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**New Acylated Phenolic Glycosides with ROS Scavenging Activity from *Psidium guajava* Leaves**

Yuanyuan Li, † Dongli Li, ‡ Qi An, † Hang Ma, ‡,§ Yu Mu, † Wenjun Qiao, ¶ Zengguang Zhang, † Jingsheng Zhang, ¶ Xueshi Huang, \*,† and Liya Li\*,†

† Institute of Microbial Pharmaceuticals, College of Life and Health Sciences, Northeastern University, Shenyang 110819, P. R. China

‡ School of Biotechnology and Health Sciences, Wuyi University, Jiangmen 529020, P. R. China

§ Bioactive Botanical Research Laboratory, Department of Biomedical and Pharmaceutical Sciences, College of Pharmacy, University of Rhode Island, Kingston, RI, 02881, United States

¶ Affiliated Hospital of Liaoning University of Traditional Chinese Medicine, Shenyang 110032, P. R. China

*\*Author to whom correspondence should be addressed*

Phone/Fax: 401-874-9367/5787; E-mail: [huangxs@mail.neu.edu.cn](mailto:huangxs@mail.neu.edu.cn)

Phone/Fax: 0086-24-83656122; E-mail: [lyli@mail.neu.edu.cn](mailto:lyli@mail.neu.edu.cn)

1 **ABSTRACT:** ROS and subsequent oxidative stress are reported to play important roles  
2 in chronic metabolic diseases. Plant-derived polyphenols, especially food-derived  
3 phenolics, have attracted a lot of attention due to their potential usage against oxidative  
4 stress-related diseases. The leaf of *Psidium guajava* (known as guava) is regarded as a  
5 good resource of polyphenols and its products are commercially available in Japan as  
6 functional foods against multiple chronic metabolism disorders. In the course of finding  
7 novel polyphenols with antioxidative activities from guava leaf, eleven acylated  
8 phenolic glycosides (**1-11**), including five new oleuropeic acid conjugated phenolic  
9 glycosides, named guajanosides A-E (**1**, **2**, and **5-7**), along with 17 known  
10 meroterpenoides (**12-28**) were isolated and identified. Their structures were determined  
11 by spectroscopic data analysis, chemical degradation and acid hydrolysis. Compounds  
12 **1**, **2**, and **5-11** displayed potent reactive oxygen species (ROS) scavenging activity in  
13 LPS-stimulated RAW 264.7 macrophage cells. Western blot revealed that compound  
14 **6** markedly increased the expression levels of nuclear factor-erythroid 2-related  
15 factor 2 (Nrf2), NAD(P)H quinone dehydrogenase 1 (NQO1) and glutamate-  
16 cysteine ligase catalytic subunit (GCLC). The current study revealed the presence  
17 of oleuropeic acid-derived phenolic glycosides in guava leaf, and highlighted the  
18 potential usage of this type of phenolics against oxidative stress-related metabolic  
19 diseases via activation of Nrf2 signaling pathway.

20 **KEYWORDS:** *Psidium guajava*, guava leaf, acylated phenolic glycosides, ROS  
21 scavenging activity, Nrf2 signaling pathway

## 22 INTRODUCTION

23       Reactive oxygen species (ROS) are a group of oxygen-containing redox-active  
24 intermediates released by endogenous and exogenous processes.<sup>1</sup> Although ROS have  
25 been reported to play important roles in signaling cascade for normal physiological  
26 functions, elevated cellular ROS levels which overcome endogenous antioxidant  
27 capacity would cause oxidative stress condition and damage multiple biomolecules.<sup>2,3</sup>  
28 ROS and subsequent oxidative stress are reported to be implicated in multiple chronic  
29 metabolic diseases, such as diabetes, cardiovascular disease, and neurodegenerative  
30 disease.<sup>1,4</sup> Thus, maintaining redox homeostasis is essential for normal physiological  
31 process in human body. Emerging evidences have demonstrated that natural  
32 antioxidants could play positive effects against oxidative stress-induced chronic  
33 metabolic diseases by acting as ROS scavengers.<sup>5,6</sup> Nuclear factor erythroid 2-related  
34 factor 2 (Nrf2), which is expressed in a wide range of tissues and cell types, is a crucial  
35 transcription factor involved in oxidation defensive system.<sup>7</sup> Although Nrf2 in the  
36 cytoplasm undergoes rapidly proteasomal degradation in the physiological condition, it  
37 will be accumulated and translocated into nucleus under oxidative stress condition,  
38 leading to the up-regulation of many antioxidant related enzymes, such as quinine  
39 oxidoreductase 1 (NQO1) and  $\gamma$ -glutamyl cysteine synthetase catalytic subunit (GCLC)  
40 to restore cellular redox homeostasis.<sup>8,9</sup> Extensive evidences demonstrate that Nrf2  
41 signaling pathway plays a pivotal role in the prevention of cell injury resulting from  
42 oxidative stress.<sup>10</sup> A lot of polyphenols, especially food-derived polyphenols, have been  
43 found to active Nrf2 signaling pathway to reduce ROS production, and then exert health

44 beneficial effects.<sup>9</sup>

45 *Psidium guajava* L., also known as guava, belongs to the family of Myrtaceae.  
46 Being important food crop, guava is cultivated in the tropical and subtropical regions  
47 worldwide, including south of China.<sup>11, 12</sup> Besides the popular usage of its edible fruit  
48 in food industry, guava leaf is also widely consumed as beverage in Japan, China and  
49 Korea against multiple chronic diseases, such as diabetes and obesity.<sup>13, 14</sup> Guava Leaf  
50 Tea (Bansoureicha®, Yakult Honsha, Tokyo, Japan) has been approved and  
51 commercially available as one of the Foods for Specified Health Uses in Japan,<sup>11</sup> and  
52 regarded as a potential natural source for finding novel bioactive compounds against  
53 metabolic diseases.<sup>10, 13</sup> Previous phytochemical studies have revealed that abundant  
54 polyphenols, including flavonoids,<sup>15</sup> tannins,<sup>16</sup> meroterpenoides<sup>17</sup> and benzophenones<sup>11</sup>  
55 are present in guava leaves. During the process of finding bioactive chemical  
56 constituents against chronic metabolic diseases from guava leaves, five new oleuropeic  
57 acid conjugated phenolic glycosides (**1**, **2**, and **5-7**) and six known analogues (**3**, **4**, and  
58 **8-11**), as well as 17 known meroterpenoides (**12-28**) were isolated and identified from  
59 guava leaves. Oleuropeic acid conjugated phenolic glycosides are a group of phenolics  
60 mainly found from Myrtaceous plants, especially from the genus *Eucalyptus*.<sup>18, 19, 20</sup>  
61 The numbers of this type of phenolics reported are very limited (less than 40). They  
62 usually possess a 5,7-dihydroxychromone, flavonone, or gallic acid skeleton, and are  
63 conjugated with oleuropeic acid fragment. Previous pharmacological studies revealed  
64 that this type of compounds possessed anti-tumor,<sup>21</sup> anti-inflammatory,<sup>22</sup> anti-  
65 melanogenesis,<sup>22</sup> and antibacterial activities.<sup>23</sup> To the best of our knowledge, this is the

66 first report of oleuropeic acid conjugated phenolic glycosides from guava. Meanwhile,  
67 to date, there is scarce of data about the ROS scavenging effect and mechanism of action  
68 of this type of compounds.

69 Thus, we initiate the current study with following objectives: (1) to investigate the  
70 underlying mechanisms of antioxidant action of these new acylated phenolic glycosides,  
71 as well as (2) to better understand the relationship between the present acylated phenolic  
72 glycosides and the health beneficial effects attributed to guava leaf. Macrophages are  
73 crucial for the immune system and are vulnerable to ROS. Bacterial lipopolysaccharide  
74 (LPS), an important component of the cell walls of gram-negative bacteria, can promote  
75 the production of ROS in macrophages and impact Nrf2 signaling pathway.<sup>24</sup> LPS-  
76 stimulated RAW 264.7 macrophage is widely accepted as a suitable model for the ROS-  
77 scavenging activity evaluation and related mechanism study.<sup>7, 24, 25</sup>

78 Herein, we reported the isolation and structure elucidation of acylated phenolic  
79 glycosides from guava leaves. In addition, ROS scavenging activities of these isolates  
80 were evaluated with LPS-stimulated RAW 264.7 cells. We also attempted to elucidate  
81 the underlying mechanisms of antioxidant action of these acylated phenolic glycosides  
82 by investigating the involvement of the Nrf2 signaling pathway.

### 83 **MATERIALS AND METHODS**

84 **General Experimental Procedures.** Optical rotations were determined in MeOH  
85 or CHCl<sub>3</sub> using an Anton Paar MCP200 automatic polarimeter (Graz, Austria). IR  
86 spectra were measured with a Bruker Tensor 27 FT-IR spectrometer (film). Ultraviolet  
87 spectra were recorder with a Beckman Coulter DU 730 nucleic acid/protein analyzer

88 (Brea, CA, USA). One- and two-dimensional NMR spectra were collected on a Bruker  
89 Advance III-600 MHz spectrometer (Bruker Co., Rheinstetten, Germany) with  
90 dimethyl sulfoxide (DMSO-*d*<sub>6</sub>) as the solvent. Semi-preparative HPLC was performed  
91 using an ODS column (250 mm × 10 mm, 5 μm, YMC-ODS-A, YMC Co. Ltd., Kyoto,  
92 Japan). ESI-MS were recorded on an Agilent 1290-6420 Triple Quadrupole LC-MS  
93 spectrometer (Santa Clara, CA, USA). HRESIMS was measured either with a Bruker  
94 Micro TOF-Q mass spectrometer (Bruker Daltonics, Billerica, MA) or a Shimadzu  
95 MALDI-TOF mass spectrometer (Shimadzu Corporation, Kyoto, Japan). Absorbance  
96 and fluorescence values were analyzed using a microplate reader (BioTek Synergy H1,  
97 BioTek Instruments, Winooski, VT, USA). Silica gel (100-200 mesh, 300-400 mesh,  
98 Qingdao Marine Chemical Ltd., Qingdao, China), Sephadex LH-20 (GE Healthcare  
99 Biosciences AB, Uppsala, Sweden), MCI gel (CHP-20P, Mitsubishi Chemical Corp.,  
100 Tokyo, Japan), and ODS-A (S-50 μm, 12 nm, YMC Co., Ltd., Kyoto, Japan) were used  
101 for column chromatography. D-glucose, L-glucose, D-galactose, and L-galactose were  
102 got from Sinopharm Chemical Reagent Co., Ltd. (Shenyang, People's Republic of  
103 China). L-cysteine methyl ester hydrochloride was purchased from Maya Reagent  
104 (Jiaxing, China). LPS, 3-[4,5-Dimethylthiazol-2-yl]-2,5-diphenyltetrazolium bromide  
105 (MTT), and minocycline (MINO) were obtained from Sigma Chemical Co. (St. Louis,  
106 MO, USA). All other chemicals and solvents were purchased from Sinopharm  
107 Chemical Reagent Co., Ltd.

108 **Plant Material.** The leaves of *P. guajava* were collected from Guangdong  
109 Province, in August 2016. The plant was identified by Professor Xiaoji Zheng

110 (Guangdong Jiangmen Chinese Traditional Medicine College, Jiangmen, China). A  
111 voucher specimen (NO 1610056) was deposited at the Institute of Microbial  
112 Pharmaceuticals, Northeastern University.

113 **Extraction and Isolation.** The air-dried leaves (4.0 kg) of *P. guajava* were  
114 powdered and defatted with petroleum ether ( $2 \times 10$  L) at room temperature. The  
115 residue was then extracted with 75% ethanol ( $3 \times 20$  L, 3 days each) at room  
116 temperature for three times. The supernatant was combined and evaporated under  
117 reduced pressure to afford a crude ethanol extract (1.0 kg). The ethanol extract was then  
118 suspended in water and partitioned successively with petroleum ether ( $3 \times 20$  L), ethyl  
119 acetate ( $3 \times 20$  L), and *n*-BuOH ( $3 \times 20$  L). After solvent removal, the ethyl acetate  
120 extract (200.0 g) was subjected to a MCI gel CHP-20P column, eluting with MeOH-  
121 H<sub>2</sub>O system (0, 20, 40, 60, 80, and 100%) to afford six fractions A-F. Fraction D (30.0  
122 g, 60% MeOH eluted fraction) was subjected to silica gel column chromatography  
123 (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 50:1, v/v) to yield three subfractions (D1-3). Subfraction D1 was  
124 separated by ODS column (MeOH/H<sub>2</sub>O 60%, v/v) and purified by Sephadex LH-20  
125 column chromatography with MeOH to yield compound **1** (5.9 mg). Subfraction D2  
126 was chromatographed over an ODS column with (MeOH/H<sub>2</sub>O 60%, v/v) elution system  
127 to yield compounds **2** (10.0 mg) and **10** (15.0 mg). Subfraction D3 was successively  
128 subjected to silica gel (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 30:1, v/v), ODS (MeOH/H<sub>2</sub>O 60%, v/v) and  
129 Sephadex LH-20 column (MeOH) to yield compounds **3** (59.2 mg) and **11** (20.6 mg).  
130 Fraction E (35.0 g, 80% MeOH eluted fraction) was put on a Sephadex LH-20 column  
131 and eluted with MeOH to yield three subfractions (E1-3). Subfraction E1 was

132 successively chromatographed over silica gel column (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 30:1, v/v), ODS  
133 column (MeOH/H<sub>2</sub>O 60%, v/v), and semi-preparative HPLC (55% MeOH/H<sub>2</sub>O, v/v) to  
134 yield compounds **4** (20.0 mg), **6** (6.8 mg), and **7** (7.5 mg). Subfraction E2 was separated  
135 over ODS column (MeOH/H<sub>2</sub>O 65%, v/v) to afford compound **8** (100.0 mg).  
136 Subfraction E3 was chromatographed over a silica gel column (CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 20:1,  
137 v/v), and further purified by semi-preparative HPLC (45% MeOH/H<sub>2</sub>O, v/v) to yield  
138 compound **9** (13.6 mg). In addition, the petroleum ether extract (160.0 g) of guava  
139 leaves was chromatographed over silica gel, ODS, Sephadex LH-20, and semi-  
140 preparative reversed-phase HPLC to afford 17 known meroterpenoides (**12-28**).  
141 Detailed isolation procedures for these known compounds are provided in the  
142 Supporting Information.

143 *Compound 1*: yellow amorphous powder;  $[\alpha]_{\text{D}}^{20} +8.0$  (c 1.0, MeOH); UV (MeOH)  
144  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 296 (3.54), 258 (4.03), 214 (4.31) nm; IR (film)  $\nu_{\text{max}}$  3346 , 2960, 2879,  
145 1702, 1661, 1620, 1591, 1424, 1390, 1357, 1259, 1178, 1079, 1025, 917 cm<sup>-1</sup>; <sup>1</sup>H and  
146 <sup>13</sup>C NMR data (in DMSO-*d*<sub>6</sub>) see Table 1; HRESI-MS *m/z* 1039.3823 [2M-H]<sup>-</sup> (calcd  
147 for C<sub>52</sub>H<sub>63</sub>O<sub>22</sub>, *m/z* 1039.3811).

148 *Compound 2*: yellow amorphous powder;  $[\alpha]_{\text{D}}^{20} -27.5$  (c 1.8, MeOH); UV (MeOH)  
149  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 296 (3.89), 258 (4.46), 214 (4.81) nm; IR (film)  $\nu_{\text{max}}$  3357 , 2961, 2929,  
150 2873, 1710, 1658, 1621, 1588, 1429, 1369, 1273, 1248, 1079, 1025, 1003, 916 cm<sup>-1</sup>;  
151 <sup>1</sup>H and <sup>13</sup>C NMR data (in DMSO-*d*<sub>6</sub>) see Table 1; HRESI-MS *m/z* 1123.4754 [2M-H]<sup>-</sup>  
152 (calcd for C<sub>58</sub>H<sub>75</sub>O<sub>22</sub>, *m/z* 1123.4750).

153 *Compound 5*: white amorphous powder;  $[\alpha]_{\text{D}}^{20} -20.0$  (c 1.0, MeOH); UV (MeOH)

154  $\lambda_{\max}$  (log  $\epsilon$ ) 258 (4.13), 218 (4.78), 196 (4.75) nm; IR (film)  $\nu_{\max}$  3371, 2967, 2931,  
155 1699, 1649, 1596, 1435, 1389, 1351, 1256, 1220, 1147, 1109, 924  $\text{cm}^{-1}$ ;  $^1\text{H}$  and  $^{13}\text{C}$   
156 NMR data (in  $\text{DMSO-}d_6$ ) see Table 1; HRESI-MS  $m/z$  511.1812  $[\text{M-H}]^-$  (calcd for  
157  $\text{C}_{24}\text{H}_{31}\text{O}_{12}$ ,  $m/z$  511.1816).

158 *Compound 6*: yellow amorphous powder;  $[\alpha]_{\text{D}}^{20}$  -12.0 (c 1.0, MeOH); UV (MeOH)  
159  $\lambda_{\max}$  (log  $\epsilon$ ) 354 (4.53), 258 (3.77), 192 (4.61) nm; IR (film)  $\nu_{\max}$  3274, 2924, 2855,  
160 1701, 1652, 1604, 1496, 1443, 1358, 1300, 1269, 1201, 1170, 1069, 1022, 997, 935  $\text{cm}^{-1}$ ;  
161  $^1\text{H}$  and  $^{13}\text{C}$  NMR data (in  $\text{DMSO-}d_6$ ) see Table 2; HRESI-MS  $m/z$  1223.3628  $[2\text{M-}$   
162  $\text{H}]^-$  (calcd for  $\text{C}_{62}\text{H}_{63}\text{O}_{26}$ ,  $m/z$  1223.3608).

163 *Compound 7*: yellow amorphous powder;  $[\alpha]_{\text{D}}^{20}$  +20.0 (c 0.6, MeOH); UV (MeOH)  
164  $\lambda_{\max}$  (log  $\epsilon$ ) 354 (3.96), 258 (4.17), 192 (4.73) nm; IR (film)  $\nu_{\max}$  3339, 2923, 2854,  
165 1698, 1651, 1604, 1507, 1445, 1359 1303, 1263, 1201, 1169, 1088, 1024, 999, 935  $\text{cm}^{-1}$ ;  
166  $^1\text{H}$  and  $^{13}\text{C}$  NMR data (in  $\text{DMSO-}d_6$ ) see Table 2; HRESI-MS  $m/z$  1223.3604  $[2\text{M-H}]^-$   
167 (calcd for  $\text{C}_{62}\text{H}_{63}\text{O}_{26}$ ,  $m/z$  1223.3608).

168 **Acid Hydrolysis and Monosaccharide Determination of Compounds 5-7.** Acid  
169 hydrolysis of compounds **5-7** were performed according to a previously reported  
170 method.<sup>26</sup> Briefly, compounds **5-7** (each 2 mg) were dissolved in 0.5 mL of MeOH and  
171 then hydrolyzed with 2 M HCl (2 mL) at 80 °C for 4 h. After cooling, the yield mixture  
172 was washed with  $\text{CH}_2\text{Cl}_2$  (2 mL) for three times, and the aqueous layer was dried with  
173 evaporator *in vacuo*. The dried powder was then dissolved in pyridine (1 mL) and  
174 reacted with L-cysteine methyl ester hydrochloride (2 mg) at 80 °C with stirring for 2  
175 h. Then *N*-trimethylsilylimidazole (0.3 mL) was added and reacted at 80 °C for another

176 2 h. After that, 1 M HCl (5 mL) was added to the mixture and then extracted with  
177 hexane (5 mL) for three times. The hexane extract was combined and submitted for  
178 GC-MS analysis. Gas chromatography was performed with a Shimadzu GCMS-  
179 QP2020 gas chromatograph mass spectrometer (Shimadzu Corporation, Kyoto, Japan)  
180 equipped with an AOC-20i auto injector (Shimadzu Corporation). The separation was  
181 conducted over a SH-Rxi-5Sil MS capillary column (30 m × 0.25 mm × 0.25 μm,  
182 Shimadzu Corporation) at a flow rate of 1 mL/min with a split ratio of 12:1. Helium  
183 was used as carrier gas and the injection volume was 1 μL. The inlet temperature was  
184 set as 270 °C. The initial oven temperature was 180 °C and raised to 280 °C by 5 °C/min,  
185 hold for 3 min, and then reached 300 °C by 20 °C/min, and hold for 4 min. The retention  
186 time of derivatives of standard D-glucose, L-glucose, D-galactose, and L-galactose  
187 were 13.45, 13.74, 13.84 and 14.21 min, respectively. The configurations of D-glucose  
188 for compounds **5** and **6**, and D-galactose for compound **7** were determined based on  
189 comparison of the retention time of corresponding derivatives with those of authentic  
190 sugars derivatives prepared in the same method.

191 **Methanolysis of Compound 8.** A solution of compound **8** (6.7 mg) in MeOH (1  
192 mL) containing 1% NaOMe was stirred at room temperature for 12 h. Then two drops  
193 of AcOH was added and the yield mixture was dried under vacuum. The residue was  
194 subjected to preparative thin layer chromatography (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 15:1, v/v) to yield  
195 (+)-oleuropeic acid methyl ester **1b** (2.6 mg):  $[\alpha]_D^{20} +76.0$  (c 1.0, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (in  
196 CDCl<sub>3</sub>, 600 MHz): δ<sub>H</sub> 7.01 (m, 1H, H-2), 3.75 (s, 3H, -OCH<sub>3</sub>), 2.56 (m, 1H, H-6a), 2.36  
197 (m, 1H, H-3a), 2.20 (m, 1H, H-6b), 2.04 (m, 1H, H-3b), 2.01 (m, 1H, H-5a), 1.57 (m,

198 1H, H-5b), 1.57 (m, 1H, H-4), 1.24 (s, 3H, H-10), 1.23 (s, 3H, H-9). <sup>13</sup>C-NMR (in  
199 CDCl<sub>3</sub>, 150 MHz): δ<sub>C</sub> 130.1 (C-1), 139.4 (C-2), 27.3 (C-3), 44.1 (C-4), 23.3 (C-5), 25.2  
200 (C-6), 167.8 (C-7), 72.4 (C-8), 26.6 (C-9), 27.4 (C-10), 51.6 (-OCH<sub>3</sub>).

201 **DPPH Assay.** The DPPH free radical scavenging assay was performed according  
202 to previously reported method.<sup>25</sup> Briefly, a series of different concentrations of the  
203 tested samples and the positive control ascorbic acid (6.25 to 200 μM) were mixed with  
204 freshly prepared DPPH (0.2 mM) in a 96-well microplate with total volume of 150 μL.  
205 After leaving the mixture reacted in the dark for 30 min, its absorbance was determined  
206 at 520 nm. All the wells with sample only were set as blank to obtain an absorbance  
207 value which was subtracted from the test sample readings. The free radical scavenging  
208 capacity was calculated using the following equation: % SC =  $\frac{A_{\text{control}} - (A_{\text{sample}} - A_{\text{blank sample}})}{A_{\text{control}}}$   
209 × 100%. The IC<sub>50</sub> values were calculated by SPSS 17.0 statistic software. Results were  
210 expressed as the mean ± SD of three replicates.

211 **Reactive Oxygen Species Scavenging Assay.** ROS scavenging activity of  
212 compounds **1-11** were evaluated with LPS-induced RAW 264.7 cell model.  
213 Intracellular ROS were determined by using a fluorescent probe 2',7'-dichlorofluoresin  
214 diacetate (DCFH-DA) (Beyotime Biotechnology, Shanghai, China), according to the  
215 manufacturer's instructions. Briefly, RAW 264.7 cells cultured in DMEM medium  
216 were seeded in 96-well-plates with 3 × 10<sup>4</sup> cells/well. After incubation for 24 h, LPS (1  
217 μg/mL) and the tested samples (final concentrations at 30 μM) were added and co-  
218 incubated for another 6 h. Then the supernatant of each well was removed and serum-  
219 free medium with DCFH-DA (10 μM) were added into wells for 60 min treatment in

220 the dark. After that, cells were washed with phosphate buffered saline and the  
221 fluorescence of each well was measured by using a microplate reader at an emission  
222 wavelength of 525 nm and an excitation wavelength of 488 nm. Minocycline (MINO)  
223 was used as the positive control. All samples were tested in triplicate and showed no  
224 cell toxicity against RAW 264.7 cells at 30  $\mu$ M as shown in MTT assay (See Supporting  
225 Information Figure S51).

226 **Western Blotting.** Compound **6** was selected for antioxidative mechanism study  
227 due to its superior activity in ROS scavenging assay. The effects of compound **6** on  
228 Nrf2 as well as its target downstream genes GCLC and NQO1 were investigated by  
229 western blot experiment. RAW 264.7 cells were seeded at a  $1 \times 10^6$  cells per well in a  
230 6-well plate. After 24 h incubation, the cells were pretreated with different  
231 concentrations of compound **6** (3.3, 10 or 30  $\mu$ M) for 2 h, followed by exposure to LPS  
232 (1  $\mu$ g/mL) for additional 6 h. After that, cells were lysed with a RIPA buffer and then  
233 centrifuged at 15 000 g for 15 min at 4 °C. The protein lysates were resolved by 10%  
234 sodium dodecyl sulfate polyacrylamide gel electrophoresis (SDS-PAGE), and  
235 electrophoretically transferred to PVDF membrane. The membrane was then blocked  
236 with 5% (w/v) non-fat dry milk in TBST, and then incubated with specific primary  
237 antibodies to Nrf2, GCLC, and NQO1 (Cell Signaling Technology, Danvers, MA, USA)  
238 at 4 °C overnight. B-actin was used as a loading control. Next day, the membrane was  
239 incubated with goat anti-rabbit IgG-HRP secondary antibody (Cell Signaling  
240 Technology, Danvers, MA, USA) at room temperature for 1 h. Chemiluminescence  
241 using ECL detection kit (GE Healthcare, Piscataway, NJ, USA) was detected by using

242 Bio-Rad ChemiDoc™ XRS+ System (Bio-Rad, Hercules, CA, USA). The experiment  
243 was repeated for three times.

244 **Statistical Analysis.** Quantitative data was presented as mean  $\pm$  S.E. The  
245 statistical analysis was performed by GraphPad Prism 4.0 (GraphPad Software Inc.,  
246 USA) and SPSS 17.0 (IBM., USA). The statistical significance between groups was  
247 interpreted by one-way ANOVA followed by Tukey's test. All statistical tests with  $p <$   
248 0.05 were considered significantly different.

## 249 **RESULTS AND DISCUSSION**

250 The ethyl acetate extract of *P. guajava* leaves was subject to column  
251 chromatography over MCI gel, silica gel, Sephadex LH-20 and semi-preparative HPLC  
252 to yield 28 phenolics, including 11 acylated phenolics glycosides (**1-11**) and 17  
253 meroterpenoides (**12-28**). Their structures were determined on the basis of  
254 spectroscopic data analysis, chemical degradation, acid hydrolysis, as well as by  
255 comparison of these to literature data (Figure 1). Among these isolates, five oleuropeic  
256 acid conjugated phenolic glycosides (**1**, **2**, and **5-7**) are new compounds, and another  
257 14 compounds (**3**, **4**, **8-10**, and **12-20**) are reported from this plant for the first time.

258 **Structure Elucidation of Isolated Phenolics.** Compound **1** was obtained as  
259 yellow amorphous powder,  $[\alpha]_D^{20} +8.0$  (c 1.0, MeOH). The negative HR-ESIMS of **1**  
260 gave a quasimolecular peak at  $m/z$  1039.3823  $[2M-H]^-$ , corresponding to the molecular  
261 formula of  $C_{26}H_{32}O_{11}$  with 11 indices of hydrogen deficiency. The IR spectrum of **1**  
262 showed absorption bands of hydroxyl ( $3346\text{ cm}^{-1}$ ), carbonyl ( $1702$  and  $1661\text{ cm}^{-1}$ ), and  
263 aromatic ring ( $1620$  and  $1591\text{ cm}^{-1}$ ) functionalities. The UV spectrum of **1** showed

264 absorption maximum at 292 and 256 nm, indicating its 5,7-dihydroxychromone  
265 skeleton.<sup>27</sup> This conclusion was supported by <sup>1</sup>H NMR data of **1** (in DMSO-*d*<sub>6</sub>, 600  
266 MHz, Table 1) which showed the presence of a hydroxyl group ( $\delta_{\text{H}}$  13.00, s, 1H, OH-  
267 5), two singlet aromatic protons ( $\delta_{\text{H}}$  6.14, s, 1H, H-3;  $\delta_{\text{H}}$  6.18, s, 1H, H-6), and a singlet  
268 methyl group at  $\delta_{\text{H}}$  2.31 (s, 3H, 2-Me). <sup>1</sup>H NMR spectrum of **1** also revealed the  
269 presence of an olefinic proton ( $\delta_{\text{H}}$  6.90, m, H-2''), two singlet methyl groups ( $\delta_{\text{H}}$  1.05,  
270 s, 6H, H<sub>3</sub>-9'' and H<sub>3</sub>-10''), and an anomeric proton of a glucosyl group ( $\delta_{\text{H}}$  4.64, d, 1H,  
271  $J = 10.2$  Hz, H-1'). The <sup>13</sup>C NMR and DEPT spectra of **1** displayed 26 resonances  
272 ascribed to two carbonyl carbons, two olefinic carbons, eight aromatic carbons, one  
273 oxygenated tertiary carbon, six methines (five oxygenated), four methylenes (one  
274 oxygenated), and three methyls. All of the proton signals were assigned to the  
275 corresponding carbons by HSQC spectrum analysis. The NMR data of **1** resembled  
276 those of the known compound 8- $\beta$ -C-glucopyranosyl-5,7-dihydroxy-2-  
277 methylchromone (isobiflorin), which was isolated from Myrtaceae family previously,<sup>27</sup>  
278 indicating compound **1** also possessed a 5,7-dihydroxy-2-methylchromone skeleton  
279 and C-glucosyl residue. The linkage of the glucosyl moiety to chromone skeleton  
280 through C-8 was deduced based on HMBC correlation signals from anomeric proton of  
281 glucose (H-1') to C-7 ( $\delta_{\text{C}}$  165.1), C-8 ( $\delta_{\text{C}}$  104.6) and C-9 ( $\delta_{\text{C}}$  157.1). The  $\beta$ -configuration  
282 of glucose was determined by the large  $J$  value of H-1' (10.2 Hz).<sup>28</sup> Further comparison  
283 of NMR data of **1** with isobiflorin revealed additional signals of a monoterpene, which  
284 was deduced as oleuropeic acid based on COSY and HMBC experiments. The linkage  
285 of oleuropeic acid moiety to  $\beta$ -glucose at C-6' was established by the HMBC

286 correlations from H-6'a ( $\delta_{\text{H}}$  4.30, m, 1H) to carbonyl carbon at  $\delta_{\text{C}}$  166.8 (C-7''), and  
287 supported by the remarkable downfield shift of C-6' (ca. 3 ppm), as well as upfield shift  
288 of C-5' (ca. 3 ppm) when compared to isobiflorin.<sup>27</sup> Although determination of the  
289 absolute stereochemistry of oleuropeic acid unit in **1** with chemical degradation was  
290 difficult due to its limited quantity, the presence of I-oleuropeoyl was deduced based  
291 on the co-isolation of (+)-I-oleuropeic acid (**1a**) ( $[\alpha]_{\text{D}}^{20} +76.0$ ,  $\text{CHCl}_3$ ) from guava leaves  
292 in the current project. This assumption was further supported by the methanolysis of  
293 compound **8**, another oleuropeic acid derivative isolated in the current project, which  
294 gave (+)-I-oleuropeic acid methyl ester (**1b**) ( $[\alpha]_{\text{D}}^{20} +76.0$ ,  $\text{CHCl}_3$ ).<sup>19</sup> On the basis of  
295 above evidence, the structure of compound **1** was determined as 8- $\beta$ -C-(6'-O-  
296 oleuropeoyl)-glucopyranosyl-5,7-dihydroxy-2-methylchromone and named as  
297 guajanoside A.

298 Compound **2** was obtained as yellow amorphous powder,  $[\alpha]_{\text{D}}^{20} -27.5$ , (c 1.8,  
299 MeOH). The molecular formula of **2** was deduced as  $\text{C}_{29}\text{H}_{38}\text{O}_{11}$  based on the  
300 quasimolecular peak at  $m/z$  1123.4754  $[\text{2M-H}]^-$  in negative HRESIMS and  $^{13}\text{C}$  NMR  
301 data. The IR spectrum of **2** highly resembled to that of **1**. Comparison of the 1D NMR  
302 data with those of compound **1** suggested the structure of compound **2** was also  
303 composed of oleuropeoyl and 8- $\beta$ -C-glucopyranosyl-5,7-dihydroxychromone moieties.  
304 However, a group of proton signals for isobutyl group ( $\delta_{\text{H}}$  2.52, d, 2H,  $J = 7.2$  Hz, H-  
305 1''';  $\delta_{\text{H}}$  2.26, m, 1H, H-2''';  $\delta_{\text{H}}$  1.00, d, 6H,  $J = 6.6$  Hz, H<sub>3</sub>-3''' and H<sub>3</sub>-4'''), which was  
306 deduced based on the  $^1\text{H}$ - $^1\text{H}$  COSY experiment (as shown in Figure 2), rather than  
307 methyl group were observed in **2**, indicating the substitute at C-2 in **2** was replaced by

308 isobutyl group. This conclusion was confirmed by the HMBC correlations from H-1'''  
309 to C-2 ( $\delta_C$  170.0), C-3 ( $\delta_C$  108.5), C-2''' ( $\delta_C$  26.9), C-3''' ( $\delta_C$  22.7) and C-4''' ( $\delta_C$  22.7),  
310 and supported by HRESIMS data. In addition, HMBC correlation signal from glucosyl  
311 H-2' ( $\delta_H$  5.42, t, 1H,  $J = 10.2$  Hz) to oleuropeoyl carboxylic carbon at  $\delta_C$  166.1  
312 suggested the C-2' linkage of oleuropeoyl group. This conclusion was supported by  
313 observed upfield chemical shifts of C-1', C-3', C-6', as well as downfield chemical  
314 shifts of C-2' and C-5' in the glucosyl moiety relative to **1** due to esterification effect  
315 (Table 1). The absolute stereochemistry of oleuropeoyl group in **2** was also assigned as  
316 I like **1** as discussed above. Consequently, the structure of compound **2** was determined  
317 as 8- $\beta$ -C-(2'-O-oleuropeoyl)-glucopyranosyl-5,7-dihydroxy-2-isobutylchromone and  
318 named as guajanoside B.

319 Compound **5** was obtained as a white amorphous powder. Its molecular formula,  
320 C<sub>24</sub>H<sub>32</sub>O<sub>12</sub>, was established on the negative HR-ESIMS ( $m/z$  511.1812 [M-H]<sup>-</sup>, calcd  
321 511.1816) and <sup>13</sup>C NMR data. The IR spectrum of **5** suggested the presence of hydroxyl  
322 group (3371 cm<sup>-1</sup>), carbonyl (1699 and 1649 cm<sup>-1</sup>), and benzene ring (1597 cm<sup>-1</sup>)  
323 functionalities. Analysis of the <sup>1</sup>H and <sup>13</sup>C NMR spectra of **5** revealed it also had  $\beta$ -  
324 glucopyranosyl and oleuropeoyl groups. In addition, <sup>1</sup>H NMR spectrum of **5** showed  
325 the presence of two singlet aromatic protons ( $\delta_H$  7.04, H-2;  $\delta_H$  7.08, H-6) and a  
326 methoxyl group ( $\delta_H$  3.72, s, 3H). The <sup>13</sup>C NMR spectrum of **5** showed six characteristic  
327 aromatic carbon signals at  $\delta_C$  105.3, 111.1, 136.9  $\times$  2, 150.4, and 152.9. The 1D NMR  
328 data of **5** resembled to known compound eucalmaidin B,<sup>19</sup> which was an oleuropeic  
329 acid derivative possessed a gallic acid unit in the structure. By comparison of 1D NMR

330 data of **5** with those of eucalmaidin B, an extra methoxyl group ( $\delta_{\text{H}}$  3.72, s;  $\delta_{\text{C}}$  56.5)  
331 was observed, which suggested one of the 17ethane17 groups on gallic acid fragment  
332 in eucalmaidin B was methylated. In the HMBC spectrum, the correlation signals from  
333 methoxyl protons to C-3 ( $\delta_{\text{C}}$  152.9) indicated methoxyl group was connected to C-3 of  
334 gallic acid. Additionally, the  $\beta$ -glucopyranose moiety located at C-4 was determined  
335 based on weak but detectable HMBC correlation between the aromatic proton ( $\delta_{\text{H}}$  4.78,  
336 d, 1H,  $J = 7.2$  Hz, H-1') and C-4 ( $\delta_{\text{C}}$  136.9). Acid hydrolysis of **5** with 2 M HCl liberated  
337 D-glucose, which was determined by subsequent GC-MS analysis following  
338 derivatization to trimethylsilylated L-cysteine adduct. The presence of *R*-oleuropeoyl  
339 unit was determined based on the co-isolation of **1a** and methanolysis result of known  
340 compound **8**. Therefore, the structure of **5** was determined as 4- $\beta$ -*O*-(6'-*O*-oleuropeoyl)-  
341 glucopyranosyl-5-hydroxy-3-methoxybenzoate and named as guajanoside C.

342 Guajanoside D (**6**) was obtained as a yellow amorphous powder with  $[\alpha]_{\text{D}}^{20} -12.0$   
343 (c 1.0, MeOH). Its molecular formula  $\text{C}_{31}\text{H}_{32}\text{O}_{13}$  was elucidated from the HRESIMS  
344 ( $m/z$  1223.3628 [2M-H]<sup>-</sup>) and  $^{13}\text{C}$  NMR data (Table 2). The IR spectrum of **6** showed  
345 absorption bands of hydroxyl ( $3274\text{ cm}^{-1}$ ), carbonyl ( $1701$  and  $1652\text{ cm}^{-1}$ ), and aromatic  
346 ring ( $1604$  and  $1496\text{ cm}^{-1}$ ) functionalities. The  $^1\text{H}$  NMR spectrum of **6** showed the  
347 presence of a hydroxyl group at  $\delta_{\text{H}}$  12.65 (5-OH), an ABX system at  $\delta_{\text{H}}$  6.82 (d, 1H,  $J$   
348 = 8.4 Hz, H-5'), 7.50 (s, 1H, H-2'), and 7.52 (d, 1H,  $J = 8.4$  Hz, H-6'), as well as two  
349 singlet aromatic protons at  $\delta_{\text{H}}$  6.17 (s, 1H, H-6) and 6.38 (s, 1H, H-8), suggesting the  
350 presence of a quercetin moiety. This conclusion was supported by UV and  $^{13}\text{C}$  NMR

351 spectral data. In addition, 1D NMR data (Table 2) revealed the presence of  $\beta$ -glucosyl  
352 with anomeric proton at  $\delta_{\text{H}}$  5.53 (d, 1H,  $J = 7.2$  Hz, H-1'') and anomeric carbon at  $\delta_{\text{C}}$   
353 100.7 (C-1''). Acid hydrolysis of **6** liberated D-glucose residue. Apart from these two  
354 moieties, the  $^{13}\text{C}$  NMR and DEPT data of **6** showed another 10 carbon signals,  
355 including one carbonyl carbon ( $\delta_{\text{C}}$  166.3), four olefinic carbons ( $\delta_{\text{C}}$  149.0, 139.3, 129.6,  
356 109.7), one 18ethane ( $\delta_{\text{C}}$  39.5), two methylenes ( $\delta_{\text{C}}$  30.7, 26.8, 24.4), and one methyl  
357 ( $\delta_{\text{C}}$  21.1). The 1D NMR data of **6** highly resembled to those of known compound  
358 **8**,<sup>29</sup> except that the signals for isopropanol unit in **8** were replaced by isopropenyl group  
359 in **6**. This indicated that the oleuropeic acid unit in **8** was changed to perillic acid and  
360 compound **6** was the dehydration derivative of **8**. This conclusion was confirmed by  
361 HMBC correlation signals from H<sub>2</sub>-9''' ( $\delta_{\text{H}}$  4.67, s, 1H; 4.72, s, 1H) to C-4''' ( $\delta_{\text{C}}$  39.5)  
362 and C-10''' ( $\delta_{\text{C}}$  21.1), and from H<sub>3</sub>-10''' ( $\delta_{\text{H}}$  1.69, s, 3H) to C-4''' and C-9''' ( $\delta_{\text{C}}$  109.7).  
363 The absolute stereochemistry of C-4''' in **6** was also assigned as I based on biogenesis  
364 consideration. Thus, the structure of guajanoside D (**6**) was established as shown in  
365 Figure 1.

366 Compound **7**, a yellow amorphous powder, had the same molecular formula of **6**  
367 as C<sub>31</sub>H<sub>32</sub>O<sub>13</sub> deduced from its HRESIMS data ( $m/z$  1223.3604 [2M-H]<sup>-</sup>, calcd  
368 1223.3608). The IR spectrum of **7** closely resembled to those of **6**. Analysis of 1D and  
369 2D NMR data suggested high similarities between the structures of **6** and **7** except for  
370 the sugar moiety. The monosaccharide in **7** was identified as  $\beta$ -D-galactopyranose by  
371 comparison of its NMR data with literature,<sup>30</sup> along with acid hydrolysis and  
372 subsequent GC-MS analysis. The HMBC correlations from H<sub>2</sub>-6'' ( $\delta_{\text{H}}$  4.05, m, 2H) to

373 C-7''' ( $\delta_C$  166.4) inferred that the monoterpene also located at C-6'' of  $\beta$ -D-galactose.  
374 The absolute stereochemistry of C-4''' in **7** was also assigned as I like **6** based on  
375 biogenesis consideration. Thus, the structure of guajanoside E (**7**) was determined as  
376 depicted in Figure 1.

377 Besides guajanosides A-E, six known acylated phenolic glycosides, including  
378 cypellocarpin C (**3**),<sup>31</sup> eucamalduside A (**4**),<sup>31</sup> cypellogin A (**8**),<sup>29</sup> cypellogin B (**9**),<sup>29</sup>  
379 quercetin-3-*O*- $\beta$ -D-(6''-*O*-*p*-coumaroyl)-galactopyranoside (**10**),<sup>30</sup> and guavaric A  
380 (**11**)<sup>32</sup> were also isolated and identified from guava leaves. Compounds **1-4** were  
381 glycosides bearing 5,7-dihydroxychromone, **5** was glycoside bearing galloyl group,  
382 whereas **6-11** were quercetin glycosides. All of the above phenol glycosides were  
383 esterified with (+)-oleuropeic acid except compounds **10** and **11**. In addition, 17  
384 meroterpenoides, including euglobal B1-1 (**12**),<sup>33</sup> euglobal Ib (**13**),<sup>33</sup> euglobal Ic (**14**),<sup>33</sup>  
385 euglobal III (**15**),<sup>34</sup> euglobal Iib (**16**),<sup>34</sup> euglobal-Iva (**17**),<sup>35</sup> euglobal Ivb (**18**),<sup>35</sup>  
386 euglobal V (**19**),<sup>36</sup> ecalrobusone E (**20**),<sup>37</sup> guajadial C (**21**),<sup>38</sup> guajadial D (**22**),<sup>38</sup>  
387 guajadial E (**23**),<sup>38</sup> guajudial (**24**),<sup>39</sup> macrocarpal A (**25**),<sup>40</sup> psiguajadial H (**26**),<sup>41</sup>  
388 psiguajadial I (**27**),<sup>41</sup> psiguajadial J (**28**)<sup>41</sup> were isolated and identified from the leaves  
389 of guava. Among these known isolates, 14 compounds (**3**, **4**, **8-10**, and **12-20**) are  
390 reported from this plant for the first time. Although Myrtaceous plants were reported to  
391 be rich in phenol glycosides esterified with oleuropeic acid, to the best of our  
392 knowledge, this is the first report of this type of compounds from *P. guajava*.

393 **Guava Leaf Phenolics Scavenged Free Radicals in the DPPH Assay and ROS**  
394 **in LPS-induced RAW264.7 Macrophages.** Since guava leaf tea has been widely used

395 to treat oxidative stress related diseases,<sup>42</sup> antioxidative activities of these isolates were  
396 evaluated with diphenylpicrylhydrazyl (DPPH) free radical scavenging assay. As  
397 shown in Table 3, acylated phenolic glycosides (**1-11**) showed superior antioxidant  
398 activity ( $IC_{50} < 200 \mu\text{M}$  except compound **3**) compared with the meroterpenoides ( $IC_{50}$   
399  $> 200 \mu\text{M}$ , see Supporting Information Table S1). It was noteworthy that the antioxidant  
400 activities of compounds **1**, **2**, and **5-11** ( $IC_{50}$  values ranged from 84 to 131  $\mu\text{M}$ ) were  
401 comparable to that of positive control ascorbic acid ( $IC_{50} = 108.60 \mu\text{M}$ ). The DPPH  
402 free radical scavenging activity difference among compounds **1-11** might be due to the  
403 hydroxyl group numbers, which was in agreement with previous report.<sup>25</sup> ROS  
404 overproduction would break the balance of oxidative and antioxidative system, thereby  
405 causing damage to healthy tissue cells.<sup>2</sup> To further confirm the antioxidant activities of  
406 these acylated phenolic glycosides, ROS scavenging activity of them were evaluated in  
407 RAW 264.7 macrophages. As shown in Figure 3, upon exposure to LPS for 6 h, ROS  
408 levels were significantly elevated (around 2-folds) in LPS-treated negative control  
409 group when compared to the non-treated control group. All of the tested compounds (**1-**  
410 **11**), as well as positive control minocycline, displayed potent ROS scavenging activity  
411 at 30  $\mu\text{M}$  in LPS-induced RAW 264.7 cells. Moreover, compounds **5-11** exhibited  
412 comparable ROS scavenging activity to the positive control minocycline (See support  
413 information Table S2). The above results suggested that acylated phenolic glycosides  
414 in guava leaf could reduce oxidative stress in LPS-stimulated RAW 264.7 macrophages,  
415 which was in agreement with the data obtained from the DPPH free radical scavenging  
416 assay. The aforementioned results highlighted the potential usage of these acylated

417 phenolic glycosides from guava leaves against oxidative stress-related chronic  
418 metabolic diseases.

419 **Effects of Guava Leaf Phenolics on Nrf2 Signaling Pathway.** Nrf2 is a major  
420 redox-sensitive transcription factor involved in modulating the expression of multiple  
421 antioxidant target genes, and plays a crucial role in protecting macrophages from  
422 damage due to elevated ROS.<sup>24, 43</sup> NQO1, a NAD(P)H-dependent reductase capable of  
423 reducing a broad range of substrates, is an important downstream gene of Nrf2.<sup>44</sup>  
424 Besides NQO1, GCLC is also known as a target gene of Nrf2 and plays a key role in  
425 the synthesis of glutathione, as well as related cellular redox process.<sup>45</sup> To understand  
426 the antioxidant mechanism of acylated phenolic glycosides in guava leaf, the effect of  
427 compound **6** (which showed superior ROS-scavenging activity) on Nrf2 signaling  
428 pathway was investigated. As showed in Figure 4, compound **6** significantly enhanced  
429 the total protein levels of Nrf2, NQO1 and GCLC in LPS-stimulated RAW264.7 cells  
430 at 10 and 30  $\mu$ M. Our results indicated that compound **6** could act as an Nrf2 activator  
431 to up-regulate the expression of its target antioxidant genes, and then reduce the  
432 production of ROS in macrophages, thereby exerting antioxidant effects.

433 In summary, 28 phenolics, including 11 acylated phenolic glycosides and 17  
434 meroterpenoids were isolated and identified from *P. guava* leaves. Among these  
435 isolates, five are new compounds, and 14 are being reported from *P. guava* for the first  
436 time. Isolated acylated phenolic glycosides display potent antioxidative activities in  
437 both DPPH and ROS scavenging assay. Preliminary molecular mechanism  
438 investigation reveals that acylated phenolic glycosides could active Nrf2 signaling

439 pathway, which reduce ROS level and exert antioxidative effects. The current study  
440 highlighted bioactive polyphenols present in guava leaf supporting its nutraceutical and  
441 functional food applications. Further *in vivo* studies are warranted to confirm the  
442 potential of these bioactive phenolics as dietary agents for the prevention and/or therapy  
443 of oxidative stress-related chronic metabolic diseases.

444 **ASSOCIATED CONTENT**

445 **Supporting Information**

446 IR, HRMS, and NMR spectra of compounds **1, 2, 5-7**, detailed information for isolation  
447 procedure and DPPH free radical scavenging activities of known compounds **12-28**, as  
448 well as cell proliferation assay result and ROS scavenging ratio of compounds **1-11** are  
449 available free of charge via the Internet at <http://pubs.acs.org>.

450 **AUTHOR INFORMATION**

451 **Corresponding Authors**

452 \* Tel/Fax: 0086-24-83656122; E-mail: [lyli@mail.neu.edu.cn](mailto:lyli@mail.neu.edu.cn) (L.-Y. Li)

453 \* Tel /Fax: 0086-24-83656106; E-mail: [huangxs@mail.neu.edu.cn](mailto:huangxs@mail.neu.edu.cn) (X.-S. Huang)

454 **Notes**

455 The authors declare no competing financial interest.

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592 **FIGURE CAPTIONS**593 **Figure 1.** Structures of compounds **1-11**.594 **Figure 2.** Key  $^1\text{H}$ - $^1\text{H}$  COSY and HMBC correlations of compounds **2** and **5**595 **Figure 3.** Effect of compounds **1-11** on the production of ROS in LPS-stimulated RAW596 264.7 macrophages. Data are the mean  $\pm$  SD for three independent experiments. ###  $p$ 597  $< 0.001$  vs. control group; \*\*  $p < 0.01$ , \*\*\*  $p < 0.001$  vs. LPS group.598 **Figure 4.** Effects of compound **6** on the protein expression of Nrf2, GCLC and NQO1

599 in LPS-stimulated RAW 264.7 macrophages. RAW 264.7 cells were pretreated with

600 different concentrations of compound **6** (3.3, 10 or 30  $\mu\text{M}$ ) for 2 h, followed by exposure601 to LPS (1  $\mu\text{g}/\text{mL}$ ) for additional 6 h. The cells were lysed with RIPA buffer and the

602 protein levels for total Nrf2, GCLC and NQO1 were measured by using immunoblot

603 analysis.  $\beta$ -actin was used as a loading control. MINO was used as positive control.

604 And all the experiments have been repeated three times independently. Data presented

605 as mean  $\pm$  S.E.,  $n = 3$ . #  $p < 0.05$ , ##  $p < 0.01$  as compared with the control group; \*  $p$ 606  $< 0.05$ , \*\*  $p < 0.01$ , \*\*\*  $p < 0.001$  as compared with the LPS group.

Table 1. <sup>1</sup>H (600 MHz) and <sup>13</sup>C NMR (150 MHz) Data for Compounds **1**, **2** and **5**

Position	<b>1</b>		<b>2</b>		<b>5</b>	
	<sup>13</sup> C, type	<sup>1</sup> H ( <i>J</i> in Hz)	<sup>13</sup> C, type	<sup>1</sup> H ( <i>J</i> in Hz)	<sup>13</sup> C, type	<sup>1</sup> H ( <i>J</i> in Hz)
<i>Aglycon</i>						
1					136.9 C	
2	167.1, C		170.0, C		105.3 CH	7.04 s
3	107.9, CH	6.14 s	108.5, CH	6.17 s	152.9 C	
4	182.2, C		182.4, C		136.9 C	
5	160.9, C		161.1, C		150.4 C	
6	99.4, CH	6.18 s	98.2, CH	6.16 s	111.1 CH	7.08 s
7	165.1, C		163.0, C		168.4 C	
8	104.6, C		103.1, C			
9	157.1, C		157.4, C			
10	103.4, C		104.1, C			
<i>2-isobutyl</i>						
			43.0, CH <sub>2</sub>	2.52 d (7.2)		
			26.9, CH	2.26 m		
			22.7, CH <sub>3</sub>	1.00 d (6.6)		
			22.7, CH <sub>3</sub>	1.00 d (6.6)		
2-CH <sub>3</sub>	20.1, CH <sub>3</sub>	2.31 s				
3-OCH <sub>3</sub>					56.5 CH <sub>3</sub>	3.72 s
5-OH		13.00 s		12.99 s		
<i>Sugar residue</i>						
1'	74.1, CH	4.64 d (10.2)	70.9, CH	4.82 d (10.2)	104.3 CH	4.78 d (7.2)
2'	71.3, CH	3.93 t (9.6)	72.6, CH	5.42 t (10.2)	74.4 CH	3.32 m
3'	78.9, CH	3.24 m	76.2, CH	3.47 m	76.5 CH	3.26 m
4'	70.6, CH	3.30 m	71.4, CH	3.20 m	70.7 CH	3.19 m
5'	78.5, CH	3.42 m	82.2, CH	3.25 m	74.6 CH	3.39 m
6'	64.3, CH <sub>2</sub>	4.10 m	62.1, CH <sub>2</sub>	3.47 m	63.9 CH <sub>2</sub>	4.02 m
		4.30 m		3.75 m		4.35 dd (11.4; 1.8)
<i>Monoterpene residue</i>						
1''	130.1, C		129.9, C		129.9 C	
2''	140.4, CH	6.90 m	139.8, CH	6.69 m	140.6 CH	6.81 s
3''	27.5, CH <sub>2</sub>	1.94 m	27.3, CH <sub>2</sub>	1.84 m	27.5 CH <sub>2</sub>	1.91 m
		2.23 m		2.14 m		2.22 m
4''	44.2, CH	1.39 m	44.1, CH	1.28 m	44.2 CH	1.37 m
5''	23.4, CH <sub>2</sub>	1.07 m	23.3, CH <sub>2</sub>	1.30 m	23.4 CH <sub>2</sub>	1.06 m
		1.90 m		1.81 m		1.89 m
6''	25.4, CH <sub>2</sub>	2.01 m	25.4, CH <sub>2</sub>	1.85 m	25.3 CH <sub>2</sub>	1.96 m
		2.37 m		2.13 m		2.31 m
7''	166.8, C		166.1, C		166.6 C	
8''	70.6, C		70.6, C		70.7 C	
9''	27.0, CH <sub>3</sub>	1.05 s	26.9, CH <sub>3</sub>	1.01 s	26.9 CH <sub>3</sub>	1.06 s
10''	27.5, CH <sub>3</sub>	1.05 s	27.5, CH <sub>3</sub>	1.01 s	27.6 CH <sub>3</sub>	1.06 s

Table 2.  $^1\text{H}$  (600 MHz) and  $^{13}\text{C}$  NMR (150 MHz) Data for Compounds **6** and **7**

Position	<b>6</b>		<b>7</b>	
	$^{13}\text{C}$ , type	$^1\text{H}$ ( <i>J</i> in Hz)	$^{13}\text{C}$ , type	$^1\text{H}$ ( <i>J</i> in Hz)
<i>Aglycon</i>				
2	156.7, C		157.0, C	
3	133.2, C		133.3, C	
4	177.8, C		177.3, C	
5	161.7, C		161.5, C	
6	99.2, CH	6.17 s	99.9, CH	6.09 s
7	164.8, C		166.4, C	
8	93.9, CH	6.38 s	94.3, CH	6.31 s
9	156.8, C		157.0, C	
10	104.2, C		103.1, C	
1'	121.5, C		121.0, C	
2'	115.6, CH	7.50 s	115.6, CH	7.48 d (1.8)
3'	145.3, C		145.5, C	
4'	148.9, C		148.8, C	
5'	116.5, CH	6.82 d (8.4)	115.9, CH	6.80 d (8.4)
6'	121.9, CH	7.52 d (8.4)	122.4, CH	7.63 dd (8.4; 1.8)
5-OH		12.65 s		12.63 s
<i>Sugar residue</i>				
1''	100.7, CH	5.53 d (7.2)	101.9, CH	5.45 d (8.4)
2''	74.3, CH	3.28 m	73.5, CH	3.64 m
3''	76.8, CH	3.26 m	71.4, CH	3.58 m
4''	70.7, CH	3.15 m	68.9, CH	3.62 m
5''	74.7, CH	3.33 m	73.5, CH	3.41 m
6''	63.5, CH <sub>2</sub>	3.95 m 4.25 d (11.4)	63.9, CH <sub>2</sub>	4.05 m
<i>Monoterpene residue</i>				
1'''	129.6, C		129.6, C	
2'''	139.3, CH	6.60 m	139.3, CH	6.59 br s
3'''	30.7, CH <sub>2</sub>	1.81 m 2.09 m	30.6, CH <sub>2</sub>	1.79 m 2.07 m
4'''	39.5, CH	1.92 m	39.1, CH	1.90 m
5'''	26.8, CH <sub>2</sub>	1.16 m 1.62 m	26.8, CH <sub>2</sub>	1.15 m 1.60 m
6'''	24.4, CH <sub>2</sub>	1.85 m 2.09 m	24.4, CH <sub>2</sub>	1.84 m 2.07 m
7'''	166.3, C		166.4, C	
8'''	149.0, C		148.8, C	
9'''	109.7, CH <sub>2</sub>	4.67 s 4.72 s	109.7, CH <sub>2</sub>	4.65 s 4.71 s
10'''	21.1, CH <sub>3</sub>	1.69 s	21.0, CH <sub>3</sub>	1.68 s

Table 3. DPPH free radical scavenging activity of compounds **1–11**

No.	IC <sub>50</sub> (μM)
<b>1</b>	131.87 ± 4.12
<b>2</b>	113.33 ± 3.44
<b>3</b>	> 200
<b>4</b>	180.00 ± 3.57
<b>5</b>	114.94 ± 3.19
<b>6</b>	97.68 ± 1.88
<b>7</b>	97.05 ± 7.83
<b>8</b>	103.95 ± 0.06
<b>9</b>	98.75 ± 1.22
<b>10</b>	92.55 ± 4.03
<b>11</b>	84.28 ± 4.68
Ascorbic acid <sup>a</sup>	108.60 ± 5.63

<sup>a</sup>Positive control substance

Figure 1.

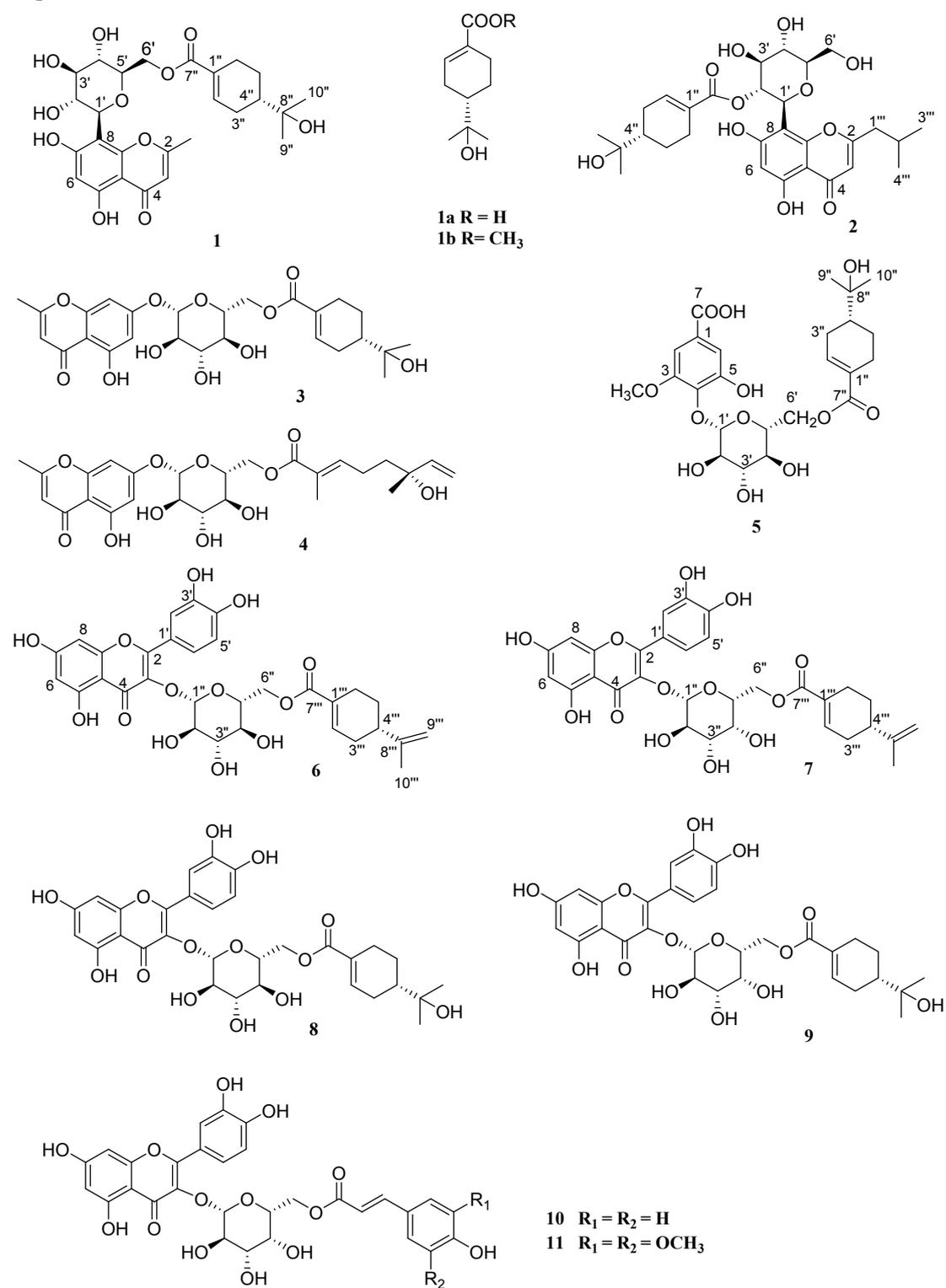


Figure 2.

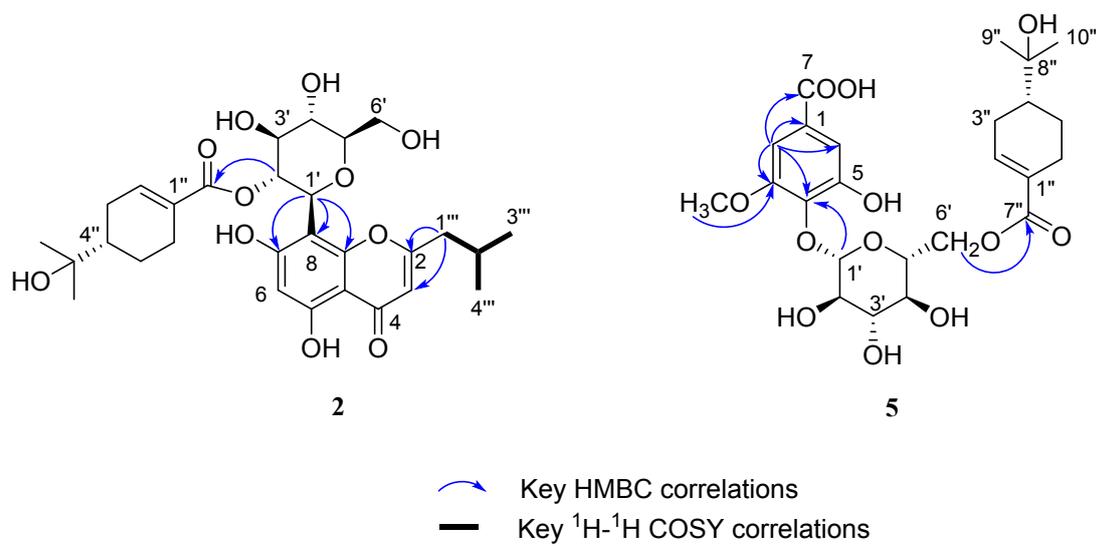


Figure 3.

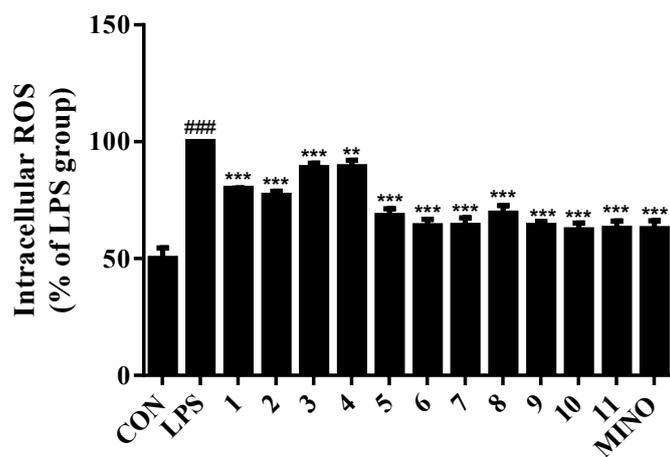
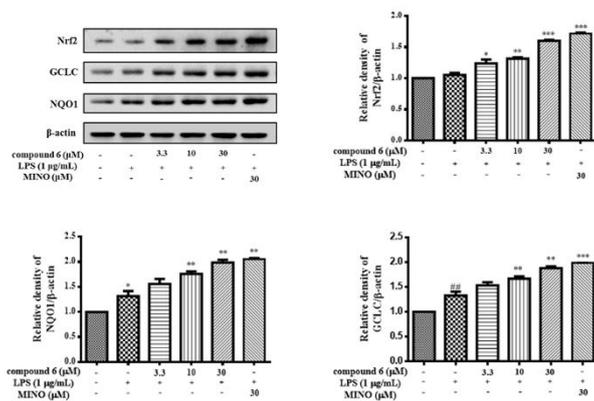


Figure 4.



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