Facile Synthesis of 2-Aminocyclobutenylphosphonates

A. A. A. Al Quntar^{*a*, *b*,*}, H. Dweik^{*b*}, and V. Dembitsky^{*c*}

 ^a Department of Material Engineering, Faculty of Engineering, Al-Quds University, Jerusalem, 90612 Palestine
^b Faculty of Chemistry and Chemical Technology, Al-Quds University, Jerusalem, 90612 Palestine
^c Centre for Applied Research, Innovation and Entrepreneurship, Lethbridge College, South Lethbridge, Canada AB T1K 1L6
*e-mail: abedalaziz@staff.alguds.edu

Received August 21, 2019; revised November 17, 2019; accepted November 22, 2019

Abstract—The addition of various amines to diethyl 4-chlorobut-1-yn-1-ylphosphonate produced novel biologically potent substituted diethyl 2-aminocyclobut-1-en-1-ylphosphonates in 70–83% isolated yield. This regioselective reaction was carried out at room temperature in the absence of solvent and catalyst.

Keywords: vinylphosphonates, cyclobutenes, β-aminophosphonates, amine addition.

DOI: 10.1134/S1070428020010212

Cyclobutene derivatives are of particular interest as organic units not only due to their intriguing molecular skeleton suggesting specific properties such as a high ring strain and consequent enhanced electrophilicity but also because of their reactivities and importance as intermediates in organic synthesis [1–3]. They undergo electrocyclic ring-opening reaction to produce highly electron-deficient 1,3-dienes [4], thermal aromatization [5], and other transformations [6–18]. In addition, cyclobutene derivatives have wide utility as biologically active compounds [19, 20]. For instance, they possess anti-inflammatory [21–23], herbicidal [24], and antitumor activity [25]. Moreover, they showed protective properties against UV radiation [26], and other useful properties [27].

During the last few decades several methods have been developed for the preparation of polysubstituted cyclobutenes. For example, they can be obtained by photochemical [2+2]-cycloaddition of olefines and acetylenes to cyclic enones and unsaturated lactones [28–32], titanium-mediated intramolecular cyclization of bis-propargyl alcohols [33], reaction of dilithiated benzylacetylene with isothiocyanates [34], metallocupration of allenes and acetylenes [35], zirconiummediated inter- or intramolecular cyclodimerization of alkynes [36], gold-catalyzed intermolecular reaction of terminal alkynes with alkenes [37], cycloalkylation reactions involving (4-halo-1-alkenyl)metals [38], intramolecular Wittig reaction of vinyl(triphenyl)phosphonium salt with dioxobutanoates [4], thermolysis– oxidation [39], and other reactions [40, 41].

Interestingly, the presence of a phosphonate group on the cyclobutene ring enhances its biological activity [42]. Despite that, not many methods have been yet reported for the synthesis of cyclobutenylphosphonates. Generally, they are produced from readily available halogenated cyclobutenes via Arbuzov reaction [43, 44] or by cyclizations of diazophosphonate intermediates [45–48].

We previously synthesized cyclobutenylphosphonates by zirconation of 1-alkynylphosphonates, followed by treatment with copper(I) chloride (Scheme 1) [49]. In addition, cyclization reactions of ω -chloro-1alkynylphosphonates with amines were also found to produce cyclic β -aminophosphonates as shown in Scheme 2 [50].

Being encouraged by these results, we focused on the reaction of amines with an alkynylphosphonate having a shorter alkynyl chain, in particular, with



Scheme 1.





7a

diethyl (4-chlorobut-1-yn-1-yl)phosphonate (6) which was successfully prepared by substituting the hydroxy group in but-3-yn-1-ol using thionyl chloride under reflux. The product was isolated by distillation, dissolved in diethyl ether, lithiated with *n*-BuLi, and reacted with diethyl chlorophosphate (Scheme 3).

We have developed an effective and facile method for the synthesis of novel cyclobutenes. 2-Aminocyclo-

Table 1. Synthesis	of diethyl 2-a	minocyclobutenylphospho-
nates 7 a –7i	NR^1R^2	

P(O)(OEt) ₂				
7a–7i				
Compound no.	Amine	Yield, ^a %		
7a	Propylamine	83		
7b	Isopropylamine	81		
7c	tert-Butylamine	80		
7d	Benzylamine	75		
7e	Aniline	73		
7f	Methylamine ^b	80		
7g	2-Phenylethylamine	78		
7h	2-Aminoethanol	77		
7i	Diisopropylamine	70		

^a Isolated vield (after silica gel chromatography); in all cases, the conversion was higher than 98% according to the GC/MS and ³¹P NMR data.

^b Methylamine was used as a 40% aqueous solution.

butenylphosphonates were obtained by the addition of various amines to diethyl 4-chlorobut-1-yn-1-ylphosphonate (6). When the latter was allowed to react with propylamine at 25°C for 15 h, diethyl [2-(propylamino)cyclobut-1-en-1-yl]phosphonate (7a) was formed as the only product in 83% yield (Scheme 4). Likewise, other primary and secondary amines were reacted with phosphonate 6 under similar conditions to obtain 2-aminocyclobutenylphosphonates 7b-7i which were isolated by silica gel column chromatography in good yields (70-83%; Table 1) and were characterized by NMR, GC/MS, and elemental analyses. The ¹H NMR spectra of **7a**–**7i** showed two broad triplets in the region δ 2.18–2.93 ppm along with vinylic carbon signals at $\delta_{\rm C}$ 68.9–217.2 ppm in the ¹³C NMR spectra. The phosphorus atom of 7 resonated in ³¹P NMR spectra at $\delta_P \sim 30$ ppm.

8a

This process represents a general and facile one-pot method for the synthesis of novel oily 2-aminocyclobutenylphosphonates 7. Compounds 7a-7i are stable on exposure to air at room temperature and are soluble in most organic solvents. Apart from primary and secondary amines, the described cyclization is also tolerant to alkyl (7a-7c, 7f, 7i), aryl (7d, 7e, 7g), and hydroxy groups (7h) as shown in Table 1.

Compound 7a was smoothly hydrogenated over Pd/C as a catalyst to afford cyclobutane derivative 8a in a high yield. On the basis of our observations and



previous theoretical calculations, we presumed that the reaction proceeds by a stepwise mechanism where the first step involves initial addition of the amine to the triple bond to give zwitterionic intermediate, and the next steps involve cyclization followed by elimination of proton, leading to 2-aminocyclobut-1-en-1-ylphos-phonates 7 [50] (Scheme 5).

In summary, novel biologically potent diethyl 2-aminocyclobut-1-en-1-ylphosphonates 7a-7i were smoothly prepared by addition of amines to diethyl 4-chlorobut-1-yn-1-ylphosphonate. The reaction is facile, general, and selective, it requires neither solvent nor catalyst, and the products are formed in high yield. This makes the described reaction a convenient method for the synthesis of 2-aminocyclobutenylphosphonates with the phosphorus and nitrogen substituents linked to the neighboring double-bonded carbon atoms (β -aminophosphonates).

EXPERIMENTAL

The ¹H, ¹³C, and ³¹P NMR spectra were recorded from solutions in CDCl₃ on a Varian Mercury 300 spectrometer at 300, 75.5, and 121.4 MHz, respectively; the chemical shifts were measured relative to TMS (¹H, ¹³C) and H₃PO₄. The mass spectra (EI) were recorded on an HP G1800A GCD GC/MS instrument using a 30-m methyl silicone column.

Diethyl (4-chlorobut-1-yn-1-yl)phosphonate (6) was prepared in our lab by refluxing but-3-yn-1-ol with thionyl chloride. The product, 4-chlorobut-1-yne, was isolated by distillation and dissolved in diethyl ether, the solution was cooled to -78° C, an equivalent amount of butyllithium was added, and the mixture was allowed to gradually warm up to room temperature. The mixture was then cooled again to -78° C, and an equivalent amount of diethyl phosphorochloridate was added. The reaction was quenched with 1 M aqueous HCl, and the product was extracted into diethyl ether, followed by evaporation of the solvent on a rotary evaporator and distillation of the residue under reduced pressure.

Diethyl [2-(propan-2-ylamino)cyclobut-1-en-1yl]phosphonate (7a). Diethyl (4-chlorobut-1-yn-1-yl)-

phosphonate (0.22 g, 1 mmol) was mixed with isopropylamine (0.07 g, 1.1 mmol) in a 10-mL roundbottom flask, and the mixture was stirred at 25°C for 15 h. The mixture was then washed with 0.1 N aqueous sodium hydroxide and extracted with methylene chloride (2×20 mL), the extract was dried over MgSO₄ and concentrated on a rotary evaporator, and the residue was purified by silica gel column chromatography (CH₂Cl₂-MeOH, 9:1). ¹H NMR spectrum, δ, ppm: 1.12 d [6H, $J_{\rm HH}$ = 6.3 Hz, CH(CH₃)₂], 1.26 t (6H, $J_{\rm HH} = 6.9$ Hz, OCH₂CH₃), 2.63 br.t (2H, $J_{\rm HH} = 6.9$ Hz, CH_2CH_2), 2.86 br.t (2H, J_{HH} = 7.2 Hz, CH_2CH_2), 2.92 m [1H, CH(CH₃)₂], 4.11 m (4H, OCH₂CH₃). ^{31}P NMR spectrum: δ_P 30.38 ppm. ^{13}C NMR spectrum, $δ_{\rm C}$, ppm: 16.2 d (${}^{3}J_{\rm PC}$ = 6.2 Hz, OCH₂CH₃), 21.4 s $[(CH_3)_2C)]$ 23.7 s (CH₂CP), 32.2 d (³J_{PC} = 5.8 Hz, CH₂CN), 42.8 s [CH(CH₃)₂], 48.8 d ($^{2}J_{PC}$ = 30.2 Hz, OCH_2CH_3), 72.9 d (${}^1J_{PC}$ = 201.6 Hz, CP), 172.8 d $({}^{2}J_{PC} = 20.1$ Hz, CNH). Mass spectrum, m/z (I_{rel} , %): 247 (20.5), 232 (14.8), 218 (31.0), 207 (1.1), 155 (100), 144 (6.0), 104 (10.3), 91 (80.1), 77 (1.2), 65 (21.3), 41 (10.3). Found, %: C 53.58; H 9.11; N 5.50; P 12.38. C₁₁H₂₂NO₃P. Calculated, %: C 53.43; H 8.97; N 5.66; P 12.53.

Compounds **7b**–**7i** were synthesized in a similar way from compound **6** and the corresponding amine.

Diethyl [2-(propylamino)cyclobut-1-en-1-yl]**phosphonate (7b).** ¹H NMR spectrum, δ , ppm: 0.84 t $(3H, J_{HH} = 8.4 \text{ Hz}, CH_2CH_2CH_3), 1.26 \text{ t} (6H, J_{HH} =$ 7.5 Hz, OCH₂CH₃), 2.05 m (2H, CH₂CH₃), 2.40 br.t and 2.61 br.t (2H each, $J_{\rm HH}$ = 7.5 Hz, CH₂CH₂), 3.06 t $(2H, J_{HH} = 6.5 \text{ Hz}, \text{NHCH}_2), 4.00 \text{ m} (4H, \text{OCH}_2\text{CH}_3).$ 31 P NMR spectrum: δ_P 28.83 ppm. 13 C NMR spectrum, $\delta_{\rm C}$, ppm: 11.44 s (CH₂CH₃), 15.9 d (³J_{PC} = 6.6 Hz, OCH₂CH₃), 20.3 s (CH₂CH₃), 20.5 s (CH₂CH₂), 43.9 s (CH₂N), 44.8 d (${}^{3}J_{PC}$ = 33.5 Hz, CH₂CN), 60.2 d $({}^{2}J_{PC} = 5.4 \text{ Hz}, \text{ OCH}_{2}), 73.0 \text{ d} ({}^{1}J_{PC} = 217.2 \text{ Hz}, \text{ CP}),$ 161.3 d (${}^{2}J_{PC}$ = 21.2 Hz, CN). Mass spectrum, m/z(*I*_{rel}, %): 247 (4.2), 232 (10.2), 207 (11.1), 158 (100), 144 (6.0), 133 (21.7), 104 (10.8), 85 (5.7), 43 (35.6). Found, %: C 53.66; H 9.06; N 5.54; P 12.39. C₁₁H₂₂NO₃P. Calculated, %: C 53.43; H 8.97; N 5.66; P 12.53.

Diethyl [2-(tert-butylamino)cyclobut-1-en-1-yl]**phosphonate (7c).** ¹H NMR spectrum, δ , ppm: 1.20 s (9H, *t*-Bu), 1.26 t (6H, $J_{\rm HH}$ = 6.7 Hz, OCH₂CH₃), 2.74 br.t (2H, $J_{\rm HH}$ = 6.5 Hz, CH₂CH₂), 2.93 br.t (2H, $J_{\rm HH} = 6.3$ Hz, CH₂CH₂), 3.96 m (4H, OCH₂CH₃). ³¹P NMR spectrum: δ_P 27.44 ppm. ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 16.3 d (${}^{3}J_{\rm PC} = 6.0$ Hz, OCH₂CH₃), 23.4 s (CH_2CH_2) , 28.9 s $[C(CH_3)]$, 32.4 d $({}^{3}J_{PC} = 5.7 \text{ Hz},$ CH₂CN), 50.4 s [C(CH₃)], 61.3 d (${}^{2}J_{PC}$ = 4.6 Hz, OCH_2CH_3), 72.2 d (${}^1J_{PC}$ = 215.6 Hz, CP), 159.8 d $(^{2}J_{PC} = 19.6 \text{ Hz}, \text{CN})$. Mass spectrum, m/z (I_{rel} , %): 261 (34.5), 246 (66.5), 233 (6.7), 218 (17.6), 206 (50.7), 178 (22.7), 160 (31.7), 132 (35.6), 108 (29.3), 95 (55.9), 69 (100), 57 (29.9). Found, %: C 54.97; H 9.17; N 5.51; P 11.98. C₁₂H₂₄NO₃P. Calculated, %: C 55.16; H 9.26; N 5.36; P 11.85.

Diethyl [2-(benzylamino)cyclobut-1-en-1-yl]**phosphonate (7d).** ¹H NMR spectrum, δ , ppm: 1.26 t $(6H, J_{HH} = 6.2 \text{ Hz}, \text{OCH}_2\text{CH}_3), 2.82 \text{ br.t and } 2.92 \text{ br.t}$ $(2H \text{ each}, J_{HH} = 6.9 \text{ Hz}, CH_2CH_2), 3.89 \text{ s} (2H, CH_2Ph),$ 3.96 m (4H, OCH₂CH₃), 7.27–7.34 m (5H, Ph). ³¹P NMR spectrum: δ_P 28.1 ppm. ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 15.9 d (${}^{3}J_{\rm PC} = 6.2$ Hz, OCH₂CH₃), 21.8 s (CH_2CH_2) , 45.6 s (CH_2Ph) , 47.1 d $({}^3J_{PC} = 31.7 \text{ Hz})$, CH₂CH₂), 60.2 d (${}^{2}J_{PC}$ = 7.8 Hz, OCH₂CH₃), 71.8 d $({}^{1}J_{PC} = 215.6 \text{ Hz}, \text{ CP}), 126.6 (C_{arom}), 127.3 (C_{arom}),$ 128.2 (C_{arom}), 141.3 (C_{arom}), 162.4 d (${}^{2}J_{PC} = 19.2$ Hz, CN). Mass spectrum, *m/z* (*I*_{rel}, %): 295 (3.9), 280 (22.1), 207 (5.3), 208 (100), 155 (3.8), 137 (8.6), 91(70.8), 77 (18.8), 57 (13.5), 41 (15.5). Found, %: C 60.88; H 7.44; N 4.83; P 10.61. C₁₅H₂₂NO₃P. Calculated, %: C 61.01; H 7.51; N 4.74; P 10.49.

Diethyl (2-anilinocyclobut-1-en-1-yl)phospho**nate (7e).** ¹H NMR spectrum, δ , ppm: 1.24 t (6H, $J_{\rm HH} = 7.2$ Hz, OCH₂CH₃), 2.18 br.t (2H, $J_{\rm HH} = 6.9$ Hz, CH_2CH_2), 2.81 br.t (2H, $J_{HH} = 6.5$ Hz, CH_2CH_2), 3.96 m (4H, OCH₂CH₃), 7.17–7.32 (5H, Ph). ³¹P NMR spectrum: δ_P 31.27 ppm. ¹³C NMR spectrum, δ_C , ppm: 16.0 d (${}^{3}J_{PC} = 6.0$ Hz, OCH₂CH₃), 23.4 s (CH₂CH₂), 46.4 d $({}^{3}J_{PC} = 31.7 \text{ Hz}, \text{ CH}_{2}\text{CH}_{2}), 61.3 \text{ d} ({}^{2}J_{PC} =$ 6.2 Hz, OCH₂CH₃), 70.2 d (${}^{1}J_{PC}$ = 205.6 Hz, CP), 126.0 (C_{arom}), 128.3 (C_{arom}), 140.3 (C_{arom}), 160.2 d $({}^{2}J_{PC} = 18.8 \text{ Hz}, \text{ CN}).$ Mass spectrum, m/z (I_{rel} , %): 281 (6.3), 264 (12.4), 207 (1.1), 168 (100), 144 (5.8), 135 (23.7), 91 (78.5), 77 (15.4), 65 (17.7), 41 (18.9). Found, %: C 59.91; H 7.26; N 4.83; P 10.87. C₁₄H₂₀NO₃P. Calculated, %: C 59.78; H 7.17; N 4.98; P 11.01.

Diethyl [2-(methylamino)cyclobut-1-en-1-yl]phosphonate (7f). ¹H NMR spectrum, δ , ppm: 1.25 t (6H, $J_{\text{HH}} = 7.2$ Hz, OCH₂CH₃), 2.52 s (3H, CH₃), 2.70 br.t (2H, $J_{\rm HH}$ = 6.5 Hz, CH₂CH₂), 2.88 br.t (2H, $J_{\rm HH}$ = 6.3 Hz, CH₂CH₂), 4.11 m (4H, OCH₂CH₃). ³¹P NMR spectrum: $\delta_{\rm P}$ 27.85 ppm. ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 15.9 d (${}^{3}J_{\rm PC}$ = 6.2 Hz, OCH₂CH₃), 25.7 s (CH₂CH₂), 33.6 d (${}^{3}J_{\rm PC}$ = 5.6 Hz, CH₂CH₂), 39.4 s (NCH₃), 62.1 d (${}^{2}J_{\rm PC}$ = 6.4 Hz, OCH₂CH₃), 68.9 d (${}^{1}J_{\rm PC}$ = 205.2 Hz, CP), 200.3 d (${}^{2}J_{\rm PC}$ = 20.1 Hz, CN). Mass spectrum, *m/z* ($I_{\rm rel}$, %): 219 (8.5), 206 (2.1), 133 (15.1), 132 (100), 172 (50.5), 144 (6.0), 104 (10.3), 91 (80.1), 77 (1.2), 65 (21.3), 41 (10.3). Found, %: C 49.68; H 8.38; N 6.25; P 13.94. C₉H₁₈NO₃P. Calculated, %: C 49.31; H 8.28; N 6.39; P 14.13.

Diethyl {2-[(2-phenylethyl)amino]cyclobut-1-en-**1-yl}phosphonate (7g).** ¹H NMR spectrum, δ, ppm: 1.26 t (6H, $J_{\rm HH}$ = 6.9 Hz, OCH₂CH₃), 2.68 br.t (2H, $J_{\rm HH} = 6.3$ Hz, CH₂CH₂), 2.74 br.t (2H, $J_{\rm HH} = 6.7$ Hz, CH_2CH_2), 2.81 br.t (2H, $J_{HH} = 6.5$ Hz, CH_2CH_2Ph), 2.92 br.t (2H, $J_{\rm HH}$ = 6.9 Hz, CH₂CH₂Ph), 3.96 m (4H, OCH₂CH₃), 7.10–7.30 m (5H, Ph). ³¹P NMR spectrum: δ_P 27.50 ppm. ¹³C NMR spectrum, δ_C , ppm: 16.1 d $({}^{3}J_{PC} = 6.8 \text{ Hz}, \text{ OCH}_{2}\text{CH}_{3}), 27.7 \text{ s} (CH_{2}CH_{2}), 38.5 \text{ d}$ $({}^{3}J_{PC} = 4.6 \text{ Hz}, \text{CH}_{2}\text{CH}_{2}), 38.5 \text{ s} (\text{CH}_{2}\text{CH}_{2}\text{Ph}), 42.7 \text{ s}$ (CH_2CH_2Ph) , 60.4 d $(^2J_{PC} = 5.7 \text{ Hz}, \text{ OCH}_2CH_3)$, 70.2 d $({}^{1}J_{PC} = 216.6 \text{ Hz}, \text{ CP}), 126.0 (C_{arom}), 128.2 (C_{arom}),$ 128.5 (C_{arom}), 139.0 (C_{arom}), 160.8 d ($^{2}J_{PC} = 19.6$ Hz, CN). Mass spectrum, m/z (I_{rel} , %): 309 (0.5), 297 (1.1), 252 (29.2), 206 (100), 178 (17.2), 160 (29.1), 132 (2.1), 105 (11.5), 104 (11.4), 91 (12.0), 77 (5.2), 57 (1.0). Found, %: C 61.94; H 7.98; N 4.67; P 9.88. C₁₆H₂₄NO₃P. Calculated, %: C 62.12; H 7.82; N 4.53; P 10.01.

Diethyl {2-[(2-hydroxyethyl)amino]cyclobut-1en-1-yl}phosphonate (7h). ¹H NMR spectrum, δ , ppm: 1.22 t (6H, $J_{\rm HH}$ = 7.3 Hz, $J_{\rm PH}$ = 1.3 Hz, OCH_2CH_3), 2.38 br.t (2H, $J_{HH} = 7.5$ Hz, CH_2CH_2), 2.57 br.t (2H, $J_{\rm HH}$ = 7.5 Hz, CH₂CH₂), 3.16 t (2H, $J_{\rm HH} = 7.7$ Hz, CH₂CH₂OH), 3.28 t (2H, $J_{\rm HH} = 7.7$ Hz, CH₂CH₂OH), 4.06 m (4H, OCH₂CH₃). ³¹P NMR spectrum: δ_P 27.98 ppm. ¹³C NMR spectrum, δ_C , ppm: 16.5 d (${}^{3}J_{PC}$ = 6.6 Hz, OCH₂CH₃), 22.7 s (CH₂CH₂), 45.6 d (${}^{3}J_{PC}$ = 31.1 Hz, CH₂CH₂), 48.5 s (CH₂CH₂OH), 54.0 s (CH_2CH_2OH), 60.4 d ($^2J_{PC} = 5.2$ Hz, OCH_2CH_3), 73.8 d (${}^{1}J_{PC}$ = 211.6 Hz, CP), 160.6 d $({}^{2}J_{PC} = 21.6 \text{ Hz, CN})$. Mass spectrum, m/z (I_{rel} , %): 249 (2.2), 234 (8.4), 263 (15.6), 205 (13.8), 176 (10.3), 144 (100), 133 (21.7), 128 (50.5), 97 (33.5), 44 (28.4).Found, %: C 47.00; H 7.95; N 5.73; P 12.59. C₁₀H₂₀NO₄P. Calculated, %: C 48.19; H 8.09; N 5.62; P 12.43.

Diethyl {2-[di(propan-2-yl)amino]cyclobut-1-en-1-yl}phosphonate (7i). ¹H NMR spectrum, δ, ppm: 1.14 d [12H, $J_{\text{HH}} = 6.2$ Hz, CH(CH₃)₂], 1.26 t (6H, $J_{\text{HH}} = 6.9$ Hz, OCH₂CH₃), 2.68 br.t and 2.91 br.t (2H each, $J_{\text{HH}} = 7.2$ Hz, CH₂CH₂), 2.90 m [2H, CH(CH₃)₂], 4.09 m (4H, OCH₂CH₃). ³¹P NMR spectrum: δ_{P} 30.32 ppm. ¹³C NMR spectrum, δ_{C} , ppm: 16.4 d (³ $J_{\text{PC}} = 6.2$ Hz, OCH₂CH₃), 22.6 s [CH(CH₃)₂], 23.3 s (CH₂CH₂), 32.7 d (³ $J_{\text{PC}} = 5.6$ Hz, CH₂CH₂), 48.8 s [CH(CH₃)₂], 60.1 d (² $J_{\text{PC}} = 5.2$ Hz, OCH₂CH₃), 73.3 d (¹ $J_{\text{PC}} = 200.8$ Hz, CP), 168.6 d (² $J_{\text{PC}} = 19.9$ Hz, CN). Mass spectrum, m/z (I_{rel} , %): 289 (18.8), 274 (22.4), 259 (15.4), 246 (100), 231 (15.4), 218 (33.4), 207 (8.8), 188 (39.8), 155 (70.5), 144 (5.8), 76 (9.3), 41 (15.7). Found, %: C 57.96; H 9.66; N 4.94; P 10.8. C₁₄H₂₈NO₃P. Calculated, %: C 58.11; H 9.75; N 4.84; P 10.70.

Diethyl [2-(propylamino)cyclobutyl]phosphonate (8). ¹H NMR spectrum, δ , ppm: 0.88 t (3H, $J_{HH} =$ 7.5 Hz), 1.31 t (6H, $J_{HH} =$ 7.2 Hz), 1.60 m (2H), 1.92– 2.40 m (5H), 2.55 br.t (2H, $J_{HH} =$ 7.3 Hz), 2.95 m (1H), 4.10 m (4H, OCH₂CH₃). ³¹P NMR spectrum: δ_P 32.66 ppm. ¹³C NMR spectrum, δ_C , ppm: 11.4, 16.0 d (³ $J_{PC} =$ 6.3 Hz), 18.1, 21.6, 27.8, 51.2, 42.7 d (¹ $J_{PC} =$ 215.6 Hz), 54, 61.2 d (² $J_{PC} =$ 4.6 Hz). Mass spectrum, m/z (I_{rel} , %): 247 (4.2), 232 (10.2), 207 (11.1), 158 (100), 144 (6.0), 133 (21.7), 104 (10.8), 85 (5.7), 43 (35.6). Found, %: C 53.66; H 9.06; N 5.54; P 12.39. C₁₁H₂₂NO₃P. Calculated, %: C 53.43; H 8.97; N 5.66; P 12.53.

FUNDING

This research was supported by Al-Quds University funding.

CONFLICT OF INTEREST

The authors declare no conflict of interest.

REFERENCES

- Moore, H.W. and Decker, O.H.W., *Chem. Rev.*, 1986, vol. 86, p. 821. https://doi.org/10.1021/cr00075a006
- Misale, A., Niyomchon, S., and Maulide, N., Acc. Chem. Res., 2016, vol. 49, p. 2444. https://doi.org/10.1021/acs.accounts.6b00375
- Namyslo, J.C. and Kaufmann, D.E., *Chem. Rev.*, 2003, vol. 103, p. 1485. https://doi.org/10.1021/cr010010y
- Yavari, I. and Asghari, S., *Tetrahedron*, 1999, vol. 55, p. 11853. https://doi.org/10.1016/S0040-4020(99)00671-7

- Bozkaya, U. and Özkan, İ., J. Org. Chem., 2012, vol. 77, p. 5714. https://doi.org/10.1021/jo300877w
- Kitayama, T., Kawauchi, T., Nakamura, M., Sufi, B., Padias, A.B., and Hall, H.K., *Polymer*, 2004, vol. 45, p. 5085. https://doi.org/10.1016/j.polymer.2004.05.024
- Chen, X.P., Sufi, B.A., Padias, A.B., and Hall, H.K., Macromolecules, 2002, vol. 35, p. 4277. https://doi.org/10.1021/ma012104f
- Perrott, M.G. and Novak, B.M., *Macromolecules*, 1996, vol. 29, p. 1817. https://doi.org/10.1021/ma951516j
- Wu, Z. and Grubbs, R.H., *Macromolecules*, 1995, vol. 28, p. 3502. https://doi.org/10.1021/ma00114a002
- 10. Casey, P.K. and Mathias, L.J., *Polym. Commun.*, 1991, vol. 32, p. 27.
- Hall, H.K. and Padias, A.B., J. Polym. Sci., Part A-1: Polym. Chem., 2003, vol. 41, p. 625. https://doi.org/10.1002/pola.10618
- Ingham, S., Turner, R.W., and Wallace, T.W., J. Chem. Soc., Chem. Commun., 1985, p. 1664. https://doi.org/10.1039/C39850001664
- Binns, F., Hayes, R., Ingham, S., Saengchantara, S.T., Turner, R.W., and Wallace, T.W., *Tetrahedron*, 1992, vol. 48, p. 515. https://doi.org/10.1016/S0040-4020(01)89013-X
- Hayes, R., Ingham, S., Saengchantara, S.T., and Wallace, T.W., *Tetrahedron Lett.*, 1991, vol. 32, p. 2953. https://doi.org/10.1016/0040-4039(91)80660-x
- Binns, F., Hayes, R., Hodgetts, K.J., Saengchantara, S.T., Wallace, T.W., and Wallis, C.J., *Tetrahedron*, 1996, vol. 52, p. 3631. https://doi.org/10.1016/0040-4020(96)00039-7
- Hodgetts, K.J., Saengchantara, S.T., Wallis, C.J., and Wallace, T.W., *Tetrahedron Lett.*, 1993, vol. 34, p. 6321. https://doi.org/10.1016/s0040-4039(00)73742-7
- Gourdel-Martin, M.E. and Huet, F., *Tetrahedron Lett.*, 1996, vol. 37, p. 7745. https://doi.org/10.1016/0040-4039(96)01763-7
- Youcef, R.A., Boucheron, C., Guillarme, S., Legoupy, S., Dubreuil, D., and Huet, F., *Synthesis*, 2006, p. 633. https://doi.org/10.1055/s-2006-926303
- 19. Dembitsky, V.M., *J. Nat. Med.*, 2008, vol. 62, p. 1. https://doi.org/10.1007/s11418-007-0166-3
- Hansen, T.V. and Stenstrøm, Y., Organic Synthesis: Theory and Applications, Amsterdam: Elsevier, 1996, vol. 5, p. 1. https://doi.org/10.1016/s1047-773x(01)80002-0
- 21. Zarghi, A. and Arfaei, S., Iran. J. Pharm. Res., 2011, vol. 10, p. 655.
- Talley, J.J., Prog. Med. Chem. Res., 1999, vol. 36, p. 201. https://doi.org/10.1016/s0079-6468(08)70048-1

RUSSIAN JOURNAL OF ORGANIC CHEMISTRY Vol. 56 No. 1 2020

- 23. Praveen, P.N., Rao, E.E., and Knaus, J., *J. Pharm. Pharm. Sci.*, 2008, vol. 11, p. 81. https://doi.org/10.18433/j3t886
- Smith, S.C., Clarke, E.D., Ridley, S.M., Bartlett, D., Greenhow, D.T., Glithro, H., Klong, A.Y., Mitchell, G., and Mullier, G.W., *Pest Manage. Sci.*, 2005, vol. 61, p. 16. https://doi.org/10.1002/ps.980
- Liu, Z.Y., Wang, Y.M., Han, Y.X., Liu, L., Jin, J., Yi, H., Li, Z.R., Jiang, J.D., and Boykin, D., *Eur. J. Med. Chem.*, 2013, vol. 65, p. 187. https://doi.org/10.1016/j.ejmech.2013.04.046
- Ortuño, R., Moglioni, M., and Moltrasio, A.G., *Curr.* Org. Chem., 2005, vol. 9, p. 237. https://doi.org/10.2174/1385272053369088
- Sergeiko, A., Poroikov, V., Hanuš, L., and Dembitsky, V., *Open Med. Chem. J.*, 2008, vol. 2, p. 26. https://doi.org/10.2174/1874104500802010026
- Owsley, D.C. and Bloomfield, J.J., J. Chem. Soc. C, 1971, p. 3445. https://doi.org/10.1039/j39710003445
- Kosugi, H., Sekiguchi, S., Sekita, R., and Uda, H., Bull. Chem. Soc. Jpn., 1976, vol. 7, p. 520. https://doi.org/10.1002/chin.197620094
- 30. Baldwin, S.W., *Organic Photochemistry*, Padwa A., Ed., New York: Marcel Dekker, 1981, vol. 5, p. 123.
- 31. Demuth, M., and Gamal, M., *Synthesis*, 1989, p. 145. https://doi.org/10.1055/s-1989-27181
- 32. Alibés, R., March, P., Figueredo, M., Font, J., Racamonde, M., Rustullet, A., Alvarez-Larena, A., Piniella, J.F., and Parella, T., *Tetrahedron Lett.*, 2003, vol. 44, p. 69.
 - https://doi.org/10.1016/s0040-4039(02)02528-5
- Delas, C., Urabe, H., and Sato, F., *Tetrahedron Lett.*, 2001, vol. 42, p. 4147. https://doi.org/10.1016/s0040-4039(01)00666-9
- Brandsma, L., Spek, A.L., Trofimov, B.A., Tarasova, O.A., Nedolya, N.A., Afonin, A.V., and Zinshenko, S.V., *Tetrahedron Lett.*, 2001, vol. 42, p. 4687. https://doi.org/10.1016/s0040-4039(01)00787-0
- Barbero, A., Cuadrado, P., Garcia, C.A., Rincon, J.A., and Pulido, F.J., *J. Org. Chem.*, 1998, vol. 63, p. 7531. https://doi.org/10.1021/jo980874s
- Liu, Y., Liu, M., and Song, Z., J. Am. Chem. Soc., 2005, vol. 127, p. 3662. https://doi.org/10.1021/ja042636m

- López-Carrillo, V. and Echavarren, A.M., J. Am. Chem. Soc., 2010, vol. 132, p. 9292. https://doi.org/10.1021/ja104177w
- Negishi E., Liu F., Choueiry D., and Mohamud M.M., J. Org. Chem., 1996, vol. 61, p. 8325. https://doi.org/10.1021/jo961277d
- Perri, S.T., Dyke, H.K., and Moore, H.W., *J. Org. Chem.*, 1989, vol. 54, p. 2032. https://doi.org/10.1021/jo00270a004
- Song, Z.Q., Li, Y.X., Liu, M., Cong, L.Q., and Liu, Y.H., *Organometallics*, 2006, vol. 25, p. 5035. https://doi.org/10.1021/om060505j
- Chen, C., Xi, C., Lai, C., Wang, R., and Hong, X., *Eur. J.* Org. Chem., 2004, p. 647. https://doi.org/10.1002/ejoc.200300485
- Al Quntar, A.A., Srebnik, M., Terent'ev, A.O., and Dembitsky, V., *Mini-Rev. Org. Chem.*, 2014, vol. 11, p. 445. https://doi.org/10.2174/1570193x1104140926170132
- 43. Bauer, G. and Haegele, G., Z. Naturforsch., B: Anorg. Chem., Org. Chem., 1979, vol. 34, p. 1252. https://doi.org/10.1515/znb-1979-0918
- Ueda, T., Inukai, K., and Muramatsu, H., *Bull. Chem.* Soc. Jpn., 1969, vol. 42, p. 1684. https://doi.org/10.1246/bcsj.42.1684
- Tomioka, H., Watanabe, M., Kobayashi, N., and Hirai, K., *Tetrahedron Lett.*, 1990, vol. 31, p. 5061. https://doi.org/10.1016/s0040-4039(00)97806-7
- Facklam, T., Hoffmann, K.L., and Regitz, M., *Chem. Ber.*, 1987, vol. 120, p. 1397. https://doi.org/10.1002/cber.19871200817
- Andriamiadanarivo, R., Pujol, B., Chantegrel, B., Dehayes, C., and Doutheau, A., *Tetrahedron Lett.*, 1993, vol. 34, p. 7923. https://doi.org/10.1016/s0040-4039(00)61512-5
- Darling, S.D. and Subramanian, N., *Tetrahedron Lett.*, 1975, vol. 38, p. 3279. https://doi.org/10.1016/s0040-4039(00)91425-4
- Sinelnikove, Y., Rubinstein, A., Srebnik, M., and Al Quntar, A.A., *Tetrahedron Lett.*, 2009, vol. 50, p. 867. https://doi.org/10.1016/j.tetlet.2008.11.108
- Srivastava, H.K., Al Quntar, A.A., Azab, A., Srebnik, M., and Shurki, A., *Tetrahedron*, 2009, vol. 65, p. 4389. https://doi.org/10.1016/j.tet.2009.03.053