252.1025; found: 252.1024.

4-Amino-6-fluoro-3-phenyl-2H-chromen-2-one (3ac). Applying the general experimental procedure using 2-(trimethylsilyl) phenyl trifluoromethanesulfonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-amino-6-fluorocoumarin (0.045 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-6-fluoro-3-phenyl-2H-chromen-2-one 3ac was obtained as a white solid (0.041 g, 64% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 282–284 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.08 (dd, J_1 = 2.9 Hz, J_2 = 10.0 Hz, 1H), 7.31–7.51 (m, 7H), 6.85 (brs, 2H); ¹³C **NMR (126 MHz, DMSO-d₆):** δ 160.6, 157.7 (d, J = 239.0 Hz), 150.1 (d, J = 1.9 Hz), 148.9, 133.4, 130.8, 128.6, 127.2, 119.1 (d, J = 23.3 Hz), 118.5 (d, J = 8.7 Hz), 115.6 (d, J = 8.9 Hz), 109.4 (d, J = 25.7 Hz), 97.9; IR (CHCl₃): 3468, 3337, 2915, 2847, 1669, 1633, 1610, 1556, 1504, 1435, 1410, 1212, 771 cm⁻¹; HRMS (+ESI) calcd for $C_{15}H_{11}NO_2F$ [M + H]⁺: 256.0774; found: 256.0770.

4-Amino-6-chloro-3-phenyl-2*H*-chromen-2-one (3ad). Applying the general experimental procedure using 2-(trimethylsilyl) phenyl trifluoromethanesulfonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-amino-6-chlorocoumarin (0.049 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-6chloro-3-phenyl-2*H*-chromen-2-one **3ad** was obtained as a white solid (0.047 g, 69% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 292–294 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.33 (d, *J* = 2.4 Hz, 1H), 7.65 (dd, J_1 = 2.4 Hz, J_2 = 8.8 Hz, 1H), 7.30–7.47 (m, 6H), 6.96 (br s, 2H); ¹³C NMR (126 MHz, DMSO-d₆): δ 160.5, 151.2, 149.9, 133.4, 131.6, 130.9, 128.6, 127.7, 127.3, 123.2, 118.6, 116.1, 97.9; IR (CHCl₃): 3444, 3323, 2927, 2845, 1674, 1626, 1605, 1552, 1498, 1428, 1399, 1219, 772 cm⁻¹; HRMS (+ESI) calcd for $C_{15}H_{11}NO_2Cl [M + H]^+$: 272.0478; found: 272.0474.

4-Amino-6-bromo-3-phenyl-2H-chromen-2-one (3ae). Applying the general experimental procedure using 2-(trimethylsilyl) phenyl trifluoromethanesulfonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-amino-6-bromocoumarin (0.060 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-6bromo-3-phenyl-2H-chromen-2-one 3ae was obtained as a white solid (0.051 g, 64% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 270-272 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.45 (d, J = 2.3 Hz, 1H), 7.76 (dd, J₁ = 2.2 Hz, J₂ = 8.8 Hz, 1H), 7.44-7.47 (m, 2H), 7.30–7.47 (m, 4H), 6.82 (br s, 2H); ¹³C NMR (**126 MHz**, **DMSO-d**₆): δ 160.5, 151.6, 149.8, 134.4, 133.4, 130.9, 128.7, 127.3, 126.1, 118.8, 116.5, 115.4, 98.0; IR (CHCl₃): 3454, 3332, 2922, 2850, 1672, 1628, 1601, 1545, 1496, 1428, 1398, 1219, 773 cm⁻¹; HRMS (+ESI) calcd for $C_{15}H_{11}NO_2Br [M + H]^+$: 315.9973; found: 315.9974.

4-Amino-7-methoxy-3-phenyl-2H-chromen-2-one (3af). Applying the general experimental procedure using 2-(trimethylsilyl) phenyl trifluoromethanesulfonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-amino-7-methoxycoumarin (0.048 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-7methoxy-3-phenyl-2H-chromen-2-one 3af was obtained as a white solid (0.048 g, 72% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 210-212 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.07 (d, J = 8.8 Hz, 1H), 7.30-7.45 (m, 5H), 6.89-6.93 (m, 2H), 6.87 (brs, 2H), 3.85 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 162.1, 161.0, 154.2, 151.0, 133.9, 131.0, 128.5, 126.9, 124.8, 111.2, 107.7, 100.4, 95.4, 55.7; IR (CHCl₃): 3454, 3342, 2922, 2850, 1638, 1602, 1546, 1513, 1431, 1272, 1219, 772 cm⁻¹; HRMS (+ESI) calcd for $C_{16}H_{14}NO_3$ [M + H]⁺: 268.0974; found: 268.0974.

4-Amino-6,7-dimethyl-3-phenyl-2H-chromen-2-one (3ag). Applying the general experimental procedure using 2-(trimethylsilyl)phenyl trifluoromethanesulfonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-amino-6,7-dimethylcoumarin (0.047 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-6,7-dimethyl-3-phenyl-2H-chromen-2-one 3ag was obtained as a white solid (0.046 g, 69% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 220-222 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 7.94 (s, 1H), 7.29-7.44 (m, 5H), 7.13 (s, 1H), 6.54 (br s, 2H), 2.31 (s, 3H), 2.28 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.0, 150.8 (d, J = 1.7 Hz), 141.2, 133.9, 131.7, 130.9, 128.5, 126.9, 123.5, 116.7, 111.9, 96.8, 19.4, 18.4; IR (CHCl₃): 3458, 3342, 2922, 2845, 1664, 1613, 1546, 1431, 1216, 772 cm⁻¹; HRMS (+ESI) calcd for C₁₇H₁₆NO₂[M + H]⁺: 266.1181; found: 266.1178.

4-Amino-7-methyl-3-phenyl-2*H***-chromen-2-one (3ah).** Applying the general experimental procedure using 2-(trimethylsilyl) phenyl trifluoromethanesulfonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-amino-7-methylcoumarin (0.044 g; 0.25 mmol,

1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-7methyl-3-phenyl-2*H*-chromen-2-one **3ah** was obtained as a white solid (0.044 g, 70% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 223–225 °C; ¹H NMR (500 MHz, DMSO-d_6): δ 8.04 (d, *J* = 8.2 Hz, 1H), 7.45 (t, *J* = 7.6 Hz, 2H), 7.33–7.35 (m, 3H), 7.13–7.15 (m, 2H), 6.88 (brs, 2H), 2.40 (s, 3H); ¹³C NMR (126 MHz, DMSO-d_6): δ 161.0, 152.6, 150.9, 142.4, 133.9, 131.0, 128.6, 127.0, 124.5, 123.4, 116.5, 112.1, 96.8, 20.9; IR (CHCl₃): 3414, 3344, 2922, 2850, 1636, 1615, 1598, 1544, 1514, 1435, 1218, 770 cm⁻¹; HRMS (+ESI) calcd for C₁₆H₁₄NO₂ [M + H]⁺: 252.1025; found: 252.1024.

4-Amino-3-(3,4-dimethoxyphenyl)-2H-chromen-2-one (3ba). Applying the general experimental procedure using 4,5dimethoxy-2-(trimethylsilyl)phenyl trifluoromethanesulfonate (0.07 mL; 0.25 mmol, 1 equiv.), 4-aminocoumarin (0.040 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-3-(3,4-dimethoxyphenyl)-2H-chromen-2-one 3ba was obtained as a white solid (0.054 g, 72% yield) after purification by flash chromatography on silica gel (60% EtOAc/hexane); mp 256-258 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.14 (d, J = 7.7 Hz, 1H), 7.60 (t, J = 7.2 Hz, 1H), 7.30–7.32 (m, 2H), 7.02 (d, J = 8.2 Hz, 1H), 6.87 (d, J = 1.6 Hz, 1H), 6.84 (dd, $J_1 = 1.8$ Hz, $J_2 = 8.2$ Hz, 1H), 6.74 (br s, 2H), 3.80 (s, 3H), 3.75 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 160.8, 152.4, 150.8, 148.7, 147.8, 131.6, 125.9, 123.4, 123.3, 123.2, 116.4, 114.6, 114.4, 111.9, 97.4, 55.4, 55.3; IR (CHCl₃): 3423, 3342, 2921, 2849, 1674, 1623, 1605, 1546, 1511, 1241, 1023, 751 cm⁻¹; HRMS (+ESI) calcd for C₁₇H₁₆NO₄ [M + H]⁺: 298.1079; found: 298.1080.

4-Amino-3-(3,4-dimethoxyphenyl)-6-methyl-2H-chromen-2-one (3bb). Applying the general experimental procedure using 4,5dimethoxy-2-(trimethylsilyl)phenyl trifluoromethanesulfonate (0.07 mL; 0.25 mmol, 1 equiv.), 4-amino-6-methylcoumarin (0.044 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-3-(3,4-dimethoxyphenyl)-6-methyl-2Hchromen-2-one 3bb was obtained as a white solid (0.055 g, 70% yield) after purification by flash chromatography on silica gel (60% EtOAc/hexane); mp 248-250 °C; ¹H NMR (500 MHz, **DMSO-d**₆): δ 7.97 (s, 1H), 7.40 (dd, *J*₁ = 1.5 Hz, *J*₂ = 8.4 Hz, 1H), 7.21 (d, J = 8.4 Hz, 1H), 7.01 (d, J = 8.2 Hz, 1H), 6.81–6.85 (m, 2H), 6.76 (br s, 2H), 3.79 (s, 3H), 3.74 (s, 3H), 2.37 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.2, 150.9, 150.7, 148.8, 147.9, 132.6, 132.5, 126.1, 123.4, 123.3, 116.3, 114.5, 114.4, 112.0, 97.5, 55.5, 55.4, 20.6; IR (CHCl₃): 3421, 3340, 2922, 2855, 1667, 1638, 1603, 1549, 1510, 1219, 1025, 992, 771 cm⁻¹; HRMS (+ESI) calcd for $C_{18}H_{18}NO_4$ [M + H]⁺: 312.1236; found: 312.1239.

4-Amino-3-(3,4-dimethoxyphenyl)-7-methoxy-2H-chromen-2-one (**3bf**). Applying the general experimental procedure using 4,5-dimethoxy-2-(trimethylsilyl)phenyl trifluoromethanesulfonate (0.07 mL; 0.25 mmol, 1 equiv.), 4-amino-7-methoxycoumarin (0.048 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in

THF (3 mL), 4-amino-3-(3,4-dimethoxyphenyl)-7-methoxy-2Hchromen-2-one 3bf was obtained as a white solid (0.057 g, 69% yield) after purification by flash chromatography on silica gel (60% EtOAc/hexane); mp 250-252 °C; ¹H NMR (500 MHz, **DMSO-d₆**): δ 8.04 (d, J = 8.7 Hz, 1H), 7.01 (d, J = 8.3 Hz, 1H), 6.89-6.92 (m, 2H), 6.80-6.85 (m, 2H), 6.78 (br s, 2H), 3.84 (s, 3H), 3.78 (s, 3H), 3.74 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 162.0, 161.2, 154.3, 151.3, 148.6, 147.7, 126.1, 124.7, 123.3, 114.5, 111.9, 111.2, 107.8, 100.4, 95.4, 55.7, 55.4, 55.3; IR (CHCl₃): 3409, 3338, 2923, 2850, 1633, 1597, 1539, 1510, 1219, 1025, 994, 770 cm^{-1} ; **HRMS** (+ESI) calcd for $C_{18}H_{18}NO_5[M + H]^+$: 328.1185; found: 328.1182.

4-Amino-3-(3,4-dimethoxyphenyl)-7-methyl-2H-chromen-2-one (3bh). Applying the general experimental procedure using 4,5dimethoxy-2-(trimethylsilyl)phenyl trifluoromethanesulfonate (0.07 mL; 0.25 mmol, 1 equiv.), 4-amino-7-methylcoumarin (0.044 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-3-(3,4-dimethoxyphenyl)-7-methyl-2Hchromen-2-one 3bh was obtained as a white solid (0.052 g, 67% yield) after purification by flash chromatography on silica gel (60% EtOAc/hexane); mp 247–249 °C; ¹H NMR (500 MHz, **DMSO-d**₆): δ 8.01 (d, J = 8.1 Hz, 1H), 7.00–7.15 (m, 3H), 6.81-6.85 (m, 2H), 6.74 (br s, 2H), 3.78 (s, 3H), 3.74 (s, 3H), 2.40 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.0, 152.5, 151.0, 148.6, 147.8, 142.2, 126.0, 124.4, 123.3, 123.2, 116.4, 114.4, 112.1, 111.9, 96.7, 55.4, 55.3, 20.9; IR (CHCl₃): 3450, 3338, 2923, 2850, 1636, 1616, 1541, 1512, 1220, 1025, 992, 772 cm⁻¹; HRMS (+ESI) calcd for $C_{18}H_{18}NO_4[M + H]^+$: 312.1236; found: 312.1240.

4-Amino-3-(naphthalen-2-yl)-2H-chromen-2-one (3ca). Applying the general experimental procedure using 3-(trimethylsilyl)-2naphthyl trifluoromethanesulfonate (0.07 mL; 0.25 mmol, 1 equiv.), 4-aminocoumarin (0.040 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-3-(naphthalen-2-yl)-2H-chromen-2-one 3ca was obtained as a white solid (0.049 g, 68% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 252-254 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.19 (d, J = 7.7 Hz, 1H), 7.94–7.96 (m, 3H), 7.87 (s, 1H), 7.63 (t, J = 7.7 Hz, 1H), 7.52-7.54 (m, 2H), 7.44 (dd, J₁ = 1.3 Hz, J₂ = 8.4 Hz, 1H), 7.35–7.38 (m, 2H), 6.92 (br s, 2H); 13 C NMR (126 MHz, DMSO-d₆): δ 160.9, 152.6, 151.1, 133.3, 133.2, 131.8, 131.4, 129.8, 129.2, 127.9, 127.8, 127.4, 125.9, 125.7, 123.6, 123.4, 116.6, 114.6, 97.3; IR (CHCl₃): 3391, 3227, 2917, 2845, 1633, 1597, 1538, 1493, 1425, 1218, 1110, 768 cm⁻¹; HRMS (+ESI) calcd for C₁₉H₁₄NO₂[M + H]⁺: 288.1025; found: 288.1023.

4-Amino-7-methoxy-3-(naphthalen-2-yl)-2*H*-chromen-2-one (3cf). Applying the general experimental procedure using 3-(trimethylsilyl)-2-naphthyl trifluoromethanesulfonate (0.07 mL; 0.25 mmol, 1 equiv.), 4-amino-7-methoxycoumarin (0.048 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-7-methoxy-3-(naphthalen-2-yl)-2*H*-chromen-2-one **3cf** was obtained as a white solid (0.051 g, 64% yield) after purification by flash chromatography on silica gel (40% EtOAc/ hexane); mp 281–283 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.11 (d, *J* = 9.6 Hz, 1H), 7.92–7.96 (m, 3H), 7.86 (s, 1H), 7.50–7.54 (m, 2H), 7.44 (dd, *J*₁ = 1.6 Hz, *J*₂ = 8.4 Hz, 1H), 6.94–6.96 (m, 2H), 6.92 (br s, 2H), 3.87 (s, 3H); ¹³C NMR (126 MHz, DMSOd₆): δ 162.2, 161.2, 154.4, 151.5, 133.3, 132.1, 131.6, 129.9, 129.5, 127.9, 127.8, 127.4, 125.9, 125.8, 124.8, 111.3, 107.8, 100.5, 95.2, 55.8; IR (CHCl₃): 3318, 3178, 2922, 2855, 1672, 1635, 1593, 1535, 1509, 1423, 1219, 772 cm⁻¹; HRMS (+ESI) calcd for C₂₀H₁₆NO₃[M + H]⁺: 318.1130; found: 318.1131.

4-Amino-7-methyl-3-(naphthalen-2-yl)-2H-chromen-2-one (3ch). Applying the general experimental procedure using 3-(trimethylsilyl)-2-naphthyl trifluoromethanesulfonate (0.07 mL; 0.25 mmol, 1 equiv.), 4-amino-7-methylcoumarin (0.044 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-7-methyl-3-(naphthalen-2-yl)-2*H*-chromen-2-one 3ch was obtained as a white solid (0.050 g, 66% yield) after purification by flash chromatography on silica gel (40% EtOAc/ hexane); mp 229–231 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.08 (d, J = 8.2 Hz, 1H), 7.93-7.97 (m, 3H), 7.87 (s, 1H), 7.50-7.54(m, 2H), 7.45 (dd, $J_1 = 1.7$ Hz, $J_2 = 8.4$ Hz, 1H), 7.16–7.19 (m, 2H), 6.92 (br s, 2H), 2.43 (s, 3H); ¹³C NMR (126 MHz, DMSO- \mathbf{d}_6): δ 161.1, 152.7, 151.3, 142.4, 133.3, 132.2, 131.6, 129.9, 129.4, 127.9, 127.8, 127.4, 125.9, 125.7, 124.5, 123.4, 116.5, 112.2, 96.6, 20.9; IR (CHCl₃): 3454, 3346, 2922, 2845, 1633, 1612, 1541, 1515, 1488, 1436, 1218, 772 cm⁻¹; HRMS (+ESI) calcd for $C_{20}H_{16}NO_2[M + H]^+$: 302.1181; found: 302.1178.

4-Amino-3-(benzo[d][1,3]dioxol-5-yl)-2H-chromen-2-one (3da). Applying the general experimental procedure using 5-(trimethylsilyl)benzo[d][1,3]dioxol-6-yl trifluoromethanesulphonate (0.05 mL; 0.25 mmol, 1 equiv.), 4-aminocoumarin (0.040 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-3-(benzo[d][1,3]dioxol-5-yl)-2H-chromen-2-one 3da was obtained as a white solid (0.051 g, 72% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 252–254 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.10 (dd, J_1 = 1.2 Hz, J_2 = 8.3 Hz, 1H), 7.57–760 (m, 1H), 7.29–7.32 (m, 2H), 6.97 (d, J = 7.9 Hz, 1H), 6.82 (d, J = 1.5 Hz, 1H), 6.46 (dd, J₁ = 1.6 Hz, J₂ = 7.9 Hz, 1H), 6.73 (br s, 2H), 6.02 (s, 2H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.1, 152.6, 151.3, 147.5, 146.4, 131.9, 127.2, 124.4, 123.6, 123.5, 116.6, 114.7, 111.4, 108.7, 100.9, 97.4; IR (CHCl₃): 3422, 3340, 2923, 2850, 1634, 1601, 1550, 1502, 1442, 1220, 772 cm⁻¹; HRMS (+ESI) calcd for C₁₆H₁₂NO₄[M + H]⁺: 282.0766; found: 282.0765.

4-Amino-3-(3,4-difluorophenyl)-2*H***-chromen-2-one (3ea).** Applying the general experimental procedure using 4,5difluoro-2-(trimethylsilyl)phenyl trifluoromethanesulfonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-aminocoumarin (0.040 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-3-(3,4-difluorophenyl)-2*H*-chromen-2-one **3ea** was obtained as a white solid (0.046 g, 67% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 287–289 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.16 (dd, J_1 = 1.4 Hz, $J_2 = 8.4$ Hz, 1H), 7.60–7.64 (m, 2H), 7.48 (dt, $J_1 = 8.6$ Hz, $J_2 = 11.0$ Hz, 1H), 7.32–7.39 (m, 2H), 7.14–7.15 (m, 1H), 6.98 (br s, 2H); ¹³C NMR (126 MHz, DMSO-d_6): δ 160.6, 151.9 (d, J = 142.4 Hz), 150.5 (d, J = 12.7 Hz), 149.6 (d, J = 12.5 Hz), 148.5 (d, J = 12.7 Hz), 147.7 (d, J = 12.7 Hz), 132.1, 131.4 (dd, $J_1 = 3.7$ Hz, $J_2 = 6.5$ Hz), 128.3 (dd, $J_1 = 2.9$ Hz, $J_2 = 6.3$ Hz), 123.5 (d, J = 11.8 Hz), 120.3 (d, J = 16.4 Hz), 117.5 (d, J = 17.0 Hz), 116.6, 114.5, 95.6; IR (CHCl₃): 3405, 3357, 2922, 2850, 1633, 1607, 1549, 1515, 751 cm⁻¹; HRMS (+ESI) calcd for C₁₅H₁₀NO₂F₂[M + H]⁺: 274.0680; found: 274.0680.

3-Phenyl-4-(phenylamino)-2H-chromen-2-one (3ai). Applying the general experimental procedure using 2-(trimethylsilyl) phenyl trifluoromethanesulphonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-(phenylamino)-2H-chromen-2-one (0.059 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 3-phenyl-4-(phenylamino)-2H-chromen-2-one 3ai was obtained as a white solid (0.049 g, 62% yield) after purification by flash chromatography on silica gel (30% EtOAc/hexane); mp 206-208 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.71 (s, 1H), 8.03 $(dd, J_1 = 1.1 Hz, J_2 = 8.1 Hz, 1H), 7.63 (t, J = 7.4 Hz, 1H), 7.43$ (d, J = 7.5 Hz, 1H), 7.34 (t, J = 7.6 Hz, 1H), 6.93-7.16 (m, 7H), 6.72–6.77 (m, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.4, 152.3, 146.8, 140.7, 134.1, 131.7, 130.5, 127.9, 127.2, 126.6, 124.3, 123.6, 122.1, 120.9, 116.9, 116.6, 106.8; IR (CHCl₃): 3322, 2922, 2850, 1673, 1594, 1560, 1524, 1495, 1439, 1380, 1170, 761 cm⁻¹; HRMS (+ESI) calcd for $C_{21}H_{16}NO_2[M + H]^+$: 314.1181; found: 314.1180.

4-(Isobutylamino)-3-phenyl-2H-chromen-2-one (3aj). Applying the general experimental procedure using 2-(trimethylsilyl) phenyl trifluoromethanesulphonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-(isobutylamino)-2H-chromen-2-one (0.054 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-(isobutylamino)-3-phenyl-2H-chromen-2-one 3aj was obtained as a yellow solid (0.014 g, 19% yield) after purification by flash chromatography on silica gel (20% EtOAc/hexane); mp 196–198 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.25 (d, J = 7.3 Hz, 1H), 8.14(d, J = 7.6 Hz, 1H), 7.90 (d, J = 8.3 Hz, 1H), 7.36–7.66 (m, 7H), 4.63 (d, J = 7.6 Hz, 2H), 2.19–2.27 (m, 1H), 0.94 (d, J = 6.6 Hz, 6H); ¹³C NMR (126 MHz, DMSO-d₆): δ 157.7, 152.8, 140.0, 139.8, 130.7, 124.9, 124.6, 123.4, 123.3, 122.8, 120.3, 117.9, 113.4, 111.8, 101.1, 51.4, 29.4, 19.5; IR (CHCl₃): 3399, 2958, 2924, 1648, 1590, 1524, 1460, 1219, 1025, 772 cm⁻¹; HRMS (+ESI) calcd for $C_{19}H_{20}NO_2[M + H]^+$: 294.1494; found: 294.1503.

4-Amino-3-(*p***-tolyl)-2***H***-chromen-2-one & 4-amino-3-(***m***-tolyl)-2***H***-chromen-2-one (3fa) & (3f'a). Applying the general experimental procedure using 4-methyl-2-(trimethylsilyl)phenyl trifluoromethanesulphonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-aminocoumarin (0.040 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-3-(***p***-tolyl)-2***H***-chromen-2one & 4-amino-3-(***m***-tolyl)-2***H***-chromen-2-one 3fa** & **3f'a** were obtained as a white solid (0.044 g, 70% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 202–204 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.15 (dd, J_1 = 3.2 Hz, J_2 = 7.8 Hz, 1H), 7.58–7.62 (m, 1H), 7.10–7.35 (m, 6H), 6.91 (br s, 2H), 2.34 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 160.9, 160.8, 152.6, 152.5, 150.8, 150.7, 137.6, 136.2, 133.6, 131.8, 131.7, 131.5, 130.8, 130.7, 129.2, 128.5, 127.9, 127.8, 123.5, 123.4, 116.6, 116.5, 114.7, 114.6, 97.6, 97.4, 21.0, 20.9; IR (CHCl₃): 3458, 3342, 2922, 2850, 1672, 1627, 1605, 1548, 1500, 1423, 1286, 1215, 970, 753 cm⁻¹; HRMS (+ESI) calcd for C₁₆H₁₄NO₂[M + H]⁺: 252.1025; found: 252.1022.

4-Amino-3-(4-methoxyphenyl)-2H-chromen-2-one & 4-amino-3-(3-methoxyphenyl)-2H-chromen-2-one (3ga) & (3g'a). Applying the general experimental procedure using 4-methoxy-2-(trimethylsilyl)phenyl trifluoromethanesulphonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-aminocoumarin (0.040 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-3-(4methoxyphenyl)-2H-chromen-2-one & 4-amino-3-(3-methoxyphenyl)-2H-chromen-2-one 3ga & 3g'a were obtained as a white solid (0.044 g, 65% yield) after purification by flash chromatography on silica gel (60% EtOAc/hexane); mp 208-210 °C; ¹H **NMR (500 MHz, DMSO-d₆):** δ 8.15 (t, J = 8.1 Hz, 1H), 7.56–7.62 (m, 1H), 7.23-7.38 (m, 3H), 6.87-7.01 (m, 3H), 6.85 (br s, 2H), 3.79 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.0, 160.7, 159.4, 158.3, 152.5, 152.4, 150.9, 150.8, 135.0, 132.0, 131.9, 131.7, 129.6, 125.6, 123.6, 123.5, 123.4, 123.3, 123.1, 116.5, 116.4, 116.3, 114.7, 114.6, 114.0, 112.8, 97.4, 97.2, 55.0, 54.3; IR (CHCl₃): 3454, 3342, 2922, 2855, 1674, 1626, 1603, 1547, 1281, 1219, 956, 773 cm⁻¹; HRMS (+ESI) calcd for $C_{16}H_{14}NO_3[M + H]^+$: 268.0974; found: 268.0972.

4-Amino-3-(o-tolyl)-2H-chromen-2-one (3ha). Applying the general experimental procedure using 2-methyl-6-(trimethylsilyl)phenyl trifluoromethanesulphonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-aminocoumarin (0.040 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-3-(o-tolyl)-2Hchromen-2-one 3ha was obtained as a white solid (0.042 g, 67% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 204-206 °C; ¹H NMR (500 MHz, **DMSO-d₆**): δ 8.15 (d, J = 8.4 Hz, 1H), 7.58–7.62 (m, 1H), 7.31-7.35 (m, 3H), 7.10-7.16 (m, 3H), 6.87 (br s, 2H), 2.34 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 160.8, 152.5, 150.8, 137.6, 133.6, 131.8, 131.5, 128.5, 127.9, 127.8, 123.5, 123.4, 116.5, 114.6, 97.6, 21.0; IR (CHCl₃): 3391, 3338, 2922, 2850, 1631, 1601, 1540, 1498, 1430, 1216, 1117, 773 cm⁻¹; HRMS (+ESI) calcd for $C_{16}H_{14}NO_2[M + H]^+$: 252.1025; found: 252.1027.

4-Amino-3-(3-methoxyphenyl)-2*H***-chromen-2-one** (3ia). Applying the general experimental procedure using 3-methoxy-2-(trimethylsilyl)phenyl trifluoromethanesulphonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-aminocoumarin (0.040 g; 0.25 mmol, 1 equiv.), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-3-(3-methoxyphenyl)-2*H*-chromen-2-one **3ia** was obtained as a white solid (0.046 g, 69% yield) after purification by flash chromatography on silica gel (60% EtOAc/hexane); mp 214–216 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.13 (dd, J_1 =

1.4 Hz, $J_2 = 8.4$ Hz, 1H), 7.59–7.62 (m, 1H), 7.31–7.37 (m, 3H), 6.85–6.92 (m, 3H), 6.82 (br s, 2H), 3.76 (s, 3H); ¹³C NMR (126 MHz, DMSO-d₆): δ 160.9, 159.5, 152.6, 151.0, 135.0, 132.0, 129.8, 123.6, 123.5, 123.2, 116.7, 116.4, 114.6, 113.0, 97.5, 55.0; IR (CHCl₃): 3398, 3340, 2927, 2850, 1664, 1625, 1597, 1259, 1219, 1025, 989, 772 cm⁻¹; HRMS (+ESI) calcd for C₁₆H₁₄NO₃[M + H]⁺: 268.0974; found: 268.0974.

4-Amino-3-(naphthalen-2-yl)-2H-chromen-2-one (3ja). Applying the general experimental procedure using 1-(trimethylsilyl)-2naphthyl trifluoromethanesulfonate (0.07 mL; 0.25 mmol, 1 equiv.), 4-aminocoumarin (0.040 g; 0.25 mmol, 1equiv), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL), 4-amino-3-(naphthalen-2-yl)-2H-chromen-2-one 3ja was obtained as a white solid (0.048 g, 67% yield) after purification by flash chromatography on silica gel (40% EtOAc/hexane); mp 274-276 °C; ¹H NMR (500 MHz, DMSO-d₆): δ 8.18 (dd, J_1 = 1.2 Hz, J_2 = 8.0 Hz, 1H), 7.93-7.97 (m, 3H), 7.88 (s, 1H), 7.61-7.65 (m, 1H), 7.44 (dd, *J*₁ = 1.7 Hz, *J*₂ = 8.4 Hz, 2H), 7.33–7.38 (m, 3H), 6.97 (br s, 2H); ¹³C NMR (126 MHz, DMSO-d₆): δ 161.2, 152.7, 151.3, 133.4, 132.7, 132.1, 131.5, 129.9, 129.3, 128.1, 128.0, 127.6, 126.1, 125.9, 123.7, 123.6, 116.7, 114.7, 97.5; IR (CHCl₃): 3463, 3352, 2922, 2850, 1632, 1602, 1549, 1501, 1219, 772 cm⁻¹; HRMS (+ESI) calcd for $C_{19}H_{14}NO_2[M + H]^+$: 288.1025; found: 288.1025.

4-Amino-3-(phenyl-2-d)-2H-chromen-2-one {3aa-(D)}. Applying the general experimental procedure using 2-(trimethylsilyl) phenyl trifluoromethanesulfonate (0.06 mL; 0.25 mmol, 1 equiv.), 4-aminocoumarin (0.040 g; 0.25 mmol, 1equiv), KF (0.043 g; 0.75 mmol, 3 equiv.), and 18-crown-6 (0.100 g; 0.37 mmol, 1.5 equiv.) in THF (3 mL) and D_2O (5.0 μ L, 0.25 mmol), 4-amino-3-(phenyl-2-d)-2H-chromen-2-one 3aa-D was obtained as a white solid (0.043 g, 72% yield) after purification by flash chromatography on silica gel (40% EtOAc/ hexane); mp 208–210 °C; ¹H NMR (500 MHz, CDCl₃): δ 8.17 (d, J = 8.0 Hz, 1H), 7.59–7.62 (m, 1H), 7.44–7.47 (m, 2H), 7.31–7.36 (m, 4H), 6.78 (br s, 2H); 13 C NMR (126 MHz, CDCl₃): δ 160.9, 152.5, 150.8, 133.7 (d, J = 11.7 Hz), 131.9, 130.9, 128.6, 128.5, 127.1, 123.5 (d, J = 21.8 Hz), 116.5, 114.6, 97.4 (d, J = 4.1 Hz); IR (CHCl₃): 3468, 3342, 2917, 2855, 1672, 1628, 1607, 1548, 1500, 1427, 967, 771 cm⁻¹; HRMS (+ESI) calcd for C₁₅H₁₁DNO₂ **[M + H]**⁺: 239.0931; found: 239.0925.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

We are grateful to the Director, CSIR-NEIST, Jorhat, India, for his interest in this work and facilities. AS acknowledges the DST, New Delhi, India, for DST-Inspire fellowship grants.

Notes and references

- 1 (a) R. L. Augustine, Carbon-Carbon Bond Formation, Dekker, New York, 1979, vol. 1; (b) J. E. Baldwin and P. D. Magnus, Tetrahedron Organic Synthesis Series, Pergamon, New York, 1998, vol. 11; (c) Current Trends in Organic Synthesis, ed. C. Scolastico and F. Nocotra, Plenum, New York, 1999; (d) B. M. Trost and I. Fleming, Comprehensive Organic Synthesis, Pergamon, Oxford, 1991, vol. 3.
- 2 E. J. Corey and X. M. Cheng, *The Logic of Chemical Synthesis*, John Wiley & Sons, New York, 1989.
- 3 Selected reviews and books: (a) F. Diederich and P. J. Stang, Metal-Catalyzed Cross-Coupling Reactions, Wiley-VCH, Weinheim, 1998; (b) S. Bräse and A. de Meijere, in Metal-Catalyzed Cross-Coupling Reactions, ed. A. de Meijere and F. Diederich, Wiley-VCH, Weinheim, 2nd edn, 2004; (c) G. Dyker, Handbook of C-H Transformations: Applications in Organic Synthesis, Wiley-VCH, Weinheim, 2005.
- 4 (a) D. Nair, J. T. Scarpello, L. S. White, L. M. Freitos dos Santos, I. F. J. Vankelecom and A. G. Livingston, *Tetrahedron Lett.*, 2001, 42, 8219–8222; (b) J. Rivera-Utrilla, I. Bautista-Toledo, M. Ferro-Garcia and C. Moreno-Catilla, *Carbon*, 2003, 41, 323–330; (c) C. Garett and K. Prasad, *Adv. Synth. Catal.*, 2004, 346, 889–900; (d) The European Agency for the Evaluation of Medicinal Products, Committee for Proprietary Medicinal Products; London, 2002.
- 5 S. Yanagisawa, K. Ueda, T. Taniguchi and K. Itami, *Org. Lett.*, 2008, **10**, 4673–4676.
- 6 E. Shirakawa, Y. Hayashi, K.-I. Itoh, R. Watabe, N. Uchiyama, W. Konagaya, S. Masui and T. Hayashi, *Angew. Chem., Int. Ed.*, 2012, **51**, 218–221.
- 7 D. Wang, B. Ge, L. Li, J. Shan and Y. Ding, *J. Org. Chem.*, 2014, **79**, 8607–8613.
- 8 G. Li, Q. Yan, X. Gong, X. Dou and D. Yang, ACS Sustainable Chem. Eng., 2019, 7, 14009–14015.
- 9 (a) L.-Y. Xie, S. Peng, T.-G. Fan, Y.-F. Liu, M. Sun, L.-L. Jiang, X.-X. Wang, Z. Cao and W.-M. He, *Sci. China: Chem.*, 2019, **62**, 460–464; (b) L.-Y. Xie, L.-L. Jiang, J.-X. Tan, Y. Wang, X.-Q. Xu, B. Zhang, Z. Cao and W.-M. He, *ACS Sustainable Chem. Eng.*, 2019, 7, 14153–14160; (c) L.-Y. Xie, T.-G. Fang, J.-X. Tan, B. Zhang, Z. Cao, L.-H. Yang and W.-M. He, *Green Chem.*, 2019, **21**, 3858–3863.
- 10 For selected review: (a) C.-J. Li, Acc. Chem. Res., 2009, 42, 335–344; (b) S. A. Girard, T. Knauber and C.-J. Li, Angew. Chem., Int. Ed., 2014, 53, 74–100. For selected examples: (c) Z. Li and C.-J. Li, J. Am. Chem. Soc., 2004, 126, 11810–11811; (d) G. Deng and C.-J. Li, Org. Lett., 2009, 11, 1171–1174; (e) Y.-X. Jia and E. P. Kundig, Angew. Chem., Int. Ed., 2009, 48, 1636–1639; (f) X. Guo and C.-J. Li, Org. Lett., 2011, 13, 4977–4979; (g) A. P. Antonchick and L. Burgmann, Angew. Chem., Int. Ed., 2013, 52, 3267–3271; (h) Y. Aihara, M. Tobisu, Y. Fukumoto and N. Chatani, J. Am. Chem. Soc., 2014, 136, 15509–15512 and reference cited therein.
- 11 (a) T. Roy and A. T. Biju, Chem. Commun., 2018, 54, 2580–2594; (b) M. Feng and X. Jiang, Synthesis, 2017, 4414–4433;

(c) A. E. Goetz, T. K. Shah and N. K. Garg, *Chem. Commun.*, 2015, 51, 34–45; (d) A. Bhunia, S. R. Yetra and A. T. Biju, *Chem. Soc. Rev.*, 2012, 41, 3140–3152; (e) P. M. Tadross and B. M. Stoltz, *Chem. Rev.*, 2012, 112, 3550–3557; (f) A. V. Dubrovskiy and R. C. Larock, *Org. Biomol. Chem.*, 2013, 11, 191–218 and references cited therein.

- 12 Y. Himeshima, T. Sonoda and H. Kobayashi, *Chem. Lett.*, 1983, **12**, 1211–1214.
- 13 (a) K. M. Allan, C. D. Gilmore and B. M. Stoltz, Angew. Chem., Int. Ed., 2011, 50, 4488–4491; (b) E. Yoshioka, S. Kohtani and H. Miyabe, Angew. Chem., Int. Ed., 2011, 50, 6638–6642; (c) S. S. Bhojgude, D. R. Baviskar, R. G. Gonnade and A. T. Biju, Org. Lett., 2015, 17, 6270–6273.
- 14 For selected examples on cycloaddition reactions: (a) D. Pérez, D. Peña and E. Guitián, Eur. J. Org. Chem., 2013, 5981-6013; (b) T. Kaicharla, S. S. Bhojgude and Biju, Org. Lett., 2012, 14, 6238-6241; Α. Т. (c) S. S. Bhojgude, T. Kaicharla, A. Bhunia and A. T. Biju, Org. Lett., 2012, 14, 4098-4101; (d) Z. Liu, F. Shi, P. D. G. Martinez, C. Raminelli and R. C. Larock, J. Org. Chem., 2008, 73, 219-226; (e) T. Jin and Y. Yamamoto, Angew. Chem., Int. *Ed.*, 2007, 46, 3323-3325; (f) C. Dockendorff, S. Sahli, M. Olsen, L. Milhau and M. Lautens, J. Am. Chem. Soc., 2005, 127, 15028-15029; (g) J.-C. Castillo, J. Quiroga, R. Abonia, J. Rodriguez and Y. Coquerel, Org. Lett., 2015, 17, 3374-3377.
- 15 For selected examples on insertion reactions: (a) M. M. Ahire, R. Khan and S. B. Mhaske, Org. Lett., 2017, 19, 2134-2137; (b) H. Yoshida, R. Yoshida and K. Takaki, Angew. Chem., Int. Ed., 2013, 52, 8629-8632; (c) N. Qi, N. Zhang, S. R. Allu, J. Gao, J. Guo and Y. He, Org. Lett., 2016, 18, 6204-6207; (d) U. K. Tambar and B. M. Stoltz, J. Am. Chem. Soc., 2005, 127, 5340-5341; (e) Z. Liu and R. C. Larock, J. Am. Chem. Soc., 2005, 127, 13112-13113; (f) D. Rodríguez-Lojo, A. Cobas, D. Peña, D. Pérez and E. Guitián, Org. Lett., 2012, 14, 1363-1365; (g) R. Li, H. Tang, H. Fu, H. Ren, X. Wang, C. Wu, C. Wu and F. Shi, J. Org. Chem., 2014, 79, 1344-1355.
- 16 (a) A. Bhunia, T. Roy, P. Pachfule, P. R. Rajamohanan and A. T. Biju, Angew. Chem., Int. Ed., 2013, 52, 10040-10043; (b) H. Yoshida, E. Shirakawa, Y. Honda and T. Hiyama, Angew. Chem., Int. Ed., 2002, 41, 3247-3249; (c) F. Nawaz, K. Mohanan, L. Charles, M. Rajzmann, D. Bonne, O. Chuzel, J. Rodriguez and Y. Coquerel, Chem. - Eur. J., 2013, 19, 17578–17583; (d) Y. Chen and M. C. Willis, Org. Lett., 2015, 17, 4786-4789; (e) T. Roy, S. S. Bhojgude, T. Kaicharla, M. Thangaraj, B. Garai and A. T. Biju, Org. Chem. Front., 2016, 3, 71-76; (f) K. Biswas and F. Greaney, Org. Lett., 2011, 13, 4946-4949; M. (g) H. Yoshida, H. Fukushima, J. Ohshita and A. Kunai, J. Am. Chem. Soc., 2006, 128, 11040-11041; (h) S. Yoshida, T. Yano, Y. Misawa, Y. Sugimura, K. Igawa, S. Shimizu, K. Tomooka and T. Hosoya, J. Am. Chem. Soc., 2015, 137, 14071-14074.
- 17 Y. K. Ramtohul and A. Chartrand, *Org. Lett.*, 2007, **9**, 1029–1032.

- 18 C. Ni, L. Zhang and J. Hu, *J. Org. Chem.*, 2008, 73, 5699–5713.
- 19 R. A. Dhokale, P. R. Thakare and S. B. Mhaske, *Org. Lett.*, 2012, **14**, 3994–3997.
- 20 K. Mohanan, Y. Coquerel and J. Rodriguez, *Org. Lett.*, 2012, **14**, 4686–4689.
- 21 T. Pirali, F. Zhang, A. H. Miller, J. L. Head, D. Mc Ausland and M. F. Greaney, *Angew. Chem., Int. Ed.*, 2012, **51**, 1006– 1009.
- 22 C. M. Holden, S. M. A. Sohel and M. F. Greaney, *Angew. Chem., Int. Ed.*, 2016, **55**, 2450–2453.
- 23 (a) Z. Liu and R. C. Larock, Org. Lett., 2003, 5, 4673-4675;
 (b) Z. Liu and R. C. Larock, J. Org. Chem., 2006, 71, 3198-3209.
- 24 (a) K. Neog, A. Borah and P. Gogoi, J. Org. Chem., 2016, 81, 11971–11977; (b) K. Neog, D. Dutta, B. Das and P. Gogoi, Org. Lett., 2017, 19, 730–733; (c) K. Neog, B. Das and P. Gogoi, Org. Biomol. Chem., 2018, 16, 3138–3150; (d) A. Sharma and P. Gogoi, ChemistrySelect, 2017, 2, 11801–11805; (e) A. Sharma and P. Gogoi, Org. Biomol. Chem., 2019, 17, 333–346; (f) H. Hazarika, K. Neog, A. Sharma, B. Das and P. Gogoi, J. Org. Chem., 2019, 84, 5846–5854.
- 25 A. Furstner, D. N. Jumbam and N. Shi, Z. Naturforsch., 1995, 50, 326–332.
- 26 S. Peng, L. Wang, J. Huang, S. Sun, H. Guo and J. Wang, *Adv. Synth. Catal.*, 2013, **355**, 2550–2557.
- 27 M. Saha, K. Pradhan and A. R. Das, *RSC Adv.*, 2016, 6, 55033–55038.