This article was downloaded by: [McGill University Library] On: 19 August 2012, At: 18:22 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



# Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lsyc20

Heterocyclizations via POCl<sub>3</sub>-Based Multicomponent Reactions: A New Approach to One-Pot Synthesis of a New Spirosystem, 7-Methyl-5-[4-(aryl/heteryl)thiazol-2-yl]-5,6diazaspiro[2,4]hept-6-en-4-ones

V. Rajeswar Rao<sup>a</sup> & V. Ravinder Reddy<sup>a</sup> <sup>a</sup> Department of Chemistry, National Institute of Technology, Warangal, India

Version of record first published: 04 Oct 2010

To cite this article: V. Rajeswar Rao & V. Ravinder Reddy (2010): Heterocyclizations via POCI<sub>3</sub>-Based Multicomponent Reactions: A New Approach to One-Pot Synthesis of a New Spirosystem, 7-Methyl-5-[4-(aryl/heteryl)thiazol-2-yl]-5,6-diazaspiro[2,4]hept-6-en-4-ones, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 40:21, 3186-3195

To link to this article: http://dx.doi.org/10.1080/00397910903372341

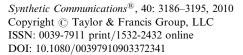
## PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <u>http://www.tandfonline.com/page/terms-and-conditions</u>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings,

demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.





## HETEROCYCLIZATIONS VIA POCI<sub>3</sub>-BASED MULTICOMPONENT REACTIONS: A NEW APPROACH TO ONE-POT SYNTHESIS OF A NEW SPIROSYSTEM, 7-METHYL-5-[4-(ARYL/HETERYL)THIAZOL-2-YL]-5,6-DIAZASPIRO[2,4]HEPT-6-EN-4-ONES

V. Rajeswar Rao and V. Ravinder Reddy

Department of Chemistry, National Institute of Technology, Warangal, India

A novel multicomponent reaction involving phenacylbromide, thiosemicarbazide, and  $\alpha$ -acetyly-butyrolactone in the presence of POCl<sub>3</sub>, forming 7-methyl-5-[4-(aryllheteryl)thiazole-2yl]-5,8-diaza spiro[2,4]hept-6-en-4-ones in good yields, is described.

Keywords: α-Acetyl-γ-butyrolactone; 3-(2-bromoacetyl)coumarin; phenacylbromides; spiro compound; thiazole

## INTRODUCTION

In view of the wide range of biological activates exhibited by 2H-1benzopyran-2-ones, pyrazoles, and thiazoles, we became interested in synthesizing these moieties in one molecule, which is expected to enhance their biological activity as compared to simple pyrazolothiazoles. Based on these observations, we report the synthesis of a new spirosystem, 7-methyl-5-[4-(aryl/heteryl)thiazol-2-yl]-5,6diazaspiro[2,4]hept-6-en-4-ones.

Thiazoles are generally synthesized by Hantzsch's thiazole synthesis from  $\alpha$ -halogenoketones and thioureas or thioamides. Dodson and King<sup>[1]</sup> and others<sup>[2]</sup> synthesized aminothiazoles by a modification of the method. The method still remains a cumbersome and time-consuming process.

A literature survey<sup>[3]</sup> clearly revealed that normally 2-pyrazol-4-yl-substituted thiazoles are prepared in two distinct methods, one involving preparation of thiazole first and then building a pyrazole in the next step or vice versa. Unlike the literature methods, we have synthesized a 2-pyrazol-4-yl-substituted thiazole system in one step.

Though the previous methodologies are quite useful, they have some limitations, such as the requirement to isolate intermediates and longer reaction times, and the overall yields are poor. None of these methods are simple, nor can they be usefully applied for the generation of functionally substituted pyrazolothiazoles. It is thus evident that there remains scope for the development of clean and efficient methodologies involving single-step reactions for the preparation of the title compounds. In the

Received July 21, 2009.

Address correspondence to V. Rajeswar Rao, Department of Chemistry, National Institute of Technology, Warangal 506 004, AP, India. E-mail: vrajesw@yahoo.com

present investigation, there is a simultaneous selective ring-closure reaction involving the formation of three rings at time, such as thiazole, pyrazole, and carbocyclic rings, with spiro linkages at the third position of coumarin in an acidic medium.

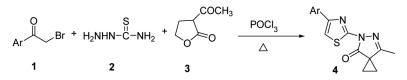
## **RESULTS AND DISCUSSION**

In continuation of our earlier work on the synthesis of heterocyclic systems from  $\alpha$ -halo carbonyl compounds,<sup>[4]</sup> we report herein a facile, novel route for the synthesis of 7-methyl-5-[4-(aryl/heteryl)-2-yl]-5,6-diazaspiro[2.4]hept-6-en-4-ones in a single step from easily available starting materials. The reaction involves simultaneous cyclization, leading to the formation of thiazole, pyrazolone, and cyclopropane rings. As shown in Scheme 1 and Table 1, the reaction of aryl bromomethylketones/heterylbromomethylketones with thiosemicarbazide and  $\alpha$ -bromoacetyl- $\gamma$ -butyrolactone in POCl<sub>3</sub> under heating yielded the title compounds in one step.

Our experiment was first conducted by adding equimolar quantities of phenacyl bromide, thiosemicarbazide, and  $\alpha$ -acetyl- $\gamma$ -butyrolactone in POCl<sub>3</sub> under heating, which yielded 7-methyl-5-[4-phenyl-thiazol-2-yl]-5,6-diazaspiro[2.4]hept-6-one rather than an expected compound such as {2-[5-hydroxy-4-(2-hydroxy-ethyl)-3-methyl-pyrazol-4-yl]-thiazol-4-yl}benzene, as confirmed by spectral analysis. The infrared (IR) spectra of **4a** displayed bands in the region 1621 cm<sup>-1</sup> (amide, -C=O), 1602 cm<sup>-1</sup> (-C=N). The <sup>1</sup>H NMR spectrum of **4a** displayed characteristic signals at 1.83 (ABq, 2H, spiro  $-CH_2-CH_2$ ), 1.94 (ABq, 2H, CH<sub>2</sub> $-CH_2$ ), and 2.10 (s, 3H, CH<sub>3</sub>). In the mass spectrum of **4a**, molecular ion was recorded at m/z 306 (M + Na), 284 (M + 1).

Further examination of the scope of the reaction revealed that substituted 3-(2-bromoacetyl)chromenes, thiosemicarbazide, and  $\alpha$ -acetyl- $\gamma$ -butyrolactone can also participate in the reaction and produce substituted 7-methyl-5-[4-(2-oxo-2*H*-chromen-3-yl)thiazol-2-yl]-5,6-diazaspiro[2.4]hept-6-en-4-one. The formation of the product (**4b**) was confirmed by spectral and analytical data. For the conformation of this product, we recorded 2D <sup>1</sup>H and <sup>13</sup>C correlation NMR spectra. In 2D NMR, each spot on the HETCOR plot has been labeled. The carbon peak at 13.1 ppm and the proton singlet at 2.1 ppm correspond to the methyl group, the carbon peaks at 20.2 and 20.2 ppm and the protons of AB quartets at 1.83 and 1.95 ppm correspond to the methylene group, and the carbon peak at 116.7 ppm and the proton doublet at 7.37 ppm correspond to H-8 of coumarin. The carbon peak at 125.0 ppm and the proton doublet of doublets at 7.28 ppm correspond to the H-6 of coumarin. The carbon peak at 128.8 ppm and the proton doublet at 7.65 ppm correspond to H-5 of coumarin.

The carbon peak at 131.8 ppm and the proton doublet of doublets correspond to the H-7 of coumarin at 7.53 ppm. The carbon peak at 140.1 ppm and the proton



Scheme 1. One-step preparation of 7-methyl-5-[4-aryl/heteryl-thiazol-2-yl]-5,6-diazaspiro[2,4]hept-6-one.

No.	Ar	Yield <sup>a</sup> (%)	
		Method 1	Method 2
<b>4</b> a		60	45
4b		55	54
4c	CI C	60	55
4d	Br	40	48
<b>4</b> e	Br Br	40	46
4f	OCH3	55	43
4g		65	45

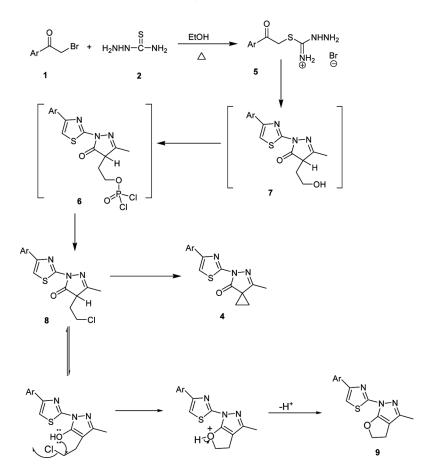
 Table 1. Synthesis of 7-methyl-5-[4-phenyl-thiazol-2-yl]-5,6-diazaspiro 

 [2.4]hept-6-one

 $^{a}$ Yields refer to isolated yield. Compounds are characterized by  $^{1}$ H,  $^{13}$ C NMR, and IR spectra.

singlet at 8.8 ppm correspond to the H–C<sub>4</sub> of coumarin. The proton of C<sub>4</sub> of coumarin is deshielded. Therefore, a spot on the HETCOR plot for this group appears at 140.1 ppm on the carbon axis and at 8.8 ppm on the proton axis. It is interesting that the methyl group of the diazaspiro[2.4]hept-6-en-4-one appears downfield of the methylene group (2) in the proton spectrum (2.18 ppm). In the carbon spectrum, however, the carbon peak for methyl appears upfield of methylene carbons. Thus, these HETCOR values confirm the assigned structure. Based on these observations, we concluded that the product was 7-methyl-5-[4-(2-oxo-2*H*-chromen-3-yl)thiazol-2-yl]-5,6-diazaspiro[2.4]hept-6-en-4-one rather than  $\{2-[5-hydroxy-4-(2-hydroxy-ethyl)-3-methyl-pyrazol-4-yl]-chromen-2-one.$ 

Mechanistically, a second possible reaction pathway could be envisaged (Scheme 2) in this process: condensation of aryl/heteryl bromomethylketones with thiosemicarbazide in anhydrous ethanol at room temperature, resulting in the formation of uncyclized 2-oxo-2-(2-aryl/2-oxo-2*H*-chromen-3-yl)ethylhydrazi-necarbimidathioatehydrobromide (5) (Table 2). Condensation of these uncyclized



Scheme 2. Method 2: stepwise synthesis of 7-methyl-5-[4-phenyl-thiazol-2-yl]-5,6-diazaspiro[2.4]hept-6one and mechanism of the reaction.

compounds (5) with  $\alpha$ -acetyl- $\gamma$ -butyrolactone (3) in POCl<sub>3</sub> under reflux resulted in the formation of title compounds (4) instead of the expected 6 or 7 or both (Scheme 2). It is believed that during the reaction of 5 with  $\alpha$ -acetyl- $\gamma$ -butyro lactone in POCl<sub>3</sub>, the compound 6 formed changes into the corresponding chloroethyl derivatives 8, which undergo in situ intramolecular cyclization with the loss of HCl to give spiro compound 4. Cyclization of intermediate 7 or 8 leads to either product 4 or 9 or both, depending upon the mode of cyclization. In our case, only one product, 4, was obtained as evidenced by thin-layer chromatography (TLC). The formation 9 can be ruled out on the basis of spectral evidence. It is interesting to note that during the course of the reaction, selective intramolecular ring closure occurs to give 4.

#### **EXPERIMENTAL**

All the reagents and solvents were purchased from commercial sources and were used without further purification unless otherwise stated. 3-(2-Bromoacetyl)

No.	Ar	Yield <sup>a</sup> (%)
5a		92
5b		92
5c	CI C	88
5d	Br	90
5e	Br Br	88
5f	OCH3 OCH3	90
5g		91

 
 Table 2. Synthesis of 2-oxo-2-(phenyl)ethylhydrazine carbimidazo thioatehydrobromide

<sup>a</sup>Yields refer to isolated yield. Compounds are characterized by <sup>1</sup>H, <sup>13</sup>C NMR, and IR spectra.

coumarins<sup>[5]</sup> were prepared by the literature procedure. Melting points were determined in open capillaries with a Cintex melting-point apparatus (Mumbai, India) and were uncorrected. CHNS analysis was done on a Carlo Erba EA 1108 automatic elemental analyzer. The purity of the compounds was checked by TLC plates (E. Merek, Mumbai, India). IR spectra (KBr) were recorded on a BrukerWM-4(X) spectrometer (577 model). <sup>1</sup>H NMR spectra were recorded on a Bruker WM-300 spectometer in  $\delta$  ppm using tetramethylsilane (TMS) as internal standard. Mass spectra (EI-MS) were determined on a Perkin-Elmer instrument (SCIEX API-2000, ESI) at 12.5 eV.

#### **General Procedure for 4**

A mixture of aryl/3-coumarinyl bromomethyl ketone (10 mmol), thiosemicarbazide (10 mmol), and  $\alpha$ -acetyl- $\gamma$ -butyrolactone (10 mmol) in POCl<sub>3</sub> (12 ml) was refluxed for 1 h. The reaction mixture was poured in ice-cold water (60 ml). The product formed was purified by column chromatography on silica (eluent, ethyl acetate/hexanes 1.5:8.5) to give the title compound in 40–65% yield. All the other compounds were prepared by a similar procedure.

#### 7-Methyl-5-[4-phenyl-thiazol-2-yl]-5,6-diazaspiro[2.4]hept-6-one (4a)

Mp 120–122 °C. IR (KBr)  $\nu$  (cm<sup>-1</sup>), 1720 (lactone, -C=O), 1621 (amide, -C=O), 1602 (-C=N). <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>)  $\delta$  ppm: 1.83 (ABq, 2H, spiro -CH<sub>2</sub>-CH<sub>2</sub>), 1.94 (ABq, 2H, CH<sub>2</sub>-CH<sub>2</sub>), 2.10 (s, 3H, CH<sub>3</sub>), 7.2–7.45 (m, 4H, Ar-H), 7.97 (d, J=7.8 Hz, 2H, Ar-H); <sup>13</sup>C NMR spectrum (300 MHz, CDCl<sub>3</sub>), 13.06, 20.0 (for two carbons), 33.7, 108.1, 126.9 (for two carbons), 128.5, 128.9 (for two carbons), 134.5, 151.4, 155.9, 162.8, 172.1. ESI-MS: m/z 306 (M<sup>+</sup>+Na), 284 (M<sup>+</sup>+1). Anal. calcd. for C<sub>15</sub>H<sub>13</sub>N<sub>3</sub>OS: C, 63.58; H, 4.62; N, 14.83; S,11.32. Found: C, 63.52; H, 4.53; N, 14.79; S, 11.36.

## 7-Methyl-5-[4-(2-oxo-2H-chromen-3-yl)thiazol-2-yl]-5,6diazaspiro[2.4]hept-6-en-4-one (4b)

Crystalline solid. 55% yield, mp 270–271 °C. IR (KBr)  $\nu$  (cm<sup>-1</sup>), 1720 (lactone, –C=O), 1635 (amide, –C=O), 1610 (–C=N–). <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>)  $\delta$  ppm: 1.83 (ABq, 2H, spiro CH<sub>2</sub>–CH<sub>2</sub>), 1.95 (ABq, 2H, spiro CH<sub>2</sub>–CH<sub>2</sub>), 2.11 (s, 3H, CH<sub>3</sub>), 7.37 (d, J = 8.1 Hz, 1H, C<sub>8</sub> of coumarin), 7.28 (dd, J = 8.1 Hz, 1H, C<sub>6</sub> of coumarin), 7.65 (d, J = 7.8 Hz, 1H, C<sub>5</sub> of coumarin), 7.53 (dd, J = 7.8 Hz, 1H, C<sub>7</sub> of coumarin), 8.29 (s, 1H, thiazole), 8.8 (s, 1H, C<sub>4</sub> of coumarin); <sup>13</sup>C NMR spectrum (300 MHz, CDCl<sub>3</sub>): 13.1, 20.2 (for two carbon atoms), 33.8, 115.5, 116.7, 119.9, 120.9, 125, 128.8, 131.8, 140.1, 144.1, 153.4, 155.5, 160.2, 163.2, 172.2. ESI-MS: m/z 352 (M<sup>+</sup> + 1). Anal. calcd. for C<sub>18</sub>H<sub>13</sub>N<sub>3</sub>O<sub>3</sub>S: C, 61.53; H, 3.73; N, 11.96; S, 9.12. Found: C, 61.59; H, 3.67; N, 11.91; S, 9.08.

## 5-[4-(6-Chloro-2-oxo-2*H*-chromen-3-yl)thiazol-2-yl]-7-methyl-5,6diazaspiro[2.4]hept-6-en-4-one (4c)

Crystalline solid, yield 60%, mp 253–255 °C. IR (KBr),  $\nu$  (cm<sup>-1</sup>), 1721 (lactone, –C=O), 1635 (amide, –C=O), 1604 (–C=N). <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>),  $\delta$  ppm: 1.83 (ABq, 2H, spiro CH<sub>2</sub>–CH<sub>2</sub>), 1.93 (ABq, 2H, spiro CH<sub>2</sub>–CH<sub>2</sub>), 2.11 (s, 3H, C<sub>7</sub> methyl of spiro system), 7.30–7.61 (m, 3H, Ar-H), 8.31 (s, 1H, C<sub>5</sub> of thiazole), 8.73 (s, 1H, C<sub>4</sub> of coumarin). <sup>13</sup>C NMR spectrum (300 MHz, CDCl<sub>3</sub>): 13.1, 20.3 (for two carbon atoms), 33.8, 116.3, 118.2, 121.0, 121.9, 127.8, 130.3, 131.7, 138.7, 143.7, 151.8, 155.6, 159.6, 163.3, 172.1. Anal. calcd. for C<sub>18</sub>H<sub>12</sub>ClN<sub>3</sub>O<sub>3</sub>S: C, 56.03; H, 3.13; N, 10.89; S, 8.31. Found: C, 56.11; H, 3.17; N, 10.82; S, 8.34.

## 5-[4-(6-Bromo-2-oxo-2*H*-chromen-3-yl)thiazol-2-yl]-7-methyl-5,6diazaspiro[2.4]hept-6-en-4-one (4d)

Crystalline solid, yield 40%, mp 237–240 °C. IR (KBr),  $\nu$  (cm<sup>-1</sup>), 1720 (lactone, –C=O), 1635 (amide, –C=O), 1601 (–C=N). <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>),  $\delta$  ppm: 1.77 (ABq, 2H, spiro CH<sub>2</sub>–CH<sub>2</sub>), 2.09 (ABq, 2H, spiro CH<sub>2</sub>–CH<sub>2</sub>), 2.37 (s, 3H). 7.42 (d, J = 8.4 Hz, 1H, Ar-H), 7.76–7.89 (m, 2H, Ar-H), 8.28 (s, 1H, thiazole), 8.72 (s, 1H, C<sub>4</sub> of coumarin). ESI-MS: m/z 430 (M<sup>+</sup> + 1). Anal. calcd. for C<sub>18</sub>H<sub>12</sub>BrN<sub>3</sub>O<sub>3</sub>S: C, 50.25; H, 2.81; N, 9.77; S, 7.45. Found: C, 50.17; H, 2.77; N, 9.83; S, 7.41.

## 5-[4-(6,8-Dibromo-2-oxo-2*H*-chromen-3-yl)thiazol-2-yl]-7-methyl-5,6diazaspiro[2.4]hept-6-en-4-one (4e)

Crystalline solid, yield 40%, mp 240–243 °C. IR (KBr),  $\nu$  (cm<sup>-1</sup>), 1732 (lactone, -C=O), 1635 (amide, -C=O), 1615 (-C=N). <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>),  $\delta$  ppm: 1.85 (ABq, 2H, spiro CH<sub>2</sub>–CH<sub>2</sub>), 1.95 (ABq, 2H, spiro CH<sub>2</sub>–CH<sub>2</sub>), 2.11 (s, 3H, C<sub>7</sub> methyl of spiro system), 7.68 (m, 1H, Ar-H), 7.92 (s, 1H, aromatic). 8.32 (s, 1H, C<sub>5</sub> thiazole), 8.68 (s, 1H, C<sub>4</sub> of coumarin). Anal. calcd. for C<sub>18</sub>H<sub>11</sub>Br<sub>2</sub>N<sub>3</sub>O<sub>3</sub>S: C, 42.46; H, 2.81; N, 8.25; S, 6.30. Found: C, 42.39; H, 2.16; N, 8.21; S, 6.26.

## 5-[4-(8-Methoxy-2-oxo-2*H*-chromen-3-yl)thiazol-2-yl]-7-methyl-5,6diazaspiro[2.4]hept-6-en-4-one (4f)

Crystalline solid, yield 55%, mp 228–230 °C. IR (KBr),  $\nu$  (cm<sup>-1</sup>), 1723 (lactone, -C=O), 1625 (amide, -C=O), 1607 (-C=N). <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>),  $\delta$  ppm: 1.77 (ABq, 2H, spiro CH<sub>2</sub>–CH<sub>2</sub>), 2.06 (ABq, 2H, spiro CH<sub>2</sub>–CH<sub>2</sub>), 2.27 (s, 3H, C<sub>7</sub> methyl of spiro system), 3.94 (s, 3H, CH<sub>3</sub> of methoxy group), 7.27–7.53 (m, 3H, Ar-H), 8.19 (s, 1H, C<sub>5</sub> thiazole), 8.69 (s, 1H, C<sub>4</sub> of coumarin). Anal. calcd. for C<sub>19</sub>H<sub>15</sub>N<sub>3</sub>O<sub>4</sub>S: C, 59.83; H, 3.96; N, 11.02; S, 8.41. Found: C, 59.87; H, 3.93; N, 11.10; S, 8.44.

## 5-[4-(5,6-Benzo-2-oxo-2*H*-chromen-3-yl)thiazol-2-yl]-7-methyl-5,6diazaspiro[2.4]hept-6-en-4-one (4g)

Crystalline solid, yield 65%, mp 245–248 °C. IR (KBr),  $\nu$  (cm<sup>-1</sup>), 1716 (lactone, –C=O), 1627 (amide, –C=O), 1610 (–C=N). <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>),  $\delta$  ppm: 1.85 (ABq, 2H, spiro CH<sub>2</sub>–CH<sub>2</sub>), 1.97 (ABq, 2H, spiro CH<sub>2</sub>–CH<sub>2</sub>), 2.13 (s, 3H, C<sub>7</sub> methyl of spiro system), 7.50–7.73 (m, 3H, Ar-H), 7.92 (d, J=8.1 Hz, 1H, aromatic), 7.99 (d, J=9 Hz, 1H, aromatic), 8.36 (s, 1H, C<sub>5</sub> thiazole), 8.56 (d, J=6 Hz, 1H, aromatic), 9.55 (s, 1H, C<sub>4</sub> of coumarin). ESI-MS: m/z 402 [M<sup>+</sup>+1], 421 [M<sup>+</sup> + Na]. Anal. calcd. for C<sub>22</sub>H<sub>15</sub>N<sub>3</sub>O<sub>3</sub>S: C, 65.82; H, 3.77; N, 10.47, S, 7.99. Found: C, 65.87; H, 3.75; N, 10.44; S, 7.96.

#### **General Procedure for Compound 5**

A mixture of arylbromomethyl ketone/3-(2-bromoacetyl)chromen-2-one (10 mmol) and thiosemicarbazide (10 mmol) was taken in anhydrous ethanol (25 ml) and stirred for 2 h at 20-25 °C. The crystalline solid thus obtained was filtered and washed with ethanol (5 ml). Compound **5** was isolated in 88–92% yield.

#### 2-Oxo-2-(phenyl)ethylhydrazine Carbimidozothioatehydrobromide (5a)

Yield 92%, mp 205–207 °C. IR (KBr),  $\nu$  (cm<sup>-1</sup>), 1607 (br, -C=N), 1660 (-C=O). <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  ppm: 3.61–3.71 (ABq, 2H, SCH<sub>2</sub>–), 7.33–7.54 (m, 6H, Ar-H and 1H of -NH, D<sub>2</sub>O exchangeable), 7.71 (m, 2H,  $-NH_2$ , D<sub>2</sub>O exchangeable), 9.39 (s, 1H, -NH, D<sub>2</sub>O exchangeable), 10.07 (s, 1H, -C=NH, D<sub>2</sub>O exchangeable). <sup>13</sup>C NMR spectrum (300 MHz, DMSO- $d_6$ )  $\delta$  ppm: 41.4, 97.1,

127.2 (for two carbons), 129.6 (for two carbons), 130, 139.5, 171.2. ESI-MS: m/z 210 (M<sup>+</sup>+1). Anal. calcd. for C<sub>9</sub>H<sub>11</sub>N<sub>3</sub>OS · HBr: C, 37.25; H, 4.17; N, 14.48; S, 11.05. Found: C, 37.16; H, 4.09; N, 14.39; S, 11.08.

## 2-Oxo-2-(2-oxo-2*H*-chromen-3-yl)ethylhydrazinecarbimidothioate Hydrobromide (5b)

Yield 92%, mp 218–220 °C. IR (KBr),  $\nu$  (cm<sup>-1</sup>) 1607 (C=C), 1625 (–C=N), 1654 (–C=O), 1711 (lactone, –C=O). <sup>1</sup>H NMR spectrum (DMSO-*d*<sub>6</sub>)  $\delta$  ppm: 3.40 (d, *J*=12 Hz, 1H, SCH<sub>2</sub>-), 3.97 (d, *J*=12 Hz, 1H, SCH<sub>2</sub>-), 5.14 (s, 2H, –NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 7.37–7.49 (m, 2H, Ar-H), 7.62 (m, 1H, Ar-H), 7.83 (d, *J*=7.2 Hz, 1H, Ar-H), 8.09 (s, 1H, –NH, D<sub>2</sub>O exchangeable), 8.41 (s, 1H, C<sub>4</sub> of coumarin), 9.17 (s, 1H, –C=NH, D<sub>2</sub>O exchangeable), 9.75 (s, 1H, C=NH<sub>2</sub><sup>+</sup>, D<sub>2</sub>O exchangeable). ESI-MS: *m*/*z* 278 (M<sup>+</sup> + 1). Anal. calcd. for C<sub>12</sub>H<sub>11</sub>N<sub>3</sub>O<sub>3</sub>S · HBr: C, 40.24; H, 3.38; N, 11.73; S, 8.95. Found: C, 40.47; H, 3.04; N, 11.67; S, 8.89.

## 2-Oxo-2-(6-chloro-2-oxo-2*H*-chromen-3yl)ethylhydrazinecarbimidothioate Hydrobromide (5c)

Yield 88%, mp 205–207 °C. IR (KBr),  $\nu$  (cm<sup>-1</sup>) 1593 (C=C), 1625 (–C=N), 1651 (ketone –C=O), 1712 (lactone–C=O). <sup>1</sup>H NMR spectrum (DMSO- $d_6$ )  $\delta$ ppm: 3.39 (d, J = 12 Hz, 1H, SCH<sub>2</sub>-), 3.97 (d, J = 12 Hz, 1H, SCH<sub>2</sub>–), 5.14 (s, 2H, –NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 7.52 (d, J = 9 Hz, 1H, Ar-H), 7.73 (dd, 1H, Ar-H), 8.09–8.16 (m, 2H, aromatic and –NH, D<sub>2</sub>O exchangeable), 8.39 (s, 1H, C<sub>4</sub> of coumarin), 9.17 (s, 1H, –C=NH, D<sub>2</sub>O exchangeable), 9.77 (s, 1H, C=NH<sub>2</sub><sup>+</sup>, exchangeable). ESI-MS: m/z 312 (M<sup>+</sup> + 1). Anal. calcd. for C<sub>12</sub>H<sub>10</sub>ClN<sub>3</sub>O<sub>3</sub>S · HBr: C, 36.72; H, 2.82; N, 9.03; S, 8.16. Found: C, 36.75; H, 2.84; N, 9.01; S, 8.11.

## 2-Oxo-2-(6-bromo-2-oxo-2*H*-chromen-3yl)ethylhydrazinecarbimidothioate Hydrobromide (5d)

Yield 90%, mp 210–212 °C. IR (KBr),  $\nu$  (cm<sup>-1</sup>) 1594 (C=C), 1622 (–C=N), 1654 (–C=O), 1713 (lactone–C=O). <sup>1</sup>H NMR spectrum (DMSO- $d_6$ )  $\delta$  ppm: 3.42 (d, J=15 Hz, 1H, SCH<sub>2</sub>-), 3.97 (d, J=12 Hz, 1H, SCH<sub>2</sub>-), 5.14 (s, 2H, –NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 7.47 (d, J=9 Hz, 1H, Ar-H), 7.85 (d, J=9 Hz, 1H, Ar-H), 8.14 (s, 1H, –NH, D<sub>2</sub>O exchangeable), 8.23 (s, 1H, Ar-H), 8.38 (s, 1H, C<sub>4</sub> of coumarin), 9.18 (s, 1H, –C=NH, D<sub>2</sub>O exchangeable), 9.75 (s, 1H, C=NH<sub>2</sub><sup>+</sup>, exchangeable). <sup>13</sup>C NMR spectrum (DMSO- $d_6$ )  $\delta$  ppm: 37.4, 93.7, 117.0, 119.1, 121.4, 127.0, 132.2, 135.9, 143.1, 153.6, 159.0, 170.4. Anal. calcd. for C<sub>12</sub>H<sub>10</sub>BrN<sub>3</sub>O<sub>3</sub>S · HBr: C, 32.97; H, 2.54; N, 9.61; S, 7.33. Found: C, 32.91; H, 2.49; N, 9.67; S, 7.39.

## 2-Oxo-2-(6,8-dibromo-2-oxo-2*H*chromen-3-yl)ethylhydrazinecarbimidothioate Hydrobromide (5e)

Yield 88%, mp 210–212 °C. IR (KBr),  $\nu$  (cm<sup>-1</sup>) 1603 (C=C), 1627 (–C=N), 1660 (–C=O), 1727 (lactone–C=O). <sup>1</sup>H NMR spectrum (DMSO-*d*<sub>6</sub>)  $\delta$  ppm: 3.38 (d, *J*=12 Hz, 1H, SCH<sub>2</sub>–), 3.99 (d, *J*=12 Hz, 1H, SCH<sub>2</sub>–), 5.15 (s, 2H, –NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 8.19 (s, 1H, –NH, D<sub>2</sub>O exchangeable), 8.23 (d, *J*=3 Hz, 1H, Ar-H), 8.26 (d, J = 3 Hz, 1H, Ar-H), 8.37 (s, 1H, C<sub>4</sub> of coumarin), 9.21 (s, 1H, -C=NH, D<sub>2</sub>O exchangeable), 9.79 (s, 1H,  $C=NH_2^+$ , exchangeable). Anal. calcd. for C<sub>12</sub>H<sub>9</sub>Br<sub>2</sub>N<sub>3</sub>O<sub>3</sub>S · HBr: C, 27.93; H, 1.95; N, 8.14; S, 6.21. Found: C, 27.88; H, 1.93; N, 8.14; S, 6.21.

## 2-Oxo-2-(8-methoxy-2-oxo-2*H*chromen-3-yl)ethylhydrazinecarbimidothioate Hydrobromide (5f)

Yield 90%, mp 228–230 °C. IR (KBr),  $\nu$  (cm<sup>-1</sup>) 1611 (C=C), 1623 (–C=N), 1656 (ketone–C=O), 1714 (lactone–C=O). <sup>1</sup>H NMR spectrum (DMSO- $d_6$ )  $\delta$ ppm: 3.37 (d, J = 12 Hz, 1H, SCH<sub>2</sub>–), 3.93 (s, 3H, methoxy group), 3.97 (d, J = 12 Hz, 1H, SCH<sub>2</sub>–), 5.13 (s, 2H, –NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 7.29–7.46 (m, 3H, Ar-H), 8.08 (s, 1H, –NH, D<sub>2</sub>O exchangeable), 8.37 (s, 1H, C<sub>4</sub> of coumarin), 9.17 (s, 1H, –C=NH, D<sub>2</sub>O exchangeable), 9.74 (s, 1H, C=NH<sub>2</sub><sup>+</sup>, exchangeable). Anal. calcd. for C<sub>13</sub>H<sub>13</sub>N<sub>3</sub>O<sub>3</sub>S · HBr: C, 40.22; H, 3.53; N, 10.82; S, 8.26. Found: C, 40.12; H, 3.57; N, 10.77; S, 8.21.

## 2-Oxo-2-(5,6-benzo-2-oxo-2*H*-chromen-3yl)ethylhydrazinecarbimidothioate Hydrobromide (5g)

Yield 91%, mp 240–242 °C. IR (KBr),  $\nu$  (cm<sup>-1</sup>) 1606 (–C=N), 1659 (ketone –C=O), 1704 (lactone–C=O), 3345 (–NH). <sup>1</sup>H NMR spectrum (DMSO- $d_6$ )  $\delta$  ppm: 3.46 (d, J = 12 Hz, 1H, SCH<sub>2</sub>–), 4.06 (d, J = 12 Hz, 1H, SCH<sub>2</sub>-), 5.20 (s, 2H, –NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 7.65–7.84 (m, 3H, Ar-H), 8.12 (d, J = 9 Hz, 1H, Ar-H), 8.23 (s, 1H, –NH, D<sub>2</sub>O exchangeable), 8.31 (d, J = 9 Hz, 1H, Ar-H), 8.51 (d, J = 9 Hz, 1H, Ar-H), 9.10 (s, 1H, C<sub>4</sub> of coumarin), 9.22 (s, 1H, –C=NH, D<sub>2</sub>O exchangeable), 9.80 (s, 1H, C=NH<sub>2</sub><sup>+</sup>, exchangeable). Anal. calcd. for C<sub>16</sub>H<sub>13</sub>N<sub>3</sub>O<sub>3</sub>S · HBr: C, 47.07; H, 3.46; N, 10.29; S, 7.85. Found: C, 47.01; H, 3.49; N, 10.25; S, 7.81.

## General Procedure for Cyclization of 5 with POCl<sub>3</sub>, Leading to 4

2-Acetylbutyrolactone (5 mmol) was added to a suspension of 2-oxo-2-(aryl-2-oxo-2*H*-chromen-3-yl)ethylhydrazinecarbimidathioate hydrobromide 5 (5 mmol) in POCl<sub>3</sub> (7.2 ml) at 20–25 °C. The suspension was heated to 95 °C, and the stirring continued for 2 h. The reaction mixture was cooled to 20 °C, and anhydrous ethanol was added (20 ml). The product was filtered and purified by chromatography on silica gel (eluent, ethyl acetate/hexane 1.5:8.5) to give **4** in 40–50% yield. The compounds obtained by method 2 were found to be identical to those obtained by method 1. This was confirmed by their mixed mp measurements, co-TLC, and spectral data.

#### CONCLUSION

In summary, we have described an elegant and simple methodology for the synthesis of 7-methyl-5-[4-(aryl/2-oxo-2*H*-chromen-3-yl)thiazol-2-yl]-5,6-diaza-spiro[2.4]hept-6-en-4-one. The synthesis involves a one-step selective ring closure leading to simultaneous formation of three rings at a time. Advantages of the present protocol are (i) ready availability of the starting materials and mild reaction

conditions, (ii) environmentally safe, (iii) simple operational procedure, and (iv) good yields. The protocol is certainly superior to classical methods available for the preparation individual cyclic systems and other multistep syntheses. This reaction can be extended to other heterocyclic  $\alpha$ -halo ketones.

## ACKNOWLEDGMENT

The authors thank Dr. G. K. A. S. S. Narayan, APL Research Centre, Hyderabad, India, for his help in molecular design.

#### REFERENCES

- Dodson, R. M.; King, L. C. The reaction of ketones with halogens and thiourea. J. Am. Chem. Soc. 1945, 67, 2242–2243; (b) King, L. C.; Hlavacek, R. J. The reaction of ketones with iodine and thiourea. J. Am. Chem. Soc. 1950, 72, 3722–3725.
- Vardhan, V. A.; Rajeswar Rao, V. Photoalogenation of 3-acetylcoumarins: Facile synthesis of 3-(2-amino-4-thiazolyl)coumarins and their coversion into 3-(2,5-dimethylpyrrol-1-yl)thiazol-4-yl)coumarins. *Ind. J. Chem.* 1997, 36B, 1085.
- Harode, R.; Sharma, T. C. Synthesis of 2-(3-methylpyrazol-5-one-1-yl)-4-ylthiazoles. J. Indian Chem. Soc. 1989, 66, 282–284.
- Srimanth, K.; Rajeswar Rao, V.; Krishna, D. R. Synthesis and evaluation of anticancer activity of some imidazothiazolyl, imidazobenzothiazolyl, and dihydroimidazothiazolyl coumarins. Arzneim. Forsch. Drug Res. 2002, 52(5), 388–392; (b) Rajeswar Rao, V.; Srimanth, V. A facile one-step synthesis of 3-[2-(3,5-dimethyl-1H-pyrazol-1-yl)-4-thiazolyl]-2H-1-benzopyran-2-ones under solvent-free conditions. J. Chem. Res. 2002, 5, 420–421; (c) Rajeswar Rao, V.; Vijaya Kumar, P. Facile one-pot synthesis of 3-{2-[5-hydroxy-4-(2-hydroxy-ethyl)-3-methyl-pyrazol-1-yl]-thiazol-4-yl}-chromen-2-ones via a three-component reaction. Synth. Commun. 2006, 36, 2157–2161; (d) Rajeswar Rao, V.; Seethalakshmi, T.; Vijaya Kumar, P.; Kalianman, P. N-[4-(2-Oxo-2H-chromen-3-yl)-3-phenyl-3H-thiazol-2-ylidene]anilinium bromide methanol solvate. Acta Cryst. E 2006, E62, 3771–3773; (e) Rajeswar Rao, V.; Guravaiah, N. A facile one-pot synthesis of 2-pyrazolyl-4-aryl-thiazoles in a three-component reaction. J. Chem. Res. 2008, 4, 195.
- 5. Koelsch, C. F. Bromination of 3-acetocoumarin. J. Am. Chem. Soc. 1950, 72, 2993-2995.