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Fine-tuning the electronic structure of heavy-atom-free BODIPY photosensitizers for fluorescence imaging and mitochondria-targeted photodynamic therapy

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Theranostics that combines both diagnosis and therapy into a single platform has recently emerged as a promising biomedical approach for cancer treatment; however, the development of efficient theranostic agents with excellent optical properties remains a challenge. Here, we report novel mitochondria-targeting BODIPY photosensitizers (**R-BODs**) that possess considerable singlet oxygen generation capabilities and good fluorescence properties for imaging-guided photodynamic therapy (PDT). The incorporation of sulfur atoms into the π -conjugated skeleton of BODIPY along with the introduction of different functional groups to the *meso*-position of the BODIPY core is essential for tuning the photophysical and photosensitizing properties. Notably, the MeOPh-substituted thiophene-fused BODIPY (**MeO-BOD**, R = *p*-methoxyphenyl) displayed the highest singlet oxygen generation capability ($\Phi_{\Delta} \approx 0.85$ in air-saturated acetonitrile) and a moderate fluorescence quantum yield ($\Phi_f = 17.11$). Furthermore, **MeO-BOD** showed good biocompatibility, low dark toxicity and superior fluorescence imaging properties in living cells. More importantly, the PDT efficacy of mitochondria-specific anchoring of **MeO-BOD** was remarkably amplified with an extremely low half-maximal inhibitory concentration (IC₅₀) value of 95 nM. We believe that the incorporation of an electron-donating group at the *meso*-position of the thiophene-fused BODIPY platform may be an effective approach for developing theranostic agents for precision cancer therapy.

Introduction

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Photodynamic therapy (PDT) is an effective clinical treatment strategy for malignant tumor.¹ In PDT process, a photosensitizer (PS) is activated under light irradiation, and the excited PS subsequently interacts with molecular oxygen to generate cytotoxic reactive oxygen species (ROS), which can oxidize biomolecules resulting in cancer cell death.^{2,3} In particular, theranostics has recently been recognized as a promising medical technology that combines diagnostic and therapeutic capabilities in one dose to achieve the real-time and precise monitoring of the therapeutic effect of the drug.⁴ Moreover, PDT combined with image-guided diagnosis has attracted considerable attention owing to its prominent advantages such as high spatiotemporal selectivity, noninvasiveness, and fewer side effects.5-7 Therefore, the design of novel PSs that can effectively produce both fluorescence and ROS is in high demand.

To date, numerous organic dyes, such as porphyrins, phthalocyanines, cyanine, squaraine, diketopyrrolopyrrole (DPP), and boron dipyrromethane (BODIPY) derivatives, have been developed as theranostic agents.⁷⁻¹³ Among these, BODIPY

dyes have attracted great interest as theranostic agents in photodynamic cancer therapy due to their excellent photochemical stability, good biocompatibility, high molar extinction coefficients, high quantum efficiencies of fluorescence and facile modification.¹¹⁻¹⁷ At present, for the sake of remarkable PDT efficiency of BODIPY dyes, the most popular approach is the introduction of heavy halogen atoms (Br and I) to promote spin-orbit coupling (SOC), which enhances intersystem crossing (ISC) and improves the singlet oxygen $({}^{1}O_{2})$ generation capability.¹⁸⁻²⁴ However, the incorporation of heavy atoms increases the dark-toxicity and quench fluorescence.²⁵ Thus, BODIPY PSs without heavy halogen atoms are preferred as theranostic agents. Recently, several approaches to enhance the ISC, such as the use of double excited states,^{26,27} spin converters,²⁸ and photoinduced electron transfer (PET),²⁹⁻³¹ have been implemented in the development of PSs without heavy atoms; however, the search for novel PSs is still required for PDT to reach its full potential.

Intriguingly, one alternative strategy to strengthen the ISC with an efficient triplet population is to increase the SOC while decreasing the singlet-triplet energy gaps; to do this, a thiophene moiety was introduced into the π -conjugated system of BODIPY,³²⁻³⁵ but the oncological applications of this approach have not been fully studied.³⁵ Furthermore, the introduction of electron-donating groups at the *meso*-site of the BODIPY platform is also an emerging strategy for achieving efficient PSs.^{36,37} The abovementioned factors inspired us to consider designing PSs with different functional substituents incorporated at the *meso*-position of the thiophene-fused

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BODIPY to achieve compounds suitable for clinical use. To the best of our knowledge, there have been no attempts to design efficient theranostic agents by taking advantages of the above two strategies.

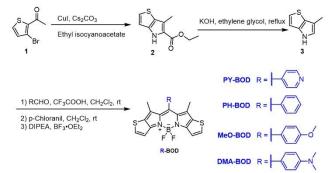
Herein, we designed and synthesized a series of heavy-atomfree thiophene-BODIPY derivatives by incorporating different functional groups (pyridinyl (**PY**), phenyl (**PH**), p-methoxyphenyl (**MeO**) and N,N-dimethylaminophenyl (**DMA**)) into the *meso*position of the thiophene-BODIPY platform (Scheme 1). The experimental results revealed that the ¹O₂ generation ability of these thiophene-BODIPY derivatives through type II process gradually increased with increasing electron-donating ability of the substituent (Φ_{Δ} : **PY-BOD** < **PH-BOD** < **MeO-BOD**). Fortunately, **MeO-BOD** had prominent dual functions, showing in both high ¹O₂ quantum yield and moderate fluorescence intensity. Furthermore, the cellular experimental results showed that **MeO-BOD** could be utilized as a mitochondriaspecific diagnostic agent to reinforce the PDT effect.

Results and discussion

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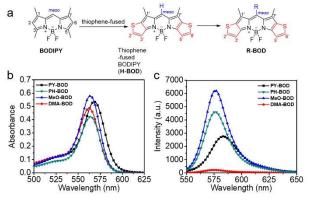
Synthesis and photophysical properties

The different thiophene-fused BODIPY derivatives were synthesized through the trifluoroacetic acid-catalyzed condensation of thiophene-fused pyrroles with different benzaldehydes, followed by oxidation and complexation (Scheme 1). Detailed experimental procedures are provided in the Supporting Information. The structures of all these thiophene-fused BODIPY derivatives were fully confirmed by ¹H and ¹³C NMR spectroscopy and high-resolution mass spectrometry (Figs. S1-S14). UV-vis absorption and fluorescence spectra of the new thiophene-BODIPY derivatives were acquired (Figs. 1, S15 and S16). These key photophysical property data are summarized in Table 1.



Scheme 1. Synthetic route to the thiophene-fused BODIPY derivatives.

The absorption and fluorescence spectra of all the thiophenefused BODIPY derivatives were similar in shape to those of **H**-**BOD**, indicating that the R substituents had little effect on the HOMO–LUMO energy gap of the thiophene-fused BODIPY chromophore because they are nearly orthogonal to the thiophene-fused BODIPY core (Fig. 1a).³² All the studied compounds retained high absorption molar extinction coefficients (ε~100,000 M⁻¹ cm⁻¹), and the maximum absorptions were shifted to longer wavetlengths (λ²/₂₀₅°565¹Am) compared with that of unmodified **BODIPY**. The absorption spectra of all the thiophene-BODIPY derivatives showed a slight blueshift with increasing solvent polarity. This phenomenon indicates that the dipole moments of the ground state of these thiophene-BODIPY derivatives might be larger than those of the first singlet excited states.³² On the other hand, **PY-BOD**, **PH-BOD** and **MeO-BOD** was poorly emissive due to a PET mechanism.^{38,39} The fluorescence quantum yields of all the thiophene-BODIPY derivatives were evaluated in toluene (Table 1), and **PY-BOD**, **PH-BOD** and **MeO-BOD** were shown to be



excellent candidates for fluorescence imaging.

Fig. 1 (a) Design of thiophene-fused BOIDPY derivatives, (b) UV-Vis absorption and (c) emission spectra of **PY-BOD**, **PH-BOD**, **MeO-BOD** and **DMA-BOD** in toluene at c = 5.0 μ M and λ_{ex} = 540 nm.

¹O₂ generation ability

Subsequently, the singlet oxygen generation capability of PY-BOD, PH-BOD, MeO-BOD and DMA-BOD was assessed in airsaturated acetonitrile (ACN) under 560 nm irradiation. A commercial ¹O₂ probe, 1,3-diphenylisobenzofuran (DPBF), was used as an indicator, and rose bengal (RB, Φ_{Δ} = 0.54 in ACN) was used as the reference.⁴⁰ As shown in Figs. 2a and S17, the absorbance at 410 nm of DPBF decreased gradually in the presence of the thiophene-BODIPY derivatives under continuous light irradiation. According to the linear relationship of the decay curves (Fig. 2b), the ¹O₂ quantum yields of **PY-BOD**, PH-BOD, MeO-BOD and DMA-BOD were calculated to be 0.52, 0.69, 0.85 and 0.04, respectively (Table 1). The strongest ¹O₂ generation ability of MeO-BOD among all the R-BODs suggested the electron donor group (MeOPh-) in the thiophene-fused BODIPY derivatives played an important role for enhancing the ¹O₂ generation. Based on these results, we propose a plausible mechanism for MeO-BOD. The introduction of the donating group probably favors ¹O₂ generation first by increasing the formation of the charge transfer (CT) state via photoinduced charge transfer (PCT) (¹BOD-Donor \rightarrow BOD^[\delta-]-Donor^[\delta+]). The charge recombination of CT state further triggers the production of T₁ of the thiophene-fused BODIPY derivatives

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ble 1 . Photophysical Properties of the synthesized compounds.					DOI: 10.1039/D0SC011714	
Compounds	$\lambda_{abs}[nm]^{a}$	ε×10 ⁻³ [M ⁻¹ cm ⁻¹] ^a	$\lambda_{em}[nm]^a$	Stokes shift[nm]	⊅ ƒ(%) ^ь	${\boldsymbol{\varPhi}}_{\!\scriptscriptstyle \Delta}{}^{c}$
BODIPY ^d	502	120.00	508	6	70.0	≈0
PY-BOD	568	107.09	585	17	9.82	0.52
PH-BOD	564	111.60	576	12	18.83	0.69
MeO-BOD	564	115.78	579	15	17.11	0.85
DMA-BOD	562	96.99	577	15	2.50	0.04

^{*a*}In toluene (5.0 × 10⁻⁶ M), ^{*b*}fluorescence quantum yield estimated relative to rhodamine 101 as the standard (Φ_f = 1.0 in methanol), ^{*c*}singlet oxygen quantum yield was determined with respect to rose bengal (Φ_{Δ} (RB) = 0.54 in ACN), ^{*d*}Literature value.¹⁹

 $(BOD^{[\delta-]}-Donor^{[\delta+]} \rightarrow {}^{3}BOD$ -Donor). Finally, ${}^{1}O_{2}$ is generated by energy transfer from T₁ to molecular oxygen (${}^{3}BOD$ -Donor + O₂ \rightarrow BOD-Donor + ${}^{1}O_{2}$) (Scheme S1).^{29-31,37,41,42} Unfortunately, DMA-BOD showed almost no ${}^{1}O_{2}$ generating ability in ACN, which implied that the electron-donating group could also lead to a PET process, which triggered the quenching of fluorescence and forbidding of the non-radiative transitions such as ISC, therefore, prohibiting the ${}^{1}O_{2}$ generation.⁴³ Thus, the ${}^{1}O_{2}$ formation efficiency could be controlled by finely tuning the electronic properties of the *meso*-substituent on the thiophene-BODIPY platform, especially by the introduction of a suitable electron-donating group. Taken together, these results indicate that **MeO-BOD** has potential as a theranostic agent for cancer treatment.

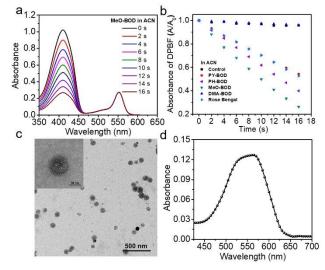


Fig. 2 (a) Time-dependent photodegradation of DPBF with MeO-BOD; (b) the DPBF degradation rate curves with PY-BOD, PH-BOD, MeO-BOD, DMA-BOD and rose bengal in ACN; (c) a TEM image of MeO-BOD NPs (inset: high magnification TEM image); and (d) UV-Vis absorption spectrum of MeO-BOD (5 μ M) in deionized water.

Cellular fluorescence imaging and subcellular colocalization

Given the above inspiring results, we attempted to investigate the fluorescence (FL) imaging and PDT efficiency of **MeO-BOD** in living cells. To this end, **MeO-BOD** nanoparticles (NPs) were

prepared by dropping the DMSO stock solution of MeO-BOD into water.44 The transmission electron microscopy (TEM) image in Fig. 2c indicated that the MeO-BOD NPs had a regular spherical morphology with a diameter of approximately 72 nm. The size distribution of the nanoparticles was determined using dynamic light scattering (DLS), which showed that the average size was 68 ± 7 nm (Fig. S18). These particles are suitably sized for passive targeting through the enhanced permeability and retention (EPR) effect.⁴⁵ In addition, the evident decrease in absorbance and broad absorption spectrum, and the redshift of the fluorescence spectrum of the MeO-BOD in deionized water (DW) suggested that the formation of MeO-BOD NPs might be due to J-aggregation (Fig. 2d and S19b).46 The MeO-BOD NPs could be disassembled and the instinctive fluorescent peak of MeO-BOD could restore in the presence of FBS (10%) in DW (Fig. S19).

Then, we further explored the cellular uptake of MeO-BOD in HeLa cells by using confocal laser scanning microscopy. As illustrated in Fig. 3a, the MeO-BOD could be rapidly internalized by living cells, and the FL images showed strong emission in the cell cytoplasm. Therefore, the MeO-BOD could be employed as an imaging-guided PDT agent. Furthermore, to test the main organelle locations of the MeO-BOD, we co-stained HeLa cells with MeO-BOD and commercial MitoTracker Green (MTG) or LysoTracker Green (LTG). The colocalization experiments indicated that MeO-BOD were mainly localized in mitochondria, as indicated by the high Pearson's coefficient (0.98), instead of the lysosomes (Pearson's coefficient 0.62) (Figs. 3b-e). In addition, the time-dependent colocalization fluorescence imaging of MeO-BOD with MTG in HeLa cells was performed. As shown in Fig. S20, MeO-BOD almost internalize into the mitochondria of HeLa cells after 1 h. The recent reports have demonstrated that the BODIPY dye itself could localize into the mitochondria of cells due to its low electron density character.47 The unmodified **BODIPY** platform has a low electron density $(+\delta)$ character. Also, the incorporation of sulfur atoms into the π conjugated skeleton of BODIPY perhaps led to more reduction of electron density of BODIPY core,48 which might facilitate the mitochondrial accumulation of MeO-BOD. Subcellular organelles are indispensable in maintaining cellular biological function.⁴⁹ In particular, the generation of ¹O₂ in mitochondria can induce direct dysfunction and trigger cell apoptosis. Thus,

mitochondria-targeted theranostic agents could maximize cancer treatment efficiency.

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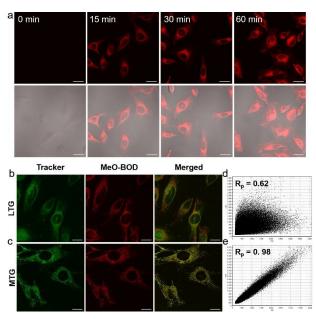


Fig. 3 (a) Confocal laser scanning microscopy images of HeLa cells incubated with **MeO-BOD** (1.0 μ M) for different times (0, 15, 30 and 60 min), scale bar: 30 μ m. Confocal laser scanning microscopy colocalization fluorescence images of **MeO-BOD** (1.0 μ M) with (b) LTG and (c) MTG (500 nM) in HeLa cells, respectively. The fluorescence intensity correlation **MeO-BOD** with (d) LTG and (e) MTG, respectively. R_p is Pearson's coefficient, scale bar: 20 μ m.

In vitro PDT efficacy evaluation

To demonstrate the PDT efficacy of MeO-BOD in living cells, we first tested its cytotoxicity by the methyl thiazolyltetrazolium (MTT) assay in HeLa cells. As indicated in Fig. 4a, the MeO-BOD had negligible cytotoxicity in the dark, revealing its excellent biocompatibility in vitro. Under 560 nm light irradiation (0.1 W/cm², 5 or 10 min), the viability of HeLa cells gradually decreased with the increasing concentration of MeO-BOD, and the growth inhibition ratio reached ~88% even at a very low concentration of 0.15 μ M (Fig. 4b). The half-maximal inhibitory concentration (IC₅₀) of MeO-BOD for HeLa cells was as low as 95 nM under 560 nm light irradiation (0.1 W/cm², 10 min). The extremely low IC₅₀ of MeO-BOD could be attributed to the high ¹O₂ quantum yield and efficient mitochondria-specific ROS generation upon light irradiation. To clarify the cytotoxicity in the PDT process, we tracked the morphological variations of HeLa cells in the presence of MeO-BOD under 559 nm laser irradiation using confocal laser scanning microscopy. As described in Fig. S21, with increasing irradiation time (0-10 min), the morphology of the HeLa cells preincubated with MeO-BOD obviously changed; gradual thinning of the cell membrane and the formation of numerous blebs (red line) were observed. In contrast, the cells not exposed to the MeO-BOD exhibited no appreciable morphological changes under the same laser irradiation.

Additionally, to intuitively establish the PDT efficacy of MeOe BOD (0.5 and 1.0 μ M), live/dead cell co-staining of MeOe cells with calcein AM (green, live cells) and PI (red, dead cells) was performed to visualize the cell viability (Fig. 4c). The control group showed strong green fluorescence for live cells, which also verified the low dark toxicity and good biocompatibility of the MeO-BOD. In comparison, almost all the HeLa cells treated with MeO-BOD were killed, and intense red fluorescence was observed. All these results confirmed that MeO-BOD could be employed as a theranostic agent for cancer treatment in vitro.

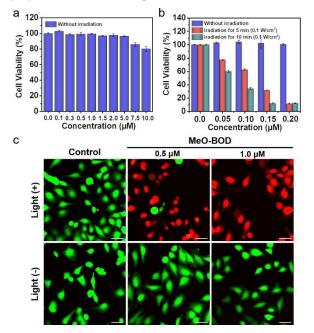


Fig. 4 Concentration–dependent changes in cell viability of HeLa cells treated with **MeO-BOD** using a typical MTT assay (a) in the dark and (b) under light irradiation. Cells were irradiated with 560 nm light (0.1 W/cm², 5 and 10 min). (c) Fluorescence images of calcein AM/PI-stained HeLa cells after preincubation with **MeO-BOD** (0.5 and 1.0 μ M) for 1 h and irradiation with 560 nm light (0.1 W/cm², 10 min). Scale bar: 40 μ m.

Apoptosis mechanism of MeO-BOD-mediated PDT

Finally, to elucidate the potential therapeutic mechanism at the cellular level, we further evaluated the cellular ¹O₂ generation capability of MeO-BOD by using 2,7-dichlorodihydrofluorescein diacetate (DCFH-DA) as the ¹O₂ indicator in HeLa cells (Fig. 5a). The HeLa cells treated with only DCFH-DA and only MeO-BOD showed almost no fluorescence, whereas the group pretreated with MeO-BOD before light irradiation and then incubated with DCFH-DA showed apparent green fluorescence from oxidized 2,7-dichlorofluorescein (DCF). These experimental results validated the ¹O₂ generation ability of MeO-BOD in living cells. As mentioned above, the production of ¹O₂ in mitochondria causes mitochondrial destruction, resulting in cell apoptosis. The reduction of mitochondrial membrane potential (MMP) is a crucial signal of mitochondrial damage. Hence, the MMP changes were monitored by confocal fluorescence images using JC-1 dye, as its fluorescence color changed between its aggregates (red, high MMP) and monomers (green, low MMP).

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The untreated control group under 560 nm light irradiation (0.1 W/cm^2 for 10 min) displayed strong red emission and weak green emission, indicating that the cells were healthy with a high MMP. By comparison, the cells being treated with **MeO-BOD** suffered from depolarization of the mitochondrial membrane, as demonstrated by the increase of green fluorescence intensity (Fig. 5b). These observations further suggested that the cell death was induced by a mitochondria-associated pathway.

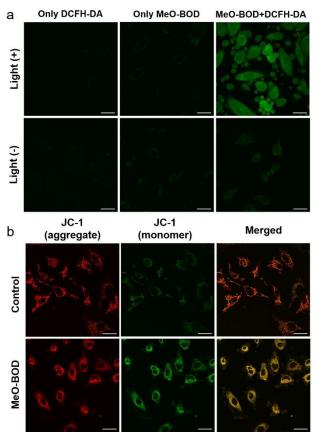


Fig. 5 (a) Fluorescence images of ROS generation in HeLa cells after incubation with **MeO-BOD** (1.0 μ M) for 1 h using DCFH-DA as an indicator (10 μ M). The green fluorescence indicates DCFH-DA is oxidized to DCF. (b) Fluorescence images of mitochondrial membrane potential of HeLa cells after incubation without or with **MeO-BOD** (1.0 μ M) using JC-1 as the indicator (2 μ M, λ_{ex} = 473 nm, red channel for aggregates (healthy cells): 575-675 nm, green channel for monomers (apoptotic cells): 490-540 nm. Samples were irradiated with 560 nm light (0.1 W/cm², 10 min). Scale bar: 30 μ m.

Conclusions

We synthesized thiophene-fused BODIPY analogues with different functional groups at the *meso*-position of the BODIPY core as mitochondria-targeted theranostic agents for imaging and PDT. The varying R-substituents had little effect on the absorption and emission maxima. **PY-BOD**, **PH-BOD** and **MeO-BOD** exhibited moderate fluorescence quantum yields, while that of **DMA-BOD** was much lower due to a PET process. Among

the analogues, **MeO-BOD** exhibited excellent dual functionality both considerable ${}^{1}O_{2}$ generation ability and high brighthess. The cell experiments manifested that the **MeO-BOD** offered many advantages: good biocompatibility, mitochondria-specific fluorescence imaging, and a very low IC₅₀ value (\approx 95 nM). Therefore, the **MeO-BOD** could be employed as imaging-guided PDT agents for cancer treatment. In addition, the apoptosis mechanism of light-induced PDT might be a result of ROSinduced damage to mitochondria, which was demonstrated by detecting changes of the mitochondrial membrane potential. Briefly, finely tuning the electron structure of the substituent at the *meso*-site of heavy-atom-free thiophene-fused BODIPY core is a promising strategy for developing highly efficient theranostic agents with minimal side effects to accomplish the integration of diagnosis and therapy.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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Notes and references

- 1 X. Li, S. Lee and J. Yoon, Chem. Soc. Rev., 2018, 47, 1174-1188.
- P. Agostinis, K. Berg, K. A. Cengel, T. H. Foster, A. W. Girotti, S. O. Gollnick, S. M. Hahn, M. R. Hamblin, A. Juzeniene, D. Kessel, M. Korbelik, J. Moan, P. Mroz, D. Nowis, J. Piette, B. C. Wilson and J. Golab, *Ca-Cancer J. Clin.*, 2011, **61**, 250-281.
- 3 D. E. J. G. J. Dolmans, D. Fukumura and R. K. Jain, *Nat. Rev. Cancer*, 2003, **3**, 380-387.
- 4 J. Zhang, L. Ning, J. Huang, C. Zhang and K. Pu, *Chem. Sci.*, 2020, **11**, 618-630.
- 5 J. Dai, Y. Li, Z. Long, R. Jiang, Z. Zhuang, Z. Wang, Z. Zhao, X. Lou, F. Xia and B. Z. Tang, *ACS Nano*, 2020, **14**, 854-866.
- 6 Y. Yuan, C.-J. Zhang, M. Gao, R. Zhang, B. Z. Tang and B. Liu, *Angew. Chem. Int. Ed.*, 2015, **54**, 1780-1786.
- 7 Y. Cai, W. Si, W. Huang, P. Chen, J. Shao and X. Dong, Small, 2018, 14, 1704247.
- 8 R. Kumar, W. S. Shin, K. Sunwoo, W. Y. Kim, S. Koo, S. Bhuniya and J. S. Kim, *Chem. Soc. Rev.*, 2015, **44**, 6670-6683.
- 9 X. Yi, F. Wang, W. Qin, X. Yang and J. Yuan, *Int. J. Nanomedicine*, 2014, **9**, 1347-1365.
- 10 X. Li, C. y. Kim, S. Lee, D. Lee, H.-M. Chung, G. Kim, S.-H. Heo, C. Kim, K.-S. Hong and J. Yoon, *J. Am. Chem. Soc.*, 2017, **139**, 10880-10886.
- 11 M. Li, S. Long, Y. Kang, L. Guo, J. Wang, J. Fan, J. Du and X. Peng, *J. Am. Chem. Soc.*, 2018, **140**, 15820-15826.
- 12 R. Wang, K. Dong, G. Xu, B. Shi, T. Zhu, P. Shi, Z. Guo, W.-H. Zhu and C. Zhao, *Chem. Sci.*, 2019, **10**, 2785-2790.
- 13 H. Wang, W. Zhao, X. Liu, S. Wang and Y. Wang, ACS Appl. Bio Mater., 2020, 3, 593-601.
- 14 J. Zhao, W. Wu, J. Sun and S. Guo, *Chem. Soc. Rev.*, 2013, **42**, 5323-5351.
- 15 S. G. Awuah and Y. You, RSC Advances, 2012, 2, 11169-11183.

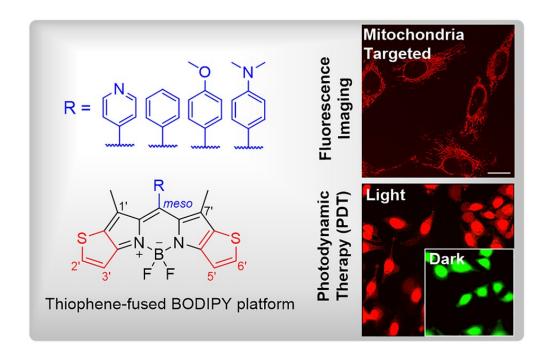
16 A. Turksoy, D. Yildiz and E. U. Akkaya, *Coord. Chem. Rev.*, 2019, **379**, 47-64.

ARTICLE

- 17 W. Sun, X. Zhao, J. Fan, J. Du and X. Peng, Small, 2019, 15, 1804927.
- 18 A. Gorman, J. Killoran, C. O'Shea, T. Kenna, W. M. Gallagher and D. F. O'Shea, J. Am. Chem. Soc., 2004, **126**, 10619-10631.
- 19 T. Yogo, Y. Urano, Y. Ishitsuka, F. Maniwa and T. Nagano, *J. Am. Chem. Soc.*, 2005, **127**, 12162-12163.
- 20 J. Tian, J. Zhou, Z. Shen, L. Ding, J.-S. Yu and H. Ju, *Chem. Sci.*, 2015, **6**, 5969-5977.
- 21 Q.-J. Hu, Y.-C. Lu, C.-X. Yang and X.-P. Yan, *Chem. Commun.*, 2016, **52**, 5470-5473.
- 22 L. Huang, Z. Li, Y. Zhao, Y. Zhang, S. Wu, J. Zhao and G. Han, J. Am. Chem. Soc., 2016, **138**, 14586-14591.
- 23 H. S. Jung, J. Han, H. Shi, S. Koo, H. Singh, H.-J. Kim, J. L. Sessler, J. Y. Lee, J.-H. Kim and J. S. Kim, *J. Am. Chem. Soc.*, 2017, **139**, 7595-7602.
- X. Miao, W. Hu, T. He, H. Tao, Q. Wang, R. Chen, L. Jin, H. Zhao,
 X. Lu, Q. Fan and W. Huang, *Chem. Sci.*, 2019, **10**, 3096-3102.
- 25 S. Xu, W. Wu, X. Cai, C.-J. Zhang, Y. Yuan, J. Liang, G. Feng, P. Manghnani and B. Liu, *Chem. Commun.*, 2017, **53**, 8727-8730.
- 26 Y. Cakmak, S. Kolemen, S. Duman, Y. Dede, Y. Dolen, B. Kilic, Z. Kostereli, L. T. Yildirim, A. L. Dogan, D. Guc and E. U. Akkaya, *Angew. Chem. Int. Ed.*, 2011, **50**, 11937-11941.
- 27 S. Kolemen, M. Işık, G. M. Kim, D. Kim, H. Geng, M. Buyuktemiz, T. Karatas, X.-F. Zhang, Y. Dede, J. Yoon and E. U. Akkaya, Angew. Chem. Int. Ed., 2015, 54, 5340-5344.
- 28 M. Üçüncü, E. Karakuş, E. Kurulgan Demirci, M. Sayar, S. Dartar and M. Emrullahoğlu, Org. Lett., 2017, 19, 2522-2525.
- 29 J. T. Buck, A. M. Boudreau, A. DeCarmine, R. W. Wilson, J. Hampsey and T. Mani, *Chem*, 2019, **5**, 138-155.
- M. A. Filatov, S. Karuthedath, P. M. Polestshuk, H. Savoie, K. J. Flanagan, C. Sy, E. Sitte, M. Telitchko, F. Laquai, R. W. Boyle and M. O. Senge, *J. Am. Chem. Soc.*, 2017, **139**, 6282-6285.
- 31 X.-F. Zhang, N. Feng, *Chem. Asian J.* 2017, **12**, 2447-2456.
- 32 S. Ji, J. Ge, D. Escudero, Z. Wang, J. Zhao and D. Jacquemin, *J. Org. Chem.*, 2015, **80**, 5958-5963.
- 33 S. G. Awuah, S. K. Das, F. D'Souza and Y. You, *Chem. Asian J.*, 2013, 8, 3123-3132.
- 34 K. Tanaka, H. Yamane, R. Yoshii and Y. Chujo, *Bioorg. Med. Chem.*, 2013, **21**, 2715-2719.
- 35 R. L. Watley, S. G. Awuah, M. Bio, R. Cantu, H. B. Gobeze, V. N. Nesterov, S. K. Das, F. D'Souza and Y. You, *Chem. Asian J.*, 2015, **10**, 1335-1343.
- 36 X. Xia and Y. Qian, Analyst, 2018, **143**, 5218–5224.
- 37 C. Wang and Y. Qian, Biomater. Sci., 2020, 8, 830-836.
- 38 W. Lin, W. Zhang, S. Liu, Z. Li, X. Hu, Z. Xie, C. Duan and G. Han, ACS Appl. Mater. Interfaces, 2019, 11, 43928-43935.
- 39 X. Liu, Q. Qiao, W. Tian, W. Liu, J. Chen, M. J. Lang and Z. Xu, J. Am. Chem. Soc., 2016, **138**, 6960-6963.
- 40 V.-N. Nguyen, S. Qi, S. Kim, N. Kwon, G. Kim, Y. Yim, S. Park and J. Yoon, *J. Am. Chem. Soc.*, 2019, **141**, 16243-16248.
- 41 V. N. Nguyen, Y. Yim, S. Kim, B. Ryu, K. M. K. Swamy, G. Kim, N. Kwon, C. Y. Kim, S. Park and J. Yoon, *Angew. Chem. Int. Ed.*, 2020, DOI: 10.1002/anie.202002843.
- 42 W. Hu, Y. Lin, X.-F. Zhang, M. Feng, S. Zhao and J. Zhang, *Dyes Pigm.*, 2019, **164**, 139-147.
- Y. Liu, C. Xu, L. Teng, H.-W. Liu, T.-B. Ren, S. Xu, X. Lou, H.Guo, L. Yuan and X.-B. Zhang, *Chem. Commun.*, 2020, **56**, 1956-1959.
- 44 17. X. Li, D. Lee, J.-D. Huang and J. Yoon, Angew. Chem. Int. Ed., 2018, **57**, 9885-9890.
- 45 N. Kamaly, Z. Xiao, P. M. Valencia, A. F. Radovic-Moreno and O. C. Farokhzad, *Chem. Soc. Rev.*, 2012, **41**, 2971-3010.
- 46 C. Duan, Y. Zhou, G.-G. Shan, Y. Chen, W. Zhao, D. Yuan, L. Zeng, X. Huang and G. Niu, J. Mater. Chem. C, 2019, 7, 3471-3478.

- 47 T. Gayathri, S. Karnewar, S. Kotamraju and S. P. Singh, ACS Med. Chem. Lett., 2018, 9, 618-622. DOI: 10.1039/D0SC01171A
- 48 K. Tanaka, H. Yamane, R. Yoshii and Y. Chujo, Bioorg. Med. Chem. 2013, **21**, 2715–2719
- 49 P. Gao, W. Pan, N. Li and B. Tang, ACS Appl. Mater. Interfaces, 2019, **11**, 26529-26558.

6 | J. Name., 2012, **00**, 1-3



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