Nucleoside Diphosphate Prodrugs: Non-symmetric Di*PP*ro-Nucleotides

Lina Weinschenk¹, Dominique Schols², Jan Balzarini² and Chris Meier¹*

¹Organic Chemistry, Department of Chemistry, Faculty of Sciences, University of Hamburg, Martin-Luther-King-Platz 6, 20146 Hamburg, Germany and ²Katholieke Universiteit Leuven, Rega Institute for Medical Research, Minderbroedersstraat 10, 3000 Leuven, Belgium.

* Correspondence to: chris.meier@chemie.uni-hamburg.de

KEYWORDS antiviral agents – nucleoside analogs – pronucleotides – bioreversible protection – drug delivery – anti-HIV – nucleoside diphosphate delivery

ABSTRACT Non-symmetric Di*PP*ro-nucleotides are described as nucleoside diphosphate (NDP) delivery systems. The concept is to attach different bis(acyloxybenzyl)-moieties at the β -phosphate moiety of a NDP. Di*PP*ro-compounds bearing two alkanoylbenzyl residues and Di*PP*ro-compounds bearing an alkanoylbenzyl or a benzoylbenzyl group as bioreversible prodrug moieties were studied. Compounds bearing short chain alkanoyl esters led to a fast

hydrolysis by chemical or enzymatic means. The ester group in the second prodrug group comprised of a long lipophilic aliphatic or an aromatic residue. The lipophilicity of this group enabled the prodrug to penetrate the cell membrane. The introduction of two different groups allowed a controlled step-wise removal of the prodrug moieties to achieve a highly selective delivery of the NDP in CEM cell extracts. The compounds were highly active against HIV even in thymidine kinase-deficient CEM cells. Thus, the compounds although charged at the α -phosphate group were up-taken by the cells and released NDPs.

INTRODUCTION

 Nucleoside analogues are extensively used as agents in antiviral and antitumor chemotherapy. However, their biological activity is strongly dependent on their intracellular activation by host cell kinases to give – via the mono- and the diphosphate – the ultimately active triphosphate derivative which inhibits the function of the viral DNA polymerase. However, in the case of nucleoside analogues cellular kinases often catalyze these metabolic steps insufficiently.^{1,2} The result might be a partial loss of antiviral activity or for some nucleoside analogues even a complete failure to exhibit any antiviral activity. Moreover, inefficient phosphorylation led in some cases to adverse effects.^{3,4} The use of prodrug forms of phosphorylated nucleoside analogues (pronucleotides) can circumvent rate limiting steps within the metabolic pathway by bypassing at least one, if not several involved activating enzymes.^{5,6} This task has been successfully achieved in the past for the intracellular delivery of monophosphates of nucleoside analogues using prodrug strategies such as the *cyclo*Sal-,⁷⁻⁹ Sate-,^{10,11} bisPOM-¹² and phosphoramidate-nucleotide approaches.^{13,14}

Remarkably, in contrast to numerous examples of successful nucleoside monophosphate prodrug approaches for the bypass of nucleoside kinases, the development of nucleoside

Journal of Medicinal Chemistry

diphosphate (NDP) prodrugs has been very rarely addressed. However, such a prodrug approach might be desirable, for instance, in the case of 3'-azido-3'-deoxythymidine (AZT), which is very poorly phosphorylated from AZTMP to yield AZTDP by thymidylate kinase (TMP-K) in human cells.^{15,16}

Hostetler et al. synthesized different nucleoside diphosphate diglycerides as potential NDP prodrugs.¹⁷⁻²⁰ However, the hydrolysis of these compounds delivered the monophosphates instead of the diphosphates as a result of the cleavage of the diphosphate moiety. A second approach was reported by Huynh-Dinh et al.²¹⁻²³ In their approach, the β -phosphate was acylated with fatty acids forming to a mixed anhydride bond. In chemical hydrolysis studies the NDP was formed. However, in cell extracts an undefined decomposition of the compounds was observed.

Since a few years, we turned our interest towards the bioreversible protection of nucleoside diphosphates although, as compared to nucleoside monophosphate prodrugs, the design of such compounds is more demanding due to a chemical reason. In NDPs, a complete lipophilic modification of the charged and therefore highly polar diphosphate unit led to a considerable decrease in chemical stability of this part of the molecule. As a result, the phosphate anhydride bond of the nucleoside diphosphate was cleaved leading to two monophosphorylated fragments.

Earlier, we reported on the Di*PP*ro-approach. In this approach, esterification of the β -phosphate of a NDP with two bis(acyloxybenzyl)-moieties neutralizes the charges at this phosphate group. In previous studies this was achieved by masking the β -phosphate using two identical bis(alkanoyloxybenzyl)-^{24,25} or bis(benzoyloxybenzyl)-moieties²⁶ (symmetrical modification, R¹ = R²). The concept was based on an enzyme-driven cleavage of these masking units, which ensured that the release of NDPs preferably took place intracellularly due to the higher concentrations of the esterases/lipases in the target cells. In contrast to the low stability in

the presence of esterases, the chemical stability at physiological pH was still found to be high. By choosing the appropriate substituent R in the ester moiety within the mask, it was possible to "tune" both stability and lipophilicity of the prodrugs (Scheme 1).²⁵



Scheme 1. Hydrolysis of earlier reported symmetric bis(acyloxybenzyl)-DiPPro-compounds 1.

Although we were able to prove that this approach works, an unwanted property of the symmetric aliphatic or aromatic Di*PP*ro-compounds was that also some nucleoside monophosphate (NMP) was detected in addition to the wanted nucleoside diphosphate e.g. in PBS solution, pH 7.3 or cell extract hydrolysis studies. Interestingly, it was shown that the amount of NMP clearly correlated with the stability of the Di*PP*ro-compounds and that the stability of the compounds was dependent on the length of the alkyl residues R attached via an ester linkage to the 4-hydroxybenzyl group: the longer the alkyl group R, the higher the stability of the Di*PP*ro-compounds but the higher was also the amount of NMP formed in the chemical hydrolyses and in the cell extract studies. As the reason for the formation of NMP a concurrence

Journal of Medicinal Chemistry

reaction was identified in which water/hydroxide reacted at the phosphorus atom of the β phosphate group. This led to the cleavage of the phosphate anhydride bond releasing the NMP. This side reaction became more relevant with increasing stability of the ester linkage in the prodrug moiety. On the other hand, we observed that after the removal of the first mask to give a mono-masked intermediate such as **2**, no further increase of the amount of NMP was observed (Scheme 1). The reason here was that after the first removal an additional charge appeared at the β -phosphate and this obviously prevented a nucleophilic reaction at this moiety.

Taking all previous results together, these findings led to the development of a second generation of Di*PP*ro-compounds. Here, we disclose a study on series of *non-symmetric* Di*PP*ronucleotides **1** bearing two different masking units (\mathbb{R}^1 not equal to \mathbb{R}^2). The design was the following: One masking group contained a short alkyl chain carboxylic acid ester which - due to our previously reported data²⁵ - will be rapidly cleaved by chemical or enzymatic means. The second masking group contained a long alkyl residue carboxylic acid ester or a substituted benzoic acid ester, which adds high lipophilicity to the molecule. It was expected that such a construct would allow a rapid conversion of the Di*PP*ro-compound **1** into the mono-masked intermediate **2** and thereby should avoid the side reaction to form the unwanted nucleoside monophosphate. The second mask will be cleaved subsequently from the intermediate to form the nucleoside diphosphate. It was expected that with these non-symmetric compounds a highly selective conversion of the Di*PP*ro-compounds into nucleoside diphosphates can be achieved.

RESULTS AND DISCUSSION

Synthesis

For the synthesis of non-symmetric Di*PP*ro-NDPs **1**, we first tried to use the protocol of the symmetric counterparts.²⁴⁻²⁶ According to that protocol two equivalents of an appropriate 4-acyloxybenzylalcohol were reacted with dichloro-*N*,*N*-di*iso*propylaminophosphoramidite to give the corresponding phosphoramidite like **5** ($R^1=R^2$) which was then coupled by an acid-mediated reaction with a nucleoside monophosphate followed by oxidation. For the new non-symmetric compounds it was tried to introduce two different 4-acyloxybenzyl masking units **3** and **4** by the same way. However, this approach failed.

Finally, in a new protocol 4-acyloxybenzylalcohol **3** was reacted with phosphorus trichloride followed by the addition of two equivalents of *N*,*N*-di*iso*propylamine (DIPA) which led to bis(di*iso*propylamino)-phosphoramidites **6** in yields up to 56% in a *one-pot* reaction. Then, one of the di*iso*propylamino moieties was replaced by the second 4-acyloxybenzylalcohol **4** after activation with dicyanoimidazol (DCI) to yield the non-symmetric phosphoramidites **5** in very high yields.

Next, a second acid-activated coupling of these phosphoramidites **5** with the corresponding nucleoside monophosphates was carried out. At the beginning, the conversions obtained in this coupling reaction were surprisingly low, which was proven by NMR- and HPLC-analysis of the crude mixtures. Changing the equivalents of phosphoramidite **5** and the DCI-activator did not led to an improvement. Finally, the problem was solved by the successive addition of small amounts of the DCI-activator solution. Starting with the addition of 0.5 equiv. DCI-activator solution to a solution of 1 equiv. of the nucleoside monophosphate and 1.5 equiv. of the phosphoramidite **5**, followed by the stepwise addition of 0.25 equiv. every 5 min up to a total of 1.25 or 1.75 equiv. of DCI led to a quantitative conversion of the nucleotide. Di*PP*ro-nucleotides **7-10** were successfully isolated by using automated flash RP-18-chromatography. Using the new protocol

Journal of Medicinal Chemistry

numerous combinations of Di*PP*ro-nucleotides were accessible in quantitative conversions, high isolated yields and high purity, which was checked by means of ¹H-, ¹³C- and ³¹P-NMR spectroscopy, mass spectrometry and HPLC analysis. The route for the synthesis of non-symmetric Di*PP*ro-NDPs **7-10** is summarized in Scheme 2.



Scheme 2. Synthesis of non-symmetric Di*PP*ro-NDPs 7-10. i) a) 1 equiv. 3, 1 equiv. pyridine, -78 °C to rt, 24 h, THF. b) 6.1 equiv. *N*,*N*-di*iso*propylamine, -10 °C to rt, 24 to 48 h, THF. ii) 0.67 equiv. 4, 0.67 equiv. DCI-activator solution (0.25 molar in acetonitrile), 4 °C to rt, 30 to 60 min, acetonitrile. iii) a) 1 equiv. $[N(nC_4H_9)_4]d4TMP$ or $[N(nC_4H_9)_4]AZTMP$, 1.5 equiv. 5, 1.25 to 1.75 equiv. DCI-activator solution (0.25 molar in acetonitrile), rt, 30 min, acetonitrile; b) 1.5 equiv. *tert*BuOOH solution (5.5 molar in *n*-decane), rt, 15 min, acetonitrile. (indices shown in compound 5,6 and 7-10 are used for NMR assignment in Experimental Section)

Structural formulae of DiPPro-d4TDPs 7,8 and DiPPro-AZTDPs 9,10 bearing different aliphatic ($R^2 = C_n H_{2n+1}$) and aromatic ($R^2 = X$ -Ph) acyl moieties as the lipophilic masks in

combination with short aliphatic acyl moieties ($R^1 = CH_3$, C_4H_9) in the acyloxybenzyl-masking group are summarized in Figure 1. Also three symmetric Di*PP*ro-AZTDPs **11a-c** were prepared by the earlier published method.



Figure 1. Non-symmetric DiPPro-d4TDPs 7,8 and DiPPro-ATZDPs 9,10.

Hydrolysis Studies

All prepared compounds were studied with regard to their stabilities and their hydrolysis products in different media.

Chemical stability in phosphate buffer, pH 7.3: We expected that the more labile mask would be cleaved faster to form intermediate **2** bearing the long, lipophilic and hydrolytically more stable mask. However, surprisingly in the chemical hydrolysis studies in most of the cases both

Journal of Medicinal Chemistry

intermediates were observed by means of HPLC analysis. In the mono-masked intermediates 2, the short alkanoyl ester moieties were more rapidly hydrolyzed as compared to the long alkyl bearing moieties. Furthermore, the half-lives determined in phosphate buffer, pH 7.3, of the non-symmetric Di*PP*ro-compounds were surprisingly higher as compared to those of the symmetric Di*PP*ro-NDPs. This can be seen for example for compound **7b** (CH₃/C₉H₁₉-Di*PP*ro-d4TDP). In this case the half-life of the cleavage of the first masking group bearing the acetate group was 40 h while the same masking group in the symmetrical di-CH₃-Di*PP*ro-d4TDP was cleaved with a half-life of 10 h only.²⁴ Steric hindrance or aggregation due to the second, more lipophilic moiety might be a reason for this increase in stability. The determined half-lives of the formation of the mono-masked intermediate were between 15 h and 69 h as summarized in Table 1.

The chemical stabilities of Di*PP*ro-AZTDPs **9**,**10** were in the same range as those of the Di*PP*ro-d4TDPs **7**,**8** having identical prodrug moieties. Only $CH_3/C_{11}H_{23}$ -Di*PP*ro-AZTDP **9c** showed a lower stability compared to the corresponding d4TDP derivative **7c** for some unknown reasons. Even in comparison with the Di*PP*ro-compounds with shorter alkyl chains the stability of $CH_3/C_{11}H_{23}$ -Di*PP*ro-AZTDP **9c** was markedly low. Nevertheless, all Di*PP*ro-compounds **7**-**10** hydrolyzed in buffer to form almost exclusively the nucleoside diphosphate.

As reported before, some NMP was formed as well probably due to the unexpected high hydrolytic stability compared to the symmetric Di*PP*ro-compounds bearing short alkyl chains in the masking group. It is important to mention, that the ratio of NDP to NMP in these chemically driven hydrolyses was higher as described for the symmetric compound with longer chains.^{24,25}

 Table 1. Half-lives of DiPPro-d4TDPs 7,8 and DiPPro-AZTDPs 9,10 in PBS and CEM/0 cell

 extracts.

	NDP = d4TDP 7		NDP = AZTDP 9	
Compound	t _{1/2} [h] in PBS	t _{1/2} [h] in CEM/0	t _{1/2} [h] in PBS	t _{1/2} [h] in CEM/0
CH ₃ /C ₇ H ₁₅ -DiPPro-NDP 7a/9a	36	0.04	33	0.02
CH ₃ /C ₉ H ₁₉ -Di <i>PP</i> ro-NDP 7b/9b	40	0.04	35	0.04
$CH_3/C_{11}H_{23}$ -Di <i>PP</i> ro-NDP 7c/9c	48	0.16	32	0.11
CH ₃ /CF ₃ -Ph-Di <i>PP</i> ro-NDP 7d/9d	15	0.03	13	0.04
CH ₃ /CH ₃ -Ph-DiPPro-NDP 7e	58	0.04	-	-
	NDP = d4TDP 8		NDP = AZTDP 10	
C ₄ H ₉ /C ₇ H ₁₅ -Di <i>PP</i> ro-NDP 8a /10a	36	0.71	46	0.83
C ₄ H ₉ /C ₉ H ₁₉ -Di <i>PP</i> ro-NDP 8b /10b	40	1.05	50	1
C ₄ H ₉ /C ₁₁ H ₂₃ -Di <i>PP</i> ro-NDP 8c/10c	52	1.91	63	1.65
C ₄ H ₉ /CF ₃ -Ph-DiPPro-NDP 8d/10d	15	0.88	19	0.83
C ₄ H ₉ /CH ₃ -Ph-DiPPro-NDP 8e	50	0.96	-	-

We also evaluated the chemical stability of Di*PP*ro-AZTDP **9b** and **10b** in Roswell Park Memorial Institute (RPMI) medium in addition to PBS solution, pH 7.3 as described above. Both compounds showed also under these conditions high stability ($t_{1/2} = 19$ h for **9b** and $t_{1/2} = 77$ h for **10b**) and were found to be markedly more stable as compared to the CEM/0 cell extracts.

Incubations in cell extracts of wild type CD_4^+ T-lymphocyte cells *(CEM/0 cell extracts)*: As in the above described studies in PBS and RPMI-culture medium, the half-lives of the Di*PP*rocompounds were found to be independent on the nucleoside analogue attached to the molecule. In contrast to the chemical hydrolysis studies, the prodrugs delivered the corresponding

Journal of Medicinal Chemistry

nucleoside diphosphate selectively within 2 to 7 hours only. In the case of the CH_3/R^2 -DiPPro-NDPs 7 and 9 the hydrolysis led exclusively to the intermediates bearing the long lipophilic acyl group with half-lives of minutes (Table 1). In all cases the acetate ester was selectively cleaved first. However, in comparison the cell extract hydrolysis of the C_4H_0/R^2 -DiPPro-NDPs 8 and 10 released in addition to the intermediate bearing the long alkyl chain acyl group also the C_4H_9 intermediate, although in a markedly smaller ratio. This is due to the higher stability of the pentanovl ($R^1 = C_4H_9$) ester moiety compared to the acetate group present in compounds 7 and 9. But also in these cases the diphosphate was finally formed from both intermediates selectively. The more lipophilic mask had no influence on the half-lives of the cleavage of the short chain acyl group ($R^1 = CH_3$ and $R^1 = C_4H_9$, respectively, Table 1) except for the four DiPPronucleotides with $R^2 = C_{11}H_{23}$ (7c-10c) which showed different stabilities. Figure 2 summarizes as an example the hydrolysis of CH₃/C₉H₁₉-DiPPro-d4TDP 7b in CEM/0 cell extracts followed by means of RP-18-HPLC-chromatography. After 60 min virtually all of the prodrug 7b was consumed. Only the C₉H₁₉-intermediate **11b** and d4TDP appeared as hydrolysis products at that time point. Whether traces of d4TMP were also formed remained unclear because the detection of d4TMP was not possible by RP-18-HPLC-chromatography due to remaining cell extract components that eluted in the chromatogram with identical retention time. On the other hand, by using a HILIC-column (HILIC: hydrophilic interaction liquid chromatography) in the chromatography the separation of d4TMP from cell extract components was successful. With this second analytical method it was proven that very small amounts of d4TMP were also present in the CEM cell extract hydrolysis samples of the DiPPro-compounds.



Figure 2. RP-18-HPLC-chromatograms of the hydrolysis of CH_3/C_9H_{19} -Di*PP*ro-d4TDP **7b** in CEM/0 cell extracts.

In order to study if the detected d4TMP resulted either from the side reaction described above or from a dephosphorylation of d4TDP to give d4TMP due to phosphatases present in the extracts, both d4TDP and AZTDP were incubated as well in the CEM/0 cell extracts and aliquots of the hydrolysis mixtures were analyzed after 2 h, 4 h, 7 h and 10 h of incubation. In both cases d4TMP and AZTMP were detected. The half-life of dephosphorylation of d4TDP was 10 h while that of AZTDP was 19 h in the experimental set-up. For this reason, we concluded that the trace amounts of the nucleoside monophosphates found in the above described hydrolysis studies of the Di*PP*ro-compounds in CEM/0 cell extracts were most likely a result of a dephosphorylation of the formed diphosphates from the prodrugs.

In the case of the C₄H₉/R²-Di*PP*ro-nucleotides **8** and **10** the hydrolysis proceeded via both possible intermediates and then led to the corresponding NDPs. As the half-lives for the starting Di*PP*ro-compounds **1** bearing an acetyl (R¹= CH₃) or a pentanoyl (R¹= C₄H₉) residue, the half-lives for the formed mono-masked intermediates of type **2** were dependent on the acyl ester group. The half-lives of mono-masked intermediates were determined by the decrease of the corresponding peak areas in the HPL-chromatograms of these compounds in the hydrolysis mixtures of the CH₃/R²-Di*PP*ro-nucleotides **7** and **9** in CEM/0 cell extracts after full consumption of the starting Di*PP*ro-compounds. The results are summarized in Table 2.

Table 2. Half-lives of the intermediates 2 and 2' in CEM/0 cell extracts.

	$2 \text{ NDP} = \mathbf{d} 4 \text{TDP}$	2' NDP = AZTDP
Intermediate (with R ²)	t _{1/2} [min] in CEM/0	t _{1/2} [min] in CEM/0
C ₇ H ₁₅ 2a/2'a	99	99
C ₉ H ₁₉ 2b/2'b	50	50
C ₁₁ H ₂₃ 2c/2'c	69	63
Ph-CF ₃ 2d/2'd	690	230
Ph-CH ₃ 2e	870	-

We assumed that the stability would increase with longer chain length. Interestingly, intermediates **2a** and **2'a** were more stable than the ones with longer chain length. Therefore, it seemed that the octanoyl group was a worse substrate for the ester cleaving enzymes.

In the case of the CF_3 -Ph-intermediate **2d** it is surprising that the attached nucleoside had obviously an influence on the stability. The AZT intermediate **2'd** was markedly more labile than the d4T intermediate **2d** for unknown reasons.

Incubations with pig liver esterase: The non-symmetric aliphatic Di*PP*ro-AZTDPs **9a-c** and **10a-c** bearing a R^1 = CH₃ or C₄H₉ in combination with long alkyl chains as the R^2 residue were incubated with pig liver esterase (PLE) to investigate the influence of the chain length on the enzymatic cleavage by this esterase. Furthermore, CH₃/C₉H₁₉-Di*PP*ro-d4TDP **7b** and C₄H₉/C₉H₁₉-Di*PP*ro-d4TDP **8b** were also included to study whether the attached nucleoside analog has an effect on the hydrolysis pathway or on the compound stability.

The results of the hydrolysis experiments in the presence of PLE are summarized in Table 3.

Compound	t _{1/2} [min]	t _{1/2} [min] of R ¹ -	
	of Di <i>PP</i> ro-NDP	intermediate	
CH ₃ /C ₉ H ₁₉ -Di <i>PP</i> ro-d4TDP 7b	0.14	nm.*	
CH ₃ /C ₇ H ₁₅ -DiPPro-AZTDP 9a	0.26	nm.*	
CH ₃ /C ₉ H ₁₉ -DiPPro-AZTDP 9b	0.21	nm.*	
CH ₃ /C ₁₁ H ₂₃ -DiPPro-AZTDP 9c	0.45	nm.*	
C ₄ H ₉ /C ₉ H ₁₉ -Di <i>PP</i> ro-d4TDP 8b	0.19	28	
C ₄ H ₉ /C ₇ H ₁₅ -DiPPro-AZTDP 10a	0.27	15	
C ₄ H ₉ /C ₉ H ₁₉ -DiPPro-AZTDP 10b	0.27	19	
C ₄ H ₉ /C ₁₁ H ₂₃ -DiPPro-AZTDP 10c	1.2	38	

Table 3. Half-lives of the DiPPro-compounds 7-10 in the presence of PLE.

*nm. not measurable

All compounds were rapidly hydrolyzed and delivered the nucleoside diphosphates d4TDP or AZTDP exclusively within a few minutes. The half-lives determined for the CH_3/C_7H_{15} - and CH_3/C_9H_{19} -compounds **9a,b** as well as the C_4H_9/C_7H_{15} and C_4H_9/C_9H_{19} -masked pronucleotides **10a,b** were found to be all in the same range (0.21 min to 0.27 min). The influence of the nucleoside on the half-live was small although Di*PP*ro-d4TDPs **7b,8b** proved to be somewhat

Journal of Medicinal Chemistry

less stable. In contrast, both compounds 9c, 10c (CH₃/C₁₁H₂₃- and C₄H₉/C₁₁H₂₃-Di*PP*ro-AZTDP) bearing the longest aliphatic chain showed higher stability against hydrolysis by PLE.

Remarkably, in the case of the hydrolysis of all CH_3/R^2 -Di*PP*ro **9a-c**,**7b**, exclusively the hydrolysis intermediate bearing the long aliphatic chain was formed although in very small amounts. Therefore, it was not possible to determine the half-lives of these intermediates from the hydrolysis mixtures. This means that both alkanoyl esters were cleaved in almost identical rate although with a slight preference of the C₁-alkanoyl group. As an example the HPL-chromatograms of the incubations of compound **9b** with PLE are shown in Figure 3.



Figure 3. RP-18-HPLC-chromatograms of the PLE-mediated hydrolysis of CH₃/C₉H₁₉-Di*PP*ro-AZTDP **9b**.



Figure 4. RP-18-HPLC-chromatograms of the PLE-mediated hydrolysis of C_4H_9/C_9H_{19} -Di*PP*ro-AZTDP **10b**.

In contrast, incubation of C_4H_9/R^2 -Di*PP*ro-NDPs **10a-c**, **8b** with PLE initially showed high amounts of the C_4H_9 -intermediate (Figure 4, e.g. the 1 min-HPLC chromatogram). In addition, as for the C_7H_{15} - and $C_{11}H_{23}$ -intermediate also the C_9H_{19} -chain bearing intermediate **2'b** was detected in the HPL-chromatograms although to a minor extent. In contrast to the above study with the CH₃-bearing Di*PP*ro-compounds, here obviously the longer lipophilic chain acyl groups are cleaved faster than the C_4H_9 -alkanoyl ester. This can be concluded from the markedly different amounts of the two possible intermediates in the HPL-chromatograms (Figure 4). From both formed intermediates finally AZTDP was released with a half-life of about 19 min as the sole hydrolysis product.

These observations contrasted also with the results obtained in CEM/0 cell extracts and point to a different substrate specificity of the esterase from pig liver and the esterases present in

Journal of Medicinal Chemistry

human CEM/0 cells or other enzymes that play a significant role in the hydrolysis process of the Di*PP*ro-compounds described here.

Comparative hydrolysis studies: Finally, we compared the non-symmetric DiPPro-nucleotides described here with compounds 11 bearing identical masking groups. Therefore, non-symmetric and in addition to the symmetric counterparts DiPPro-AZTDP derivatives with $R=C_4H_9$ (11a), C_6H_{13} (11b) and C_9H_{19} (11c) were incubated in phosphate buffer at pH 7.3 and in CEM/0 cell extracts. The hydrolysis studies in PBS led to significant differences in the ratios of formed NDP and NMP. In all cases the symmetric compounds 11 formed markedly more NMP and less NDP as compared to the studies of the non-symmetric DiPPro-compounds disclosed here. Even more interesting was the comparison of the hydrolysis in CEM/0 cell extracts. Also under these conditions the hydrolysis of the symmetric C₄H₉-DiPPro-AZTDP **11a** and C₆H₉-DiPPro-AZTDP 11b resulted in high NDP/NMP ratios. In contrast, this ratio decreased significantly in the case of the more lipophilic C_9H_{19} -DiPPro-AZTDP **11c** which was correlated to the difference in the rate of metabolism. As mentioned above, the short chain alkyl esters are rapidly cleaved by the esterases present in the extracts and therefore the side reaction did not interfere as in the case of the more stable long chain aliphatic esters. Figure 5 summarizes a comparison of the hydrolysis of CH₃/C₉H₁₉-DiPPro-AZTDP **9b**, C₉H₁₉-DiPPro-AZTDP **11c** and the nucleoside diphosphate AZTDP after 10 hours of incubation. The analysis of the hydrolysis aliquots was performed by means of HILIC-HPL-chromatography. The NDP/NMP ratio in the case of the non-symmetric prodrug 9b was found to be 5:1, for the symmetric prodrug 11c the ratio was only 1.5:1. Interestingly, starting from independently synthesized AZTDP, the ratio was found to be 3:1 after 10 h of incubation in CEM/0 cell extracts.



Figure 5. HPL-chromatograms (HILIC-column) of the incubations of CH_3/C_9H_{19} -Di*PP*ro-AZTDP **9b**, C_9H_{19} -Di*PP*ro-AZTDP **11c** and AZTDP in CEM/0 cell extracts after 10 hours. *components from the cell extracts

The markedly higher ratio found in the case of the non-symmetric Di*PP*ro-prodrug **9b** compared to AZTDP implies that compound **9b** is not susceptible to cleavage at the pyrophosphate moiety. The formation of the NMP only took place by the enzymatic dephosphorylation <u>after</u> AZTDP delivery and thus after cleavage of the masking units. As a consequence, the NDP/NMP ratio in case of the prodrug **9b** was even higher as compared to that of AZTDP for which the dephosphorylation immediately starts at the beginning of the incubation. On the other hand, due to the high stability of the symmetric Di*PP*ro-compound **11c** against enzymatic cleavage of the prodrug moieties, the chemical cleavage of the diphosphate unit in this compound occurred by the above described concurrence reaction which led to AZTMP and consequently to a low NDP/NMP ratio (1.5:1).

Antiviral Activity

Journal of Medicinal Chemistry

All Di*PP*ro-compounds **7-11** were evaluated for their anti-HIV activity in HIV-1- and HIV-2infected wild-type CEM/0 cell cultures and in HIV-2-infected mutant thymidine kinase-deficient CEM/TK⁻ cell cultures. Table 4 summarizes the antiviral and cytostatic data of the Di*PP*rocompounds **7-10** and the parent nucleoside analogues d4T and AZT, respectively, as reference compounds.

D4TDP derivatives **7**,**8** bearing aliphatic ester functions in the acyloxybenzyl units (**a-c**) proved to be antivirally active against HIV-1 and HIV-2 at least in the same concentration range as compared to the parent compound d4T in wild-type CEM/0 cell cultures. Often the inhibitory potency was even somewhat higher as compared to that of the parent nucleoside particularly for compound **8c**. The antiviral activity observed in the wild-type CEM/0 cell cultures was completely retained in the case of the lipophilic Di*PP*ro-compounds **7b** and **8b**,**c** in mutant thymidine-deficient CEM cells (TK⁻). Moreover, the antiviral activity increased with the increase of lipophilicity associated with the introduced acyl-moieties of the Di*PP*ro-prodrugs.

Disappointingly, Di*PP*ro-d4TDPs 7,8 bearing benzoyl ester functions in the masking units (d,e) proved to be less potent as compared to the alkanoyl counterparts. They also were less active as compared to d4T in the wild-type CEM/0 cell cultures and lost more or less all antiviral activity in the TK-deficient CEM cell cultures (EC₅₀: $\geq 10 \ \mu$ M).

Alike the d4TDP-containing Di*PP*ro-compounds, the AZTDP derivatives **9**,**10** were able to inhibit the HIV-1 and HIV-2 induced cytopathogenic effect at very low concentrations in wild-type CEM/0 cell cultures.

Table 4: Anti-HIV and cytostatic activity of the DiPPro-NDPs 7-11.

 $EC_{50}^{a)}[\mu M]$

CC₅₀^{b)} [µM]

-				
	CEM/0	CEM/0	CEM/TK ⁻	
Compound	HIV-1 (HE)	HIV-2 (ROD)	HIV-2 (ROD)	CEM/0
CH ₃ /C ₇ H ₁₅ -DiPPro-d4TDP 7a	0.33 ± 0.11	1.05 ± 0.30	6.31 ± 0.00	33 ± 27
CH ₃ /C ₉ H ₁₉ -DiPPro-d4TDP 7b	0.33 ± 0.11	1.36 ± 0.13	2.32 ± 1.6	15 ± 1
CH ₃ /C ₁₁ H ₂₃ -DiPPro-d4TDP 7c	0.29 ± 0.16	0.28 ± 0.04	1.77 ± 0.33	26 ± 0
CH ₃ /CF ₃ -Ph-DiPPro-d4TDP 7d	0.72 ± 0.16	1.38 ± 0.16	> 10	51 ± 15
CH ₃ /CH ₃ -Ph-DiPPro-d4TDP 7e	0.29 ± 0.057	1.47 ± 0.29	10 ± 0.00	11 ± 6
C ₄ H ₉ /C ₇ H ₁₅ -DiPPro-d4TDP 8a	0.33 ± 0.11	1.47 ± 0.29	6.31 ± 0.00	66 ± 15
C ₄ H ₉ /C ₉ H ₁₉ -DiPPro-d4TDP 8b	0.36 ± 0.046	1.17 ± 0.40	0.83 ± 0.00	20 ± 1
$C_4H_9/C_{11}H_{23}$ -Di <i>PP</i> ro-d4TDP 8c	0.10 ± 0.03	0.28 ± 0.04	0.13 ± 0.07	30 ± 17
C ₄ H ₉ /CF ₃ -Ph-DiPPro-d4TDP 8d	0.40 ± 0.00	0.72 ± 0.16	>10	37 ± 31
C ₄ H ₉ /CH ₃ -Ph-Di <i>PP</i> ro-d4TDP 8e	0.3 ± 0.014	1.45 ± 0.00	10 ± 0.00	14 ± 1
CH ₃ /C ₇ H ₁₅ -DiPPro-AZTDP 9a	0.042 ± 0.008	0.16 ± 0.08	>10	29 ± 3
CH ₃ /C ₉ H ₁₉ -Di <i>PP</i> ro-AZTDP 9b	0.016 ± 0.0	0.14 ± 0.08	>10	20 ± 9
CH ₃ /C ₁₁ H ₂₃ -Di <i>PP</i> ro-AZTDP 9c	0.042 ± 0.008	0.08 ± 0.00	10 ± 2.61	18 ± 3
CH ₃ /CF ₃ -Ph-DiPPro-AZTDP 9d	0.012 ± 0.007	0.06 ± 0.03	>10	100 ± 8
C ₄ H ₉ /C ₇ H ₁₅ -DiPPro-AZTDP 10a	0.036 ± 0.0	0.07 ± 0.01	>10	73 ± 12
C ₄ H ₉ /C ₉ H ₁₉ -Di <i>PP</i> ro-AZTDP 10b	0.034 ± 0.018	0.29 ± 0.18	10 ± 0.00	41 ± 23
$C_4H_9/C_{11}H_{23}$ -Di <i>PP</i> ro-AZTDP 10c	0.013 ± 0.005	0.06 ± 0.02	8.16 ± 2.61	36 ± 28
C ₄ H ₉ /CF ₃ -Ph-DiPPro-AZTDP 10d	0.036 ± 0.0	0.06 ± 0.02	>10	58 ± 54
C4H9-DiPPro-AZTDP 11a	0.011 ± 0.0018	0.054 ± 0.01	>10	60 ± 7
C ₆ H ₁₃ -Di <i>PP</i> ro-AZTDP 11b	0.016 ± 0.0	0.058 ± 0.03	>10	47 ± 8
C ₉ H ₁₉ -Di <i>PP</i> ro-AZTDP 11c	0.026 ± 0.014	0.23 ± 0.15	1.3 ± 0.0	38 ± 10
d4T	0.52 ± 0.32	2.23 ± 1.51	150 ± 9	79 ± 3
AZT	0.044 ± 0.01	0.49 ± 0.57	>250	>250

[a] Antiviral activity in CEM T-lymphocyte cultures: 50% effective concentration; [b] Cytostatic activity: 50% cytostatic concentration to inhibit CEM/0 cell proliferation by 50%.

In mutant thymidine kinase-deficient CEM/TK⁻ cell cultures only the lipophilic compounds 9c, 10b and 10c showed some antiviral activity (EC₅₀: ~ 10 μ M). Interestingly, the symmetric C₉H₁₉-Di*PP*ro-compound 11c was active at 1.3 μ M in the TK-deficient CEM cell cultures. One reason for this pronounced retention of antiviral activity of 11c might be that the lipophilicity of this compound was sufficiently high to allow a passage through the cell membrane. In the hydrolysis studies discussed above it was shown that all compounds bearing an acetyl ester as an acyl moiety in the masking group were cleaved extremely fast. Thus, if these compounds are not

Journal of Medicinal Chemistry

lipophilic enough they will be cleaved in the cell culture medium before the uptake into the cells. Then AZTDP will be formed extracellularly which will be enzymatically dephosphorylated to give AZT. AZT is able to migrate through the membrane and this result in antiviral activity in the wild-type cell cultures. It has been reported that the phosphorylation of AZTMP to AZTDP by dTMP kinase is rate-limiting.^{15,16} However, another reason for the failure of the DiPPro-AZTDP compounds might be that even the delivered AZTDP is not efficiently metabolized into the corresponding triphosphate. In such a case the delivery of AZTDP will not lead to increased antiviral activity. It was already reported that in biochemical activity studies using nucleoside diphosphate kinase (NDP-K), which is generally the enzyme that is involved in the phosphorylation of nucleoside diphosphates into their triphosphate forms, the phosphorylation to AZTTP was rather poor.²⁷ So, maybe there is not only a bottle-neck in the metabolism of AZTMP to AZTDP²⁸ but also a bottle-neck in the conversion of AZTDP to form AZTTP which becomes more prominent if the concentration of AZTDP is increased due to its delivery from a prodrug. From our studies in CEM/0 cell extracts we clearly have proven that the exclusively formed product from the prodrug was AZTDP.

SUMMARY AND CONCLUSION

Non-symmetric Di*PP*ro-NDPs **1** of the anti-HIV active nucleoside analogues d4T and AZT are disclosed here bearing two different masking units attached to the β -phosphate group of the corresponding nucleoside diphosphate. For the synthesis of the title compounds, a new protocol for the synthesis of non-symmetric bis(acyloxybenzyl)-phosphoramidites was developed, which yielded the phosphoramidites in high chemical yields. These were coupled by weak acid catalysis with nucleoside monophosphates followed by oxidation. Using this route, Di*PP*ro-

d4TDP and DiPPro-AZTDP compounds with different combinations of aliphatic and aromatic acyl moieties were obtained in good yields. Hydrolysis studies in phosphate buffer, RPMIculture medium, CEM/0 cell extracts and pig liver esterase confirmed the designed delivery mechanism described above. Half-lives in PBS and RPMI-culture medium were markedly higher than those in CEM/0 cell extracts and incubations with PLE. Although in PBS still both the nucleoside diphosphates and the monophosphates were formed, the exclusive products formed by enzymatic hydrolysis of the prodrugs (cell extracts and PLE) were the nucleoside diphosphates. As compared to the symmetric DiPPro-NDPs with two identical masks reported previously, the ratio of formed NDP to NMP was markedly higher in all media. Most of the nonsymmetric DiPPro-compounds were as active as or even more active than the corresponding parent nucleosides in wild-type CEM/0 cell cultures. Moreover, also high activities were obtained in dependence on the lipophilicity of the DiPPro-d4TDP prodrugs against HIV-2 in mutant CEM/TK⁻ cell cultures. This confirmed that these compounds were efficiently taken-up by the cells and delivered intracellularly a phosphorylated form of d4T, most likely d4T diphosphate. Surprisingly, the good antiviral activities of DiPPro-AZTDPs in the wild-type cell cultures were only weakly retained in the mutant TK-deficient cell cultures. This may point to a possible further bottle-neck in the phosphorylation of the AZTDP to AZTTP. A clear conclusion from the antiviral activity data was that DiPPro-compounds bearing short acetyl ester or benzoyl ester groups in one of the prodrug moieties were not suitable for further development. The first ester group is cleaved too fast and is not lipophilic enough to allow a successful uptake into the target cells while the second type of ester moiety is also not lipophilic enough although we have shown before that the stability of such compounds could be adjusted over a wide range.²⁶

Page 23 of 60

Journal of Medicinal Chemistry

Nevertheless, this report on non-symmetric Di*PP*ro-compounds proved the successful and selective nucleoside diphosphate delivery as a further development of the Di*PP*ro-approach.

EXPERIMENTAL SECTION

General. All experiments involving water-sensitive compounds were conducted under anhydride conditions and nitrogen atmosphere. All solvents were dried over an appropriate drying agent. Triethylamine and acetonitrile were dried by heating under reflux over calcium hydride for several days followed by distillation. Acetonitrile was stored over 3Å molecular sieves. THF was dried by heating under reflux over potassium followed by distillation. Petroleum ether 50-70 for chromatography was distilled before use. The evaporation of solvents was carried out on a rotary evaporator under reduced pressure or using a high-vacuum pump. Preparative radial chromatography was performed by chromatotron (Harrison Research, Model 7924T) with glass plates coated with 1, 2, or 4 mm layers of VWR 60 PF₂₅₄ silica gel containing a fluorescent indicator (VWR no. 7749) were used. Analytical thin-layer chromatography was performed on pre-coated aluminum plates with a 0.2-mm layer of silica gel containing a fluorescent indicator (Macherey & Nagel Xtra Sil UV₂₅₄); sugar-containing compounds were visualized by wetting the plates with 10% sulphuric acid in water and heating with a fan.

Instrumentation. ¹H-NMR Spectroscopy was carried out using a Bruker AMX 400 or AV 400 at 400 MHz or a Bruker AV 600 at 600 MHz with CDCl₃ or MeOD-*d*₄ as internal standards. ¹³C-NMR spectra were recorded on a Bruker AMX 400 or AV 400 at 101 MHz or on a Bruker AV 600 at 150 MHz (CDCl₃ or MeOD-*d*₄ as internal standard). ³¹P-NMR spectra were recorded on a Bruker AV 600 at 243 MHz (H₃PO₄ as internal standard). ¹⁹F-NMR spectra were recorded on a Bruker AV 600 MHz. All ¹H-, ³¹P- and ¹³C-

NMR chemical shifts are quoted in parts per million (ppm). All ¹³C- and ³¹P-NMR spectra were recorded in the proton-decoupled mode. High resolution ESI mass spectra were recorded with an Agilent 6224 ESI-TOF spectrometer in positive or negative modus. MALDI mass spectra were recorded on an ultraflextreme MALDI-TOF-TOF mass spectrometer by Bruker Daltonik with 9-AA as matrix. All HPLC measurements were carried out using a VWR Hitachi 'Elite LaChrom' (pump L-2130, auto sampler L-2200, column oven L-2300, diode array detector L-2455). Analytical Rp18-HPLC: Macherey & Nagel EC 125/3 NUCLEODUR 100-5 C18 EC. As eluents $CH_3CN/[(C_4H_9)_4N]^+CH_3COO^-$ buffer (2 mM, pH 6.0) gradient was used. Method A: 0-24 min CH₃CN (5-80%), 24-29 min CH₃CN (80%), flow rate 1 mL/min. Method B: 0-27 min CH₃CN (5-90%), 27-30 min CH₃CN (90%), flow rate 1 mL/min, column temperature: 25 °C. UV detection at 265 nm. The purities of the DiPPro nucleotides were evaluated by means of HPLC and were in all cases \geq 95%. Analytical HILIC-HPLC: Macherey & Nagel EC 125/4 NUCLEODUR HILIC, 5µm. As eluents CH₃CN/NH₄⁺HCOO⁻ buffer (10 mM, pH 8.3) gradient was used. Method C: 0-5 min CH₃CN (95%), 5-15 min CH₃CN (95-70%), 15-30 min CH₃CN (70%) flow rate 0.6 mL/min, column temperature: 25 °C. UV detection at 265 nm.

Synthesis.

General procedure A: Preparation of acyloxybenzyl bis(di*iso***propylamino)phosphoramidite 6**: To a low concentrated solution of 1 equivalent phosphorus trichloride and 1 equivalent pyridine in THF 1 equivalent 4-acyloxybenzyl alcohol, dissolved in THF, was added dropwise over 1.5 h at -78 °C. After removing of the cold bath the reaction was stirred for about 20 h at room temperature until the total reaction of 4-acyloxybenzyl alcohol. Then 6.1 equivalents *N,N*-di*iso***p**ropylamine were added dropwise at -10 °C. After removing of the

Journal of Medicinal Chemistry

cold bath the reaction was stirred for about one to two days at room temperature. Salts were removed by filtration. The crude product was purified by radial chromatography at the chromatotron with petroleum ether/triethylamine as eluent.

General procedure B: Preparation of non-symmetric bis(acyloxybenzyl)phosphoramidites 5: To a solution of 1.5 equivalents phosphordiamidite in acetonitrile 1 equivalent 4-acyloxybenzyl alcohol and 1 equivalent of a DCI-activator solution (0.25 molar in acetonitrile) also dissolved in acetonitrile were added dropwise at 4 °C. After removing of the cold bath the reaction mixture was stirred for 30 to 60 min. The reaction was stopped by removing the solvent under reduced pressure. The residue was dissolved in petroleum ether/triethylamine (9:1) and filtrated. The crude product was purified by radial chromatography at the chromatotron with petroleum ether/triethylamine as eluent.

General procedure C: Preparation of Di*PP***ro-nucleoside diphosphates**: 1.5 equivalents phosphoramidite **5** were added to a solution of 1 equivalent of the bis(tetra-*n*-butylammonium) nucleoside monophosphate in acetonitrile. The reaction was started by addition of 0.5 equivalents 4,5-dicyanoimidazole (0.25 M solution in acetonitrile) at room temperature. Every 5 min again 0.25 equivalents 4,5-dicyanoimidazole were added up to a total of 1.25 to 1.75 equivalents. After stirring another 5 min after the last addition the mixture was oxidized by addition of 1.5 equivalents *tert*butylhydroperoxid (5.5 molar solution in *n*-decane). The solvents were removed under reduced pressure after 15 min. The compounds were isolated by automatic flash RP18- chromatography (acetonitrile/water). After the tetra-*n*-butylammonium ions were changed with ammonia by elution over DOWEX 50WX8 (NH₄⁺), a second automatic flash RP18- chromatography (acetonitrile/water) followed.

Syntheses of 4-(hydroxymethyl)phenylacetate 3a,²⁴ 4-(hydroxymethyl)phenylpentanoate 3b,²⁵ 4-(hydroxymethyl)phenyloctanoate 4a,²⁴ 4-(hydroxymethyl)phenyldecanoate 4b,²⁵ 4-(hydroxymethyl)phenyldodecanoate 4c,²⁵ 4-(hydroxymethyl)phenyl-4'-trifluoromethyl-benzoate 4d,²⁶ 4-(hydroxymethyl)phenyl-4'-methylbenzoate 4e,²⁶ 4-(hydroxymethyl)phenyl-4' and the series and the se

Syntheses of $(N[n-Bu]_4)_2$ -d4TMP and $(N[n-Bu]_4)_2$ -AZTMP have been described before.^{24,25} The analytical data were identical to those reported before.

Acetoxybenzyl-bis(diisopropylamino)phosphoramidite 6a: General procedure B: 4-Acyloxybenzylalcohol **3a** (2.36 g, 14.4 mmol), dissolved in 15 mL THF and phosphorus trichloride (1.26 mL, 14.4 mmol) and pyridine (1.17 mL, 14.4 mmol) in 35 mL THF. In the following step DIPA (12.4 mL, 87.8 mmol) was added. The product (3.02 g, 7.62 mmol, 53%) was obtained as a colorless wax-like solid. ¹H-NMR: (400 MHz, CDCl₃): δ [ppm] = 7.39-7.37 (m, 2 H, H-2); 7.05-7.04 (m, 2 H, H-3); 4.64-4.63 (m, 2 H, Ph-CH₂); 3.61-3.54 (m, 4 H, CN-H); 2.29 (s, 3 H, H-6); 1.19 (d, ${}^{3}J_{HH} = 6.9$ Hz, 12 H, *iso*Pr); 1.19 (d, ${}^{3}J_{HH} = 7.0$ Hz, 12 H, *iso*Pr); 13 C-NMR: (101 MHz, CDCl₃): δ [ppm] = 169.6 (C-5); 149.6 (C-4); 138.3 (d, {}^{3}J_{CP} = 10.6 Hz, C-1); 127.9 (C-2); 121.3 (C-3); 65.6 (d, ${}^{2}J_{CP}$ = 23.2 Hz, Ph-CH₂); 44.5 (d, ${}^{1}J_{CN}$ = 12.3 Hz, 2x N-CH); 24.7 (d, ${}^{2}J_{CN} = 8.0$ Hz, *iso*Pr); 24.0 (d, ${}^{2}J_{CN} = 5.6$ Hz, *iso*Pr); 21.2 (C-6). 31 P-NMR (162 MHz, CDCl₃, decoupled): $\delta/\text{ppm} = 123.4$; IR $\tilde{\nu}$ [cm⁻]= 2970, 2923, 2901, 1766, 1506, 1454, 1302, 1114, 1049, 950, 741. MS (MALDI) $m/z = \text{calcd for } C_{34}H_{46}N_3O_6P$: 589.331 [M-H⁺+9aminoacridine (9-AA) used as matrix¹, found: 589.309.

Pentanoyloxybenzyl-bis(di*iso***propylamino)phosphoramidite 6b**: General procedure B; 4-Acyloxybenzylalcohol 3b (2.25 g, 10.8 mmol), dissolved in 15 mL THF and phosphorus

trichloride (0.94 mL, 10.8 mmol) and pyridine (0.88 mL, 10.8 mmol) in 35 mL THF. In the following step (9.24 mL, 65.8 mmol) was added. The product (2.65 g, 6.04 mmol, 56%) was obtained as a colorless oil. ¹H-NMR: (400 MHz, CDCl₃): δ [ppm] = 7.38-7.36 (m, 2 H, H-2); 7.04-7.02 (m, 2 H, H-3); 4.64-4.62 (m, 2 H, Ph-CH₂); 3.62-3.52 (m, 4 H, CN-H); 2.29 (t, ³*J*_{HH} = 7.5 Hz, 2 H, H-6); 1.78-1.70 (m, 2H, H-7); 1.50-1.40 (m, 2H, H-8); 1.18 (d, ³*J*_{HH} = 6.8 Hz, 12 H, *iso*Pr); 1.18 (d, ³*J*_{H,H} = 6.9 Hz, 12 H, *iso*Pr); 0.97 (t, ³*J*_{H,H} = 7.4 Hz, 2 H, H-9). ¹³C-NMR: (101 MHz, CDCl₃): δ [ppm] = 172.5 (C-5); 149.7 (C-4); 138.2 (d, ³*J*_{C,P} = 10.7 Hz, C-1); 127.9 (C-2); 121.3 (C-3); 65.7 (d, ²*J*_{C,P} = 23.0 Hz, Ph-CH₂); 44.6 (d, ¹*J*_{C,N} = 12.4 Hz, 2x N-CH); 34.3 (C-6); 27.2 (C-7); 24.8 (d, ²*J*_{C,N} = 7.9 Hz, *iso*Pr); 24.0 (d, ²*J*_{C,N} = 5.7 Hz, *iso*Pr); 22.4 (C-8); 13.9 (C-9). ³¹P-NMR (162 MHz, CDCl₃, decoupled): δ /ppm = 123.5. IR $\tilde{\nu}$ [cm]= 2964, 2930, 2870, 1760, 1507, 1390, 1195, 1182, 1144, 1016, 951, 779. MS (MALDI) *m/z* = calcd for C₃₇H₅₂N₃O₆P: 631.386 [M+ 9-aminoacridine (9-AA) used as matrix], found: 631.368.

CH₃/C₇H₁₅-phosphoramidite 5a: General procedure C; Acetoxybenzyl-bis(di*iso*propylamino)phosphoramidite 6a (1.24 g, 3.12 mmol) dissolved in 15 mL acetonitrile and 4-(hydroxymethyl)phenyloctanoate 4a (520 mg, 2.08 mmol) and DCI-activator solution (8.32 mL, 2.08 mmol) dissolved in 5 mL acetonitrile. The product (932 mg, 1.71 mmol, 82%) was obtained as a colorless oil. ¹H-NMR: (400 MHz, CDCl₃): δ [ppm] = 7.36-7.34 (m 4H, H-2, H-2^c); 7.05-7.03 (m, 4H, H-3, H-3^c); 4.77-4.73 (m, 2H, Ph-CH₂); 4.70-4.66 (m, 2H, Ph-CH₂); 3.72-3.66 (m, 2H, 2x NC-H); 2.56-2.53 (m, 2H, H-6^c); 2.29 (s, 3H, -CH₃); 1.78-1.73 (m, 2H, H-7^c); 1.44-1.29 (m, 12H, H-8^c – H-11^c); 1.20 (d, ³*J*_{H,H} = 6.8 Hz, 12H, *iso*Pr); 0.90 (d, ³*J*_{H,H} = 6.8 Hz, 3H, H-12^c). ¹³C-NMR: (101 MHz, CDCl₃): δ [ppm] = 172.5 (C-5^c); 169.6 (C-5); 150.1 (C-4^c)</sup>; 150.0 (C-4); 137.3 (d, ³*J*_{C,P} = 7.6 Hz, C-1^c); 137.1 (d, ³*J*_{C,P} = 7.5 Hz, C-1); 128.1, 128.1 (C-2,

C-2[°]); 121.5, 121.5 (C-3, C-3[°]); 65.0 (d, ${}^{2}J_{C,P}$ = 4.7 Hz, Ph-CH₂); 64.9 (d, ${}^{2}J_{C,P}$ = 4.8 Hz, Ph-CH₂); 43.3 (d, ${}^{1}J_{C,N}$ = 12.5 Hz, N-CH); 34.6 (C-6[°]); 31.8, 29.2, 29.6, 22.7, (C-8[°] – C-11[°]); 25.1 (C-7[°]); 24.8, 24.8 (2x *iso*Pr); 21.3 (C-6); 14.2 (C-12[°]).³¹P-NMR (162 MHz, CDCl₃, decoupled): δ /ppm = 148.0. IR $\tilde{\nu}$ [cm⁻]= 2965, 2929, 2859, 1760, 1608, 1507, 1366, 1192, 1163, 1005, 973, 756, 504. HRMS (ESI⁺) *m/z* = calcd for C₃₀H₄₅NO₆P: 546.2979 [M+H⁺]⁺, found: 546.2951.

CH₃/C₉H₁₉-phosphoramidite 5b: General procedure C; Acetoxybenzyl-bis(di*iso*propylamino)phosphoramidite 6a (520 mg, 1.31 mmol) dissolved in 10 mL acetonitrile and 4-(hydroxymethyl)phenyldecanoate 4b (244 mg, 0.88 mmol) and DCI-activator solution (3.52 mL, 0.88 mmol) dissolved in 6 mL acetonitrile. The product (437 mg, 0.76 mmol, 87%) was obtained as a colorless oil. ¹H-NMR: (400 MHz, CDCl₃): δ [ppm] = 7.36-7.34 (m 4H, H-2, H-2'); 7.05-7.02 (m, 4H, H-3, H-3'); 4.77-4.64 (m, 4H, 2x Ph-CH₂); 3.73-3.64 (m, 2H, 2x NC-H); 2.56-2.52 (m, 2H, H-6'); 2.29 (s, 3H, -CH₃); 1.79-1.71 (m, 2H, H-7'); 1.44-1.27 (m, 12H, H-8' - H-13'); 1.20 (d, ${}^{3}J_{HH}$ = 6.8 Hz, 12H, *iso*Pr); 0.89 (d, ${}^{3}J_{HH}$ = 6.8 Hz, 3H, H-14'). 13 C-NMR: (101 MHz, CDCl₃): δ [ppm] = 172.5 (C-5'); 169.6 (C-5); 150.2 (C-4'); 150.0 (C-4); 137.2 (d, ${}^{3}J_{CP} = 7.4$ Hz, C-1, C-1'); 128.2, 128.1 (C-2, C-2'); 121.5, 121.5 (C-3, C-3'); 65.1 (d, ${}^{2}J_{CP} = 3.2$ Hz, Ph-CH₂); 64.9 (d, ${}^{2}J_{CP}$ = 3.4 Hz, Ph-CH₂); 43.3 (d, ${}^{1}J_{CN}$ = 12.4 Hz, N-CH); 34.6 (C-6'); 32.0, 29.6, 29.4, 29.3, 22.8, (C-8' - C-13'); 25.1 (C-7'); 24.8, 24.8 (2x isoPr); 21.3 (C-6); 14.2 (C-14').³¹P-NMR (162 MHz, CDCl₃, decoupled): $\delta/\text{ppm} = 148.0$. IR $\tilde{\nu}$ [cm⁻]= 2964, 2926, 2855, 1759, 1507, 1192, 1005, 972, 755, 503. HRMS (ESI⁺) m/z = calcd for C₃₂H₄₉NO₆P: 574.3292 [M+H⁺]⁺, found: 574.3196.

CH₃/C₁₁H₂₃-phosphoramidite 5c: General procedure C: Acetoxybenzyl-bis(di*iso*propylamino)phosphoramidite 6a (620 mg, 1.57 mmol) dissolved in 12 mL acetonitrile and 4-(hydroxymethyl)phenyl-dodecanoate 4c (319 mg, 1.04 mmol) and DCI-activator solution (4.16 mL, 1.04 mmol) in 5 mL acetonitrile. The product (453 mg, 0.75 mmol, 72%) was obtained as a colorless oil. ¹H-NMR: (400 MHz, CDCl₃): δ [ppm] = 7.36-7.33 (m 4H, H-2, H-2'); 7.05-7.02 (m, 4H, H-3, H-3'); 4.78-4.64 (m, 4H, 2x Ph-CH₂); 3.73-3.64 (m, 2H, 2x NC-H), 2.56-2.52 (m, 2H, H-6'); 2.29 (s, 3H, -CH₃); 1.78-1.71 (m, 2H, H-7'); 1.43-1.26 (m, 16H, H-8' -H-15'); 1.20 (d, ${}^{3}J_{HH} = 6.8$ Hz, 12H, *iso*Pr); 0.89 (d, ${}^{3}J_{HH} = 6.8$ Hz, 3H, H-16'). 13 C-NMR: (101 MHz, CDCl₃): δ [ppm] = 172.5 (C-5'); 169.5 (C-5); 150.2 (C-4'); 150.0 (C-4); 137.2 (C-1, C-1'); 128.2 (C-2, C-2'); 121.5 (C-3, C-3'); 65.0 (d, ${}^{2}J_{CP} = 18.4$ Hz, Ph-CH₂); 65.0 (d, ${}^{2}J_{CP} = 18.9$ Hz, Ph-CH₂); 43.3 (d, ${}^{1}J_{CN}$ = 12.2 Hz, N-CH); 34.6 (C-6'); 29.8, 29.7, 29.6, 29.5, 29.4, 29.4, 29.3 (C-8' - C-15'); 25.1 (C-7'); 24.8, 24.8 (2x isoPr); 21.3 (C-6); 14.4 (C-16').³¹P-NMR (162 MHz, CDCl₃, decoupled): δ /ppm = 148.0. IR \tilde{v} [cm⁻]= 2964, 2925, 2854, 1760, 1608, 1507, 1365, 1192, 1163, 1005, 973, 756, 503. HRMS (ESI⁺) m/z = calcd for C₃₄H₅₃NO₆P: 602.3605 $[M+H^+]^+$, found: 602.4100.

CH₃/CF₃-Ph-phosphoramidite 5d: General procedure C; Acetoxybenzyl-bis(di*iso*propylamino)phosphoramidite 6a (544 mg, 1.37 mmol) dissolved in 12 mL acetonitrile and 4-(hydroxymethyl)phenyl-4'-trifluoromethylbenzoate 4d (271 g, 0.915 mmol) and DCI-activator solution (3.66 mL, 0.915 mmol) in 5 mL acetonitrile. The product (360 mg, 0.61 mmol, 67%) was obtained as a colorless oil. ¹H-NMR: (400 MHz, CDCl₃): δ [ppm] = 8.33-8.32 (m 2H, H-7'); 7.79-7.78 (m, 2H, H-8'); 7.43-7.41 (m, 2H, H-2'); 7.37-7.36 (m, 2H, H-2); 7.19-7.18 (m, 2H, H-3'); 7.06-7.05 (m, 2H, H-3); 4.81- 4.67 (m, 4H, 2x Ph-CH₂); 3.74- 3.68 (m, 4H, CN-H); 2.29 (s, 3H, CH₃); 1.22 (d, ${}^{3}J_{H,H} = 6.7$ Hz, 6H, *iso*Pr); 1.22 (d, ${}^{3}J_{H,H} = 6.5$ Hz, 6H, *iso*Pr). 13 C-NMR (101 MHz, CDCl₃): δ [ppm] = 169.7 (C-5); 164.2 (C-5'); 150.0 (C-4); 149.9 (C-4); 137.7 (d, ${}^{3}J_{C,P} = 7.5$ Hz, C-1'); 137.2 (d, ${}^{3}J_{C,P} = 7.8$ Hz, C-1); 135.2 (d, ${}^{3}J_{C,F} = 33$ Hz, C-9'), 133.0 (C-6'); 130.7 (C-8'); 128.3 (C-2'); 128.3 (C-2); 125.8 (d, ${}^{3}J_{C,F} = 3.6$ Hz, C-7'); 125.7 (CF₃); 121.5, 121.5 (C-3, C-3'); 65.0 (d, ${}^{2}J_{C,P} = 18.5$ Hz, Ph-CH₂); 65.0 (d, ${}^{2}J_{C,P} = 18.6$ Hz, Ph-CH₂); 43.2 (d, ${}^{1}J_{C,N} = 12.3$ Hz, N-CH); 24.8, 24.8 (d, ${}^{2}J_{C,N} = 7.3$ Hz, 2x *iso*Pr); 21.3 (C-6). 19 F-NMR (565 MHz, CDCl₃): δ /ppm = 67.1. 31 P-NMR (162 MHz, CDCl₃, decoupled): δ /ppm = 148.1. IR $\tilde{\nu}$ [cm⁻] = 2968, 2932, 2869, 1762, 1741, 1608, 1506, 1324, 1264, 1193, 1129, 1075, 1013, 770, 552. HRMS (ESI⁺) *m/z* = calcd for C₃₀H₃₄F₃NO₆P: 592.2071 [M+H⁺]⁺, found: 592.2209.

CH₃/CH₃-Ph-phosphoramidite 5e: General procedure C; Acetoxybenzyl-bis(d*iso*propylamino)phosphoramidite 6a (505 mg, 1.27 mmol) dissolved in 10 mL acetonitrile and 4-(hydroxymethyl)phenyl-4'-methylbenzoate 4e (205 mg, 0.85 mmol) and DCI-activator solution (3.40 mL, 0.85 mmol) in 5 mL acetonitrile. The product (323 mg, 0.60 mmol, 71%) was obtained as a colorless oil. ¹H-NMR: (400 MHz, CDCl₃): δ [ppm] = 8.11-8.09 (m 2H, H-7'/H-8'); 7.42-7.36 (m, 4H, H-2, H-2'); 7.32-7.30 (m, 2H, H-7'/H-8'); 7.19-7.17 (m, 2H, H-3'); 7.07-7.05 (m, 2H, H-3); 4.81- 4.67 (m, 4H, 2x Ph-CH₂); 3.76-3.67 (m, 4H, CN-H); 2.45 (s, 3H, -Ph-CH₃); 2.29 (s, 3H, -CH₃); 1.22 (d, ³J_{H,H} = 6.8 Hz, 6H, *iso*Pr); 1.22 (d, ³J_{H,H} = 6.8 Hz, 6H, *iso*Pr). ¹³C-NMR (101 MHz, CDCl₃): δ [ppm] = 169.6 (C-5); 165.4 (C-5'); 150.3 (C-4); 150.0 (C-4'); 144.5 (C-9'); 137.3 (d, ²J_{C,P} = 8.0 Hz, C-1/C-1'); 137.2 (d, ²J_{C,P} = 7.8 Hz, C-1/C-1'); 130.3 (C-7'); 129.4 (C-8'); 128.2, 128.2 (C-2/C-2'); 127.0 (C-6'); 121.7, 121.5 (C-3, C-3'); 65.0 (d, ²J_{C,P} = 18.4 Hz, Ph-CH₂); 65.1 (d, ²J_{C,P} = 18.3 Hz, Ph-CH₂); 43.3 (d, ¹J_{C,N} = 12.4 Hz, N-CH); 24.8, 24.8 (d, ²J_{C,N} = 7.2 Hz, 2x *iso*Pr); 21.9 (Ph-CH₃); 21.3 (C-6). ³¹P-NMR (162 MHz, CDCl₃)

Journal of Medicinal Chemistry

decoupled): $\delta/\text{ppm} = 147.9$. IR $\tilde{\nu}$ [cm⁻]= 2966, 2930, 2868, 1762, 1735, 1610, 1507, 1366, 1192, 1177, 1003, 972, 783, 505. HRMS (ESI⁺) m/z = calcd for C₃₀H₃₇NO₆P: 538.2553 [M+H⁺]⁺, found: 538.2359.

 C_4H_9/C_7H_{15} -phosphoramidite 5f: General procedure C; Pentanoyloxybenzyl-bis(diisopropylamino)phosphoramidite 6b (350 mg, 0.798 mmol) dissolved in 8 mL acetonitrile and 4-(hydroxymethyl)phenyloctanoate 4a (133 mg, 0.532 mmol) and DCI-activator solution (2.13 mL, 0.532 mmol) in 4 mL acetonitrile. The product (281 mg, 0.478 mmol, 90%) was obtained as a colorless oil. ¹H-NMR: (400 MHz, CDCl₃): δ [ppm] = 7.35-7.34 (m 4H, H-2, H-2'); 7.04-7.02 (m, 4H, H-3, H-3'); 4.76-4.73 (m, 2H, Ph-CH₂); 4.69-4.65 (m, 2H, Ph-CH₂); 3.72-3.65 (m, 2H, 2x NC-H), 2.56 (t, ${}^{3}J_{H,H}$ = 6.4 Hz, 2H, H-6'); 2.54 (t, ${}^{3}J_{H,H}$ = 6.4 Hz, 2H, H-6); 1.76-1.72 (m, 4H, H-7, H-7'); 1.48-1.29 (m, 10H, H-8, H-8' – 11'); 1.20 (d, ${}^{3}J_{H,H}$ = 6.8 Hz, 12H, *iso*Pr); 0.97 (d, ${}^{3}J_{HH}$ = 7.4 Hz, 3H, H-9); 0.89 (d, ${}^{3}J_{HH}$ = 6.8 Hz, 3H, H-12⁽⁾). 13 C-NMR: (101 MHz, CDCl₃): δ [ppm] = 172.5 (C-5, C-5'); 150.1 (C-4, C-4'); 137.1 (d, ${}^{3}J_{CP}$ = 7.4 Hz, C-1, C-1'); 128.2 (C-2, C-2'); 121.5 (C-3, C-3'); 65.0 (d, ${}^{2}J_{CP}$ = 18.1 Hz, 2x Ph-CH₂); 43.3 (d, ${}^{1}J_{CN}$ = 12.4 Hz, N-CH); 34.6, 34.3 (C-6, C-6'); 31.8, 29.2, 29.1, 22.8, 22.4 (C-8, C-8' - C-11'); 27.2, 25.1 (C-7, C-7'); 24.8, 24.8 (2x isoPr); 14.2 (C-12'); 13.9 (C-9). ³¹P-NMR (162 MHz, CDCl₃, decoupled): $\delta/\text{ppm} = 148.0$. IR $\tilde{\nu}$ [cm⁻]= 2964, 2929, 2861, 1758, 1608, 1507, 1364, 1197, 1163. 1001, 973, 754, 504. HRMS (ESI⁺) m/z = calcd for C₃₃H₅₁NO₆P: 588.3449 [M+H⁺]⁺, found: 588.3379.

 C_4H_9/C_9H_{19} -phosphoramidite 5g: General procedure C; Pentanoyloxybenzyl-bis(di*iso*-propylamino)phosphoramidite 6b (427 mg, 0.974 mmol) dissolved in 8 mL acetonitrile and 4-

(hydroxymethyl)phenyldecanoate **4b** (181 mg, 0.649 mmol) and DCI-activator solution (2.60 mL, 0.649 mmol) in 4 mL acetonitrile. The product (344 mg, 0.559 mmol, 86%) was obtained as a colorless oil. ¹H-NMR: (400 MHz, CDCl₃): δ [ppm] = 7.35-7.34 (m 4H, H-2, H-2'); 7.04-7.02 (m, 4H, H-3, H-3'); 4.76-4.73 (m, 2H, Ph-CH₂); 4.69-4.65 (m, 2H, Ph-CH₂); 3.72-3.65 (m, 2H, 2x NC-H), 2.56 (t, ³*J*_{H,H} = 6.8 Hz, 2H, H-6'); 2.54 (t, ³*J*_{H,H} = 6.8 Hz, 2H, H-6); 1.77-1.72 (m, 4H, H-7, H-7'); 1.48-1.38 (m, 4H, H-8', H-8); 1.35-1.26 (m, 10H, H-9' – H-13'); 1.20 (d, ³*J*_{H,H} = 6.8 Hz, 12H, *iso*Pr); 0.97 (d, ³*J*_{H,H} = 7.4 Hz, 3H, H-9); 0.89 (d, ³*J*_{H,H} = 6.9 Hz, 3H, H-14'). ¹³C-NMR: (101 MHz, CDCl₃): δ [ppm] = 172.5 (C-5, C-5'); 150.1 (C-4, C-4'); 137.1 (d, ³*J*_{C,P} = 7.6 Hz, C-1, C-1'); 128.2 (C-2, C-2'); 121.5 (C-3, C-3'); 65.0 (d, ²*J*_{C,P} = 18.1 Hz, 2x Ph-CH₂); 43.3 (d, ¹*J*_{C,N} = 12.6 Hz, N-CH); 34.6, 34.3 (C-6, C-6'); 32.0, 29.6, 29.4, 22.8, (C-9' – C-13'); 29.3 (C-8/C-8'); 27.2, 25.1 (C-7, C-7'); 24.8, 24.8 (2x *iso*Pr); 22.4 (C-8/C-8'); 14.2, 13.9 (C-9, C-14').³¹P-NMR (162 MHz, CDCl₃, decoupled): δ /ppm = 148.0. IR $\tilde{\nu}$ [cm⁻] = 2962, 2927, 2856, 1758, 1507, 1364, 1198, 1138, 1104, 973, 754, 504. HRMS (ESI⁺) *m/z* = calcd for C₃₅H₃₅NO₆P: 616.3762 [M+H⁺]⁺, found: 616.3753.

C₄H₉/C₁₁H₂₃-phosphoramidite 5h: General procedure C; Pentanoyloxybenzyl-bis(di*iso*propylamino)phosphoramidite **6b** (376 mg, 1.08 mmol) dissolved in 8 mL acetonitrile and 4-(hydroxymethyl)-phenyldodecanoate **4c** (220 mg, 0.72 mmol) and DCI-activator solution (2.88 mL, 0.72 mmol) in 4 mL acetonitrile. The product (174 mg, 0.242 mmol, 34%) was obtained as a colorless oil. ¹H-NMR: (400 MHz, CDCl₃): δ [ppm] = 7.35-7.34 (m 4H, H-2, H-2^c); 7.04-7.02 (m, 4H, H-3, H-3^c); 4.77-4.72 (m, 2H, Ph-CH₂); 4.69-4.65 (m, 2H, Ph-CH₂); 3.72-3.65 (m, 2H, 2x NC-H), 2.57-2.53 (m, 2H, H-6, H-6^c); 1.78-1.71 (m, 4H, H-7, H-7^c); 1.49-1.26 (m, 18H, H-8, H-8^c – H-15^c); 1.20 (d, ³J_{H H} = 6.8 Hz, 12H, *iso*Pr); 0.97 (d, ³J_{H H} = 7.4 Hz, 3H, H-

9); 0.88 (d, ${}^{3}J_{H,H}$ = 6.9 Hz, 3H, H-16[°]). 13 C-NMR: (101 MHz, CDCl₃): δ [ppm] = 172.5 (C-5, C-5[°]); 150.0 (C-4, C-4[°]); 136.9 (m, C-1, C-1[°]); 128.1 (C-2, C-2[°]); 121.5 (C-3, C-3[°]); 65.0 (d, ${}^{2}J_{C,P}$ = 18.1 Hz, 2x Ph-CH₂); 43.3 (d, ${}^{1}J_{C,N}$ = 12.5 Hz, N-CH); 34.6, 34.3 (C-6, C-6[°]); 32.1, 29.7, 29.4, 29.3, 22.8, 22.4 (C8, C-8[°] - C-13[°]); 27.1, 25.1 (C-7, C-7[°]); 24.8, 24.8 (2x *iso*Pr); 14.2 (C-14[°]); 13.9 (C-9). 31 P-NMR (162 MHz, CDCl₃, decoupled): δ /ppm = 148.0. IR $\tilde{\nu}$ [cm⁻] = 2962, 2925, 2855, 1758, 1507, 1311, 1198, 1138, 1004, 973, 755, 503. HRMS (ESI⁺) *m/z* = calcd for C₃₇H₅₉NO₆P: 644.4075 [M+H⁺]⁺, found: 644.4068.

C₄H₉/CF₃-Ph-phosphoramidite 5i: General procedure C; Pentanoyloxybenzyl-bis(diisopropylamino)phosphoramidite **6b** (717 mg, 1.64 mmol) dissolved in 12 mL acetonitrile and 4-(hydroxymethyl)phenyl-4'-trifluoromethylbenzoate 4d (323 mg, 1.09 mmol) and DCI-activator solution (4.36 mL, 1.09 mmol) in 5 mL acetonitrile. The product (622 mg, 0.981 mmol, 90%) was obtained as a colorless oil. ¹H-NMR: (400 MHz, CDCl₃): δ [ppm] = 8.33-8.31 (m 2H, H-7'); 7.79-7.77 (m, 2H, H-8'); 7.43-7.41 (m, 2H, H-2'); 7.37-7.35 (m, 2H, H-2); 7.19-7.17 (m, 2H, H-3'); 7.05-7.03 (m, 2H, H-3); 4.81-4.66 (m, 4H, 2x Ph-CH₂); 3.75-3.66 (m, 4H, CN-H); 2.56 (t, ${}^{3}J_{\rm H,H}$ = 7.5 Hz, 2H, H-6); 1.78-1.70 (m, 2H, H-7); 1.49-1.40 (m, 2H, H-8); 1.22 (d, ${}^{3}J_{H,H} = 6.8$ Hz, 6H, *iso*Pr); 1.21 (d, ${}^{3}J_{H,H} = 6.8$ Hz, 6H, *iso*Pr); 0.97 (t, ${}^{3}J_{H,H} = 7.3$ Hz, 2H, H-9). ¹³C-NMR (101 MHz, CDCl₃): δ [ppm] = 172.4 (C-5); 164.1 (C-5'); 150.0 (C-4); 149.9 (C-4'); 137.7 (C-1'); 137.1 (C-1); 135.0 (C-6'), 132.9 (C-9'); 130.7 (C-7'); 128.3 (C-2'); 128.2 (C-2); 125.8 (d, ${}^{3}J_{C,F}$ = 3.7 Hz, C-8'); 125.5 (CF₃); 121.5, 121.5 (C-3, C-3'); 65.0 (d, ${}^{2}J_{C,P}$ = 18.5 Hz, Ph-CH₂); 65.0 (d, ${}^{2}J_{C,P}$ = 18.3 Hz, Ph-CH₂); 43.2 (d, ${}^{1}J_{C,N}$ = 12.4 Hz, N-CH); 34.3 (C-6); 27.2 (C-7); 24.8, 24.8 (d, ${}^{2}J_{C,N}$ = 7.3 Hz, 2x *iso*Pr); 22.4 (C-8); 13.9 (C-9). 19 F-NMR (565 MHz, CDCl₃): $\delta/\text{ppm} = 63.2$. ³¹P-NMR (162 MHz, CDCl₃, decoupled): $\delta/\text{ppm} = 148.1$. IR $\tilde{\nu}$ [cm⁻]= 2968,

2932, 2872, 1737, 1505, 1325, 1266, 1065, 1014, 969, 712, 504. HRMS (ESI⁺) m/z = calcd for C₃₃H₄₀F₃NO₆P: 616.3692 [M+H⁺]⁺, found: 616.3753.

General C₄H₉/CH₃-Ph-phosphoramidite 5j: procedure C; Pentanoyloxybenzylbis(diisopropylamino)phosphoramidite 6b (396 mg, 0.903 mmol) dissolved in 8 mL acetonitrile and 4-(hydroxymethyl)-phenyl-4'-methylbenzoate 4e (146 mg, 0.602 mmol) and DCI-activator solution (2.4 mL, 0.60 mmol) in 4 mL acetonitrile. The product (348 mg, 0.60 mmol, 99%) was obtained as a colorless oil. ¹H-NMR: (400 MHz, CDCl₃): δ [ppm] = 8.10-8.08 (m 2H, H-7'/H-8'); 7.40-7.39 (m, 2H, H-2'); 7.36-7.35 (m, 2H, H-2); 7.31-7.30 (m, 2H, H-7'/H-8'); 7.17-7.16 (m, 2H, H-3'); 7.05-7.03 (m, 2H, H-3); 4.79-4.67 (m, 4H, 2x Ph-CH₂); 3.73-3.67 (m, 4H, CN-H); 2.55 (t, ${}^{3}J_{H,H}$ = 7.5 Hz, 2H, H-6); 2.45 (s, 3H, -Ph-CH₃); 1.76-1.71 (m, 2H, H-7); 1.48-1.41 (m, 2H, H-8); 1.21 (d, ${}^{3}J_{HH} = 6.8$ Hz, 6H, *iso*Pr); 1.21 (d, ${}^{3}J_{HH} = 6.7$ Hz, 6H, *iso*Pr); 0.97 (t, ${}^{3}J_{\text{H,H}}$ = 7.4 Hz, 2H, H-9). 13 C-NMR (101 MHz, CDCl₃): δ [ppm] = 172.5 (C-5); 164.4 (C-5'); 150.3 (C-4); 150.1 (C-4'); 144.5 (C-9'); 137.2 (d, ${}^{2}J_{C,P}$ = 7.6 Hz, C-1/C-1'); 137.2 (d, ${}^{2}J_{C,P}$ = 7.6 Hz, C-1/C-1'); 130.4 (C-7'); 129.4 (C-8'); 128.2, 128.2 (C-2/C-2'); 127.0 (C-6'); 121.7, 121.5 (C-3, C-3'); 65.0 (d, ${}^{2}J_{CP} = 18.2$ Hz, Ph-CH₂); 65.0 (d, ${}^{2}J_{CP} = 18.1$ Hz, Ph-CH₂); 43.2 (d, ${}^{1}J_{CN} =$ 12.4 Hz, N-CH); 34.3 (C-6); 27.2 (C-7); 24.8, 24.8 (d, ${}^{2}J_{CN}$ = 7.3 Hz, 2x isoPr); 22.4 (C-8); 21.9 (Ph-CH₃); 13.9 (C-9). ³¹P-NMR (162 MHz, CDCl₃, decoupled): $\delta/\text{ppm} = 144.0$. IR $\tilde{\nu}$ [cm⁻]= 2965, 2931, 2871, 1758, 1736, 1610, 1507, 1264, 1197, 1007, 972, 756, 506. HRMS (ESI⁺) $m/z = \text{calcd for } C_{33}H_{43}NO_6P: 616.3692 [M+H^+]^+, \text{ found: } 616.3753.$

Journal of Medicinal Chemistry

Syntheses of symmetric bis(4-pentanoyloxymethyl)-, bis(4-heptanoyloxymethyl)- and bis(4-decanoyloxymethyl)-*N*,*N*-di*iso*propylaminophosphoramidite have been described before.^{24,25} The analytical data were identical.

Ammonium-CH₃/C₇H₁₅-Di*PP*ro-d4TDP 7a: General procedure E: Phosphoramidite 5a (144 mg, 0.26 mmol), d4T monophosphate (110 mg, 0.18 mmol) and 4,5-dicyanoimidazole activator solution (1.04 µL, 0.26 mmol) in 4 mL acetonitrile. Oxidation by addition of tertbutylhydroperoxide (5.5 molar solution in n-decane, 47 µL, 0.26 mmol). The product (108 mg, 0.14 mmol, 79%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 7.76 (s, 1H, H-6); 7.41-7.37 (m, 4H, H-c, H-c'); 7.08-7.04 (m, 4H, H-c'); 7.08-7.04 (m, H-d, H-d'); 6.95-6.95 (m, 1H, H-1'); 6.39-6.38 (m, 1H, H-3'); 5.84-5.83 (m, 1H, H-2'); 5.12-5.09 (m, 4H, H-a, H-a'); 4.95 (bs, 1H, H-4'); 4.22-4.15 (m, 2H, H-5'); 2.58 (t, ${}^{3}J_{HH} = 7.4$ Hz, 2H, Hg'); 2.27 (s, 3H, CH₃); 1.89 (s, 3H, H-7); 1.73 (dt, ${}^{3}J_{H,H} = 7.4$ Hz, ${}^{3}J_{H,H} = 7.4$ Hz, 2H, H-h'); 1.45-1.29 (m, 8H, H-i' – H-l'); 0.92 (t, ${}^{3}J_{HH} = 6.8$ Hz, 3H, H-m'). 13 C-NMR: (150 MHz, MeOH-d4); δ $[ppm] = 173.9 (C-f^{\circ}); 171.2 (C-f); 166.7 (C-4); 152.8 (C-2); 152.5 (C-e, C-e^{\circ}); 138.7 (C-6); 135.3$ (C-3'); 135.0 (C-b', C-b); 130.4 (2xd, ${}^{4}J_{C,P} = 4.8 \text{ Hz}$, ${}^{4}J_{C,P} = 4.8 \text{ Hz}$, C-c, C-c'); 127.6 (C-2'); 122.9 (C-d, C-d'); 112.1 (C-5); 90.8 (C-1'); 86.9 (d, ${}^{3}J_{CP} = 9.0$ Hz, C-4'); 70.2 (d, ${}^{2}J_{CP} = 5.2$ Hz, C-a, C-a'); 67.2 (d, ${}^{3}J_{CP} = 5.5$ Hz, C-5'); 35.0 (C-g'); 32.9, 30.1, 30.1, 23.7 (C-i' - C-l'); 26.0 (Ch'); 20.9 (CH₃); 14.4 (C-m'); 12.5 (C-7). ³¹P-NMR (243 MHz, MeOH-d4, decoupled): $\delta/\text{ppm} = -$ 12.07 (d, ${}^{2}J_{PP} = 20.0$ Hz, 1P, P_B); -12.92 (d, ${}^{2}J_{PP} = 20.7$ Hz, 1P, P_a). HRMS: (ESI⁺, m/z) calcd. for $C_{34}H_{42}N_2O_{14}P_2Na$: 787.2009 [M+Na]⁺, found: 787.1983. RP18-HPLC: Method A, $R_t = 17.1 \text{ min.}$

Ammonium-CH₃/C₉H₁₉-Di*PP***ro-d4TDP 7b**: General procedure E; Phosphoramidite **5b** (165 mg, 0.29 mmol), d4T monophosphate (120 mg, 0.19 mmol) and 4,5-dicyanoimidazole activator solution (1.14 mL, 0.29 mmol) in 4 mL acetonitrile. Oxidation by addition of tertbutylhydroperoxide (5.5 molar solution in *n*-decane, 53 µL, 0.29 mmol). The product (106 mg, 0.031 mmol, 54%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 7.66 (s, 1H, H-6); 7.41-7.37 (m, 4H, H-c, H-c'); 7.08-7.04 (m, 4H, H-d, H-d'); 6.95-6.95 (m, 1H, H-1'); 6.39-6.38 (m, 1H, H-3'); 5.84-5.83 (m, 1H, H-2'); 5.12-5.09 (m, 4H, H-a, H-a'); 4.95 (bs, 1H, H-4'); 4.22-4.15 (m, 2H, H-5'); 2.58 (t, ${}^{3}J_{HH} = 7.4$ Hz, 2H, Hg'); 2.27 (s, 3H, CH₃); 1.89 (s, 3H, H-7); 1.72 (dt, ${}^{3}J_{H,H} = 7.4$ Hz, ${}^{3}J_{H,H} = 7.5$ Hz, 2H, H-h'); 1.44-1.30 (m, 12H, H-i' – H-n'); 0.91 (t, ${}^{3}J_{H,H} = 6.6$ Hz, 3H, H-o'). 13 C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.9 (C-f'); 171.1 (C-f); 166.7 (C-4); 152.8 (C-2); 152.5 (C-e, C-e'); 138.7 (C-6); 135.3 (C-3'); 135.0 (C-b', C-b); 130.4 (2xd, ${}^{4}J_{CP} = 4.5$ Hz, ${}^{4}J_{CP} = 5.0$ Hz, C-c, C-c'); 127.6 (C-2'); 122.9 (C-d, C-d'); 112.1 (C-5); 90.8 (C-1'); 86.9 (d, ${}^{3}J_{C,P} = 9.4 \text{ Hz}, \text{ C-4'}); 70.2 (C-a, C-a');$ 68.0 (d, ${}^{3}J_{CP} = 6.2$ Hz, C-5'); 35.0 (C-g'); 33.1, 30.6, 30.4, 30.2, 23.7 (C-i' – C-n'); 26.0 (C-h'); 20.9 (CH₃); 14.4 (C-o²); 12.5 (C-7). ³¹P-NMR (243 MHz, MeOH-d4, decoupled): $\delta/\text{ppm} = -12.07$ $(d, {}^{2}J_{P,P} = 20.6 \text{ Hz}, 1P, P_{\beta});$ -12.92 $(d, {}^{2}J_{P,P} = 20.5 \text{ Hz}, 1P, P_{\alpha})$. HRMS: $(ESI^{+}, m/z)$ calcd. for $C_{36}H_{46}N_2O_{14}P_2Na: 815.2322 [M+Na]^+$, found: 815.2312. RP18-HPLC: Method A, R_t= 18.5 min.

Ammonium-CH₃/C₁₁H₂₃-Di*PP***ro-d4TDP 7c:** General procedure E; Phosphoramidite 5c (108 mg, 0.18 mmol), d4T monophosphate (77 mg, 0.12 mmol) and 4,5-dicyanoimidazole activator solution (720 μ L, 0.18 mmol) in 4 mL acetonitrile. Oxidation by addition of *tert*butylhydroperoxide (5.5 molar solution in *n*-decane, 33 μ L, 0.18 mmol). The product (24 mg, 0.028 mmol, 23%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz,

 MeOH-d4): δ [ppm] = 7.66 (d, ${}^{4}J_{H,H}$ = 1.0 Hz, 1H, H-6); 7.41-7.37 (m, 4H, H-c, H-c'); 7.08-7.04 (m, 4H, H-d, H-d'); 6.94 (ddd, ${}^{3}J_{H,H}$ = 3.5 Hz, ${}^{3/4}J_{H,H}$ = 1.7 Hz, ${}^{3/4}J_{H,H}$ = 1.6 Hz, 1H, H-1'); 6.38 (ddd, ${}^{3}J_{H,H}$ = 6.0 Hz, ${}^{3}J_{H,H}$ = 3.2 Hz, ${}^{4}J_{H,H}$ = 3.2 Hz, 1H, H-3'); 5.83 (ddd, ${}^{3}J_{H,H}$ = 5.9 Hz, ${}^{3}J_{H,H}$ = 3.4 Hz, ${}^{4}J_{H,H}$ = 3.4 Hz, 1H, H-2'); 5.12-5.09 (m, 4H, H-a, H-a'); 4.94 (bs, 1H, H-4'); 4.23-4.15 (m, 2H, H-5'); 2.58 (t, ${}^{3}J_{H,H}$ = 7.4 Hz, 2H, H-g'); 2.27 (s, 3H, CH₃); 1.89 (d, ${}^{4}J_{H,H}$ = 1.0 Hz 3H, H-7); 1.73 (m, 2H, H-h'); 1.45-1.29 (m, 12H, H-i' – H-p'); 0.90 (t, ${}^{3}J_{H,H}$ = 7.1 Hz, 3H, H-q'). ¹³C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.8 (C-f'); 171.1 (C-f); 166.5 (C-4); 152.8 (C-2); 152.3 (C-e, C-e'); 138.7 (C-6); 135.3 (C-3'); 134.9 (C-b', C-b); 130.4 (2xd, ${}^{4}J_{C,P}$ = 4.6 Hz, ${}^{4}J_{C,P}$ = 5.2 Hz, C-c, C-c'); 127.6 (C-2'); 122.9 (C-d, C-d'); 112.1 (C-5); 90.8 (C-1'); 86.9 (d, ${}^{3}J_{C,P}$ = 9.1 Hz, C-4'); 70.2 (C-a, C-a'); 68.1 (d, ${}^{3}J_{C,P}$ = 6.6 Hz, C-5'); 35.0 (C-g'); 33.1, 30.7, 30.6, 30.5, 30.4, 30.2, 23.7 (C-i' – C-p'); 26.0 (C-h'); 20.9 (CH₃); 14.4 (C-q'); 12.5 (C-7). ³¹P-NMR (243 MHz, D₂O, decoupled): δ/ppm = -12.08 (d, ${}^{2}J_{P,P}$ = 20.8 Hz, P_β); -12.92 (d, ${}^{2}J_{P,P}$ = 20.2 Hz, P_α). HRMS: (ESΓ, m/z) calcd. for C₃₈H₄₉N₂O₁₄P₂: 819.2665 [M-H⁺]⁻, found: 819.2604. RP18-HPLC: Method A, R_i = 19.5 min.

Ammonium-CH₃/CF₃-Ph-Di*PP***ro-d4TDP 7d**: General procedure E; Phosphoramidite 5d (140 mg, 0.24 mmol), d4T monophosphate (100 mg, 0.16 mmol) and 4,5-dicyanoimidazole activator solution (960 μ L, 0.24 mmol) in 4 mL acetonitrile. Oxidation by addition of *tert*butylhydroperoxide (5.5 molar solution in *n*-decane, 44 μ L, 0.24 mmol). The product (40 mg, 0.048 mmol, 30%) was obtained as a colorless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 8.38-8.37 (m, 2H, H-h'/H-i'); 7.90-7.89 (m, 2H, H-h'/H-i'); 7.66 (d, ⁴J_{H,H} = 1.2 Hz, 1H, H-6); 7.48-7.39 (m, 4H, H-c', H-c); 7.25-7.22 (m, 2H, H-d'); 7.09-7.07 (m, 2H, H-d); 6.96 (ddd, ³J_{H,H} = 3.5 Hz, ³J_{H,H} = 1.8 Hz, ⁴J_{H,H} = 1.8 Hz, 1H, H-1'); 6.40-6.39 (m 1H,

H-3'); 5.84 (ddd, ${}^{3}J_{H,H} = 6.0$ Hz, ${}^{3}J_{H,H} = 3.4$ Hz, ${}^{4}J_{H,H} = 3.4$ Hz, 1H, H-2'); 5.16-5.11 (m, 4H, H-a, H-a'); 4.96 (bs, 1H, H-4'); 4.24-4.17 (m, 1H, H-5'); 2.25 (s, 3H, CH₃); 1.90 (s, 3H, H-7). ${}^{13}C$ -NMR: (150 MHz, MeOH-d4): δ [ppm] = 171.1 (C-f); 166.6 (C-4); 165.3 (C-f'); 152.9 (C-2); 152.4 (C-e/Ce'); 152.3 (C-e/C-e'); 138.7 (C-6); 135.9 (C-j'); 135.4 (C-b'); 135.3 (C-3'); 135.0 (C-b); 134.4 (C-g'); 131.8 (C-h'/C-i'); 130.5 (d, ${}^{4}J_{C,P} = 4.6$ Hz, C-c'); 130.4 (d, ${}^{4}J_{C,P} = 5.5$ Hz, C-c); 127.6 (C-2'); 126.9 (C-h'/C-i'); 124.3 (CF₃); 122.9 (C-d'); 122.9 (C-d); 112.0 (C-5); 90.8 (C-1'); 86.9 (d, ${}^{3}J_{C,P} = 9.5$ Hz, C-4'); 70.2 (C-a, C-a'); 68.1 (C-5'); 20.6 (CH₃); 12.5 (C-7). ${}^{19}F$ -NMR (565 MHz, MeOH-d4): δ/ppm = -60.75. ${}^{31}P$ -NMR (243 MHz, D₂O, decoupled): δ/ppm = -12.1 (d, ${}^{2}J_{P,P} = 20.4$ Hz, P_β); -12.9 (d, ${}^{2}J_{P,P} = 19.8$ Hz, P_α). HRMS: (ESΓ, m/z) calcd. for C₃₄H₃₀F₃N₂O₁₄P₂: 809.1130 [M-H⁺]⁺, found: 809.1096. Rp18-HPLC: Method A, R₄= 15.8 min.

TetrabutyImmonium-CH₃/CH₃-Ph-Di*PP*ro-d4TDP 7e: General procedure E; Phosphoramidite 5e (190 mg, 0.35 mmol), d4T monophosphate (150 mg, 0.24 mmol) and 4,5dicyanoimidazole activator solution (1.20 mL, 0.30 mmol) in 6 mL acetonitrile. Oxidation by addition of *tert*butylhydroperoxide (5.5 molar solution in *n*-decane, 64 µL, 0.35 mmol). The product (115 mg, 0.115 mmol, 48%) was obtained as a colourless wad. ¹H-NMR: (600 MHz, D₂O): δ [ppm] = 8.07-8.06 (m 2H, H-h'); 7.67 (d, ⁴J_{H,H} = 1.1 Hz, 1H, H-6); 7.47-7.39 (m 4H, Hc, H-c'); 7.38-7.37 (m, 2H, H-d/H-d'); 7.20-7.18 (m, 2H, H-d/H-d'); 7.09-7.07 (m, 2H, H-i'); 6.96 (ddd, ³J_{H,H} = 3.6 Hz, ³J_{H,H} = 1.7 Hz, ⁴J_{H,H} = 1.7 Hz, 1H, H-1'); 6.40 (ddd, ³J_{H,H} = 6.0 Hz, ³J_{H,H} = 3.2 Hz, ⁴J_{H,H} = 3.2 Hz, 1H, H-3'); 5.84 (ddd, ³J_{H,H} = 6.0 Hz, ³J_{H,H} = 3.4 Hz, ⁴J_{H,H} = 3.4 Hz, 1H, H-2'); 5.16-5.11 (m, 4H, H-a, H-a'); 4.96 (bs, 1H, H-4'); 4.25-4.17 (m, 2H, H-5'); 3.24-3.21 (m, 8H, H-A); 2.45 (s, 3H, H-k'); 2.25 (s, 3H, CH₃); 1.90 (d, ⁴J_{H,H} = 1.1 Hz, 3H, H-7); 1.68-1.62 (m, 8H, H-B); 1.41 (dq, ³J_{H,H} = 7.4 Hz, ³J_{H,H} = 7.4 Hz, 8H, H-C); 1.02 (t, ³J_{H,H} = 7.4 Hz, 12H, H-

ACS Paragon Plus Environment

 D). ¹³C-NMR: (150 MHz, D₂O): δ [ppm] = 171.0 (C-f); 166.6 (C-4); 166.5 (C-f'); 152.8 (C-2); 152.6 (C-e/C-e'); 152.4 (C-e/C-e'); 146.2 (C-g'); 138.7 (C-6); 135.3 (C-3'); 135.1 (dd, ³*J*_{C,P} = 7.5 Hz, ⁵*J*_{C,P} = 3.3 Hz, C-b/C-b'); 135.1 (dd, ³*J*_{C,P} = 7.2 Hz, ⁵*J*_{C,P} = 2.8 Hz, C-b/C-b'); 131.2 (Ch'); 130.5 (C-i'); 130.5 (d, ⁴*J*_{C,P} = 5.5 Hz, C-c'); 130.4 (d, ⁴*J*_{C,P} = 4.7 Hz, C-c); 128.0 (C-j'); 127.6 (C-2'); 123.0 (C-d); 122.9 (C-d'); 112.1 (C-5); 90.8 (C-1'); 87.0 (d, ³*J*_{C,P} = 9.4 Hz, C-4'); 70.2 (d, ²*J*_{C,P} = 5.6 Hz, C-a); 70.2 (d, ²*J*_{C,P} = 5.0 Hz, C-a'); 68.1 (d, ²*J*_{C,P} = 6.2 Hz, C-5'); 59.5, 59.5, 59.5 (C-A); 24.8 (C-B); 21.7 (C-k'); 20.9 (CH₃); 20.7 (C-C); 13.9 (C-D); 12.5 (C-7). ³¹P-NMR (243 MHz, D₂O, decoupled): δ /ppm = -12.1 (d, ²*J*_{P,P} = 20.6 Hz, P_β); -12.9 (d, ²*J*_{P,P} = 20.8 Hz, P_α). HRMS: (ESΓ, m/z) calcd. for C₃₄H₃₃N₂O₁₄P₂: 755.1413 [M-H⁺]⁻, found.: 755.1413. RP18-HPLC: Method A, R_i= 15.4 min.

Ammonium-C₄H₉/C₇H₁₅-Di*PP*ro-d4TDP 8a: General procedure E; Phosphoramidite 5f (112 mg, 0.19 mmol), d4T monophosphate (79 mg, 0.13 mmol) and 4,5-dicyanoimidazole activator solution (760 μL, 0.19 mmol) in 3 mL acetonitrile. Oxidation by addition of *tert*butylhydroperoxide (5.5 molar solution in *n*-decane, 35 μL, 0.19 mmol). The product (76 mg, 0.092 mmol, 71%) was obtained as a colourless wad after ion exchange. 71%. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 7.76 (s, 1H, H-6); 7.40-7.37 (m, 4H, H-c, H-c²); 7.06-7.04 (m, 4H, H-d, H-d²); 6.95 (bs, 1H, H-1²); 6.39-6.38 (m, 1H, H-3²); 5.84-5.83 (m, 1H, H-2²); 5.11-5.09 (m, 4H, H-a, H-a²); 4.95 (bs, 1H, H-4²); 4.22-4.15 (m, 2H, H-5²); 2.60-2.56 (m, 4H, H-g, H-g²); 1.89 (s, 3H, H-7); 1.71 (dt, ³*J*_{H,H} = 7.7 Hz, ³*J*_{H,H} = 7.5 Hz, 4H, H-h, H-h²); 1.48-1.30 (m, 10H, H-i, H-i² – H-1²); 0.99 (t, ³*J*_{H,H} = 7.4 Hz, 3H, H-j); 0.92 (t, ³*J*_{H,H} = 6.7 Hz, 3H, H-m²). ¹³C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.8 (C-f, C-f²); 166.7 (C-4); 152.4 (C-2); 152.4 (C-e, C-e²); 138.7 (C-6); 135.3 (C-3²); 134.9 (d, ³*J*_{C,P} = 2.4 Hz, C-b²/C-b); 134.9 (d, ³*J*_{C,P} = 3.2 Hz, C-b²/C-b); 130.4

(d, ${}^{4}J_{C,P} = 4.5$ Hz, C-c/C-c'); 127.6 (C-2'); 122.9 (C-d, C-d'); 112.1 (C-5); 90.8 (C-1'); 86.9 (d, ${}^{3}J_{C,P} = 9.0$ Hz, C-4'); 70.2 (2xd, ${}^{2}J_{C,P} = 5.0$ Hz, ${}^{2}J_{C,P} = 4.2$ Hz, C-a, C-a'); 68.1 (d, ${}^{3}J_{C,P} = 6.1$ Hz, C-5'); 35.0 (C-g'); 34.8 (C-g); 32.9, 30.1, 30.1, 23.7, (C-i' – C-l'); 28.1 (C-h); 26.0 (C-h'); 23.5 (C-i); 14.4 (C-m'); 14.1 (C-j); 12.5 (C-7). 31 P-NMR (243 MHz, MeOH-d4, decoupled): δ /ppm = -12.07 (d, ${}^{2}J_{P,P} = 20.5$ Hz, 1P, P_{β}); -12.91 (d, ${}^{2}J_{P,P} = 20.5$ Hz, 1P, P_{α}). HRMS: (ESI⁺, m/z) calcd. for C₃₇H₄₈N₂O₁₄P₂Na: 829.2478 [M+Na]⁺, found: 829.2411. RP18-HPLC: Method A, R_t= 18.8 min.

Ammonium-C₄H₉/C₉H₁₉-DiPPro-d4TDP 8b: General procedure E; Phosphoramidite 5g (164 mg, 0.27 mmol), d4T monophosphate (111 mg, 0.18 mmol) and 4,5-dicyanoimidazole activator solution (1.07 mL, 0.27 mmol) in 4 mL acetonitrile. Oxidation by addition of tertbutylhydroperoxide (5.5 molar solution in n-decane, 49 µL, 0.27 mmol). The product (115 mg, 0.14 mmol, 75%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 7.65 (s, 1H, H-6); 7.40-7.37 (m, 4H, H-c, H-c'); 7.06-7.04 (m, 4H, H-d, H-d'); 6.95-9.95 (m, 1H, H-1'); 6.38-6.37 (m, 1H, H-3'); 5.84-5.83 (m, 1H, H-2'); 5.11-5.09 (m, 4H, H-a, H-a'); 4.95 (bs, 1H, H-4'); 4.22-4.14 (m, 2H, H-5'); 2.58, 2.58 (2xt, ${}^{3}J_{HH} = 7.4$ Hz, ${}^{3}J_{H,H} = 7.3 \text{ Hz } 4\text{H}, \text{H-g}, \text{H-g}'); 1.89 \text{ (s, 3H, H-7)}; 1.72 \text{ (dt, } {}^{3}J_{H,H} = 7.6 \text{ Hz}, {}^{3}J_{H,H} = 7.4 \text{ Hz}, 4\text{H}, \text{H-h},$ H-h'); 1.49-1.31 (m, 14H, H-i, H-i' – H-n'); 0.99 (t, ${}^{3}J_{HH} = 7.4$ Hz, 3H, H-j); 0.90 (t, ${}^{3}J_{HH} =$ 6.7 Hz, 3H, H-o'). ¹³C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.8 (C-f/C-f'); 173.8 (C-f/Cf'); 166.5 (C-4); 152.8 (C-2); 152.4 (C-e/C-e'); 152.4 (C-e/C-e'); 138.7 (C-6); 135.3 (C-3'); 134.9 (d, ${}^{3}J_{CP} = 7.7$ Hz, C-b'/C-b); 134.9 (d, ${}^{3}J_{CP} = 6.8$ Hz, C-b'/C-b); 130.4 (d, ${}^{4}J_{CP} = 3.8$ Hz, C-c/C-c'); 127.6 (C-2'); 122.9 (C-d, C-d'); 112.1 (C-5); 90.8 (C-1'); 86.9 (d, ${}^{3}J_{CP} = 9.3$ Hz, C-4'); 70.3 (d, ${}^{2}J_{CP} = 5.2$ Hz, C-a/C-a'); 70.2 (d, ${}^{2}J_{CP} = 5.3$ Hz, C-a/C-a'); 68.1 (d, ${}^{3}J_{CP} = 6.3$ Hz,

Journal of Medicinal Chemistry

C-5'); 35.0 (C-g'); 34.8 (C-g); 33.0, 30.6, 30.4, 30.2, 23.7, (C-i' – C-n'); 28.1 (C-h); 26.0 (C-h'); 23.2 (C-i); 14.4 (C-o'); 14.1 (C-j); 12.5 (C-7). ³¹P-NMR (243 MHz, MeOH-d4, decoupled): $\delta/\text{ppm} = -12.06$ (d, ${}^{2}J_{P,P} = 20.5$ Hz, 1P, P_{β}); -12.88 (d, ${}^{2}J_{P,P} = 20.5$ Hz, 1P, P_{α}). HRMS: (ESI⁻, m/z) calcd. for C₃₉H₅₁N₂O₁₄P₂: 833.2821 [M-H⁺]⁻, found: 833.2638. RP18-HPLC: Method A, R_t= 20.2 min.

Ammonium-C₄H₉/C₁₁H₂₃-DiPPro-d4TDP 8c: General procedure E; Phosphoramidite 5h (42 mg, 0.058 mmol), d4T monophosphate (33 mg, 0.053 mmol) and 4.5-dicyanoimidazole activator solution (230 µL, 0.053 mmol) in 2 mL acetonitrile. Oxidation by addition of tertbutylhydroperoxide (5.5 molar solution in *n*-decane, 11 µL, 0.058 mmol). The product (33 mg, 0.038 mmol, 71%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, D₂O): δ [ppm] = 7.66 (d, ⁴J_{H,H} = 1.2 Hz, 1H, H-6); 7.40-7.37 (m, 4H, H-c, H-c'); 7.06-7.04 (m, 4H, H-d, H-d'); 6.95 (ddd, ${}^{3}J_{HH} = 3.5$ Hz, ${}^{3}J_{HH} = 1.7$ Hz, ${}^{4}J_{HH} = 1.7$ Hz, 1H, H-1'); 6.38 $(ddd, {}^{3}J_{H,H} = 6.0 \text{ Hz}, {}^{3}J_{H,H} = 3.4 \text{ Hz}, {}^{4}J_{H,H} = 3.4 \text{ Hz}, 1\text{ H}, \text{ H-3'}); 5.83 (ddd, {}^{3}J_{H,H} = 5.6 \text{ Hz}, 10.3 \text{ Hz}); 5.83 (ddd, {}^{3}J_{H,H} = 5.6 \text{ Hz}); 5.$ ${}^{3}J_{\text{H.H}} = 1.8 \text{ Hz}, {}^{4}J_{\text{H.H}} = 1.8 \text{ Hz}, 1\text{H}, \text{H-2'}); 5.11-5.08 (m, 4\text{H}, \text{H-a}, \text{H-a'}); 4.95 (bs, 1\text{H}, \text{H-4'}); 4.23-$ 4.15 (m, 2H, H-5'); 2.60-2.56 (m, 4H, H-g, H-g'); 1.89 (d, ${}^{4}J_{HH} = 1.1$ Hz, 3H, H-7); 1.76-1.69 (m, 4H, H-h, H-h'); 1.49-1.1.27 (m, 18H, H-i, H-i'-H-p'); 0.99 (t, ${}^{3}J_{H,H} = 7.4$ Hz, 3H, H-j); 0.99 (t, ${}^{3}J_{HH} = 6.7$ Hz, 3H, H-q'). 13 C-NMR: (150 MHz, D₂O): δ [ppm] = 173.6 (C-f, C-f'); 166.5 (C-f) 4); 152.8 (C-2); 152.3 (C-e, C-e'); 138.7 (C-6); 133.9 (C-3'); 134.8 (C-b', C-b); 130.4 (C-c, Cc'); 127.3 (C-2'); 122.9 (C-d, C-d'); 112.1 (C-5); 90.8 (C-1'); 86.9 (C-4'); 70.2 (C-a, C-a'); 68.1 (C-5'); 35.0 (C-g'); 34.8 (C-g); 32.9, 30.7, 30.6, 30.4, 30.2, 23.5, (C-i' - C-p'); 28.0 (C-h); 25.8 (C-h'); 22.9 (C-i); 14.2 (C-q'); 13.8 (C-i); 12.2 (C-7). ³¹P-NMR (243 MHz, D₂O, decoupled):

 δ /ppm = -12.1 (d, ²*J*_{P,P} = 20.6 Hz, P_β); -12.9 (d, ²*J*_{P,P} = 21.5 Hz, P_α). HRMS: (ESI⁺, m/z) calcd. for C₄₁H₅₃N₂O₁₄P₂Na: 885.3104 [M+Na]⁺, found: 885.3022. RP18-HPLC: Method A, R_t= 21.4 min.

Ammonium-C₄H₉/CF₃-Ph-DiPPro-d4TDP 8d: General procedure E; Phosphoramidite 5i (180 mg, 0.29 mmol), d4T monophosphate (121 mg 0.19 mmol) and 4,5-dicyanoimidazole activator solution (950 µL, 0.24 mmol) in 5 mL acetonitrile. Oxidation by addition of tertbutylhydroperoxide (5.5 molar solution in *n*-decane, 53 µL, 0.29 mmol). The product (89 mg, 0.10 mmol. 53%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz. MeOH-d4): δ [ppm] = 8.38-8.37 (m, 2H, H-h'/H-i'); 7.90-7.89 (m, 2H, H-h'/H-i'); 7.67 (d, ${}^{4}J$ = 1.0 Hz, 1H, H-6); 7.47-7.44 (m, 2H, H-c'); 7.42-7.38 (m, 2H, H-c); 7.24-7.22 (m, 2H, H,d'); 7.07-7.05 (m, 2H, H-d); 6.96 (ddd, ${}^{3}J_{H,H} = 3.5 \text{ Hz}$, ${}^{3}J_{H,H} = 1.7 \text{ Hz}$, ${}^{4}J_{H,H} = 1.7 \text{ Hz}$, 1H, H-1'); 6.41-6.39 (m 1H, H-3'); 5.85-5.84 (m, 1H, H-2'); 5.16-5.11 (m, 4H, H-a, H-a'); 4.96 (bs, 1H, H-4'); 4.25-4.17 (m, 1H, H-5'); 2.56 (t, ${}^{3}J_{HH} = 7.4$ Hz, 2H, H-g); 1.90 (s, 3H, H-7); 1.70-1.66 (m, 2H, H-h); 1.46-1.39 (m, 2H, H-i); 0.96 (t, ${}^{3}J_{H,H} = 7.4$ Hz, 3H, H-j). 13 C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.9 (C-f); 166.9 (C-4); 165.3 (C-f'); 152.9 (C-2); 152.6 (C-e); 152.4 (C-e'); 138.9 (C-6); 136.1 (C-j'); 135.6 (C-b'); 135.1 (C-b); 135.0 (C-3'); 134.5 (C-g'); 131.8 (C-h'/C-i'); 130.5 (d, ${}^{4}J_{CP}$ = 5.5 Hz, C-c'); 130.4 (d, ${}^{4}J_{CP}$ = 5.6 Hz, C-c); 127.6 (C-2'); 126.9 (C-h'/C-i'); 124.3 (CF₃); 122.9 (C-d'); 122.9 (C-d); 112.4 (C-5); 90.6 (C-1'); 87.0 (d, ${}^{3}J_{CP} = 9.3$ Hz, C-4'); 70.2 (C-a, C-a'); 68.1 (C-5'); 34.7 (C-g); 28.0 (C-h); 23.1 (C-i); 14.0 (C-j); 12.5 (C-7). ¹⁹F-NMR (565 MHz, MeOH-d4): $\delta/\text{ppm} = -60.75$. ³¹P-NMR (243 MHz, MeOH-d4, decoupled): $\delta/\text{ppm} = -60.75$. 12.1 (bs, P_{β}); -12.9 (d, ${}^{2}J_{P,P}$ = 20.8 Hz, P_{α}). 19 F-NMR (565 MHz, MeOH-d4): δ /ppm = -60.75. HRMS: (ESI', m/z) calcd. for $C_{37}H_{36}F_{3}N_{2}O_{14}P_{2}$: 851.1559 [M-H⁺]⁻, found: 851.1597. RP18-HPLC: Method A, $R_t = 17.4$ min.

Ammonium-C₄H₉/CH₃-Ph-DiPPro-d4TDP 8e: General procedure E; Phosphoramidite 5j (123 mg, 0.21 mmol), d4T monophosphate (89 mg, 0.14 mmol) and 4,5-dicyanoimidazole activator solution (860 µL, 0.21 mmol) in 4 mL acetonitrile. Oxidation by addition of tertbutylhydroperoxide (5.5 molar solution in n-decane, 39 µL, 0.21 mmol). The product (36 mg, 0.044 mmol, 31%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, D₂O): δ [ppm] = 8.07-8.06 (m 2H, H-h'); 7.67 (d, ${}^{4}J_{HH}$ = 1.2 Hz, 1H, H-6); 7.45-7.38 (m 6H, Hc, H-c', H-d/H-d'); 7.20-7.18 (m, 2H, H-d/H-d'); 7.07-7.04 (m, 2H, H-i'); 6.96 (ddd, ${}^{3}J_{\rm H,H} = 3.5 \text{ Hz}, {}^{3}J_{\rm H,H} = 1.7 \text{ Hz}, {}^{4}J_{\rm H,H} = 1.7 \text{ Hz}, 1\text{ H}, \text{H-1}^{2}$; 6.40 (ddd, ${}^{3}J_{\rm H,H} = 6.0 \text{ Hz}, {}^{3}J_{\rm H,H} = 3.3 \text{ Hz},$ ${}^{4}J_{H,H} = 3.3 \text{ Hz}, 1\text{H}, \text{H}-3'$; 5.84 (ddd, ${}^{3}J_{H,H} = 5.9 \text{ Hz}, {}^{3}J_{H,H} = 3.5 \text{ Hz}, {}^{4}J_{H,H} = 3.5 \text{ Hz}, 1\text{H}, \text{H}-2'$); 5.15-5.10 (m, 4H, H-a, H-a'); 4.96 (bs, 1H, H-4'); 4.25-4.16 (m, 2H, H-5'); 2.56 (t, 2H, H-g); 2.46 (s, 3H, H-k²); 1.90 (d, ${}^{4}J_{HH} = 1.1$ Hz, 3H, H-7); 1.72-1.65 (m, 2H, H-h); 1.41 (m, 2H, H-i); 0.96 (t, ${}^{3}J_{HH} = 7.4$ Hz, 3H, H-j). 13 C-NMR: (150 MHz, D₂O): δ [ppm] = 173.6 (C-f); 166.6 (C-4); 166.5 (C-f'); 152.7 (C-2); 152.5 (C-e/C-e'); 152.3 (C-e/C-e'); 146.2 (C-g'); 138.6 (C-6); 135.2 (C-3'); 135.1 (C-b/C-b'); 134.9 (C-b/C-b'); 131.2 (C-h'); 131.2 (C-c, C-c'); 130.5 (C-i'); 130.4 (C-j'); 127.6 (C-2'); 123.0 (C-d); 122.9 (C-d'); 112.1 (C-5); 90.8 (C-1'); 87.0 (d, ${}^{3}J_{CP} = 9.4$ Hz, C-4'); 70.3 (C-a, C-a'); 68.1 (C-5'); 34.7 (C-g); 27.9 (C-h); 23.2 (C-i); 21.6 (C-k'); 13.8 (C-j); 12.4 (C-7). ³¹P-NMR (243 MHz, D₂O, decoupled): $\delta/\text{ppm} = -12.1$ (d, ² $J_{\text{P}P} = 20.2$ Hz, P₆); -12.9 (d, ${}^{2}J_{P,P} = 21.0 \text{ Hz}$, P_{α}). HRMS: (ESI⁻, m/z) calcd. for $C_{37}H_{39}N_2O_{14}P_2$: 797.1882 [M-H⁺]⁻, found: 797.1855. RP18-HPLC: Method A, R_t= 17.0 min.

Ammonium-CH₃/C₇H₁₅-Di*PP*ro-AZTDP 9a: General procedure E; Phosphoramidite 5a (130 mg, 0.24 mmol), AZT monophosphate (82 mg, 0.16 mmol) and 4,5-dicyanoimidazole

activator solution (960 µL, 0.24 mmol) in 3 mL acetonitrile. Oxidation by addition of tertbutylhydroperoxide (5.5 molar solution in n-decane, 44 µL, 0.24 mmol). The product (79 mg, 0.096 mmol, 60%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 7.721 (s, 1H, H-6); 7.42-7.40 (m, 4H, H-c, H-c'); 7.08-7.05 (m, 4H, H-d, H-d'); 6.21 (dd, ${}^{3}J_{H,H} = 6.8$ Hz, ${}^{3}J_{H,H} = 6.8$ Hz, 1H, H-1'); 5.14-5.11 (m, 4H, H-a, H-a'); 4.41-4.39 (m, 1H, H-3'); 4.21-4.18 (m, 1H, H-5'a); 4.13-4.10 (m, 1H, H-5'b); 4.04-4.03 (m, 1H, H-4'); 2.58 (t, ${}^{3}J_{H,H}$ = 7.4 Hz, 2H, H-g'); 2.36-2.31 (m, 2H, H-2'a); 2.30-2.26 (m, 2H, H-2'b); 2.27 (s, 3H, CH₃); 1.90 (s, 3H, H-7); 1.73 (dt, ${}^{3}J_{H,H} = 7.4$ Hz, ${}^{3}J_{H,H} = 7.4$ Hz, 2H, H-h'); 1.44-1.30 (m, 8H, H-i' – H-l'); 0.92 (t, ${}^{3}J_{HH} = 6.8$ Hz, 3H, H-m'). 13 C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.8 (C-f'); 171.0 (C-f); 166.4 (C-4); 152.5 (C-e/Ce'); 152.4 (C-e/Ce'); 152.3 (C-2); 137.7 (C-6); 134.9 (C-b', C-b); 130.5 (d, ${}^{4}J_{CP} = 3.2$ Hz, C-c/C-c'); 130.4 (d, ${}^{4}J_{CP} = 3.3$ Hz, C-c/C-c'); 122.9 (C-d, C-d'); 112.2 (C-5); 85.8 (C-1'); 84.4 (d, ${}^{3}J_{CP} = 9.4$ Hz, C-4'); 70.3 (d, ${}^{2}J_{CP} = 5.2$ Hz, C-a/C-a'); 67.2 (d, ${}^{3}J_{CP} = 5.5$ Hz, C-5'); 62.5 (C-3'); 38.0 (C-2'); 35.0 (C-g'); 32.9, 30.1, 30.1, 23.7 (C-i' - C-l'); 26.0 (C-h'); 20.9 (CH₃); 14.4 (C-m'); 12.6 (C-7). ³¹P-NMR (243 MHz, MeOHd4, decoupled): $\delta/\text{ppm} = -12.14$ (d, ${}^{2}J_{PP} = 20.1$ Hz, 1P, P_B); -12.76 (d, ${}^{2}J_{PP} = 20.3$ Hz, 1P, P_g). HRMS: (ESI⁺, m/z) calcd. for $C_{34}H_{44}N_5O_{14}P_2$: 808.2355 [M+H⁺]⁺, found: 808.2354. RP18-HPLC: Method B, $R_t = 17.4$ min.

Ammonium-CH₃/C₉H₁₉-Di*PP***ro-AZTDP 9b**: General procedure E; Phosphoramidite 5b (100 mg, 0.18 mmol), AZT monophosphate (73 mg, 0.12 mmol) and 4,5-dicyanoimidazole activator solution (700 μ L, 0.18 mmol) in 3 mL acetonitrile. Oxidation by addition of *tert*butylhydroperoxide (5.5 molar solution in *n*-decane, 32 μ L, 0.18 mmol). The product (45 mg, 0.053 mmol, 45%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz,

MeOH-d4): δ [ppm] = 7.72 (s, 1H, H-6); 7.42-7.40 (m, 4H, H-c, H-c'); 7.08-7.05 (m, 4H, H-d, Hd'); 6.21 (dd, ${}^{3}J_{HH} = 6.8$ Hz, ${}^{3}J_{HH} = 6.8$ Hz, 1H, H-1'); 5.14-5.11 (m, 4H, H-a, H-a'); 4.40 (dt, ${}^{3}J_{\text{H,H}} = 6.6 \text{ Hz}, {}^{3}J_{\text{H,H}} = 6.8 \text{ Hz}, 1\text{H}, \text{H-3'}); 4.21-4.18 \text{ (m, 1H, H-5'a)}; 4.13-4.10 \text{ (m, 1H, H-5'b)};$ 4.04-4.03 (m, 1H, H-4'); 2.57 (t, ${}^{3}J_{HH} = 7.4$ Hz, 2H, H-g'); 2.36-2.31 (m, 2H, H-2'a); 2.30-2.26 (m, 2H, H-2'b); 2.27 (s, 3H, CH₃); 1.90 (s, 3H, H-7); 1.73 (dt, ${}^{3}J_{H,H} = 7.4$ Hz, ${}^{3}J_{H,H} = 7.4$ Hz, 2H, H-h'); 1.45-1.29 (m, 12H, H-i' – H-n'); 0.99 (t, ${}^{3}J_{H,H} = 7.4$ Hz, 3H, H-j); 0.90 (t, ${}^{3}J_{H,H} = 6.8$ Hz, 3H, H-o'). ¹³C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.7 (C-f'); 170.1 (C-f); 166.3 (C-4); 152.5 (C-e, Ce'); 152.4 (C-2); 137.8 (C-6); 134.9 (C-b', C-b); 130.5 (C-c, C-c'); 122.9 (C-d, Cd'); 112.1 (C-5); 85.8 (C-1'); 84.4 (d, ${}^{3}J_{CP} = 9.6$ Hz, C-4'); 70.3 (d, ${}^{2}J_{CP} = 6.4$ Hz, C-a/C-a'); 67.3 (d, ${}^{3}J_{C,P} = 5.8$ Hz, C-5'); 62.5 (C-3'); 38.0 (C-2'); 35.0 (C-g'); 33.0, 30.6, 30.4, 30.2, 23.8 (C-i' – C-n'); 26.0 (C-h'); 14.5 (C-o'); 12.6 (C-7). ³¹P-NMR (243 MHz, MeOH-d4, decoupled): $\delta/\text{ppm} = -12.16 \text{ (d, }^{2}J_{P,P} = 20.2 \text{ Hz}, 1P, P_{\beta}\text{)}; -12.79 \text{ (d, }^{2}J_{P,P} = 19.8 \text{ Hz}, 1P, P_{\alpha}\text{)}. \text{ HRMS: (ESI⁺, m/z)}$ calcd. for $C_{36}H_{48}N_5O_{14}P_2$: 863.2668 $[M+H^+]^+$, found: 836.2854. RP18-HPLC: Method B, $R_{t} = 18.6 \text{ min.}$

Ammonium-CH₃/C₁₁H₂₃-Di*PP***ro-AZTDP 9c:** General procedure E; Phosphoramidite 5c (125 mg, 0.21 mmol), AZT monophosphate (86 mg, 0.14 mmol) and 4,5-dicyanoimidazole activator solution (830 µL, 0.21 mmol) in 4 mL acetonitrile. Oxidation by addition of *tert*butylhydroperoxide (5.5 molar solution in *n*-decane, 38 µL, 0.21 mmol). The product (58 mg, 0.066 mmol, 48%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 7.72 (s, 1H, H-6); 7.42-7.40 (m, 4H, H-c, H-c'); 7.08-7.05 (m, 4H, H-d, H-d'); 6.21 (dd, ³J_{H,H} = 6.8 Hz, ³J_{H,H} = 6.8 Hz, 1H, H-1'); 5.14-5.12 (m, 4H, H-a, H-a'); 4.4 (dt, ³J_{H,H} = 6.5 Hz, ³J_{H,H} = 6.7 Hz, 1H, H-3'); 4.21-4.18 (m, 1H, H-5'a); 4.13-4.10 (m, 1H, H-5'b);

4.04-4.03 (m, 1H, H-4'); 2.58 (d, ${}^{3}J_{H,H} = 7.4$ Hz, 1H, H-g'); 2.36-2.26 (m, 2H, H-2'a, H-2'b); 2.27 (s, 3H, CH₃); 1.90 (s, 3H, H-7); 1.76-1.71 (m, 2H, H-h'); 1.45-1.25 (m, 16H, H-i' – H-p'); 0.90 (t, ${}^{3}J_{H,H} = 6.9$ Hz, 3H, H-q'). 13 C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.8 (C-f'); 171.0 (C-f); 166.3 (C-4); 152.5 (C-e, Ce'); 152.3 (C-2); 137.8 (C-6); 134.8 (C-b', C-b); 130.5 (2xd, ${}^{4}J_{C,P} = 2.3$ Hz, C-c, C-c'); 122.9 (C-d, C-d'); 112.2 (C-5); 85.8 (C-1'); 84.4 (d, ${}^{3}J_{C,P} =$ 8.8 Hz, C-4'); 70.3 (d, ${}^{2}J_{C,P} = 5.3$ Hz, C-a/C-a'); 67.3 (d, ${}^{3}J_{C,P} = 5.8$ Hz, C-5'); 62.5 (C-3'); 38.0 (C-2'); 35.0 (C-g'); 33.1, 30.7, 30.6, 30.5, 30.4, 30.2, 23.7 (C-i' – C-p'); 26.0 (C-h'); 20.9 (CH₃); 12.6 (C-7). 31 P-NMR (243 MHz, MeOH-d4, decoupled): δ /ppm = -12.17 (d, ${}^{2}J_{P,P} = 20.5$ Hz, 1P, P_{β}); -12.8 (d, ${}^{2}J_{P,P} = 20.4$ Hz, 1P, P_{α}). HRMS: (ESI⁺, m/z) calcd. for C₃₈H₅₁N₅O₁₄P₂Na: 864.2981 [M+Na]⁺, found: 864.2982. RP18-HPLC: Method B, R_i= 20.1 min.

Ammonium-CH₃/Ph-CF₃-Di*PP*ro-AZTDP 9d: General procedure E; Phosphoramidite 5d (140 mg, 0.24 mmol), AZT monophosphate (98 mg, 0.16 mmol) and 4,5-dicyanoimidazole activator solution (940 μL, 0.24 mmol) in 4 mL acetonitrile. Oxidation by addition of *tert*butylhydroperoxide (5.5 molar solution in *n*-decane, 44 μL, 0.24 mmol). The product (83 mg, 0.095 mmol, 61%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 8.38-8.36 (m, 2H, H-h'/H-i'); 7.90-7.89 (m, 2H, H-h'/H-i'); 7.73 (s, 1H, H-6); 7.50-7.48 (m, 2H, H-c'); 7.45-7.42 (m, 2H, H-c); 7.25-7.23 (m, 2H, H-d'); 7.10-7.08 (m, 2H, H-d); 6.22 (dd, ${}^{3}J_{H,H}$ = 6.8 Hz, ${}^{3}J_{H,H}$ = 6.8 Hz, 1H, H-1'); 5.18-5.14 (m, 4H, H-a, H-a'); 4.43-4.40 (m, 1H, H-3'); 4.23-4.20 (m, 1H, H-5'a); 4.15-4.12 (m, 1H, H-5'b); 4.06-4.04 (m, 1H, H-4'); 2.36-2.26 (m, 2H, H-2'a, H-2'b); 2.25 (s, 3H, CH₃); 1.90 (s, 3H, H-7). ¹³C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 171.0 (C-f); 166.4 (C-4); 165.4(C-f'); 152.4 (C-2); 152.3 (C-e); 152.3 (C-e'); 137.8 (C-6); 135.9 (C-j'); 135.3 (C-b'); 135.1 (C-b); 134.4 (C-g'); 131.8 (C-h'/C-i'); 130.6

 (d, ${}^{5}J_{C,P} = 4.2 \text{ Hz}, \text{ C-c}{}^{2}$); 130.1 (d, ${}^{5}J_{C,P} = 2.5 \text{ Hz}, \text{ C-c}$); 126.5 (C-h²/C-i²); 124.4 (CF₃); 123.0 (C-d²); 122.9 (C-d); 112.2 (C-5); 85.8 (C-1²); 84.4 (d, ${}^{3}J_{C,P} = 9.5 \text{ Hz}, \text{ C-4}{}^{2}$); 70.4 (d, ${}^{2}J_{C,P} = 5.3 \text{ Hz}, \text{ C-a}/\text{C-a}{}^{2}$); 70.3 (d, ${}^{2}J_{C,P} = 5.5 \text{ Hz}, \text{ C-a}/\text{C-a}{}^{2}$); 67.3 (d, ${}^{3}J_{C,P} = 5.8 \text{ Hz}, \text{ C-5}{}^{2}$); 62.5 (C-3²); 38.0 (C-2²); 20.9 (CH₃); 12.6 (C-7). ¹⁷F-NMR (565 MHz, MeOH-d4): δ /ppm = -64.7. ³¹P-NMR (243 MHz, MeOH-d4, decoupled): δ /ppm = -12.14 (d, ${}^{2}J_{P,P} = 20.1 \text{ Hz}, 1P, P_{\beta}$); -12.68-12.83 (m, 1P, P_a). HRMS: (ESI⁺, m/z) calcd. for C₃₄H₃₃F₃N₅O₁₄P₂: 854.1451 [M+H⁺]⁺, found: 854.1442. RP18-HPLC: Method B, R_t= 16.4 min.

Ammonium-C₄H₉/C₇H₁₅-DiPPro-AZTDP 10a: General procedure E; Phosphoramidite 5f (132 mg, 0.23 mmol), AZT monophosphate (96 mg, 0.15 mmol) and 4,5-dicyanoimidazole activator solution (900 µL, 0.23 mmol) in 4 mL acetonitrile. Oxidation by addition of tertbutylhydroperoxide (5.5 molar solution in n-decane, 42 µL, 0.23 mmol). The product (37 mg, 0.043 mmol, 29%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz. MeOH-d4): δ [ppm] = 7.71 (s, 1H, H-6); 7.42-7.40 (m, 4H, H-c, H-c'); 7.07-7.05 (m, 4H, H-d, Hd'); 6.20 (dd, ${}^{3}J_{H,H} = 6.7$ Hz, ${}^{3}J_{H,H} = 6.7$ Hz, 1H, H-1'); 5.13-5.11 (m, 4H, H-a, H-a'); 4.40-4.38 (m, 1H, H-3'); 4.21-4.18 (m, 1H, H-5'a); 4.13-4.09 (m, 1H, H-5'b); 4.03 (bs, 1H, H-4'); 2.59-2.56 (m, 4H, H-g, H-g'); 2.36-2.31 (m, 2H, H-2'a); 2.30-2.26 (m, 2H, H-2'b); 1.90 (s, 3H, H-7); 1.72 (dt, ${}^{3}J_{H,H} = 7.6$ Hz, ${}^{3}J_{H,H} = 7.6$ Hz 4H, H-h, H-h'); 1.48-1.29 (m, 10H, H-i, H-i' – H-l'); 0.98 (t, ${}^{3}J_{H,H} = 7.4$ Hz, 3H, H-j); 0.92 (t, ${}^{3}J_{H,H} = 6.6$ Hz, 3H, H-m'). 13 C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.7 (C-f, C-f'); 166.4 (C-4); 152.5 (C-e, Ce'); 152.3 (C-2); 137.8 (C-6); 134.9 (Cb/C-b'); 134.8 (C-b/C-b'); 130.5 (C-c, C-c'); 122.9 (C-d, C-d'); 112.2 (C-5); 85.8 (C-1'); 84.4 (d, ${}^{3}J_{CP} = 9.4 \text{ Hz}, \text{ C-4'}$; 70.4 (d, ${}^{2}J_{CP} = 5.6 \text{ Hz}, \text{ C-a/C-a'}$); 67.2 (d, ${}^{3}J_{CP} = 6.1 \text{ Hz}, \text{ C-5'}$); 62.5 (C-3'); 38.0 (C-2'); 35.0 (C-g/C-g'); 34.8 (C-g/c-g'); 32.9, 30.1, 30.1, 23.7 (C-i' - C-l'); 28.1 (C-h, C-

h'); 26.0 (C-h, C-h'); 23.2 (C-i) 14.4 (C-m'); 14.1 (C-j); 12.6 (C-7). ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 7.71 (s, 1H, H-6); 7.42-7.40 (m, 4H, H-c, H-c'); 7.07-7.05 (m, 4H, H-d, H-d'); 6.20 (dd, ${}^{3}J_{\text{H,H}}$ = 6.7 Hz, ${}^{3}J_{\text{H,H}}$ = 6.7 Hz, 1H, H-1'); 5.13-5.11 (m, 4H, H-a, H-a'); 4.40-4.38 (m, 1H, H-3'); 4.21-4.18 (m, 1H, H-5'a); 4.13-4.09 (m, 1H, H-5'b); 4.03 (bs, 1H, H-4'); 2.59-2.56 (m, 4H, H-g, H-g'); 2.36-2.31 (m, 2H, H-2'a); 2.30-2.26 (m, 2H, H-2'b); 1.90 (s, 3H, H-7); 1.72 (dt, ${}^{3}J_{\text{H,H}}$ = 7.6 Hz, ${}^{3}J_{\text{H,H}}$ = 7.6 Hz 4H, H-h, H-h'); 1.48-1.29 (m, 10H, H-i, H-i' – H-1'); 0.98 (t, ${}^{3}J_{\text{H,H}}$ = 7.4 Hz, 3H, H-j); 0.92 (t, ${}^{3}J_{\text{H,H}}$ = 6.6 Hz, 3H, H-m'). HRMS: (ESI⁺, m/z) calcd. for C₃₇H₄₉N₅O₁₄P₂Na: 872.2643 [M+Na]⁺, found: 872.2568.

Ammonium-C₄H₉/C₉H₁₉-Di*PP***ro-AZTDP 10b: General procedure E; Phosphoramidite 5g (117 mg, 0.191 mmol), AZT monophosphate (79 mg, 0.13 mmol) and 4,5-dicyanoimidazole activator solution (760 μL, 0.19 mmol) in 3 mL acetonitrile. Oxidation by addition of** *tert***butylhydroperoxide (5.5 molar solution in** *n***-decane, 35 μL, 0.91 mmol). The product (97 mg, 0.11 mmol, 85%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 7.72 (s, 1H, H-6); 7.42-7.40 (m, 4H, H-c, H-c'); 7.06-7.05 (m, 4H, H-d, H-d'); 6.21 (dd, {}^{3}J_{H,H} = 6.8 Hz, {}^{3}J_{H,H} = 6.8 Hz, 1H, H-1'); 5.14-5.11 (m, 4H, H-a, H-a'); 4.40 (dt, {}^{3}J_{H,H} = 6.7 Hz, {}^{3}J_{H,H} = 6.8 Hz, 1H, H-3'); 4.21-4.18 (m, 1H, H-5'a); 4.13-4.10 (m, 1H, H-5'b); 4.04-4.03 (m, 1H, H-4'); 2.59-2.56 (m, 4H, H-g, H-g'); 2.36-2.31 (m, 2H, H-2'a); 2.30-2.26 (m, 2H, H-2'b); 1.90 (s, 3H, H-7); 1.72 (dt, {}^{3}J_{H,H} = 7.4 Hz, {}^{3}J_{H,H} = 7.6 Hz 4H, H-h, H-h'); 1.48-1.29 (m, 14H, H-i, H-i' – H-n'); 0.99 (t, {}^{3}J_{H,H} = 7.4 Hz, 3H, H-j); 0.91 (t, {}^{3}J_{H,H} = 6.8 Hz, 3H, H-o'). ¹³C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.7 (C-f, C-f'); 166.5 (C-4); 152.5 (C-e, Ce'); 152.4 (C-2); 137.8 (C-6); 134.8 (C-b', C-b); 130.5 (C-c, C-c') 122.9 (C-d, C-d'); 112.2 (C-5); 85.8 (C-1'); 84.4 (d, {}^{3}J_{C,P} = 8.8 Hz, C-4'); 70.4 (d, {}^{2}J_{C,P} = 5.5 Hz, C-a/C-a'); 67.3 (d, {}^{3}J_{C,P} =**

5.5 Hz, C-5'); 62.5 (C-3'); 38.0 (C-2'); 35.0 (C-g/C-g'); 34.8 (C-g/c-g'); 33.0, 30.6, 30.4, 30.2, 23.7 (C-i' – C-n'); 28.1 (C-h, C-h'); 26.0 (C-h, C-h'); 23.3 (C-i); 14.4 (C-o'); 14.1 (C-j); 12.6 (C-7). ³¹P-NMR (243 MHz, MeOH-d4, decoupled): δ /ppm = -12.16 (d, ²J_{P,P} = 20.6 Hz, 1P, P_β); -12.78 (d, ²J_{P,P} = 21.7 Hz, 1P, P_a). HRMS: (ESI⁺, m/z) calcd. for C₃₉H₅₃N₅O₁₄P₂Na: 900.2962 [M+Na]⁺, found: 900.2946. RP18-HPLC: Method B, R_t= 20.5 min.

Ammonium-C₄H₉/C₁₁H₂₃-DiPPro-AZTDP 10c: General procedure E; Phosphoramidite 5h (120 mg, 0.17 mmol), AZT monophosphate (69 mg, 0.11 mmol) and 4,5-dicyanoimidazole activator solution (550 µL, 0.14 mmol) in 3 mL acetonitrile. Oxidation by addition of tertbutylhydroperoxide (5.5 molar solution in *n*-decane, 31 µL, 0.17 mmol). The product (37 mg, 0.040 mmol, 37%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 7.75 (s, 1H, H-6); 7.21-7.12 (m, 4H, H-c, H-c'); 6.85-6.84 (m, 4H, H-d, Hd'); 6.00 (dd, ${}^{3}J_{HH} = 6.8$ Hz, ${}^{3}J_{HH} = 6.8$ Hz, 1H, H-1'); 4.92-4.90 (m, 4H, H-a, H-a'); 4.19 (dt, ${}^{3}J_{\rm H,H} = 6.6 \,\text{Hz}, \,{}^{3}J_{\rm H,H} = 6.8 \,\text{Hz}, \,1\text{H}, \,\text{H-3'}); \,4.00-3.97 \,(\text{m}, \,1\text{H}, \,\text{H-5'a}); \,3.92-3.89 \,(\text{m}, \,1\text{H}, \,\text{H-5'b});$ 3.83-3.82 (m, 1H, H-4'); 2.38-2.35 (m, 4H, H-g, H-g'); 2.15-2.10 (m, 2H, H-2'a); 2.08-2.05 (m, 2H, H-2'b); 1.69 (s, 3H, H-7); 1.55-1.48 (m, 4H, H-h, H-h'); 1.28-1.08 (m, 18H, H-i, H-i' – Hp'); 0.78 (t, ${}^{3}J_{HH} = 7.4$ Hz, 3H, H-j); 0.69 (t, ${}^{3}J_{HH} = 6.9$ Hz, 3H, H-q'). 13 C-NMR: (150MHz, MeOH-d4): δ [ppm] = 173.3 (C-f, C-f'); 166.4 (C-4); 152.5 (C-e, Ce'); 152.4 (C-2); 137.8 (C-6); 134.9 (C-b', C-b); 130.5 (d, C-c ,C-c'); 122.9 (C-d, C-d'); 112.1 (C-5); 85.8 (C-1'); 84.4 (d, ${}^{3}J_{CP} = 9.4 \text{ Hz}, \text{ C-4'}$; 70.3 (d, ${}^{2}J_{CP} = 5.5 \text{ Hz}, \text{ C-a/C-a'}$); 67.3 (d, ${}^{3}J_{CP} = 5.9 \text{ Hz}, \text{ C-5'}$); 62.5 (C-3'); 38.0 (C-2'); 35.0 (C-g/c-g'); 34.8 (C-g/c-g'); 33.1, 30.7, 30.6, 30.5, 30.4, 30.2, 23.3 (C-i,C-i' -C-p'); 28.1 (C-h, C-h'); 26.0 (C-h, C-h'); 14.1 (C-j); 14.5 (C-q'); 12.6 (C-7). ³¹P-NMR (243 MHz, MeOH-d4, decoupled): $\delta/ppm = -12.16$ (d, ${}^{2}J_{PP} = 20.3$ Hz, 1P, P_B); -12.78 (d, {}^{2}J_{PP} = 20.3 Hz, 1P, P_B); -12.78 (d, {}^{

20.4 Hz, 1P, P_{α}). HRMS: (ESI⁺, m/z) calcd. for $C_{41}H_{57}N_5O_{14}P_2Na$: 928.3275 [M+Na]⁺, found: 928.3252. RP18-HPLC: Method B, R_t = 21.9 min.

Ammonium-C₄H₉/CF₃-Ph-DiPPro-AZTDP 10d: General procedure E; Phosphoramidite 5i (179 mg, 0.282 mmol), AZT monophosphate (117 mg, 0.188 mmol) and 4,5-dicyanoimidazole activator solution (1.13 mL, 0.282 mmol) in 5 mL acetonitrile. Oxidation by addition of tertbutylhydroperoxide (5.5 molar solution in n-decane, 51 µL, 0.282 mmol). The product (80 mg, 0.088 mmol, 47%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 8.38-8.36 (m, 2H, H-h'/H-i'); 7.90-7.89 (m, 2H, H-h'/H-i'); 7.73 (s, 1H, H-6); 7.49-7.47 (m, 2H, H-c'); 7.44-7.41 (m, 2H, H-c); 7.25-7.23 (m, 2H, H-d'); 7.08-7.06 (m, 2H, H-d); 6.22 (dd, ${}^{3}J_{HH} = 6.8$ Hz, ${}^{3}J_{HH} = 6.8$ Hz, 1H, H-1'); 5.18-5.13 (m, 4H, H-a, H-a'); 4.43-4.41 (m, 1H, H-3'); 4.23-4.21 (m, 1H, H-5'a); 4.15-4.12 (m, 1H, H-5'b); 4.06-4.05 (m, 1H, H-4'); 2.56 (t, ${}^{3}J_{HH} = 7.4$ Hz, 2H, H-g); 2.37-2.32 (m, 1H, H-2'a); 2.31-2.27 (m, 1H, H-2'b); 1.90 (s, 3H, H-7); 1.68 (dt, ${}^{3}J_{HH} = 7.4$ Hz, ${}^{3}J_{HH} = 7.6$ Hz, 2H, H-h); 1.43 (dq, ${}^{3}J_{HH} = 7.5$ Hz, ${}^{3}J_{HH} = 7.5$ Hz, 7.5 Hz, 2H, H-i); 0.96 (t, ${}^{3}J_{HH} = 7.4$ Hz, 2H, H-i). 13 C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.6 (C-f); 166.3 (C-4); 166.3 (C-f'); 152.4 (C-2); 152.3 (C-e); 152.2 (C-e'); 137.4 (C-6); 135.9 (C-j'); 135.3 (C-b'); 134.8 (C-b); 134.3 (C-g'); 131.6 (C-h'/C-i'); 130.6 (d, ${}^{4}J_{CP} = 4.2$ Hz, C-c'); 130.1 (d, ${}^{4}J_{CP}$ = 2.5 Hz, C-c); 126.5 (C-h'/C-i'); 124.3 (CF₃); 122.6 (C-d'); 122.6 (C-d); 112.1 (C-5); 85.4 (C-1'); 84.4 (d, ${}^{3}J_{CP} = 9.2$ Hz, C-4'); 70.4 (2xd, ${}^{2}J_{CP} = 6.3$ Hz, ${}^{2}J_{CP} = 4.9$ Hz C-a, Ca'); 67.3 (d, ${}^{3}J_{CP} = 5.6$ Hz, C-5'); 62.2 (C-3'); 37.8 (C-2'b); 37.7 (C-2'a); 34.4 (C-g); 27.7 (C-h); 22.9 (C-i); 14.1 (C-j); 12.4 (C-7). ¹⁹F-NMR (565 MHz, MeOH-d4): $\delta/ppm = -65.7$. ³¹P-NMR (243 MHz, MeOH-d4, decoupled): $\delta/\text{ppm} = -12.14$ (d, ${}^{2}J_{P,P} = 20.2$ Hz, 1P, P_b); -12.67-12.81 (m,

1P, P_{α}). HRMS: (ESI⁺, m/z) calcd. for $C_{37}H_{39}F_3N_5O_{14}P_2$: 896.1921 [M+H]⁺, found: 896.1914. RP18-HPLC: Method B, R_t = 18.0 min.

bis(4-Ammonium-C₄H₉-Di*PP*ro-AZTDP 11a: General E: Symmetric procedure pentanoyloxymethyl)-*N*,*N*-di*iso*propylaminophosphoramidite (133 mg, 0.243 mmol), AZT monophosphate (101 mg, 0.162 mmol) and 4,5-dicyanoimidazole activator solution (970 µL, 0.243 mmol) in 4 mL acetonitrile. Oxidation by addition of tertbutylhydroperoxide (5.5 molar solution in *n*-decane, 44 µL, 0.243 mmol). The product 84 mg (0.10 mmol, 63%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 7.72 (s, 1H, H-6); 7.42-7.40 (m, 4H, H-c); 7.07-7.05 (m, 4H, H-d); 6.21 (dd, ${}^{3}J_{H,H} = 6.8$ Hz, ${}^{3}J_{H,H} = 6.8$ Hz, 1H, H-1'); 5.14-5.11 (m, 4H, H-a); 4.41-4.39 (m, 1H, H-3'); 4.21-4.18 (m, 1H, H-5'a); 4.13-4.10 (m, 1H, H-5'b); 4.04-4.03 (m, 1H, H-4'); 2.58 (t, ${}^{3}J_{HH} = 7.4$ Hz, 2H, H-g); 2.36-2.26 (m, 2H, H-2'); 1.90 (s, 3H, H-7); 1.74-1.69 (m 2H, H-h); 1.49-1.43 (m, 2H, H-i); 0.99 (t, ${}^{3}J_{HH} = 7.4$ Hz, 6H, H-j). ¹³C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.7 (C-f); 166.4 (C-4); 152.5 (C-e); 152.3 (C-2); 137.8 (C-6); 134.9 (C-b); 130.5 (d, ${}^{4}J_{CP} = 2.3$ Hz, C-c); 122.9 (C-d'); 122.6 (C-d); 112.2 (C-5); 85.8 (C-1'); 84.4 (d, ${}^{3}J_{CP} = 9.0$ Hz, C-4'); 70.3 (d, ${}^{2}J_{CP} = 6.1$ Hz C-a); 67.3 (d, ${}^{3}J_{CP} = 6.1$ Hz C-a); 67.3 (d, {}^{3}J_{CP} = 6.1 5.8 Hz, C-5'); 62.5 (C-3'); 38.0 (C-2'); 34.8 (C-g); 28.1 (C-h); 23.3 (C-i); 14.1 (C-j); 12.6 (C-7). $[\alpha]_{D}^{23} = 7$ (c = 0.1 mg/mL; CH₃OH). ³¹P-NMR (243 MHz, MeOH-d4, decoupled): δ /ppm = -12.2 $(d, {}^{2}J_{PP} = 20.4 \text{ Hz}, P_{\beta});$ -12.8 $(d, {}^{2}J_{PP} = 20.5 \text{ Hz}, P_{\alpha}).$ HRMS: $(\text{ESI}^{+}, \text{m/z})$ calcd. for $C_{34}H_{44}N_5O_{14}P_2$: 808.2355 [M+H]⁺, found: 808.2352. RP18-HPLC: Method B, R_t= 17.4 min.

Ammonium-C₆H₁₃-Di*PP*ro-AZTDP 11b: General procedure E: Symmetric bis(4heptanoyloxymethyl)-*N*,*N*-di*iso*propylaminophosphoramidite (217 mg, 0.36 mmol), AZT monophosphate (150 mg, 0.24 mmol) and 4,5-dicyanoimidazole activator solution (1.44 mL, 0.36 mmol) in 6 mL acetonitrile. Oxidation by addition of *tert* butylhydroperoxide (5.5 molar solution in *n*-decane, 66 µL, 0.36 mmol). The product 83 mg (0.09 mmol, 39%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 7.72 (s, 1H, H-6); 7.42-7.40 (m, 4H, H-c); 7.07-7.05 (m, 4H, H-d); 6.21 (dd, ${}^{3}J_{H,H} = 6.8$ Hz, ${}^{3}J_{H,H} = 6.8$ Hz, 1H, H-1'); 5.14-5.11 (m, 4H, H-a); 4.41-4.39 (m, 1H, H-3'); 4.21-4.18 (m, 1H, H-5'a); 4.13-4.10 (m, 1H, H-5'b); 4.03 (bs, 1H, H-4'); 2.58 (t, ${}^{3}J_{HH} = 7.4$ Hz, 2H, H-g); 2.36-2.26 (m, 2H, H-2'); 1.90 (s, 3H, H-7); 1.75-1.70 (m 2H, H-h); 1.44-1.41 (m, 2H, H-i); 1.37-1.36 (m, 8H, H-j, H-k); 0.93 (t, ${}^{3}J_{H,H} = 6.6$ Hz, 6H, H-l). 13 C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.7 (C-f); 166.4 (C-f) 4); 152.5 (C-e); 152.4 (C-2); 137.8 (C-6); 134.9 (C-b); 130.5 (d, ${}^{4}J_{CP} = 2.2$ Hz, C-c); 122.9 (C-d'); 122.6 (C-d); 112.2 (C-5); 85.8 (C-1'); 84.4 (d, ${}^{3}J_{C,P} = 9.1$ Hz, C-4'); 70.3 (d, ${}^{2}J_{C,P} =$ 6.0 Hz C-a); 67.3 (d, ${}^{3}J_{C,P} = 6.1$ Hz, C-5'); 62.5 (C-3'); 38.0 (C-2'); 35.0 (C-g); 32.7 (C-j/C-k); 29.9 (C-i); 25.9 (C-h); 23.6 (C-j/C-k); 14.4 (C-l); 12.6 (C-7). $[\alpha]_D^{23} = 12$ (c = 0.1 mg/mL; CH₃OH). ³¹P-NMR (243 MHz, MeOH-d4, decoupled): $\delta/\text{ppm} = -12.2$ (d, ² $J_{P,P} = 20.9$ Hz, P_{B}); -12.8 (d, ${}^{2}J_{PP} = 21.4 \text{ Hz}$, P_a). HRMS: (ESI⁺, m/z) calcd. for C₃₈H₅₂N₅O₁₄P₂: 864.2991 [M+H]⁺, found: 864.2978. RP18-HPLC: Method B, R_t= 20.0 min.

Ammonium-C₉H₁₉-Di*PP***ro-AZTDP 11c**: General procedure E; Symmetric bis(4decanoyloxymethyl)-*N*,*N*-di*iso*propylaminophosphoramidite (120 mg, 0.18 mmol), AZT monophosphate (72 mg, 0.12 mmol) and 4,5-dicyanoimidazole activator solution (720 μ L, 0.18 mmol) in 3 mL acetonitrile. Oxidation by addition of *tert* butylhydroperoxide (5.5 molar

solution in *n*-decane, 33 μL, 0.18 mmol). The product (58 mg, 0.060 mmol, 50%) was obtained as a colourless wad after ion exchange. ¹H-NMR: (600 MHz, MeOH-d4): δ [ppm] = 7.72 (s, 1H, H-6); 7.42-7.39 (m, 4H, H-c); 7.06-7.04 (m, 4H, H-d); 6.21 (dd, ${}^{3}J_{H,H} = 6.8$ Hz, ${}^{3}J_{H,H} = 6.8$ Hz, ${}^{1}H_{H}$ H-1'); 5.13-5.11 (m, 4H, H-a); 4.41-4.39 (m, 1H, H-3'); 4.21-4.18 (m, 1H, H-5'a); 4.13-4.10 (m, 1H, H-5'b); 4.04 (bs, 1H, H-4'); 2.57 (t, ${}^{3}J_{H,H} = 7.4$ Hz, 4H, H-g); 2.36-2.29 (m, 4H, H-2'); 1.90 (s, 3H, H-7); 1.76-1.71 (m 4H, H-h); 1.43-1.31 (m, 24H, H-i - H-n); 0.91 (t, ${}^{3}J_{H,H} = 6.8$ Hz, 6H, H-o). ¹³C-NMR: (150 MHz, MeOH-d4): δ [ppm] = 173.7 (C-f); 166.5 (C-4); 152.6 (C-e); 152.5 (C-2); 137.8 (C-6); 134.9 (C-b); 130.5 (C-c); 122.9 (C-d); 112.2 (C-5); 85.8 (C-1'); 84.4 (d, ${}^{3}J_{C,P} = 9.5$ Hz, C-4'); 70.4 (d, ${}^{2}J_{C,P} = 5.8$ Hz C-a); 67.3 (d, ${}^{3}J_{C,P} = 5.9$ Hz, C-5'); 62.5 (C-3'); 38.0 (C-2'); 35.0 (C-g); 33.1, 30.6, 30.4, 30.2, 23.7 (C-i - C-n); 26.0 (C-h); 14.4 (C-o); 12.6 (C-7). [α]_D²³ = 37 (c = 0.1 mg/mL; CH₃OH). ³¹P-NMR (243 MHz, MeOH-d4, decoupled): δ/ppm = -12.2 (d, ${}^{2}J_{P,P} = 20.4$ Hz, P_β); -12.8 (d, ${}^{2}J_{P,P} = 20.4$ Hz, P_α). HRMS: (ESI⁺, m/z) calcd. for C₄₄H₆₄N₅O₁₄P₂: 948.3930 [M+H]⁺, found: 948.3915. RP18-HPLC: Method B, R_t = 24.4 min.

Compounds evaluation methods.

Hydrolysis studies in phosphate buffer at pH 7.3: Hydrolysis was started by addition of 300 μ L 50 mM phosphate buffer (547 mg (3.85 mol) disodium hydrogen phosphate and 155 mg (0.890 mol) potassium dihydrogen phosphate, filled-up with water to 100 mL, pH-value: by addition of NaOH or phosphoric acid) to 300 μ L of 1.9 mM solution of the Di*PP*ro-NDP in DMSO. The hydrolysis solution was incubated at 37 °C. Aliquots were taken at certain points of time and frozen in liquid nitrogen. Directly after unfreezing analysis was followed by analytical RP-18-HPLC.

Hydrolysis studies in CEM/0 cell extracts: $50 \ \mu\text{L}$ CEM/0 cell extracts were mixed with $10 \ \mu\text{L}$ of water. By addition of $10 \ \mu\text{L}$ of a 6 mM solution of the Di*PP*ro-NDP in DMSO the hydrolysis was started. 6 to 8 of these samples were prepared and incubated at 37 °C for different periods of time. Hydrolysis was stopped by quenching with 150 μ L methanol. After 5 min on ice, 5 min in an ultra-sonic bath and centrifugation at 13000 rmp (Heraeus, Biofuge Pico) the samples were filtered and frozen in liquid nitrogen until 35 μ L were injected to the analytical RP-18-HPLC. Hydrolysis was monitored for a maximum of 10 hours.

Hydrolysis studies in RPMI-medium: $50 \ \mu L$ RPMI-culture medium without FCS were mixed with $10 \ \mu L$ of water. By addition of $10 \ \mu L$ of a 6 mM solution of the Di*PP*ro-NDP in DMSO the hydrolysis was started. 6 to 8 of these samples were prepared and incubated at 37 °C for different periods of time. After a period of time 150 μL methanol were added and the sample were frozen in liquid nitrogen until 35 μL were injected to the analytical RP-18-HPLC.

Hydrolysis studies with PLE: 10 μ L of a 6 mM solution of the Di*PP*ro-NDP in DMSO were mixed with 15 μ L DMSO and 100 μ L 50 mM phosphate buffer. By addition of 7.5 μ L PLEsolution (3 mg/mL in PBS, 17 units/mg) the hydrolysis was started. Hydrolysis was stopped by quenching with 132.5 μ L methanol. After 5 min on ice, 5 min in an ultra-sonic bath and centrifugation at 13000 rmp (Heraeus, Biofuge Pico) the samples were filtered and frozen in liquid nitrogen until 35 μ L were injected to the analytical RP-18-HPLC.

Preparation of CEM cell extracts: Human $CD4^+$ T-lymphocyte CEM cells were grown in RPMI-1640-based cell culture medium to a final density of about 3 x 10⁶ cells/mL. After centrifugation for 10 min at 1250 rpm (Heraeus, Megafuge 3.0R) at 4 °C, washing twice with cold phosphate buffered saline (PBS) the pellet was resuspended at 10⁸ cells/mL and sonicated

Journal of Medicinal Chemistry

three times for 10 sec. After a second centrifugation of this suspension at 10000 rpm (Heraeus, Megafuge 3.0R) the supernatant was divided into aliquots and frozen to -80 °C until used.

Antiviral assay: Inhibition of HIV-1(III_B) and HIV-2(ROD)-induced cytopathogenicity in wild-type CEM/0 or thymidine kinase-deficient CEM/TK⁻ cell cultures was measured in microtiter 96-well plates containing ~3 x 10^5 CEM cells/mL infected with 100 CCID₅₀ of HIV per milliliter and containing appropriate dilutions of the test compounds. After 4-5 days of incubation at 37 °C in a CO₂-controlled humidified atmosphere, CEM giant (syncytium) cell formation was examined microscopically. EC₅₀ (50% effective concentration) was defined as the compound concentration required to inhibit HIV-induced giant cell formation by 50%.

ASSOCIATED CONTENT

Supporting Information Detailed information for the syntheses, ¹³C-NMR-data, ¹⁹F-NMR-data, IR-data and specific rotation are listed as supporting information. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

*chris.meier@chemie.uni-hamburg.de

Author Contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

ACKNOWLEDGMENT

The work conducted by CM has been supported by the Deutsche Forschungsgemeinschaft (DFG; Me1161/13-1). We thank Leen Ingels, Evelyne Van Kerckhove, Sandra Claes, Ria Van Berwaer and Lizette van Berckelaer for excellent technical assistance. The biological assays were performed with the financial support of the KU Leuven (GOA 10/014).

ABBREVIATIONS

AZT, 3'-azido-3'-deoxythymidine; d4T, 3'-deoxy-2',3'-didehydrothymidine; DCI, 4,5dicyanoimidazole; DIPA, *N,N*-di*iso*propylamine; FCS, fetal calf serum; HILIC, hydrophilic interaction liquid chromatography; NDP, nucleoside diphosphate; NMP, nucleoside monophosphate; PLE, pig live resterase; R_t, retention time; RP, reversed phase; TK, thymidine kinase

REFERENCES

1. Balzarini, J.; Herdewijn, P.; De Clercq, E. Differential patterns of intracellular metabolism of 2',3'-didehydro-2',3'-dideoxythymidine and 3'-azido-2',3'-didesoxythymidine, two potent anti-human immunodeficiency virus compounds. *J. Biol. Chem.* **1989**, *264*, 6127-6133.

2. Balzarini, J. Metabolism and mechanism of antiretroviral action of purine and pyrimidine derivatives. *Pharm. World Sci.* **1994**, *16*, 113-126.

3. Harrington, J. A.; Reardon, J. E.; Spector, T. 3'-Azido-3'-deoxythymidine (AZT) monophosphate: an inhibitor of exonucleolytic repair of AZT-terminated DNA. *Antimicrob. Agents Chemother.* **1993**, *37*, 918-920.

4. Sommadossi, J. P.; Carlisle, R.; Zhou, Z. Cellular pharmacology of 3'-azido-3'deoxythymidine with evidence of incorporation into DNA of human bone marrow cells. *Mol. Pharmacol.* **1989**, *36*, 9-14.

5. Hecker, S. J.; Erion, M. D. Prodrugs of phosphates and phosphonates (perspective). *J. Med. Chem.* **2008**, *51*, 2328-2345.

6. Pradere, U.; Garnier-Amblard, E. C.; Coats, S. J.; Amblard, F.; Schinazi, R. F. Synthesis of nucleoside phosphate and phosphonate prodrugs. *Chem. Rev.* **2014**, *114*, 9154-9218.

7. Meier, C. *cyclo*Sal phosphates as chemical Trojan horses for intracellular nucleotide and glycosyl-monophosphate delivery - chemistry meets biology. *Eur. J. Org. Chem.* **2006**, *5*, 1081-1102.

8. Gisch, N.; Pertenbreiter, F.; Balzarini, J.; Meier C. 5-(1-Acetoxyvinyl)-*cyclo*Saligenyl-2',3'dideoxy-2',3'- didehydrothymidine monophosphates, a second type of new, enzymatically activated *cyclo*Saligenyl pronucleotides *J. Med. Chem.* **2008**, *51*, 8115-8123.

9. Rios Morales, E. H.; Balzarini, J.; Meier, C. Stereoselective synthesis and antiviral activity of methyl-substituted *cyclo*Sal-pronucleotides *J. Med. Chem.* **2012**, *55*, 7245-7252.

10. Peyrottes, S.; Egron, D.; Lefebvre, I.; Gosselin, G.; Imbach, J. L.; Perigaud, C. SATE pronucleotide approaches: an overview. *Mini-Rev. Med. Chem.* **2004**, *4*, 395-408.

Gouy, M.-H.; Jordheim, L. P.; Lefebvre, I.; E.; Cros, C.; Dumontet, S.; Peyrottes, Périgaud
 C. Special feature of mixed phosphotriester derivatives of cytarabine. *Bioorg. Med. Chem.* 2009, *17*, 6340-6347.

12. Farquhar, D.; Khan, S.; Srivastva, D. N.; Saunders, P. P. Synthesis and antitumor evaluation of bis[(pivaloyloxy)methyl] 2'-deoxy-5-fluorouridine 5'-monophosphate (FdUMP): a strategy to introduce nucleotides into cells. *J. Med. Chem.* **1994**, *37*, 3902-3909.

13. McGuigan, C.; Pathirana, R. N.; Mahmood, N.; Devine, K. G.; Hay, A. J. Aryl phosphate derivatives of AZT retain activity against HIV1 in cell lines which are resistant to the action of AZT. *Antiviral Res.* **1992**, *17*, 311-321.

14. Mehellou, Y.; Balzarini, J.; McGuigan, C. Aryloxy phosphoramidate triesters: a technology for delivering monophosphorylated nucleosides and sugars into cells. *ChemMedChem* **2009**, *4*, 1779-1791.

15. Furman, P. A.; Fyfe, J. A.; St Clair, M. H.; Weinhold, K.; Rideout, J. L.; Freeman, G. A.; Lehrman, S. N.; Bolognesi, D. P.; Broder, S.; Mitsuya, H. Phosphorylation of 3'-azido-3'deoxythymidine and selective interaction of the 5'-triphosphate with human immunodeficiency virus reverse transcriptase. *Proc. Natl. Acad. Sci., USA* **1986**, *83*, 8333-8337.

16. Lavie, A.; Schlichting, I.; Vetter, I. R.; Konrad, M.; Reinstein, J.; Goody, R. S. The bottleneck in AZT activation. *Nature Med.* **1997**, *3*, 922-924.

17. Hostetler, K. Y.; Stuhmiller, L. M.; Lenting, H. B. M.; van den Bosch, H.; Richman, D. D. Synthesis and antiretroviral activity of phospholipid analogs of azidothymidine and other antiviral nucleosides. *J. Biol. Chem.* **1990**, *265*, 6112-6117.

18. van Wijk, G. M. T.; Hostetler, K. Y.; van den Bosch, H. Lipid conjugates of antiretroviral: release of antiretroviral nucleoside monophosphate by a nucleoside diphosphate diglyceride hydrolase activity from rat liver mitochondria. *Biochim. Biophys. Acta, Lipids Lipid Metab.* **1991,** *1084,* 307-310.

19. Hostetler, K. Y.; Richman, D. D.; Carson, D. A.; Stuhmiller, L. M.; van Wijk, G. M. T.; van den Bosch, H. Cytidine diphosphate diglygeride analogs. *Antimicrob. Agents Chemother.* **1992,** *36*, 2025-2029.

20. Hostetler, K. Y.; Parker, S.; Sridhar, C. N.; Martin, M. J.; Li, J. L.; Stuhmiller, L. M.; van Wijk, G. M. T.; van den Bosch, H.; Gardner, M. F. Acyclovir diphosphate dimyristoylglycerol: a phospholipid prodrug with activity against acyclovir-resistant herpes simplex virus. *Proc. Natl. Acad. Sci., USA* **1993**, *90*, 11835-11839.

21. Bonnaffé, D.; Dupraz, B.; Ughetto-Monfrin, J.; Namane, A.; Huynh Dinh, T. Synthesis of acyl pyrophosphates. Application to the synthesis of nucleotide lipophilic prodrugs. *Tetrahedron Lett.* **1995**, *36*, 531 - 534.

22. Bonnaffé, D.; Dupraz, B.; Ughetto-Monfrin, J.; Namane, A.; Huynh Dinh, T. Synthesis of lipophilic prodrugs containing two inhibitors targeted against different phases of the HIV replication cycle. *Nucleos. Nucleot.* **1995**, *14*, 783 - 787.

23. Bonnaffé, D.; Dupraz, B.; Ughetto-Monfrin, J.; Namane, A.; Huynh Dinh, T. Potential lipophilic nucleotide prodrugs: synthesis, hydrolysis, and antiretroviral activity of AZT and d4T acyl nucleotides. *J. Org. Chem.* **1996**, *61*, 895 - 902.

24. Jessen, H. J.; Schulz, T.; Balzarini, J.; Meier, C. Bioreversible protection of nucleoside diphosphates. *Angew. Chem. Int. Ed.* **2008**, *47*, 8719-8722.

25. Schulz, T.; Balzarini, J.; Meier, C. The Di*PP*ro Approach: synthesis, hydrolysis, and antiviral activity of lipophilic d4T diphosphate prodrugs. *ChemMedChem* **2014**, *9*, 762-775.

26. Weinschenk, L.; Gollnest, T.; Meier, C.; Schols, D.; Balzarini, J. Bis(benzoyloxybenzyl)-Di*PP*ro nucleoside diphosphates of anti-HIV active nucleoside analogues. *ChemMedChem* **2015**, *10*, 891-900.

27. Schneider, B.; Sarfati, R.; Deville-Bonne, D.; Véron, M. Role of nucleoside diphosphate kinase in the activation of anti-HIV nucleoside analogs. *J. Bioenerg. Biomembr.* **2000**, *32*, 317-324.

28. Balzarini, J.; Naesens, L.; Aquaro, S.; Knispel, T.; Perno, C.-F.; De Clercq, E.; Meier. C. Intracellular metabolism of *cyclo*Saligenyl 3'-azido-2',3'-dideoxythymidine monophosphate, a prodrug of 3'-azido-2',3'-dideoxythymidine (zidovudine). *Mol. Pharmacol.* **1999**, *56*, 1354-1361.

Table of contents graphic

