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Facile synthesis of 3-arylpyrroles by tandem Suzuki–dehydrogenation reaction

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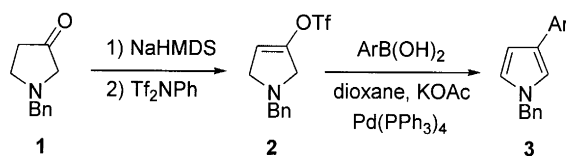
Abstract

Facile method for the synthesis of 3-arylpyrroles is described based on the palladium-catalyzed coupling of trifluoromethanesulfonic acid 1-benzyl-2,5-dihydro-1*H*-pyrrol-3-yl ester with aryl halides. © 2000 Elsevier Science Ltd. All rights reserved.

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The introduction of substituents into the 3-position of pyrrole has been actively pursued in recent years due to the importance of this class of compound in natural product synthesis.¹ In contrast to the significant number of preparations of polysubstituted 3-arylpyrroles,² there are relatively few syntheses of simple 3-arylpyrroles. A literature survey on the preparation of 3-arylpyrrole showed that some of the methods reported involve the reductive cyclization of 2-arylsuccinonitriles,³ the base-induced ring closure of arylvinamidinium salts,⁴ and rhodium-catalyzed hydroformylation of β -alkynylamines with CO/H₂.⁵

We now report that reaction of trifluoromethanesulfonic acid 1-benzyl-2,5-dihydro-1*H*-pyrrol-3-yl ester (**2**) with aryl halides in the presence of palladium-catalyzed leads to the formation of 3-arylpyrroles (**3**) in good yields (see Scheme 1 and Table 1).



Scheme 1.

The synthesis was carried out using commercially available 1-benzyl-3-pyrrolidinone (**1**) as the starting material. Vinyl triflate (**2**)⁶ could be prepared in 55% yield from 1-benzyl-3-pyrrolidinone

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Table 1
Tandem Suzuki–dehydrogenation reaction

entry	boronic acid	product (3) ^{a,b}	yield ^c
1	4-fluorophenylboronic acid	3a	69 %
2	4-methoxyphenylboronic acid	3b	65 %
3	3-nitrophenylboronic acid	3c	73 %
4	1-naphthaleneboronic acid	3d	74 %
5	2-furanboronic acid	3e	66 %

^a Reaction was conducted for overnight by using vinyl triflate **2** (0.3 mmol), boronic acid (1.1 eq), Pd(PPh₃)₄ (3 mol %), KOAc (3 eq) and dioxane (5 mL). ^b All compounds were characterized by FAB-MS, ¹H-NMR, and ¹³C-NMR. ^c These isolated yields are not optimized.

(**1**) by trapping the enolate with a triflating reagent, *N*-phenyltrifluoromethanesulfonimide (Tf₂NPh).⁷ The regiochemistry of enolization of α -amino ketone has been investigated by A. R. Martin et al.⁸ In accordance with his result, when the *N*-substituents are electron-donating groups (methyl, benzyl), only allylamine-type isomers were detected, and if the substituents are electron-withdrawing groups (Cbz, Boc), the enamine-type isomers were exclusively observed. As expected, the reaction took place in such a manner that it gives exclusively allylamine-type triflate (**2**) rather than enamine-type isomer, determined by NOE experiments. We have treated the vinyl triflate (**2**) with boronic acid derivatives in the presence of a catalytic amount of Pd(PPh₃)₄ and KOAc in dioxane at 100°C.⁹ The reaction mixture provided a compound, which was isolated and characterized to be 3-arylpyrroles. Apparently, after Suzuki coupling, subsequent dehydrogenation was occurred providing 3-arylpyrroles. In general, dehydrogenative aromatizations can be achieved by heating (300~350°C) hydroaromatic compounds with catalytic amounts of palladium on activated carbon.¹⁰ In our reaction, under the Suzuki coupling condition, the dehydrogenative aromatization¹¹ of the pyrroline (**2**) took place in tandem mode. This could be the first example of tandem Suzuki–dehydrogenation reaction to prepare 3-arylpyrroles from pyrroline.

References

- Anderson, H. J.; Loader, C. E. *Synthesis* **1985**, 353–364 and references cited therein.
- Bean, G. T. In *Pyrroles. Part 1: The Synthesis of 1H-Pyrroles*; Jones, R. A., Ed.; John Wiley and Sons: New York, 1990; pp. 105–294. Gupta, R. R.; Kumar, M.; Gupta, V. *Heterocyclic Chemistry*; Springer: Heidelberg, 1999; Vol. 2, pp. 34–43.
- Babler, J. H.; Spina, K. P. *Tetrahedron Lett.* **1984**, 25, 1659–1660.
- Gupton, J. T.; Krolikowski, D. A.; Yu, R. H.; Riesinger, S. W.; Sikorski, J. A. *J. Org. Chem.* **1990**, 55, 4735–4740. Gallagher, P. T.; Palmer, J. L.; Morgan, S. E. *J. Chem. Soc., Perkin Trans. 1* **1990**, 3212–3214.
- Campi, E. M.; Jackson, W. R.; Nilsson, Y. *Tetrahedron Lett.* **1991**, 32, 1093–1094.
- Ritter, K. *Synthesis* **1993**, 735–762.
- Preparation of vinyl triflate (2)*: To a stirred solution of 1-benzyl-3-pyrrolidinone (**1**) in anhydrous THF under nitrogen at –78°C was added sodium bis(trimethylsilyl)amide (1.0 M solution in THF) dropwise. After stirring at –78°C for 1 h, a solution of *N*-phenyltrifluoromethanesulfonimide in anhydrous THF was quickly added to the mixture. After agitating at –78°C for 10 min, the slurry was warmed slowly to 0°C and stirred at 0°C for 3 h. The solvent was removed, and the crude

- residue was taken up in CH_2Cl_2 and washed with water. Drying and evaporation of the organic extracts gave an oil, which was purified by flash chromatography. ^1H NMR (300 MHz, CDCl_3) δ 3.62 (s, 4H), 3.82 (s, 2H), 5.69 (s, 1H), 7.25–7.35 (m, 5H); ^{13}C NMR (75 MHz, CDCl_3) δ 56.4, 56.5, 60.2, 113.9, 118.5 (d, $J=318.9$ Hz), 127.4, 128.52, 128.54, 137.9, 145.1.
8. Zheng, Q.; Yang, Y.; Martin, A. R. *Heterocycles* **1994**, 37, 1761–1772.
 9. *Preparation of 3-arylpyrroles (3)*: A mixture of vinyl triflate, boronic acid, $\text{Pd}(\text{PPh}_3)_4$ and KOAc in dioxane was heated to 100°C overnight. The product was extracted with CH_2Cl_2 , washed with brine and dried over Na_2SO_4 . Isolation by flash chromatography gave following compounds. *Representative data for the selected compound (3a)*: ^1H NMR (300 MHz, CDCl_3) δ 5.08 (s, 2H), 6.46 (m, 1H), 6.73 (m, 1H), 6.94 (m, 1H), 7.00–7.49 (m, 9H); ^{13}C NMR (75 MHz, CDCl_3) δ 54.0, 106.9, 115.8 (d, $J=21.2$ Hz), 118.1, 122.8, 124.7, 126.8 (d, $J=7.7$ Hz), 127.5, 128.4, 129.2, 132.4 (d, $J=3.2$ Hz), 138.2, 161.5 (d, $J=242.0$ Hz); FAB-MS m/z 252.0 ($[\text{M}+\text{H}]^+$, calcd 252.3).
 10. Hudlický, M. *Oxidations in Organic Chemistry*; American Chemical Society: Washington, DC, 1990; pp. 38, 50–51.
 11. There is literature about the oxidation of pyrroline to pyrrole by $\text{Pd}(\text{PPh}_3)_4$: Bäckvall, J.-E.; Nyström, J.-E. *J. Chem. Soc., Chem. Commun.* **1981**, 59–61.