# Efficient synthesis of tri- and difluoroacetyl hydrazides as useful building blocks for non-symmetrically substituted, fluoroalkylated 1,3,4-oxadiazoles

# Grzegorz Mlostoń<sup>1</sup>, Emilia Obijalska<sup>1</sup>\*, Alicja Żurawik<sup>1</sup>, Heinz Heimgartner<sup>2</sup>

<sup>1</sup> Department of Organic and Applied Chemistry, University of Łódź, Tamka 12, PL-91-403 Łódź, Poland; e-mail: emilkaobijalska@gmail.com

<sup>2</sup> Department of Chemistry, University of Zürich, Winterthurerstrasse 190, CH-8057 Zürich, Switzerland e-mail: heinz.heimgartner@chem.uzh.ch

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A convenient and efficient approach to 2-arylamino-5-fluoroalkyl-1,3,4-oxadiazoles has been established via heterocyclization of tri- and difluoroacetylated thiosemicarbazides using dicyclohexylcarbodiimide. A heterocyclization performed with selected thiosemicarbazides under basic conditions led to 4-aryl-5-fluoroalkyl-2,4-dihydro-3H-1,2,4-triazole-3-thiones in moderate yields. The starting fluoroacetylated thiosemicarbazides were prepared by fluoroacetylation of benzyloxycarbonyl-protected hydrazine with a corresponding anhydride, followed by hydrogenolytic deprotection and reaction with arylisothiocyanates. Fluoroacetylated semicarbazides were prepared similarly, but all attempts to achieve their heterocyclization were unsuccessful.

Keywords: fluoroalkylated heterocycles, hydrazides, 1,3,4-oxadiazoles, protected hydrazines, 1,2,4-triazole-3-thiones, heterocyclization.

Fluorine-containing heterocyclic compounds are widely applied as biologically active substances for the preparation of new drugs, agrochemicals, and materials with special properties.1 The development of methods for stereoselective synthesis of diverse fluorinated N-heterocycles is also of current interest.<sup>2</sup> Among heterocyclic systems which attract great attention, an important class comprises 1.3.4-oxadiazole derivatives<sup>3</sup> including fluorinated examples.<sup>4</sup>

Hydrazides of carboxylic acids (carbohydrazides) are known as versatile building blocks used for the preparation of heterocyclic compounds with diverse ring size. The target heterocycles can be synthesized starting either from substituted carbohydrazides or their derivatives, such as hydrazones, semicarbazides, thiosemicarbazides, etc.<sup>5</sup> A known method for the preparation of 1,3,4-oxadiazoles is the treatment of semicarbazides derived from carbohydrazides with sulfuric acid<sup>6</sup> or another acidic agent. Alternatively, 1,3,4-oxadiazoles can be obtained directly from carbohydrazides by the reaction with orthoesters.<sup>7</sup>

Due to our continuing interest in the development of methods for fluoroalkylation and the synthesis of fluorinated heterocycles,8 we focused attention on the exploration of the relatively little known tri- and difluoroacetyl hydrazides **1a**,**b** as potential precursors for the preparation

of fluorinated 1,3,4-oxadiazoles. The method selected for the synthesis of the latter was heterocyclization of semicarbazides of fluorinated carbohydrazides using sulfuric acid. In addition, analogous heterocyclizations could be performed starting with thiosemicarbazides and selenosemicarbazides.

Typically, the access to carbohydrazides is based on the reaction of the corresponding carboxylic esters with an equimolar amount of hydrazine hydrate.<sup>5</sup> In our case, the treatment of methyl trifluoroacetate with 1.0-5.0 equivalents of hydrazine hydrate in methanolic solution led to bis(trifluoroacetyl)hydrazine (2a) as a single product (Scheme 1). Its structure was confirmed by the ESI-MS registered for the crude product, which displayed a single signal for m/z 223. Increasing the excess of hydrazine

## Scheme 1







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hydrate to 1:10 or 1:20 resulted in a mixture of comparable amounts of compounds 2a and 1a or in the exclusive formation of compound 1a, respectively. Apparently, the initially formed compound 1a reacts rapidly with the starting ester yielding bis-adduct 2a as the final product. The formation of the analogous product 2b was observed using methyl difluoroacetate in the reaction with hydrazine hydrate.

In the course of the attempted synthesis of compound 1a, the tested protocol with 20-fold excess of hydrazine hydrate was practically useless, as the expected product could not be efficiently separated from the excess of hydrazine hydrate. In the light of the presented results, the reported syntheses of compound 1a from trifluoroacetates and hydrazine hydrate used in nearly equimolar amounts seem to be non-reproducible protocols.<sup>9</sup> However, instability of compound 1a was observed during the storage of the isolated product already at room temperature.9a

For that reason, the Boc- and Cbz-protected hydrazines 3a and 3b, respectively, were prepared and converted into the N-protected non-symmetrical tri- and difluoroacetyl hydrazines 4a-c by treatment with tri- or difluoroacetic anhydride (Scheme 2). Similar approaches for the synthesis of compound 1a based on the application of N-Bocprotected hydrazine have been only rarely reported.<sup>10</sup>

However, the attempted deprotection of the Bochydrazide 4a performed typically by treatment with trifluoroacetic acid in CH2Cl2 at 0°C or with HCl in dioxane at 0°C yielded exclusively the symmetrical product 2a. On the other hand, hydrogenolysis of compounds 4b and 4c over Pd/C in THF at room temperature afforded the desired tri- and difluoroacetyl hydrazides 1a and 1b, respectively, as the sole products (Scheme 2). In the case of compound 1a, the attempted isolation led again to its conversion into the symmetrical diacylhydrazine 2a. Therefore, freshly prepared compounds 1a and 1b in THF solution were treated with isocyanates, isothiocyanates, and isoselenocyanates 5 (X = O, S, Se, respectively, Scheme 2). The most reactive isocyanates 5a-d formed the expected semicarbazides 6a-f in high yields (Scheme 2, Table 1). Similarly, aromatic isothiocyanates 5e-h gave thiosemicarbazides 6g-j in good yields, and only in the case of the less reactive cyclohexylisothiocyanate the corresponding products 6k and 6l were formed in low yields and could not be isolated in pure form. Finally, the attempted reactions of compounds 1a,b with phenyl isoselenocyanate led to blackcolored, complex mixtures of unidentified products.

First experiments aimed at the preparation of fluorinated 1,3,4-oxadiazoles were performed with compound 6a using

Table 1. Reactants and product yields in the synthesis of N-acylated semicarbazides 6a-f and thiosemicarbazides 6g-l

Hydrazide	$R_{\rm F}$	Reagent	R	Х	Product	Yield, %
1a	CF <sub>3</sub>	5a	Ph	0	6a	88
1a	CF <sub>3</sub>	5b	$4\text{-FC}_6\text{H}_4$	0	6b	91
1a	CF <sub>3</sub>	5c	4-MeOC <sub>6</sub> H <sub>4</sub>	0	6c	86
1a	CF <sub>3</sub>	5d	<i>c</i> Hex	0	6d	93
1b	$\mathrm{CHF}_2$	5a	Ph	0	6e	90
1b	$\mathrm{CHF}_2$	5d	cHex	0	6f	94
1a	$CF_3$	5e	Ph	S	6g	77
1a	CF <sub>3</sub>	5f	$4\text{-FC}_6\text{H}_4$	S	6h	74
1a	CF <sub>3</sub>	5g	4-MeOC <sub>6</sub> H <sub>4</sub>	S	6i	68
1b	$\mathrm{CHF}_2$	5e	Ph	S	6j	80
1a	CF <sub>3</sub>	5h	cHex	S	6k	ca. 30*
1b	$\mathrm{CHF}_2$	5h	cHex	S	61	ca. 30*

Symmetrical N,N-bis(trifluoroacetyl)hydrazine (2a) and N,N-bis (difluoroacetyl)hydrazine (2b) were formed as by-products, and separation of these mixtures could not be achieved. The yields of compounds 6k and 61 were estimated based on the registered <sup>19</sup>F NMR spectra and the calculated ratios of 6k/2a and 6l/2b, respectively.

ethyl orthoacetate and catalytic amounts of p-TsOH, but in this case, a complex mixture of unidentified products was formed. Similarly, treatment of compound 6a with concd.  $H_2SO_4$  led to the same result. Finally, the reaction of thiosemicarbazides 6g-j with DCC in boiling THF afforded selectively 2-amino-1,3,4-oxadiazoles 7a-d in excellent yields (Scheme 2, Table 2). Analytically pure products were isolated after flash column chromatography. Their structures were established by means of spectroscopic methods. For example, the <sup>13</sup>C NMR spectrum of compound 7a revealed a singlet for the C-2 atom at 161.7 ppm and a quartet for the C-5 atom with  ${}^{2}J_{CF} = 42.9$  Hz at 147.9 ppm. The characteristic quartet with  ${}^{1}J_{CF} = 268.1$  Hz at 116.8 ppm was attributed to the CF<sub>3</sub> group. In addition, the <sup>19</sup>F NMR spectrum showed the signal of the CF<sub>3</sub> group at -64.6 ppm.

Table 2. Yields of fluoroalkylated 2-amino-1,3,4-oxadiazoles 7a-d

Thiosemicarbazide	$R_{\rm F}$	R	1,3,4-Oxadiazole	Yield, %
6g	CF <sub>3</sub>	Ph	7a	92
6h	$CF_3$	$4-FC_6H_4$	7b	95
6i	$CF_3$	$4-MeOC_6H_4$	7c	80
6j	$\mathrm{CHF}_2$	Ph	7d	82

It is worth mentioning that an efficient method to obtain 2-amino-1,3,4-oxadiazoles comprises the reaction of thiosemicarbazides of type 6g-l or analogous with TsCl and pyridine.<sup>11</sup> In some cases, the same product could be obtained from the corresponding semicarbazide, albeit in significantly lower yield and after longer reaction time. The reported formation of compound **7a** as the exclusive product was confirmed also in our laboratory.

In an extension of the study, fluorinated thiosemicarbazides **6g** and **6j** were heated in an aqueous solution of NaOH (2%), and 1,2,4-triazole-3-thiones **8a** and **8b**, respectively, were obtained in moderate yields (Scheme 2). This observation confirms the behavior of thiosemicarbazides under basic conditions.<sup>12</sup> Again, the attempted cyclization of the corresponding semicarbazides **6a**–**f** (X = O) under analogous conditions was unsuccessful.

The present study has shown that, in addition to other methods for the preparation of 2-amino-1,3,4-oxadiazoles based on the heterocyclization of semicarbazides, the corresponding thiosemicarbazides, including fluorinated derivatives, can also be used for this purpose. The heterocyclization of fluorinated thiosemicarbazides can be efficiently performed using DCC as the activating and H<sub>2</sub>Sbinding reagent. This protocol supplements the already reported method with p-TsCl and pyridine in THF solution.<sup>11</sup> According to our observation, yields of the final products are comparable in both methods. However, the elimination of pyridine (or triethylamine) as an additional component of the reaction mixture can be considered as an advantageous feature of the present protocol. In both procedures, the heterocyclization occurs chemoselectively *via* the conversion of the more nucleophilic C=S function into a good leaving group and the subsequent nucleophilic attack of the oxygen atom leading to 2-amino-1,3,4oxadiazoles as the final products. This type of chemoselectivity is preserved also in thiosemicarbazides containing an electron-deficient R<sub>F</sub>CO group. Another method for the synthesis of symmetrical 2,5-bis-(perfluoroalkyl)substituted 1,3,4-oxadiazoles from 1,2-diacylhydrazines, prepared stepwise by twofold acylation of hydrazine, is the cyclization by treatment with POCl<sub>3</sub> as activating and dehydrating agent.<sup>9b</sup> This approach can be considered as a pioneering method for the synthesis of 1.3.4-oxadiazoles.<sup>13</sup>

The presented method for the preparation of fluorinated 2-amino-1,3,4-oxadiazoles requires an efficient access to fluorinated monoacetylhydrazines. Our study showed that they can be synthesized from Cbz-protected hydrazine and the corresponding fluorinated acetanhydrides. After the deprotection, the obtained carbohydrazides have to be reacted immediately with isocyanates or isothiocyanates to give the required semicarbazides and thiosemicarbazides, respectively. Otherwise they undergo spontaneous disproportionation yielding symmetrical 1,2-diacylhydrazines, which are of limited importance for further applications.

### **Experimental**

IR spectra were recorded on a NEXUS FT-IR spectrophotometer in KBr pellets. The <sup>1</sup>H, <sup>13</sup>C, and <sup>19</sup>F NMR spectra were recorded on a Bruker Avance III 600 spectrometer (600, 150, and 565 MHz, respectively) in DMSO- $d_6$ , using the solvent signal as reference (2.50 and 39.51 ppm for <sup>1</sup>H and <sup>13</sup>C nuclei, respectively). Assignments of signals in <sup>13</sup>C NMR spectra were made using <sup>1</sup>H–<sup>13</sup>C HMQC and HMBC techniques. ESI mass spectra were obtained using a Varian 500 MS LS Ion Trap spectrometer. High-resolution mass spectra (electron ionization) were obtained on a Finningan MAT-95 spectrometer. Elemental analyses were recorded on an Elementar vario Micro cube apparatus. Melting points were determined on a Melt-Temp II apparatus (Aldrich) in capillaries.

All solvents were used as commercial products. Methyl trifluoroacetate, methyl difluoroacetate, trifluoroacetic acid anhydride, and difluoroacetic acid anhydride, were purchased from Fluorochem. Chloroform was dried over  $P_2O_5$  and freshly distilled prior to use. Tetrahydrofuran was dried over sodium in the presence of benzophenone and freshly distilled from the violet-colored solution prior to use. Anhydrous ethanol was dried over calcium oxide and  $CH_2Cl_2$  over calcium hydride. (*tert*-Butoxycarbonyl)-hydrazine (**3a**)<sup>14a</sup> and (benzyloxycarbonyl)hydrazine (**3b**)<sup>14b</sup> were obtained according to the known protocols.

Synthesis of the symmetrical 1,2-bis(trifluoroacetyl)hydrazine (2a). Ethyl trifluoroacetate (142 mg, 1.0 mmol) was added dropwise to the solution of hydrazine hydrate (50 mg, 1.0 mmol) in MeOH (~5.0 ml). The reaction mixture was stirred over 1 h at room temperature. Next, the solvent was evaporated, and the solid product was recrystallized from EtOH. Yield 160 mg (71%), mp 173-175°C (EtOH) (mp 176°C<sup>9b</sup>). IR spectrum, v , cm<sup>-1</sup>: 3346 (N-H, w), 3293 (w), 3213 (w), 3142 (w), 2989 (m), 2957 (w), 2858 (w), 2756 (w), 1612 (C=O, vs), 1520 (w), 1414 (m), 1203 (vs), 1190 (vs), 1146 (vs), 1098 (s), 973 (m). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 7.45 (2H, br. s, 2NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (*J*, Hz): 118.1 (q, <sup>1</sup>*J*<sub>CF</sub> = 284.4, 2CF<sub>3</sub>); 159.1 (q,  ${}^{2}J_{CF} = 31.0$ , 2C=O).  ${}^{19}F$  NMR spectrum, δ, ppm: -72.9 (6F, s, 2CF<sub>3</sub>). Found, m/z: 224.0013 [M]<sup>+</sup>. C<sub>4</sub>H<sub>2</sub>F<sub>3</sub>N<sub>2</sub>O<sub>2</sub>. Calculated, *m/z*: 224.0015.

Reactions of protected hydrazines 3a,b with fluorinated acid anhydrides (General method). A solution of BocNHNH<sub>2</sub>(3a) (3.0 mmol, 498 mg) or CbzNHNH<sub>2</sub>(3b) (3.0 mmol, 396 mg) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (~30 ml) was placed in an ice bath (~0°C). Next, triethylamine (4.5 mmol, 455 mg) was added, and after ca. 20 min, the appropriate fluorinated anhydride (3.2 mmol) was added dropwise while cooling the reaction flask in an ice bath. The reaction mixture was stirred at room temperature for 2 h. Then the solvent was evaporated, and the crude products were purified by column chromatography (SiO<sub>2</sub>, hexane–AcOEt, 1:1). Analytically pure samples were obtained after crystallization from an appropriate solvent.

*tert*-Butyl 2-(trifluoroacetyl)hydrazinecarboxylate (4a). Yield 608 mg (89%), colorless crystals, mp 129–131°C (Et<sub>2</sub>O–hexane) (mp 130–132°C<sup>10a</sup>). IR spectrum, v, cm<sup>-1</sup>: 3294 (N–H, br., s), 3196 (N–H, br., m), 3031 (m), 3000 (m), 2989 (m), 2914 (w), 1741 (C=O, s), 1690 (C=O, vs), 1571 (w), 1506 (s), 1457 (w), 1400 (w), 1369 (m), 1247 (s), 1207 (vs), 1162 (s), 1137 (s), 1052 (w), 1024 (w), 901 (w), 857 (w), 732 (m), 675 (w). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.43 (9H, s, 3CH<sub>3</sub>); 9.28 (1H, s, NH); 11.23 (1H, s, NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (*J*, Hz): 28.4 (3CH<sub>3</sub>); 80.6 (<u>C</u>(CH<sub>3</sub>)<sub>3</sub>); 116.3 (q, <sup>1</sup>*J*<sub>CF</sub> = 286.5, CF<sub>3</sub>); 154.9 (C=O); 156.7 (q, <sup>2</sup>*J*<sub>CF</sub> = 35.1, CF<sub>3</sub><u>C</u>=O). <sup>19</sup>F NMR spectrum,  $\delta$ , ppm: -74.0 (3F, s, CF<sub>3</sub>). Mass spectrum (ESI-), *m/z* (*I*<sub>rel</sub>, %): 227 [M-H]<sup>-</sup> (100%). Found, %: C 37.02; H 4.96; N 12.30. C<sub>7</sub>H<sub>11</sub>F<sub>3</sub>N<sub>2</sub>O<sub>3</sub>. Calculated, %: C 36.85; H 4.86; N 12.28.

Benzyl 2-(trifluoroacetyl)hydrazinecarboxylate (4b). Yield 605 mg (77%), colorless crystals, mp 109–111°C (Et<sub>2</sub>O–hexane) (mp 106–108°C<sup>10b</sup>). IR spectrum, v, cm<sup>-1</sup>: 3335 (N–H, br., m), 3268 (N–H, br., m), 3041 (m), 2968 (w), 2915 (w), 1761 s (C=O), 1689 s (C=O), 1549 (m), 1511 (m), 1357 (w), 1243 (vs), 1209 (vs), 1158 (s), 1046 (w), 969 (w), 746 (m), 697 (m). <sup>1</sup>H NMR spectrum, δ, ppm (*J*, Hz): 5.14 (2H, s, CH<sub>2</sub>); 7.35–7.39 (5H, m, H Ph); 9.75 (1H, br. s, NH); 11.44 (1H, br. s, NH). <sup>13</sup>C NMR spectrum, δ, ppm (*J*, Hz): 66.9 (CH<sub>2</sub>); 116.3 (q, <sup>1</sup>*J*<sub>CF</sub> = 286.5, CF<sub>3</sub>); 128.4, 128.6, 128.9 (C Ph); 136.7 (C-1 Ph); 155.8 (C=O); 156.7 (q, <sup>2</sup>*J*<sub>CF</sub> = 35.8, CF<sub>3</sub>⊆=O). <sup>19</sup>F NMR spectrum, δ, ppm (*J*, Hz): –74.0 (3F, s, CF<sub>3</sub>). Mass spectrum (ESI–), *m*/*z* (*I*<sub>rel</sub>, %): 261 [M–H]<sup>-</sup> (100). Found, %: C 45.67; H 3.56; N 10.85. C<sub>10</sub>H<sub>9</sub>F<sub>3</sub>N<sub>2</sub>O<sub>3</sub>. Calculated, %: C 45.81; H 3.46; N 10.68.

Benzyl 2-(difluoroacetyl)hydrazinecarboxylate (4c). Yield 682 mg (93%), colorless crystals, mp 116-117°C (Et<sub>2</sub>O-hexane). IR spectrum, v,  $cm^{-1}$ : 3316 (2N-H, br., s), 3091 (w), 3064 (w), 3040 (w), 3017 (w), 2996 (w), 2965 (w), 2903 (w), 1739 s (C=O), 1683 (C=O, s), 1546 (m), 1506 (m), 1464 (w), 1384 (w), 1358 (w), 1257 (m), 1205 (w), 1106 (w), 1056 (m), 960 (w), 750 (m), 701 (m), 580 (m). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (J, Hz): 5.13 (2H, s, CH<sub>2</sub>); 6.37 (1H, t,  ${}^{2}J_{\text{HF}}$  = 52.9, CHF<sub>2</sub>); 7.35–7.39 (5H, m, H Ph); 9.51 (1H, s, NH); 10.74 (1H, s, NH). <sup>13</sup>C NMR spectrum, δ, ppm (*J*, Hz): 66.7 (CH<sub>2</sub>); 108.6 (t,  ${}^{1}J_{CF}$  = 245.4, CHF<sub>2</sub>); 128.4, 128.5, 128.9 (C Ph); 136.9 (C-1 Ph); 156.0 (C=O); 162.3 (t,  ${}^{2}J_{CF} = 25.0$ , CHF<sub>2</sub>C=O).  ${}^{19}F$  NMR spectrum, δ, ppm (J, Hz): -126.8 (2F, d,  ${}^{2}J_{\text{HF}}$  = 52.9, CHF<sub>2</sub>). Mass spectrum (ESI+), m/z ( $I_{rel}$ , %): 267 [M+Na]<sup>+</sup> (100). Mass spectrum (ESI-), *m/z* (*I*<sub>rel</sub>, %): 243 [M-H]<sup>-</sup> (100). Found, %: C 49.32; H 4.13; N 11.60. C<sub>10</sub>H<sub>10</sub>F<sub>2</sub>N<sub>2</sub>O<sub>3</sub>. Calculated, %: C 49.19; H 4.13; N 11.47.

Preparation of fluorinated acid hydrazides via hydrogenolysis of N'-(benzyloxy)carbohydrazides 4b,c and their reactions with isocyanates and isothiocyanates (General method). A portion of the catalyst (10% Pd/C (60 mg)) was added to a solution of the CBz-protected carbohydrazide 4b,c (2.0 mmol) in anhydrous THF (~50 ml). Next, the reaction flask was equipped with a septum, a long syringe, and a balloon filled with hydrogen. Progress of the reaction was monitored by TLC (SiO<sub>2</sub>, hexane-AcOEt, 1:1), and as soon as the reaction was finished, the solution was filtered through a Celite pad. Next, the appropriate isocyanate 5a-d or isothiocyanate 5e-h (2.2 mmol) was added dropwise to the filtrate while cooling the reaction flask in an ice bath ( $\sim 0^{\circ}$ C). The mixture was stirred overnight at room temperature. After evaporation of the solvent, the crude products were purified by column chromatography (SiO<sub>2</sub>, hexane, then with increasing amount (40-90%) of AcOEt in hexane). Semicarbazides **6a**–**f** and thiosemicarbazides **6g**–**j** were obtained as amorphous solids by trituration with  $Et_2O$ .

N-Phenyl-2-(trifluoroacetyl)hydrazinecarboxamide (6a). Yield 436 mg (88%), colorless crystals, mp 176–178°C (mp 265.8–266.6<sup>11</sup>). IR spectrum, v, cm<sup>-1</sup>: 3297 (3N–H, br., vs), 3057 (m), 2890 (w), 1727 (C=O, vs), 1667 (C=O, vs), 1608 (s), 1574 (s), 1558 (m), 1360 (m), 1317 (m), 1303 (m), 1216 (s), 1189 (vs), 1165 (vs), 1128 (s), 1078 (w), 1044 (w), 1033 (w), 920 (m), 857 (w), 745 (m), 695 (m), 633 (m), 581 (w). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 6.98–7.00 (1H, m, H Ph); 7.26–7.29 (2H, m, H Ph); 7.45–7.46 (2H, m, H Ph); 8.44 (1H, br. s, NH); 8.96 (1H, br. s, NH); 11.24 (1H, br. s, NH). <sup>13</sup>C NMR spectrum, δ, ppm (*J*, Hz): 116.5  $(q, {}^{1}J_{CF} = 289.4, CF_{3}); 119.1, 122.7, 129.1 (C Ph); 139.8$ (C-1 Ph); 154.7 (C=O); 156.8 (q,  ${}^{2}J_{CF} = 34.8$ , CF<sub>3</sub>C=O).  $^{19}$ F NMR spectrum,  $\delta$ , ppm: -73.6 (3F, s, CF<sub>3</sub>). Mass spectrum (ESI+), m/z ( $I_{rel}$ , %): 248 [M+H]<sup>+</sup> (100). Mass spectrum (ESI–), m/z ( $I_{rel}$ , %): 246 [M–H]<sup>-</sup> (100). Found, m/z: 247.0571 [M]<sup>+</sup>. C<sub>9</sub>H<sub>8</sub>F<sub>3</sub>N<sub>3</sub>O<sub>2</sub>. Calculated, *m/z*: 247.0563.

N-(4-Fluorophenyl)-2-(trifluoroacetyl)hydrazinecarboxamide (6b). Yield 483 mg (91%), colorless crystals, mp 190-191°C. IR spectrum, v, cm<sup>-1</sup>: 3378 (N-H, br., m), 3319 (N-H, br., m), 3227 (N-H, br., m), 3065 (w), 2908 (w), 1720 (m), 1690 (C=O, s), 1664 (C=O, s), 1620 (w), 1571 (s), 1514 (s), 1411 (w), 1363 (w), 1277 (w), 1219 (s), 1198 (s), 1170 (s), 1124 (w), 1099 (w), 1053 (w), 1015 (w), 923 (w), 832 (w), 793 (w), 739 (w), 635 (w). <sup>1</sup>H NMR spectrum, δ, ppm: 7.10-7.13 (2H, m, H Ar); 7.45-7.48 (2H, m, H Ar); 8.49 (1H, br. s, NH); 9.04 (1H, br. s, NH); 11.23 (1H, br. s, NH). <sup>13</sup>C NMR spectrum, δ, ppm (*J*, Hz): 115.6, 115.7 (C Ar); 116.4 (q,  ${}^{1}J_{CF} = 286.6$ , CF<sub>3</sub>); 136.1 (C-1 Ar); 154.8 (C=O); 157.1 (q,  ${}^{2}J_{CF} = 35.7$ , F<sub>3</sub>C<u>C</u>=O); 158.0 (d,  ${}^{1}J_{CF} = 237.0$ , C-4 Ar).  ${}^{19}F$  NMR spectrum,  $\overline{\delta}$ , ppm: -117.3 (1F, s, CF); -73.7 (3F, s, CF<sub>3</sub>). Mass spectrum (ESI+), m/z $(I_{rel}, \%)$ : 266  $[M+H]^+$  (100). Mass spectrum (ESI–), m/z $(I_{\text{rel}}, \%)$ : 264  $[M-H]^-$  (100). Found, m/z: 265.0473  $[M]^+$ . C<sub>9</sub>H<sub>7</sub>F<sub>4</sub>N<sub>3</sub>O<sub>2</sub>. Calculated, *m*/*z*: 265.0469.

N-(4-Methoxyphenyl)-2-(trifluoroacetyl)hydrazinecarboxamide (6c). Yield 475 mg (86%), colorless crystals, mp 186-188°C. IR spectrum, v, cm<sup>-1</sup>: 3289 (3N-H, br., vs), 3146 (w), 3088 (w), 3056 (w), 3008 (w), 2954 (w), 2914 (w), 2837 (w), 1728 (C=O, s), 1668 (C=O, vs), 1608 (s), 1575 (s), 1540 (m), 1515 (s), 1465 (w), 144 (w), 1361 (w), 1304 (w), 1251 (s), 1218 (s), 1190 (s), 1173 (vs), 1128 (s), 1104 (m), 1031 (m), 920 (m), 836 (m), 799 (w), 836 (m), 739 (m), 636 (m). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 3.72 (3H, s, CH<sub>3</sub>); 6.85–6.88 (2H, m, H Ar); 7.33–7.36 (2H, m, H Ar); 8.36 (1H, br. s, NH); 8.79 (1H, br. s, NH); 11.18 (1H, br. s, NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (*J*, Hz): 55.6 (OCH<sub>3</sub>); 114.4, 114.5, 120.4, 120.5 (C Ar); 116.4 (q,  ${}^{1}J_{CF} = 286.6$ , CF<sub>3</sub>); 132.7 (C-1 Ar); 155.0 (C-4 Ar); 155.2 (C=O); 157.0 (q,  ${}^{2}J_{CF} = 35.4$ , CF<sub>3</sub>C=O).  ${}^{19}F$  NMR spectrum,  $\delta$ , ppm: -73.7 (3F, s, CF<sub>3</sub>). Mass spectrum (ESI+), m/z ( $I_{rel}$ , %): 300  $[M+Na]^+$  (100); 278  $[M+H]^+$  (75). Mass spectrum (ESI–), m/z ( $I_{rel}$ , %): 276 [M–H]<sup>-</sup> (100). Found, m/z: 277.0673 [M]<sup>+</sup>.  $C_{10}H_{10}F_3N_3O_3$ . Calculated, *m/z*: 277.0669.

*N*-Cyclohexyl-2-(trifluoroacetyl)hydrazinecarboxamide (6d). Yield 471 mg (93%), colorless crystals, mp  $151-152^{\circ}$ C. IR spectrum, v, cm<sup>-1</sup>: 3331 (N–H, br., m), 3267 (2N–H, br., m),

3116 (w), 3027 (w), 2935 (m), 2858 (m), 1748 (C=O, s), 1675 (C=O, s), 1636 (m), 1589 (m), 1551 (w), 1489 (w), 1465 (w), 1456 (w), 1358 (w), 1317 (w), 1253 (m), 1239 (m), 1204 (s), 1164 (s), 1133 (m), 1078 (m), 1018 (w), 914 (w), 891 (w), 795 (w), 737 (m), 656 (w). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.11–1.30 (5H, m) and 1.54–1.76 (5H, m, 5CH<sub>2</sub>); 3.39–3.40 (1H, m, CH); 6.38 (1H, br. s, NH); 7.99 (1H, br. s, NH); 11.02 (1H, s, NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (*J*, Hz): 25.0, 25.6, 33.3 (5 CH<sub>2</sub>); 48.8 (CH); 116.4 (q, <sup>1</sup>*J*<sub>CF</sub> = 286.5, CF<sub>3</sub>); 156.5 (C=O); 156.9 (q, <sup>2</sup>*J*<sub>CF</sub> = 35.5, CF<sub>3</sub><u>C</u>=O). <sup>19</sup>F NMR spectrum,  $\delta$ , ppm: –73.7 (3F, s, CF<sub>3</sub>). Mass spectrum (ESI+), *m/z* (*I*<sub>rel</sub>, %): 276 [M+Na]<sup>+</sup> (100), 254 [M+H]<sup>+</sup> (25). Mass spectrum (ESI–), *m/z* (*I*<sub>rel</sub>, %): 252 [M–H]<sup>-</sup> (100). Found, *m/z*: 253.1028 [M]<sup>+</sup>. C<sub>9</sub>H<sub>14</sub>F<sub>3</sub>N<sub>3</sub>O<sub>2</sub>. Calculated, *m/z*: 253.1033.

2-(Difluoroacetyl)-N-phenylhydrazinecarboxamide (6e). Yield 410 mg (90%), colorless crystals, mp 172-174°C. IR spectrum, v, cm<sup>-1</sup>: 3366 (N-H, br., s), 3300 (N-H, br., s), 3199 (N-H, br., s), 3099 (m), 3039 (m), 2926 (w), 2857 (w), 1946 (w), 1929 (w), 1717 (C=O, vs), 1693 (C=O, vs), 1600 (vs), 1564 (s), 1498 (s), 1444 (s), 1383 (w), 1320 (m), 1254 (m), 1218 (m), 1174 (m), 1151 (w), 1109 (s), 973 (w), 906 (w), 837 (w), 745 (s), 692 (m), 645 (m). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (*J*, Hz): 6.32 (1H, t,  ${}^{1}J_{CF} = 52.1$ , CHF<sub>2</sub>); 6.96-6.98 (1H, m, H Ph); 7.25-7.27 (2H, m, H Ph); 7.43-7.45 (2H m, H Ph); 8.39 (1H, br. s, NH); 8.89 (1H, br. s, NH); 10.64 (1H, br. s, NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (J, Hz): 108.4 (t,  ${}^{1}J_{\text{CF}}$  = 244.5, CHF<sub>2</sub>), 118.8, 122.2, 128.8 (C Ph); 139.0 (C-1 Ph); 154.7 (C=O); 161.9 (t,  ${}^{2}J_{CF} = 25.2$ , CHF<sub>2</sub>C=O). <sup>19</sup>F NMR spectrum,  $\delta$ , ppm (J, Hz): -126.5  $(2F, d, {}^{2}J_{FH} = 52.1, CHF_{2})$ . Mass spectrum (ESI+), m/z ( $I_{rel}, \%$ ): 252  $[M+Na]^+$  (100). Mass spectrum (ESI–), m/z ( $I_{rel}$ , %): 228  $[M-H]^-$  (100). Found, m/z: 229.0664  $[M]^+$ . C<sub>9</sub>H<sub>9</sub>F<sub>2</sub>N<sub>3</sub>O<sub>2</sub>. Calculated, *m/z*: 229.0657.

N-Cyclohexyl-2-(difluoroacetyl)hydrazinecarboxamide (6f). Yield 442 mg (94%), colorless crystals, mp 178–179°C. IR spectrum, v, cm<sup>-1</sup>: 3391 (N–H, br., m), 3351 (N–H, s), 3231 (N-H, s), 3104 (w), 3038 (w), 2935 (s), 2856 (m), 1731 (C=O, s), 1656 (C=O, vs), 1564 (vs), 1464 (w), 1454 (m), 1399 (w), 1370 (w), 1346 (w), 1320 (w), 1277 (w), 1258 (w), 1238 (m), 1190 (w), 1125 (s), 1095 (m), 1076 (m), 978 (w), 967 (w), 893 (w), 845 (w), 824 (w), 773 (w), 670 (w), 639 (w). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (J, Hz): 1.13– 1.28 (5H, m) and 1.65-1.76 (5H, m, 5CH<sub>2</sub>); 3.31-3.42 (1H, m, CH); 6.26 (1H, t,  ${}^{1}J_{CF} = 53.2$ , CHF<sub>2</sub>); 6.28 (1H, br. s, NH); 7.88 (1H, br. s, NH); 10.44 (1H, br. s, NH). <sup>13</sup>C NMR spectrum, δ, ppm (J, Hz): 25.0, 25.7, 33.4 (5 CH<sub>2</sub>); 48.6 (CH); 108.6 (t,  ${}^{1}J_{CF} = 244.2$ , CHF<sub>2</sub>); 156. 8 (C=O); 162.0 (t,  ${}^{2}J_{CF} = 25.2$ , CHF<sub>2</sub>C=O).  ${}^{19}F$  NMR spectrum,  $\delta$ , ppm (J, Hz): -126.5 (2F,  $\overline{d}, {}^{2}J_{\text{FH}} = 53.2$ , CHF<sub>2</sub>). Mass spectrum (ESI+), m/z ( $I_{rel}$ , %): 258 [M+Na]<sup>+</sup> (100), 236 [M+H]<sup>+</sup> (50). Mass spectrum (ESI–), m/z ( $I_{rel}$ , %): 234 [M–H]<sup>-</sup> (100). Found, m/z: 235.1133 [M]<sup>+</sup>. C<sub>9</sub>H<sub>15</sub>F<sub>2</sub>N<sub>3</sub>O<sub>2</sub>. Calculated, m/z: 235.1127.

*N*-Phenyl-2-(trifluoroacetyl)hydrazinecarbothiamide (6g). Yield 405 mg (77%), colorless crystals, mp 128–129°C (mp 132.0–133.0°C<sup>15</sup>). IR spectrum, v, cm<sup>-1</sup>: 3310 (N–H, br., m), 3158 (2N–H, br., s), 3034 (m), 2891 (w), 1735 (C=O, vs), 1628 (w), 1594 (m), 1538 (s), 1520 (s), 1498 (m), 1487 (m), 1451 (m), 1418 (w), 1362 (w), 1270 (w), 1217 (s), 1166 (vs), 1155 (vs), 1117 (m), 1029 (w), 1005 (w), 935 (w), 898 (w), 745 (m), 697 (m), 638 (w), 615 (w), 526 (m). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 7.18–7.46 (5H, m, H Ph); 9.75 (2H, br. s, 2NH); 11.52 (1H, br. s, NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (*J*, Hz): 116.8 (q, <sup>1</sup>*J*<sub>CF</sub> = 291.7, CF<sub>3</sub>); 125.6 (C-1 Ph); 128.7, 139.4 (C Ph); 155.8 (q, <sup>2</sup>*J*<sub>CF</sub> = 30.0, CF<sub>3</sub><u>C</u>=O); 179.9 (C=S). <sup>19</sup>F NMR spectrum,  $\delta$ , ppm: -73.6 (s, CF<sub>3</sub>). Mass spectrum (ESI+), *m/z* (*I*<sub>rel</sub>, %): 286 [M+Na]<sup>+</sup> (100), 264 [M+H]<sup>+</sup> (75). Mass spectrum (ESI–), *m/z* (*I*<sub>rel</sub>, %): 262 [M–H]<sup>-</sup> (100). Found, *m/z*: 263.0342 [M]<sup>+</sup>. C<sub>9</sub>H<sub>8</sub>F<sub>3</sub>N<sub>3</sub>OS. Calculated, *m/z*: 263.0334.

*N*-(4-Fluorophenyl)-2-(trifluoroacetyl)hydrazinecarbothiamide (6h). Yield 415 mg (74%), colorless crystals, mp 138–139°C. IR spectrum, v, cm<sup>-1</sup>: 3238 (3N–H, br., s), 3148 (w), 3063 (m), 1726 (C=O, m), 1616 (w), 1559 (m), 1518 (m), 1416 (w), 1362 (m), 1304 (w), 1249 (w), 1208 (m), 1191 (m), 1170 (m), 1126 (w), 1097 (w), 918 (w), 830 (w), 803 (w), 743 (w), 691 (w). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 7.17–7.20 (2H, m, H Ar); 7.45 (2H, m, H Ar); 9.85 (2H, br. s, 2NH); 11.53 (1H, br. s, NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (*J*, Hz): 116.0, 115.5 (C Ar); 116.9 (q, <sup>1</sup>*J*<sub>CF</sub>= 289.5, CF<sub>3</sub>); 136.1 (C-1 Ar); 157.3 (q, <sup>2</sup>*J*<sub>CF</sub> = 35.7, F<sub>3</sub>C<u>C</u>=O); 158.1 (d, <sup>1</sup>*J*<sub>CF</sub> = 237.1, C-4 Ar); 178.1 (C=S). <sup>19</sup>F NMR spectrum,  $\delta$ , ppm: –117.3 (1F, s, CF); –73.4 (3F, s, CF<sub>3</sub>). Mass spectrum (ESI+), *m*/*z* (*I*<sub>rel</sub>, %): 280 [M+H]<sup>+</sup> (100). Mass spectrum (ESI-), *m*/*z* (*I*<sub>rel</sub>, %): 280 [M–H]<sup>-</sup> (100). Found, *m*/*z*: 281.0243 [M]<sup>+</sup>. C<sub>9</sub>H<sub>7</sub>F<sub>4</sub>N<sub>3</sub>OS<sup>+</sup>. Calculated, *m*/*z*: 281.0240.

N-(4-Methoxyphenyl)-2-(trifluoroacetyl)hydrazinecarbothiamide (6i). Yield 401 mg (68%), colorless crystals, mp 140–141°C. IR spectrum, v, cm<sup>-1</sup>: 3214 (3N–H, br., s), 3046 (w), 2966 (w), 2943 (w), 2918 (w), 2845 (w), 1716 (C=O, m), 1613 (w), 1589 (w), 1545 (m), 1516 (m), 1469 (w), 1357 (m), 1302 (w), 1240 (m), 1215 (m), 1187 (m), 1168 (m), 1130 (w), 1101 (w), 1028 (m), 925 (m), 841 (w), 783 (w), 742 (w), 688 (w), 658 (w). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 3.77 (3H, s, CH<sub>3</sub>); 6.93-6.94 (2H m, H Ar); 7.26-7.27 (2H, m, H Ar); 9.73 (1H, br. s, NH); 9.88 (1H, br. s, NH); 11.47 (1H, br. s, NH).  $^{13}C$  NMR spectrum,  $\delta,$  ppm (J, Hz): 55.7 (OCH<sub>3</sub>); 114.0, 132.1 (C Ar); 127.8 (C-1 Ar); 116.4 (q,  ${}^{2}J_{CF} = 286.5$ , CF<sub>3</sub>); 156.8 (q,  ${}^{2}J_{CF} = 35.7$ , CF<sub>3</sub>C=O); 157.5 (C-4 Ar); 181.9 (C=S).  ${}^{19}$ F NMR spectrum, δ, ppm: -73.6 (3F, s, CF<sub>3</sub>). Mass spectrum (ESI+), m/z ( $I_{rel}$ , %): 294 [M+H]<sup>+</sup> (100). Mass spectrum (ESI–), m/z ( $I_{rel}$ , %): 292 [M–H]<sup>-</sup> (100). Found, m/z: 293.0437  $[M]^+$ . C<sub>10</sub>H<sub>10</sub>F<sub>3</sub>N<sub>3</sub>OS. Calculated, *m/z*: 293.0440.

**2-(Difluoroacetyl)-***N***-phenylhydrazinecarbothiamide** (6j). Yield 394 mg (80%), colorless crystals, mp 143–144°C. IR spectrum, v, cm<sup>-1</sup>: 3284 (N–H, br., s), 3163 (2N–H, br., vs), 3033 (m), 2886 (w), 1725 (C=O, vs), 1592 (w), 1523 (s), 1500 (s), 1452 (m), 1374 (w), 1335 (w), 1290 (w), 1241 (w), 1201 (m), 1150 (s), 1120 (m), 1090 (s), 1027 (w), 1004 (w), 970 (w), 935 (w), 911 (w), 820 (w), 742 (m), 697 (s), 668 (w), 636 (w), 616 (w). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (*J*, Hz): 6.30 (1H, t, <sup>2</sup>*J*<sub>CF</sub> = 50.8, CHF<sub>2</sub>); 7.19–7.43 (5H, m, H Ph); 9.79 (2H, br. s, 2NH); 10.91 (1H, br. s, NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (*J*, Hz): 108.6 (t, <sup>1</sup>*J*<sub>CF</sub> = 244.8, CHF<sub>2</sub>); 125.8 (C-1 Ph); 128.7, 139.4 (C Ph); 162.3 (t, <sup>2</sup>*J*<sub>CF</sub> = 25.0, CHF<sub>2</sub><u>C</u>=O); 181.4 (C=S). <sup>19</sup>F NMR spectrum, δ, ppm (*J*, Hz): -126.9 (2F, d,  ${}^{2}J_{FH} = 50.8$ , CHF<sub>2</sub>). Mass spectrum (ESI+), *m/z* (*I*<sub>rel</sub>, %): 268 [M+23]<sup>+</sup> (100), 246 [M+H]<sup>+</sup> (50). Found, *m/z*: 245.0312 [M]<sup>+</sup>. C<sub>9</sub>H<sub>9</sub>F<sub>2</sub>N<sub>3</sub>OS. Calculated, *m/z*: 245.0429.

Cyclization of thiosemicarbazides 6g-j to 1,3,4-oxadiazoles 7a–d (General method). Method A. Dicyclohexylcarbodiimide (305 mg, 1.48 mmol) was added to a solution of the appropriate thiosemicarbazide 6g-j (1.0 mmol) in anhydrous THF (~15 ml). The mixture was heated at reflux for 9 h, and then the solvent was evaporated. The obtained crude products were purified by column chromatography (SiO<sub>2</sub>, hexane–Et<sub>2</sub>O, 6:4).

Method B. Tosyl chloride (229 mg, 1.20 mmol) and  $Et_3N$  (222 mg, 2.20 mmol) were added to a solution of thiosemicarbazide **7g** (263 mg, 1.0 mmol) in *N*-methyl-pyrrolidin-2-one (NMP, 8 ml). The mixture was stirred for 3 h at room temperature, and, afterwards, H<sub>2</sub>O (~10 ml) and CH<sub>2</sub>Cl<sub>2</sub>(15 ml) were added. The organic layer was separated and the aqueous layer was washed with dichloromethane (2×15 ml). The organic layers were combined, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and filtered. Then the solvent was evaporated, and the product was purified by column chromatography (SiO<sub>2</sub>, hexane–Et<sub>2</sub>O, 6:4).

N-Phenyl-5-(trifluoromethyl)-1,3,4-oxadiazol-2-amine (7a). Yield 210 mg (92%, method A), 140 mg (61%, method B), colorless crystals, mp 129-131°C (Et<sub>2</sub>Ohexane) (mp 125.7–128.5°C<sup>11</sup>). IR spectrum, v, cm<sup>-1</sup>: 3186 (w), 3082 (w), 2996 (w), 2944 (w), 2879 (w), 1676 (s), 1603 (m), 1588 (m), 1503 (m), 1407 (w), 1346 (w), 1333 (w), 1231 (w), 1210 (m), 1165 (m), 1142 (m), 1119 (m), 1058 (w), 1032 (w), 968 (w), 890 (w), 729 (w), 749 (m), 688 (w), 627 (w). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 7.08–7.10 (1H, m, H Ph); 7.40–7.42 (2H, m, H Ph); 7.56–7.57 (2H, m, H Ph); 11.1 (1H, br. s, NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (*J*, Hz): 116.8 (q,  ${}^{1}J_{CF} = 268.1$ , CF<sub>3</sub>); 118.2, 123.4, 129.7 (C Ph); 138.1 (C-1 Ph); 147.9 (q,  ${}^{2}J_{CF} = 42.9$ , <u>C</u>CF<sub>3</sub>); 161.7 (C-2). <sup>19</sup>F NMR spectrum, δ, ppm: –64.6 (3F, s, CF<sub>3</sub>). Mass spectrum (ESI+), m/z ( $I_{rel}$ , %): 230 [M+H]<sup>+</sup> (100). Mass spectrum (ESI–), m/z ( $I_{rel}$ , %): 228 [M–H]<sup>-</sup> (100). Found, m/z: 229.0459 [M]<sup>+</sup>. C<sub>9</sub>H<sub>6</sub>F<sub>3</sub>N<sub>3</sub>O. Calculated, *m*/*z*: 229.0457.

*N*-(4-Fluorophenyl)-5-(trifluoromethyl)-1,3,4-oxadiazol-2-amine (7b). Yield 219 mg (95%, method A), colorless crystals, mp 147–149°C (Et<sub>2</sub>O–hexane). IR spectrum, v, cm<sup>-1</sup>: 3337 (w), 3191 (w), 3074 (w), 2998 (w), 2945 (w), 2874 (w), 1679 (s), 1668 (s), 1607 (m), 1596 (m), 1511 (vs), 1441 (w), 1387 (m), 1328 (m), 1232 (s), 1164 (m), 1116 (s), 1057 (m), 968 (w), 830 (m), 750 (m). <sup>1</sup>H NMR spectrum, δ, ppm: 7.24–7.27 (2H, m, H Ar); 7.58–7.61 (2H, m, H Ar); 11.11 (1H, br. s, NH). <sup>13</sup>C NMR spectrum, δ, ppm (*J*, Hz): 116.8 (q, <sup>1</sup>*J*<sub>CF</sub> = 268.0, CF<sub>3</sub>); 116.2, 116.3, 119.8, 119.9 (C Ar); 134.5 (C-1 Ar); 147.9 (q, <sup>2</sup>*J*<sub>CF</sub> = 43.0, <u>C</u>CF<sub>3</sub>), 158.4 (d, <sup>1</sup>*J*<sub>CF</sub> = 238.0, C-4 Ar), 161.7 (<u>C-</u>2). <sup>19</sup>F NMR spectrum, δ, ppm: –120.2 (1F, s, CF); –64.5 (3F, s, CF<sub>3</sub>). Mass spectrum (ESI+), *m*/*z* (*I*<sub>rel</sub>, %): 248 [M+H]<sup>+</sup> (100). Found, *m*/*z*: 247.0369 [M]<sup>+</sup>. C<sub>9</sub>H<sub>5</sub>F<sub>4</sub>N<sub>3</sub>O. Calculated, *m*/*z*: 247.0363.

*N*-(4-Methoxyphenyl)-5-(trifluoromethyl)-1,3,4-oxadiazol-2-amine (7c). Yield 193 mg (80%, method A), colorless crystals, mp 153–154°C (Et<sub>2</sub>O–hexane). IR spectrum, v, cm<sup>-1</sup>: 3281 (m), 3131 (w), 3061 (w), 3008 (w), 2954 (w), 2920 (w), 2836 (w), 1632 (vs), 1586 (s), 1540 (m), 1518 (s), 1457 (m), 1426 (m), 1383 (s), 1229 (s), 1171 (s), 1133 (s), 1060 (m), 1034 (m), 987 (w), 831 (w). <sup>1</sup>H NMR spectrum, δ, ppm: 3.75 (3H, s, OCH<sub>3</sub>); 6.98–6.99 (2H, m, H Ar); 7.48–7.49 (2H, m, H Ar); 10.85 (1H, br. s, NH). <sup>13</sup>C NMR spectrum, δ, ppm (*J*, Hz): 55.7 (OCH<sub>3</sub>); 114.9, 119.9 (C Ar); 116.8 (q, <sup>1</sup>*J*<sub>CF</sub> = 267.9, CF<sub>3</sub>); 131.2 (C-1 Ar); 147.6 (q, <sup>2</sup>*J*<sub>CF</sub> = 42.7, <u>C</u>CF<sub>3</sub>); 155.7 (C-4 Ar); 162.0 (C-2). <sup>19</sup>F NMR spectrum, δ, ppm: –64.5 (3F, s, CF<sub>3</sub>). Mass spectrum (ESI+), *m/z* (*I*<sub>rel</sub>, %): 260 [M+H]<sup>+</sup> (100). Found, *m/z*: 259.0564 [M]<sup>+</sup>. C<sub>10</sub>H<sub>8</sub>F<sub>3</sub>N<sub>3</sub>O<sub>2</sub>. Calculated, *m/z*: 259.0563.

5-(Difluoromethyl)-N-phenyl-1,3,4-oxadiazol-2-amine (7d). Yield 160 mg (82%, method A), colorless crystals, mp 157–158°C (Et<sub>2</sub>O–hexane). IR spectrum, v, cm<sup>-1</sup>: 3329 (w), 3252 (w), 3189 (w), 3073 (w), 3023 (m), 2955 (m), 2882 (m), 2748 (w), 2660 (w), 1668 (s), 1602 (s), 1589 (s), 1504 (s), 1479 (m), 1394 (w), 1375 (m), 1355 (m), 1292 (m), 1230 (w), 1108 (m), 1050 (s), 987 (m), 885 (w), 838 (m), 748 (s), 730 (m), 686 (m). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (J, Hz): 7.07 (1H, t,  ${}^{2}J_{\text{HF}} = 7.2$ , CHF<sub>2</sub>); 7.38–7.40 (3H, m, H Ph); 7.57-7.59 (2H, m, H Ph); 10.87 (1H, br. s, NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (*J*, Hz): 107.1; (t, <sup>1</sup>*J*<sub>CF</sub> = 235.2, CHF<sub>2</sub>); 117.9, 123.0, 129.6 (C Ph); 138.5 (C-1 Ph); 153.2 (t,  ${}^{2}J_{CF} = 28.6$ , <u>CCHF<sub>2</sub></u>); 161.7 (C-2).  ${}^{19}F$  NMR spectrum, δ, ppm (J, Hz): -119.9 (2F, d,  ${}^{2}J_{FH} = 51.4$ , CHF<sub>2</sub>). Mass spectrum (ESI+), *m/z* (*I*<sub>rel</sub>, %): 274 [M+Na]<sup>+</sup> (100). Mass spectrum (ESI–), m/z ( $I_{rel}$ , %): 250 [M–H]<sup>-</sup> (100). Found, m/z: 211.0561  $[M]^+$ . C<sub>9</sub>H<sub>7</sub>F<sub>2</sub>N<sub>3</sub>O. Calculated, *m/z*: 211.0552.

Cyclization of thiosemicarbazides 6g,i to 1,2,4triazole-3-thiones 8a,b (General method). An aqueous solution of NaOH (2%, ~2 ml) and an appropriate thiosemicarbazide 6g,i were heated at reflux for 3 h. Next, the reaction mixture was neutralized with acetic acid, the product was extracted with AcOEt ( $3 \times 25$  ml), and the combined organic layers were dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. After evaporation of the solvent, the crude products were purified by column chromatograpy (SiO<sub>2</sub>, hexane–AcOEt, 6:4).

**4-Phenyl-5-(trifluoromethyl)-2,4-dihydro-3***H***-1,2,4-<b>triazole-3-thione (8a)**.<sup>16</sup> Yield 42 mg (34%), pale-yellow crystals, mp 170–172°C. IR spectrum, v, cm<sup>-1</sup>: 3145 (m), 3091 (m), 3033 (m), 2911 (m), 2853 (m), 2741 (m), 1597 (m), 1498 (s), 1474 (s), 1446 (s), 1458 (s), 1344 (w), 1277 (s), 1238 (s), 1219 (s), 1176 (s), 1149 (s), 1093 (m), 1036 (m), 984 (m), 775 (m). <sup>1</sup>H NMR spectrum, δ, ppm: 7.50– 7.51 (2H, m, H Ph); 7.59–7.60 (3H, m, H Ph); 14.69 (1H, br. s, NH). <sup>13</sup>C NMR spectrum, δ, ppm (*J*, Hz): 117.1 (q, <sup>1</sup>*J*<sub>CF</sub> = 269.1, CF<sub>3</sub>); 128.9, 129.9, 130.8 (C Ph); 132.9 (C-1 Ph); 140.7 (q, <sup>2</sup>*J*<sub>CF</sub> = 39.9, <u>C</u>CF<sub>3</sub>), 171.3 (C=S). <sup>19</sup>F NMR spectrum, δ, ppm: – 62.7 (3F, s, CF<sub>3</sub>). Mass spectrum (ESI+), *m/z* (*I*<sub>rel</sub>, %): 244 [M–H]<sup>+</sup> (100). Mass spectrum (ESI–), *m/z* (*I*<sub>rel</sub>, %): 244 [M–H]<sup>-</sup> (100). Found, *m/z*: 245.0234 [M]<sup>+</sup>. C<sub>9</sub>H<sub>6</sub>F<sub>3</sub>N<sub>3</sub>S. Calculated, *m/z*: 245.0229.

**5-Difluoromethyl-4-phenyl-2,4-dihydro-3***H***-1,2,4-tri-azole-3-thione (8b)**.<sup>16</sup> Yield 41 mg (36%), colorless crystals, mp 187–189°C. IR spectrum, v , cm<sup>-1</sup>: 3249 (w), 3149 (w), 3089 (m), 3037 (m), 2999 (w), 2902 (m), 2848 (m), 2745 (m), 1594 (w), 1541 (m), 1497 (s), 1487 (s), 1435 (m), 1340 (w), 1323 (m), 1270 (s), 1229 (m), 1193 (w), 1176

(w), 1126 (m), 1077 (s), 1056 (s), 1015 (w), 823 (m), 776 (m), 738 (w), 724 (w), 690 (m). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (*J*, Hz): 7.04 (1H, t, <sup>2</sup>J<sub>HF</sub> = 51.6, CHF<sub>2</sub>); 7.46–7.47 (2H, m, H Ph); 7.57–7.62 (3H, m, H Ph); 14.41 (1H, br. s, NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (*J*, Hz): 107.8 (t, <sup>1</sup>J<sub>CF</sub> = 235.9, CHF<sub>2</sub>); 128.7, 129.8, 130.4 (C Ph); 133.3 (C-1 Ph); 144.9 (t, <sup>2</sup>J<sub>CF</sub> = 27.9, CHF<sub>2</sub>C); 170.5 (C=S). <sup>19</sup>F NMR spectrum,  $\delta$ , ppm (*J*, Hz): –119.2 (2F, d, <sup>2</sup>J<sub>FH</sub> = 50.8, CHF<sub>2</sub>). Mass spectrum (ESI+), *m/z* (*I*<sub>rel</sub>, %): 226 [M+H]<sup>+</sup> (100). Mass spectrum (ESI–), *m/z* (*I*<sub>rel</sub>, %): 226 [M–H]<sup>-</sup> (100). Found, *m/z*: 227.0329 [M]<sup>+</sup>. C<sub>9</sub>H<sub>7</sub>F<sub>2</sub>N<sub>3</sub>S. Calculated, *m/z*: 227.0323.

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