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CHEMOSELECTIVE REMOVAL OF ALLYLIC FORMYLOXY GROUP USING Sml₂.

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Abstract: A variety of allylic formates are chemoselectively transformed into the corresponding olefins in excellent yields on treatment with Sml2 in THF-HMPA-H₂O (20:5:1), even though allylic acetate and allylic ether groups are present in the same molecule.

Although a number of synthetic methods for deoxygenation have been reported¹, there are few methods for the chemoselective removal of allylic oxygen group under mild conditions.

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Herein, we wish to report a highly efficient method for the chemoselective transformation of allylic formates into the corresponding olefins using Sml2. In fact, this reaction did readily take place chemoselectively at the allylic formate center leaving the other allylic ester or ether intact when the latter functionality was present in the same molecule. The reactions were simply carried out by treating the substrates with a four molar excess of Sml2 in a mixture of THF, HMPA, and H2O (20:5:1 v/v) at room temperature. It was found that the reaction could be terminated within a few minutes to furnish the corresponding olefins in excellent yields with some extent of double bond migration (Table 1). Myrtenyl formate and perillyl formate were smoothly converted into pinene, limonene and their corresponding isomers, respectively in excellent yield (entries 1 and 2)². In order to elucidate the reaction process, we selected the optically active carvyl formate derived from (R)-(-)-carvone. While starting from (-)-carvyl formate, the value of the optical rotation of generated limonene was extremely low (entry 3)3. It was suggested that the generated anion was delocalized and racemized this molecule, and then trapped a proton from H2O to produce olefins⁴ as shown in Scheme 1.

Most noteworthy from the synthetic viewpoint was that the other functional groups such as allylic acetate and allylic ether in the same molecules were inert (entries 4-6). Therefore, this indicated that the present reaction was essentially different from the palladium-mediated reaction⁵ in view of chemoselectivity. Interestingly, two formyloxy groups were removed from cholesten-3,4-diformate **7** to afford cholest-3,5-diene **21**⁶ in 96% yield. On the other hand, cholesten-3,4-diacetate **8**



Table 1. Deoxygenation of allylic formates.

^a Yield after purification of product by SiO₂ chromatography.

^b Isomeric ratio was determined by ¹H-NMR (270 MHz).

(continued)



 Table 1 (continued). Deoxygenation of allylic formates.





did not react at all (entries 7 and 8). Upon exposure to 1,4methanonaphthalene-5,8-diformate **9** and 5-formyloxy-1,4-methanonaphthalene-8-acetate **10**, similar reductive eliminations were observed to give the 1,3-diene compound **23**⁷ as a common product (entries 9 and 10). The pathway of the present reactions was presumed to start by the generated anion migrating to eliminate acyloxy group as a leaving group and produce the 1,3-diene compound **23** as shown in Scheme 2. In the case of diformate **9**, the reaction might predominantly proceed *via* pathway b and c.

However, even in the same substitution system, the benzyl ether group was inert under the same conditions (entry 11).

In conclusion, we found a specific and efficient removal of allylic formate functionality using Sml₂ without affecting other O-allylic functionalities.

Experimental Section

¹H-NMR spectra were recorded at 270 MHz on a JEOL GSX-270 spectrometer. ¹³C-NMR spectra were recorded at 67.5 MHz on a JEOL



Scheme 2

GSX-270 spectrometer. IR spectra were recorded on a JASCO VALOR-III. Low resolution mass spectra (MS) and high resolution mass spectra (HRMS) were recorded on a JEOL JMS-D-300 mass spectrometer instrument. Elemental analyses were performed by Yanako MT-3. Optical rotations were measured on a JASCO DIP-370 polarimeter. Melting points were determined with a Yanagimoto micro melting point apparatus. All melting points are uncorrected. Dichloromethane (CH₂Cl₂) was distilled from P₂O₅. Hexamethylphosphoramide (HMPA) was distilled from calcium hydride. Tetrahydrofuran (THF) was freshly distilled from sodium benzophenone ketyl before use. (+)-Limonene (purity 95%) as standard sample was purchased from Tokyo Chemical Industry Co., Ltd.. Column chromatography was carried out with Merck silica gel 0.063-0.2 mm, 70-230 mesh.

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Preparation of the allylic formates.

The allylic formates used in this study were prepared from the corresponding allylic alcohols following the typical procedure.

 $(1\alpha, 4\alpha, 4a\alpha, 5\beta, 8\beta, 8a\alpha)$ -1, 4, 4a, 5, 8, 8a-Hexahydro-1,4methanonaphthalene-5, 8-diol formate (9) (Typical Procedure) To an ice-cooled and stirred solution of 3β,7β-dihydroxy-endotricyclo[6.2.1.0^{2,7}]undeca-4,9-diene⁷ (178 mg, 1.0 mmol) in CH₂Cl₂ (2 ml) and pyridine (0.4 ml) was added dropwise acetyl formate⁸ (211 mg, 2.4 mmol). After 20 min, acetyl formate (211 mg, 2.4 mmol) and pyridine (0.4 ml) was more added. The reaction mixture was stirred at the same temperature for 30 min and dilluted with ether and water. The organic layer was sequentially washed with water, ag. 10% KHSO4, sat. aq. NaHCO3 and sat. aq. NaCl, dried over MgSO4, filtered and concentrated under reduced pressure. The residue was column chromatographed on silica gel (eluted with hexane-ether=5:1) to give 232 mg (99% yield) of diformate 9 as a colorless oil. IR (CHCl3) v max (cm⁻¹): 3029, 2983, 2934, 1727, 1173; ¹H-NMR (CDCl₃) δ: 1.29 (dt, J=8.1, 1.7 Hz, 1H), 1.36 (dt, J=8.1, 1.7 Hz, 1H), 2.89 (ddd, J=5.4, 3.4, 1.7 Hz, 2H), 3.06 (dd, J=3.4, 1.7 Hz, 1H), 3.09 (dd, J=3.4, 1.7 Hz, 1H), 5.39 (s, 2H), 5.52 (brt, J=5.4 Hz, 1H), 5.55 (brt, J=5.4 Hz, 1H), 5.84 (t, J=1.7 Hz, 2H) 8.15 (s, 2H); ¹³C-NMR (CDCl₃) δ: 37.9 (d), 46.0 (d), 48.5 (t), 69.3 (d), 127.9 (d), 135.6 (d), 160.5 (d); EIMS m/z (relative intensity): 234 (M+, 1.2), 189 (16), 66 (100); Anal. Calcd. for C13H14O4: C, 66.66; H, 6.02. Found: C, 66.69; H, 6.10.

(-)-Myrtenyl formate (1)²

Prepared in 95% yield as a colorless oil from (-)-myrtenol. $[\alpha]_D^{25}$ = -55° (c 1.5, CHCl3). Anal. Calcd. for C11H16O2: C, 73.30; H, 8.95. Found: C, 73.13; H, 8.88.

(-)-Perillyl formate (2)²

Prepared in 97% yield as a colorless oil from (-)-perillyl alcohol. $[\alpha]_D^{25}$ = -81.3° (c 1.5, CHCl₃). Anal. Calcd. for C₁₁H₁₆O₂: C, 73.30; H, 8.95. Found: C, 73.03; H,8.80.

(-) -Carvyl formate (3)

Prepared in 93% yield as a colorless oil from a 3:1 mixture of diastereomers of (-)-carveol. $[\alpha]_D^{25}$ = -102.5° (c 1.0, CHCl3). Anal. Calcd. for C11H16O2: C, 73.30; H, 8.95. Found: C, 73.29; H, 9.08.

4-Ethoxycarbonyl-3-methyl-2-cyclohexen-1-ol formate (4)

Prepared in 91% yield as a colorless oil from 4-ethoxycarbonyl-3methyl-2-cyclohexen-1-ol (a 3:2 mixture of stereoisomers), which was easily obtained upon reduction of 4-ethoxycarbonyl-3-methyl-2cyclohexanone with a combination of NaBH4 and CeCl3 in methanol at 0°C. IR (CHCl3) v max (cm⁻¹): 3026, 2983, 2940, 1722, 1446, 1259, 1178; ¹H-NMR (CDCl3) δ : 1.26 (t, J=7.3 Hz, 1.2H), 1.28 (t, J=7.1 Hz, 1.8H), 1.76 (d, J=1.0 Hz, 1.8H), 1.78 (d, J=1.0 Hz, 1.2H), 1.80-2.12 (m, 7H), 2.96 (m, 1H), 4.16 (q, J=7.3 Hz, 0.8H), 4.18 (q, J=7.1 Hz, 1.2H), 5.34 (m, 1H), 5.65 (m, 1H), 8.04 (d, J=1.0 Hz, 0.4H), 8.06 (d, J=1.0 Hz, 0.6H); Anal. Calcd. for C11H16O4: C, 62.25; H, 7.60. Found: C, 62.13; H, 7.60. (*2E,6E*)-8-Acetoxy-2,6-dimethyl-2,6-octadien-1-ol formate (5) Prepared in 75% yied as a colorless oil from (*2E,6E*)-8-acetoxy-2,6dimethyl-2,6-octadien-1-ol⁹. Anal. Calcd. for C₁₃H₂₀O₄: C, 64.98; H, 8.39. Found: C, 65.19; H, 8.45.

(1*S*, 2*S*, 5*R*, 6*S*, 9*R*)-6-Benzyloxy-9-benzyloxymethyl-1methylbicyclo[3.3.1]-3-nonen-2-ol formate (6)

Prepared in 92% yield as a colorless oil from (1*S*, 2*S*, 5*R*, 6*S*, 9*R*)-6benzyloxy-9-benzyloxymethyl-1-methybicyclo[3.3.1]-3-nonen-2-ol,

which was obtained upon exposure of (1S, 5R, 6S, 9R)-6-benzyloxy-9-benzyloxymethyl-1-methybicyclo[3.3.1]-3-nonen-2-one¹⁰ with diisobutyl aluminum hydride (DIBAL) in CH₂Cl₂ at -78°C. [α]_D²⁵= -45.0° (c 1.0, CHCl₃); IR (CHCl₃) v max (cm⁻¹): 3011, 2935, 2876, 1717, 1454, 1367, 1190; ¹H-NMR (CDCl₃) & 0.86 (s, 3H), 1.28 (m, 1H), 1.64-1.91 (m, 3H), 2.20 (m, 1H), 3.03 (m, 1H), 3.48 (t, J=9.3 Hz, 1H), 3.58 (dd, J=9.3, 5.9 Hz, 1H), 3.64 (ddd, J=9.3, 5.9, 3.9 Hz, 1H), 4.43 (d, J=12.2 Hz, 1H), 4.48 (d, J=12.2 Hz, 1H), 4.53 (d, J=12.2 Hz, 1H), 4.55 (d, J=12.2 Hz, 1H), 5.39 (brs, 1H), 5.76 (dd, J=9.8, 1.9 Hz, 1H), 5.97 (ddd, J=9.8, 6.1, 1.9 Hz, 1H), 7.23-7.38 (m, 10H), 8.20 (s, 1H); ¹³C-NMR (CDCl₃) &: 24.72 (t), 24.74 (q), 28.5 (t), 35.4 (s), 35.9 (d), 45.3 (d), 68.0 (t), 70.0 (t), 73.2 (t), 74.2 (d), 132.6 (d), 138.3 (s), 138.9 (s), 161.2 (d); EIMS m/z (relative intensity): 406 (M⁺, 3), 360 (3), 315 (99), 91 (100); Anal. Calcd for C26H₃₀O4: C, 76.82; H, 7.44. Found: C, 76.81; H, 7.48.

(3S, 4R)-3,4-Diformyloxy-5-cholestene (7)

Prepared in 94% yield from (3*S*, 4*R*)-cholest-5-en-3,4-diol¹¹. A sample was recrystalized from acetone-hexane to give a colorless needle (mp 138-139°C). [α]_D²⁵= -122.7° (c 1.0, CHCl₃); IR (CHCl₃) v max (cm⁻¹):

2962, 2935, 2870, 1716, 1467, 1171; ¹H-NMR (CDCl₃) δ : 0.68 (s, 3H), 0.85 (d, J=6.6 Hz, 3H), 0.87 (d, J=6.6 Hz, 3H), 0.91 (d, J=6.6 Hz, 3H), 0.94-1.40 (m, 14H), 1.14 (s, 3H), 1.41-1.67 (m, 6H), 1.71-2.18 (m, 6H), 4.89 (ddd, J=11.7, 4.2, 3.6 Hz, 1H), 5.64 (d, J=3.2 Hz, 1H), 5.87 (dd, J=5.1, 2.7 Hz, 1H), 8.00 (d, J=0.7 Hz, 1H), 8.12 (s, 1H); ¹³C-NMR (CDCl₃) δ : 11.9 (q), 18.7 (q), 20.4 (q), 20.5 (t), 22.5 (t), 22.6 (q), 22.8 (q), 23.8 (t), 24.2 (t), 28.0 (d), 28.2 (t), 31.6 (d), 32.0 (t), 35.8 (d), 36.2 (t), 36.1 (s), 36.6 (t), 39.5 (t), 39.6 (t), 42.3 (s), 50.2 (d), 56.1 (d), 56.8 (d), 72.4 (d), 75.7 (d), 132.8 (d), 137.7 (s), 160.1 (d), 160.3 (d); EIMS m/z (relative intensity): 458 (M⁺, 0.8), 412 (100); Anal. Calcd for C₂₉H₄₆O₄: C, 75.94; H, 10.11. Found: C, 75.76; H, 10.10.

(3S, 4R)-3,4-Diacetoxy-5-cholestene $(8)^{11}$

1S-(1 α , 4 α , 4a α , 5 β , 8 β , 8a α)-1, 4, 4a, 5, 8, 8a-Hexahydro-5acetoxy-1,4-methanonaphthalene-8-ol formate (10)

Prepared in 97% yield as a colorless oil from (+)-1R-(1 α , 4 α , 4 α , 5 β , 8 β , 8 α)-1, 4, 4 α , 5, 8, 8 α -hexahydro-8-acetoxy-1,4-methanonaphthalen-5-ol¹². IR (CHCl₃) v max (cm⁻¹): 3027, 2980, 2938, 1725, 1248, 1182; ¹H-NMR (CDCl₃) δ : 1.28 (dt, J=8.4, 1.4 Hz, 1H), 1.33 (dt, J=8.4, 1.7 Hz, 1H), 2.13 (s, 3H), 2.84 (td, J=3.2, 1.7 Hz, 1H), 2.88 (td, J=3.2, 1.7 Hz, 2H), 3.03 (ddd, J=8.1, 3.2, 0.9 Hz, 1H), 3.09 (ddd, J=8.1, 3.2, 0.9 Hz, 1H), 5.38 (s, 2H), 5.40 (m, 1H), 5.53 (m, 1H), 5.83 (t, J=1.7 Hz, 1H), 8.15 (s, 1H); ¹³C-NMR (CDCl₃) δ : 21.1 (q), 38.0 (d), 45.9 (d), 48.5 (t), 69.3 (d), 69.5 (d), 127.3 (d), 128.5 (d), 135.5 (d), 160.6 (d), 170.6 (s); EIMS m/z (relative intensity): 248 (M⁺, 6), 203 (54), 66 (100); Anal. Calcd for C14H16O4: C, 67.73; H, 6.50. Found: C, 67.59; H, 6.55.

(Z)-4-Benzyloxy-2-buten-1-ol formate (11)

Prepared in 90% yield as a colorless oil from (*Z*)-4-benzyloxy-2-buten-1-ol¹³. IR (CHCl₃) v max (cm⁻¹): 3031, 3012, 2935, 2861, 1724, 1169; ¹H-NMR (CDCl₃) δ : 4.14 (ddd, J=6.1, 1.5, 0.7 Hz, 2H), 4.53 (s, 2H), 4.73 (dd, J=6.1, 1.2 Hz, 2H), 5.73 (dtt, J=11.3, 6.1, 1.5Hz, 1H), 5.87 (dtt, J=11.3, 6.1, 1.2 Hz), 7.25-7.40 (m, 5H), 8.05 (t, J=0.7 Hz, 1H); EIMS m/z (relative intensity): 177 (M⁺-CHO, 1.7%), 105 (100), 91 (80); Anal. Calcd for C1₂H1₄O₃: C, 69.89; H, 6.84. Found: C, 69.68; H, 6.94.

Removal of allylic formyloxy group from substrates

The removal of an allylic formyloxy group from various substrates was carried out by following the typical procedure.

Cholest-3,5-diene (21) (Typical procedure)

A 30 ml flask, equipped with a closed stopcock-controlled argon inlet, a magnetic stirrer and a vacuum adapter, was charged with HMPA (1.6 ml) and water (0.32 ml). The vacuum was replaced with argon to keep the entering system oxygen free. A solution of samarium diiodide (0.25 M solution in THF 6.4 ml, 1.6 mmol) was added. To this mixture was added in one portion a solution of diformate 7 (183 mg, 0.4 mmol) in THF (1.6 ml). After being stirred at room temperature for 5 min, the reaction mixture was diluted with sat. aq. NH4Cl (20ml) and ether (10 ml). The organic layer was taken up and the aqueous layer was extracted with ether (2x 5ml). The combined organic layers were washed with water (3x 10 ml) and sat. aq. NaCl (10 ml), dried over MgSO4 and concentrated under reduced pressure. The residue was chromatographed on silica gel (eluted with hexane) to give 140 mg

(96%) of diene **21** as a white solid (mp 79-80°C). $[\alpha]_D^{25}$ = -124.9° (c 1.0, CHCl₃); IR (CHCl₃) v max (cm⁻¹): 3010, 2949, 2869, 1468, 909; ¹H-NMR (CDCl₃) δ : 0.70 (s, 3H), 0.86 (d, J=6.6 Hz, 6H), 0.92 (d, J=6.6 Hz, 3H), 0.95 (s, 3H), 0.96-2.25 (m, 26H), 5.39 (m, 1H), 5.59 (m, 1H), 5.92 (brd, J=9.8 Hz, 1H); ¹³C-NMR (CDCl₃) δ : 12.0 (q), 18.7 (q), 18.8 (q), 21.0 (t), 22.6 (q), 22.8 (q), 23.1 (t), 23.8 (t), 24.2 (t), 28.0 (d), 28.3 (t), 31.8 (t), 33.8 (t), 35.2 (s), 35.8 (d), 36.2 (t), 39.5 (t), 39.8 (t), 42.5 (s), 48.4 (d), 56.2 (d), 57.0 (d), 123.2 (d), 125.0 (d), 129.0 (d), 141.5 (s); EIMS m/z (relative intensity): 368 (M⁺, 100), 353 (M⁺-Me, 28); HRMS Found: 368.3422, C₂₇H44 (M⁺) requires; 368.3433; Anal. Calcd for C₂₇H44: C, 87.97; H, 12.03. Found: C, 87.95; H, 12.02.

β -Pinene (12) and α -Pinene (13)

Isolated as a 9:1 mixture of inseparable regioisomers **12** and **13** in 90% yield after column chromatography (eluted with pentane).

Limonene (14) and p-Mentha-7,8-diene (15)

Isolated as a 6:4 mixture of inseparable regioisomers **14** and **15** in 93% yield after column chromatography (eluted with pentane).

Limonene (15)

Isolated in 98% yield after column chromatography (eluted with pentane). $[\alpha]_D^{25}$ = -0.9° (c 2.0, CHCl3) [standard sample: (+)-limonene $[\alpha]_D^{25}$ = +108.5° (c 2.0, CHCl3)].

Ethyl 2-methyl-3-cyclohexen-carboxylate (16), trans-Ethyl 2methyl-2-cyclohexen-carboxylate (17a) and cis-Ethyl 2methyl-2-cyclohexen-carboxylate (17b)

Isolated as a 4:6 mixture of regioisomers 16 and 17 in 94% yield after

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column chromatography (eluted with hexane-ether=20:1). Compound **17** was a 2:1 mixture of stereoisomers **17a** and **17b**. IR (CHCl₃) v max (cm⁻¹): 3025, 2940, 2874, 1724, 1179, 1034; ¹H-NMR (CDCl₃) δ : 0.90 (d, J=6.8 Hz, 0.4H), 0.99 (d, J=7.1 Hz, 0.8H), 1.27 (t, J=7.1 Hz, 3H), 2.50 (m, 0.26H), 2.95 (m, 0.6H), 4.16 (q, J=7.1 Hz, 2H), 5.49 (dq, J=10.8, 2.2 Hz, 0.26H), ¹³C-NMR (CDCl₃) major isomer **16**: δ : 14.3 (q), 19.7 (t), 22.5 (q), 25.0 (t), 26.7 (t), 45.7 (d), 60.4 (t), 125.2 (d), 130.3 (s), 174.9 (s); trans isomer **17a**: 14.3 (q), 19.2 (t), 20.3 (q), 24.5 (t), 32.6 (d), 47.9 (d), 60.2 (t), 125.5 (d), 132.0 (d), 176.1 (s); cis isomer **17b**: 14.3 (q), 16.4 (q), 19.2 (t), 25.8 (t), 31.0 (d), 43.5 (d), 60.0 (t), 125.9 (d), 131.8 (d), 176.1 (s); EIMS m/z (relative intensity): 168 (M⁺, 16), 95 (100). HRMS Found: 168.1156, C10H16O2 (M⁺) requires; 168.1150.

(2E)-3,7-Dimethyl-2,7-octadien-1-ol acetate (18) and (2E)-3.7-Dimethyl-2,6-octadien-1-ol acetate (19)

Isolated as a 7:3 mixture of inseparable regioisomers **18** and **19** in 71% yield after column chromatography (eluted with hexane-ether=20:1). IR (CHCl₃) v max (cm⁻¹): 2971, 2939, 2861, 1728, 1444, 1240; ¹H-NMR (CDCl₃) δ : 2.05 (s, 3H), 4.59 (d, J=7.1 Hz, 2H), 4.67 (m, 0.7 H), 4.71 (m, 0.7 H), 5.09 (m, 0.3 H), 5.35 (tdt, J=7.1, 1.5, 1.2 Hz, 2H); HRMS Found: 136.1262, C10H16 (M⁺-AcOH) requires; 136.1252. (*2E, 6E*)-8-acetoxy-2,6-dimethyl-2,6-octadien-1-ol was isolated as sub-product in 17% yield (eluted with hexane-ether=1:1).

(1R, 2S, 5S, 9R)-2-Benzyloxy-9-benzyloxymethyl-5-methylbicyclo[3.3.1]-6-nonene (20a) and (1R, 2S, 5S, 9R)-2-

Benzyloxy-9-benzyloxymethyl-5methylbicyclo[3.3.1]-7-

nonene (20b)

Isolated as a 3:7 mixture of separable regioisomers 20a and 20b in 99% yield after column chromatography (eluted with hexaneether=20:1). The physical data of compound 20a was reported in the literature¹⁰. Compound **20b** showed the following physical data. (a colorless oil). $[\alpha]_D^{25}$ -102.1° (c 0.9, CHCl3); IR (CHCl3) v max (cm⁻¹): 3010, 2950, 2874, 1454, 1095, 1072; ¹H-NMR (CDCl₃) δ: 0.87 (s, 3H), 1.31 (dddd, J=13.4, 5.4, 2.4, 1.9 Hz, 1H), 1.45 (tdd, J=13.4, 6.8, 1.4 Hz, 1H), 1.62-1.81 (m, 2H), 1.90 (ddd, J=18.6, 3.9, 1.9 Hz, 1H), 1.99-2.14 (m, 2H), 3.01(m, 1H), 3.50 (t, J=9.3 Hz, 1H), 3.59 (dd, J=10.8, 5.6 Hz, 1H), 3.61 (dd, J=10.8, 5.6 Hz, 1H), 4.43 (d, J=12.2 Hz, 1H), 4.48 (d, J=12.2 Hz, 1H), 4.53 (d, J=12.2 Hz, 1H), 4.57 (d, J=12.2 Hz, 1H), 5.78 (ddt, J=9.8, 5.8, 1.9 Hz, 1H), 5.87 (ddd, J=9.8, 3.9, 2.9 Hz, 1H), 7.20-7.38 (m, 10H); ¹³C-NMR (CDCl₃) δ: 25.3 (t), 28.6 (q), 31.4 (s), 35.6 (d), 35.7 (t), 42.9 (t), 44.7 (d), 68.6 (t), 69.8 (t), 73.1 (t), 74.8 (d), 127.3 (d), 127.5 (d), 128.2 (d), 128.3 (d), 128.4 (d), 129.8 (d), 138.6 (s), 139.2 (s); EIMS m/z (relative intensity): 362 (M⁺, 3), 271 (100); Anal. Calcd for C₂₅H₃₀O₂: C, 82.83; H, 8.34. Found: C, 82.84; H, 8.49.

 $(1\alpha, 4\alpha, 4a\alpha, 5\beta, 8a\alpha)$ -1, 4, 4a, 5, 6, 8a-Hexahydro-1,4methanonaphthalene-5-ol formate (22)

After work-up, the residue was chromatographed on silica gel (eluted with pentane) to give **23** in 59% yield as a colorless oil and (eluted with hexane-ether=200:1) to give **22** in 26% yield as a colorless oil. Compound **22** showed the following physical data. IR (CHCl₃) v_{max}

 (cm^{-1}) : 3028, 2972, 2931, 1712, 1190; ¹H-NMR (CDCl₃) δ : 1.34 (dt, J=8.3, 1.4 Hz, 1H), 1.47 (dt, J=8.3, 1.7 Hz, 1H), 1.90 (dddd, J=15.7, 10.0, 2.7, 1.7 Hz, 1H), 2.11 (dtd, J=15.7, 6.4, 1.4 Hz, 1H), 2.82 (m, 3H), 3.02 (brs, 1H), 5.28 (dt, J=10.0, 6.4 Hz, 1H), 5.45 (ddd, J=9.8, 6.4, 1.9 Hz, 1H), 5.73 (dddd, J=9.8, 4.3, 2.7, 0.9 Hz, 1H), 6.06 (dd, J=5.6, 2.7 Hz, 1H), 6.10 (dd, J=5.6, 2.7 Hz, 1H), 8.14 (d, J=0.9 Hz, 1H); ¹³C-NMR (CDCl₃) δ : 28.3 (t), 41.6 (d), 41.6 (d), 45.8 (d), 46.0 (d), 50.3 (t), 72.8 (d), 123.3 (d), 130.7 (d), 135.0 (d), 136.0 (d), 160.9 (d); EIMS m/z (relative intensity): 191 (M⁺+1, 48), 144 (6), 66 (100); HRMS Found: 190.0976, C12H14O2 (M⁺) requires; 190.0994.

 $(1\alpha, 4\alpha, 4a\alpha, 5\beta, 8\beta, 8a\alpha)$ -1, 4, 4a, 8a-Tetrahydro-1,4methanonaphthalene (23)

Compound **23** was unstable because the retro Diels-Alder reaction gradually occurred and decomposed into benzene and cyclopentadiene, which were observed by ¹H-NMR during measurement. IR (CHCl₃) v max (cm⁻¹): 3063, 3031, 3007, 2970, 2939, 2891, 2867, 1339, 1210, 908; ¹H-NMR (CDCl₃) δ : 0.99 (dddd, J=7.8, 3.4, 1.4, 0.7 Hz, 1H), 1.14 (ddd, J=7.8, 1.9, 1.4 Hz, 1H), 2.91 (dt, J=3.4, 1.4 Hz, 2H), 3.05 (ddd, J=3.4, 1.9, 1.4 Hz, 2H), 5.55 (ddd, J=9.5, 3.4, 1.9 Hz, 2H), 5.60 (dt, J=9.5, 3.4 Hz, 2H), 6.08 (t, J=1.7 Hz, 2H); ¹³C-NMR (CDCl₃) δ : 39.9 (d), 44.1 (t), 48.5 (d), 121.4 (d), 128.8 (d), 135.9 (d). Triene **23** was further characterized by its Diels-Alder adduct with N-phenyltriazoline-3,5-dione⁷, which was recrystalized from toluene-pentane (mp 216-217°C, Lit¹⁴. 214-215°C). Anal. Calcd. for C19H17N3O2: C, 71.46; H, 5.37. Found: C, 71.51; H,5.39. (1S)- $(1\alpha, 4\alpha, 4a\alpha, 5\beta, 8a\alpha)$ -1, 4, 4a, 5, 8, 8a-Hexahydro-1,4methanonaphthalene-5-ol acetate (24a) and (1S)- $(1\alpha, 4\alpha, 4a\alpha, 5\beta, 8a\alpha)$ -1, 4, 4a, 5, 6, 8a-Hexahydro-1,4methanonaphthalene-5-ol acetate (24b)

After work-up, the residue was chromatographed on silica gel (eluted with pentane) to give **23** in 30% yield as a colorless oil and (eluted with hexane-ether=200:1) to give a 1:1 mixture of inseparable regioisomers **24a** and **24b** in 54% yield. IR (CHCl₃) v max (cm⁻¹): 3025, 2968, 2937, 1724, 1371, 1253, 1027; ¹H-NMR (CDCl₃) δ : 1.29-1.38 (m, 1.5H), 1.45 (dt, J=8.1, 1.7 Hz, 0.5H), 1.78-1.90 (m, 1H), 199-2.24 (m, 4H), 2.54 (tt, J=9.3, 3.7 Hz, 0.5H), 2.75-2.84 (m, 2.5H), 2.85-3.10 (m, 1.5H), 5.14 (dt, J=10.5, 6.3 Hz, 0.5H), 5.38-5.57 (m, 1H), 5.59-5.75 (m, 1H), 5.93 (ddd, J=5.4, 3.4, 1.0 Hz, 0.5H), 6.01 (ddd, J=5.4, 3.4, 1.0 Hz, 0.5H), 6.07 (m, 1H); ¹³C-NMR (CDCl₃) δ : 21.3 (q), 21.5 (q), 25.6 (t), 28.3 (t), 34.8 (d), 41.6 (d), 45.7 (d), 46.0 (d), 46.3 (d), 48.8 (d), 49.7 (t), 50.2 (t), 69.9 (d), 135.9 (d), 137.2 (d); EIMS m/z (relative intensity): 144 (M⁺-AcOH, 75), 66 (100); HRMS Found: 144.0943, C11H12 (M⁺-AcOH) requires; 144.0939.

Benzyl-3-butenyl ether (25) , Benzyl-2-(*E*)-butenyl ether (26a) and Benzyl-2-(*Z*)-butenyl ether (26b)

Isolated as a 6:2:2 mixture of inseparable regioisomers **25**, **26a** and **26b** in 94% yield after column chromatography (eluted with hexaneether=20:1). IR (CHCl₃) v max (cm⁻¹): 3032, 3012, 2928, 2861, 1454, 1093, 909; ¹H-NMR (CDCl₃) major isomer **25a**: δ : 2.36 (dt, J=6.8, 1.5 Hz, 1H), 2.41 (dt, J=6.8, 1.5 Hz, 1H), 3.53 (t, J=6.8 Hz, 2H), 4.52 (s, 2H), 5.05 (ddd, J=10.8, 1.7, 1.5 Hz, 1H), 5.10 (ddt, J=16.2, 1.7, 1.5 Hz, 1H), 5.68 (dddd, J=16.2, 10.8, 6.8, 1.5 Hz, 1H), 7.24-7.38 (m, 5H); minor isomer **25b**: δ : 1.64 (brd, J=6.8 Hz, 3H), 4.09 (brd, J=6.8 Hz, 2H), 5.60 (dddd, J=12.2, 6.8, 1.5, 1.2 Hz, 1H), 5.85 (dd, J=12.2, 6.8 Hz, 1H); minor isomer **25c**: 5.79 (t, J=6.8 Hz, 1H), 5.90 (t, J=6.8 Hz, 1H); EIMS m/z (relative intensity): 162 (M+, 33), 91 (100); HRMS Found: 162.1042, C11H14O (M+) requires; 162.1045.

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References and Notes

- Entwistle, I. D. and Wood, W. W., "Comprehensive Organic Synthesis ", Trost, B. M., Ed.; Pergamon Press, Oxford, 1991, Vol. 8, pp 955-981, and references cited therein.
- (2) Tsuji reported the palladium-catalyzed hydrogenolysis of the same substrate with excellent regioselectivity. See reference. Mandai,T., Matsumoto, T. and Tsuji, T., Synlett, 1993, 113.
- (3) Inanaga reported a similar result that optical activity was completely lost in the formation of limonene from (-)-carvyl acetate.

See reference. Tabuchi, T., Inanaga, J. and Yamaguchi, M., *Tetrahedron Lett.*, **1986**, <u>27</u>, 601.

- (4) Shibuya, K., Nagaoka, H. and Yamada, Y., J. Chem. Soc. Chem. Commun., 1991, 1541.
- (5) (a) Hutchins, R. O., Learn, K. and Fulton, R. P., *Tetrahedron Lett.*, **1980**, <u>21</u>, 27. (b) Keinan, E. and Greenspoon, N., *ibid*, **1982**, <u>23</u>, 241. (c) Hutchins, R. O. and Learn , K., *J. Org. Chem.*, **1982**, <u>47</u>, 4382. (d) Keinan, E. and Greenspoon, N., *J. Org. Chem.*, **1983**, <u>48</u>, 3545. (e) *idemn.*, *ibid.*, **1988**, <u>53</u>, 3723. and see references (3), (6) and (7).
- (6) (a) Mandai, T., Matsumoto, T., Kawada, M. and Tsuji, T., J. Org. Chem., 1992, <u>57</u>, 1326. (b) idemn., Tetrahedron., 1993, <u>49</u>, 5483.
- (7) Trost, B. M. and Tometzki, G. B., Synthesis, 1991, 1235.
- (8) Muramatsu, L., Murakami, M., Yoneda, T. and Hagitani, A., Bull. Chem. Soc. Japan, 1965, <u>38</u>, 244.
- (9) Umbreit, M., A. and Sharpless, K., B., J. Am. Chem. Soc., 1977, <u>99</u>, 5526.
- (10) Nagaoka, H., Shibuya, K. and Yamada, Y., *Tetrahedron*, **1994**, <u>50</u>, 661.
- (11) Rosenheim, O. and Staring, W. W., J. Chem. Soc., 1937, 377.

- Takano, S., Moriya, M., Higashi, K. and Ogasawara, K., J. Chem. Soc. Chem. Commun., 1993, 177.
- (13) Naruta, Y., Nagai, N. and Maruyama, K., Chem. Lett., 1983, 1383.
- (14) Rye, A. R. and Wege, D., Aust. J. Chem., 1974, 27, 4943.

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