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The synthesis and biological evaluation of a kabiramide C fragment modified with a WH2 consensus actin-binding motif as a potential disruptor of the actin cytoskeleton[†]

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The F-actin depolymerisation potency of a fragment of kabiramide C was increased when modified with a WH2 consensus actin-binding motif LKKV. Despite its low affinity for actin monomers, a shorter analogous fragment not bearing LKKV was identified as a potent inhibitor of actin polymerisation and a promoter of its depolymerisation, resulting in a loss of actin stress fibres in cells.

The control of the actin cytoskeleton by actin-binding proteins (ABPs) is vital to a large number of fundamental cellular processes.¹ The Wiskott-Aldrich syndrome protein-homology domain 2 (WH2) is an actin-binding motif that binds actin between its subdomains 1 and 3.² WH2 is widespread among ABPs and a representative example of the actin/WH2 complex is depicted in Fig. 1a with the WH2 of the missing in metastasis protein (MIM, cyan and red cartoon).³ In WH2, a key actinrecognition element consists of two basic residues (usually lysine), flanked by two hydrophobic residues.^{2,4} Thus, a typical WH2 consensus actin-binding motif is LKKV or LKKT (Fig. 1a, red section of the cartoon).²⁻⁴ Moreover, the actin cytoskeleton can be disrupted by several natural products,^{5,6} among which kabiramide C (1) (Fig. 1b, green tubes),⁷ aplyronine A and reidispongiolide A are prime examples. These marine toxins are structurally related and consist of a well-conserved aliphatic side chain attached to a macrolactone of varying size and structure. The crystal structures of the 1:1 complexes of these compounds and the actin monomer (G-actin) show a considerable overlap of their binding site on actin with those of WH2 in actin/WH2 complexes, as illustrated for **1** in Fig. 1a.^{7,8} Importantly, previous studies suggest that these compounds disrupt the actin cytoskeleton by slowly severing actin filaments (F-actin) and capping the shortened filament thus formed at their so-called barbed end.⁸

† Electronic supplementary information (ESI) available: Characterisation of synthetic procedures of compounds, bioassays results. See DOI: 10.1039/c5cc06081h

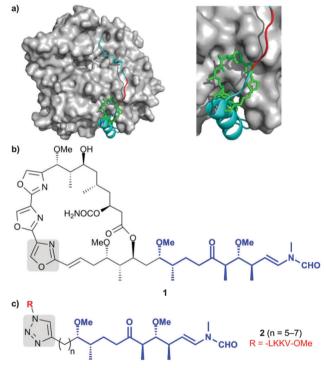


Fig. 1 (a) Superimposition of actin/WH2 of the MIM (1:1) complex (PDB: 2D1K, cyan and red cartoon) and the actin/kabiramide C (1:1) complex (PDB: 1QZ5, green tubes). (b) Structure of kabiramide C (1). (c) Structure of hybrids 2 designed from the truncated side chain of 1 and the key actin-binding tetrapeptide LKKV.

In addition, the sub-nanomolar affinity of these natural toxins for G-actin suggests that the elongation of the filaments is also prevented by the incorporation of a G-actin/toxin complex at the barbed end of the filaments.⁸ Remarkably, when truncated from the macrolactone ring, the lateral chain of aplyronine A and analogues thereof can still promote the depolymerisation of F-actin but with at least a ten-fold reduced potency as compared to the natural product.⁹ Similarly, analogues of the lateral chain of reidispongiolide A, whereby the macrolactone ring is

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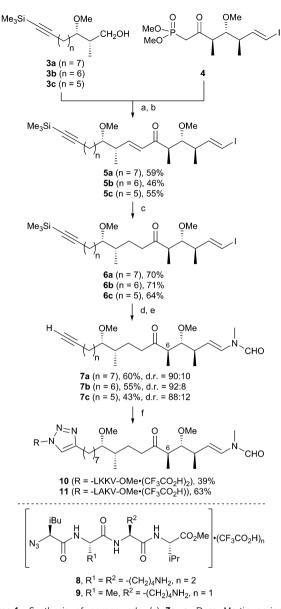
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replaced with small apolar groups to enhance the hydrophobic interactions with actin, still display a sub-micromolar affinity for G-actin and are reported to sever F-actin.¹⁰

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In contrast to the strategy previously followed with the modified fragments of reidispongiolide A^{10} and in view of the proximity of the WH2 consensus actin-binding motifs and 1 when in a complex with G-actin (Fig. 1a),^{3,7} we hypothesised that replacing the macrolactone moiety of 1 with the key actinbinding tetrapeptide LKKV in compound 2 (Fig. 1c) would deliver hybrids that could trigger similar effects on the actin cytoskeleton as compared to 1. According to this design, both actin-binding sections of 2 would be tethered by a chain of adjustable length (n = 5-7) and a triazole moiety that would replace one of the oxazole rings of 1 (highlighted in grey). We anticipated that this design could serve as a new basis for the future development of molecular tools to study the actin cytoskeleton, or drugs to fight diseases (*e.g.* cancer, viral infection, intraocular pressure and outflow) by targeting actin.¹¹

Our approach towards hybrids 2 is illustrated in Scheme 1. Horner-Wadsworth-Emmons olefination between advanced fragments 3 and 4^{12,13} afforded 5. The reduction of the carbon-carbon double bond thus formed using Stryker's reagent¹⁴ afforded intermediate 6, with partial desilylation of the terminal alkyne (4%). The crude material was used without further purification in the next step. The N-methyl formamide moiety was installed using a copper-catalysed amidation reaction,^{9b,15} and those basic conditions caused partial isomerisation at C(6).^{16,17} The diastereomers were not separated and the synthesis was pursued nonetheless, as we assumed that the ratio of diastereomers would not prevent a meaningful evaluation of our hypothesis about the biological activity of the final compounds. Thus, the cleavage of the trimethylsilyl group using AgNO₃¹⁸ afforded the terminal alkyne 7a. Attempts to achieve the same cleavage using other conditions such as K₂CO₃ in MeOH or tetrabutylammonium fluoride in THF were unsuccessful and led to either an incomplete reaction or decomposition, respectively. Analogues 7b and 7c were prepared in the same fashion, whereby uncontrolled desilylation of the triple bond (6b and 6c, 12%) during the Stryker reduction and epimerisation at C(6) during the amidation reaction were also observed. Although the copper-catalysed formation of the triazole 2 using alkyne 7a and azide 8 initially enabled us to obtain 10 as free bases,¹⁹ we noticed a relatively rapid decomposition upon storage. Although the products of this decomposition were not identified, we suspected that it was linked to the presence of the free primary amino groups on the side chains of the two lysine residues. Accordingly, we modified the work-up procedure of the coppercatalysed 1,3-dipolar addition between 7a and 8 in order to isolate 10 as a bis-TFA salt. In order to assess whether a putative enhancement of actin-binding properties could be attributed to the specific tetrapeptide LKKV, the analogous control substrate 11 was prepared from 7a and 9, whereby the LKKV fragment was replaced with LAKV. Finally, although we initially intended to prepare analogues of 10 from 7b and 7c, the results from the initial comparison of the affinity for G-actin of 7a and 10 did not encourage us to pursue this endeavour.



Scheme 1 Synthesis of compounds. (a) 3a-c, Dess-Martin periodinane (1.85 equiv.), NaHCO₃ (2 equiv.), CH₂Cl₂, rt, 0.25 h; (b) (i) **4** (1 equiv.), Ba(OH)₂ (1.1 equiv.), THF, 0.5 h then (ii) aldehyde obtained from 3a-c (1 equiv.), THF/ water (40:1), rt, 1 h; (c) (CuHPPh₃)₆ (0.33 equiv.), water (10 equiv.), toluene, rt, 1 h; (d) Cul (0.2 equiv.), 1,2-*trans*-cyclohexyldiamine (0.4 equiv.), K₃PO₄ (2 equiv.), MeNCHO (10 equiv.), 1,4-dioxane, 80 °C, 16 h; (d) AgNO₃ (4 equiv.), THF/EtOH/water/2,6-lutidine (1:1:1:0.1), rt, 48 h; (e) CuSO₄ (0.1 equiv.), sodium L-ascorbate (0.2 equiv.), **8** or **9**, t-BuOH, 35 °C, 48 h.

Thus, a rapid comparison of the G-actin binding properties of **7a**, **10** and **11** was conducted by scanning the fluorescence intensity of an 1 μ M solution of prodan-G-actin in the absence or presence of 25 μ M solutions of the compounds (Fig. 2).^{10,20} Contrary to our initial hypothesis that modifying **7a** with the LKKV fragment would increase the affinity of **10** for G-actin, the scan did not reveal any evidence of such an effect. In contrast, a minor increase of affinity was observed for control compound **11**, which bears the LAKV fragment. Moreover, shortening the aliphatic chain in **7a–7c** led to compounds with increasingly lower affinity for G-actin.

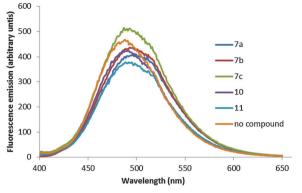


Fig. 2 Fluorescence emission spectra of 1 μ M prodan-G-actin in the presence of 25 μ M compounds **7a–7c**, **10**, and **11**.

The potency of all compounds was nevertheless evaluated in G-actin polymerisation²¹ and F-actin depolymerisation assays²² as well as whole cell assays.²³ In the polymerisation assay of 2 μ M pyrenyl G-actin, the initial rates of polymerisation were calculated for compounds in the initial linear phase of polymerisation for up to 20 minutes. The minimum concentration that reduced the polymerisation rate by 10% compared to DMSO control was deemed significant. The quantification of the depolymerisation assay was performed by measuring the amount of actin in the supernatant obtained after incubation of 6 µM F-actin with 7a-c, 10, and 11. A compound was deemed to have a significant effect at a concentration that resulted in 5% of actin in the supernatant fraction whereas between 1 and 2% of actin was in the supernatant fraction when F-actin was incubated with DMSO only. The results of those assays are summarised in Table 1. They demonstrate that despite its low affinity for G-actin, 7c is a strong inhibitor of G-actin polymerisation whilst it also promotes the depolymerisation of F-actin. Treating REF52 rat embryo fibroblast cells with compound 7c led to a concentrationdependent loss of actin stress fibres and actin stress fibre organisation (Fig. 3). Compared to DMSO alone, 800 nM 7c caused an increased accumulation of disorganised F-actin clouds in the perinuclear region, with a progressive loss of stress fibres and stress fibre organisation with increased concentrations of up to 20 µM. Furthermore, the numbers of cells remaining attached to the culture dish was also reduced as the concentration of 7c increased, probably as a consequence of the loss of stress

Table 1 Potency of compounds **7a–c**, **10** and **11** in regard to the inhibition of G-actin polymerisation (A), promotion of F-actin depolymerisation (B) and to the disruption of actin cytoskeleton in rat fibroblast cells after 2 h (C) and 24 h $(D)^a$

Compound	A^b	B^b	С	D
7a 7b	10 (-)	5 (+)	\pm^d	20
7b 7c	1 (+) 1 (-)	5 (+) 5 (+)	20 + d	20 0.8
10	10 (+)	1.5 (+)	20	\pm^{d}
11	\pm^{c}	5 (+)	20	\pm^d

^{*a*} Minimal concentration (μ M) of compound for which a significant effect was observed. ^{*b*} Inhibiting and promoting effects are indicated by a minus and plus sign, respectively. ^{*c*} No effect at 10 μ M. ^{*d*} No effect at 20 μ M.

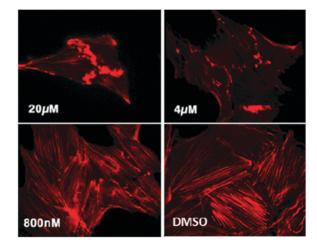


Fig. 3 Effect of 7c on the actin cytoskeleton in REF52 rat embryo fibroblast cells 24 h after they have been treated with 0.8–20 μ M solutions of the compound.

fibre-mediated cell adhesion contacts to the substratum. At 20 μ M 7c, in any remaining adherent cells, F-actin was mostly concentrated, with dense accumulation at the cell periphery adjacent to other cells. Among compounds 7a–7c, 7c is the most potent inhibitor of G-actin polymerisation and the most potent promoter of F-actin depolymerisation (Table 1). Moreover, attaching the LKKV fragment to the polypropionate chain of 7a increased the potency of 10 to act as a promoter of F-actin depolymerisation. This effect appears to be attributable to the LKKV fragment as replacing even one of the two lysine residues with a non-basic residue (11, Table 1) led to a complete loss of potency in all assays. However, both 10 and 11 failed to elicit any significant effects in whole-cell assays, which might be due to issues of cell penetration.

In conclusion, and contrary to our initial hypothesis, modifying a simplified side chain of kabiramide C with a WH2 actin-binding motif does not enhance the affinity of the compound thus obtained for G-actin. However, the F-actin depolymerisation potency is increased by this modification and this effect appears to be specific to the LKKV fragment. Moreover, the remarkable effects observed for 7c in the assays described above cannot be explained by the sequestration of actin monomers due to its low affinity for G-actin, in contrast to the modified side chains of reidispongiolide A that exhibit a remarkable sub-micromolar affinity for G-actin.^{10b} Further investigations are required to delineate more precisely the mechanism of action of this compound.

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