Tetrahedron 64 (2008) 8010-8015

Contents lists available at ScienceDirect

Tetrahedron

journal homepage: www.elsevier.com/locate/tet

Regioselective synthesis of 6-halomethyl-5,6-dihydro-4*H*-1,2-oxazines based on cyclizations of arylalkenyl-oximes

Vahuni Karapetyan ^{a,b}, Satenik Mkrtchyan ^{a,b}, Tung T. Dang ^a, Alexander Villinger ^a, Helmut Reinke ^a, Peter Langer ^{a,c,*}

^a Institut für Chemie, Universität Rostock, Albert-Einstein-Strasse 3a, 18059 Rostock, Germany ^b Faculty of Chemistry, Yerevan State University, Alex Manoogian 1, 0025 Yerevan, Armenia ^c Leibniz-Institut für Katalyse e. V. an der Universität Rostock, Albert-Einstein-Strasse 29a, 18059 Rostock, Germany

ARTICLE INFO

Article history: Received 6 March 2008 Received in revised form 26 May 2008 Accepted 29 May 2008 Available online 3 June 2008

ABSTRACT

6-Iodo- and 6-bromomethyl-5,6-dihydro-4*H*-1,2-oxazines were prepared by condensation of dilithiated acetophenone-oximes with allylbromide and subsequent regioselective iodine- or NBS-mediated cyclization.

© 2008 Elsevier Ltd. All rights reserved.

1. Introduction

1,2-Oxazines are of pharmacological relevance and represent useful synthetic building blocks. They have been used, for instance, as intermediates during the synthesis of glycosidase inhibitor analogues¹ and functionalized pyrroles.² 1,2-Oxazines have been prepared, for example, by hetero-Diels-Alder reactions of alkenes with ene-nitroso compounds derived from α -haloximes³ and by hetero-Diels-Alder reactions of dienes with nitroso compounds.⁴ 1,2-Oxazines are also available by NBS-,⁵ diphenyldiselenide-.⁶ acid-⁷ and UV-mediated⁸ cvclization of alkenvl-substituted oximes. 1,2-Oxazines have also been prepared by base-mediated cyclizations of γ -chloroximes⁹ and γ -sulfonyloximes.¹⁰ Other synthetic approaches to 1,2-oxazines rely on Lewis acid-catalyzed reactions of allenoximes,¹¹ acid-catalyzed cyclization of cyclopropyloximes¹² and on cyclizations of γ -nitroketones.¹³ Recently, we have reported¹⁴ the synthesis of 1,2-oxazines by cyclization¹⁵ of oxime dianions with epibromohydrin. Herein, we report what are, to the best of our knowledge, the first syntheses of 6-iodomethyl-5,6-dihydro-4H-1,2-oxazines by condensation of oxime dianions with allylbromide and subsequent O-regioselective iodine-mediated cyclization.

2. Results and discussion

The reaction of the dianions of oximes **1a–k**, generated by means of *n*-BuLi (2.5 equiv), afforded the arylalkenyl-oximes **3a–k** in good yields (Scheme 1, Table 1). The reaction of the latter with iodine

* Corresponding author. Fax: +49 381 4986412.

E-mail address: peter.langer@uni-rostock.de (P. Langer).

afforded the 6-iodomethyl-5,6-dihydro-4*H*-1,2-oxazines **4a**-**k** in moderate to excellent yields. The best yields were obtained when the reaction was carried out in dichloromethane using a saturated aqueous solution of sodium bicarbonate as the base. The reaction of **3e**,**f**,**j**,**k** with *N*-bromosuccinimide (NBS) afforded the 6-bromomethyl-5,6-dihydro-4*H*-1,2-oxazines **4I**-**o**. The tricyclic oxazine **4p** was prepared in high yield from tetralone (**1I**) (Scheme 2). The structure of all products was established by spectroscopic methods. The structure analyses (Figs. 1–3).¹⁶ Products **4j**,**k** and **4n**-**p** were isolated as 1:1 mixtures of diastereomers. In case of **4j**, one of the two diastereomers could be separated by crystallization (Fig. 3).

Scheme 1. Synthesis of 1,2-oxazines **4a–o**. Reagents and conditions: (i) (1) **1** (1.0 equiv), *n*-BuLi (2.5 equiv), THF, 1 h, $-78 \degree C$, then 10 min, 20 °C, (2) **2** (2.0 equiv), $-78 \rightarrow 20 \degree C$, 16 h; (ii) **4a–k**: I₂ (2.0 equiv), CH₂Cl₂, NaHCO₃ (saturated aqueous solution), 20 °C, 12 h, **4l–o**: NBS (1.0 equiv), CH₂Cl₂, 20 °C, 2 h.

The regioselectivity of cyclization requires some discussion. Oximes are ambident nucleophiles, which can react with electrophiles either at the oxygen or at the nitrogen atom. Grigg and







^{0040-4020/\$ –} see front matter \odot 2008 Elsevier Ltd. All rights reserved. doi:10.1016/j.tet.2008.05.116

 Table 1

 Products and vields

1,3	4	Х	R	Ar	Yield %	
					3 ^a	4 ^a
a	a	Ι	Н	Ph	85	95
b	b	Ι	Н	4-MeC ₆ H ₅	69	83
с	с	Ι	Н	3-(MeO)C ₆ H ₅	68	66
d	d	Ι	Н	4-(MeO)C ₆ H ₅	71	67
e	e	Ι	Н	2-(EtO)C ₆ H ₅	64	96
f	f	Ι	Н	4-(EtO)C ₆ H ₅	69	61
g	g	Ι	Н	4-FC ₆ H ₅	67	81
h	h	Ι	Н	4-ClC ₆ H ₅	60	52
i	i	Ι	Н	1-Naphthyl	65	66
j	j	Ι	Me	Ph	63	50 ^b
k	k	I	Me	4-(MeO)C ₆ H ₅	60	43 ^b
e	1	Br	Н	2-(EtO)C ₆ H ₅	64	57
f	m	Br	Н	4-(EtO)C ₆ H ₅	69	87
j	n	Br	Me	Ph	63	73 ^b
k	0	Br	Me	4-(MeO)C ₆ H ₅	60	25 ^b

^a Yields of isolated product.

^b dr=1:1.



Scheme 2. Synthesis of 1,2-oxazine **4p**. Reagents and conditions: (i) (1) **1** (1.0 equiv), *n*-BuLi (2.5 equiv), THF, 1 h, -78 °C, then 10 min, 20 °C, (2) **2** (2.0 equiv), $-78 \rightarrow 20$ °C, 16 h; (ii) I₂ (2.0 equiv), CH₂Cl₂, NaHCO₃ (saturated aqueous solution), 20 °C, 12 h, dr=1:1.

co-workers showed that the regioselectivity is controlled by the E/Z-configuration of the oxime and by the rate of E/Z-isomerization with respect to the N- or O-nucleophilic attack.^{17–19} The intramolecular reaction of oximes with halonium ions has been reported to result in N-alkylation and formation of nitrones. For example, treatment of a CH₂Cl₂ solution of alkenyl-oxime **5** with iodine and



Figure 1. ORTEP plot of 4d (50% probability level).



Figure 2. ORTEP plot of 4f (50% probability level).



Figure 3. ORTEP plot of 4j (50% probability level).

anhydrous potassium carbonate quantitatively afforded nitrone **6**, which was trapped by a subsequent [3+2] cycloaddition (Scheme 3).¹⁹ Similar results were obtained for the oxime of ethyl 2-homoallyl-cyclohexanone-2-carboxylate. The N-regioselectivity was explained by a rapid $Z \rightarrow E$ isomerization and subsequent attack of the nitrogen atom onto the iodonium ion. The reaction of **5** with *N*-bromosuccinimide (NBS) was reported to give a 2:1 mixture of nitrone and 1,2-oxazine, which reflects the E/Z ratio of **5**.⁵ In this reaction, the E/Z isomerization was slow compared to the N- and O-cyclization. Similar results have been reported for diphenyl diselenide-mediated cyclizations.⁶



Scheme 3. Synthesis of nitrone **6** by Grigg and co-workers (Ref. 19). Reagents and conditions: (i) l_2 (2.0 equiv), CH_2Cl_2 , K_2CO_3 (anhydrous), 25 °C, 12 h.

In contrast to **5**, the aryl-substituted oximes **3a–1** contain an *E*-configured C—N group, due to the steric effect of the aryl group.²⁰ The excellent O-regioselectivity of the formation of 1,2-oxazines **4a–p** can be explained by the assumption that the $E \rightarrow Z$ isomerization is slow compared to the O-regioselective 1,2-oxazine formation.

3. Conclusions

In conclusion, 6-iodo- and 6-bromomethyl-5,6-dihydro-4*H*-1,2oxazines were prepared by condensation of dilithiated acetophenone-oximes with allylbromide and subsequent regioselective iodine- or NBS-mediated cyclization. The results reported herein show that oxazines are available from alkenyl-oximes containing sterically demanding substituents.

4. Experimental section

4.1. General comments

All solvents were dried by standard methods and all reactions were carried out under an inert atmosphere. For ¹H and ¹³C NMR spectra the deuterated solvents indicated were used. Mass spectrometric data (MS) were obtained by electron ionization (EI, 70 eV), chemical ionization (CI, H_2O) or electrospray ionization (ESI). For preparative scale chromatography, silica gel (60–200 mesh) was used. Melting points are uncorrected.

4.2. General procedure for the preparation of oximes 3

To a THF solution (20 mL) of oxime **1** (2.0 mmol) was added *n*butyllithium (5.0 mmol, 2.5 M) at -78 °C. After stirring for 1 h at -78 °C, the mixture was warmed to 20 °C and stirred for 10 min. Subsequently, allylbromide (0.484 g, 4.0 mmol) was added at -78 °C. After warming of the mixture to 20 °C for 16 h, a saturated aqueous solution of NH₄Cl (30 mL) was added. The organic and the aqueous layer were separated and the latter was extracted with ethyl acetate (3×50 mL). The combined organic layers were dried (Na₂SO₄), filtered and the solvent of the filtrate was removed in vacuo. The residue was purified by column chromatography (silica gel, *n*-heptane/EtOAc=5:1).

4.3. General procedure for the synthesis of 1,2-oxazines 4a-k and 4p

To a CH₂Cl₂ solution (15 mL) of **3a–l** (0.81 mmol) and l₂ (0.406 g, 1.6 mmol) was added a saturated aqueous solution of NaHCO₃ (16 mL) and the solution was stirred for 12 h at 20 °C. The excess of iodine was removed by addition of a saturated aqueous solution of Na₂SO₃ (40 mL). The organic and the aqueous layer were separated and the latter was extracted with ethyl acetate (3×50 mL). The combined organic layers were dried (Na₂SO₄), filtered and the solvent of the filtrate was removed in vacuo. The residue was purified by column chromatography (silica gel, *n*-heptane \rightarrow *n*-heptane/EtOAc=4:1).

4.4. General procedure for the synthesis of 1,2-oxazines 41-o

To a CH₂Cl₂ solution (10 mL) of **3e**,**f**,**j**,**k** (2.0 mmol) was portionwise added NBS (0.356 g, 2.0 mmol) over 15 min at 0 °C. The resultant solution was stirred for 2 h at room temperature. The residue was purified by column chromatography (silica gel, *n*-heptane \rightarrow *n*-heptane/EtOAc=4:1).

4.4.1. 6-Iodomethyl-3-phenyl-5,6-dihydro-4H-[1,2]oxazine (4a)

Starting with 1-phenyl-pent-4-en-1-one oxime 3a (0.141 g, 0.81 mmol), I₂ (0.412 g, 1.62 mmol) and a saturated aqueous solution of NaHCO₃ (8.1 mL) in CH₂Cl₂ (14 mL), 4a was isolated as a brownish solid (0.232 g, 95%); mp 126–128 °C. ¹H NMR (300 MHz, CDCl₃): δ =1.86 (m, 1H, CHCH₂), 2.34 (m, 1H, CHCH₂), 2.68 (m, 2H, CCH₂), 3.27 (dd, ${}^{2}J=10.3$ Hz, ${}^{3}J=7.3$ Hz, 1H, CHCH₂I), 3.42 (dd, ²*I*=10.3 Hz, ³*I*=5.0 Hz, 1H, CHCH₂I), 3.85 (m, 1H, OCHCH₂), 7.38 (m, 3H, CH_{Ar}), 7.68 (m, 2H, CH_{Ar}). ¹³C NMR (75.5 MHz, CDCl₃): δ =5.3 (CH₂I), 21.6, 24.2 (CHCH₂CH₂C), 74.1 (CHO), 125.4 (CH_{Ar}), 128.5 (CH_{Ar}) , 129.7 (CH_{Ar}) , 135.1 (C), 154.8 (C). IR (ATR, cm^{-1}) : $\tilde{\nu}$ =3039 (br, cm^{-1}) w), 2959 (w), 2905 (br, w), 2853 (w), 1589 (w), 1563 (w), 1490 (w), 1443 (w), 1404 (w), 1378 (w), 1330 (w), 1296 (w), 1260 (w), 1231 (w), 1195 (m), 1161 (w), 1086 (m), 1012 (m), 997 (m), 982 (m), 913 (m), 799 (m), 750 (s), 685 (s), 603 (m). MS (GC/MS, 70 eV): m/z (%)=301 (M⁺, 100), 207 (6), 174 (17), 156 (48), 144 (30), 128 (38), 118 (59), 104 (51), 77 (70). HRMS (EI): calcd for C₁₁H₁₂NOI (M⁺): 300.99581, found: 300.995322.

4.4.2. 6-Iodomethyl-3-p-tolyl-5,6-dihydro-4H-[1,2]oxazine (4b)

Starting with 1-(3-tolyl)pent-4-en-1-one oxime **3b** (0.342 g, 1.80 mmol), I₂ (0.914 g, 3.60 mmol) and a saturated aqueous solution of NaHCO₃ (18 mL) in CH₂Cl₂ (30 mL), **4b** was isolated as a colourless solid (0.471 g, 83%); mp 120–122 °C. ¹H NMR (250 MHz, CDCl₃): δ =1.85 (m, 1H, CH₂), 2.32 (m, 1H, CHCH₂), 2.36 (s, 3H, C_{Ar}CH₃), 2.68 (m, 2H, CCH₂), 3.25 (dd, ²*J*=10.5 Hz, ³*J*=7.3 Hz, 1H, CHCH₂I), 3.41 (dd, ²*J*=10.5 Hz, ³*J*=5.0 Hz, 1H, CHCH₂I), 3.84 (m, 1H, OCHCH₂), 7.18 (d, ³*J*=8.1 Hz, 2H, CH_{Ar}), 7.57 (d, ³*J*=8.1 Hz, 2H, CH_{Ar}). ¹³C NMR (75.5 MHz, CDCl₃): δ =5.5 (CH₂I), 21.3 (C_{Ar}CH₃), 21.6, 24.3 (CHCH₂CH₂C), 74.0 (CHO), 125.3 (CH_{Ar}), 129.2 (CH_{Ar}), 132.3 (C), 139.8

(C), 154.8 (C). IR (KBr, cm⁻¹): $\tilde{\nu}$ =3431 (br, w), 3034 (m), 2935 (br, w), 2910 (w), 1611 (w), 1592 (w), 1510 (w), 1418 (w), 1407 (m), 1380 (m), 1335 (s), 1297 (m), 1233 (m), 1198 (s), 1163 (w), 1111 (w), 1088 (w), 1062 (w), 1013 (s), 936 (w), 915 (s), 812 (s), 760 (w), 710 (w). MS (GC/MS, 70 eV): m/z (%)=315 (M⁺, 100), 188 (9), 174 (20), 143 (14), 132 (32), 117 (33), 105 (9), 91 (40), 77 (7), 65 (17). HRMS (EI): calcd for C₁₂H₁₄NOI (M⁺): 315.01146, found: 315.011582.

4.4.3. 6-Iodomethyl-3-(3-methoxyphenyl)-5,6-dihydro-4H-[1,2]oxazine (**4c**)

Starting with 1-(4-methoxyphenyl)pent-4-en-1-one oxime **3c** (0.205 g, 1 mmol), I₂ (0.508 g, 2 mmol) and a saturated aqueous solution of NaHCO₃ (5 mL) in CH₂Cl₂ (10 mL), **4c** was isolated as a brownish oil (0.219 g, 66%). ¹H NMR (300 MHz, CDCl₃): δ =1.76 (m, 1H, CH₂), 2.22 (m, 1H, CH₂), 2.55 (m, 2H, CH₂), 3.19, 3.35 (t, 2H, CH₂), 3.75 (m, 1H, CH), 3.80 (s, 3H, OCH₃), 7.12–7.25 (m, 3H, Ar), 7.85 (s, 1H, Ar). ¹³C NMR (75 MHz, CDCl₃): δ =6.7, 22.0, 24.5 (3C, CH₂), 55.4 (C, CH₃), 74.1 (C, CH), 110.4, 116.3, 119.5, 129.6 (4C, ArCH), 136.6, 154.6 (2C, ArC), 159.6 (C, CN). IR (KBr, cm⁻¹): $\tilde{\nu}$ =3067 (w), 2994 (w), 2954 (w), 2932 (w), 2832 (w), 1597 (m), 1568 (m), 1425 (m), 1287 (m), 1234 (m), 1175 (m), 1038 (s), 924 (m), 823 (m), 779 (m), 688 (m). MS (EI, 70 eV): m/z (%)=332 (M⁺, 14), 331 (100), 204 (14), 190 (21), 187 (16), 186 (18), 178 (20), 133 (23). HRMS (EI, 70 eV): calcd for C₁₂H₁₄NO₂I (M⁺): 331.00637; found: 331.006082.

4.4.4. 6-Iodomethyl-3-(4-methoxyphenyl)-5,6-dihydro-4H-[1,2]oxazine (**4d**)

Starting with 1-(4-methoxyphenyl)pent-4-en-1-one oxime 3d (0.478 g, 2.33 mmol), I₂ (1.184 g, 4.66 mmol) and a saturated aqueous solution of NaHCO3 (23.3 mL) in CH2Cl2 (40.0 mL), 4d was isolated as a red solid (0.517 g, 67%); mp 140 °C. ¹H NMR (300 MHz, CDCl₃): δ =1.84 (m, 1H, CH₂), 2.31 (m, 1H, CHCH₂), 2.66 (m, 2H, CCH₂), 3.26 (dd, ²*J*=10.6 Hz, ³*J*=7.2 Hz, 1H, CHCH₂I), 3.41 (dd, ²*J*=10.6 Hz, ³*J*=5.0 Hz, 1H, CHCH₂I), 3.81 (s, 3H, OCH₃), 3.83 (m, 1H, OCHCH₂), 6.88 (d, ³*J*=9.0 Hz, 2H, CH_{Ar}), 7.63 (d, ³*J*=9.0 Hz, 2H, CH_{Ar}). ¹³C NMR (75.5 MHz, CDCl₃): δ =5.4 (CH₂I), 21.6 (CH₂), 24.3 (CH₂), 55.3 (OCH₃), 74.0 (CHO), 113.8, 126.8 (CH_{Ar}), 127.5, 154.5, 160.9 (C). IR (KBr, cm⁻¹): $\tilde{\nu}$ =2951 (w), 2905 (w), 2834 (w), 1611 (s), 1514 (s), 1462 (w), 1334 (w), 1295 (m), 1258 (s), 1199 (m), 1175 (s), 1030 (m), 1016 (m), 920 (m), 823 (s), 646 (w). MS (EI, 70 eV): m/z (%)=331 (M⁺, 100), 187 (11), 172 (11), 133 (26), 90 (8), 77 (11). HRMS (EI): calcd for C₁₂H₁₄NO₂I (M⁺): 331.00637, found: 331.006054. Anal. Calcd for C₁₂H₁₄NO₂I (331.15): C, 43.52; H, 4.26; N, 4.23. Found: C, 43.68; H, 4.30; N, 3.91.

4.4.5. 3-(2-Ethoxyphenyl)-6-iodomethyl-5,6-dihydro-4H-[1,2]oxazine (**4e**)

Starting with 1-(2-ethoxyphenyl)pent-4-en-1-one oxime 3e (0.438 g, 2.00 mmol), I₂ (1.016 g, 4.00 mmol) and a saturated aqueous solution of NaHCO₃ (20 mL) in CH₂Cl₂ (34 mL), 4e was isolated as a brownish viscous (0.656 g, 95%). ¹H NMR (250 MHz, CDCl₃): $\delta = 1.40$ (t, ³*J*=7.0 Hz, 3H, OCH₂CH₃), 1.84 (m, 1H, CHCH₂CH₂), 2.27 (m, 1H, CHCH₂CH₂), 2.69 (dd, ²J=8.3 Hz, ³J=5.5 Hz, 2H, CCH₂), 3.28 (dd, ²*J*=10.2 Hz, ³*J*=7.7 Hz, 1H, CHCH₂I), 3.44 (dd, ²*J*=10.2 Hz, ${}^{3}J$ =4.9 Hz, 1H, CHCH₂I), 3.96 (m, 1H, OCHCH₂), 4.05 (q, ${}^{3}J$ =7.0 Hz, 2H, OCH₂CH₃), 6.92 (m, 2H, CH_{Ar}), 7.32 (m, 2H, CH_{Ar}). ${}^{13}C$ NMR (62.9 MHz, CDCl₃): δ =5.8 (CH₂I), 14.8 (OCH₂CH₃), 24.1, 24.2 (CHCH₂CH₂C), 63.8 (OCH₂CH₃), 74.2 (CHO), 111.9 (CH_{Ar}), 120.6 (CH_{Ar}), 125.7 (C), 129.6 (CH_{Ar}), 130.4 (CH_{Ar}), 156.6, 158.5 (C). IR (ATR, cm⁻¹): $\tilde{\nu}$ =3061 (br, w), 2975 (w), 2927 (w), 2875 (w), 1600 (m), 1490 (m), 1446 (s), 1391 (m), 1290 (m), 1237 (s), 1161 (m), 1122 (s), 1039 (s), 1010 (m), 892 (s), 802 (w), 750 (s), 681 (m), 640 (w), 604 (m). MS (EI, 70 eV): m/z (%)=345 (M⁺, 9), 313 (54), 256 (2), 204 (27), 185 (28), 158 (22), 145 (100), 128 (36), 119 (24), 91 (22), 77 (13). HRMS (EI): calcd for C₁₃H₁₆NO₂I (M⁺): 345.02202, found: 345.022935.

4.4.6. 3-(4-Ethoxyphenyl)-6-iodomethyl-5,6-dihydro-4H-[1,2]oxazine (**4f**)

Starting with 1-(4-ethoxyphenyl)pent-4-en-1-one oxime 3f (0.438 g, 2.00 mmol), I_2 (1.016 g, 4.00 mmol) and a saturated aqueous solution of NaHCO3 (20 mL) in CH2Cl2 (34 mL), 4f was isolated as a brownish solid (0.352 g, 51%); mp 131-135 °C. ¹H NMR (250 MHz, CDCl₃): δ =1.41 (t, ³*J*=7.0 Hz, 3H, OCH₂CH₃), 1.84 (m, 1H, CHCH2CH2), 2.32 (m, 1H, CHCH2CH2), 2.65 (m, 2H, CCH2), 3.25 (dd, ²*J*=10.8 Hz, ³*J*=7.3 Hz, 1H, CHCH₂I), 3.41 (dd, ²*J*=10.8 Hz, ³*J*=5.1 Hz, 1H, CHCH₂I), 3.82 (m, 1H, OCHCH₂), 4.04 (q, ${}^{3}J$ =7.0 Hz, 2H, OCH₂CH₃), 6.87 (d, ${}^{3}J$ =9.0 Hz, 2H, CH_Ar), 7.61 (d, ${}^{3}J$ =9.0 Hz, 2H, CH_{Ar}). ¹³C NMR (62.9 MHz, CDCl₃): δ =5.5 (CH₂I), 14.7 (OCH₂CH₃), 21.2, 24.3 (CHCH₂CH₂C), 63.5 (OCH₂CH₃), 73.9 (CHO), 114.3 (CH_{Ar}), 126.7 (CH_{Ar}), 127.4 (C), 154.4 (C), 160.2 (C). IR (ATR, cm⁻¹): $\tilde{\nu}$ =3047 (w), 2976 (w), 2929 (w), 1608 (s), 1590 (m), 1510 (s), 1481 (m), 1444 (w), 1394 (m), 1334 (m), 1296 (w), 1253 (s), 1232 (m), 1196 (m), 1174 (m), 1116 (m), 1046 (m), 1015 (s), 984 (w), 916 (m), 817 (s), 759 (w), 662 (m), 553 (m). MS (EI, 70 eV): m/z (%)=345 (M⁺, 100), 256 (6), 201 (20), 172 (31), 147 (11), 119 (21), 97 (16), 69 (24). Anal. Calcd for C₁₃H₁₆INO₂ (345.176): C, 45.23; H, 4.67; N, 4.06. Found: C, 45.17; H, 4.58; N, 3.81. HRMS (EI): calcd for C₁₃H₁₆NO₂I (M⁺): 345.02202, found: 345.022376.

4.4.7. 3-(4-Fluorophenyl)-6-iodomethyl-5,6-dihydro-4H-[1,2]oxazine (**4g**)

Starting with 1-(4-fluorophenyl)pent-4-en-1-one oxime 3g (0.290 g, 1.50 mmol), I₂ (0.762 g, 3.00 mmol) and a saturated aqueous solution of NaHCO₃ (15 mL) in CH₂Cl₂ (25 mL), 4g was isolated as a brownish solid (0.388 g, 81%); mp 129-131 °C. ¹H NMR (250 MHz, CDCl₃): δ=1.86 (m, 1H, CHCH₂), 2.34 (m, 1H, CHCH₂), 2.65 (m, 2H, CCH₂), 3.27 (dd, ${}^{2}J$ =10.4 Hz, ${}^{3}J$ =7.2 Hz, 1H, CHCH₂I), 3.42 (dd, ²/=10.4 Hz, ³/=5.0 Hz, 1H, CHCH₂I), 3.84 (m, 1H, OCHCH₂), 7.05 $(m, 2H, CH_{Ar}), 7.67 (m, 2H, CH_{Ar}).$ ¹³C NMR (62.9 MHz, CDCl₃): $\delta = 5.2$ (CH₂I), 21.6, 24.1 (CHCH₂CH₂C), 74.1 (CHO), 115.5 (d, ²*J*=22.0 Hz, CHCHCF_{Ar}), 127.2 (d, ³*J*=8.5 Hz, CHCHCF_{Ar}), 131.3 (d, ⁴*J*=3.3 Hz, CCHCHCF_{Ar}), 153.8 (CN), 163.6 (d, ¹*J*=249.7 Hz, CHCF_{Ar}). IR (ATR, cm⁻¹): $\tilde{\nu}$ =3053 (w), 2933 (br, w), 2904 (w), 2872 (w), 1606 (m), 1508 (s), 1444 (w), 1405 (w), 1379 (w), 1331 (m), 1294 (w), 1232 (s), 1197 (s), 1158 (m), 1099 (m), 1061 (w), 1012 (m), 1002 (m), 986 (m), 913 (s), 852 (m), 829 (s), 785 (w), 758 (w), 644 (m), 552 (s). MS (EI, 70 eV): m/z (%)=319 (M⁺, 100), 192 (16), 174 (37), 162 (16), 148 (14), 136 (47), 121 (40), 95 (19), 83 (18). HRMS (EI): calcd for C₁₁H₁₁INOF (M⁺): 318.98639, found: 318.985435.

4.4.8. 3-(4-Chlorophenyl)-6-iodomethyl-5,6-dihydro-4H-[1,2]oxazine (**4h**)

Starting with 1-(4-chlorophenyl)pent-4-en-1-one oxime **3h** (0.209 g, 1.0 mmol), I₂ (0.508 g, 2.0 mmol) and a saturated aqueous solution of NaHCO₃ (5 mL) in CH₂Cl₂ (10 mL), **4h** was isolated as a brownish oil (0.172 g, 52%). ¹H NMR (300 MHz, CDCl₃): δ =1.82 (m, 1H, CH₂), 2.22 (m, 1H, CH₂), 2.65 (q, 2H, CH₂), 3.19 (t, 1H, CH₂), 3.31 (t, 1H, CH₂), 3.75 (m, 1H, CH), 7.25 (d, ³*J*=8.2 Hz, 1H, Ar), 7.46 (d, ³*J*=8.2 Hz, 1H, Ar). ¹³C NMR (75 MHz, CDCl₃): δ =5.4, 23.0, 26.0 (CH₂), 74.4 (CH), 127.7, 129.3 (4C, ArCH), 134.0, 136.0 (2C, ArC), 153.7 (C, CN). IR (KBr, cm⁻¹): $\tilde{\nu}$ =3049 (w), 3005 (w), 2928 (w), 1548 (m), 1456 (m), 1349 (m), 1075 (m), 924 (s), 819 (s), 752 (m). MS (EI, 70 eV): *m*/*z* (%)=336 (M⁺, ³⁷Cl, 7), 334 (M⁺, ³⁵Cl, 19), 192 (55), 180 (100), 177 (38), 143 (61), 137 (29), 112 (35), 101 (19). HRMS (EI, 70 eV): calcd for C₁₁H₁₁INOCl (M⁺, ³⁵Cl): 334.95738; found: 334.95742.

4.4.9. 6-Iodomethyl-4-(naphthalen-1-yl)-5,6-dihydro-4H-[1,2]oxazine (**4i**)

Starting with 1-(1-naphthyl)pent-4-en-1-one oxime **3i** (0.225 g, 1.0 mmol), I_2 (0.508 g, 2.0 mmol) and a saturated aqueous solution of NaHCO₃ (5 mL) in CH₂Cl₂ (10 mL), **4i** was isolated

as a brownish oil (0.232 g, 66%). ¹H NMR (300 MHz, CDCl₃): δ =1.92 (m, 1H, CH₂), 2.31 (m, 1H, CH₂), 2.35 (t, 1H, CH₂), 2.41 (t, 1H, CH₂), 2.61 (t, 2H, CH₂), 3.94 (m, 1H, CH), 74.2 (CH), 7.31 (t, ³*J*=8.2 Hz, 2H, Ar), 7.38 (d, ³*J*=8.2 Hz, 2H, Ar), 7.73 (d, ³*J*=8.2 Hz, 1H, Ar), 7.79 (d, ³*J*=8.2 Hz, 1H, Ar), 7.87 (t, ³*J*=8.2 Hz, 1H, Ar). ¹³C NMR (75 MHz, CDCl₃): δ =5.6, 24.6, 26.3 (CH₂), 74.4 (CH), 125.0, 125.1, 125.7, 126.1, 126.8, 128.5, 129.6 (ArCH), 131.3, 133.8, 133.9 (ArC), 158.0 (C, CN). IR (KBr, cm⁻¹): $\tilde{\nu}$ =3044 (w), 2953 (w), 2932 (m), 2852 (m), 1673 (m), 1506 (m), 1368 (m), 1280 (m), 1127 (m), 999 (m), 895 (s), 772 (s). MS (EI, 70 eV): *m/z* (%)=351 (M⁺, 100), 350 (11), 224 (15), 206 (15), 165 (15), 153 (46), 152 (33), 127 (38). HRMS (EI, 70 eV): calcd for C₁₅H₁₄NO₂I (M⁺): 351.01146; found: 351.011650.

4.4.10. 6-Iodomethyl-4-methyl-3-phenyl-5,6-dihydro-4H-[1,2]oxazine (**4i**)

Starting with 2-methyl-1-phenyl-pent-4-en-1-one oxime 3j (0.378 g, 2.00 mmol), I₂ (1.016 g, 4.00 mmol) and a saturated aqueous solution of NaHCO3 (20 mL) in CH2Cl2 (34 mL), 4j was isolated as a colourless solid (0.315 g, 50%); mp 70-72 °C. The product was isolated as an inseparable mixture of diastereomers (dr=1:1). Therefore, most signals are splitted. ¹H NMR (250 MHz, CDCl₃): δ =1.08, 1.17 (d, ³*J*=7.3 Hz, 3H, CHCH₃), 1.57, 1.88 (m, 1H, CHCH2CH), 2.06, 2.47 (m, 1H, CHCH2CH), 3.05 (m, 1H, CHCH3), 3.28 (m, 1H, CH₂I), 3.42 (m, 1H, CH₂I), 3.84 (m, 1H, OCHCH₂), 7.37 (m, 3H, CH_{Ar}), 7.48, 7.61 (m, 2H, CH_{Ar}). ¹³C NMR (62.9 MHz, CDCl₃): δ=5.6, 5.9 (CH₂I), 19.1, 20.2 (CHCH₃), 25.7, 28.2 (CHCH₃), 31.6, 34.7 (CHCH2CH), 70.9, 74.7 (OCH), 126.2, 126.8 (CHAr), 128.3, 128.5 (CH_{Ar}), 129.1, 129.4 (CH_{Ar}), 134.5, 134.6 (C), 159.1, 161.8 (C). IR (ATR, cm^{-1}): $\tilde{\nu}$ =3047 (br, w), 2964 (w), 2929 (w), 2867 (w), 1589 (w), 1493 (w), 1440 (m), 1410 (w), 1376 (w), 1258 (m), 1219 (w), 1187 (m), 1090 (w), 1034 (s), 1007 (s), 965 (s), 928 (w), 897 (s), 765 (s), 739 (m), 689 (s), 609 (m). MS (EI, 70 eV): *m*/*z* (%)=315 (M⁺, 100), 256 (27), 239 (11), 188 (20), 170 (27), 132 (52), 117 (45), 104 (35), 77 (46), 55 (52). HRMS (EI): calcd for C₁₂H₁₄INO (M⁺): 315.01146, found: 315.011677. Anal. Calcd for C₁₂H₁₄NOI (315.01): C, 45.73; H, 4.48; N, 4.44. Found: C, 45.80; H, 4.42; N, 4.32.

4.4.11. 6-Iodomethyl-3-(4-methoxyphenyl)-4-methyl-5,6-dihydro-4H-[1,2]oxazine (**4k**)

Starting with 1-(4-methoxyphenyl)-2-methylpent-4-en-1-one oxime **3k** (0.657 g, 3.00 mmol), I₂ (1.524 g, 6.00 mmol) and a saturated aqueous solution of NaHCO₃ (30 mL) in CH₂Cl₂ (51 mL), 4k was isolated as a colourless solid (0.445 g, 43%); mp 100-102 °C. The product was isolated as an inseparable mixture of diastereomers (dr=1:1). Therefore, most signals are splitted. For most of the diastereomers, the difference of the chemical shifts is very small. Therefore, only a single value is given. ¹H NMR (250 MHz, CDCl₃): δ =1.18 (d, ³*J*=7.3 Hz, 3H, CHCH₃), 1.86 (m, 1H, CHCH₂CH), 2.06 (m, 1H, CHCH₂CH), 3.03 (m, 1H, CHCH₃), 3.27 $(dd, {}^{2}J=10.5 \text{ Hz}, {}^{3}J=6.9 \text{ Hz}, 1\text{ H}, CHCH_{2}\text{I}), 3.44 (dd, {}^{2}J=10.5 \text{ Hz},$ ³/=5.1 Hz, 1H, CHCH₂I), 3.82 (s, 3H, OCH₃), 3.89 (m, 1H, OCHCH₂), 6.90 (d, ³*J*=9.0 Hz, 2H, CH_{Ar}), 7.55 (d, ³*J*=9.0 Hz, 2H, CH_{Ar}). ¹³C NMR (62.9 MHz, CDCl₃): δ=6.02 (CH₂I), 20.4 (CHCH₃), 25.7 (CHCH₃), 31.8 (CHCH₂CH), 55.3 (OCH₃), 70.9 (OCHCH₂), 113.9 (CH_{Ar}), 127.0 (C), 127.6 (CH_{Ar}), 158.6 (C), 160.6 (C). IR (ATR, cm⁻¹): $\tilde{\nu}$ =2960 (w), 2930 (w), 2881 (w), 2837 (w), 1606 (m), 1585 (w), 1511 (m), 1456 (m), 1411 (w), 1374 (w), 1346 (w), 1295 (m), 1244 (s), 1178 (m), 1129 (w), 1110 (w), 1093 (w), 1074 (m), 1031 (m), 1007 (m), 961 (m), 927 (m), 899 (s), 860 (m), 831 (s), 814 (s), 749 (m), 725 (w), 639 (m), 628 (s), 608 (m), 551 (m). MS (EI, 70 eV): m/z (%)=345 (M⁺, 91), 256 (50), 239 (19), 201 (17), 186 (16), 133 (19), 111 (25), 102 (45), 83 (64), 69 (69), 57 (100). HRMS (EI): calcd for C₁₃H₁₆INO₂ (M⁺): 345.02202, found: 345.022506. Anal. Calcd for C₁₃H₁₆NO₂I (345.18): C, 45.23; H, 4.67; N, 4.06. Found: C, 45.38; H, 4.60; N, 3.86.

4.4.12. 6-Bromomethyl-3-(2-ethoxyphenyl)-5,6-dihydro-4H-[1,2]oxazine (**4**I)

Starting with 1-(2-ethoxyphenyl)pent-4-en-1-one oxime 3e (0.329 g, 1.50 mmol) and NBS (0.267 g, 1.50 mmol) in CH₂Cl₂ (7.5 mL), **4I** was isolated as a brownish viscous oil (0.256 g, 57%). ¹H NMR (250 MHz, CDCl₃): δ =1.40 (t, ³*J*=7.0 Hz, 3H, OCH₂CH₃), 1.89 (m, 1H, CHCH₂CH₂), 2.23 (m, 1H, CHCH₂CH₂), 2.69 (dd, ²J=8.2 Hz, ${}^{3}J=5.7$ Hz, 2H, CCH₂), 3.47 (dd, ${}^{2}J=10.4$ Hz, ${}^{3}J=7.3$ Hz, 1H, CHCH₂Br), 3.62 (dd, ²*J*=10.4 Hz, ³*J*=4.9 Hz, 1H, CHCH₂Br), 4.05 (q, ³*J*=7.0 Hz, 2H, OCH2CH3), 4.11 (m, 1H, OCHCH2), 6.92 (m, 2H, CHAr), 7.32 (m, 2H, CH_{Ar}). ¹³C NMR (62.9 MHz, CDCl₃): δ =14.8 (OCH₂CH₃), 22.7 (CH₂), 23.9 (CH₂), 32.5 (CH₂Br), 63.8 (OCH₂CH₃), 73.9 (CHO), 111.9 (CH_{Ar}), 120.6 (CH_{Ar}), 125.8 (C), 129.6 (CH_{Ar}), 130.4 (CH_{Ar}), 156.6, 158.5 (C). IR (ATR, cm⁻¹): $\tilde{\nu}$ =3061 (br, w), 2976 (w), 2929 (w), 1600 (m), 1491 (m), 1475 (w), 1446 (s), 1391 (w), 1291 (m), 1236 (s), 1161 (w), 1122 (m), 1039 (s), 1023 (s), 924 (w), 897 (s), 800 (w), 750 (s), 682 (w), 656 (w). MS (GC/MS, 70 eV): *m*/*z* (%)=299 (M⁺, ⁸¹Br, 7), 297 (M⁺, ⁷⁹Br, 7), 267 (4), 265 (4), 204 (34), 174 (24), 158 (60), 145 (100), 132 (21), 103 (9), 91 (35), 77 (18). HRMS (EI): calcd for C₁₃H₁₆ O₂NBr (M⁺): 297.03589, found: 297.035775.

4.4.13. 6-Bromomethyl-3-(4-ethoxyphenyl)-5,6-dihydro-4H-[1,2]oxazine (**4m**)

Starting with 1-(4-ethoxyphenyl)pent-4-en-1-one oxime 3f (0.438 g, 2.00 mmol) and NBS (0.356 g, 2.00 mmol) in CH₂Cl₂ (10.0 mL), 4m was isolated as a colourless solid (0.519 g, 87%); mp 130–135 °C. ¹H NMR (250 MHz, CDCl₃): δ =1.41 (t, ³*I*=7.0 Hz, 3H, OCH₂CH₃), 1.91 (m, 1H, CHCH₂CH₂), 2.28 (m, 1H, CHCH₂CH₂), 2.65 (m, 2H, CCH₂), 3.44 (dd, ²*I*=10.8 Hz, ³*I*=7.0 Hz, 1H, CHCH₂Br), 3.60 (dd, ${}^{2}I=10.8$ Hz, ${}^{3}I=5.0$ Hz, 1H, CHCH₂Br), 3.98 (m, 1H, OCHCH₂), 4.04 $(q, {}^{3}J=7.0 \text{ Hz}, 2\text{H}, \text{ OCH}_{2}\text{CH}_{3}), 6.88 \text{ (d, }^{3}J=9.0 \text{ Hz}, 2\text{H}, \text{ CH}_{Ar}), 7.61$ $(d, {}^{3}J=9.0 \text{ Hz}, 2\text{H}, C\text{H}_{Ar})$. ${}^{13}C \text{ NMR} (62.9 \text{ MHz}, CDCl_{3})$: $\delta = 14.7 (OCHCH_{3})$, 21.2 (CH₂), 22.9 (CH₂), 32.4 (CH₂Br), 63.5 (OCH₂CH₃), 73.7 (OCHCH₂), 114.3 (CH_{Ar}), 126.7 (CH_{Ar}), 127.6, 154.5, 160.2 (C). IR (ATR, cm^{-1}): $\tilde{\nu}$ =2977 (w), 2909 (br, w), 1608 (w), 1590 (m), 1511 (m), 1479 (m), 1449 (br, w), 1414 (w), 1392 (m), 1384 (m), 1356 (m), 1337 (m), 1292 (m), 1247 (s), 1225 (m), 1170 (m), 1116 (m), 1093 (w), 1064 (w), 1043 (m), 1022 (m), 988 (m), 941 (w), 911 (m), 852 (m), 816 (s), 763 (m), 664 (m), 621 (m), 547 (s). MS (GC/MS, 70 eV): m/z (%)=299 (M⁺, ⁸¹Br, 98), 297 (M⁺, ⁷⁹Br, 100), 268 (3), 204 (28), 176 (17), 148 (22), 147 (20), 134 (21), 119 (56), 91 (22), 77 (11), 65 (20). HRMS (EI): calcd for C13H16 BrNO2 (M⁺): 297.03589, found: 297.035839. Anal. Calcd for C₁₃H₁₆NO₂Br (298.18): C, 52.36; H, 5.41; N, 4.70. Found: C, 52.01; H, 5.32; N, 4.52.

4.4.14. 6-Bromomethyl-4-methyl-3-phenyl-5,6-dihydro-4H-[1,2]oxazine (**4n**)

Starting with 2-methyl-1-phenyl-pent-4-en-1-one oxime 3j (0.567 g, 3.00 mmol) and NBS (0.534 g, 3.00 mmol) in CH₂Cl₂ (15 mL), **4n** was isolated as a brownish viscous oil (0.584 g, 73%). The product was isolated as an inseparable mixture of diastereomers (dr=1:1). Therefore, most signals are splitted. ¹H NMR (250 MHz, CDCl₃): δ =1.10, 1.18 (d, ³*J*=7.2 Hz, 3H, CHCH₃), 1.98 (m, 2H, CHCH₂CH), 3.07 (m, 1H, CHCH₃), 3.47 (m, 1H, CH₂I), 3.62 (m, 1H, CH₂I), 4.04 (m, 1H, OCHCH₂), 7.36-7.41 (m, 3H, CH_{Ar}), 7.41-7.62 (m, 2H, CH_{Ar}). ¹³C NMR (62.9 MHz, CDCl₃): δ=19.2, 20.2 (CHCH₃), 25.4, 27.9 (CHCH₃), 30.3, 32.6 (CHCH₂CH), 32.8, 33.2 (CH₂Br), 70.8, 74.5 (OCH), 126.3, 126.8 (CH_{Ar}), 128.4, 128.6 (CH_{Ar}), 129.2, 129.5 (CH_{Ar}), 134.7 (C), 159.2, 161.9 (C). IR (ATR, cm⁻¹): $\tilde{\nu}$ =3055 (br, w), 2964 (w), 2932 (w), 2893 (w), 1588 (w), 1563 (w), 1493 (w), 1452 (w), 1441 (m), 1417 (m), 1376 (m), 1340 (w), 1290 (w), 1262 (m), 1226 (m), 1177 (w), 1138 (w), 1096 (w), 1083 (m), 1043 (m), 1011 (m), 973 (m), 919 (s), 902 (s), 863 (m), 817 (w), 765 (s), 745 (s), 689 (s), 656 (s), 628 (m), 557 (m). MS (GC/MS, 70 eV): *m*/*z* (%)=269 (M⁺, ⁸¹Br, 92), 267 (M⁺, ⁷⁹Br, 93), 188 (7), 174 (90), 146 (17), 132 (45), 117 (90), 104 (78), 91 (42), 77 (98), 55 (100). HRMS (EI): calcd for C₁₂H₁₄NOBr (M⁺): 267.01146, found: 267.011677.

4.4.15. 6-Bromomethyl-3-(4-methoxyphenyl)-4-methyl-5,6dihydro-4H-[1,2]oxazine (**40**)

Starting with 1-(4-methoxyphenyl)-2-methylpent-4-en-1-one oxime 3k (0.519 g, 2.70 mmol) and NBS (0.480 g, 2.70 mmol) in CH₂Cl₂ (13.5 mL), 40 was isolated as a colourless solid (0.201 g, 25%); mp 88-91 °C. The product was isolated as an inseparable mixture of diastereomers (dr=1:1). Therefore, most signals are splitted. For most of the diastereomers, the difference of the chemical shifts is very small. Therefore, only a single value is given. ¹H NMR (250 MHz, CDCl₃): δ =1.19 (d, ³J=7.3 Hz, 3H, CHCH₃), 1.96 (m, 2H, CHCH₂CH), 3.03 (m, 1H, CHCH₃), 3.40 (dd, ²J=10.6 Hz, ${}^{3}J=6.4$ Hz, 1H, CHCH₂Br), 3.56 (dd, ${}^{2}J=10.6$ Hz, ${}^{3}J=5.1$ Hz, 1H, CHCH2Br), 3.82 (s, 3H, OCH3), 4.04 (m, 1H, OCHCH2), 6.90 (d, $^{3}J=9.0$ Hz, 2H, CH_{Ar}), 7.55 (d, $^{3}J=9.0$ Hz, 2H, CH_{Ar}). ^{13}C NMR (75.5 MHz, CDCl₃): δ=20.3 (CHCH₃), 25.3 (CHCH₃), 30.4 (CHCH₂), 32.9 (CH₂Br), 55.3 (OCH₃), 70.7 (OCHCH₂), 114.0 (CH_{Ar}), 127.1 (C), 127.6 (CH_{Ar}), 158.7 (C), 160.7 (C). IR (ATR, cm⁻¹): $\tilde{\nu}$ =2965 (w), 2931 (w), 2877 (w), 1606 (s), 1510 (s), 1461 (m), 1374 (w), 1294 (m), 1245 (s), 1176 (m), 1109 (w), 1095 (w), 1079 (w), 1044 (m), 1029 (m), 1013 (s), 969 (m), 911 (s), 831 (s), 813 (s), 755 (m), 655 (m), 624 (m). MS (GC/MS, 70 eV): *m*/*z* (%)=299 (M⁺, ⁸¹Br, 100), 297 (M⁺, ⁷⁹Br, 99.7), 218 (2), 204 (23), 186 (11), 162 (15), 160 (15), 147 (32), 134 (55), 133 (50), 115 (16), 103 (20), 91 (16), 77 (21). HRMS (EI): calcd for C₁₃H₁₆NO₂Br (M⁺): 297.03589, found: 297.035394.

4.4.16. 3-Iodomethyl-4,4a,5,6-tetrahydro-3H-naphtho[1,2-c] [1,2]oxazine (**4p**)

Starting with 2-allyl-3,4-dihydro-2H-naphthalen-1-one oxime **31** (0.221 g, 1.1 mmol), I₂ (0.559 g, 2.2 mmol) and a saturated aqueous solution of NaHCO₃ (11.0 mL) in CH₂Cl₂ (18.0 mL), 4p was isolated as a colourless solid (0.345 g, 96%); mp 105-107 °C. The product was isolated as an inseparable mixture of diastereomers (dr=1:1). Therefore, most signals are splitted. For most of the diastereomers, the difference of the chemical shifts is very small. Therefore, only a single value is given. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.37 - 1.65$ (m, 2H, CH₂), 2.09 (m, 1H, CH₂), 2.33 (m, 1H, CH₂), 2.48 $(m, 1H, CCHCH_2), 2.85 (m, 2H, CH_2), 3.20 (dd, {}^2J=10.6 Hz, {}^3J=7.0 Hz,$ 1H, CHCH₂I), 3.36 (dd, ²J=10.6 Hz, ³J=5.0 Hz, 1H, CHCH₂I), 3.88 (m, 1H, OCH), 7.14 (m, 3H, CH_{Ar}), 7.96 (d, ³*J*=7.8 Hz, 1H, CH_{Ar}). ¹³C NMR (75.5 MHz, CDCl₃): δ=5.8 (CH₂I), 28.9 (CH₂), 29.0 (CH₂), 32.0 (CH₂), 33.0 (CCHCH₂), 73.6, 75.3 (OCH), 124.7 (CH_{Ar}), 126.5 (CH_{Ar}), 129.0 (CH_{Ar}), 129.6 (CH_{Ar}), 129.5 (C), 138.1 (C), 154.4 (C). IR (ATR, cm⁻¹): $\tilde{\nu}$ =3016 (br, w), 2924 (w), 2855 (w), 2831 (w), 1728 (w), 1610 (w), 1479 (w), 1445 (w), 1431 (w), 1372 (w), 1323 (w), 1307 (w), 1291 (w), 1259 (w), 1198 (s), 1151 (w), 1125 (w), 1098 (w), 1079 (w), 1009 (s), 968 (m), 945 (m), 919 (s), 880 (s), 763 (s), 728 (s), 677 (m), 646 (m), 620 (w). MS (EI, 70 eV): m/z (%)=327 (M⁺, 100), 297 (4), 182 (13), 170 (15), 144 (16), 128 (50), 116 (23), 89 (11), 77 (13). HRMS (EI): calcd for C₁₃H₁₄INO (M⁺): 327.01146, found: 327.010903.

Acknowledgements

We are grateful to Dr. Tuan T. Dang for his help. Financial support by the State of Mecklenburg-Vorpommern (scholarship for V.K.) and by the State of Vietnam (MOET scholarship for T.T.D.) is gratefully acknowledged.

References and notes

- (a) Streith, J.; Defoin, A. Synlett **1996**, 189; (b) Defoin, A.; Sarazin, H.; Streith, J. Tetrahedron **1997**, 53, 13769; (c) Bach, P.; Bols, M. Tetrahedron Lett. **1999**, 40, 3460.
- 2. Miyashita, M.; Awen, B. Z. E.; Yoshikoshi, A. Tetrahedron 1990, 46, 7569.
- (a) Gilchrist, T. L. J. Chem. Soc., Chem. Rev. **1983**, 12, 53; (b) Gilchrist, T. L.; Roberts, T. G. J. Chem. Soc., Chem. Commun. **1978**, 847; (c) Zimmer, R.; Reißig, H. U. Liebigs Ann. Chem. **1991**, 553; (d) Yoon, S. C.; Kim, K.; Park, Y. J. J. Org. Chem. **2001**, 66, 7334.
- 4. Naruse, M.; Aoyagi, S.; Kibayashi, C. Tetrahedron Lett. 1994, 35, 595.

- (a) Bowman, W. R.; Davies, R. V.; Slawin, A. M. Z.; Sohal, G. S.; Titman, R. B.; Wilkins, D. J. J. Chem. Soc., Perkin Trans. 1 1997, 155.
- (a) Tiecco, M.; Testaferri, L.; Tingoli, M.; Bagnoli, L.; Marini, F. J. Chem. Soc., Perkin Trans. 1 1993, 1989; (b) Grigg, R.; Hadjisoteriou, M.; Kennewell, P.; Markandu, J. J. Chem. Soc., Chem. Commun. 1992, 1537.
- 7. Bishop, R.; Hawkins, S. C.; Quibuyen, T. A. O.; Brooks, P. R. *Tetrahedron Lett.* **1988**, 29, 6805.
- Armesto, D.; Austin, M. A.; Griffiths, O. J.; Horspool, W. M.; Carpintero, M. Chem. Commun. 1996, 2715.
- (a) Kaiser, A.; Mayer, K. K.; Sellmer, A.; Wiegrebe, W. Monatsh. Chem. 2003, 3, 343; (b) Ellames, G. J.; Hewkin, C. T.; Jackson, R. F. W.; Smith, D. I.; Standen, S. P. Tetrahedron Lett. 1989, 30, 3471.
- 10. Saiki, H.; Mukai, T. Chem. Lett. **1981**, 1561.
- 11. Lathbury, D.; Gallagher, T. J. Chem. Soc., Chem. Commun. 1986, 1017.
- Castin, V.; Rentzea, C. N. Angew. Chem. **1980**, 92, 195.
 Boberg, F.; Ruhr, M.; Garburg, K. H.; Garming, A. J. Heterocycl. Chem. **1986**, 23, 759
- Dang, T. T.; Albrecht, U.; Gerwien, K.; Siebert, M.; Langer, P. J. Org. Chem. 2006, 71, 2293.

- 15. For a review of cyclization reactions of dianions, see: Langer, P.; Freiberg, W. Chem. Rev. 2004, 104, 4125.
- 16. CCDC-680927 (4d), CCDC-682000 (4f) and CCDC-682001 (4j) contain all crystallographic details of this publication and is available free of charge at www. ccdc.cam.ac.uk/conts/retrieving.html or can be ordered from the following address: Cambridge Crystallographic Data Centre, 12 Union Road, GB-Cambridge CB21E2; fax: +44 1223 336 033 or deposit@ccdc.cam.ac.uk.
- Grigg, R.; Markandu, J.; Perrior, T.; Surendrakumar, S.; Warnock, W. J. Tetrahedron 1992, 48, 6929.
- Dondas, H. A.; Grigg, R.; Hadjisoteriou, M.; Markandu, J.; Thomas, W. A.; Kennewell, P. Tetrahedron 2000, 56, 10087.
- Dondas, H. A.; Grigg, R.; Hadjisoteriou, M.; Markandu, J.; Kennewell, P.; Thornton-Pett, M. Tetrahedron 2001, 57, 1119.
- 20. The reaction of acetophenone with hydroxylamine results in the formation of the *E*-configured oxime: (a) Janny, A. *Ber. Dtsch. Chem. Ges.* **1882**, *15*, 2781; For the synthesis of the *Z*-configured acetophenone oxime, see: (b) Smith, J. H.; Kaiser, E. T. *J. Org. Chem.* **1974**, 39, 728; For the stereochemistry of acetophenone oximes, see also: (c) Moehrle, H.; Wehefritz, B.; Steigel, A. *Tetrahedron* **1987**, *43*, 2255.